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Current challenges on the widespread adoption of new bio-based fertilizers: insights to move forward toward more circular food systems

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To meet global food demands sustainably, it is necessary to safeguard finite natural resources and reduce harmful emissions to the environment. Nutrients in biowastes are often not managed appropriately. Instead, they can be recovered, recycled into bio-based fertilizers (BBFs) and reincorporated into food production systems. This review addresses three critical issues for developing and adopting new BBFs, focusing on the European context: (1) BBFs should match the agronomic efficiency of mineral fertilizers. We propose that the agronomic efficiency of BBFs can be increased through pre-treating the inputs in biowaste transformation processes (e.g., anaerobic digestion), chemical treatments of existing BBFs, organo-mineral combinations, and soil placement strategies. (2) Production and use of new BBFs is not free of environmental impacts, and these are influenced by regional conditions. (3) Public perception and end-user preferences play a significant role in the adoption of BBFs. Therefore, it is vital to address the requirements of end-users of BBFs. Our findings indicate that for widespread adoption, BBFs need sufficient and reliable nutrient amounts and crop-adequate ratios, as well as competitive pricing. A key advantage of BBFs over mineral fertilizers is their ability to improve soil fertility. However, farmers also require fertilizers that can be handled and applied with existing machinery and offer the practicality of commercial products. Another important aspect is the willingness of consumers to buy products fertilized with BBFs. Designing and promoting BBFs requires a careful assessment of environmental impacts and regional conditions, as the sustainability of BBFs depends on factors like energy sources and biowaste transport distances. Ultimately, the goal is to promote a circular economy and not just to substitute mineral fertilizers with new products. This review aims to guide researchers, policymakers, and stakeholders by highlighting key innovations and addressing critical barriers.

KEYWORDS

sewage sludge, bio-based fertilizer, compost, circular economy, fertilizer

1 Introduction

An increase in agricultural productivity is necessary to properly feed a growing human population. However, improving the sustainability of global food production is also a priority. Agriculture is responsible for 17 % of global greenhouse gas (GHG) emissions (FAO, 2021). Additionally, current agriculture significantly contributes to the degradation of ecosystems and causes the depletion of freshwater and nutrient reserves (Foley et al., 2011; Poore and Nemecek, 2018). Making agriculture more sustainable can be achieved through recovering and recycling nutrients into bio-based fertilizers (BBFs), which can be reincorporated into food production systems.

The concept of planetary boundaries (Rockström et al., 2009) defines the limits of nine key systems crucial for maintaining Earth's safety and habitability. One of these key systems is the biogeochemical flow of phosphorus (P) and nitrogen (N). Currently, 22.6 Tg of P and 190 Tg of N are emitted yearly, doubling and tripling, respectively, the planetary boundaries for each nutrient (Richardson et al., 2023). The main driver of exceeding the system's boundaries is the continuous and excessive P and N inputs in agriculture (Sandström et al., 2023). Correcting this issue requires recycling, improving retention, and utilizing nutrients more efficiently (Carpenter and Bennett, 2011; Sandström et al., 2023).

Fertilizer nutrient use occurs mostly within an inefficient linear economy: a take-make-dispose approach (Chojnacka et al., 2020; Donner et al., 2020). *Take* refers to mineral fertilizers, which are derived, *taken*, from phosphate rock deposits or synthesized through energy-intensive industrial processes, as in the case of Haber-Bosch N. Then, to *make* food and grow crops, mineral fertilizers are applied to agricultural land. As consumption of agricultural products is centralized in urban areas, nutrients concentrate in urban wastewater and end up being *lost* to the atmosphere, aquatic environments or *disposed* in landfills (Witek-Krowiak et al., 2022). This perpetuates the inefficient system as new mineral fertilizers need to be continually produced to maintain agricultural outputs.

On a global scale, P fertilizer use efficiency is between 9 and 12% for cereal crops (Yu et al., 2021), while N fertilizer use efficiency averages between 48 and 78% in croplands (You et al., 2023). In Europe, P fertilizer use efficiency is 57% (Schoumans et al., 2015). Despite improvements in recent decades, significant P and N surpluses persist, leading to eutrophication and other forms of ecosystem damage (Bouwman et al., 2013; Reid et al., 2018; Ural-Janssen et al., 2023; Muntwyler et al., 2024). For example, extracting mineral P generates hazardous mine tailings (Silva et al., 2022). Production of synthetic Haber-Bosch N is responsible for *ca.* 2% of carbon emissions globally and relies on non-renewable natural gas (Osorio-Tejada et al., 2022). These problems highlight an urgent need to shift to circular nutrient economies (Foley et al., 2011).

A circular economy is a system where materials are kept inside, instead of disposed, thus minimizing external resource usage and energy requirements (Ritzén and Sandström, 2017). Achieving this involves reducing nutrient losses during food production and consumption and recovering and reutilizing nutrients (Harder et al.,

2021). Sourcing fertilizers from P and N that otherwise would be wasted, significantly contributes to shaping circular nutrient economies, as opposed to mined P or synthetic N.

The concept of Doughnut Economics (Raworth, 2017) emphasizes that circular economies should exist within the boundaries of meeting human economic and social needs, while safeguarding the planet's ecosystems. In consequence, the production and use of fertilizers with recovered nutrients must develop in conjunction with the agricultural, economic, and social needs of the specific context in which it takes place, involving stakeholders in the decision-making process (Schengel and Goehlich, 2024). Furthermore, it is essential to consider the available P and N sources, the existing infrastructure, and the regulatory and market conditions (Moshkin et al., 2023; Garmendia-Lemus et al., 2024).

In the European Union (EU), approximately 1 Tg of mineral P fertilizers and 10 Tg of synthetic N fertilizers are applied annually (Eurostats, 2023). Meanwhile, considerable sources of P and N, such as sewage sludge, are underutilized compared to the United States and Australia (Marchuk et al., 2023). Therefore, the EU is attempting to transition to a circular economy and maximize the recovery of nutrients (Sporchia and Caro, 2023). At the forefront of the EU's efforts is the European Green Deal, a set of policy initiatives with the goal of making the EU climate neutral by 2050 (European Commission, 2019). The food system and agricultural core of the European Green Deal is the Farm to Fork Strategy, which aims to overhaul policy and regulation to create a more circular European food system. From an agricultural perspective, one of the central aims of the Farm to Fork Strategy is to reduce nutrient losses by at least 50% (Heyl et al., 2023).

The required changes are gradually being incorporated into policy and legislation, reflected in the Common Agricultural Policy reforms, which now promote a precision approach to plant nutrition to improve nutrient use efficiency (Heyl et al., 2023). The Circular Economy Action Plan (European Commission, 2020) has underlined the need to foster nutrient recycling and facilitate the reincorporation of waste streams into agriculture. Reincorporation and upcycling/recycling of nutrient-rich wastes into new circular fertilizer products replacing synthetic fertilizers has been supported by the recent updates in the Fertilizing Product Regulation (Regulation (EU) 2019/2009). These policy and regulation changes have caused a renewed interest in approaches for incorporating waste-derived nutrients into fertilizers.

In this context, the EU has committed significant research efforts through initiatives such as Horizon 2020 and Horizon Europe. Among these initiatives is the EU-funded FertiCycle project, a Marie Skłodowska-Curie Actions Innovative Training Network, in which all authors of this review paper were actively involved. The FertiCycle project aimed at proposing solutions for nutrient recovery and recycling within the European framework, addressing agronomical, environmental, and economic challenges associated with the development and adoption of improved and new bio-based fertilizers, and thus contributing to a more circular nutrient economy in the EU. The aim of this review is to present the contributions of the FertiCycle project within the context of critical challenges and current state-of-the-art literature on BBFs in the EU, guiding researchers,

