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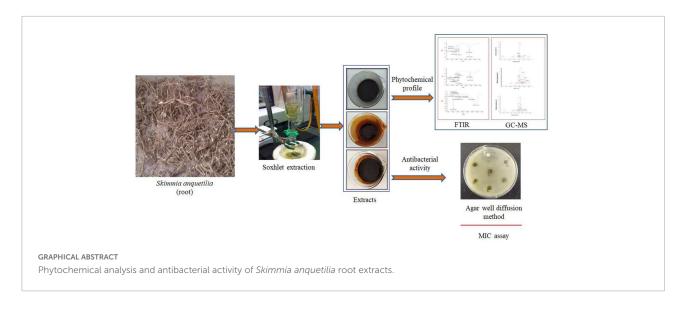
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# Phytochemical screening and antibacterial activity of *Skimmia anquetilia* N.P. Taylor and Airy Shaw: A first study from Kashmir Himalaya

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The present study aimed to explore the antibacterial activity of various organic root extracts of Skimmia anguetilia N.P. Taylor and Airy Shaw and the identification of major functional groups and phytoconstituents through fourier transform infrared spectrometer (FTIR) and gas chromatography-mass spectrometer (GC-MS). The extracts were evaluated for antibacterial activity against multidrug-resistant (MDR) strains viz., Pseudomonas aeruginosa (MTCC424), Escherichia coli (MTCC739), Klebsiella pneumoniae (MTCC139), Salmonella typhi (MTCC3224), and Staphylococcus aureus (MTCC96). ESKAPE pathogens such as S. aureus, K. pneumoniae, and P. aeruginosa are responsible for a majority of all healthcare acquired infections. The ethyl acetate extract showed the highest zone of inhibition against P. aeruginosa (18 mm) followed by S. aureus (17 mm). The minimum inhibitory concentration (MIC) of ethyl acetate extract against strain of S. aureus (4 mg mL $^{-1}$ ) demonstrated therapeutically significant antibacterial activity. The FTIR spectra of root extracts revealed the occurrence of functional characteristic peaks of alcohols, carboxylic acids, aromatic compounds, alkanes, alkenes, and amines that indicates the presence of various metabolites in the extracts. The GC-MS investigation led to the identification of diverse phytoconstituents in each of the extracts with varying concentrations and molecular masses. The highest number of compounds were identified from the methanol extract (112), followed by n-hexane extract (88) and ethyl acetate extract (74). The most predominant compounds were 5, 10-pentadecadien-1-ol, (Z,Z)-(33.94%), n-hexadecanoic acid (13.41%) in n-hexane extract, 5,10-pentadecadien-1-ol, (Z,Z)-(10.48%), 1-hexyl-2-nitrocyclohexane (7.94%)



in ethyl acetate extract, and 1-hexyl-2-nitrocyclohexane (15.43%), *cis,cis,cis*-7,10,13-hexadecatrienal (13.29%) in methanol extract. The results of the present study will create a way for the invention of plant-based medicines for various life-threatening microbial infections using *S. anquetilia*, which may lead to the development of novel drugs against drug-resistant microbial infections.

KEYWORDS

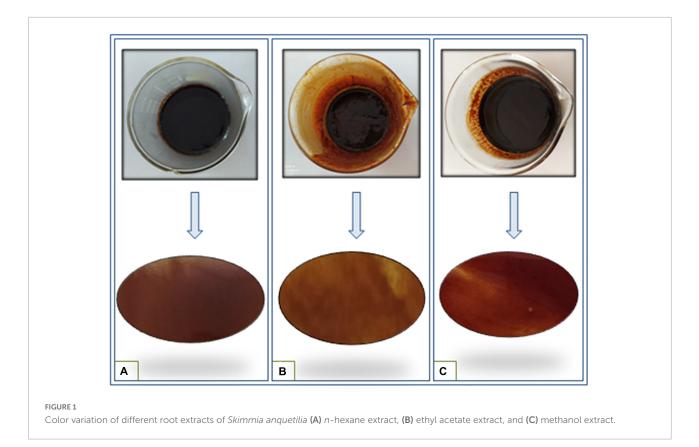
antibacterial activity, FTIR, GC-MS, Kashmir Himalaya, multiresistant, plant extracts, *Skimmia anquetilia* 

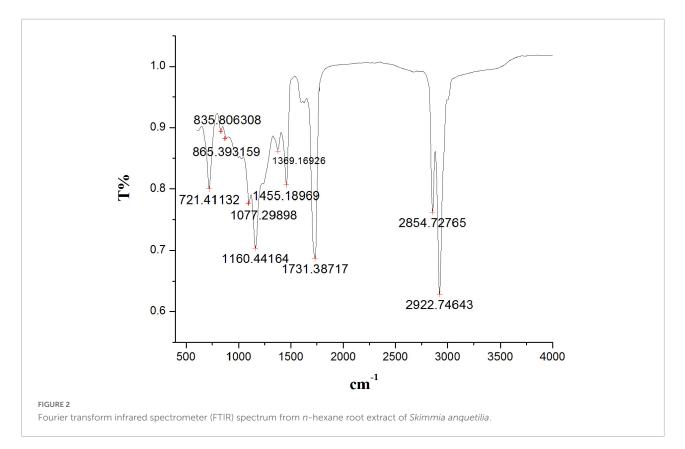
# Introduction

Multidrug-resistant strains of pathogens including Enterococcus faecium, Staphylococcus aureus, Klebsiella pneumoniae, Acinetobacter baumannii, Pseudomonas aeruginosa, and Enterobacteriaceae (commonly known as ESKAPE pathogens) cause many life-threatening infections. These infections are not treatable with currently available antibiotics if left unaddressed, this will surpass cancer as a cause of death by 2050 (O'Neil, 2014). Even in developed countries like the United States, 23,000 people die each year due to drug-resistant microbial infections. The scenario is similar in Europe and much worse in the developing countries of Asia including India, Latin America, and Africa (Reardon, 2014). Unfortunately, the pipeline for developing novel antibiotics has drained, and clinical approval of new antibiotics is declining. Therefore, it is a need of an hour to discover new lead molecules with a novel mechanism of action and can combat antimicrobial resistance. Further, the use of plant sources and natural products can be used for the rational development of new lead molecules with better efficacy against ESKAPE pathogens. Medicinal plants and natural products have been considered as a great source of medicines to benefit mankind from time immemorial (Benarba and Pandiella, 2020; Javed et al., 2021).

They perform an important role in the prevention of disease and treatment of various ailments across the globe (Banaras et al., 2021). They are a source of many active principles that are both biologically as well as pharmaceutically significant including alkaloids, flavonoids, glycosides, lignans, monoterpenes, lipids (phyto-sterols, toco-pherols, saturated and un-saturated fatty acids), and vitamins (Mazurek et al., 2017; Javaid et al., 2021). Therefore, medicinal plants are a vital source of many drugs, almost a quarter of prescribed medicines (Pan et al., 2013). As

Abbreviations: MDR, multidrug-resistant; FTIR, fourier transform infrared spectrometer; GC-MS, gas chromatography-mass spectrometer; MIC, minimum inhibitory concentration; WHO, world health organization; NIST, national institute of standards and technology; MTCC, microbial type culture collection; MHB, mueller hinton broth; DMSO, dimethyl sulfoxide; ANOVA, analysis of variance.





