



Differential Response to NaCl Osmotic Stress in Sequentially Harvested Hydroponic Red and Green Basil and the Role of Calcium

Michele Ciriello^{1†}, Luigi Formisano^{1†}, Georgios A. Soteriou², Angelos Kyratzis², Stefania De Pascale¹, Marios C. Kyriacou² and Youssef Rouphael^{1*}

Agricultural Research Institute, Nicosia, Cyprus

¹ Department of Agricultural Sciences, University of Naples Federico II, Naples, Italy, ² Department of Vegetable Crops,

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*Correspondence:

Youssef Rouphael youssef.rouphael@unina.it †These authors have contributed equally to this work

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Ciriello M, Formisano L, Soteriou GA, Kyratzis A, De Pascale S, Kyriacou MC and Rouphael Y (2022) Differential Response to NaCl Osmotic Stress in Sequentially Harvested Hydroponic Red and Green Basil and the Role of Calcium. Front. Plant Sci. 13:799213. doi: 10.3389/fpls.2022.799213 Basil (Ocimum basilicum L.) is a heterogeneous reservoir of bioactive compounds that provide recognized benefits to human health, rendering it a model aromatic herb. Notwithstanding the application of nutritional stress, such as sodium chloride (NaCl) salinity, which mainly affects the primary metabolism, it also triggers adaptive mechanisms that involve the production of bioactive secondary metabolites. Genotype selection and the exogenous application of calcium chloride (CaCl₂) help minimize salinity's suppressive effects on growth. In the present study, we hypothesize that the ratio of different salt types may induce differential responses in the function of preharvest factors in hydroponic basil culture. In this perspective, the stock nutrient solution (Control) was supplemented with 12.5 mm NaCl + 8.33 mm CaCl₂ (Moderate Mix), 25 mm NaCl (Moderate NaCl), 25 mm NaCl + 16.66 of CaCl₂ (High Mix), or 50 mM of NaCl (High NaCl) with the objective of evaluating the different impact of salinity on yield, sensory quality (color and aroma profile), and the accumulation of minerals and bioactive compounds in two successive harvests of green and red basil cultivars. Although more productive (+39.0% fresh weight) than the red one, the green cultivar exhibited higher susceptibility to salinity, especially under the High Mix and High NaCl treatments. The addition of CaCl₂ to the High Mix solution reduced the sodium by 70.4% and increased the total polyphenols by 21.5% compared to the equivalent isomolar solution (High NaCl). The crop performance in terms of fresh and dry yield improved for both cultivars at the second cut. Regardless of cultivar and salt treatment, successive harvests also increased the concentration of phenols and vitamin C (29.7 and 61.5%, respectively) while reducing (-6.9%) eucalyptol, the most abundant aromatic compound in both cultivars. Salinity, as well as the mechanical stress induced by cutting, improved the functional quality of basil. However, the productive responses to the conditions imposed in our work once again highlighted the importance of genetic background. Specifically, CaCl₂ in the Moderate Mix solution preserved fresh leaf weight in the most stress-sensitive green cultivar.

Keywords: Ocimum basilicum L., nutritional stress, successive harvests, isomolar concentrations, volatile compounds, calcium chloride

INTRODUCTION

Basil (Ocimum basilicum L.) is a native herbaceous plant of southern Asia, also known as the "king of herbs" for its fragrant, delightful, and colorful leaves (Ciriello et al., 2020; Formisano et al., 2021b). As a culinary herb, it is preferably used fresh to improve the aroma and appearance of salads, pizza, meat, and fish dishes (Corrado et al., 2020a,b). According to vernacular knowledge, basil has wound-healing properties and its leaves and flowering tops are used for preparing infusions claimed to have sedative, antispasmodic, stomachic, diuretic, and antimicrobial effects (Ahmed et al., 2019; Amor et al., 2021). Furthermore, it is also used to treat indigestion and as a vermifuge due to its bioactive content and high nutraceutical value (Ch et al., 2015). These secondary metabolites of aromatic herbs are a precious reservoir of bioactive molecules that the pharmaceutical industry could use to develop new drugs and improve their natural chemical diversity (David et al., 2015). In this respect, recent scientific research has highlighted how genotype and agronomic practices, such as successive harvests, can be a valuable tool toward increasing the levels of desired secondary metabolites (Nicoletto et al., 2013; Ciriello et al., 2021a,b).

In the previous three decades, research efforts were focused mainly on boosting the desirable phytochemicals (e.g., ascorbate, carotenoids, glucosinolates, polyamines, and polyphenols) in fruits and vegetables through plant breeding and biotechnology, but have yet achieved limited commercial success because of safety issues (Poiroux-Gonord et al., 2010). For instance, the disproportional increase in some secondary metabolites generated through breeding and biotechnology (e.g., a 36fold increase in kaempferol-rutinoside, a 20-fold increase in carotenoids, and glucosinolates) has been found genotoxic (Rouphael and Kyriacou, 2018). Moreover, the improvement of plant phytochemicals through breeding constitutes a costlier approach than crop management practices. Therefore, the manipulation of phytochemicals through controlled agroenvironmental conditions (air and root zone temperature, light quality and intensity, carbon dioxide enrichment, and vapor pressure deficit), as well as innovative crop management practices (nutritional eustress, nutrient solution management, and biofortification) constitutes an approach balancing safety, cost, and effectiveness (Poiroux-Gonord et al., 2010). Unsurprisingly, this approach has attracted the interest of food technologists and extension specialists as it additionally quells consumers' concerns over genetic modification of food products (Kyriacou and Rouphael, 2018). In this perspective, nutrient solution management posits as a significant toolbox for improving the functional quality of vegetal food (Rouphael and Kyriacou, 2018). As demonstrated in the current work, soilless culture may facilitate the effective and precise control of the plant-nutrient interface to achieve target concentrations in key secondary metabolites (Fallovo et al., 2009a,b; Colla et al., 2013a; Borgognone et al., 2016). Moreover, accurate application of chemical eustress without soil interaction, such as the controlled deprivation of macronutrients, exposure to mild or moderate salinity [Sodium Chloride (NaCl)], and sustained exposure of the roots to variable cationic and

anionic proportions or single ions, may boost the bioactive phytochemical content of vegetables within the optimal range (Rouphael and Kyriacou, 2018).

A prerequisite to achieving these targets is to identify the optimal balance between exposure time and phenological stage of stress application, as there is strong crop intraspecific and interspecific genetic variability toward stress tolerance (Rouphael et al., 2018). Moderate salt concentrations in the soil solution effectively trigger metabolic pathways involved in the production of secondary metabolites in the context of a concerted adaptive mechanism against oxidative damage induced by salinity (Tarchoune et al., 2013). However, salinity levels exceeding the crop's tolerance threshold can induce morphophysiological and metabolic changes capable of retarding and/or inhibiting plant growth (Ntatsi et al., 2017). Salinity damage is primarily co-induced by excess NaCl, which jointly triggers osmotic (low water potential) and ion stress (toxicity and ion-specific imbalance) (Munns and Tester, 2008; Colla et al., 2013b). The combination of both stressors leads to chlorophyll and carotenoids pigment degradation, stomatal closure, and photosystem II (PSII) blockage, with a consequent reduction in carbon organization and impaired translocation and uptake of critical nutrients (das Neves et al., 2008; Liu et al., 2012).

Calcium, due to its role as a structural component of membranes and cell walls, is crucial for plant survival under suboptimal environmental conditions (Colla et al., 2013b; Cheruth et al., 2016). Its regulatory role in nutrient uptake processes, implicated in ensuring the proper integrity and permeability of cell membranes, mitigates the ionic effect of NaCl (Borgognone et al., 2014a). Several authors have suggested that the exogenous application of calcium may alleviate the harmful effects induced by excess NaCl in model crops such as wheat (Triticum aestivum L.), tomatoes (Solanum lycopersicum L.), peas (Pisum sativum L. cv. Argona), and sunflowers (Helianthus annuus L.) (Bonilla et al., 2004; Zhou and Xiao, 2010; Borgognone et al., 2014a). Specifically, calcium drives ion-selective transport and maintains the key K/Na and Ca/Na ratios at optimal levels by reducing the cytosolic Na uptake (Nedjimi and Daoud, 2009).

Most studies on the salt stress tolerance of vegetable crop species have focused on the role of NaCl as the predominant salt (Bernstein et al., 2010; Zahedi et al., 2011; Barbieri et al., 2012). Only a few of these studies have additionally evaluated the calcium-induced ameliorative effects on the growth and physiology of crops, such as cucumber (Cucumis sativus L. cv. Palmera) (Trajkova et al., 2006), basil (Ocimum basilicum L. cv. Thai Siam Queen) (Scagel et al., 2017), marjoram (Satureja montana) (Hegazy et al., 2019), chicory (Cichorium spinosum) (Ntatsi et al., 2017), lettuce (Lactuca sativa L. cv. Lollo Rossa) (Borghesi et al., 2013; Cheruth et al., 2016), sorghum (Sorghum bicolor L. Monench, cv. ICI-5521) (Dashti et al., 2009), and wheat (Nemat Alla et al., 2014). However, these authors mainly used variable NaCl molar concentrations without distinguishing the osmotic from the ionic effect and without assessing the natural sensitivity of the cultures to sodium (Na) and chlorine (Cl). To date, few studies have investigated the ameliorative effects of CaCl₂ on NaCl-induced salt stress using isomolar concentrations of the two salts (Colla et al., 2013b; Borgognone et al., 2014a,b).

In the present study, we hypothesize that not only different kinds of salt but also their ratio may induce differentiated responses in basil crops. Understanding these responses could help growers pursue appropriate agricultural practices to improve plant tolerance to salinity, enhance the plant secondary metabolome and enable efficient production under prevailing suboptimal conditions. Based on this hypothesis, the objective of the present work was to evaluate the effect of two salts (NaCl and CaCl₂), added at isomolar concentrations to the stock nutrient solution (Control) in a hydroponic greenhouse system, with respect to the growth, quality, and volatile aromatic components of green (cv. Mammolo) and red (cv. Red Rubin) basil (*O. basilicum* L.), harvested twice during the crop cycle (first and second cut).

MATERIALS AND METHODS

Experimental Site and Design and Management of the Nutrient Solution

The experimental trial was carried out at the "Torre Lama" experimental farm of the University of Naples Federico II -Department of Agriculture (DIA) located in Bellizzi (SA, Italy; lat. 43° 31' N, long. 14° 58' E; alt. 60 m above sea level). The trial was conducted in a glass greenhouse 10 m wide, 30 m long, and 3 and 4.5 m high at the eaves and ridge, respectively, from April 29, 2015, to July 2, 2015. Twenty days after sowing (April 29, 2015; DAT 1), two basil (O. basilicum L.) cultivars with different pigmentation "Mammolo" (hereafter "Green"; L'ortolano, Cesena, FC, Italy) and "Red Rubin" (hereafter "Red"; Ortomio, Forlì, FC, Italy) were transplanted into square antispiral pots (0.1 m \times 0.1 m \times 0.15 m) filled with 1.3 L volume of a mixture (v/v) consisting of 1/3 perlite (Vigorplant, Fombio, LO, Italy) and 2/3 peat (Vigorplant, Fombio, LO, Italy). The seedlings were arranged in rows with a spacing of $0.30 \text{ m} \times 0.15 \text{ m}$ yielding a density of 22 plants m^{-2} .

