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## EDITED BY

Muhammad Shaukat,  
Allama Iqbal Open University, Pakistan

## REVIEWED BY

Sadam Hussain,  
Northwest A&F University, China

## \*CORRESPONDENCE

Jenny Kao-Kniffin  
✉ jtk57@cornell.edu

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# Three important roles and chemical properties of glomalin-related soil protein

Yejin Son, Carmen Enid Martínez and Jenny Kao-Kniffin\*

School of Integrative Plant Science, Cornell University, Ithaca, NY, United States

The consequences of climate change urgently demand the reduction of atmospheric carbon, including by sequestering carbon in soil. The glomalin-related soil proteins (GRSP) of arbuscular mycorrhizal fungi (AMF) are renowned for their soil aggregation and carbon sequestration properties. With their considerable binding abilities, GRSP can also adsorb various cations and sequester heavy metals in soil, thereby assisting in soil fertilization and remediation efforts. However, despite its benefits for soil health and climate change, the mechanisms underlying these traits in the context of soil chemistry remain unexplored. In this review, we focus on three crucial roles of GRSP—long-term carbon sequestration, soil aggregation, and soil remediation and fertility—in the context of the chemical characteristics elucidated by previous research, namely hydrophobicity, amid group glycosylation (N-glycosylation), and metal adsorption. Based on the proposed chemical mechanisms, the current review also offers insight into soil factors that may influence the persistence of GRSP. We conclude by proposing a working model for GRSP, aiming to establish a conceptual platform for future research to examine GRSP in terms of their known or novel chemical and biochemical reactions, thereby improving our understanding of this important group of soil proteins.

## KEYWORDS

glomalin-related soil protein, soil aggregation, climate change, arbuscular mycorrhizal fungi, carbon sequestration, soil health, soil remediation, microbial protein

## 1 Introduction

Global warming is a threat to future food production worldwide. Soil carbon (C) sequestration from the atmosphere has received much attention from both the public and researchers as it has the potential to meet the goal of reducing atmospheric C while enhancing soil fertility (1, 2). Soil C sequestration can also improve the storage of soil organic carbon (SOC), soil physical structure and reduce the risk of soil degradation and erosion while supplying nutrients and energy for soil microorganisms (1).

Arbuscular mycorrhizal fungi (AMF) are important fungi that form symbiosis with more than 80% of terrestrial plants and can effectively aid soil C sequestration and storage

in the form of SOC (3). Wright and Upadhyaya (4) first discovered that AMF produced a heat-stable glycoprotein, glomalin, using an enzyme-linked immunosorbent assay (ELISA) and a monoclonal antibody, MAb32B11, raised against crushed AMF spores of *Rhizophagus irregularis*. Using mass spectrometry analysis, Gadkar and Rillig (5) found that glomalin shared similarities in the protein sequences with a 60 kDa heat shock protein (HSP60) of *R. irregularis* (> 80%), thus showing that glomalin was a putative gene product of AMF origin. In fact, Magurno et al. (6) successively utilized glomalin gene sequences (HSP 60 homologue) of 15 different AMF isolates as a gene marker to study a diversity of glomalin genes expressed in soils.

The extraction methods of glomalin extracts include: 1) high-temperature citrate extraction (HTCE) through the processes of HTCE, purification and decanting, and quantification using the Bradford assay with a reference of bovine serum albumin (BSA); and 2) the immunoreactivity test to the antibody MAb32B11 using ELISA, western blots/immunoblots, or indirect immunofluorescence assays (7–10). However, HTCE is a nonspecific protein extraction method which co-extracts other heat-stable proteins of non-AMF origin (e.g. plants, bacteria, and other soil autotrophs) along with organic compounds such as lipids, fatty acids, and humic acid (11–14). These co-extracted compounds can interfere with the Bradford assay (e.g. polyphenols) and ELISA (e.g. organic matter), raising questions about precision and accuracy of the quantification with HTCE and ELISA (8, 9).