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per World Health Organization (WHO), nearly threequarters (80%) of people of developing nations depend upon conventional homeopathic treatments for their basic therapeutic requirements (Ward, 2008). Due to the contribution of various useful phyto-compounds found in various plant components, most medicinal plants are unique in their ability to treat and cure different human diseases (Naqvi et al., 2020). Various medicinal plants (~80,000 species) have been used as conventional remedies in various indigenous medicine systems in India since ancient times for the treatment of different ailments (Konappa et al., 2020). Currently, almost 25% of active principles have been detected from medicinal plants which are being employed as prescription medicinal products (Konappa et al., 2020; Süntar, 2020). Some studies suggest that around 25,000 of the original plant-specific preparations are available in indigenous folk and conventional medicine systems that are recommended by approximately 15 lakh folk healers for preventive, convincing, and curative purposes (Sen and Chakraborty, 2015). Essential oils and crude extracts of medicinal plants possess numerous kinds of bioactive compounds that have revealed an array of bioefficacies, namely antibacterial (Zeb et al., 2016; Umaru et al., 2019; Nabi et al., 2022a), antifungal (Banaras et al., 2020), antianalgesic (Lisa et al., 2020), antioxidant

TABLE 1 Fourier transform infrared spectrometer (FTIR) peaks and their assigned functional groups of *n*-hexane root extract of *Skimmia anquetilia*.

S. No.	Wavenumber (cm <sup>-1</sup> )	Compound class	Functional group
1.	2922.74	Alcohol	O-H stretching
		Alkane	C-H stretching
		Amine salt	N-H stretching
		Carboxylic acid	O-H stretching
2.	2854.72	Alcohol	O-H stretching
		Alkane	C-H stretching
		Amine salt	N-H stretching
		Carboxylic acid	O-H stretching
3.	1731.38	Aromatic compound	C-H bending
		Aldehyde	C=O stretching
4.	1455.18	Aromatic compound	C=C stretching
5.	1369.16	Alcohol	O-H bending
		Phenol	O-H bending
6.	1160.44	Amine	C-N stretching
		Tertiary alcohol	C-O stretching
7.	1077.29	Amine	C-N stretching
		Primary alcohol	C-O stretching
8.	865.39	Alkene	C-H bending
9.	835.80	Alkene	C=C bending
10.	721.41	Alkene	C=C bending

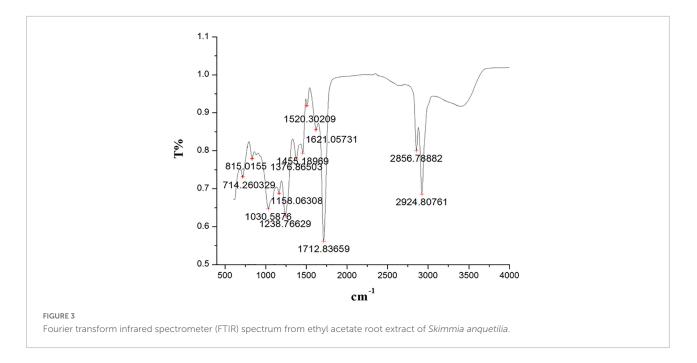
(Gondwal et al., 2012a; Umaru et al., 2019), anticancerous (Liu Y. T. et al., 2020; Oh et al., 2020), antidiabetic (Tran et al., 2020), etc.

S. anquetilia (Rutaceae) is an erect, perennial, glabrous, scented, creeping gregarious, ornamental shrub, 1.5 m in height, found in association with conifers between 1,800 and 2,715 m above msl in Western Himalaya. Traditionally the leaf of the plant has been used in the treatment of headache, smallpox, fever, and also as an anti-inflammatory and antidiabetic agent, etc. S. anquetilia is used to treat paralysis, pneumonia, lung cancer, as an insect and pest repellent and as alexipharmic against snake and scorpion poisons (Nabi et al., 2022b). The powdered bark of the plant is used to cure wounds and burn injuries. Apart from its use in conventional medicine, various extracts and bioconstituents of S. anquetilia have been broadly used in several treatments such as antioxidants (Prakash et al., 2011; Gondwal et al., 2012a; John et al., 2014), antifeedant (Gondwal et al., 2012b), anti-inflammatory (Kumar et al., 2012), and anticancerous activity (Wani et al., 2016). Although undocumented, the plant is being used to treat diabetes in some areas of Kashmir valley. In recent years, various modern techniques such as FTIR, GC-MS, high-performance liquid chromatography (HPLC), etc., have been widely used to detect functional groups and identify various biologically active curative constituents existing in medicinal plants (Koparde et al., 2019). GC-MS displays molecules extracted at different retention rates with spectral data correlating to secondary metabolites, suggesting fatty acid constituents, whereas, FTIR spectrum indicates absorption peaks with a specific wavelength associated with various functional groups (Sim et al., 2014). To date, no such study has been carried out on the identification of bioactive constituents and antibacterial potential of various root extracts of S. anquetilia. Hence, the present study aimed to perform the phytochemical screening for the identification of various functional groups and bioactive constituents through the FTIR and GC-MS techniques and to assess antibacterial activity of S. anquetilia root extracts against both the gram-positive and gram-negative bacterial strains.

## Materials and methods

### Chemicals and reagents

All chemicals and reagents used were of analytical grade. *n*-hexane, ethyl acetate, methanol, nutrient agar, mueller hinton broth (MHB), dimethyl sulfoxide (DMSO), and gentamycin were purchased from Merck (Mumbai, India) and Sigma-Aldrich (St. Louis, MO, United States).



# Collection and identification of plant material

*S. anquetilia* used for the investigation was obtained from the Gulmarg area of Baramulla District, Kashmir, India. The plant specimen was authenticated by a leading Taxonomist Dr. Akhter Hussain Malik, Professor, Centre for Biodiversity and Taxonomy (CBT), University of Kashmir. The voucher number is 2697-(KASH).

### Sample extraction

Fresh plant material of *S. anquetilia* was rinsed with running water, dried under shade, and powdered in an electric blender. Dried root powder (50 g) was successively extracted using solvents (each 500 mL) with escalating polarity, namely *n*-hexane, ethyl acetate, and methanol in a Soxhlet extractor. Repetitive extraction of the plant material was carried out before the attainment of colorless solvent. The acquired extracts were then evaporated to dryness using a rotary evaporator and stored at  $4^{\circ}$ C in airtight glass containers for further analysis.

## Determination of plant extract yield (%)

Yield percentage (w/w) of the dried extracts was calculated as:

Yield (%) =  $W1 \times 100/W2$ 

where W1 is the dry weight of extract after solvent evaporation and W2 is the weight of the dried root powder.

# Fourier transform infrared spectrometer

FTIR spectrometer (Alpha FTIR spectrometer from Bruker optic), fitted with deuterized triglycine sulfate (DTGS) and germanium as a detector and beam splitter, configured to a Windows-based device and coupled to OPUS operating system software (Version 7.0 Bruker optic), was employed throughout the attainment of FTIR spectra. Each sample was placed in direct contact with the attenuated total reflectance (ATR) plate. In spectral regions of 4,000–400 cm<sup>-1</sup>, the FTIR spectra were obtained to determine potential functional groups. The ATR plate was gently wiped with 70% ethanol twice, preceded by drying using soft tissue until filling with the succeeding sample, allowing the ATR plate to dry (Wulandari et al., 2016).