The plants were arranged according to a factorial design with three replicates in which a non-saline control (Control) vs. four saline nutrient solutions (Moderate Mix, Moderate NaCl, High Mix, High NaCl), two basil cultivars (Green and Red), and two successive harvests (Cut1 and Cut2) were considered as factors. Each experimental unit (i.e., replicate) included 15 plants. The standard nutrient solution (NS; Control) was a Hoagland modified as follows: 13.0 mM NO3-N, 1.0 mM NH4-N, 1.75 mM S, 1.5 mM P, 5.0 mM K, 4.5 mM Ca, 2 mM Mg, 20 μ M Fe, 9 μ M Mn, 0.3 μ M Cu, 1.6 μ M Zn, 20 μ M B, and 0.3 μ M Mo. Saline NS were prepared by adding isomolar concentrations of NaCl and CaCl₂ to the standard NS (Table 1). The NS was prepared using osmotized water and distributed through a drip irrigation system (1 dripper/plant with a flow rate of 2 L/h). To avoid large fluctuations in EC, pH, and the accumulation of salts in the substrates, for each pot, additional irrigations with water were provided to drain the excess salts, keeping the variation of EC within a value never higher or lower than 5% of the initial value.

Growth, Yield, and CIELab Colorimetric Index Measurements

At 42 and 65 days after transplanting (DAT) (June 9-Cut1 and July 2-Cut2), and just before flowering, the plants were harvested. At each cut, 10 plants per experimental unit were sampled, immediately weighed, and the leaves were separated. The fresh weight (FW), number, and total leaf area were determined using the LI-COR 3100C area meter (LI-COR Biosciences, Lincoln, NE, United States). The plant tissues were dried in a ventilated oven at 65°C for 72 h for total and leaf dry weight (DW) determination; the latter was used to determine leaf dry matter (DM) percentage (%) as follows: DM = 100 × leaf DW/leaf FW. Samples were finely ground and sieved with a cutting-grinding head mill (IKA[®], Staufen im Breisgau, BW, Germany). An aliquot of fresh ground leaves was immediately placed in liquid nitrogen and stored at -80° C for future qualitative analysis.

On the same dates (42 and 65 DAT), colorimetric measurements of the CIELab color space were made on the adaxial surface of 30 healthy and fully expanded leaves for treatment using a portable Minolta CR-300 Chroma Meter (Minolta Camera Co., Ltd., Tokyo, Kantō, Japan), calibrated according to the manufacturer's instructions. As defined by the Commission Internationale de l'Éclairage (CIE), color was expressed according to the three components L* (lightness), a* (greenness), and b* (yellowness) by which the Chroma (C*) and Hue angle (h°) values were determined.

Total Nitrogen, Nitrate, and Macronutrient Determination

Finely ground dry leaf samples were analyzed by the Kjeldahl method for total nitrogen determination and by ion chromatography for macronutrients [phosphorus (P), sulfur (S), potassium (K), calcium, magnesium (Mg), sodium, and chlorine] and nitrate, following the protocols described by Bremner (1965) and Formisano et al. (2021a), respectively.

Briefly, for the total nitrogen determination, 0.25 g of sample was mixed with 7 mL of 96% H_2SO_4 (Carlo Erba Reagents Srl., Milan, MI, Italy), a catalyst (Velp[®] Scientifica, Usmate Velate, MB, Italy), and 10 mL of 30% H_2O_2 (Carlo Erba Reagents Srl., Milan, MI, Italy). The obtained mixture was heated at 420°C with a DK 20 Heating Digester (Velp[®] Scientifica, Usmate Velate, MB, Italy) until mineralization (30 min). Mineralized samples were distilled with a UDK 140 distiller (Velp[®] Scientifica, Usmate

TABLE 1 | Total ion concentration and sodium chloride (NaCl) and calciumchloride (CaCl2) concentration in saline nutrient solutions.

NaCl	CaCl ₂	Total ion concentration			
mM	mM	mM			
0	0	0			
12.5	8.33	50			
25	0	50			
25	16.66	100			
50	0	100			
	mM 0 12.5 25 25	mM mM 0 0 12.5 8.33 25 0 25 16.66			

Velate, MB, Italy) and subsequently titrated with H_2SO_4 0.1 N with a methyl red and bromocresol green indicator (Carlo Erba Reagents Srl., Milan, MI, Italy) until the solution reached reddish color. The total nitrogen concentration values were expressed as g kg⁻¹ of dry weight (DW). Each treatment was analyzed in triplicate. The results obtained on the quantification of minerals, antioxidant activity, total phenols, and ascorbic acid (AsA) were expressed in terms of concentration according to the definition proposed by Bauer et al. (1997).

For the determination of macronutrient and nitrate concentrations, 0.25 g of sample was extracted in ultrapure water (Arium[®] Advance EDI pure water system, Sartorius, Göttingen, NI, Germany) and analyzed using an ICS-3000 ion chromatographic system (Thermo ScientificTM DionexTM, Sunnyvale, CA, United States), coupled with an electrical conductivity detector. The determination of anions (nitrate, sulfate, P, and Cl) and cations (K, calcium, Mg, and sodium) was performed by isocratic and gradient mode separation, respectively. The concentration of anions and cations in leaves were expressed as g kg⁻¹ of DW, except for nitrate expressed as $\mu g g^{-1}$ of FW. Each treatment was analyzed in triplicate.

Antioxidant Activity Determination, Total Phenols, and Total Ascorbic Acid

The determination of the antioxidant 2,2'-azino-bis(3-ethylbenzothiazoline-6-sulfonic acid) (ABTS) fraction was determined according to the method described by Pellegrini et al. (1998). This method is based on chromogen 2,2'-azino-bis(3-ethylbenzothiazoline-6-sulfonic acid), which registers in its radical form an absorption maximum at 734 nm. The ABTS antioxidant activity was expressed as g Trolox eq kg⁻¹ of DW.

The quantification of the total phenolic compounds was determined according to Folin-Ciocalteu's method described by Singleton et al. (1999). Briefly, 0.25 g of freeze-dried and ground leaves were added to 10 ml of 60% methanol, placed on a tilting plate for 15 min, and centrifuged. Subsequently, 125 μ l of the supernatant was taken, and 125 μ l of a mixture of phosphotungstic acid (H₃PW₁₂O₄₀) and phosphomolybdic acid (H₃Mo₁₂PO₄₀) (Folin-Ciocalteu reagent) was added to 0.5 mL of distilled water. In the presence of phenolic compounds, the Folin-Ciocalteu reagent is reduced to a mixture of tungsten (W₈O₂₃) and molybdenum (Mo₈O₂₃) oxides that has an absorption peak at 760 nm. The total phenolic concentration was expressed as mg gallic acid equivalents 100 g⁻¹ DW.

The total AsA concentration was determined by an assay based on the reduction of Fe^{3+} to Fe^{2+} by reduced AsA and the spectrophotometric detection of Fe^{2+} with 2,2'-dipyridyl (Kampfenkel et al., 1995). An aliquot of 0.4 g of frozen sample was weighed and then crushed in a mortar with 0.8 mL of 6% trichloroacetic acid (TCA). After incubation for 15 min at -20° C, 1.2 mL of 6% TCA was added to the extract, after which it was centrifuged at 4,000 rpm for 10 min. The obtained supernatant was taken to determine total AsA (AsA + dehydroascorbate). The absorbance was then read at a wavelength of 525 nm, which corresponds to the maximum absorption peak of the Fe/2,2-dipyridyl. UV-vis spectrophotometry (Hach DR 4000; Hach Co., Loveland, CO, United States) was used to read the absorbances of antioxidant activities, total phenols, and AsA. Each treatment was analyzed in triplicate.

Volatile Compounds Determination

In order to quantify volatile compounds (VOCs) *via* the solidphase microextraction (SPME) method, 0.5 g of basil leaves stored at -80° C were manually crushed and placed in a 20 ml glass vial that was shaken and heated for 10 min to promote the migration of VOCs into the headspace. The VOCs were absorbed by a divinylbenzene/carboxen/polydimethylsiloxane fiber (1 cm long, 50/30 μ m thick; Supelco[®], Bellefonte, PA, United States). VOCs were determined by GC 6890N gas chromatograph (Agilent, Santa Clara, CA, United States) coupled with MS 5973N mass spectrometer (Agilent, Santa Clara, CA, United States) equipped with a split-splitless injector and a capillary column (30 m \times 0.250 mm) coated with a 5% phenyl/95% dimethylpolysiloxane film (0.25 μ m; Supelco[®], Bellefonte, PA, United States).

Following a desorption phase of 10 min at 230°C, the oven temperature was kept at 50°C for 2 min and then ramped to 150°C at 10°C/min and from 150 to 280 at 15°C/min. The ion and injection source temperatures were 230 and 250°C, respectively. Helium was used as the carrier gas at a flow rate of 1 ml min⁻¹. The mass spectrometer was set at 70 eV. The identification of VOCs was facilitated through the use of the Atomic Spectra Database version 1.6 of the National Institute of Standards and Technology (NIST; U.S. Department of Commerce, Gaithersburg, MD, United States). Samples were analyzed in triplicates and the results were expressed as a percentage relative abundance (%).

Statistics

The experiment was set up in a randomized design (Completely Randomized Design) with three replicate plots per treatment. Thirty (2 cultivars \times 5 salinity levels \times 3 replications) fully randomized experimental plots were used for each cut (Cut 1 and Cut 2). Each experimental plot accommodated 15 plants. During each cut, all the 15 plants of each experimental plot were harvested. Following sampling, aggregates of 15 plants were made for each treatment replication. Therefore 450 plants (30 experimental plots \times 15 plants) were used during each cut and 900 plants for the whole study (Cut1 and Cut2). ANOVA was performed to assess the significance of the main effects [Cultivar (V), Salinity (S), Cut (C)] and their pair-wise interactions. During the statistical analysis, we ensured that all the assumptions of ANOVA were met. That is, the treatment replications samples were independent of each other, sample populations were normally distributed (normality), and the sample populations had the same variance (homoscedasticity). Three-way interactions were either non-significant or had a very low contribution to the total variance of the examined variables. Means for V and C were compared by the Student's *t*-test, whereas the means for S and all two-way interactions were compared using the Tukey-Kramer HSD test at the p < 0.05level. Data represent mean \pm SE of three replicates (n = 3). All

statistical analyses were performed using IBM SPSS 20 (Armonk, NY, United States) package for Microsoft Windows 10.

RESULTS

Morphometric and Yield Parameters

With respect to the main experimental factors, significant differences were observed between cultivars for leaf number, leaf area, and leaf FW (Table 2). Superior leaf number but smaller leaf area and leaf FW were observed in Red, reflecting the smaller leaves of this cultivar compared to Green, whereas leaf DM percentage and DW were not influenced by cultivar type. Salinity treatments had no effect on the leaf number and leaf DM percentage; however, the High Mix treatment and the High NaCl treatment suppressed the leaf area, leaf FW, and DW, which signals a deleterious effect on plant growth under both high-salt treatments. Cut order affected significantly all the morphometric variables examined, with leaf number, area and FW, and DW being higher at Cut2, at the expense, however, of leaf DM percentage. Significant salinity × cultivar interaction was observed for leaf area, leaf FW, and DW as the mean values of Red for these variables under the High Mix and High NaCl treatments did not differ significantly to the Control (non-salt) treatment. The rest of the interactions observed derived from changes in the scale of the observed effects but not in the ranking of treatments.

