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# $\text{NH}_3/\text{NH}_4^+$ allosterically activates SLC4A11 by causing an acidic shift in the intracellular pK that governs $\text{H}^+(\text{OH}^-)$ conductance

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SLC4A11 is the most abundant membrane transport protein in corneal endothelial cells. Its functional presence is necessary to support the endothelial fluid pump that draws fluid from the corneal stroma, preventing corneal edema. Several molecular actions have been proposed for SLC4A11 including  $\text{H}_2\text{O}$  transport and cell adhesion. One of the most reproduced actions that SLC4A11 mediates is a  $\text{H}^+$  (or  $\text{OH}^-$ ) conductance that is enhanced in the presence of  $\text{NH}_4\text{Cl}$ . The mechanism by which this occurs is controversial with some providing evidence in favor of  $\text{NH}_3\text{-H}^+$  cotransport and others providing evidence for uncoupled  $\text{H}^+$  transport that is indirectly stimulated by the effects of  $\text{NH}_4\text{Cl}$  upon intracellular pH and membrane potential. In the present study we provide new evidence and revisit previous studies, to support a model in which  $\text{NH}_4\text{Cl}$  causes direct allosteric activation of SLC4A11 by means of an acidic shift in the intracellular pK ( $\text{pK}_i$ ) that governs the relationship between intracellular pH ( $\text{pH}_i$ ) and SLC4A11  $\text{H}^+$ -conductance. These findings have important implications for the assignment of a physiological role for SLC4A11.

## KEYWORDS

acid-base, Btr1, NaBC1, cornea, proton

## 1 Introduction

The members of the SLC4 family of solute carrier proteins are mainly  $\text{Na}^+$ -independent and  $\text{Na}^+$ -dependent  $\text{Cl}^-/\text{HCO}_3^-$  exchangers or  $\text{Na}^+$ -coupled  $\text{HCO}_3^-$  (or  $\text{CO}_3^{2-}$ ) transporters (Parker and Boron, 2013; Lee et al., 2022). SLC4A11 (originally named “BTR1”: Bicarbonate Transporter Related Protein 1) was the last member of the SLC4 family to be cloned (Parker et al., 2001) and is the only member of the family that does not transport  $\text{HCO}_3^-/\text{CO}_3^{2-}$  (Jalimarada et al., 2013; Ogando et al., 2013; Loganathan et al., 2016). Instead, SLC4A11 influences intracellular pH ( $\text{pH}_i$ ) by conducting  $\text{H}^+$ , or its thermodynamic equivalent  $\text{OH}^-$  (Kao et al., 2015; Myers et al., 2016). For convenience hereafter, and in the absence of any definitive data in favor of one substrate over the other, we will assume that  $\text{H}^+$  are the transported species. SLC4A11 is expressed at low levels in many tissues, but appears to have most functional impact in the cornea and inner ear as evidenced by the corneal dystrophies and progressive hearing loss that are caused by SLC4A11 mutations in both humans and mice (Vithana et al., 2006; 2008; Desir et al., 2007; Lopez et al., 2009; Gröger et al., 2010; Han et al., 2013).

Most studies of SLC4A11 have focused on its role in the cornea. Here, SLC4A11 is expressed in the basolateral membrane of the corneal endothelial cells that line the posterior (aqueous-humor facing side) of the cornea (Vilas et al., 2013). These cells are responsible for pumping fluid from the corneal stroma to prevent it from swelling and losing its ability to optimally transmit and refract light (Hodson, 1977). SLC4A11 is clearly valuable for endothelial pumping because autosomal recessive inheritance of SLC4A11 mutations cause congenital hereditary endothelial dystrophy (CHED), a non-progressive corneal thickening and opacification (Vithana et al., 2006). Furthermore, an apparently discrete set of SLC4A11 mutations result in autosomal dominant inheritance of a late-onset form of Fuchs endothelial corneal dystrophy (FECD4), which typically manifests from the 4th decade of life, preceded by the appearance of excrescences from the Descemet's membrane that underlie the endothelial cells, which are known as guttae (Weiss et al., 2024). However, the mechanism by which SLC4A11 loss compromises the endothelial pump remains elusive because there is little consensus about SLC4A11's molecular action.