# Gas chromatography-mass spectrometer analysis

GC-MS investigation of *S. anquetilia* root extracts was conducted *via* the Thermo scientific "Chromeleon" (c) Dionex Version: 7.2.8.10783 (Agilent technologies) instrument. GC-MS investigation was performed by employing the following conditions: high electron ionization energy (70 eV) was used. Helium gas (99.99%) was used as the carrier gas with a 1 mL

ABLE 2 Fourier transform infrared spectrometer (FTIR)	) peaks and
eir assigned functional groups of ethyl acetate root ex	tract of
kimmia anquetilia.	

S. No.	Wavenumber (cm <sup>-1</sup> )	Compound class	Functional group
1.	2929.80	Alcohol	O-H stretching
		Carboxylic acid	O-H stretching
		Alkane	C-H stretching
		Amine salt	N-H stretching
2.	2856.78	Alcohol	O-H stretching
		Carboxylic acid	O-H stretching
		Alkane	C-H stretching
		Amine salt	N-H stretching
3.	1712.83	Aromatic compound	C-H bending
		Aliphatic ketone	C=O stretching
		Carboxylic acid	C=O stretching
4.	1621.05	Conjugated alkene	C=C stretching
		Amine	N-H bending
		Cyclic alkene	C=C stretching
5.	1520.30	Aromatics	C=C stretching
6.	1455.18	Aromatics	C=C stretching
7.	1376.36	Alcohol	O-H bending
		Phenol	O-H bending
8.	1238.76	Alkyl aryl ether	C-O stretching
		Amine	C-N stretching
9.	1158.06	Amine	C-N stretching
		Tertiary alcohol	C-O stretching
10.	1030.58	Amine	C-N stretching
11.	815.01	Alkene	C=C bending
12.	714.26	Alkene	C=C bending

min<sup>-1</sup> flow rate. Initially furnace temperature was maintained at 50 °C and then increased to 150 °C with a 3 °C min<sup>-1</sup> increasing rate and retention time of approximately 10 min. The temperature was eventually raised at 10 °C min<sup>-1</sup> to 300 °C. Then, 1 mL of the sample was kept in a 2 mL screw-top vial in an autoinjector, and 1  $\mu$ L of the sample was injected in split-mode (1:40). The overall run time of the GC was 33 min. The phytocompounds present in the extracts have been identified based on comparison of their mass spectral patterns with those spectral database of compounds stored in the National Institute of Standards and Technology (NIST) electronic library coupled with the GC-MS system and the data collected has been tabled.

### Antibacterial activity

#### Bacterial strains, media, and controls

Five bacterial strains viz., P. aeruginosa (MTCC424), E. coli (MTCC739), K. pneumoniae (MTCC139), S. typhi (MTCC3224), and S. aureus (MTCC96) were procured from the Microbial Type Culture Collection (MTCC), Chandigarh (India). The strains included both gram-negative as well as gram-positive strains; for agar well diffusion assay, all strains were initially sub-cultured in nutrient agar media and incubated at 37 °C for 18  $\pm$  2 h. The MIC for all strains was determined by the broth dilution method for which they were grown at 37 °C for 18  $\pm$  2 h in MHB. For antibacterial assay, gentamycin (10 µg mL<sup>-1</sup>) and DMSO were used as positive and negative controls whereas for MIC, plant extract and inoculated broth were used as positive and negative controls.

### Antibacterial screening

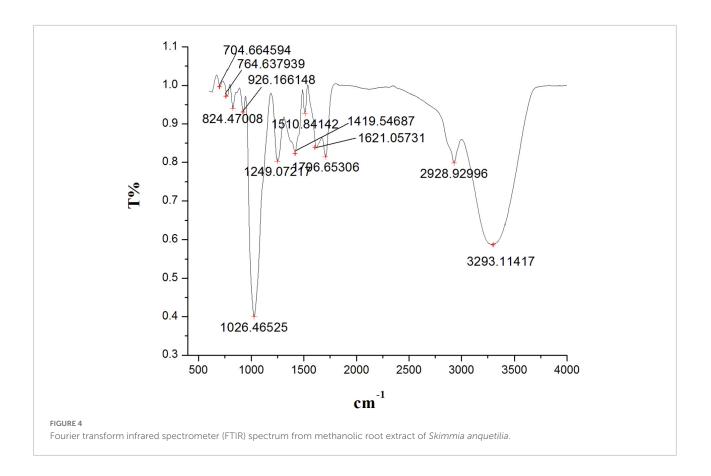
Antibacterial efficacy of n-hexane, ethyl acetate, and methanol root extracts of S. anquetilia was evaluated through the agar well diffusion technique (Clinical and Laboratory Standards Institute, 2008). The nutrient agar media tubes (20 mL) were inoculated with freshly prepared bacterial inoculums using a sterile loop in a back-and-forth motion to ensure an even distribution of inoculums. Petri plates were prepared by pouring pre-inoculated media and allowing it to solidify, and then 8 mm wells were made using a sterile cork borer. A total of 100 µL of different concentrations of each extract and an equal volume of negative control (DMSO) were poured into the wells. The plates were set aside to rest for 30 min to enable the extract to be pre-diffused into the media and were incubated at 37 °C for 17 h. Thereafter, the plates were examined for inhibition zones, and the findings were compared to gentamycin (10  $\mu$ g mL<sup>-1</sup>).

# Determination of minimum inhibitory concentration

The technique of macro-broth dilution (Clinical and Laboratory Standards Institute, 2008) was used to evaluate the antibacterial potential of ethyl acetate extract by measuring the noticeable bacterial growth in MHB. For MIC estimation in MHB, two-fold serial dilutions of the extract at varying concentrations from 64 to 4 mg mL<sup>-1</sup> with an optimized concentration of bacterial strains ( $10^8$  CFU mL<sup>-1</sup>) using 0.5 McFarland standard. The positive control included inoculated broth whereas the negative control included only plant extract and was incubated at 37 °C for 18 h. The MIC is the least concentration of extracts at which the tubes do not show any noticeable growth. To determine the value of MIC, the test tubes were observed for their visible turbidity both pre as well as post-incubation.

### Statistical analysis

All experiments were carried out in three replicates. Data were expressed as mean  $\pm$  standard deviation and evaluated by analysis of variance (ANOVA). Differences with p < 0.05 were considered significant.



# Results

### Physical properties and percent yield

The various extracts possessed varied colors. The extract of *n*-hexane appeared dark brown, the ethyl acetate extract was brown and the methanolic extract was reddish-brown (**Figure 1**). The methanol extract of *S. anquetilia* had the highest yield (15%), followed by ethyl acetate extract (8.6%), while *n*-hexane extract had the lowest yield (3.3%).

# Fourier transform infrared spectrometer analysis

The FTIR spectrum indicated the existence of functional groups in the *n*-hexane root extract of *S. anquetilia* with peak positions at 2922.74 cm<sup>-1</sup>, 2854.72 cm<sup>-1</sup> (alcohols, carboxylic acids, alkanes, and amine salts), 1731.38 cm<sup>-1</sup> (aromatic compounds, aldehydes), 1455.18 cm<sup>-1</sup> (aromatics), 1369.16 cm<sup>-1</sup> (alcohols, phenols), 1160.44 cm<sup>-1</sup> (amines, tertiary alcohols), 1077.29 cm<sup>-1</sup> (amines, primary alcohols), 865.39 cm<sup>-1</sup> (alkenes, aromatics), 835.80 cm<sup>-1</sup>

(alkenes), and 721.41  $\text{cm}^{-1}$  (alkenes) (Figure 2 and Table 1).