Leaf Colorimetric Components

Significant differences were observed between cultivars for the leaf colorimetric attributes examined, which largely reflects their apparent genotypic differences in color (Table 3). Lightness (L*) was overall higher in Green than Red and in Cut2 than Cut1, while slightly higher L* was also observed with salinity treatments compared to Control. Cultivar ranking did not change for L* across salinity treatments as the salinity \times cultivar interaction observed was due to change in scale rather than treatment rank. Similarly, cut ranking was invariable across salinity treatments despite salinity \times cut interaction. However, the significant cultivar \times cut interaction observed highlighted that leaf L* of Green remained unaltered between Cut1 and Cut2, as opposed to the sharp increase observed in Red. Significant variation in response to salinity treatments was only observed for Red as the red color intensity (+a*) decreased significantly with all salinity treatments, except for Moderate Mix when compared to the Control. Moreover, under both moderate and high salinity conditions, the NaCl treatment tended to suppress red color intensity more than the combined $NaCl + CaCl_2$ treatment. Significant cultivar \times cut interaction was also observed as the green color intensity $(-a^*)$ of green basil was not dependent on the cut, whereas the red color intensity $(+a^*)$ of Red decreased at Cut2. With respect to the overall leaf color saturation or chroma (C*), Green demonstrated higher C* than Red irrespective of salinity and cut. The cut order was not significant for leaf chroma in Green; however, Red demonstrated an increase in leaf chroma at Cut2.

Leaf Mineral Concentration

The choice of cultivar had a significant effect on the leaf concentration for all minerals analyzed. The cultivar was the only factor with a significant effect on leaf N concentration, which was found higher in Green than in Red (Table 4). P concentration was also higher in Green than Red, however, in the former cultivar, it decreased at Cut2 whereas in the latter it did not, as manifested in significant cultivar \times cut interaction. Salinity also had a significant effect on P concentration, which decreased in response to the High Mix and High NaCl treatments. For P, the interaction between the factors under consideration (salinity \times cultivar \times cut) was significant ($p \le 0.01$). In general, the lowest phosphorus concentration was obtained from the High NaCl \times Green \times Cut2 interaction (data not shown). For the same cultivar in the High Mix and Low Mix treatments, from the first to the second cut, phosphorus decreased by 40.7 and 37.1%, respectively (data not shown). S concentration incurred a significant effect with respect to the cultivar, as it was higher in Red than Green, however, it was not significantly affected by salinity or cut. K concentration was affected by cultivar and cut but not by salinity; it was found higher in Green and increased at Cut2. Calcium, Mg, sodium, and chlorine concentrations were the ones most affected by salinity treatments. Calcium concentration was lowest in the Control and highest in the High-Mix treatment, with the mix treatments resulting in higher Ca concentration than the corresponding isomolar NaCl treatments. The Ca concentration was higher in Red while in both cultivars it increased at Cut2. A salinity × cultivar interaction was observed mainly due to a change in scale between the two cultivars. Mg concentration was mostly affected by salinity and to a lesser extent by cultivar. The salinity effect was reflected solely on the High NaCl treatment which resulted in the highest Mg concentration, whereas no other salinity treatment differed from the Control. The effect of cut on Mg concentration was non-significant although it was implicated in a cultivar × cut interaction since Mg concentration in Green increased at Cut2 whereas in Red it was unaltered. Leaf sodium concentration was significantly affected by all the experimental factors examined. It was higher in Green and increased in response to all salinity treatments, except for the Moderate Mix treatment, which did not differ significantly from the non-salt Control treatment. The highest Na concentration was attained in response to the High NaCl treatment. The mean cultivar Na concentration doubled at Cut2 relative to Cut1, however, a cultivar × cut interaction revealed that the Na increase in Red was non-significant, whereas in Green a significant increase by 152.6% was noted. Chlorine concentration was significantly affected by salinity and cultivar but unlike sodium concentration, it demonstrated no significant change in response to the order of cuts. Chlorine concentration was higher in Red rather than Green and also incurred a significant increase with salinity treatments relative to the Control. The high salinity treatments induced higher Cl concentration than the moderate treatments while no significant differences were observed between the isomolar mix and plain NaCl treatments. None of the factors examined except salinity had a significant effect on the basil leaf nitrate concentration,

Source of variance	Leaf number	Leaf area	Leaf FW	Leaf DM	Dry weight (g plant ⁻¹)	
	n° plant ⁻¹	(cm ² plant ⁻¹)	(g plant ⁻¹)	(%)		
Cultivar (V)	***	*	***	ns	ns	
Salinity (S)	ns	***	***	ns	***	
Cut (C)	***	***	***	*	***	
$S \times V$	ns	***	***	ns	*	
S×C	ns	ns	ns	ns	*	
V×C	***	**	*	*	**	
			Means \pm standard error			
Cultivar						
Green	$122 \pm 8b$	$1639 \pm 89a$	$60.72 \pm 3.39a$	9.00 ± 0.20	7.96 ± 0.42	
Red	197 ± 14a	$1482 \pm 79b$	$43.40 \pm 2.48b$	9.18 ± 0.14	7.27 ± 0.48	
Salinity						
Control	159 ± 17	$1862 \pm 127a$	$64.05 \pm 6.56a$	8.58 ± 0.16	8.73 ± 0.83a	
Moderate Mix	185 ± 22	1893 ± 73a	$63.44 \pm 3.72a$	8.88 ± 0.21	$9.15 \pm 0.54a$	
Moderate NaCl	159 ± 23	$1657 \pm 130a$	$54.16 \pm 4.08ab$	9.22 ± 0.28	8.07 ± 0.75 ab	
High Mix	148 ± 21	$1291 \pm 97b$	$43.38 \pm 3.64 \text{bc}$	8.94 ± 0.29	$6.36 \pm 0.47 bc$	
High NaCl	138 ± 18	$1042 \pm 109b$	$34.15 \pm 2.96c$	9.88 ± 0.32	$5.51 \pm 0.47c$	
Cut						
Cut1	$101 \pm 5b$	$1287 \pm 69b$	$42.96 \pm 2.69b$	$9.21 \pm 0.14a$	$5.89 \pm 0.29b$	
Cut2	$219 \pm 11a$	$1853 \pm 76a$	$62.09 \pm 3.16a$	$8.96 \pm 0.20b$	$9.44 \pm 0.41a$	
S × V	210 2 114	1000 1 100	02100 ± 01104	0.00 ± 0.200		
Control × Green	136 ± 20	$2229 \pm 126a$	85.90 ± 5.70a	8.25 ± 0.18	$10.66 \pm 1.12a$	
Moderate Mix \times Green	130 ± 19	$1951 \pm 93ab$	$73.02 \pm 3.98ab$	8.38 ± 0.20	9.09 ± 0.69 ab	
Moderate NaCl × Green	118 ± 18	$1783 \pm 142abcd$	$60.38 \pm 3.63 \text{bc}$	8.90 ± 0.38	7.83 ± 0.69 ab	
High Mix \times Green	113 ± 15	$1286 \pm 91 def$	48.40 ± 4.53 cde	9.01 ± 0.47	6.57 ± 0.42 bo	
High NaCl × Green	111 ± 16	$947 \pm 130f$	$35.89 \pm 3.93e$	10.45 ± 0.48	$5.64 \pm 0.49c$	
Control × Red	182 ± 27	1494 ± 119 bcde	42.20 ± 3.97 de	8.91 ± 0.22	6.81 ± 0.78 bc	
Moderate Mix \times Red	239 ± 29	$1835 \pm 115abc$	53.86 ± 4.14 cd	9.38 ± 0.28	9.21 ± 0.87 ab	
Moderate NaCl × Red	200 ± 20 200 ± 38	1532 ± 219 bcde	47.94 ± 6.87 cde	9.54 ± 0.41	8.31 ± 1.39 ab	
High Mix \times Red	191 ± 35	1336 ± 176 cdef	38.90 ± 5.29 de	8.81 ± 0.30	6.29 ± 0.91 bc	
High NaCl \times Red	170 ± 31	1151 ± 181ef	$32.15 \pm 4.70e$	9.23 ± 0.25	$5.37 \pm 0.88c$	
S×C						
Control × Cut1	101 ± 9	1616 ± 174	54.12 ± 8.50	8.62 ± 0.20	$6.81 \pm 0.79 bc$	
Moderate Mix \times Cut1	126 ± 18	1683 ± 82	54.88 ± 4.35	9.14 ± 0.26	7.65 ± 0.53 b	
Moderate NaCl × Cut1	91 ± 11	1251 ± 122	41.75 ± 4.59	9.36 ± 0.33	5.77 ± 0.54bc	
High Mix \times Cut1	98 ± 9	1056 ± 57	35.44 ± 3.08	9.15 ± 0.35	5.05 ± 0.23 cc	
High NaCl × Cut1	91 ± 8	828 ± 34	28.62 ± 1.84	9.76 ± 0.28	$4.20 \pm 0.19d$	
Control × Cut2	218 ± 15	2107 ± 145	73.98 ± 9.18	8.53 ± 0.27	$10.66 \pm 1.11a$	
Moderate Mix \times Cut2	244 ± 27	2103 ± 58	72.00 ± 4.40	8.61 ± 0.32	$10.64 \pm 0.56a$	
Moderate NaCl × Cut2	227 ± 28	2064 ± 102	66.57 ± 2.51	9.09 ± 0.48	$10.38 \pm 0.79a$	
High Mix \times Cut2	208 ± 30	1599 ± 107	53.70 ± 4.58	8.64 ± 0.45	8.03 ± 0.45 ak	
High NaCl \times Cut2	192 ± 25	1288 ± 197	40.45 ± 5.19	10.02 ± 0.62	7.01 ± 0.59 bc	
	102 1 20	1200 ± 131	10.10 ± 0.10	10.02 ± 0.02	1.01 ± 0.0800	
Green × Cut1	$77\pm3d$	$1442 \pm 110b$	$53.28 \pm 3.80 \mathrm{b}$	8.97 ± 0.20 ab	$6.49 \pm 0.42b$	
Green × Cut2						
	$166 \pm 5b$ $126 \pm 7c$	1837 ± 129a 1132 ± 70c	$68.15 \pm 5.18a$	9.03 ± 0.35 ab	$9.42 \pm 0.57a$ 5 30 ± 0.37b	
Red × Cut1 Red × Cut2	120 ± 70 277 ± 13a	1132 ± 700 1870 ± 76a	$32.64 \pm 1.98c$ $55.36 \pm 2.73b$	9.45 ± 0.17a 8.89 ± 0.20b	5.30 ± 0.37b 9.47 ± 0.59a	

TABLE 2 | Analysis of variance (ANOVA) and mean comparisons for leaf number, leaf area, leaf fresh weight, leaf dry matter percentage, and dry weight of the Green and Red cultivars grown hydroponically under iso-osmotic salinity treatments and harvested in two sequential cuts.

Salinity treatments: Control = non-salt; Moderate Mix = (12.5 mM NaCl + 8.3 mM CaCl₂); Moderate NaCl = 25 mM NaCl; High Mix = (25 mM NaCl + 16.6 mM CaCl₂); High NaCl = 50 mM NaCl. * Significant effect at the 0.05 level, **0.01 level, *** 0.001 level, ns = non-significant effect. Data represent means \pm standard error of 3 replicates (n = 3) consisting of 10 plants each. Treatment means within each column followed by different letters denote significant differences (p < 0.05) according to Tukey–Kramer HSD test. Three-way interactions were either non-significant or had a very low contribution to the total variance of the examined variables.

TABLE 3 Analysis of variance (ANOVA) and mean comparisons for leaf colorimetric components L*, a*, b*, Chroma (C*), and hue Angle (h°) of the Green and Red cultivars grown hydroponically under iso-osmotic salinity treatments and harvested in two sequential cuts.