Due to the co-extracting nature, glomalin is therefore an operationally defined protein of AMF origin, and the term ‘glomalin-related soil proteins’(GRSP) has been widely used to represent a mixture of proteins of both AMF and non-mycorrhizal origin extracted by HTCE (7, 15). The origin of AMF is still controversial due to the presence of the proteins of non-AMF origin in GRSP, especially those of bacteria, although it has been suggested that bacterial proteins contained in GRSP may be a bacterial transformation of AMF exudates and/or necromass, not their product (15). In contrast, many studies have reported positive relationships between the accumulation of GRSP and AMF hyphal growth (16–21), and a higher production of GRSP in soils with plants inoculated by AMF species (22), supporting the causal relationship between AMF and GRSP. Regardless of their origin, GRSP in general contain diverse elements such as C (35-40%), oxygen (33-49%), nitrogen (2-4%), aluminum (1.3-4.8%), and other elements in addition to diverse functional groups, different types of C compounds, and protein-like substances (3).

Glomalin gained popularity owing to increasing interest in their beneficial functions for long-term C sequestration, soil aggregation, and soil remediation and fertility as summarized in Figure 1 (9, 10). Once glomalin is produced within the hyphal wall and released into the surrounding environment, it can form an insoluble protective glue-like layer covering the surface of soil aggregates (19, 23). GRSP is found to contain high concentrations of alkyl C and aromatic C, which contribute to the recalcitrant structure of SOC (11, 24). GRSP