The peaks at 2929.80 cm<sup>-1</sup>, 2856.78 cm<sup>-1</sup> (alcohols, carboxylic acids, alkanes, and amine salts), 1712.83 cm<sup>-1</sup> (aromatic compounds, aliphatic ketones, and carboxylic acids), 1621.05 cm<sup>-1</sup> (conjugated alkenes, amines, and cyclic alkenes), 1520.30 cm<sup>-1</sup> (aromatics), 1455.18 cm<sup>-1</sup> (aromatics), 1376.36 cm<sup>-1</sup> (alcohols, phenols), 1238.76 cm<sup>-1</sup> (amines, alkyl aryl ether), 1158.06 cm<sup>-1</sup> (amines, tertiary alcohols), 1030.58 cm<sup>-1</sup> (amines), 815.01 cm<sup>-1</sup> (alkenes), and 714. 26 cm<sup>-1</sup> (alkenes) confirmed the presence of functional groups in ethyl acetate root extract (**Figure 3** and **Table 2**).

Similarly, the FTIR spectra of the methanol root extract of *S. anquetilia* revealed the presence of functional groups with peak ranges at 3293.11 cm<sup>-1</sup> (alcohols, carboxylic acids, and alkynes), 2928.92 cm<sup>-1</sup> (alcohols, amine salts, carboxylic acids, and alkanes), 1706.65 cm<sup>-1</sup> (aromatic compounds, aliphatic ketones, carboxylic acids, conjugated acids, and conjugated aldehydes), 1621.05 cm<sup>-1</sup> (cyclic alkenes, amines, and conjugated alkenes), 1510.84 cm<sup>-1</sup> (aromatic compounds), 1419.54 cm<sup>-1</sup> (carboxylic acids, alcohols, and aromatics), 1249.07 cm<sup>-1</sup> (acid, alkyl aryl ether, and amines), 1026.46 cm<sup>-1</sup> (amines, phosphate ion), 926.16 cm<sup>-1</sup> (alkenes), 824.47 cm<sup>-1</sup> (alkenes), 764.63 cm<sup>-1</sup> (-), and 704.66 cm<sup>-1</sup> (alkenes) (**Figure 4** and **Table 3**).

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TABLE 3 Fourier transform infrared spectrometer (FTIR) peaks an	d
their assigned functional groups of methanol root extract of	
Skimmia anguetilia.	

Carboxylic acid O-H stretching Alkyne C-H stretching Alkyne C-H stretching Amine salt N-H stretching Alkane C-H stretching Carboxylic acid O-H stretching Carboxylic acid O-H stretching Carboxylic acid C-H bending Carboxylic acid C=O stretching Carboxylic acid C=O stretching Conjugated acid C=O stretching Conjugated acid C=O stretching aldehyde 4. 1621.05 Cyclic alkene C=C stretching Conjugated acid C=O stretching alkene 5. 1510.84 Aromatic C=C stretching compound 6. 1419.54 Carboxylic acid O-H bending 7. 1249.07 Acid C-O stretching Amine C-N stretch	S. No.	Wavenumber (cm <sup>-1</sup> )	Compound class	Functional group
Alkyne C-H stretching Alkyne C-H stretching Anine salt N-H stretching Alkane C-H stretching Alkane C-H stretching Carboxylic acid O-H stretching Carboxylic acid C-H bending Carboxylic acid C=O stretching Conjugated acid C=O stretching aldehyde 4. 1621.05 Cyclic alkene C=C stretching Conjugated C=C stretching alkene 5. 1510.84 Aromatic C=C stretching compound 6. 1419.54 Carboxylic acid O-H bending 7. 1249.07 Acid C-O stretching Alkyl aryl ether C-O stretching Amine C-N stretching Amine C	1.	3293.11	Alcohol	O-H stretching
2. 2928.92 Alcohol O-H stretching Amine salt N-H stretching Alkane C-H stretching Carboxylic acid O-H stretching Carboxylic acid C-H bending compound Aliphatic ketone C=O stretching Conjugated acid C=O stretching Conjugated acid C=O stretching Conjugated acid C=O stretching aldehyde 4. 1621.05 Cyclic alkene C=C stretching Conjugated C=C stretching alkene 5. 1510.84 Aromatic C=C stretching alkene 5. 1510.84 Aromatic C=C stretching Conjugated C=C stretching Alcohol O-H bending Conjugated C=C stretching Conjugated C=C stretching alkene 5. 1510.84 Aromatic C=C stretching Conjugated C=C stretching Conjugated C=C stretching Alcohol O-H bending Alcohol O-H bending Alcohol O-H bending Alcohol O-H bending Alkyl aryl ether C-O stretching Amine C-N stretching 8. 1026.46 Phosphate ion PO <sub>3</sub> stretching Amine C-N stretching 9. 926.16 Alkene C=H bending 11. 764.63 Alkene C=H bending			Carboxylic acid	O-H stretching
Amine salt N-H stretching Alkane C-H stretching Carboxylic acid O-H stretching Carboxylic acid C-H bending compound Aliphatic ketone C=O stretching Carboxylic acid C=O stretching Conjugated acid C=O stretching Conjugated acid C=O stretching aldehyde 4. 1621.05 Cyclic alkene C=C stretching Conjugated C=C stretching Conjugated C=C stretching alkene 5. 1510.84 Aromatic C=C stretching compound 6. 1419.54 Carboxylic acid O-H bending Alcohol O-H bending 7. 1249.07 Acid C-O stretching Alkyl aryl ether C-O stretching Amine C-N stretching Amine C-N stretching Amine C-N stretching 9. 926.16 Alkene C=H bending 11. 764.63 Alkene C=H bending			Alkyne	C-H stretching
Alkane     C-H stretching       Carboxylic acid     O-H stretching       Carboxylic acid     O-H stretching       Carboxylic acid     C-H bending       Conjugated acid     C=O stretching       Conjugated     C=O stretching       Conjugated     C=C stretching       S.     1510.84     Aromatic       Alkohol     O-H bending       Alcohol     O-H be	2.	2928.92	Alcohol	O-H stretching
Carboxylic acid O-H stretching Carboxylic acid O-H stretching Aliphatic ketone C=O stretching Carboxylic acid C=O stretching Conjugated acid C=O stretching Conjugated acid C=O stretching aldehyde 4. 1621.05 Cyclic alkene C=C stretching Conjugated C=C stretching alkene 5. 1510.84 Aromatic C=C stretching compound 6. 1419.54 Carboxylic acid O-H bending Alcohol O-H bending 7. 1249.07 Acid C-O stretching Alkyl aryl ether C-O stretching Alkyl aryl ether C-O stretching Amine C-N stretching Amine C-N stretching 9. 926.16 Alkene C=H bending 11. 764.63 Alkene C=H bending			Amine salt	N-H stretching
3.     1706.65     Aromatic compound     C-H bending compound       Aliphatic ketone     C=O stretching Carboxylic acid     C=O stretching C=O stretching Conjugated acid       Carboxylic acid     C=O stretching Conjugated acid     C=O stretching C=O stretching aldehyde       4.     1621.05     Cyclic alkene     C=C stretching Conjugated       5.     1510.84     Aromatic Compound     C=C stretching Conjugated       6.     1419.54     Carboxylic acid     O-H bending Conjugated       7.     1249.07     Acid     C-O stretching Alicohol       7.     1249.07     Acid     C-O stretching Alikyl aryl ether       8.     1026.46     Phosphate ion     PO <sub>3</sub> stretching Amine       9.     926.16     Alkene     C=H bending       11.     764.63     Alkene     C=H bending			Alkane	C-H stretching
compoundAliphatic ketoneC=O stretchingCarboxylic acidC=O stretchingConjugated acidC=O stretchingConjugated acidC=O stretchingaldehydeC=O stretching4.1621.05Cyclic alkeneC=C stretchingAmineN-H bendingConjugatedC=C stretchingalkeneC=C stretching5.1510.84Aromatic compound6.1419.54Carboxylic acidO-H bending7.1249.07AcidC-O stretching8.1026.46Phosphate ionPO3 stretching9.926.16AlkeneC=H bending10.824.47AlkeneC=C bending			Carboxylic acid	O-H stretching
Carboxylic acidC=O stretchingCarboxylic acidC=O stretchingConjugated acidC=O stretchingConjugatedC=O stretchingaldehydealdehyde4.1621.05Cyclic alkeneC=C stretchingAmineN-H bendingConjugatedC=C stretchingalkeneC=C stretching5.1510.84AromaticC=C stretching6.1419.54Carboxylic acidO-H bending7.1249.07AcidC-O stretchingAmineC-N stretchingAmineC-N stretching8.1026.46Phosphate ionPO3 stretching9.926.16AlkeneC=H bending10.824.47AlkeneC=C bending11.764.63AlkeneC=H bending	3.	1706.65		C-H bending
$\begin{array}{cccc} & & & & & & & & & & & & & & & & & $			Aliphatic ketone	C=O stretching
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aldehyde       4.     1621.05     Cyclic alkene     C=C stretching       Amine     N-H bending       Conjugated     C=C stretching       alkene     C=C stretching       alkene     C=C stretching       5.     1510.84     Aromatic       C=C stretching     compound       6.     1419.54     Carboxylic acid     O-H bending       7.     1249.07     Acid     C-O stretching       Alkyl aryl ether     C-O stretching       Amine     C-N stretching       8.     1026.46     Phosphate ion     PO <sub>3</sub> stretching       9.     926.16     Alkene     C=H bending       10.     824.47     Alkene     C=C bending       11.     764.63     Alkene     C=H bending			Conjugated acid	C=O stretching
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926.16       Alkene       C=H bending         10.       824.47       Alkene       C=C bending         11.       764.63       Alkene       C=H bending	8.	1026.46	Phosphate ion	PO <sub>3</sub> stretching
10.       824.47       Alkene       C=C bending         11.       764.63       Alkene       C=H bending			Amine	C-N stretching
11. 764.63 Alkene C=H bending	9.	926.16	Alkene	C=H bending
	10.	824.47	Alkene	C=C bending
12. 704.66 Alkene C=C bending	11.	764.63	Alkene	C=H bending
	12.	704.66	Alkene	C=C bending