Source of variance	L*	a*	b*	C*	h° (0–360°)				
	(0–100)	(-60/+60)	(-60/+60)	$\sqrt{(a^2 + b^2)}$					
Cultivar (V)	***	***	***	***	***				
Salinity (S)	***	*	ns	ns	*				
Cut (C)	***	***	*	ns	***				
S×V	***	***	***	*	*				
S×C	***	ns	ns	ns	*				
V×C	***	***	***	***	***				
		Means ± standard error							
Cultivar									
Green	$46.37 \pm 0.19a$	$-6.17 \pm 0.11 b$	13.28 ± 0.36a	14.67 ± 0.37a	$115.43 \pm 0.28 b$				
Red	$40.68 \pm 0.58b$	$2.12 \pm 0.33a$	$3.64 \pm 0.48b$	$5.71 \pm 0.35b$	186.54 ± 5.70a				
Salinity									
Control	$42.38 \pm 1.25b$	$-1.26 \pm 1.40a$	7.35 ± 1.91	10.60 ± 1.47	148.75 ± 9.29a				
Moderate Mix	43.43 ± 1.11ab	-1.75 ± 1.18ab	8.20 ± 1.45	9.90 ± 1.38	159.87 ± 12.34a				
Moderate NaCl	$44.18 \pm 0.90a$	-2.34 ± 1.10 ab	9.03 ± 1.29	10.23 ± 1.35	156.41 ± 12.77a				
High Nix	43.95 ± 0.96a	-2.27 ± 1.09ab	9.03 ± 1.20	10.59 ± 1.10	143.11 ± 10.28a				
High NaCl	$43.68 \pm 0.68a$	$-2.47 \pm 0.83b$	8.52 ± 1.00	9.45 ± 1.07	146.91 ± 10.59a				
Cut									
Cut1	$41.84 \pm 0.71b$	$-1.59 \pm 0.78a$	$7.97 \pm 0.95 b$	$9.77 \pm 0.88a$	$159.47 \pm 7.86a$				
Cut2	45.21 ± 0.37a	$-2.46 \pm 0.63b$	$8.94 \pm 0.80a$	10.61 ± 0.71a	$142.49 \pm 5.65b$				
S × V									
Control × Green	$46.84 \pm 0.49a$	$-6.61 \pm 0.30d$	14.48 ± 1.02a	15.93 ± 1.05a	$114.95 \pm 0.72c$				
Moderate Mix × Green	$47.17 \pm 0.25a$	-6.23 ± 0.29 d	$13.48 \pm 0.96a$	$14.87 \pm 0.98a$	$115.20 \pm 0.70c$				
Moderate NaCl × Green	46.11 ± 0.25a	$-6.45 \pm 0.17d$	13.72 ± 0.63a	$15.18 \pm 0.64a$	$115.41 \pm 0.55c$				
High Mix \times Green	$46.30 \pm 0.35a$	-6.01 ± 0.17 d	$12.82 \pm 0.54a$	14.18 ± 0.55a	$115.38 \pm 0.45c$				
High NaCl × Green	$45.42 \pm 0.49a$	$-5.54 \pm 0.12d$	11.89 ± 0.67a	$13.18 \pm 0.64a$	$116.19 \pm 0.73c$				
Control × Red	$37.92 \pm 0.86c$	$4.09 \pm 0.34a$	$0.23 \pm 0.22c$	5.27 ± 0.31 b	182.56 ± 6.52ab				
Moderate Mix \times Red	39.70 ± 1.12c	$2.74 \pm 0.42 ab$	$2.91 \pm 0.33 bc$	$4.93 \pm 0.30 b$	204.54 ± 9.08 ab				
Moderate NaCl × Red	$42.26 \pm 1.52b$	1.77 ± 0.57 bc	$4.34 \pm 0.71b$	$5.29 \pm 0.64b$	$197.41 \pm 14.79a$				
High Mix \times Red	$41.58 \pm 1.50b$	1.40 ± 1.11 bc	$5.55 \pm 1.57b$	$7.32 \pm 1.38b$	170.58 ± 15.40 b				
High NaCl × Red	$41.94 \pm 0.94b$	$0.60 \pm 0.46c$	$5.14 \pm 0.74b$	$5.72 \pm 0.69b$	177.62 ± 14.50 ab				
S×C									
Control × Cut1	$41.33 \pm 2.13e$	-1.09 ± 2.11	7.50 ± 2.80	10.57 ± 2.19a	152.31 ± 14.71 abo				
Moderate Mix × Cut1	42.15 ± 2.04 de	-1.38 ± 1.85	8.03 ± 2.07	$9.77 \pm 1.94a$	$166.97 \pm 19.83a$				
Moderate NaCl × Cut1	42.04 ± 1.42 de	-1.79 ± 1.85	8.71 ± 2.27	$10.08 \pm 2.32a$	$157.18 \pm 20.44a$				
High Mix \times Cut1	$41.69 \pm 1.50e$	-1.51 ± 1.85	7.92 ± 2.26	$9.67 \pm 2.12a$	161.57 ± 17.94 ab				
High NaCl-Cut1	41.97 ± 0.97 de	-2.19 ± 1.39	7.69 ± 1.61	$8.75 \pm 1.72a$	$159.35 \pm 18.87a$				
Control × Cut2	43.42 ± 1.34 cd	-1.43 ± 1.98	7.21 ± 2.78	$10.63 \pm 2.13a$	145.2 ± 12.23 abc				
Moderate Mix × Cut2	44.71 ± 0.82 bc	-2.12 ± 1.60	8.37 ± 2.17	$10.04 \pm 2.09a$	152.77 ± 15.66 abo				
Moderate NaCl × Cut2	$46.33 \pm 0.37a$	-2.89 ± 1.29	9.35 ± 1.41	$10.39 \pm 1.56a$	$155.64 \pm 16.78ab$				
High Mix \times Cut2	46.19 ± 0.51 ab	-3.11 ± 1.26	10.45 ± 0.98	$11.83 \pm 0.86a$	$124.39 \pm 5.88c$				
High NaCl-Cut2	45.40 ± 0.45 ab	-2.74 ± 0.99	9.34 ± 1.21	$10.15 \pm 1.34a$	124.39 ± 0.000 134.47 ± 8.95 bc				
V x C	-00 ± 0.40ab	2.17 ± 0.00	J.UT 1 1.21	10.10 ± 1.04a	10-1.47 ± 0.3000				
Green × Cut1	$46.03 \pm 0.30a$	$-6.35 \pm 0.09 c$	13.73 ± 0.31a	15.14 ± 0.32a	$114.90 \pm 0.22c$				
Green × Cut2	$46.70 \pm 0.21a$	$-5.99 \pm 0.20c$	$12.82 \pm 0.64a$	$14.02 \pm 0.66a$	114.90 ± 0.220 $115.95 \pm 0.49c$				
Red × Cut1	$40.70 \pm 0.21a$ 37.64 ± 0.34c	-5.99 ± 0.200 3.17 $\pm 0.27a$	$12.82 \pm 0.04a$ $2.20 \pm 0.33c$	$4.39 \pm 0.20c$	115.95 ± 0.490 204.05 ± 6.68a				
Red × Cut2	37.04 ± 0.340 $43.72 \pm 0.54b$	$1.07 \pm 0.27a$	2.20 ± 0.330 $5.07 \pm 0.78b$	4.39 ± 0.200 $7.02 \pm 0.54b$	$204.05 \pm 0.08a$ 169.03 ± 7.51b				

Salinity treatments: Control = non-salt; Moderate Mix = (12.5 mM NaCl + 8.3 mM CaCl₂); Moderate NaCl = 25 mM NaCl; High Mix = (25 mM NaCl + 16.6 mM CaCl₂); High NaCl = 50 mM NaCl. * Significant effect at the 0.05 level, *** 0.001 level, ns = non-significant effect. Data represent means \pm standard error of 3 replicates (n = 3) consisting of 10 plants each. Treatment means within each column followed by different letters denote significant differences (p < 0.05) according to the Tukey–Kramer HSD test. Three-way interactions were either non-significant or had a very low contribution to the total variance of the examined variables.

TABLE 4 Analysis of variance (ANOVA) and mean comparisons for mineral [nitrogen (N), phosphorus (P), sulfur (S), potassium (K), calcium (Ca), magnesium (Mg),
sodium (Na), and chlorine (CI)] concentration of Green and Red cultivars cultivated under iso-osmotic salinity treatments and harvested in two sequential cuts.