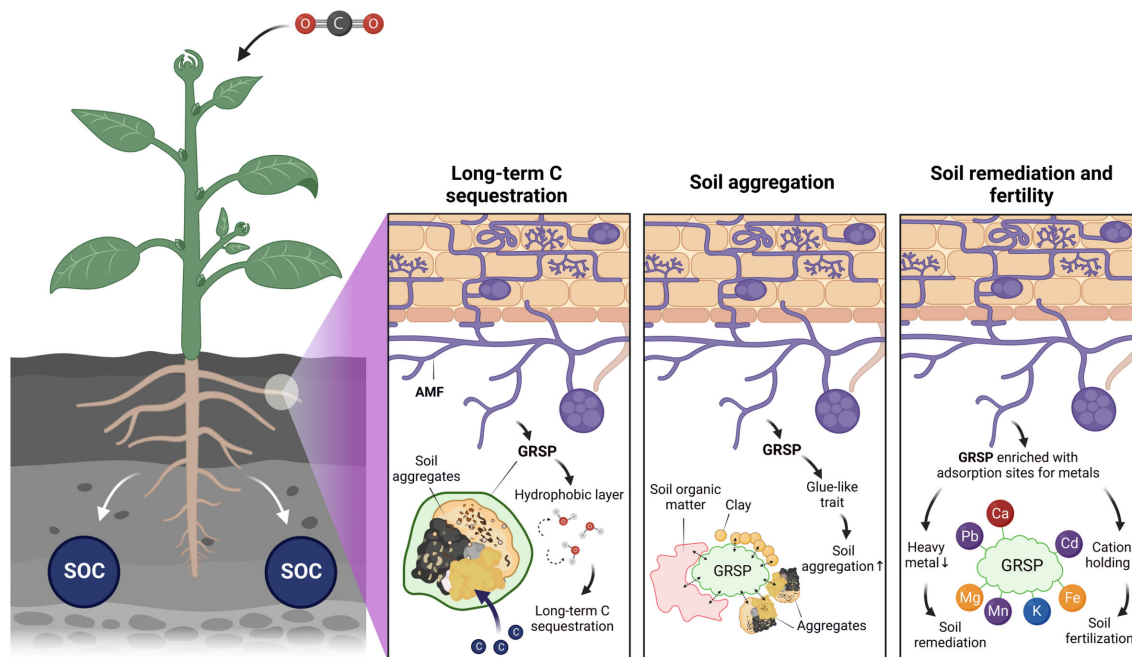


FIGURE 1

The main benefits of GRSP—long-term carbon (C) sequestration, soil aggregation, and soil remediation and fertilization. Atmospheric  $\text{CO}_2$  is fixed by plants via photosynthesis, and then the plant C is incorporated into the soil as soil organic carbon (SOC). Arbuscular mycorrhizal fungi (AMF) inoculate the roots of the plant and grow their extraradical hyphal network into the soil, releasing glomalin related soil protein (GRSP) to the soil. GRSP can promote long-term C sequestration by forming a hydrophobic layer around aggregates. GRSP can increase soil aggregation by binding soil organic matter and clays, thus forming microaggregates and macroaggregates using its glue-like trait. GRSP can bind and sequester metals, thereby decreasing their bioavailability and their toxicity in the soil.

are significant components of soil organic matter (SOM), and their hydrophobicity contributes to the chemical stability of SOM leading to slower turnover and thereby holding great importance for long-term soil C sequestration and soil fertility (11). In addition, GRSP consist of N-linked glycoproteins where a variety of carbohydrate residues are covalently linked to the GRSP protein backbone offering reaction sites between GRSP and other soil constituents such as SOM, clays, other soil particles, and nutrients (11). The glycosylated nature of GRSP is assumed to be a key factor for soil micro- and/or macro-aggregation, which nonetheless has not been tested or can be the result of high-temperature citrate extraction (9). Increased aggregation facilitated by GRSP can improve soil hydraulic properties such as water penetration, saturated hydraulic conductivity, infiltrability, porosity, and soil moisture potential as well as water-holding capacity (10, 25). Thus, GRSP can not only contribute to reducing a risk of C loss by soil erosion and water-based degradation such as rain and surface runoff (9, 10, 26) but they also increase drought tolerance in plants (27). In addition, reaction sites present in GRSP can adsorb cations (e.g., heavy metals and nutrients) via its functional groups that provide a negatively charged surface (28). This function allows the GRSP to decrease the bioavailability and toxicity of heavy metals, alongside holding soil cations and increasing soil fertility.

Although the importance of GRSP as a key biological player in soil C sequestration and other ecological benefits has been widely recognized, their chemical nature remains unclear due to current methodological challenges and extraction procedures (5, 15). Most chemical studies on GRSP have focused on revealing the individual chemical characteristics of GRSP such as chemical composition (11, 24); soil aggregation mechanism and adsorption behavior (29); heavy metal chelation (30); and organic pollutant removal (31), but less attempts have been made to summarize and interpret these characteristics in the context of soil chemistry. Hence, there is a growing need to collectively review the chemistry and molecular biology of GRSP to understand how its chemical traits relate to the known benefits of long-term C sequestration, soil aggregation, and soil health via enhancing soil remediation and fertility.

## 2 The hydrophobicity of GRSP

### 2.1 Hydrophobicity of GRSP explained by disulfide bonds

The GRSP is a homolog of heat shock protein 60 (HSP60) in terms of its secondary and tertiary structures (5). Using tandem liquid chromatography-mass spectrometry, Gadkar and Rillig (5) identified that the partial amino acid sequence of GRSP were 80% identical to the full-length open-reading frame encoding the 590 amino acids of the HSP60 gene in the cDNA library of *R. irregularis*. Brocchieri and Karlin (32) created the consensus sequences using diverse HSP60 sequences of 43 prokaryotes, called GroEL. They reported that the conserved regions of HSP60 contained ATP/ADP and magnesium binding sites, hydrophobic residues, and charged residues in the center of HSP60, facilitating substrate interaction and hydrophobic core formation. Later, other studies including both prokaryotic and