The earliest proposal for SLC4A11 action was that it, like plantal Slc4-like proteins, performed borate transport: specifically electrogenic  $2\text{Na}^+\text{-B(OH)}_4^-$  cotransport (Frommer and von Wirén, 2002; Park et al., 2004). This hypothesis fell out of favor due to an inability of others to find evidence for this mode of action with mammalian SLC4A11 (Jalimarada et al., 2013; Ogando et al., 2013; Vilas et al., 2013; Kao et al., 2015; Loganathan et al., 2016). That original characterization study also attributed an EIPA-insensitive  $\text{Na}^+/\text{H}^+$  exchanger-like activity to SLC4A11, but neither the EIPA-sensitivity nor the  $\text{Na}^+$ -dependence of this  $\text{H}^+$  transport activity has proven to be universally repeatable, leaving open the possibility that such results were influenced by endogenous activities (Park et al., 2004; Ogando et al., 2013; Kao et al., 2015; 2016; Zhang et al., 2015; Myers et al., 2016). Other reported SLC4A11 transport modes include a  $\text{H}_2\text{O}$  permeability that is disturbed by disease-causing mutations (Vilas et al., 2013), a role in extracellular matrix adhesion (Malhotra et al., 2020), and a conditional presence in mitochondria (Ogando et al., 2019). Any of these, alone or in combination, remain a feasible explanation for loss of endothelial pump function with SLC4A11 mutation. However, at present, none of these features has been investigated by more than one group.

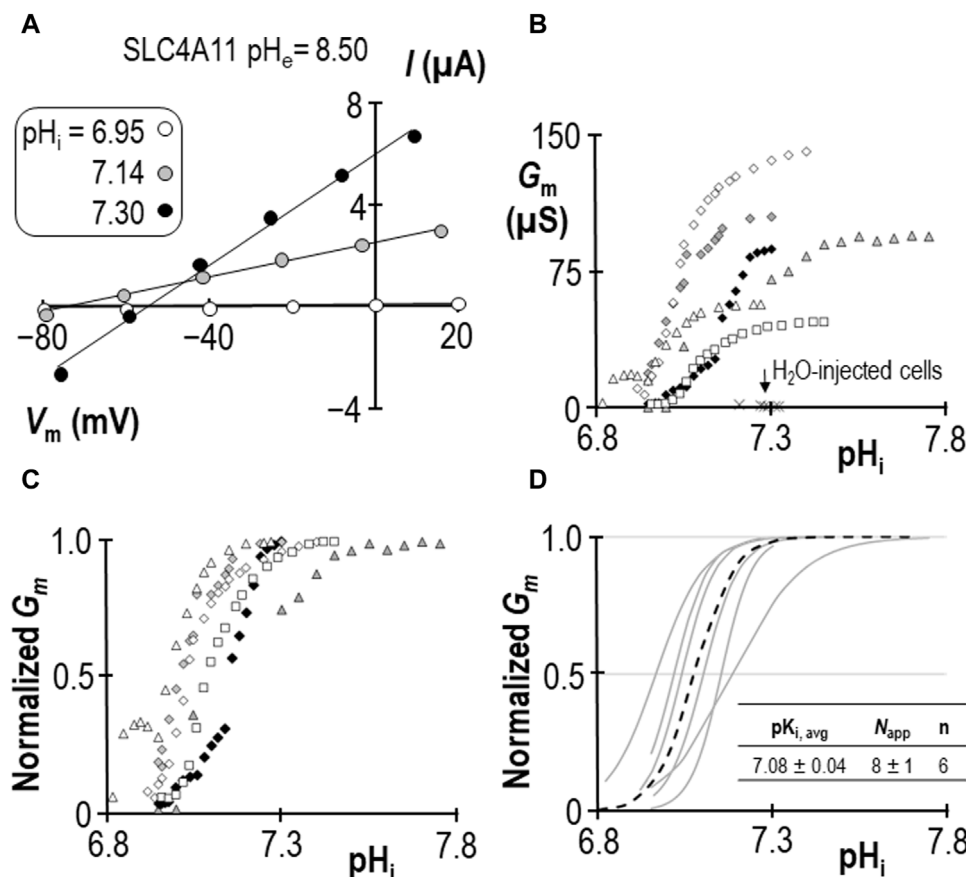
In this study, we focus on the robust and repeatable  $\text{Na}^+$ -independent  $\text{H}^+$  transport action of SLC4A11, an action that is stimulated by increases in  $\text{pH}_i$ , extracellular pH ( $\text{pH}_e$ ), and  $\text{NH}_4\text{Cl}$  (Zhang et al., 2015; Myers et al., 2016; Kao et al., 2020; Quade et al., 2020) and which is also disrupted by disease-causing mutations (Kao et al., 2015; Quade et al., 2022). However, there remains a controversy over whether the  $\text{NH}_4\text{Cl}$ -stimulated action of SLC4A11 represents SLC4A11-mediated  $\text{NH}_3\text{-H}^+$  cotransport (Zhang et al., 2015; Kao et al., 2020) or an indirect stimulation of SLC4A11-mediated  $\text{H}^+$  conductance caused by cellular depolarization and sub-membranous alkalization of  $\text{pH}_i$  due to  $\text{NH}_3$  movement across the lipid bilayer (Myers et al., 2016). A recent attempt to distinguish  $\text{NH}_3\text{-H}^+$  cotransport from  $\text{NH}_4^+$