Source of variance	Ν	Р	S	К	Ca	Mg	Na	CI
	(g kg ⁻¹ DW)							
Cultivar (V)	***	*	***	***	***	*	***	**
Salinity (S)	ns	**	ns	ns	***	***	***	***
Cut (C)	ns	***	ns	**	***	ns	***	ns
S×V	ns	ns	ns	ns	**	ns	ns	ns
S×C	ns	ns	ns	ns	ns	ns	***	ns
V×C	ns	***	ns	ns	ns	**	**	ns
$S \times V \times C$	ns	**	ns	ns	ns	ns	ns	ns
				Means \pm st	andard error			
Cultivar								
Green	$45.99\pm0.34a$	$10.45 \pm 0.65a$	$1.06\pm0.04b$	$53.84 \pm 1.13a$	$7.61\pm0.50b$	$2.77\pm0.15b$	$3.46\pm0.64a$	20.24 ± 2.19
Red	$43.00\pm0.43b$	$8.58\pm0.44b$	$1.61 \pm 0.07a$	$45.48\pm1.65b$	$10.66 \pm 0.77a$	$3.17\pm0.18a$	$2.17\pm0.47b$	24.22 ± 2.813
Salinity								
Control	45.09 ± 0.33	$11.49 \pm 0.39a$	1.49 ± 0.12	48.73 ± 2.17	$6.82\pm0.39\mathrm{c}$	$2.41\pm0.14b$	$0.86\pm0.13d$	$1.15 \pm 0.10c$
Moderate Mix	44.89 ± 0.75	$11.19 \pm 1.04a$	1.41 ± 0.11	47.79 ± 2.96	$10.03\pm0.76b$	$2.42\pm0.15b$	$1.15\pm0.18cd$	21.06 ± 1.108
Moderate NaCl	44.66 ± 0.80	$10.05\pm0.71\text{ab}$	1.38 ± 0.12	50.07 ± 2.25	$7.18\pm0.55c$	$3.06\pm0.21b$	$2.77\pm0.50b$	$19.29 \pm 1.20k$
High Mix	42.55 ± 0.88	$7.24\pm0.91b$	1.20 ± 0.12	48.15 ± 2.81	$14.16 \pm 1.44a$	$2.86\pm0.17b$	$2.23\pm0.61\text{bc}$	36.09 ± 2.293
High NaCl	45.38 ± 0.80	$7.44\pm0.75b$	1.15 ± 0.11	54.02 ± 2.36	$8.09\pm0.80\text{bc}$	$4.03\pm0.27a$	$7.05 \pm 1.17a$	33.60 ± 2.253
Cut								
Cut1	44.17 ± 0.47	$10.20\pm0.58a$	1.40 ± 0.08	$46.52 \pm 1.56b$	$8.24\pm0.61b$	2.86 ± 0.17	$1.93\pm0.26\text{b}$	22.44 ± 2.64
Cut2	45.10 ± 0.47	$8.79\pm0.56\mathrm{b}$	1.25 ± 0.07	$53.33 \pm 1.36a$	$9.99\pm0.75a$	3.06 ± 0.16	$3.80\pm0.76a$	21.80 ± 2.39
S × V								
Control \times Green	45.87 ± 0.40	11.66 ± 0.63	1.24 ± 0.08	52.58 ± 2.11	$6.21\pm0.46\mathrm{c}$	2.14 ± 0.17	0.98 ± 0.18	0.98 ± 0.10
Moderate Mix \times Green	46.69 ± 0.72	13.67 ± 1.52	1.22 ± 0.09	51.21 ± 3.14	$9.07\pm0.97 \text{bc}$	2.38 ± 0.22	1.30 ± 0.29	19.43 ± 0.70
Moderate NaCl \times Green	46.64 ± 0.65	11.66 ± 0.80	1.03 ± 0.06	54.44 ± 2.56	$6.42\pm0.84c$	2.77 ± 0.15	3.72 ± 0.85	17.74 ± 1.22
$\text{High Mix} \times \text{Green}$	44.26 ± 0.68	8.13 ± 1.64	0.94 ± 0.09	53.65 ± 2.30	$10.65\pm1.53b$	2.77 ± 0.33	2.70 ± 1.06	33.39 ± 2.66
$High\;NaCl\timesGreen$	46.17 ± 0.94	7.73 ± 1.20	0.89 ± 0.08	56.80 ± 2.51	$6.43\pm0.77\mathrm{c}$	3.63 ± 0.34	7.74 ± 1.67	30.19 ± 1.98
$Control\timesRed$	44.30 ± 0.28	11.31 ± 0.50	1.74 ± 0.18	44.87 ± 3.20	$7.42\pm0.57 \text{bc}$	2.67 ± 0.18	0.75 ± 0.20	1.36 ± 0.13
Moderate Mix \times Red	43.08 ± 0.79	9.12 ± 0.75	1.60 ± 0.19	44.38 ± 4.90	$10.99 \pm 1.11b$	2.46 ± 0.21	1.00 ± 0.22	22.69 ± 1.95
Moderate NaCl \times Red	42.69 ± 0.92	8.45 ± 0.74	1.72 ± 0.11	45.70 ± 2.84	$7.93\pm0.64\text{bc}$	3.35 ± 0.37	1.83 ± 0.10	20.84 ± 1.96
$\text{High Mix} \times \text{Red}$	40.42 ± 1.04	6.35 ± 0.78	1.46 ± 0.14	42.64 ± 3.88	$17.67 \pm 0.90a$	2.95 ± 0.14	1.76 ± 0.65	38.79 ± 3.58
$High\;NaCl\timesRed$	44.00 ± 1.35	7.03 ± 0.77	1.51 ± 0.12	50.13 ± 4.13	$10.4\pm0.84b$	4.60 ± 0.31	6.08 ± 1.66	38.38 ± 3.93
S × C								
Control \times Cut1	44.76 ± 0.41	11.29 ± 0.54	1.68 ± 0.21	45.94 ± 3.23	6.46 ± 0.57	2.26 ± 0.23	$0.98\pm0.21d$	1.08 ± 0.17
Moderate Mix \times Cut1	44.04 ± 0.97	12.41 ± 1.71	1.50 ± 0.13	43.66 ± 4.38	8.14 ± 0.56	2.37 ± 0.24	$0.87\pm0.15d$	18.92 ± 0.50
Moderate NaCl \times Cut1	44.75 ± 0.98	10.71 ± 1.00	1.33 ± 0.17	47.03 ± 2.65	6.65 ± 0.51	2.84 ± 0.16	$2.20\pm0.16bcd$	16.67 ± 1.24
High Mix \times Cut1	42.21 ± 1.24	8.23 ± 1.34	1.21 ± 0.15	45.38 ± 4.04	12.56 ± 1.97	2.64 ± 0.21	$1.48\pm0.57 \text{cd}$	35.67 ± 3.56
High NaCl \times Cut1	45.10 ± 1.33	8.34 ± 1.02	1.29 ± 0.17	50.59 ± 3.51	7.38 ± 1.01	4.21 ± 0.37	$4.09\pm0.43b$	36.29 ± 3.91
Control \times Cut2	45.41 ± 0.52	11.69 ± 0.60	1.30 ± 0.07	51.51 ± 2.67	7.17 ± 0.56	2.55 ± 0.17	$0.74\pm0.17d$	1.21 ± 0.12
Moderate Mix \times Cut2	45.73 ± 1.11	9.72 ± 0.77	1.32 ± 0.19	51.93 ± 3.54	11.92 ± 0.90	2.47 ± 0.19	$1.43\pm0.29\text{cd}$	23.21 ± 1.81
Moderate NaCl \times Cut2	44.58 ± 1.36	9.40 ± 1.03	1.42 ± 0.19	53.11 ± 3.40	7.70 ± 0.99	3.29 ± 0.38	$3.34\pm0.96\text{bc}$	21.91 ± 1.42
High Mix \times Cut2	43.23 ± 1.09	5.76 ± 0.67	1.19 ± 0.22	52.31 ± 2.95	16.56 ± 1.63	3.19 ± 0.21	3.35 ± 1.11 bcd	36.71 ± 2.70
High NaCl \times Cut2	45.73 ± 0.91	6.53 ± 1.05	1.01 ± 0.15	57.45 ± 2.73	8.79 ± 1.27	3.86 ± 0.40	$10.01 \pm 1.53a$	30.90 ± 2.04
V × C								
Green × Cut1	45.61 ± 0.47	$11.90 \pm 0.80a$	1.12 ± 0.07	50.38 ± 1.23	6.41 ± 0.34	$2.42\pm0.18b$	$1.96\pm0.40\mathrm{b}$	19.45 ± 3.15
Green × Cut2	46.37 ± 0.49	$8.90\pm0.88b$	1.00 ± 0.06	57.30 ± 1.44	8.82 ± 0.84	$3.11 \pm 0.20a$	4.95 ± 1.11a	21.03 ± 3.16
Red \times Cut1	42.73 ± 0.64	$8.49\pm0.58\mathrm{b}$	1.68 ± 0.09	42.66 ± 2.54	10.07 ± 0.98	$3.30\pm0.24a$	$1.89\pm0.36b$	25.65 ± 4.26
Red \times Cut2	43.38 ± 0.56	$8.67 \pm 0.70 b$	1.54 ± 0.09	48.75 ± 1.71	11.34 ± 1.22	$3.01 \pm 0.27a$	$2.48 \pm 0.93b$	22.68 ± 3.76

Salinity treatments: Control = non-salt; Moderate Mix = (12.5 mM NaCl + 8.3 mM CaCl₂); Moderate NaCl = 25 mM NaCl; High Mix = (25 mM NaCl + 16.6 mM CaCl₂); High NaCl = 50 mM NaCl. * Significant effect at the 0.05 level, ** 0.01 level, *** 0.001 level, ns = non-significant effect. Data represent means \pm SE of 3 replicates (n = 3) consisting of 10 plants each. Treatment means within each column followed by different letters denote significant differences (P < 0.05) according to the Tukey–Kramer HSD test.

Source of variance	Nitrate (µg g ⁻¹ FW)	ABTS (g Trolox eq kg ⁻¹ DM)	Phenolics (mg gallic acid 100 g ⁻¹ DW)) Ascorbic acid (mg 100 g^{-1} FW		
Cultivar (V)	ns	ns	***	***		
Salinity (S)	*	***	***	*		
Cut (C)	ns	***	***	***		
$S \times V$	ns	ns	ns	ns		
S×C	ns	ns	ns	ns		
$V \times C$	ns	ns	ns	***		
		Ν	Means \pm standard error			
Cultivar						
Green	2669 ± 136	9.03 ± 0.39	$52.60 \pm 3.30 \mathrm{b}$	$25.70 \pm 1.70 b$		
Red	2599 ± 147	12.81 ± 0.51	$97.50 \pm 2.90a$	$52.60 \pm 4.60a$		
Salinity						
Control	3311 ± 150a	$9.18 \pm 1.02b$	$61.40 \pm 7.20c$	$31.30 \pm 5.10b$		
Moderate Mix	$2891 \pm 164b$	11.51 ± 0.61a	75.80 ± 8.30ab	$44.60 \pm 9.00a$		
Moderate NaCl	$2803 \pm 174b$	11.63 ± 1.05a	$78.70 \pm 8.90 ab$	41.70 ± 8.50a		
High Mix	$2093 \pm 133c$	$11.99 \pm 0.91a$	$87.50 \pm 8.60a$	$44.80 \pm 4.20a$		
High NaCl	$1978\pm220c$	$10.23 \pm 0.71 \mathrm{ab}$	$72.00 \pm 7.00 bc$	$31.70\pm4.50\mathrm{b}$		
Cut						
Cut1	2753 ± 126	$9.72\pm0.47b$	$65.30 \pm 4.80 \mathrm{b}$	$30.00\pm2.30b$		
Cut2	2509 ± 153	$12.03 \pm 0.59a$	$84.80 \pm 4.90a$	$48.20 \pm 5.10a$		
S × V						
Control × Green	3133 ± 182	6.28 ± 0.26	39.40 ± 2.60	16.90 ± 1.80		
Moderate Mix \times Green	2841 ± 295	10.03 ± 0.70	50.40 ± 3.00	22.20 ± 2.80		
Moderate NaCl × Green	3052 ± 246	9.65 ± 0.90	52.90 ± 7.60	32.90 ± 2.70		
High Mix $ imes$ Green	2170 ± 195	10.76 ± 0.63	65.00 ± 10.90	34.70 ± 3.90		
High NaCl \times Green	2150 ± 368	8.46 ± 0.42	55.20 ± 7.10	21.80 ± 1.70		
Control × Red	3489 ± 231	12.09 ± 1.06	83.40 ± 5.30	45.70 ± 5.30		
Moderate Mix \times Red	2941 ± 174	12.99 ± 0.50	101.10 ± 6.20	67.10 ± 12.10		
Moderate NaCl \times Red	2553 ± 218	13.61 ± 1.58	104.40 ± 5.30	50.40 ± 16.70		
High Mix $ imes$ Red	2002 ± 190	13.48 ± 1.72	110.00 ± 2.00	54.90 ± 4.70		
High NaCl \times Red	1772 ± 212	12.00 ± 0.88	88.80 ± 7.20	43.50 ± 6.60		
S × C						
Control × Cut1	3347 ± 213	8.66 ± 1.44	53.90 ± 9.00	25.30 ± 5.10		
Moderate Mix \times Cut1	3105 ± 185	11.37 ± 1.19	69.80 ± 11.10	29.10 ± 6.30		
Moderate NaCl × Cut1	2850 ± 179	9.43 ± 0.70	66.00 ± 12.60	27.90 ± 3.40		
High Mix \times Cut1	2169 ± 206	10.04 ± 1.05	77.50 ± 13.30	39.90 ± 5.10		
High NaCl \times Cut1	2295 ± 319	9.14 ± 0.55	59.50 ± 8.50	27.70 ± 4.60		
Control × Cut2	3276 ± 231	9.71 ± 1.55	68.90 ± 11.10	37.30 ± 8.60		
Moderate Mix × Cut2	2677 ± 256	11.64 ± 0.43	81.70 ± 12.90	60.20 ± 14.80		
Moderate NaCl × Cut2	2755 ± 316	13.84 ± 1.56	91.30 ± 11.30	55.50 ± 15.10		
High Mix \times Cut2	2003 ± 172	13.62 ± 1.07	97.50 ± 10.40	49.60 ± 6.60		
High NaCl × Cut2	1597 ± 220	11.32 ± 1.19	84.40 ± 9.00	36.50 ± 8.40		
V×C						
Green × Cut1	2894 ± 158	8.29 ± 0.47	41.90 ± 1.90	$23.60 \pm 2.90c$		
Green × Cut2	2445 ± 211	9.78 ± 0.58	63.20 ± 5.00	27.80 ± 1.80bc		
Red × Cut1	2613 ± 196	11.24 ± 0.61	88.80 ± 3.90	36.40 ± 2.70 bc		
Red × Cut2	2583 ± 228	14.28 ± 0.61	106.30 ± 3.10	$70.00 \pm 6.40a$		

TABLE 5 | Analysis of variance (ANOVA) and mean comparisons for nitrate concentration, ABTS antioxidant activity, total concentration of phenolic compounds, and ascorbic acid concentration of Green and Red cultivars grown hydroponically under iso-osmotic salinity treatments and harvested in two sequential cuts.