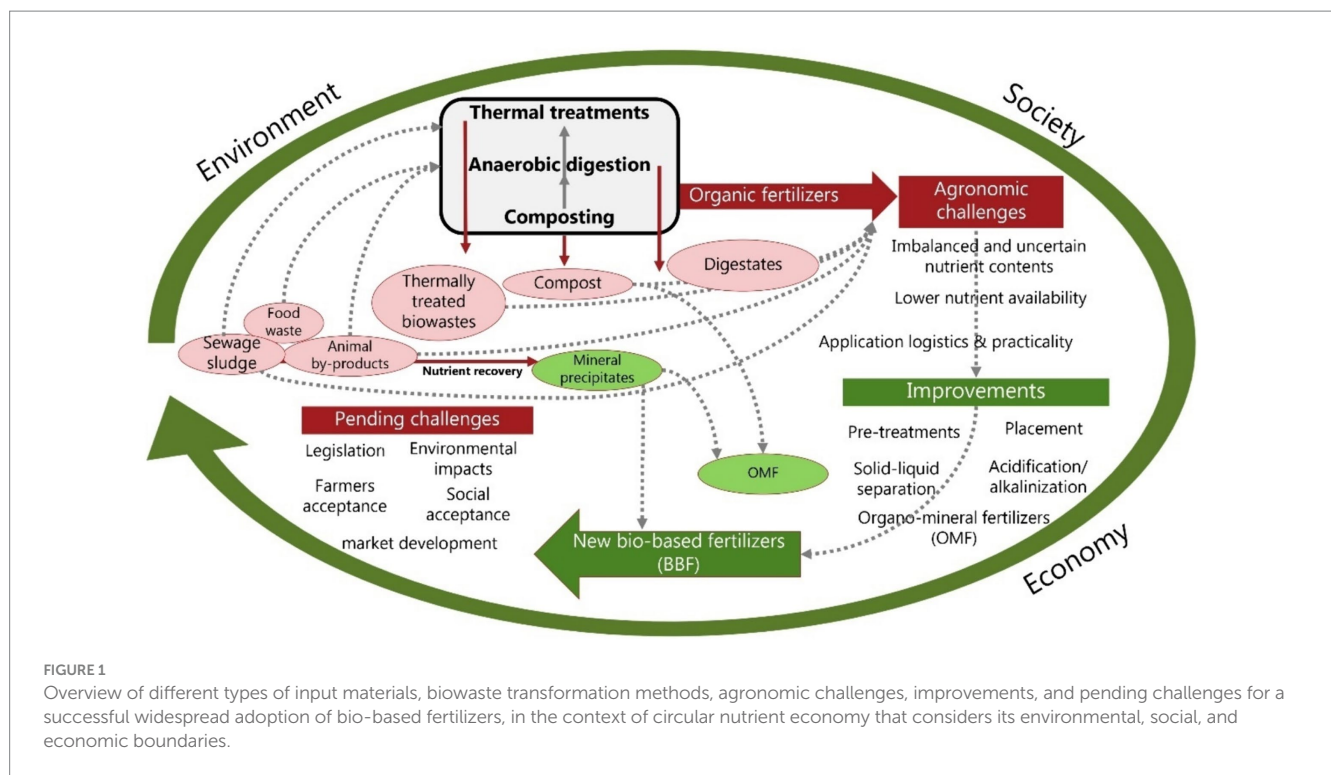


FIGURE 1 Overview of different types of input materials, biowaste transformation methods, agronomic challenges, improvements, and pending challenges for a successful widespread adoption of bio-based fertilizers, in the context of circular nutrient economy that considers its environmental, social, and economic boundaries.

policymakers, and stakeholders by highlighting key innovations and addressing critical barriers for BBF adoption. Figure 1 provides an overview of the agronomic challenges, improvements, and pending challenges presented and discussed in this review.

2 What is a bio-based fertilizer?

The use of the term bio-based fertilizer or BBF has increased lately in academic publications for describing a broad category of materials of biological origin used to supply nutrients to crops (Chojnacka et al., 2020; Wester-Larsen et al., 2022; Rodríguez-Alegre et al., 2023; Kurniawati et al., 2023a). However, distinguishing BBFs from organic fertilizers and biowastes presents a challenge.

The term “biowaste” encompasses materials of biological origin discarded as wastes, including byproducts from agriculture, food production, animal husbandry, and human life. Nutrient-rich biowastes, when applied to soils to provide nutrients for plants, are included under the category of organic fertilizers (Pain and Menzi, 2011). In everyday language, a distinction is made between a general biowaste and an organic fertilizer, only based on their actual usage. Manure, defined by the Oxford English Dictionary as “Dung, excrement or compost, esp. as spread over or mixed with soil to fertilize it” is a prevalent example.

The term “animal manure” describes a mixture of animal feces, urine and bedding material that is utilized as a fertilizer (Shober and Maguire, 2018). Animal manures often undergo some degree of processing, such as anaerobic storage, addition of water or homogenization to facilitate its application to the field. However, the term BBF is employed in literature advocating alternatives to mineral fertilizers or describing materials of enhanced fertilizer value, implying a more technologically advanced material than a basic, raw organic fertilizer such as an animal manure (Luo et al., 2022; Egas

et al., 2023; Moshkin et al., 2023; Garmendia-Lemus et al., 2024). Therefore, it is arguable that an animal manure is not a BBF, since it had no treatment specifically to improve its fertilizing qualities.

The core value of a BBF lies in its circularity (Chojnacka et al., 2020). Bio-based fertilizers are intended to allow the reincorporation of nutrients that otherwise would be wasted, have an improved nutrient use efficiency compared to a less processed organic fertilizer, or decrease possible negative environmental impacts compared to its organic or mineral fertilizer counterparts. Furthermore, organic fertilizers can be associated with containing organic matter and being capable of, e.g., improving soil structure or being a source of organic carbon. However, a BBF can also include nutrients derived from biomass, in the absence of organic matter, and resemble mineral fertilizers in their chemical and physical forms, as long as the nutrients are from biological origin.

The concept of biological origin is one of the main points of discussion for recognizing a fertilizer as bio-based according to the European Sustainable Phosphorus Platform (ESPP, 2023). In the context of bioplastics, the European Standard EN 16575 (August 2014) recognizes the biological origin, when the material is wholly or partly derived from biomass renewable during human lifespans, i.e., excluding materials fossilized and embedded in geological formations.

The European Union updated Fertilizing Product Regulation (FPR, regulation EU 2019/1009) has not yet established a precise definition for BBFs. However, it distinguishes between materials of inorganic and organic origin and introduces categories for nutrients of biological origin. The FPR incorporates so-called product function categories (PFC) into legislation, which categorize different types of fertilizing products based on their primary purpose. Products within these categories need to be composed of materials recognized within specific component material categories (CMCs). Therefore, for a bio-based product to be recognized as a fertilizer, or be included as a part of one, its components need to fit within a CMC. Relevant PFCs

TABLE 1 Product function categories (PFC) and sub-categories from the updated fertilizing product regulation (Regulation (EU) 2019/1009) pertinent to new bio-based fertilizers.

Category	Description	Sub-categories	Component material categories (CMCs)
PFC 1	Fertilizers - Products with the primary function of supplying nutrients to plants or mushrooms	PFC 1 (A): Organic Fertilizers PFC 1 (B): Organo-Mineral Fertilizers PFC 1 (C): Inorganic Fertilizers	CMC 2: Plants or Plant Extracts CMC 3: Compost CMC 4: Fresh Crop Digestate CMC 5: Non-Fresh Crop Digestate CMC 6: Food Industry By-products CMC 12: Precipitated phosphate salts & derivatives (e.g., struvite) CMC 13: Thermal oxidation materials & derivatives (e.g., ashes) CMC 14: Pyrolysis & gasification materials CMC 15: Recovered high purity materials (e.g., ammonium salts, purity >95%)
PFC 7	Product Blends - Products that fit into two PFCs, e.g., a product designed to supply nutrients to plants but also acts as a soil improver		

and CMCs are summarized in Table 1. Main novelties include CMC 12, which regulates the use of recovered phosphate from, e.g., wastewater treatment plants. This category sets limits of tolerance for chemical impurities and pathogens, and allows materials of less purity than those classified under CMC 15. Animal by-products are still outside the scope of the FPR (EC, 2022), and regulated under the Animal By-products Regulation EC 1069/2009 (EC, 2022). However, this will change once the European Commission establishes endpoints in the processing chain for individual by-products. Currently, there is a draft proposal for the inclusion of animal by-products in the FPR.¹

The clear definition of the term BBF is challenging and still debatable. One definition that has been proposed is that BBFs are materials or products derived from materials of biological origin, with a content of bioavailable nutrients high enough to be used as a fertilizer (Wester-Larsen et al., 2022). On the other hand, Rodríguez-Alegre et al. (2023) proposed that waste streams, i.e., manure, sewage sludge and food processing wastes, which can be valorized, through processing, into a fertilizer product should be termed BBFs. Both definitions indicate that some degree of processing is necessary for a biowaste or an organic fertilizer to acquire BBF status. Moreover, literature also has assigned bio-based status to fertilizers where the nutrients are recovered (e.g., via ammonia stripping or P precipitation) from biowastes, or when they are contained in new fertilizing products combining synthetic nutrients with those of biological origin.

It is not our goal to replace the term “organic fertilizer” with BBF. It is rather to provide an overview of the challenges and opportunities for reincorporating nutrients from biowastes into the food system and improving the agronomic efficiency of existing organic and bio-based fertilizers. Therefore, creating new, improved BBFs. For the purpose of this review, we will consider a BBF, any material of biological origin (e.g., a biowaste), that has been treated with the purpose of enhancing its fertilizing properties (e.g., an acidified slurry); as well as fertilizing products, where a significant proportion of its nutrients is of biological origin, for example, an

organo-mineral fertilizer containing a combination of mined rock phosphate and ammonium nitrate gained through ammonia stripping of manure (Wester-Larsen et al., 2022).

3 From biowastes to new bio-based fertilizers

The importance of BBFs in a more sustainable agriculture is underscored by the abundance of waste materials such as agricultural and food waste (e.g., olive pomace), sewage sludge and animal excreta (Kummu et al., 2012; Mateo-Sagasta et al., 2015; Shober and Maguire, 2018). These biowastes contain carbon (C), P and N, making them valuable but potentially harmful for the environment if improperly handled (Chojnacka et al., 2020).

One of the main challenges in utilizing biowastes and organic fertilizers as nutrient sources is maximizing the availability of the contained P and N, which are often only partly or slowly available (Chojnacka et al., 2020). Raw biowastes also pose challenges such as nutrient imbalances relative to crop demand, pathogen contents and high volumes, requiring chemical and physical modifications (Overmeyer et al., 2023; Zireeni et al., 2023; Kopp et al., 2023b). Biowaste transformations (e.g., composting, anaerobic digestion, pyrolysis) designed to tackle logistical or environmental challenges can negatively impact the nutrient availability (Jameson et al., 2016).