eukaryotic HSP60 sequences revealed that unlike GroEL, eukaryotic HSP60 contained cysteine residues in the conserved region which could contribute to the stability of tertiary structure by forming disulfide bonds (33, 34). Disulfide bridges are known to increase thermodynamic and mechanical stability in a protein, energetically favoring protein folding and increasing the packing of a local cluster of hydrophobic residues and a hydrophobic core (35, 36). Many filamentous fungi such as Ascomycetes and Basidiomycetes produce fungal hydrophobins consisting of eight conserved cysteine residues that form disulfide bridges and self-assemble into various kinds of aggregate as an amphiphile (37, 38). Rillig (23) proposed that GRSP was a homolog of fungal hydrophobins based on its ability to form a hydrophobic coating on soil aggregates with slow turnover and chemical recalcitrance. Compared to the hydroxyl (-OH) group, the sulfhydryl (-SH) group of cysteine prefers to form weak hydrogen bonds with water while establishing covalent bonds between two cysteine residues, and thus creating a hydrophobic layer over the surface of GRSP (Figure 2) (39). Disulfide bonds can limit the exposure of solvent-accessible surfaces such as H-bond donors and acceptors in the unfolded protein due to the hydrophobic effect and a higher entropy of the solvent surrounding the disulfide-containing protein, imposing structural elements that stabilize protein conformation (35). By forming longer covalent bonds of disulfides, the disulfide bonds can increase the intra- and inter-molecular cross-links and thermostability of a protein (40), which may explain the heat-resistance of GRSP.

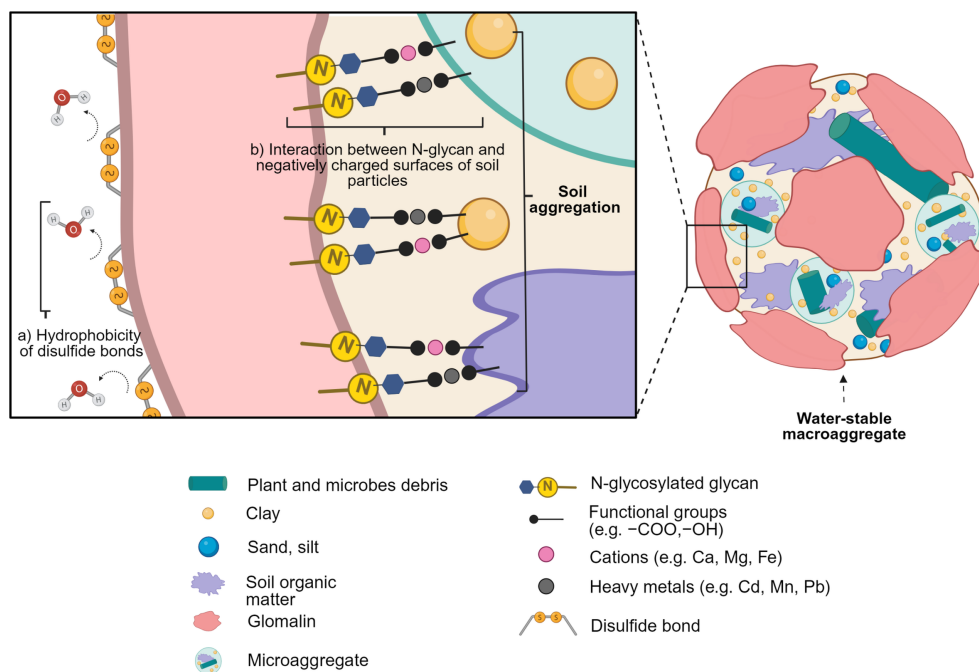
### 2.2 The hydrophobicity of GRSP in relation with soil metal composition

The functional disulfide bonds include catalytic and allosteric sites which regulate the functions of a protein, providing a ligand-exchange site, for example, polarizable and anionic thiols (-R-S<sup>-</sup>) (41). The negative charge on the ligands can be stabilized by a high amount of metal cations in GRSP; when they are coordinated with chalcophile elements such as Fe<sup>2+</sup>, Zn<sup>2+</sup>, and Cu<sup>2+</sup> (42) and form Fe-S clusters (43). These metals can react with the beta thiol group of cysteine residues, thereby producing a thiyl radical (-S•) (44). Sulfur radicals initiate a free-radical polymerization and a ring-opening polymerization of disulfide moieties both of which convert monomeric GRSP units into a multimeric complex, enhancing the hydrophobicity of GRSP units (43, 45). In addition, GRSP contains other multivalent cations (e.g., Al<sup>3+</sup>, Mg<sup>2+</sup>, and Ca<sup>2+</sup>) which can bind to non-S containing functional groups and induce intramolecular rearrangement by neutralizing the negative charge on the surface of GRSP (46). These processes create a stable GRSP structure, allowing it to increase in size and hydrophobicity.