Salinity treatments: Control = non-salt; Moderate Mix = (12.5 mM NaCl + 8.3 mM CaCl₂); Moderate NaCl = 25 mM NaCl; High Mix = (25 mM NaCl + 16.6 mM CaCl₂); High NaCl = 50 mM NaCl. * Significant effect at the 0.05 level, *** 0.001 level, ns = non-significant effect. Data represent means \pm standard error of 3 replicates (n = 3) consisting of 10 plants each. Treatment means within each column followed by different letters denote significant differences (p < 0.05) according to the Tukey–Kramer HSD test. Three-way interactions were either non-significant or had a very low contribution to the total variance of the examined variables.

where a tendency for reduced NO_3 concentration was apparent with increasing salinity treatments (Table 5).

Quality and Functional Attributes

The ABTS antioxidant activity (ABTS) was significantly affected by salinity and cut. The ABTS was invariably increased by all salinity treatments compared to the control, except for the High NaCl treatment which was not significantly different from the Control. The ABTS was also significantly higher at Cut2 than Cut1. The total phenolic concentration was significantly affected by salinity, cultivar, and cut. Phenolics were higher in Red by 85.3% relative to Green and in both cultivars, the phenolic concentration increased at Cut2 by 29.9%. Moreover, all salinity treatments increased the leaf phenolic concentration except for the High NaCl treatment which did not differ significantly from the Control. The total AsA concentration was marginally affected by salinity (p < 0.05) and more affected by cultivar and cut $(p \le 0.05)$. Higher ascorbate concentration was found in Red, and though it increased at Cut2 in both cultivars, this increase was also more pronounced in Red.

Volatile Aromatic Compounds Profile

A total of 84 VOCs were identified in the basil aromatic profile, accounting for 16 monoterpene hydrocarbons, 10 oxygenated monoterpenes, 18 sesquiterpene hydrocarbons, 11 aliphatic alcohols, 8 aliphatic aldehydes, 11 aliphatic esters, 2 aliphatic ketones, and 8 aromatic compounds (Supplementary Table 1). Among these, seven major VOCs were identified in the two cultivars, which contributed individually $\geq 2\%$ and collectively near 90% to the total VOCs relative abundance (Table 6). These included two monoterpene hydrocarbons: β-Myrcene (woody, vegetative, citrus fruity, and minty nuance) and (Z)- β -Ocimene (sweet herbal); two oxygenated monoterpenes: eucalyptol (fresh mint, spicy, cooling) and β-Linalool (floral, woody, sweet, spicy, tropical); one aliphatic aldehyde: (E)-2-hexenal (green, fruity, fresh, herbal, citrus); one aliphatic alcohol: 1-octen-3-ol (mushroom, earthy, fungal, green, oily, vegetative, umami sensation, and savory-brothy); and one sesquiterpene hydrocarbon: trans-a-bergamotene (citrusy, fresh, earthy green odor).

The relative abundance of β -Myrcene was affected only by a cultivar, being higher in red basil. Eucalyptol was only affected by the cut sequence, and it was subject to significant cultivar \times cut interaction as in Green, it increased at Cut2 whereas in Red it did not change significantly. Salinity did not have a significant effect on (E)-2-hexenal as opposed to cultivar and cut, being more abundant in Red rather than Green, and in Cut1 compared to Cut2. The relative abundance of (Z)- β -ocimene was subject to significant interactions involving the three main factors of the model examined. (Z)- β -ocimene was more abundant in Green irrespective of salinity treatment and in this cultivar, it nearly doubled from Cut1 to Cut2, as opposed to Red wherein no difference was observed between cuts. Salinity had no effect on the relative abundance of this component in Red, however, in Green, it increased significantly in the High NaCl treatment compared to the Control and Moderate Mix treatments. The relative abundance of 1-octen-3-ol was higher in Green than Red; however, this difference was apparent only in response to the High Mix and High NaCl treatments. β -linalool was relatively more abundant in Red in which also increased at Cut2, whereas in Green it did not change in response to cut order. In both cultivars, the presence of β -linalool decreased in the High Mix and High NaCl treatments. *Trans*- α -bergamotene incurred a significant effect from the cultivar as it was overall more abundant in Green than Red; however, its relative abundance in Green increased significantly from Cut1 to Cut2, whereas in Red, no change was observed.

Principal Component Analysis

A PCA was performed on all basil-analyzed data in relation to the three experimental factors: cultivars, iso-osmotic salinity concentrations, and sequential harvests. The loading plot and scores are reported in Figures 1, 2. PC1 was positively correlated with leaf FW, color components L* and C*, N, P, K, and beta-Linalool while it was negatively correlated with S, Ca, ABTS, phenolics, AsA concentrations but also with the relative percentage of beta-Ocimene, (Z)-, 2-Hexenal, (E)-, and trans-alfa-Bergamotene (Figure 1). Moreover, PC2 was positively correlated with leaf dry matter, K, Mg, Na, and Cl concentration while it was negatively correlated with P, S, and NO3 FW. The two basil cultivars were well separated but not uniformly clustered with respect to PC1 and PC2 (Figure 2). The green basil treatments from cuts 1 and 2 were distributed on the positive side of PC1 between upper and lower right quadrants, while the red basil cuts 1 and 2 treatments were distributed on the negative side of PC1, in the upper and lower left quadrants. The successive red basil cuts grown under the same salinity level were detected in the same quadrant, apart from the red basil cuts grown under high mix salinity. Similar behavior was observed for the two cuts of green basil except for green basil cuts grown under the moderate mix and moderate NaCl salinity treatments. Furthermore, the control treatments were in the negative quadrant of PC2 irrespective of cultivar type or cut. Quadrant 1 of the score plot included treatments that were characterized by a higher concentration of Na, K, or N while the treatments in quadrant 2 had the highest concentration of Cl, Mg, and phenolics. The lower negative side of PC1 (quadrant 3) included treatment combinations that produced plants characterized by high S and AsA concentration, increased antioxidant activity but also by a higher relative abundance (%) of the major aroma volatile components like *trans*-α-bergamotene, (Z)-β-ocimene, and 1-Octen-3-ol. Treatments in the lower right quadrant (quadrant 4) produced leaves with the highest FW but they were also characterized by a higher concentration of P and nitrates.

DISCUSSION

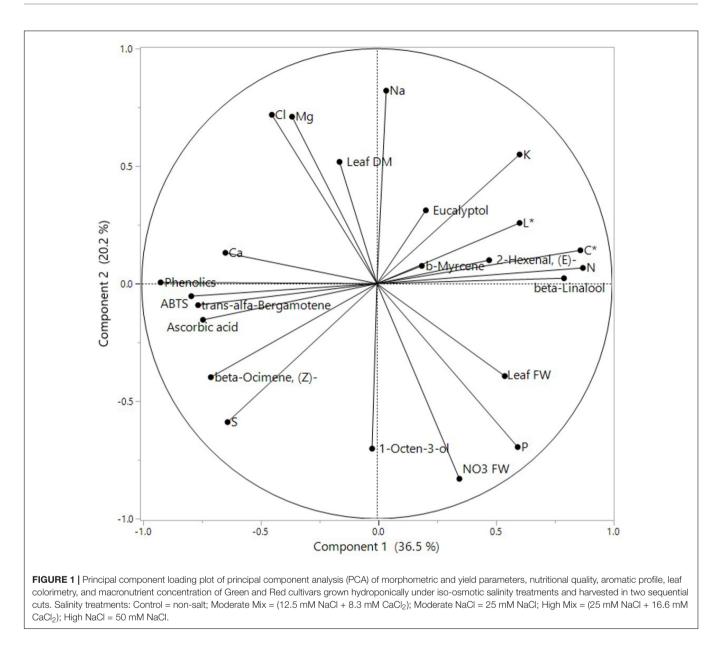
Salinity Affects Morphometric and Yields Parameters in a Cultivar-Dependent Manner

In the present work, the Red and Green basil cultivars showed different responses to moderate and high salt stress induced

TABLE 6 | Analysis of variance (ANOVA) and mean comparisons for the relative abundance (%) of major aroma volatile components [β-myrcene, eucalyptol, (E)-2-hexenal, (Z)-β-ocimene, 1-octen-3-ol, β-linalool, and *trans*-α-bergamotene] in Green and Red cultivars grown hydroponically under iso-osmotic salinity treatments and harvested in two sequential cuts.