Biowastes undergo several treatments depending on legal, economic, or environmental requirements. For example, cattle excreta are composted, or subjected to ammonia stripping; sewage sludge can be anaerobically digested, composted, dewatered, or incinerated (Kanteraki et al., 2022). An overview of available input materials for new BBFs is given in Table 2.

3.1 Phosphorus and nitrogen in raw materials for new bio-based fertilizers

3.1.1 Phosphorus

Increasing the P fertilizer value in biowastes is challenging, due to the diverse chemical forms of inorganic P (Pi) and organic P (Po), influenced by the biowaste source and processing (Meyer et al., 2018).

¹ The latest version of the draft amendment of the FPR was elaborated on October 2, 2023, and can be consulted under the document name Ares(2023)6629522.

TABLE 2 Types of potential main input materials for new bio-based fertilizers.

Material	Description	Examples of transformation methods into BBF	References
Animal by-products	E.g. Raw animal excreta, meat, and bone meal	Acidification, alkalization	Cao et al. (2020), Chrysanthopoulos et al. (2024), Fangueiro et al. (2015), Sica et al. (2023), and Zireeni et al. (2023)
Compost	Biowastes that went through a composting process	Incorporation into organo-mineral fertilizers	Sitzmann et al. (2024)
Digestates	Solid or liquid output of the anaerobic digestion process of biowastes	Biomass pre-treatments Electrokinetic pre-treatment Ensiling pre-treatment Re-digestion after solid-liquid separation	Carlsson et al. (2012) Nyang'au et al. (2023a) Nyang'au et al. (2023b) Aguirre-Villegas et al. (2019)
Mineral precipitates or concentrates	Mineral compounds extracted, separated, or precipitated from biowastes (e.g., struvite, ammonium sulfate)	Incorporation into organo-mineral fertilizers and direct application	Fangueiro et al. (2017), Hušek et al. (2022), Sena et al. (2021), Zabaleta and Rodic (2015)
Plant based materials	Plant juices, extracts, and solids, olive pomace	Composting, anaerobic digestion, solid-liquid separation	Ameziane et al. (2020), Muscolo et al. (2019) and Sorensen and Thorup-Kristensen (2011)
Sewage sludge	By-product of wastewater treatment plants that concentrates solids of sewage.	Acidification, alkalization Incorporation into organo-mineral fertilizers	Sica et al. (2023) Deeks et al. (2013)
Thermally treated biowastes	Biowastes converted into ash or biochars via incineration or pyrolysis, respectively.	Pre-treatments	Kopp et al. (2023a)

List was originally categorized by Wester-Larsen et al. (2022).

We added sewage sludge, due to the different legislation and fertilization requirements: sewage sludge is excluded from the FPR and most European countries have stricter regulations on land application of sewage sludges compared to other biowastes (Gianico et al., 2021). Some European countries ban the land application of sewage sludge entirely, as in the case of Switzerland (Verordnung Über Umweltgefährdende Stoffe, 2002). Another introduced category was “thermally treated materials”.

Phosphorus is plant available only in the form of dissolved inorganic orthophosphates (Tate, 1984).

Most of P in manure and feces is water-soluble Pi, whereas Po varies based on the feed and animal and type (Barnett, 1994; Poulsen and Kristensen, 1998). Dairy cattle manures contain between 6g·kg⁻¹ and 16g·kg⁻¹ of total P, mostly Pi (Barnett, 1994). Poultry manures contain between 13g·kg⁻¹ and 23g·kg⁻¹ of total P, have variable proportions of Pi and Po, and are rich in calcium-bound Pi forms (Barnett, 1994; Neijat et al., 2011). One of the most important challenges regarding manures are their imbalanced P to N ratios, that may result in excessive P application when fertilization aims to meet plant N demand (Sharpley, 1996). High concentrations of water-soluble Pi in and organic compounds have been linked to high P leaching in soils treated with manure (Glæsner et al., 2011). The transformation of manure with composting significantly concentrates the contained P due to dry matter loss, and reduces P solubility (Zhang et al., 2023).

Phosphorus in composts is predominantly inorganic (Frossard et al., 2002; Zhang et al., 2023). In an evaluation of 16 samples of composted Swiss solid urban biowastes and woody wastes, Frossard et al. (2002) measured P concentrations between 2g·kg⁻¹ and 7g·kg⁻¹, mostly in the form of Pi. Phosphorus in compost included both water-soluble Pi as well as relatively insoluble calcium phosphates like apatites or octacalcium phosphates. Sewage sludges contain Po and Pi forms of low solubility [e.g., associated to iron (Fe), aluminum (Al) or calcium (Ca)] influenced by the diversity of P removal techniques (e.g., flocculation with iron or aluminum salts or biological P removal) used in wastewater treatment plants (Frossard et al., 1997; Wang et al., 2022). Anaerobic digestion (AD) of sludges can increase the crystallinity of Pi forms and leave behind only recalcitrant Po (Wang et al., 2022). Alkalization by liming, done for sanitization purposes

(Anderson et al., 2015; Malinowska, 2017), produces the formation of calcium phosphates that can become plant available in acidic soil conditions (Lindsay, 1972; Meyer et al., 2018). Solubility of P in sewage sludge is strongly decreased during thermal treatments for stabilization or volume reduction (e.g., pyrolysis or incineration), causing sewage sludge ashes to contain little available Pi due to the formation of stable P mineral forms (Nanzer et al., 2014; Lemming et al., 2017). Sewage sludge ashes and meat and bone meal have low mineral P fertilizer equivalencies, of 30 and 40%, respectively; untreated sewage sludge typically ranges from 60 to 90%, depending on the method used for P removal (Möller et al., 2018). The P in meat and bone meal is mainly in the form of hydroxyapatite (Jeng et al., 2007).

3.1.2 Nitrogen

The challenges associated with the N fertilizer value from biowastes include the diversity of N forms in biowastes and the losses through collection, treatment, storage, and soil application (Lassaletta et al., 2014; Fangueiro et al., 2015; Nigussie et al., 2017). In biowastes, N is usually contained in organic forms (urea and other amides, amino acids, undigested feed and food protein residues, nucleic acids) with variable amounts of inorganic plant available forms like ammonium (NH₄⁺) and nitrate (NO₃) (Langmeier et al., 2002; de Guardia et al., 2010; Bosshard et al., 2011). Human and animal urine contain most N in the form of urea, which is enzymatically transformed shortly after excretion into NH₄⁺. Feces contain more complex N forms, for example, bacterial or endogenous debris for which availability will depend on mineralization processes (Dion et al., 2020).

Anaerobic digestion of manures and sewage sludges increases the inorganic N content (Hafner and Bisogni, 2009; Jiang et al., 2022). During AD, mineralization of N forms takes place (Möller and Müller,

TABLE 3 Aspects regarding the form and contents of phosphorus and nitrogen and agronomic challenges of biowastes and potential input materials for new bio-based fertilizers.

Biowaste	Phosphorus	Nitrogen	Agronomic challenges	References
Cattle manure	6-16 g·kg ⁻¹ total P, mostly soluble Pi	Organic N forms (urea, amino acids), mineral N (NH ₄ ⁺ -N)	Imbalanced P to N ratios, potential excessive P application, risk of P leaching.	Barnett (1994) and Poulsen and Kristensen (1998)
Poultry manure	13-23 g·kg ⁻¹ total P, variable Pi and Po, rich in calcium-bound Pi	Organic N forms (urea, amino acids)	Variable P and N content, calcium-bound Pi, insoluble at neutral to alkaline pH.	Barnett (1994) and Neijat et al. (2011)
Sewage sludge	Very rich in P (up to 40 g·kg ⁻¹)	Dissolved organic N	Impact of P flocculation or removal technique on P availability.	Frossard et al. (1997) and Wang et al. (2022)
Sewage sludge ashes	Pi of very low availability	Low quantity due to thermal treatment	Low P solubility due to formation of crystalline forms.	Lemming et al. (2017) and Nanzer et al. (2019)
Olive Pomace	Raw pomace is poor in available P (<0.02-0.04%)	High C:N ratio	Low pH, high content of organic components that can be phytotoxic, low P and N contents.	Ameziane et al. (2020) and Muscolo et al. (2019)
Meat and bone meal	20-40 g·kg ⁻¹ total P, mainly hydroxyapatite derived from bones	Proteins and amino acids derived from the meat and blood.	Low P solubility due. to stable mineral forms, having a low availability to plants, especially in soils with neutral and alkaline pH.	Brod et al. (2015) and Christiansen et al. (2020)
Mineral precipitates or concentrates, e.g., struvite:	Struvite: > 12.5 % total P (Mg associated P)	E.g. Ammoniacal form when ammonia stripping is used	Although it is already produced commercially the precipitation and recovery processes are costly, making it an expensive fertilizer.	Christiansen et al. (2020) and Li et al. (2019)
Organo-mineral fertilizers (OMF)	>2% P ₂ O ₅ – P that can come from mineral fertilizer (e.g., diammonium phosphate) and/or from P contained in the organic fraction	>2% N, minimum 0.5% organic N (if OMF contains P and/or K), >2.5% N, minimum 1% organic N (OMF with only N as macronutrient). N can come from mineral fertilizer and from the organic fraction	OMF are already commercialized, generally more expensive than equivalent mineral fertilizers. Peat is still the main organic matrix used in Europe. If nutrients come from the organic fraction, challenges are typical of organic fertilizers (low solubility and high variability among batches).	EC (2019), Paré et al. (2010) and Rodrigues et al. (2021)
Composts	2-7 g·kg ⁻¹ total P, mostly Pi, rich in calcium-bound Pi	Inorganic N (NH ₄ ⁺ , NO ₃ ⁻) and organic N forms	Dry matter loss reduces P availability.	Frossard et al. (2002) and Zhang et al. (2023)
Digestates	Depends on the biowaste. Improved (decreased) C:P ratios in, e.g., olive pomace	High inorganic N content, low organic N, decreased C:N ratio	N losses during digestion are lower than composting, but increased crystallinity reduces P availability, high N losses when surface applied.	Hafner and Bisogni (2009), Jiang et al. (2022) and Pedersen et al. (2021)

Nutrient concentrations reported in percentage or g·kg⁻¹ dry matter.