or  $\text{H}^+$  transport using the Goldman-Hodgkin-Katz approach led to the conclusion that SLC4A11 operates in competing modes of  $\text{NH}_3\text{-H}^+$  cotransport or unaccompanied  $\text{H}^+$  transport, with the inference that the presence of  $\text{NH}_3/\text{NH}_4^+$  inhibits unaccompanied  $\text{H}^+$  conduction (Kao et al., 2020). These are important distinctions as they inform the predicted direction of transport, the interpretation of SLC4A11 structure, and the ultimately the physiological role of SLC4A11. In our previous work, we demonstrated that SLC4A11-mediated  $\text{H}^+$  transport is governed by an intracellular  $\text{pK}_i$ , the value of which can be modulated by changes in  $\text{pH}_e$  and by disease-causing mutations (Quade et al., 2020; Quade et al., 2022). At  $\text{pH}_e = 7.50$ ,  $\text{pK}_i$  is too alkaline to be determined, but can be no more acidic than 7.6 (Quade et al., 2020). However, when we raise  $\text{pH}_e$ ,  $\text{pK}_i$  is shifted into measurable range. For example,  $\text{pK}_i$  for human SLC4A11  $\sim 7.04$  when  $\text{pH}_e = 8.50$  (Quade et al., 2022). This acidic-shift in  $\text{pK}_i$  manifests as a rise in SLC4A11 current. We note that no study of SLC4A11 activity has directly measured transmembrane  $\text{NH}_3/\text{NH}_4^+$  movement, relying instead on proxies such as pH and voltage changes. With that as context, here we revisit the phenomenon of  $\text{NH}_4\text{Cl}$ -stimulation of SLC4A11-mediated  $\text{H}^+$  currents to determine whether it could be explained by a direct allosteric effect of  $\text{NH}_3/\text{NH}_4^+$  upon SLC4A11  $\text{pK}_i$ .

## 2 Results

### 2.1 Determining $\text{pK}_i$ for SLC4A11 at $\text{pH}_e = 8.50$

In a previous study we determined that  $\text{pK}_i$  for human SLC4A11 at  $\text{pH}_e = 8.50$  is  $7.04 \pm 0.01$  (Quade et al., 2022). For this study we generated a contemporary set of control data to confirm the  $\text{pK}_i$  of human SLC4A11 at  $\text{pH}_e = 8.50$  in new experimental hands. As we have previously reported, SLC4A11-expressing oocytes slowly alkalize upon exposure to  $\text{pH}_e = 8.50$  solution, and the rate of alkalization can be enhanced by clamping the membrane potential ( $V_m$ ) at a value more positive than the predicted reversal potential for  $\text{H}^+$  ( $E_{\text{H}}$ ). An example of this phenomenon can be seen in the first figure of Quade et al. (2022). As  $\text{pH}_i$  rises, we gather a series of I-V plots such that each I-V plot can be assigned to a value of  $\text{pH}_i$ . Figure 1A shows a selection of these plots gathered from a single SLC4A11-expressing oocyte as  $\text{pH}_i$  is caused to rise under voltage-clamp. Note that the slope of the I-V relationship (i.e., membrane conductance,  $G_m$ ) rises as  $\text{pH}_i$  increases. The relationship between  $\text{pH}_i$  and  $G_m$  is shown for six SLC4A11-expressing cells in Figure 1B, in which the trace marked by black diamonds is the full data set from the cell shown in Figure 1A. The average  $G_{m,\text{max}}$  was  $90 \pm 14 \mu\text{S}$ . Also shown in Figure 1B are resting  $\text{pH}_i/G_m$  relationships from six  $\text{H}_2\text{O}$ -injected cells (crosses) for which the average  $G_m$  was  $2 \pm 1 \mu\text{S}$ . In order to extract  $\text{pK}_i$  values from the SLC4A11 data, we first normalize  $G_m$  data from each cell to its own  $G_{m,\text{max}}$  (Figure 1C) and fit those data to the Hill equation, generating the best-fit relationships described by a  $\text{pK}_i$  and an apparent Hill coefficient ( $N_{\text{app}}$ ) shown as gray lines in Figure 1D. The average of these relationships is represented as a black dotted-line in Figure 1. We calculate an average  $\text{pK}_i$  of  $7.08 \pm 0.04$  for human SLC4A11 at  $\text{pH}_e = 8.50$ , which is not different from the range