# Gas chromatography-mass spectrometer analysis

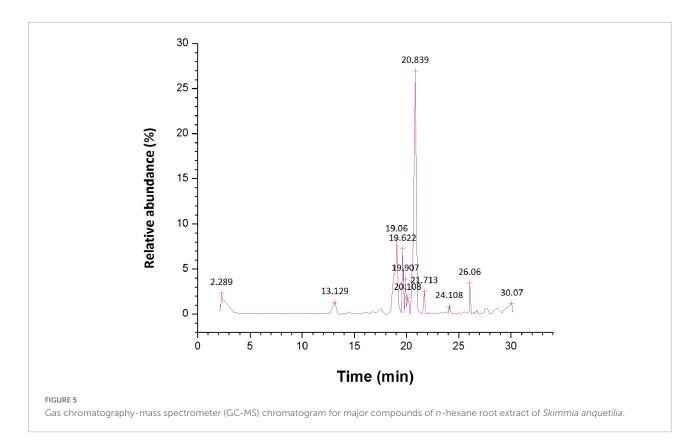
The GC-MS chromatogram of *n*-hexane, ethyl acetate, and methanol root extracts of *S. anquetilia* recorded a total of 88, 74, and 112 peaks respectively corresponding to the bioactive compounds that were recognized by relating their mass spectral fragmentation patterns to that of the known compounds described by the NIST library. The analysis of *n*-hexane extract *via* GC-MS resulted in the identification of 88 distinct phytoconstituents. The identified chemical constituents according to their retention time, peak area (%), and molecular weight are listed in **Supplementary Table 1**. The predominant organic constituents that were present in *n*-hexane extract (**Figure 5**) are 5, 10-pentadecadien-1-ol, (Z,Z)-(33.94%), *n*-hexadecanoic acid (13.41%), 8a(2H)phenanthrenol, 7-ethenyldodecahydro-1,1,4a,7- tetramethyl-, acetate, [4as-(4a.alpha.,4b.beta.,7.beta.,8a.alpha.,10a.beta.)]-(7.30%), 1-hexyl-2-nitrocyclohexane (4.35%), l-alanine, *N*-(3trifluoromethylbenzoyl)-, heptyl ester (4.02%), cyclopropane, 1-ethyl-2- methyl-, *cis*-(3.91%), squalene (3.59%), 7Hfuro(3,2-g)(1)benzopyran-7-one, 4-methoxy- (2.56%), 7H-furo[3,2-g][1]benzopyran-7-one, 4-methoxy-(1.82%), dihydro-*cis*- $\alpha$ -copaene-8-ol (1.78%), and cyclohexane (1.57%).

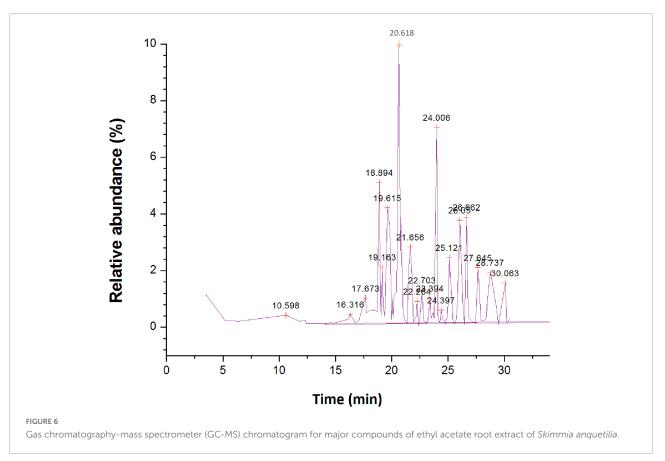
The GC-MS investigation of ethyl acetate extract led to the detection of 74 different organic constituents (**Supplementary Table 2**). The most abundant organic compounds found in the extract of ethyl acetate (**Figure 6**) are 5,10-pentadecadien-1-ol, (*Z*,*Z*)-(10.48%), 1-hexyl-2nitrocyclohexane (7.94%), phthalic acid, di(2-propylpentyl) ester (7.41%), 1-methylene-2b-hydroxymethyl-3,3-dimethyl-4b-(3-methylbut-2-enyl)-cyclohexane (5.59%), *n*-hexadecanoic acid (5.54%), 1,3,3-trimethyl-2-hydroxymethyl-3,3-dimethyl-4-(3-methylbut-2-enyl)-cyclohexene (4.65%), 2-cyclopropen-1-ol, 1,2-dicyclopentyl-3-(1-methylethyl)-, acetate (4.25%), squalene (3.79%), 1-alanine, *N*-(3-trifluoromethylbenzoyl)-, isohexyl ester (3.49%), 2R-acetoxymethyl-1,3,3-trimethyl-4t-(3-methyl-2-buten-1-yl)-1t-cyclohexanol (2.87%).