2012). Nitrogen losses during the digestion process are significantly smaller compared to composting or thermal treatments (Chojnacka et al., 2020). Anaerobic digestion leads to decreased C to N ratios in the digestate effluent (Möller and Müller, 2012) as C is converted to biomethane gas (mixture of CO₂ and CH₄), while N is conserved in the digestate. Mechanical separation of solid–liquid phases from the anaerobic digestate concentrates directly available inorganic N (as dissolved ions) in the liquid fraction and slowly available organic N in the solid fraction (Zabaleta and Rodic, 2015; Chojnacka et al., 2020).

Composting of household waste, pig slaughterhouse sludge, and green algae resulted in losses of 36 - 66% of total N to the atmosphere (de Guardia et al., 2010), predominantly caused by ammonia (NH₃) volatilization (Chowdhury et al., 2014). Volatility and speciation of ammoniacal-N in biowastes is influenced by the pH (Morales et al., 2017). Acidification of animal manures prevents N emissions during processing, storage and land spreading and reduces NH₃ volatilization by shifting the equilibrium toward the non-volatile

NH₄⁺ (Fangueiro et al., 2015). Similarly, during storage of biowastes or composting, acidification and pH management minimize N losses (Chowdhury et al., 2014; Cao et al., 2020; Kupper et al., 2020). Delayed applications of N-rich materials during composting can also reduce N losses (Nigussie et al., 2017).

Much of the existing waste management infrastructure prioritizes sanitation and emission reduction rather than recycling nutrients (Magid et al., 2006). For instance, wastewater treatment plants emit substantial amounts of N into the atmosphere through nitrification–denitrification processes aiming to reduce wastewater N (Marchuk et al., 2023). The goal of current and future waste management is to transform these facilities into nutrient recovery centers (Marchuk et al., 2023). The application of chemical and physical treatments to the outputs of waste management infrastructure, coupled with enhancements in existing processes are a step in that direction. Table 3 includes an overview of P and N forms in biowastes, as well as related agronomical challenges.

3.2 Treatments to enhance the P and N fertilizer value of bio-based fertilizers

3.2.1 Pre-treatment of inputs in anaerobic digestion

Anaerobic digestion improves the N fertilizer value of digestates compared to untreated inputs. Anaerobic digestions mineralize decomposable organic matter in slurries, reduce the C to N by converting C to biogas, and increase the $\text{NH}_4\text{-N}$ to total N ratio through organic N mineralization (Sørensen and Møller, 2009). However, with an increasing shift toward using high solid co-substrates such as lignocellulosic agricultural wastes (e.g., straw and other crop residues) with short hydraulic retention times, a smaller proportion of the organic matter (40–60 %) is degraded into biogas and the rest remains in digestates (Romio et al., 2021). Resulting digestates have reduced fertilizer values due to their high dry matter contents, higher C to N ratios and lower infiltration rates enhancing the risk of NH_3 loss (Møller et al., 2022; Pedersen and Hafner, 2023). Biogas plants prioritize biogas yields over digestate quality (Logan and Viswanathan, 2019). This is driven by the higher value of biogas compared to digestate and incentives such as renewable energy targets.

Optimization of AD for enhanced biogas production and also nutrient availability can be achieved through biomass pre-treatments. Biomass pre-treatment techniques, classified as physical, chemical, biological, or a combination of these, have variable effects on biomass utilization in the AD process, dependent on the pre-treatment mechanism and feedstock characteristics. Biomass containing lignin or bacterial cells are the most affected during pre-treatment for enhanced AD process (Carlsson et al., 2012).

Nyang'au et al. (2023a) investigated the effects of electrokinetic and ultrasonication pre-treatments of biowastes in a two-step AD process on nitrogen fertilizer replacement value of digestates obtained from two biogas plants. The electrokinetic pre-treatment step significantly increased the ratio of ammonium-N to total N in the digestates before the second AD step. However, the effect leveled off after the secondary digestion step. The study demonstrated how integrating pre-treatment technologies into biogas plants could improve the fertilizing properties of the digestates.

Another study by Nyang'au et al. (2023b) highlighted the use of ensiling as a biological pre-treatment method to enhance biogas yield and improve the fertilizer value of the digestates. Ensiling significantly impacted physico-chemical properties of straw, increased methane yield by 4 to 14 %, and increased net inorganic N and S release in the soil compared to non-ensiled straw. They attributed the positive effect to enhanced substrate biodegradation during the ensiling, which increased biochemical accessibility and nutrient solubilization during AD.

Pre-treatments can be effective measures to increase the fertilizer value of digestates. However, the selection of a pre-treatment technique should consider net effects, including cost and energy consumption (Meegoda et al., 2018). Many studies solely evaluate pre-treatment benefits by comparing extra energy output against energy consumption, overlooking other potential advantages such as increased digestate fertilizing value.

3.2.2 Acidification of slurries

Emissions derived from barns and slurry storage represent 80% of agricultural NH_3 emissions, which can be mitigated by acidification

to a pH in the range of 4.5–6.8 (Fangueiro et al., 2015). Moreover, utilization of untreated animal slurry in horticulture poses potential risks in terms of food safety. Slurry acidification can be employed to address both NH_3 emissions and food safety risks (Fangueiro et al., 2015; J. Rodrigues et al., 2021).

Utilization of sulfuric acid for slurry acidification represents an addition of available sulfur that increases the fertilizer value of the slurry (Zireeni et al., 2023). Acidification increased the concentration of water-soluble P in 20–65% compared to raw slurry (Regueiro et al., 2020). Moreover, there are benefits associated with the soil application of acidified slurry. In the study conducted by Schreiber et al. (2023) where acidified slurry (pH 5.5) was applied to a Haplic Cambisol (pH 7), a significant increase of 38% in N use efficiency was observed in plant biomass. Zireeni et al. (2023) found that the application of acidified slurry (pH 5.5) to a Cambisol (pH 6.8) temporarily reduced soil pH by at least 0.4 units for up to two months, before the pH went back to its baseline. A temporary acidification of soil can be beneficial in soil conditions where P is a limiting factor.

For slurry acidification, sulfuric acid, and to a lesser extent, nitric and hydrochloric acids are utilized (Fangueiro et al., 2015). However, other additives and processes for bio-acidification involving agro-industrial by-products were tested by Chrysanthopoulos et al. (2024) for acidifying pig slurries. It was shown that bio-acidification of pig slurries through fermentation is possible when the fermentation substrate contains sufficient organic C (Chrysanthopoulos et al., 2024). Moreover, slurry bio-acidification using rice bran, a biowaste rich in N, as a fermentation substrate significantly increased the total N content compared to untreated slurry (Prado et al., 2020). Conclusively, acidification of slurries can be used beyond addressing hygienic concerns and mitigating NH_3 emissions. Acidification can be a way to utilize agro-industrial by products and increase the fertilizer value of slurries.

3.2.3 Acidification and alkalization of sewage sludge, sewage sludge ashes and meat and bone meal

Sica et al. (2023) evaluated the impact of acidification and alkalization pre-treatments on the P solubility of sewage sludge, sewage sludge ashes and meat and bone meal. For acidification, a 2 to 1 biowaste to solution ratio was utilized, with sulfuric acid concentrations ranging from 0.25 M to 10 M. Alkalization was carried out using sodium hydroxide concentrations ranging from 1 M to 2.5 M. For alkalization with lime $[\text{Ca}(\text{OH})_2]$, lime quantities equivalent to 10 to 40% of the fresh weight of the biowaste were added. Acidification resulted in a decrease in the pH of the biowastes to a range between 1 and 4, while alkalization treatments raised the pH of the biowastes between 8 and 12. It was observed that acidification significantly increased the solubility of P, leading to a greater release of P into the soil and an increase in soil water-extractable P. Water-extractable P in the sewage sludge ashes increased up to 60 times. In meat and bone meal, water-extractable P rose from 4% to more than 80% of the total P when pH dropped below 4. Alkalization with sodium hydroxide was found effective in increasing soil P availability in sewage sludge, and potentially providing similar sanitation effects to lime. However, plant trials were not conducted by Sica et al. and they highlighted that the higher costs of sodium hydroxide compared to lime may limit the large-scale application of this pretreatment.

Similarly, Keskinen et al. (2023) used residual organic acids from cellulose extraction to acidify sewage sludge to both pH 7 and 4.5. They observed a significant increase in P solubility at pH 4.5, but it did not impact P uptake by ryegrass. It is important to mention that their results demonstrated that the acidification of sewage sludge solubilizes other elements, including harmful metals, which may harm plant growth (Imadi et al., 2016).

3.2.4 Chemical modifications of thermally treated biowastes to enhance P availability

In the Netherlands, Switzerland, and Belgium, mono-incineration has emerged as the predominant practice (Mininni et al., 2015). Recent legislative changes in Sweden, the Czech Republic, and Denmark have made possible utilizing sewage sludge biochar for agricultural purposes.