## 3 Soil aggregating ability of GRSP

### 3.1 N-glycosylation of GRSP facilitates soil aggregation

The GRSP is known to promote soil aggregation which is important to soil C sequestration (7). The N-terminal carbohydrates



**FIGURE 2**  
 The outline of the formation of a water-stable macroaggregate facilitated by GRSP. Mechanisms a and b are as follows: **(A)** The disulfide bonds within glomalin can help form a sticky, insoluble, hydrophobic layer around aggregates; **(B)** the N-glycosylation within glomalin not only expands the reaction sites for cation and heavy metal adsorption but also helps bind soil organic matter, microaggregates, and soil particles, producing a water-stable aggregate.

provide glue-like traits that are important for the sorption of proteins to mineral and soil particles (47). Understanding the roles of N-terminal interactions of GRSP with soil particles can unravel the mechanisms underlying soil aggregation. Gadkar and Rillig (5) discovered the presence of N-linked oligosaccharides at its N-terminus, showing its glycoproteinaceous nature. Schindler et al. (11) reported that GRSP was composed of 30–40% protein, with 4–11% aliphatic, 42–49% aromatic, 24–30% carboxyl, and 4–16% carbohydrate-type C content. Due to its glycosylation sites, GRSP can be stabilized by covalent bonding with soil carbohydrates at the side-chain amino groups of lysine and arginine at the N-terminus (48). Mothay and Ramesh (49) visualized the 3D structure of glomalin of *R. irregularis* based on HSP60 amino acid sequences to understand the molecular interactions and kinetic energy of glomalin with the ligands of SOM. They reported that binding of glomalin to SOM involved hydrogen bonds, electrostatic interactions, and Van der Waals forces. However, hydrophilic interactions were predominant for ligand-protein docking while hydrophobic interactions were more involved in stabilizing protein structure and activity. In fact, a wide range of glycans such as sugars, monosaccharides, oligosaccharides, or polysaccharides can attach to multiple side chains of a protein backbone, determining a diversity of GRSP (3, 50). Furthermore, N-linked sugar moieties can be diversified by phosphorylation, sulfation, methylation, or fatty acylation which can generate more binding sites for a variety of soil particles (51). According to Smith et al. (52), prions with 10 basic amino acid residues at their N-terminus interacted strongly with the negatively charged surface of clays such as montmorillonite due to the positive charges developed on the N-terminal ends. The adsorption of proteins can be greater when

oligosaccharides are added to the N-terminus because they expand their reaction sites for adsorption to the exterior of aggregates.

The adsorption process can vary depending on the constituents of the polysaccharides and their different sizes and charges. Polar or charged polysaccharides, such as polysaccharide-N moieties mainly present in proteins, can adsorb to the surface of montmorillonite and kaolinite via electrostatic interaction and H-bonding between carbonyl group in the protein and the Al-OH of montmorillonite and kaolinite edge-sites (53). In contrast, uncharged sugar moieties can increase sorption through hydrophobic interaction by increasing conformational entropy (52). Likewise, N-linked glycans that attach to GRSP can act as extracellular polymeric substances (EPS), forming a sticky and insoluble biofilm that glues surrounding SOM, minerals (i.e., oxides, layer silicates), and microbes (Figure 2). Numerous studies have reported a significant contribution of GRSP to the formation of water-stable macroaggregates, leading to improved soil quality and health by improving the resistance to soil erosion, soil C storage, and nutrient availability (4, 10, 20, 54, 55).