**FIGURE 1** SLC4A11 behavior in the absence of  $NH_4Cl$  (extracellular  $pH = 8.50$ ) (A) A representative selection of current-voltage ( $I-V$ ) relationships gathered from a single SLC4A11-expressing oocyte as intracellular  $pH$  ( $pH_i$ ) rises. (B) The relationship between  $pH_i$  and slope conductance ( $G_m$ ) from the full-set of  $I-V$  relationships gathered from the SLC4A11-expressing cell shown in panel (A) (black diamonds) and from five other SLC4A11-expressing cells (each represented by its own symbol). The  $pH_i$  versus  $G_m$  relationship at resting  $pH_i$  for five  $H_2O$ -injected oocytes is represented by crosses. (C) SLC4A11 data from panel (B), normalized to its respective maximum  $G_m$  ( $G_{m,max}$ ). (D) Best-fit lines for each cell to the Hill equation are shown in gray. The dashed black line represents the Hill equation generated using the average  $pK_i$  and apparent Hill coefficients ( $N_{app}$ ) of the  $n = 6$  replicates, as shown in the inset table.

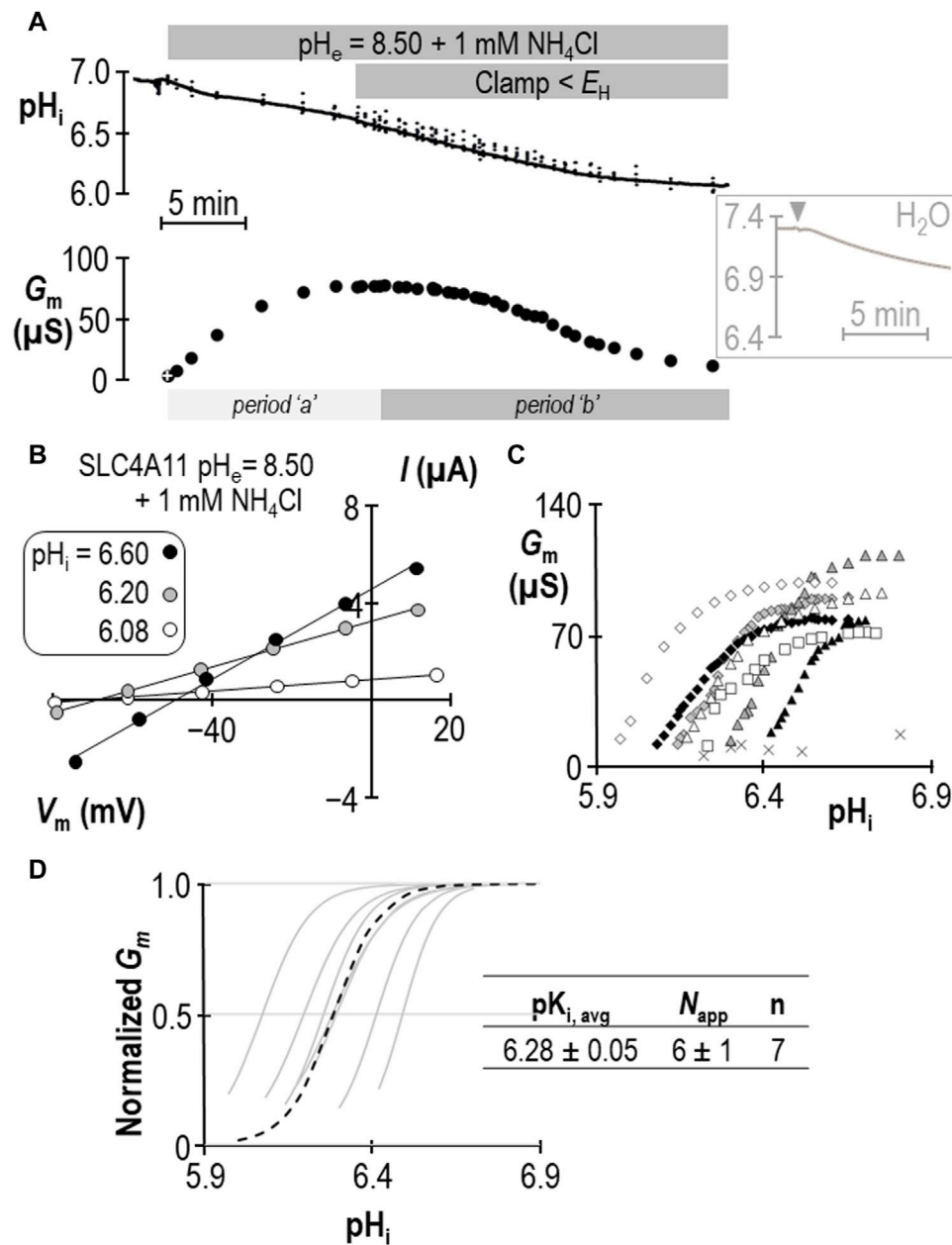
that we had previously determined in [Quade et al. \(2022\)](#) ( $p = 0.32$ , two-tailed, unpaired t-test).