Both incineration and pyrolysis offer advantages such as volume reduction, increased P concentration, pathogen elimination or reduction, and potential energy recovery. Additionally, pyrolysis forms recalcitrant C and contributes to soil C sequestration and climate change mitigation (Smith, 2016). However, as previously discussed, thermal treatment decreases the P availability.

The use of additives before pyrolysis can enhance P availability in biochar. The addition of magnesium hydroxide alters the resulting P forms after pyrolysis, mainly by avoiding the formation of crystalline Ca-P minerals in favor of amorphous and more soluble Mg-P forms (e.g., $MgNH_4PO_4$ and $Mg_3(PO_4)_2$), which has resulted in a 20% increase in the P availability of poultry litter biochar (Zwetsloot et al., 2015; Leite et al., 2023). The addition of calcium oxide (CaO) in thermal treatment may offer similar benefits. Liu et al. (2019) found

that addition of 10% CaO converted non-apatite inorganic P forms into apatitic forms, (e.g., $Ca_3(PO_4)_2$ and $Ca_3Mg_3(PO_4)_4$). This increased the sewage sludge biochar P availability and promoted plant growth in acidic soils.

Another alternative is offered by pH modification treatments after pyrolysis or incineration. The effect of acidification with sulfuric acid was assessed in three ashes (sewage sludge, poultry litter, digestate solids) and four biochars (digestate solids, sewage sludge, meat and bone meal, insect frass) (Kopp et al., 2023b). While the P availability from untreated ashes and biochars was very low, the acidification significantly increased the total plant P uptake from all materials from 2 to 35 times. Acidification solubilized mainly Ca-P and did not increase heavy metal uptake nor strongly affected soil pH which remained at 6.4. A synthesis of the effectiveness of alkalization and acidification treatments for P solubilization is provided in Figure 2.

3.2.5 Combined treatments

Integrated approaches throughout manure management chains elevate the fertilizer value and nutrient concentration and mitigate trade-offs between NH_3 and GHG emissions (Aguirre-Villegas et al., 2019). One such effective integrated approach involves coupling anaerobic digestion of biomass with solid-liquid separation of the resulting digestate. In this strategy, the solid fraction is re-digested to enhance further biogas and nutrient recovery. The liquid fraction is then used as a fertilizer, since it is enriched with a higher proportion of NH_4^+ and overall higher N. Alternatively, the solid fraction, rich in P, may be utilized to fertilize P-deficient soil or transformed into value-added products like biochar, contributing to soil C retention (Fangueiro et al., 2015), or used as livestock bedding material.

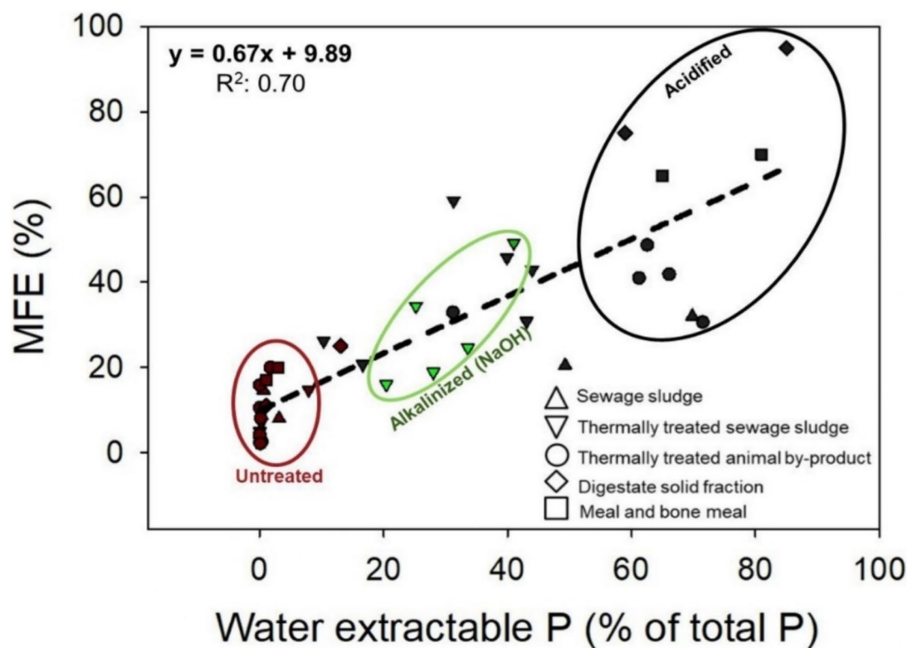


FIGURE 2
Correlation between the P solubility (water extractable P) and the mineral fertilizer equivalent of untreated, alkalinized, and acidified biowastes (Sica et al., 2023; Kopp et al., 2023a,b). The figure shows that biowastes with initially low water-soluble P (<5% of total P content) significantly increased this proportion after acidification, while thermally treated sewage sludge also did so after alkalization. Furthermore, a correlation ($R^2 = 0.7$) between the water-extractable P content of these materials and their mineral fertilizer equivalent values was observed.

Other integrated approaches that could improve the fertilizer value of digestates include AD followed by acidification of the digestates, source segregation of manure followed by anaerobic digestion and AD followed by plasma treatment of digestates. Plasma treatment of digestates fixes reactive N from the atmosphere to the slurries in nitrite and nitrate forms (Graves et al., 2019). The nitrite and nitrate fixation forms nitric acid (HNO₃) and nitrous acid (HNO₂), which lowers the slurry pH and produces the previously mentioned benefits (Winter and Chen, 2021). Plasma treatment of slurry has shown a potential to replace more mineral fertilizer and increase yields compared to untreated slurry (Cottis et al., 2023). Moreover, this treatment reduces both methane (CH₄) and NH₃ emissions during storage and field applications, although the risk of increased losses via nitrous oxide (N₂O) emissions and nitrate leaching needs to be assessed (Cottis et al., 2023). Integrated approaches offer a comprehensive solution for optimizing fertilizer values and contribute to sustainable agricultural practices by minimizing environmental impacts.

3.3 Organo-mineral combinations

Organo-mineral fertilizers (OMFs) are a mixture of materials of biological origin with one or more mineral fertilizers (Smith et al., 2020). This allows the use of BBFs rich in organic C as the organic matrix, such as compost. The use of OMF has been reported to have some advantages over mineral fertilizers alone, as addition of organic C can reduce N losses and increase N use efficiency (Richards et al., 1993; Antille et al., 2014; Florio et al., 2016). Similarly, OMFs increase plant P use efficiency by causing a prolonged release of plant available P in soil (Antille et al., 2013).

The enhanced nutrient use efficiency can be linked to various processes. For instance, there is an electrostatic attraction of phosphates to organic materials, reducing the P mobility in soil that could be otherwise fixed to clay minerals (Gwenzi et al., 2018; Luo et al., 2021). The chemical interactions between the organic matrix and mineral fertilizer can result in the formation of insoluble compounds that precipitate out of the soil solution (Mazeika et al., 2016; Carneiro et al., 2021; Luo et al., 2021). Additionally, the organic material acts as a physical barrier between the mineral fertilizer granule and the soil solution, thereby decreasing the solubility of the fertilizer granule (Limwikran et al., 2018). Lastly, the stimulation of soil microbial activity leads to the immobilization of nutrients into the microbial biomass and a subsequent gradual N release that reduces leaching (Richards et al., 1993; Mandal et al., 2007).

The FPR establishes that at least 3% of the total mass of an OMF must be composed of organic C if it is a liquid OMF, and at least 7.5% on a solid OMF, while the sum of macronutrients needs to be 6% or 8%, respectively (EC, 2019). This broad definition allows the production of OMFs with varying characteristics.

One type of OMF is where the organic matrix is used as the main source of one or more nutrients and mineral fertilizer is added to produce a specific nutrient ratio that will depend on specific soil and plant nutrient requirements (Rady, 2012; Antille et al., 2013; Anetor and Omueti, 2014). This type of solid OMF, in addition to supplying macro and micronutrients, can be used as a soil amendment to improve physical properties due to its large organic C content (Babalola et al., 2007). On the flip side, OMFs rich in organic C require

the application of large quantities to the soil - in the order of tons per hectare- to adequately fulfill crop nutrient requirements (Antille et al., 2013; Mazeika et al., 2016). Such large required application volumes could result impractical.

A second group of solid OMFs has a small proportion of organic C from a material of biological origin. This material acts mainly as a protecting matrix for the mineral nutrients and not as a relevant nutrient source. Low organic C OMFs are designed to produce a slow release of mineral nutrients into the soil solution increasing the nutrient use efficiency (Richards et al., 1993). However, although the use of low organic C OMFs has the advantage of increasing the efficiency of the mineral fertilizer - and therefore reducing its application volume - they have the disadvantage of negligibly contributing with organic C to the soil.

There is a significant tradeoff between OMFs with high and low organic C. In those with low organic C, biowastes are not a significant source of nutrients. Therefore, their potential contribution to biowaste recycling is limited. If the mineral nutrients in the OMF are not from biological origin, it is advisable not to consider them a bio-based fertilizer (ESPP, 2023). Despite this limitation, biowastes in low organic C OMF play a crucial role in enhancing the nutrient use efficiency of mineral fertilizers, which is a significant benefit.