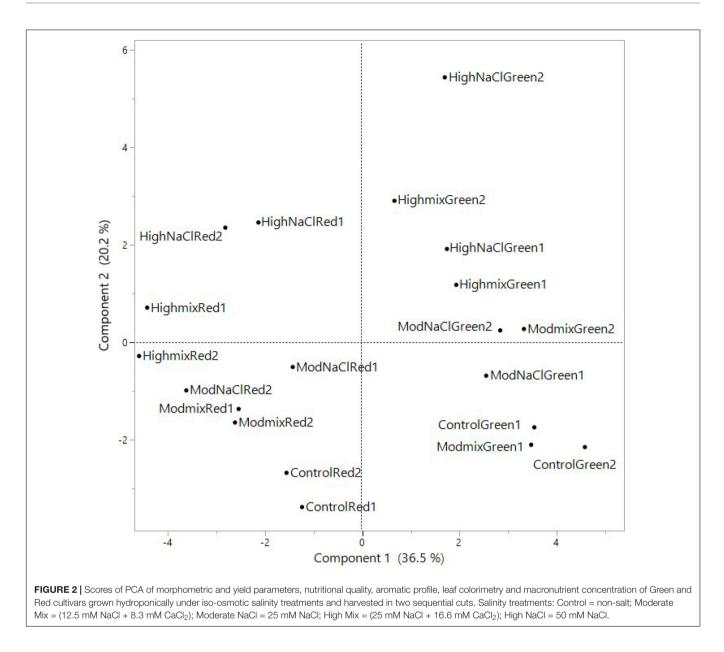
Source of variance	β-myrcene	Eucalyptol	(E)-2-hexenal	(Z)-β-ocimene	I-Octen-3-ol	β-linalool	Trans-α-bergamotene
Cultivar (V)	*	ns	*	***	**	***	***
Salinity (S)	ns	ns	ns	*	***	***	ns
Cut (C)	ns	*	**	***	ns	ns	ns
$S \times V$	ns	*	ns	*	*	ns	ns
S×C	**	ns	ns	*	ns	*	ns
$V \times C$	***	***	ns	***	ns	***	***
S × V × C	*	**	ns	**	ns	ns	*
			lean percentage	fraction of total	volatile area \pm sta	indard error	
Cultivar							
Green	$4.86\pm0.15b$	31.52 ± 1.02	$2.17\pm0.12b$	$4.80\pm0.47a$	$5.51 \pm 0.13a$	$19.5\pm0.45b$	$10.17 \pm 0.54a$
Red	$5.40\pm0.28a$	32.91 ± 0.73	$2.67 \pm 0.16a$	$1.48\pm0.05b$	$5.02\pm0.24b$	$24.3\pm0.47a$	$6.45\pm0.43b$
Salinity							
Control	4.76 ± 0.26	31.16 ± 1.63	2.26 ± 0.17	$3.56\pm0.69a$	$6.01 \pm 0.24a$	$23.17 \pm 0.99a$	8.23 ± 1.25
Moderate Mix	5.14 ± 0.29	32.84 ± 0.84	2.57 ± 0.27	$3.55\pm0.63a$	$5.42\pm0.19\text{ab}$	$22.84 \pm 0.76a$	7.02 ± 0.60
Moderate NaCl	4.83 ± 0.36	31.92 ± 1.54	2.31 ± 0.19	$3.13\pm0.70\text{ab}$	$5.86\pm0.20a$	$22.09\pm0.93\text{ab}$	8.23 ± 0.77
High Mix	5.33 ± 0.34	31.67 ± 1.17	2.41 ± 0.33	$3.37\pm0.95\text{ab}$	$4.28\pm0.31c$	$20.17\pm1.00b$	9.95 ± 1.15
High NaCl	5.60 ± 0.50	33.42 ± 1.87	2.50 ± 0.19	$2.32\pm0.74b$	$4.66\pm0.28\text{bc}$	$20.58\pm1.22b$	8.59 ± 0.80
Cut							
Cut1	5.22 ± 0.27	$33.46 \pm 0.91 a$	$2.67 \pm 0.13a$	$2.48\pm0.40b$	5.39 ± 0.18	21.69 ± 0.53	8.12 ± 0.50
Cut2	5.02 ± 0.17	$30.92\pm0.83b$	$2.14\pm0.15b$	$3.91\pm0.49a$	5.15 ± 0.21	21.94 ± 0.75	8.63 ± 0.69
S × V							
Control × Green	4.77 ± 0.28	27.70 ± 2.07	2.14 ± 0.12	$5.78\pm0.29a$	$5.69\pm0.33\text{ab}$	20.83 ± 1.26	11.24 ± 1.65
Moderate Mix \times Green	4.69 ± 0.24	32.96 ± 1.58	2.11 ± 0.28	$5.56\pm0.36a$	$5.63\pm0.14\text{ab}$	21.06 ± 0.56	8.46 ± 0.46
Moderate NaCl \times Green	4.28 ± 0.42	32.38 ± 2.70	2.21 ± 0.30	$4.79 \pm 1.03 \mathrm{ab}$	$6.03\pm0.30\text{ab}$	20.03 ± 0.94	8.86 ± 1.26
$High\;Mix\timesGreen$	5.30 ± 0.30	30.31 ± 1.57	1.81 ± 0.36	$4.80\pm1.55\text{ab}$	$4.95\pm0.30\text{bc}$	17.64 ± 0.70	12.58 ± 0.88
$High\;NaCl\timesGreen$	5.24 ± 0.33	34.26 ± 2.85	2.56 ± 0.18	$3.07 \pm 1.32 \text{bc}$	$5.25\pm0.26 \text{abc}$	17.96 ± 0.71	9.71 ± 0.93
$Control\timesRed$	4.75 ± 0.47	34.63 ± 1.62	2.37 ± 0.34	$1.33\pm0.08c$	$6.34\pm0.30a$	25.52 ± 0.70	5.22 ± 0.75
Moderate Mix-Red	5.59 ± 0.48	32.72 ± 0.75	3.03 ± 0.41	$1.55\pm0.07\mathrm{c}$	$5.21\pm0.36 \mathrm{abc}$	24.63 ± 0.96	5.58 ± 0.74
Moderate NaCl \times Red	5.37 ± 0.51	31.46 ± 1.73	2.41 ± 0.25	$1.48\pm0.10c$	$5.70\pm0.28 \text{ab}$	24.14 ± 1.13	7.60 ± 0.91
$High\:Mix\timesRed$	5.36 ± 0.71	33.31 ± 1.59	3.13 ± 0.41	$1.65\pm0.18c$	$3.46\pm0.32d$	23.20 ± 0.77	6.80 ± 1.24
$High\;NaCl\timesRed$	6.02 ± 1.06	32.41 ± 2.56	2.42 ± 0.39	$1.42\pm0.09c$	3.95 ± 0.33 cd	23.73 ± 1.71	7.25 ± 1.18
S × C							
Control × Cut1	$4.05\pm0.21b$	34.09 ± 2.16	2.09 ± 0.25	$3.41 \pm 1.03a$	5.98 ± 0.23	$24.82 \pm 0.97a$	7.19 ± 1.00
Moderate Mix \times Cut1	$5.44\pm0.46 \mathrm{ab}$	32.98 ± 1.36	2.81 ± 0.26	$3.65\pm0.90a$	5.63 ± 0.34	22.47 ± 0.92	6.96 ± 0.78
Moderate NaCl \times Cut1	$5.08\pm0.64 \text{ab}$	33.07 ± 2.37	2.56 ± 0.23	$2.38\pm0.75\text{ab}$	5.78 ± 0.33	$21.32\pm0.87\text{bc}$	8.21 ± 0.78
High Mix \times Cut1	5.64 ± 0.45 ab	32.48 ± 1.25	3.01 ± 0.38	$1.97\pm0.89 \mathrm{ab}$	4.42 ± 0.52	$20.40\pm0.77\text{bc}$	9.44 ± 1.87
High NaCl \times Cut1	$6.02\pm1.00a$	34.91 ± 3.49	2.94 ± 0.23	$0.72\pm0.37b$	5.11 ± 0.30	$19.01\pm1.00\mathrm{c}$	8.92 ± 0.65
$Control \times Cut2$	$5.48\pm0.23 \text{ab}$	28.24 ± 1.91	2.42 ± 0.25	$3.70\pm1.00a$	6.04 ± 0.44	$21.52\pm1.49 \text{abc}$	9.27 ± 2.34
Moderate Mix \times Cut2	$4.84\pm0.34\text{ab}$	32.69 ± 1.11	2.33 ± 0.49	$3.45\pm0.96a$	5.21 ± 0.17	$23.21\pm1.27 ab$	7.08 ± 0.99
Moderate NaCl \times Cut2	$4.57\pm0.35\text{ab}$	30.77 ± 2.07	2.07 ± 0.29	$3.89 \pm 1.17a$	5.94 ± 0.26	$22.86\pm1.69 \text{ab}$	8.25 ± 1.40
$\text{High Mix} \times \text{Cut2}$	$4.96\pm0.52\text{ab}$	30.70 ± 2.17	1.69 ± 0.38	$5.05 \pm 1.59a$	4.10 ± 0.34	$19.89\pm2.14\text{bc}$	10.57 ± 1.35
High NaCl \times Cut2	$5.24\pm0.43 \text{ab}$	32.18 ± 2.01	2.13 ± 0.19	$3.66\pm1.06a$	4.28 ± 0.41	$21.89\pm2.02abc$	8.32 ± 1.42
V × C							
Green × Cut1	$4.45\pm0.15\mathrm{c}$	$35.18 \pm 1.22a$	2.50 ± 0.11	$3.29\pm0.72b$	5.79 ± 0.16	$20.51\pm0.68c$	$8.89\pm0.74\text{bc}$
Green \times Cut2	$5.27\pm0.22 \text{ab}$	$27.87\pm0.94c$	1.84 ± 0.17	$6.31\pm0.25a$	5.22 ± 0.19	$18.50\pm0.48c$	$11.46 \pm 0.66a$
$\text{Red} \times \text{Cut1}$	$6.04\pm0.44\text{ab}$	$31.61 \pm 1.22b$	2.86 ± 0.24	$1.62\pm0.08c$	4.96 ± 0.31	$22.96\pm0.68b$	$7.03\pm0.64\text{bc}$
$\text{Red} \times \text{Cut2}$	4.76 ± 0.24 bc	$34.20\pm0.68 \mathrm{ab}$	2.47 ± 0.21	$1.35 \pm 0.03c$	5.07 ± 0.38	$25.63 \pm 0.44a$	$5.61\pm0.52c$

Salinity treatments: Control = non-salt; Moderate Mix = (12.5 mM NaCl + 8.3 mM CaCl₂); Moderate NaCl = 25 mM NaCl; High Mix = (25 mM NaCl + 16.6 mM CaCl₂); High NaCl = 50 mM NaCl. * Significant effect at the 0.05 level, ** 0.01 level, *** 0.001 level, ns = non-significant effect. Data represent means \pm SE of 3 replicates (n = 3) consisting of 10 plants each. Treatment means within each column followed by different letters denote significant differences (p < 0.05) according to the Tukey–Kramer HSD test.



by NaCl and the mixture of NaCl and CaCl2 at isomolar concentrations. Relative to the weight and leaf area, Red appeared more tolerant to salinity than Green (Table 2). FW and leaf area in Green were significantly reduced by Moderate NaCl, High Mix, and High NaCl treatments, compared to the Control. On the contrary, Red showed no difference between treatments for these parameters, similar to the findings of Cruz et al. (2020) on red basil cultivars (Red basil, Dark Opal, and Basilico Rosso) exposed to high salinity levels. Bione et al. (2014) reported a linear decrease in the leaf FW of basil in response to increasing salinity levels and a strong genotypic effect on the adaptive capacity to the aforementioned abiotic stress. The reason why green cultivars showed a distinct response to saline treatments could be found in the high genetic variation of the Ocimum genus (Akbari et al., 2018). As previous studies suggested (Parida and Das, 2005; Bernstein et al., 2010), the reduction in fresh leaf weight observed in Green is derived primarily from the inhibitory effect of salt stress on leaf expansion. In fact, the decrease in Green fresh leaf weight under Moderate NaCl, High Mix, and High NaCl treatments was also associated with a reduction in leaf area.

Several authors (Attia et al., 2011; Ahmadi and Souri, 2018) suggest that the reduction in leaf weight is due to the inhibition of leaf initiation; however, in the present work, no significant changes were observed in the number of leaves per plant, corroborating that the reduction in leaf FW was due solely to the reduction in leaf area (**Table 2**). It should be noted that the more sensitive the Green cultivar, when not exposed to salt stress (Control), was characterized by the highest values for leaf area, fresh leaf weight, and DW, compared to the more tolerant Red cultivar. Nonetheless, at low salinity levels, the gap between the two cultivars narrowed, with Green linearly reducing DW,



while Red did not follow the same trend. Specifically, compared to the Control, Red increased the DW with the Moderate Mix treatment. Although not significant, this increase could be due to the mitigating effect of calcium, which, as reported in the literature, could help plants preserve the integrity and functionality of membranes and cell walls by regulating the transport and ion exchange under NaCl salinity stress (Rengel, 1992; Borgognone et al., 2014a).

Chaparzadeh et al. (2004) suggested that high salinity may prompt growth reduction as an adaptive response to osmotic stress, which allows conservation of energy deployed to mitigate the risk of permanent damage. Such adaptations seem related to heat dissipation and transpiration adjustments mobilized under environmental stress conditions (Bridge et al., 2013). In light of the above, the unchanged leaf DM percentage contributes further evidence of Red's higher tolerance to salinity. As argued by Bernstein et al. (2010), the ability of basil to maintain adequate tissue hydration is key to its salt stress tolerance. In fact, in the present work, Red did not significantly vary in leaf FW under saline treatments compared to the Control and, at the same time, it did not vary in leaf DM percentage, confirming the above statement. In the Red cultivar, the leaf water was unchanged, which indicates that Na and Cl ions were likely compartmentalized in the vacuoles (Tarchoune et al., 2013). By contrast, the Green cultivar probably also accumulated Na and Cl ions in the apoplast, leading to water loss through dehydration that affected the FW of the leaves (Attia et al., 2011). This suggests the likely occurrence of toxicity or ion imbalance in Green, which limited plant growth. Regarding the effect of cut sequence, except for leaf DM percentage, the two cultivars showed similar behavior in both cuts (Table 2). As argued by several authors, the increase in

Basil Response to NaCl Stress

DW observed at the second cut could have resulted from greater root system growth and increased branching induced by the suppression of apical dominance (Zheljazkov et al., 2008; Nicoletto et al., 2013). Furthermore, the increase in cytokinins production in response to cut may have increased the number of leaf primordia in both cultivars and stimulated cell division leading to greater leaf area and fresh leaf weight (Wang et al., 2014).