### 3.2 The adsorption of cations to GRSP can facilitate soil aggregation but depends on soil pH

N-glycan attachment in GRSP provides multiple binding sites for soil cations, such as K, Ca, Mg, Fe, Al, Mn, Zn, Cu, Fe, and Na, to form metal complexes with GRSP (31, 56). Soil microorganisms can exude biopolymers composed of extracellular proteins and polysaccharides, which might attach to GRSP via N-glycosylation

(57). The polysaccharides provide negatively charged active sites, such as carboxyl and hydroxyl groups, that can form complexes with cations (58). The cations within the cation-GRSP complex can also help GRSP make bridges with clays and SOM by increasing the electrostatic flocculation process. Wang et al. (56) reported that the absolute value of the zeta potential declined when GRSP was added to a kaolin suspension, which indicates that the negative charges on the kaolin were balanced and neutralized by the adsorption of the cation-GRSP complex. Hence, this adsorption offers strong binding effects that cause soil particles to stick together and increases soil aggregation, contributing to SOC sequestration (59).

However, adsorption of GRSP onto soil minerals can be dependent on soil pH. According to Wang et al. (60), glycosylated proteins are less stable compared to unglycosylated proteins at a low pH of 4 because sugar chains interact with one another and weakly interact with the protein structure through weak interactions such as H-bonding or van der Waals forces rather than strong ion-bonding. Also, the negative charges of polysaccharides can produce repulsive electrostatic forces at higher pH, because the clay surface is mostly negatively charged and produces a larger amount of endothermic heat during the adsorption at a high pH (53). As the point of zero charge (PZC) of soils varies depending on the mineral constituents, it is a challenge to describe consistent behavior in the adsorption of GRSP. Layer silicate clays have mostly negatively charged surfaces, due to isomorphous substitution at most soil pH values (53). The researchers observed decreasing adsorption with increasing pH because of the progressive deprotonation of functional groups and negative charge development on EPS molecules causing repulsion between EPS and clay surfaces. However, the decreasing patterns of adsorption differed from the type of clay; for montmorillonite and kaolinite, the adsorption of EPS-derived C, N, and P greatly decreased when the pH was higher than their PZC values (3.0 and 3.3, respectively), while with goethite, the adsorption was still high when pH was below 8.3, owing to its high PZC ( $\approx 8.3$ ) and the positive charge on EPS at relatively high pH (53). If N-linked glycans of GRSP behave similarly as EPS, the adsorption behavior of GRSP can vary depending on the type of mineral with which they interact. In fact, Tian et al. (61) reported that soil pH was the greatest variable, among other soil parameters such as moisture, depth and salinity, in determining the deposition of GRSP in mangrove wetland soils. Similarly, Wang et al. (29) discovered that in deep and shallow soils, GRSP strongly influenced SOC accumulation, an effect that was largely determined by the soil pH in deep soil. The chemistry of soil aggregation facilitated by GRSP can, therefore, be reliant on soil pH, mineral type, and the composition of related polysaccharides at the N-terminus.

## 4 Improving soil health: Soil remediation and fertilization by GRSP

### 4.1 The remediation ability of GRSP by sequestering heavy metals and organic pollutants in the soil

AMF are known for their ability to accumulate heavy metals within the mycorrhizal structures (62). The GRSP have a high