3.4 Application methods and strategies to improve the fertilizer value of bio-based fertilizers

Various studies highlight the importance of fertilizer P placement for enhancing the fertilizer value (Quinn et al., 2020; Freiling et al., 2022). Applying mineral P fertilizers by placement close to the seeds maximizes crop P uptake in the early growth stages (Grant et al., 2001; Grant and Flaten, 2019). Another strategy is subsurface band application (10 cm to 20 cm depth). This is mainly adopted in tropical countries with highly sorbing P soils that are susceptible to long drought periods. Subsurface band application creates P-rich bands with high available P contents, that are less susceptible to the soil surface drying (Meyer et al., 2023). However, the placement of P-rich fertilizers, as would be the case with some BBFs, is not as effective as the placement of mineral P fertilizers. Two possible explanations are the lower P solubility in BBFs compared to mineral fertilizers; and nutrient imbalances, leading to the over application of elements that may be toxic to the plant (Lemming et al., 2016).

Placement of sewage sludge increased root proliferation in the placement zone but did not enhance plant growth and total P uptake (Lemming et al., 2016). It significantly reduced P uptake from the soil compared to a treatment where the fertilizer was completely mixed with the soil. Regarding sewage sludge ash, Lemming et al. (2016) observed that placement did not attract root growth to the placement zone and significantly reduced both P uptake and plant growth. Building up on that, Sica et al. (2023) suggested that biowastes may have sufficient soluble P to attract roots to the placement zone, however, available P may not be sufficient to sustain plant growth for longer periods. Therefore, treatments to increase the P solubility of bio-based fertilizers are needed, along with placement strategies.

Fertilization with organic fertilizers may lead to the overapplication of organic N in the placement zone. Accumulated

TABLE 4 Negative environmental impacts of BBFs, mitigation strategies and relevant literature.

Negative impact	Mitigation strategy	References
Overapplication of nutrients due to non-optimized nutrient ratios	Optimize N to P ratio through solid–liquid separation or source segregation; use organo-mineral combinations	Deeks et al. (2013) and Sharpley (1996)
Presence of heavy metals in sewage sludges and other BBFs	Thorough monitoring and management of biowastes; use of conversion factors to predict heavy metal concentration	de Castro et al. (2023), Geng et al. (2020), Kupper et al. (2014), and Lu et al. (2012)
Presence of microplastics and organic pollutants in soil from sewage sludge	Monitoring; improving wastewater treatment processes; pyrolysis and thermal treatments	Corradini et al. (2019), Crossman et al. (2020), and Kanteraki et al. (2022)
Emission of GHGs from BBF production and soil application	Use energy from renewable sources for BBF production, placement and seasonal fertilization strategies; use locally sourced biowastes; use of nitrification inhibitors	Kar et al. (2023) and Meneses-Quelal and Velázquez-Martí (2020)
Emissions of NH ₃	Storage with reduced exposed surface area; injection of digestates into soil rather than surface applications; use acidifying agents;	Álvarez-Gaitan et al. (2016) and Maris et al. (2021)

organic N is mineralized over time, releasing high amounts of NH₃ into the soil (Chaves et al., 2004). Plants can suffer NH₃ toxicity symptoms, such as inhibiting root growth at concentrations above 100 mg·kg⁻¹ (Nkebiwe et al., 2016). To reduce NH₃ toxicity effects, placement should be done more than 10 cm from the seeds, and/or days before sowing, allowing NH₃ to be nitrified thus lowering its concentration over time, while the plant is starting to develop (Delin et al., 2018; Baral et al., 2021). Acidification may also reduce the negative effects of NH₃ toxicity (Pedersen et al., 2017).

4 Environmental impacts of bio-based fertilizers

4.1 Environmental benefits of bio-based fertilizers production and use

Recycling P reduces phosphate rock mining, thereby avoiding the associated environmental impacts such as landscape degradation, contamination of water bodies, and emission of GHGs associated with transport (Higgins, 2001; Fayiga and Nwoke, 2016). Moreover, using phosphate rock and derivatives for fertilization can result in the contamination of water bodies and soils with heavy metals and hazardous elements like cadmium, uranium and arsenic (Fayiga and Nwoke, 2016). Similarly, reducing the reliance on synthetic N contributes to lowering net GHG emissions and energy consumption, as Haber-Bosch synthesis of NH₃ is one of the major fossil energy consuming processes worldwide (Osorio-Tejada et al., 2022; Gao and Cabrera Serrenho, 2023).

Beyond the substitution of mineral fertilizers, bio-based fertilizers offer additional environmental benefits. For example, by offering a slower release of nutrients compared to mineral fertilizers, nutrient leaching is reduced (Mandal et al., 2007). Bio-based fertilizers containing organic C enhance soil structure by increasing the organic matter content which also may be beneficial for microbial communities (Mayer et al., 2022). In soils in a Danish long-term field experiment, soils treated with compost significantly increased their organic C content compared to the mineral NPK treatment by up to 3%. Moreover, they significantly reduced their bulk density from 1.6 kg·l⁻¹ down to 1.2 kg·l⁻¹, and improved their soil structure, thus requiring less energy consumption for tillage (Peltre et al., 2015).

4.2 Negative environmental impacts associated with bio-based fertilizer production and use, and mitigation strategies

Due to non-optimized nutrient ratios relative to crop demand in many BBFs compared to mineral fertilizers, fertilization with BBF and organic fertilizers could lead to overapplication of nutrients. An overview of environmental impacts associated with BBFs is available in Table 4. For instance, fertilization with manures can lead to overapplication of P when aiming to meet crop N demand (Sharpley, 1996). Similar problems arise when using digestates (Kadam et al., 2022), sewage sludge and other biowastes (Deeks et al., 2013; Lemming et al., 2019). Mitigation strategies encompass minimizing N losses during production, storage, and application of the BBF, thus maximizing the N content and its use efficiency, for which Ndegwa et al. (2008) and Pedersen and Hafner (2023) offer extensive reviews. The N to P ratio of some BBFs such as those derived from animal manures can be optimized by separating N from P through solid–liquid separation, or through segregation at-source of feces and urine before any processing step (Hjorth et al., 2009; Vu et al., 2016). Another strategy is using an organo-mineral combination, that provides the desired N to P ratio and corrects for the nutrient imbalance (Deeks et al., 2013).

The presence of contaminants could be a problem for some BBFs. Sewage sludges can contain cadmium, lead, and other heavy metals that may accumulate in soils after fertilization (Lu et al., 2012). In a study in Chile, significant amounts of microplastics accumulated in the soil after a decade of fertilization with sewage sludge (Corradini et al., 2019). The presence of persistent organic compounds like polychlorinated biphenyls, polycyclic aromatic hydrocarbons, traces of pharmaceutical compounds and hormones in sewage sludges, is discussed by Kanteraki et al. (2022). An overview of pollutant removal techniques for sewage sludge is provided by Geng et al. (2020). Olive pomaces contain phytotoxic phenolic compounds, lipids, and organic acids. Composting pomaces can help mitigate their toxicity by degrading some of the harmful compounds and increasing their fertilizer value (Muscolo et al., 2019; Ameziane et al., 2020).

Heavy metals are of potential concern in digestates and composts depending on the local input materials (Kupper et al., 2014; Kadam et al., 2022). Therefore, there is a pressing need for thorough monitoring and management of these biowastes. A breakthrough for

the simplification of monitoring on a local scale was the obtention of conversion factors to predict the concentration of heavy metals in Belgian manure-derived digestates (de Castro et al., 2023). These conversion factors were based on process parameters of the AD processes, allowing the prediction of the concentrations of aluminum, chromium, copper, iron, manganese and zinc based on the dry matter and the biodegradable fraction content of digestates.

Emissions of GHGs can result from both the production of BBFs and their soil application. In the production stage, mineral precipitates and thermally treated biowastes specifically raise concern as they require energy-intensive treatments. Thermal treatments and separation processes such as ammonia stripping, membrane electrodialysis, ion-exchange and struvite precipitation may emit less C to the atmosphere compared to traditional mineral fertilizer synthesis. However, their operational processes also demand substantial energy consumptions (Meneses-Quelal and Velázquez-Martí, 2020; Kar et al., 2023). Therefore, the source of the utilized energy will influence the environmental impacts of such treatments.

During soil application, NH₃ emissions from manure digestates are often higher than from untreated manure (Holly et al., 2017; Emmerling et al., 2020). Furthermore, using ammonium sulphate to target sulfur fertilization has been linked with increased and overlooked NH₃ emissions in neutral to alkaline soils (Powelson and Dawson, 2022). This not only emits NH₃, contributing to undesired N deposition in natural ecosystems, but also results in losses of recovered N and the energy invested in its recovery. Likewise, fertilization with sewage sludge causes emissions of N₂O and CH₄ (Alvarez-Gaitan et al., 2016).

Implementing mitigation strategies is essential to minimize emissions from BBFs. These strategies include selecting appropriate application timing and methods, managing soils to maximize C retention, and using stabilizing additives like nitrification inhibitors (Severin et al., 2016; Tariq et al., 2022). For instance, injecting digestates into the soil reduces NH₃ emissions compared to surface applications (Hou et al., 2015). Similarly, applying slurry to grasslands during spring rather than autumn is linked to less N₂O emissions (Maris et al., 2021). Soil management practices such as minimizing or eliminating tillage and implementing crop rotations and leys, maximize the amount of C retained in agricultural soils (Jarecki et al., 2003). Acidifying digestates with, e.g., sulfuric acid can significantly reduce NH₃ emissions (Pedersen and Nyord, 2023).