Leaf Colorimetry and Macronutrient Uptake Under Salt Stress: The Ameliorative Role of Calcium

Visual quality is a key driver for consumer choice of horticultural products, with coloration playing a determining role, especially for aromatic herbs (León et al., 2006). Color in plants is affected by various factors ranging from genetics to pre- and postharvest factors (Pathare et al., 2013). In the current study, Chroma (C*) and Hue Angle (h°) in both basil cultivars were affected by the interaction between the investigated factors (Table 3). In agreement with the findings of Scagel et al. (2019), salt treatments did not change the colorimetric parameters in Green. In contrast, the leaf colorimetry of Red changed as salinity levels increased, corroborating analogous observations by Cruz et al. (2020). Red lost its distinctive pigmentation, showing a decrease in redness (a*), an increase in lightness (L*), and in yellowness (b*) in the Moderate NaCl, High Mix, and High NaCl treatments, compared to the Control (Table 3). Interestingly, plants in the Moderate Mix treatment did not vary in the above parameters compared to the Control, probably due to the mitigating effect of calcium and the low sodium levels (Table 4). However, this result was not observed in the High Mix treatment plants, where the sodium concentration was approximately 2.5 times higher than the Control (Table 4). Although the overall color intensity of Red increased at Cut2, this was probably due to the increase in yellowness (b*) rather than redness (a^{*}), which decreased at Cut2. On the contrary, the Green cultivar's colorimetry did not respond to cut sequence, which contrasts with the results reported by Ciriello et al. (2021b), possibly reflecting differences in the genetic materials used, the cultivation systems applied, and in growing seasons. The present results highlight how red-colored genotypes, despite being overall more tolerant to salinity and sequential cut stress than green cultivars, are also more prone to leaf color downgrading.

Apart from salinity and cut, the mineral profile of basil in the current study showed strong cultivar dependence (**Table 4**). The higher concentration of macroelements N, P, and K in Green, crucial to maintaining plant biological functions, may justify its higher yield compared to Red, which was characterized by a higher concentration of Ca, Mg, and less Na (**Tables 2**, **4**; Karthika et al., 2018). The lower accumulation of Na in the leaves of Red probably relates to a desalination mechanism that transports this cation to the hypogeal apparatus (Attia et al., 2009). This likely mechanism activated by Red could underlie its greater tolerance to salt stress compared to Green

(Table 2). Relative to the effect of cut, in agreement with Nicoletto et al. (2013) and Carillo et al. (2020), the second cut increased calcium and potassium concentration (Table 4). These elements play an essential role in promoting human health, such as stabilizing blood pressure, detoxification, and improving bone structure (Gharibzahedi and Jafari, 2017). Although an increase in FW was observed from the first to the second cut (Table 2), this was not associated with an increase in the leaf concentration of P and N. This could be attributed, as hypothesized for basil by Nicoletto et al. (2013), to the variation in environmental parameters during the two successive harvests. Although the detrimental impact of NaCl salinity on K uptake in different vegetables is known in the literature (Borgognone et al., 2014a), our results do not show significant differences in the concentration of this cation in leaves. As suggested by Scagel et al. (2019), this inconsistency can be explained by the fact that leaf nutrient concentration was determined in our study and not total plant uptake. Considering the reduction in leaf area, leaf FW, and DW in High Mix and High NaCl treatments (Table 2), it can be assumed that salinity reduced K uptake without altering leaf concentration. This hypothesis is confirmed by several studies on basil in which NaCl salinity, in addition to reducing fresh production, increased K concentration in leaves (Zuccarini and Okurowska, 2008; Attia et al., 2009; Scagel et al., 2017).

The addition of CaCl₂ in the Moderate Mix and High Mix solutions evidently increased Ca and decreased the Na concentration compared to the equivalent isomolar Moderate NaCl and High NaCl solutions (Table 4). What was subsequently observed was not surprising since calcium is known to mitigate NaCl salinity by reducing the efficiency of leaf sodium uptake via the reduction in the permeability of the plasmalemma to this toxic cation and the increase in the K/Na ratio that enhances tolerance to salinity (Nedjimi and Daoud, 2009; Colla et al., 2013b). However, the beneficial effect of calcium in NaCl saline treatments was only observed in the moderate salinity treatment (Moderate Mix) where leaf area, leaf FW, and DW were unchanged compared to the non-salinized Control. In the high salinity treatment (High Mix), a strong reduction was observed in the above parameters as calcium was unable to mitigate the detrimental effects of salinity, probably due to the exceedingly high concentration of Cl that reached toxic levels even for Cl-tolerant plants (Scagel et al., 2017). Not least, the reduction in FW recorded in the High Mix versus the High NaCl treatments could be attributable not only to an increase in Cl but also to a decrease in P (Table 4). This element is a limiting factor for plant development as its deficiency leads to a reduction in photosynthesis and the ATP/ADP ratio (Formisano et al., 2021b).

Moderate Salinity Improves Nutritional Quality Without Affecting the Aromatic Profile of Basil

In our study, leaf nitrate concentration was influenced only by saline treatments (**Table 5**). Nitrate decreased significantly in plants treated with moderate and high salt levels (Moderate

Mix, Moderate NaCl, High Mix, and High NaCl) compared to the Control, with the lowest values obtained in highsalinity treatments. Salinity depresses the rate of nitrate transport through xylem vessels, leading to reduced growth (Borgognone et al., 2014a), as evidenced in our results (Table 1). The reduction in nitrate was due to the increase in Cl concentration, which showed the opposite trend (Table 4). In fact, chlorine, due to its recognized antagonism to nitrate, results in a downregulation of its net uptake (Carillo et al., 2020). Several studies have shown that plant tolerance to high salinity levels is a function of the ability to implement active defense mechanisms such as the production of antioxidant compounds (Rouphael and Kyriacou, 2018; Carillo et al., 2020; Farsaraei et al., 2020). Indeed, in our study, the most salinity-tolerant Red cultivar had higher concentrations of AsA and total phenols (Table 5). The higher concentration of total phenols in the leaves of Red can be attributed to the constitutive presence of anthocyanins, belonging to the flavonoid class, which confer the characteristic red pigmentation. Regardless of the cultivar and cut, salinity significantly altered the polyphenol concentration (Table 5). Consistent with several studies (Petropoulos et al., 2017), moderate salinity levels (Moderate Mix and Moderate NaCl) increased polyphenols without affecting yield, compared to the Control (Table 2), achieving the desired compromise between yield and increase in beneficial secondary metabolites. The opposite trend was observed with high salinity levels. In particular, the High NaCl treatment reduced yield without increasing phenolic concentration and ABTS antioxidant activity, thus demonstrating the strong dependence of the synthesis of antioxidant compounds on the magnitude of salt stress (Waškiewicz et al., 2013). In contrast, the application of calcium chloride in the high salinity treatment (High Mix), although it reduced yield compared to the Control, resulted in a significant increase in total phenols. Indeed, Ca ion is a crucial ubiquitous messenger involved in plant biosignal transduction (Anil and Sankara Rao, 2001) that can influence physiological activities such as growth and development but especially resistance and adaptation to stressors, partly through the biosynthesis of phenols, known metabolites involved in stress response (Ma et al., 2019). The role of phenols as "stress mitigants" is confirmed by their increase after the cut (Table 5), as observed by several authors (Nicoletto et al., 2013; Ciriello et al., 2021b) in basil. The increase in total phenols could be attributable to the upregulation of the Phenylalanineammonia-lyase (PAL) enzyme in response to the mechanical stress induced by the cut (Ciriello et al., 2021b). Additionally, the improved production performance at the second cut (Table 2) may have resulted from increased photosynthate allocation to the shikimate metabolic pathway (Shaw et al., 1998; Crozier et al., 2007). In our study, the second cut increased ABTS antioxidant activity, contrary to the findings of Corrado et al. (2020a), while it promoted vitamin C biosynthesis exclusively in the Red cultivar (Table 5). This observation highlights how genetic material may be implicated in plant response to cut stress.

The gas chromatography/mass spectrometry (GC/MS) analysis showed that the aroma profile of Red and Green

basil was predominantly characterized by eucalyptol (Table 6), in contrast to several studies in which β-linalool was the most abundant compound (Marotti et al., 1996; Ciriello et al., 2021a,b). This variation in the volatile composition may derive from the interaction of pre-harvest factors such as cultivar, environmental conditions, and cultural practices (Farsaraei et al., 2020). Similarly, abiotic stresses, such as salinity, can alter the volatilome in a manner highly dependent on the duration and magnitude of stress and the cultivar-specific tolerance (Niinemets, 2010). In our work, high salinity levels (High Mix and High NaCl) reduced β-linalool, probably due to the depression of carbon assimilation required to produce VOCs, as inferred from the reduced DW (Table 2). As is known in the literature, mutable conditions after the cut had a greater influence on the quality of the aroma profile (Table 6; Zheljazkov et al., 2008; Corrado et al., 2020a; Ciriello et al., 2021a,b). Specifically, the two basil cultivars responded differentially to the successive cuts, with an increase in β -myrcene and β -ocimene and a reduction in eucalyptol observed in Green, while an increase in β-linalool was observed in Red. Different growth conditions and mechanical stress signaling may have altered the metabolic pathways involving mono-oxygenated terpenes that share the same precursor (i.e., geranyl pyrophosphate; GPP). However, the singular increase in β-linalool found in Red, despite being in line with Ciriello et al. (2021b), confirms once more that gene expression of critical enzymes in the biosynthesis of oxygenated monoterpenes is strongly controlled by the interaction of genotype and stress in a way that is not yet completely clear (Chang et al., 2007).

CONCLUSION

In the next decades, groundwater and soil salinization will pose a major challenge for growers in the Mediterranean coastal region. Understanding salinity thresholds could help growers select targeted crops capable of optimizing water and soil resources use to achieve high yields of premium quality vegetables. However, moderate salt stress can trigger adaptive mechanisms in plants leading to the production of the desired secondary metabolites. Our study showed that moderate salinity levels (Moderate Mix and Moderate NaCl) and high salinity levels in combination with calcium chloride (High Mix) improved the nutritional quality of the two differentially pigmented hydroponic basil genotypes by boosting phenol concentration and lowering nitrate concentration, without affecting the eucalyptol. The response of productive traits to salinity was not uniform among cultivars. Despite being more productive overall, the Green cultivar showed a linear decrease in yield as the salinity level increased. Although the Red cultivar did not show yield reduction in response to salinity, induced stress resulted in a drastic alteration of the typical pigmentation of the leaves that turned green. However, the addition of calcium to the moderate salt solution (Moderate Mix) reduced the detrimental effects mentioned above for both Red and Green cultivars, underlining its mitigating role against salt stress.

DATA AVAILABILITY STATEMENT

The raw data supporting the conclusions of this article will be made available by the authors, without undue reservation.

AUTHOR CONTRIBUTIONS

YR: conceptualization, project administration, resources, visualization, and supervision. MC, LF, MK, and YR: methodology, validation, formal analysis, investigation, and writing – original draft preparation. GS and AK: software and data curation. MC, LF, GS, AK, SD, MK, and YR: writing – review and editing. SD and YR: funding acquisition.

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SUPPLEMENTARY MATERIAL

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