affinity for toxic heavy metals such as Pb and Cd, while the immobilization of Al was also recently discovered (63). The GRSP can adsorb toxic metals on their functional groups, including the carboxyl, hydroxyl, and thiol groups that provide a negatively charged surface for heavy metals to bind to, which alleviates heavy metal toxicity (28). In fact, Cornejo et al. (30) discovered that GRSP reduced the bioavailability of total Cu and Zn by 26% and 5.8%, respectively. Similarly, González-Chávez et al. (64) extracted 1.6–4.3 mg Cu, 0.02–0.08 mg Cd, and 0.62–1.12 mg Pb per g glomalin, which were the immobilized forms of the heavy metals, whereby the maximum immobilization rate for Cu was 35%. Organic pollutants were also found adsorbed to GRSP. For example, Chen et al. (31) reported the enhanced sorption of a carcinogenic organic pollutant, phenanthrene, in cation-modified montmorillonites ( $\text{Ca}^{2+}$  and  $\text{Fe}^{3+}$ ) when treated with GRSP. They discovered that the sorption process of phenanthrene started with the hydroxyl and carboxyl functional groups of GRSP, which then formed stable complexes with metal cations in the montmorillonites via cation- $\pi$  interaction. The phenyl rings of phenanthrene then bound with aromatic hydrocarbons and hydroxyl-benzenes in GRSP via  $\pi$ - $\pi$  electron donor-accepter interaction. Likewise, GRSP, enriched with multiple functional groups, are involved in the retention of diverse heavy metals and organic molecules, thus holding significance for the bioremediation of contaminated soils.

### 4.2 Soil fertilization and plant growth promotion by GRSP

The GRSP contains numerous cations, such as K, Ca, Mg, Fe, and Al, within their glycoprotein-derived extensive networks of organic polymers (3). As GRSP possess numerous complex ligands, they can retain cations present in the soil solution (65). The cation binding sites are mostly oxidized functional groups such as carboxyl, phenolic, and hydroxyl, all of which are abundant in GRSP (31, 66). The GRSP-bound cations could be used as a nutrient source for soil microbes and plants. In fact, GRSP were found to increase soil microbial activities such as the production of extracellular polysaccharides, glucosidase, urease, catalase, peroxidase, and phosphatase by improving soil quality and providing a food substrate for the microbes all of which benefit plant growth and development (10, 28). Recently, many studies have examined the application of exogenous GRSP on plant growth and nutritional quality (67–70). Xu et al. (70) discovered that peanuts treated with GRSP fertilized with K, Fe, Mg, Zn, and Cu showed improved stomatal opening, photosynthesis, release of photosynthates to the rhizosphere, dry biomass, and pod length and yield as well as higher activity of antioxidative enzymes under drought conditions compared with GRSP alone. This suggests that GRSP-bound nutrients can play important roles in determining plant growth and biology. Liu et al. (67) also found that exogenous GRSP application enhanced the growth of trifoliolate orange seedlings, the uptake of N, P, and K, and auxin expression in the roots. Similarly, fruit nutritional quality improves after a foliar application of cation-enriched GRSP for citrus, accompanied by

better soil nutrient availability for nitrate, ammonium, P, and K (68). Meng et al. (69) discovered that sweet oranges accumulated more N, P, K, Fe, and Si in their sarcocarp after the application of exogenous GRSP to their foliar parts. The findings may explain the high capacity of GRSP to adsorb nutrients onto its functional groups, which can be subsequently used as a nutrient source for both soil microbes and plants.

## 5 Conclusion

There are still mysteries surrounding GRSP, including their origin, chemical structure and composition, full genome and amino acid sequences, and phylogenetic history. Nonetheless, the importance of studying GRSP is growing due to the growing threat of climate change alongside soil degradation and pollution. Increasing soil GRSP is crucial for long-term soil C storage, agricultural soil aggregation, and the remediation of polluted lands, all of which promote the environmental sustainability of soils. The GRSP can contribute to forming a water-repellent macroaggregate, by proposing a working model where glomalin-related functional groups interact with N-glycans, cations, heavy metals, and SOM. Although the full structure of glomalin is unknown, this soil protein conglomerate warrants significant attention to facilitate C sequestration in the soil as an important part of mitigating climate change.

## Author contributions

YS: Conceptualization, Formal analysis, Investigation, Visualization, Writing – original draft, Writing – review &

editing. CM: Supervision, Validation, Writing – original draft, Writing – review & editing. JK-K: Conceptualization, Supervision, Validation, Writing – original draft, Writing – review & editing.

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## Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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