4.3 Assessment of environmental impacts from bio-based fertilizers

Single mitigation strategies can potentially result in pollution swapping. Therefore, understanding the interaction of all processes involved and the impact of specific local conditions is necessary. A comprehensive and integrated approach such as a life cycle assessment (LCA) is optimal to assess the potential environmental impacts of fertilization with BBFs (Jensen et al., 2020; Egas et al., 2023).

For example, Styles et al. (2018) used a LCA to assess BBF production from the liquid fraction of digestate of food waste in Sweden, applying NH₃ stripping and struvite precipitation. The BBF production and field application was compared to the conventional management of digestate's liquid fraction, including storage and field application. They concluded that producing BBF from liquid digestate results in significant

environmental benefits due to the avoidance of CH₄, N₂O and NH₃ emissions compared to the conventional management of liquid digestate. Moreover, application of that BBF enhanced the substitution of synthetic fertilizer due to the targeted use of nutrients. Another LCA on digestate utilization compared four alternative BBF production scenarios in relation to mineral fertilizer production (Alengebawy et al., 2022). This study included two technologies for nutrient extraction from the solid fraction and two more for the liquid fraction. Results showed that in all scenarios, BBFs constituted environmentally beneficial alternatives compared to mineral fertilizers, provided the digestate was pretreated to remove pollutants and pathogens.

Several studies have applied LCA to assess different aspects of fertilization with sewage sludge (Yoshida et al., 2018; Ding et al., 2021). For example, Yoshida et al. (2018) assessed the long-term impacts after field application of sewage sludge by using emission factors calculated by Bruun et al. (2016). Emission factors for sewage sludges were calculated considering sewage sludges with different properties, applied to three soil types, using three precipitation regimes and varying application amounts. This approach enabled the use of region-specific emission factors in the LCA (Yoshida et al., 2018). Normalizing the LCA results to yearly *per capita* emissions showed that human toxicity and ecotoxicity impacts were of greatest concern, largely due to the zinc and copper content in sewage sludge (Yoshida et al., 2018).

Ultimately, regional and local conditions may determine the sustainability and feasibility of using one BBF over another. Walling and Vaneekhaute (2020) reviewed emission factors on organic and inorganic fertilizer production and use. This study recommended that emission factors should be estimated based on case-specific data due to the high variation in emissions depending on the composition of the fertilizer and the impact of local conditions like soil type and climate. In the LCA study of Beyers et al. (2022), the environmental impact of pig slurry acidification was assessed for the climatic, agronomic, and legislative conditions of Denmark, Spain, and the Netherlands. Slurry acidification reduced the environmental impacts related to emissions of GHG and NH₃. However, the acquisition of energy and materials for the acidification process led to increased off-farm impacts in some categories, including fossil resource depletion and human toxicity. Furthermore, the effectiveness of acidification to reduce environmental impacts varied between countries due to differences in legislative requirements and energy sources. Thus, specific regional conditions (soil, climate, legislation, and farming practices) are crucial for the overall environmental sustainability of fertilization with specific BBFs. The PLCI 2.0 model has been developed to account for regional differences in LCAs of P-containing BBFs applied in European regions (Rydgård et al., 2024). This model incorporates factors such as regional soil P concentrations, soil erosion rates and distribution of crop types. Furthermore, it enables the modeling of the impact of different fertilization practices on P losses, harvesting of P in crops and the substitution of mineral P fertilizer.

5 Bio-based fertilizer market developments – social acceptance and economic drivers

Concerns regarding biosafety and environmental impact raised in recent decades, particularly for sewage sludge, have cast a negative

perception on fertilization with some biowastes (Ekane et al., 2021). There is still a significant gap between the EU and other regions in the West in biosolids (treated sewage sludge) adoption. For instance, on average, only 35% of produced sewage sludge is reincorporated into agriculture in the EU (Hušek et al., 2022), compared to 55% in the United States and Canada, and more than 70% in Australia (Marchuk et al., 2023). However, the European average varies significantly by nation; for example, it is 0% in the Netherlands and Slovakia, but as high as 80% in Ireland (Hudcová et al., 2019). A comprehensive understanding of regulations, as well as preferences and needs of end-users and major stakeholders is necessary for a better acceptance and adoption of BBFs (Goldstein and Beecher, 2007; Ekane et al., 2021).

5.1 Stakeholders in the BBF market

In the BBF market, primary stakeholders include livestock farmers with a surplus of manure for export/processing, and crop farmers who are potential end-users of BBFs (Jensen et al., 2017; Kurniawati, et al., 2023a). Other important players are food and pharmaceutical industries, waste management companies, recycling fertilizer companies and farmers with crop residues on the supplier side, as well as garden owners and horticultural producers on the end-user's side (Jensen et al., 2017; Venegas et al., 2021). Governments, public institutions (local/national/EU), civil society, non-governmental organizations (NGO), the food industry, investors, media, and scientists are also stakeholders in the European BBF market (Nedelciu et al., 2019).

The influence of primary stakeholders (farmers) on the supply and demand of BBFs varies. While livestock farmers are incentivized to utilize surplus manure; crop farmers exert more power in accepting or declining recycled products based on their perceptions and preferences (Case et al., 2017). Thus, defining and addressing stakeholder requirements and preferences is crucial for BBF adoption.

5.2 Attributes of bio-based fertilizers that influence acceptance and perception

The acceptance of BBFs among farmers in Europe is mainly influenced by four attributes: known nutrient contents, organic matter contents, cost and ease of application (Egan et al., 2022). Negative perceptions often arise from uncertainty regarding the N, P and K contents in BBFs (Tur-Cardona et al., 2018; Egan et al., 2022), contrasting with the precision offered by mineral fertilizers. Farmers are aware of the uncertainties in nutrient contents in organic fertilizers, and they prefer BBFs with nutrient ratios that fit crop demands (Egan et al., 2022). Therefore, reliable and known amounts and ratios of nutrients are essential for facilitating adoption.

A strength of BBFs is their organic matter content, and their perceived capacity to enhance soil structure, improve soil productivity, and increase water retention capacity of soils (Case et al., 2017; Gwara et al., 2021). These benefits are well-recognized by farmers (Egan et al., 2022). However, these advantages alone are not enough to completely substitute mineral fertilizers with BBFs and need to be accompanied by the other three attributes.

The cost of BBFs and their fertilizing properties relative to mineral fertilizers play a pivotal role in social acceptance. Another advantage of BBFs is their perceived low-cost and high nutrient content (Case

et al., 2017). Farmers are more likely to adopt BBFs if they are competitively priced (Egan et al., 2022). Logistical costs and perceived higher overall expenses may deter adoption, necessitating a cost at least half that of mineral fertilizers for widespread acceptance (Tur-Cardona et al., 2018). Despite potential cost differences, the ecological co-benefits of BBFs mentioned before, could sway farmers toward their adoption with proper awareness (Egan et al., 2022). We discuss such potential in the next section.

The ease of application and the practicality of BBFs also play a major role. The form of the BBF (solid, semi-solid, liquid, or granulated) significantly influences farmer preferences and acceptance (Tur-Cardona et al., 2018; Egan et al., 2022). Solid and semi-solid forms are favored over liquids due to ease of application with existing farm machinery and improved forms, such as pellets, can enhance acceptance and willingness to pay (Hills et al., 2021). Thus, farmers prefer BBFs to be granular like mineral fertilizers.

Farmers may be reluctant to utilize BBFs due to the potential negative perception from clients and consumers of their products (Simha et al., 2017). In a Polish survey, an important source of negative perception stemmed from concerns about potential health risks associated with fertilization using nutrient-rich biowastes (Smol, 2021). Consumer acceptance of products fertilized with BBFs is greater for ornamental plants than for horticultural crops for consumption (Segrè Cohen et al., 2020). This indicates that final consumers of produce fertilized with BBFs may hold negative perceptions. Media and NGOs also influence consumer opinion and their perceptions of BBF products (Jensen et al., 2017). Even if farmers are willing to use recycled nutrients, the reluctance of final consumers and the food industry to consume products fertilized with BBFs may prevent them from doing so (Barquet et al., 2020; McConville et al., 2023). Therefore, it is important to raise awareness among produce consumers and inform them about the environmental advantages of BBFs.

5.3 Current bio-based fertilizer market

Although precise estimations of the European BBF market size are not available, it is evident that the BBF market is smaller compared to the conventional mineral fertilizer market. Approximately half of the EU's fertilizer inputs for P come from mineral P, and about two-thirds of N fertilizers are produced through the Haber-Bosch process (Schoumans et al., 2015; Einarsson et al., 2021). Both studies recognize animal manures as the most significant alternative sources of P and N aside from mineral fertilizers. Therefore, the BBF market is still in early stages, requiring substantial changes in product availability, quality, legislation, competition dynamics, and stakeholder perceptions (Kvakkestad et al., 2023).

A favorable landscape is currently emerging for the development of the BBF market. Legislative developments, as outlined in section 2, have established a legal framework for BBF products in the European market, providing a conducive environment for market growth. However, a crucial question arises: where should the focus be directed? In the previous section, we outlined five criteria important for BBF adoption. Additionally, literature has emphasized that BBFs should resemble mineral fertilizers in consistent supply, allowing for a seamless transition for farmers accustomed to using commercially available mineral fertilizers (Gregson et al., 2015; Case et al., 2017; Buysse and Cardona, 2020; Kvakkestad et al., 2023).