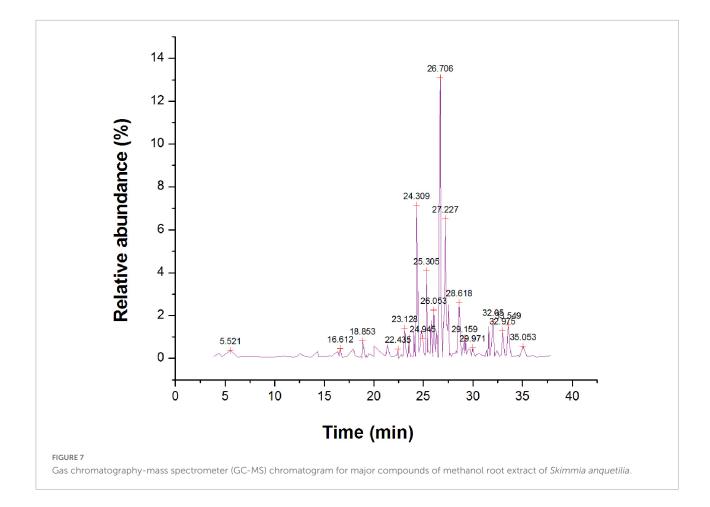
The methanol extract *via* GC-MS investigation had resulted in the detection of 112 different bioactive constituents (**Supplementary Table 3**). The main organic constituents found in methanol extract (**Figure 7**) are 1-hexyl-2-nitrocyclohexane (15.43%), *cis,cis,cis-7*,10,13-hexadecatrienal (13.29%), methyl 9-*cis*, 11-*trans*-octadecadienoate (9.62%), hexadecanoic acid, methyl ester (7.24%), 5, 10-pentadecadien-1-ol, (Z,Z)-(6.28%), tetradecanoic acid, 12 methyl-, methyl ester, (S)-(4.54%), farnesyl butanoate (1.41%), and 7H-furo[3,2-g][1]benzopyran-7-one,4,9-dimethoxy- (2.62%). The bioactive compounds with significant antibacterial activities are presented in **Table 4**.

### Antibacterial activity

The findings of the agar well diffusion assay (**Table 5**) showed that all the extracts are active against the bacterial strains tested. The most effective extract was ethyl acetate extract, which has exhibited the greatest effect against *P. aeruginosa* with an inhibition zone of 18 mm, followed by *S. aureus* and *K. pneumoniae* each having an inhibition zone of 17 mm. The methanol extract displayed a similar trend of bacterial inhibition, with the maximum zone of inhibition against *S. aureus* (17 mm), followed by *K. pneumoniae* (16 mm), and *E. coli* (15 mm). Similarly, the *n*-hexane extract exhibited strong inhibition against *K. Pneumoniae* (17 mm) and *S. typhi* (17 mm) followed by *E. coli* (16 mm). Owing to the increased antibacterial activity in agar well diffusion analysis against the tested bacterial strains,







the MIC was evaluated in the ethyl acetate extract of *S. anquetilia*. The ethyl acetate extract showed the MIC of 4 mg mL<sup>-1</sup> against *S. aureus* (**Table 6**). Analysis of variance (P < 0.05) showed that there is significant difference between the strains with respect to the concentration of plant extracts used. Dissimilar letters show significant difference and similar letters show insignificant difference (Tukey's HSD test) (**Figure 8**). The highest antibacterial activity could be attributed to the presence of bioactive constituents in the extracts.

# Discussion

In the present study, the analysis of *n*-hexane, ethyl acetate, and methanol root extracts of *S. anquetilia* revealed the existence of different bioactive principles, namely flavones, coumarins, glycosides, alkenes, carboxylic acids, tannins, phenols, amines, alkaloids, steroids, ketones, terpenoids, sesquiterpenes, fatty acid esters and alcohols, phyto-sterols, diterpenes, triterpene, etc. The therapeutic potential of different extracts of *S. anquetilia* may be attributed to the presence of these bioactive phytoconstituents. The

FTIR analysis of S. anquetilia root extracts have shown several peaks signifying the presence of different functional groups in the extracts. The absence of absorption peak at 3,000–3,500 cm<sup>-1</sup> in the IR spectrum of *n*-hexane and ethyl acetate extracts predicted the presence of a hydroxyl group (OH<sup>-</sup>). The major peaks that appeared in the extracts indicated the existence of functional groups such as alcohols, carboxylic acids, alkanes, aldehydes, amines, tertiary alcohols, aromatic compounds, aliphatic ketones, alkynes, etc. The FTIR analysis of petroleum ether seed oil extract of Ziziphus spina-christi revealed the presence of alcohols, phenols, alkanes, alkenes, carbonyls, carboxylic acids, and aromatic compounds as major functional groups (Abubaker et al., 2021). Visveshwari et al. (2017) while analyzing the methanol extract of Ceropegia juncea revealed the presence of alcohols, aldehydes, alkynes, alkenes, esters, and amines groups. These functional groups confirmed that S. anquetilia comprises a range of pharmaceutically significant bioactive constituents. The GC-MS investigation of root extracts of S. anquetilia showed the existence of 88 phyto-compounds in *n*-hexane, 74 phyto-compounds in ethyl acetate, and 112 phyto-compounds in methanol extracts, which add to the therapeutic values of this plant species. Some of the various

S. No.	Compounds	Root extracts			Biological activity	References
		<i>n</i> -hexane Ethyl acetate Methan		Methanol		
1.	1-hexyl-2- nitrocyclohexane	+	+	+	Antimicrobial activity against Salmonella suis (ATCC 13076), Pseudomonas aeruginosa (ATCC 27583), Escherichia coli (ATCC 25922), Staphyllococcus aureus (ATCC 25923), Bacillus subtilis (ATCC 6633), Shigella sonnei (ATCC 11060), and Candida albicans (ATCC 10231).	Al-Wathnani et al., 2012
2.	2R-acetoxymethyl- 1,3,3-trimethyl-4t- (3-methyl-2-buten- 1-yl)-1t- cyclohexanol	+	+	+	Antibacterial activity against <i>Pseudomonas</i> aeruginosa, Escherichia coli, Klebsiella pneumoniae, Salmonella typhi, and Staphylococcus aureus.	Nabi et al., 2022a
3.	5, 10- pentadecadien-1-ol, (Z,Z)-	+	+	+	Antibacterial activity against <i>Staphylococcus</i> aureus and <i>Pseudomonas aeruginosa</i> .	Majeed et al., 2021
4.	7H-furo[3,2- g][1]benzopyran-7- one,4,9-dimethoxy-	+	-	+	Antimicrobial activity against Bacillus subtilis, Klebsiella pneumoniae, Aspergillus niger, and Candida albicans.	AlMalki, 2016
5.	<i>n</i> -hexadecanoic acid	+	+	+	Antibacterial activity against Bacillus subtilus (NCIM 2718), Staphylococcus aureus (ATCC 25923), Pseudomonas aurginosa (ATCC 27853), Klebseilla pneumoniae (ATCC 70063), and Escherichia coli (ATCC 25922).	Mickymaray et al., 2016
6.	Squalene	+	+	+	Antibacterial activity against Escherichia coli, Klebsiella pneumoniae, Vibrio harveyi Micrococcus roseus, and Staphylococcus aureus.	Dordab et al., 2021
7.	Hexadecanoic acid, methyl ester	-	-	+	Antibacterial activity against Staphylococcus aureus (W35), Pseudomonas aeruginosa (D31), Klebsiella pneumoniae (DF30), Klebsiella pneumoniae (B45), and Bacillus subtilis.	Lalthanpuii and Lalchhandama, 2019; Shaaban et al., 2021
8.	7-hydroxycoumarin	-	-	+	Antibacterial activity against Bacillus subtilis, Staphylococcus aureus, Streptococcus pyogene, Pseudomonas aeruginosa, Salmonella typhimurium, Escherichia coli, and antifungal activity against strains of Candida albicans, Candida krusei, Candida parapsilosis, and Cryptococcus neoformans.	Farshori et al., 2011
9.	Linalool	-	-	+	Antibacterial activity against Pseudomonas aeruginosa, and Shewanella putrefaciens.	Liu X. et al., 2020; Guo et al., 2021
10.	Geraniol	-	-	+	Antimicrobial activity against <i>Candida</i> and <i>Staphylococcus</i> genera.	Lira et al., 2020
11.	<i>cis,cis,cis</i> -7,10,13- hexadecatrienal	-	-	+	Antibacterial activity against Escherichia coli.	Li et al., 2014
12.	Cyclohexane	+	_	_	Antibacterial activity against Staphylococcus aureus, Staphylococcus epidermidis, Pseudomonas aeruginosa, and Escherichia coli.	Shoaib et al., 2019
13.	Phthalic acid, di(2- propylpentyl)ester	_	-	+	Antimicrobial activity against bacteria and yeasts, such as Shigella flexneri, Escherichia coli, Klebsiella pneumoniae, Bacillus cereus, Staphylococcus aureus, Enterococcus faecalis, Bacillus subtilis, Candida albicans, and Candida glabrata.	Chakraborty et al., 2022
14.	Methyl 9- <i>cis</i> , 11- <i>trans</i> - octadecadienoate	-	-	+	Antifungal activity against <i>Microsporum canis</i> and <i>Trichophyton mentagrophytes</i> .	Ouf et al., 2022
15.	Hexadecanoic acid, ethyl ester	_	+	_	Antibacterial activity against Aeromonas hydrophila, Edwardsiella tarda, and Vibrio ordalli.	Gohar et al., 2010
16.	Campesterol	+	-	-	Antibacterial activity against the rate limiting enzyme involved in cell wall synthesis of bacteria i.e., glucosamine 6 phosphate synthase (PDB ID – 4VF5) as protein target.	Ranjith, 2019

#### TABLE 4 Bioactive compounds with significant antimicrobial activity.

"+" indicate presence and "-" absence of phytocontituents.

TABLE 5 In-vitro antibacterial activity of Skimmia anquetilia root extracts against tested bacterial strains.

Organic extract	Concentration
0	$(mg mL^{-1})$

#### Zone of inhibition (mm) (Mean $\pm$ SD)

	(					
		Gram-negative bacteria				Gram-positive bacteria
		Escherichia coli	Pseudomonas aeruginosa	Klebsiella pneumoniae	Salmonella typhi	Staphylococcus aureus
<i>n</i> -hexane	10	$14.0\pm2.64$	$14.0\pm2.0$	$11.0\pm9.84$	$14.0\pm3.0$	$12.0\pm1.73$
	20	$14.0\pm2.0$	$13.0\pm1.0$	$15.0\pm3.0$	$12.0\pm1.0$	$13.0\pm1.0$
	40	$14.0\pm3.0$	$13.0\pm2.0$	$17.0\pm1.0$	$13.0\pm2.0$	$13.0\pm1.73$
	80	$16.0\pm2.64$	$15.0\pm3.0$	$17.0\pm2.0$	$17.0\pm2.64$	$15.0\pm2.3$
	160	$15.0\pm3.6$	$15.0\pm1.0$	$17.0\pm1.73$	$17.0\pm1.0$	$16.0\pm3.0$
Ethyl acetate	10	$12.0\pm1.73$	$11.0\pm1.73$	$13.0\pm1.73$	$11.0\pm1.0$	$15.0\pm3.6$
	20	$12.0\pm1.0$	$11.0\pm1.0$	$12.0\pm1.0$	$11.0\pm1.0$	$9.0\pm7.80$
	40	$13.0\pm1.0$	$12.0\pm2.0$	$14.0\pm1.0$	$13.0\pm1.0$	$14.0\pm1.0$
	80	$14.0\pm1.0$	$15.0\pm1.73$	$15.0\pm1.73$	$15.0\pm1.73$	$15.0 \pm 0$
	160	$16.0\pm1.0$	$18.0\pm1.0$	$17.0\pm1.0$	$16.0\pm2.0$	$17.0\pm1.0$
Methanol	10	$10.0\pm1.73$	$11.0\pm1.0$	$12.0\pm1.0$	-	$7.0\pm 6.08$
	20	$8.0\pm7.0$	$11.0\pm1.0$	$13.0\pm3.0$	-	$14.0\pm2.64$
	40	$12.0 \pm 0$	$12.0\pm1.0$	$13.0\pm1.73$	-	$13.0\pm1.0$
	80	$13.0\pm1.0$	$13.0\pm1.73$	$13.0\pm1.0$	-	$14.0\pm1.0$
	160	$14.0\pm1.0$	$14.0\pm1.0$	$16.0\pm1.0$	-	$17.0\pm3.6$
Positive control	10 µg disc	$29.6 \pm 1.52$	$31.0 \pm 1.0$	$30.3 \pm 1.52$	$30.6\pm0.57$	$30.3 \pm 1.52$

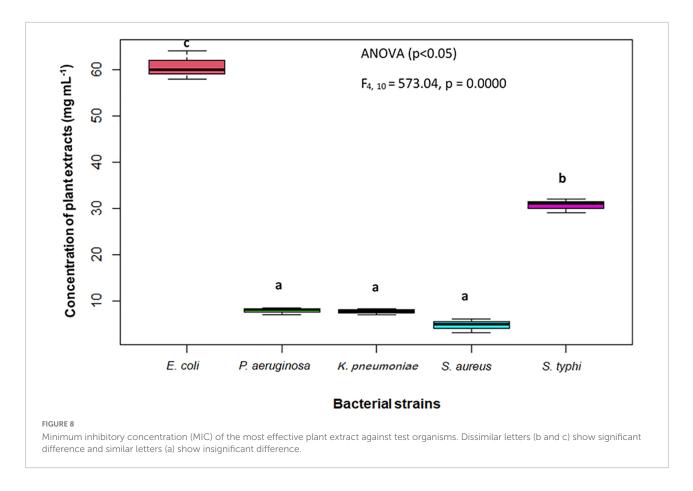
Data are means of three replicates (n = 3)  $\pm$  standard deviation.