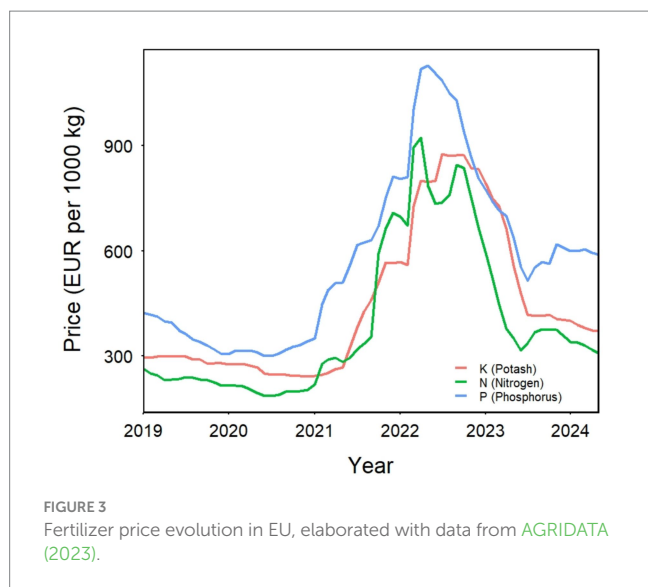


FIGURE 3 Fertilizer price evolution in EU, elaborated with data from AGRIDATA (2023).

Buyse and Cardona (2020) suggested that rising prices of mineral fertilizers would create an opportunity for the BBF industry to develop, given the higher production costs of BBFs. The fertilizer market has experienced significant volatility (Figure 3) during the period from 2021 to 2023, particularly due to rising geopolitical tensions and the war in Ukraine, leading to several increases in mineral fertilizer prices (AGRIDATA, 2023). The price increase in late 2021 and 2022 created a new reality for BBF producers, making more nutrient recovery technologies economically feasible (Hermann and Hermann, 2021). Since early 2023, fertilizer prices have been decreasing and stabilizing. At the time of writing this article, mineral fertilizer prices have returned at the levels of mid-2021. However, despite the prices being lower than in 2022, they remain higher than previous pre-2021 levels. Given the ongoing geopolitical pressures on the European Union, it is unlikely that the prices will decrease further in the foreseeable future (Alexander et al., 2022; Brownlie et al., 2023; Rabbi et al., 2023).

The pricing and competitive position of BBF compared to mineral fertilizers are influenced by many factors: demand, availability of adequate BBFs, logistics, local legislation, and regional fertilization practices (Case et al., 2017; Kvakkestad et al., 2023). Previous studies recommended pricing of BBF significantly lower than mineral fertilizers (Case et al., 2017; Tur-Cardona et al., 2018). However, Moshkin et al. (2023) considered the willingness-to-pay and perceptions of potential users in an EU farmer survey and suggested that BBFs should be priced equivalently to mineral fertilizers. While willingness-to-pay may still be lower for BBFs compared to mineral fertilizers, marketing strategies could help justify higher prices and contribute to the development of the BBF industry. Moreover, branding of BBFs plays an important role in their adoption. For example, using the term “biosolids” instead of “treated sewage sludge” positively impacts consumer attitudes and acceptance (Lu et al., 2012).

6 Discussion

The EU’s initiatives like the Farm to Fork strategy and the new EU Fertilizer Regulation signal a shift toward a more circular nutrient

economy, with BBFs playing a crucial role. However, agronomic and environmental challenges persist for BBFs. The main challenge identified is the uncertainty surrounding the quantity and availability of nutrients, as well as the lack of adequate nutrient ratios in biowastes (Egan et al., 2022).

Many biowastes have deficiencies in their nutrient compositions when used as organic fertilizers, and these deficiencies can be exacerbated by transformation treatments applied to biowastes. However, deficiencies can be corrected through chemical treatments or pre-treatments that enhance the nutritional properties of the transformation outputs (e.g., digestate). Despite these improvements, many BBFs still have limitations compared to the practicality of more concentrated and stable mineral fertilizers. To ensure adequate nutrient quantities and ratios, organo-mineral combinations are a promising alternative (Deeks et al., 2013; Sitzmann et al., 2024), as well as improved processes of nutrient recovery in waste management facilities (Marchuk et al., 2023). To maximize the agricultural efficiency of BBFs, including organo-mineral combinations, it is essential to use them alongside appropriate placement strategies. This approach ensures that the benefits of both organic and mineral components are fully realized, promoting sustainable and effective fertilizations.

The current economic conditions, marked by high prices of mineral fertilizers and changes to the EU Fertilizer Regulation, present an unique opportunity for the growth of the BBF market. However, barriers such as negative perceptions of BBF and logistical issues hinder their adoption by farmers. Improving perception could be achieved by offering BBF products that match mineral fertilizers in both agronomic efficiency and physical attributes. To improve perception and acceptance of BBFs, it is imperative to work with stakeholders and end-users of BBFs and include them in research and development processes to address their logistical and agricultural needs (Nedelciu et al., 2019; Venegas et al., 2021). While agricultural efficiency is crucial, it is also important for BBFs to provide additional benefits to the soil, like addition of organic matter (Case et al., 2017). Practicality is another key element, along with customer and societal acceptance of products fertilized with BBFs. It is one of the most crucial factors to develop to unlock the BBF market. Literature has evidenced a demand for BBFs that offer the same convenience, meaning they are always available, can be applied without major logistical complications, and are designed for the needs of a specific crop, i.e., commercial products.

Based on the literature review, we propose that to enhance BBF acceptance among farmers and growers, it is important to demonstrate that: (1) BBFs have sufficient and reliable nutrient amounts, as well as crop-adequate nutrient ratios; (2) BBF can maintain or improve soil fertility by being a supply of organic matter to soils; (3) BBFs need to be priced competitively with mineral fertilizers, but not significantly cheaper than mineral fertilizers. The close pricing ensures that BBFs remain an attractive alternative while covering costs of production and innovation.; (4) Handling and application of BBF with existing machinery is possible; (5) There is willingness among consumers of the farmer’s products to buy products fertilized with BBFs.

Increasing refinement of bio-based fertilizing products will be associated with higher energy and material consumption, and environmental impacts. For instance, pollution swapping caused by some treatment techniques may result in trade-offs between the agronomic and

environmental benefits of BBFs. To overcome this, integrated measures that consider local conditions are crucial. Multiple LCAs have shown that the sustainability of BBFs is strongly influenced by local factors such as available biowastes, electricity sources, local legislation, soil and climatic conditions, which ultimately determine the agronomic value and environmental impacts (Jensen et al., 2020; Beyers et al., 2022). Thus, local, and small to middle-scale production of BBFs utilizing regional biowastes may be a more optimal solution than centralized, large-scale national facilities. Similarly, transforming waste management facilities like wastewater treatment plants into nutrient recovery centers, is an important aspect of locally sourcing BBFs.

Strategies of fertilization with BBFs should also consider adequate timing and placement to maximize nutrient efficiency and minimize nutrient losses to the environment (Lemming et al., 2016; Maris et al., 2021). Therefore, effective agricultural extension services are needed to assist farmers in transitioning to BBFs and ensure that they are used and placed appropriately to maximize their fertilizer value.

7 Conclusion

Maximizing the potential of BBFs in the European context and overcoming existing barriers requires a comprehensive assessment. We acknowledge that the agricultural use of BBFs at present could be demanding for farmers, given the regulatory framework, the scarcity of new, innovative BBFs, the planning required to ensure proper nutrient supply to crops, and the potential reluctance of consumers to buy products that have been fertilized with BBFs. It is crucial to carefully consider legislative, logistical, economic, soil fertility, and climatic requirements, with a primary focus on the crop's nutrient needs to develop BBFs that can be adopted by farmers. Therefore, substantial research and product development efforts are still required to overcome these barriers and provide viable alternatives. However, the goal is to achieve a circular nutrient economy and not only substitute mineral fertilizers with BBFs. Therefore, it is important for all stakeholders to recognize the value of producing and consuming products fertilized with recycled nutrients, understanding how their purchase contributes to a more circular economy and sustainable development in general.

Author contributions

MÁ: Writing – review & editing, Writing – original draft, Visualization, Project administration, Methodology, Investigation, Formal analysis, Data curation, Conceptualization. PS: Writing – review & editing, Writing – original draft, Visualization, Formal analysis, Data curation, Conceptualization. TS: Writing – review & editing, Writing – original draft, Formal analysis, Data curation, Conceptualization. MR: Writing – review & editing, Writing – original

draft, Conceptualization. JN: Writing – review & editing, Writing – original draft, Methodology, Formal Analysis, Conceptualization. EM: Writing – review & editing, Writing – original draft, Visualization, Conceptualization. JM: Writing – review & editing, Writing – original draft, Validation. CK: Writing – review & editing, Writing – original draft. HC: Writing – review & editing, Writing – original draft. AU-J: Writing – review & editing, Writing – original draft. SC: Writing – review & editing, Writing – original draft. FL: Writing – review & editing, Writing – original draft. KW: Writing – review & editing, Writing – original draft. YZ: Writing – review & editing, Writing – original draft. SH: Writing – review & editing, Writing – original draft. DC: Writing – review & editing, Writing – original draft, Supervision. LJ: Writing – review & editing, Writing – original draft, Supervision, Conceptualization.

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Conflict of interest

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