bioactive compounds identified from the root extracts of S. anquetilia are known to possess significant biological activities. For instance, squalene is a triterpene and has been revealed to exhibit antitumor and antioxidant activities (Katerere et al., 2003; Amarowicz, 2009; Ganesh and Mohankumar, 2017), antimicrobial, and anticancer activity towards lung, skin, and colon oncogenesis (Rao et al., 1998; Smith, 2000). It also possesses anticancerous, gastro-preventive, hepatoprotective, and pesticidal properties (Ukiva et al., 2002; Ganesh and Mohankumar, 2017), and as an anti-inflammatory, immune-stimulant, and lipogenase inhibitor (Yamuna et al., 2017). The n-hexadecanoic acid is a fatty acid, possesses antimicrobial, antioxidant, antiatherosclerotic (Cho et al., 2010), antiandrogenic (Komansilan et al., 2012), anticancer (Sabithira and Udayakumar, 2017), and antitumor activities. The n-hexadecanoic acid identified in plant extracts of Benincasa hispida, Carissa congesta, Allium nigrum, Kielmeyera coriacea, Cyrtocarpa procera, Labisia pumila, and Rosa indica and has been reported to possess antibacterial activity (Mickymaray et al., 2016). Also, using the in-silico method, the *n*-hexadecanoic acid has been used to design inhibitors specific to phospholipase A (2) by comparing with other known inhibitors as anti-inflammatory agents (Aparna et al., 2012; Qureshi et al., 2016). Furthermore, hexadecanoic acid, methyl ester is suggested to be a fatty acid and is reported to have antibacterial, antioxidant, nematicide, insecticide properties and also helps in lowering cholesterol. Further, it possesses anticoronary, hepato-protective, antiacne, anti-inflammatory, antiarthritic, anticancer, antihistaminic, antieczemic, *a*-reductase inhibitor, and antiandrogenic activities (Krishnamoorthy and Subramaniam, 2014). Hexadecanoic acid, methyl ester has been shown to be a potent antibacterial agent against S. aureus W35, P. aeruginosa D31, K. pneumoniae DF30, and K. pneumoniae B45 strains (Shaaban et al., 2021). Lalthanpuii and Lalchhandama (2019) identified hexadecanoic acid, methyl ester as a major compound in Imperata cylindrica and reported its antibacterial properties against bacterial strains including P. aeruginosa, Bacillus subtilis, and K. pneumoniae. Various species use the antibacterial actions of fatty acid methyl ester to defend themselves against bacterial infections. Its principal focus of action is on bacterial cell membranes. It also disrupts cellular energy generation, impairs enzyme activity, and finally results in direct bacterial cell lysis. Owing to its safety and efficacy, it is a promising antibacterial therapeutic agent (Shaaban et al., 2021). (Z) 6, (Z) 9-pentadecadien-1-ol is fatty acid alcohol and possesses antibacterial activity (Sabithira and Udayakumar, 2017). 2R-acetoxymethyl-1,3,3-trimethyl-4t-(3-methyl-2-buten-1-yl)-1t-cyclohexanol is known to exhibit

TABLE 6 Minimum inhibitory concentration (MIC) of the most effective plant extract against test organisms.

S. No.	Bacterial strain	$MIC (mg mL^{-1})$
1.	Escherichia coli*	64
2.	Pseudomonas aeruginosa*	8
3.	Klebsiella pneumoniae*	8
4.	Salmonella typhi*	32
5.	Staphylococcus aureus**	4

\*Gram-negative bacteria. \*\*Gram-positive bacteria.



anticancer potential (Naine et al., 2016), anti-inflammatory, and antibacterial activities (Saravanan and Kasisankar, 2013). 1-hexyl-2-nitrocyclohexane is a ketone and exhibits antimicrobial (Al-Wathnani et al., 2012), and anti-inflammatory actions (Sivakumar and Gayathri, 2011; Ravisankar and Ester, 2017). 1-methylene-2b-hydroxymethyl-3,3-dimethyl-4b-(3-methyl but-2-enyl)-cyclohexane is sesquiterpene alcohol and it may act as an antimicrobial, anti-inflammatory, and anti-hyperlipidemic agent. 7H-furo[3,2-g][1]benzopyran-7-one,4,9-dimethoxy- also known as isopimpinellin is a furano-coumarin and it possesses antibacterial, antifungal, antibiofilm, and antioxidant potential (AlMalki, 2016).

Isopimpinellin, isolated from the hexane extract of *Peucedanum zenkeri* seeds, exhibited antibacterial property against *Cryptococcus neoformans* and *Mycobacterium intracellulare* (Ngunde Ngwendson et al., 2003; Mbah et al., 2010). In addition, it has shown significant insulinstimulated lipogenesis inhibition, suggesting that it may trigger the lipolytic hormonal behavior and specifically reduces the antilipolytic hormonal effects (Kimura et al., 1982). Furthermore, it demonstrated mild cytotoxicity 39.2  $\mu$ g mL<sup>-1</sup> (IC50 value) to Colo-205 (Yang et al., 2003). 7-hydroxycoumarin also known as umbelliferone is a hydroxy-coumarin and it has been found to have

several bioactivities viz., antibacterial (Farshori et al., 2011), anticancer, and antidiabetic particularly type-2 diabetes mellitus (Ofentse, 2017). 7-hydroxyl-coumarin derivatives demonstrated effective antifungal and antibacterial activity against bacterial strains such as B. subtilis, S. aureus, Streptococcus pyogene, P. aeruginosa, Salmonella typhimurium, E. coli, and fungal strains of Candida albicans, Candida krusei, Candida parapsilosis, and C. neoformans (Farshori et al., 2011). 1,3,3-trimethyl-2-hydroxymethyl-3,3-dimethyl-4-(3-methyl but-2-enyl)-cyclohexene exhibits antibacterial and antioxidant activities (Sen et al., 2015). In this study, the root extracts of S. anquetilia exhibited remarkable invitro antibacterial potential towards the gram-positive and gram-negative bacterial strains by inhibiting their colony and growth rate, which is the first report from this study. Recently, Nabi et al. (2022a) reported the antibacterial potential of methanol leaf extract of S. anquetilia against P. aeruginosa, E. coli, K. pneumoniae, S. typhi, and S. aureus. Previously, the antimicrobial activity using the essential oil extracted from S. laureola was documented (Jangwan et al., 2010; Irshad et al., 2012). Similarly, Zeb et al. (2016) reported the antibacterial activity of aqueous leaf extract of S. laureola. It can be interpreted from the aforementioned justification that S. anquetilia possesses a wide range of therapeutic phytoconstituents capable of an array of bioactivities such as antibacterial, antifungal, antioxidant, anti-inflammatory, antidiabetic, antiaging, anticancer, hepatoprotective, hypercholesterolemic, antihistaminic, anticoagulant, diuretic, etc. Therefore, the detection of different phytoconstituents of *n*-hexane, ethyl acetate, and methanol root extracts of *S. anquetilia* exhibits important pharmacological applications. Besides, investigations such as bioprospecting are required to sustain its pharmacological attributes and the biological significance of these novel bioconstituents will be noteworthy to be explored.

# Conclusion

The present study is the first report on the identification of various bioactive compounds by GC-MS analysis and antibacterial efficacy of root extracts of S. anquetilia. 5, 10pentadecadien-1-ol, (Z,Z)- in n-hexane and ethyl acetate extracts, and 1-hexyl-2 nitrocyclohexane in methanol extract were the major compounds, respectively. These compounds may be responsible for the different therapeutic and pharmacological properties of S. anquetilia in conventional medicine. The root extracts of S. anquetilia demonstrated promising antibacterial efficacy towards the gram-positive as well as gram-negative bacterial strains as is obvious by high inhibition-zones, and lower MICs. This study, therefore, recommends the assessment of antibacterial activity of bioactive compounds from root extracts of S. anquetilia, which could be a vital source for the development of novel antibacterial drug candidate beneficial in the management of life-threatening microbial infections.

# Data availability statement

The original contributions presented in this study are included in the article/**Supplementary material**, further inquiries can be directed to the corresponding author.

## References

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## Author contributions

MN: conceptualization, methodology, data curation, formal analysis, software, and writing—original draft, review, and editing. NT: resources, investigation, and supervision. BAG: resources, investigation, supervision, and review. All authors contributed to the article and approved the submitted version.

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# Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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## Supplementary material

The Supplementary Material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/ fpls.2022.937946/full#supplementary-material

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