## 2.2 Determining $pK_i$ for SLC4A11 at $pH_e = 8.50$ in the presence of 1 mM $NH_4Cl$

The response of SLC4A11-expressing oocytes in  $pH_e = 8.50$  solution was altered in two ways in the presence of 1 mM  $NH_4Cl$ . First, as shown in the example in [Figure 2A](#), the cell acidified rather than alkalized (initial  $dpH_i/dt = 5.1 \pm 0.9 \times 10^{-4}$  pH units/s,  $n = 6$ ). This was also a feature of  $H_2O$ -injected cells (initial  $dpH_i/dt = 5.5 \pm 0.8 \times 10^{-4}$  pH units/s,  $n = 6$ , [Figure 2A](#) inset). Second, unique to SLC4A11-expressing cells,  $G_m$  unexpectedly rose to a plateau (over period “a”) and subsequently declined to a value close to its starting value (over period “b”) when  $pH_i$  was further acidified by clamping  $V_m$  at a value more negative than  $E_H$ . Let us first focus our attention on period “b,” during which acidification causes a familiar decline. [Figure 2B](#) shows a selection of responses from the cell represented in [Figure 2A](#). The  $pH_i$  versus  $G_m$  relationships gathered from seven

SLC4A11-expressing cells during period “b” are plotted in [Figure 2C](#). The average  $G_{m,max}$  was  $89 \pm 5 \mu S$ , which is not different from that for the group of cells assayed in the absence of  $NH_4Cl$  ( $p = 0.94$ , two-tailed unpaired t-test). Also shown in [Figure 2C](#) are resting  $pH_i/G_m$  relationships from six control cells (originally injected with  $H_2O$  in place of SLC4A11 cRNA) that were acidified prior to assay by HCl injection (crosses). The average  $G_m$  of these cells was  $11 \pm 2 \mu S$ . Best-fit normalized  $G_m$  versus  $pH_i$  data for SLC4A11-expressing cells is shown in [Figure 2D](#). We calculate that the average  $pK_i$  for SLC4A11-expressing cells at  $pH_e = 8.50$  in the presence of 1 mM  $NH_4Cl$  is  $6.28 \pm 0.05$ , which is significantly more acidic than the  $pK_i$  range determined in the absence of  $NH_4Cl$  ( $p < 0.001$ , one-tailed, unpaired t-test). On the other hand, there was no significant difference in the value of  $N_{app}$  compared to its value in the absence of  $NH_4Cl$  ( $P = 0.17$ , two-tailed unpaired t-test).

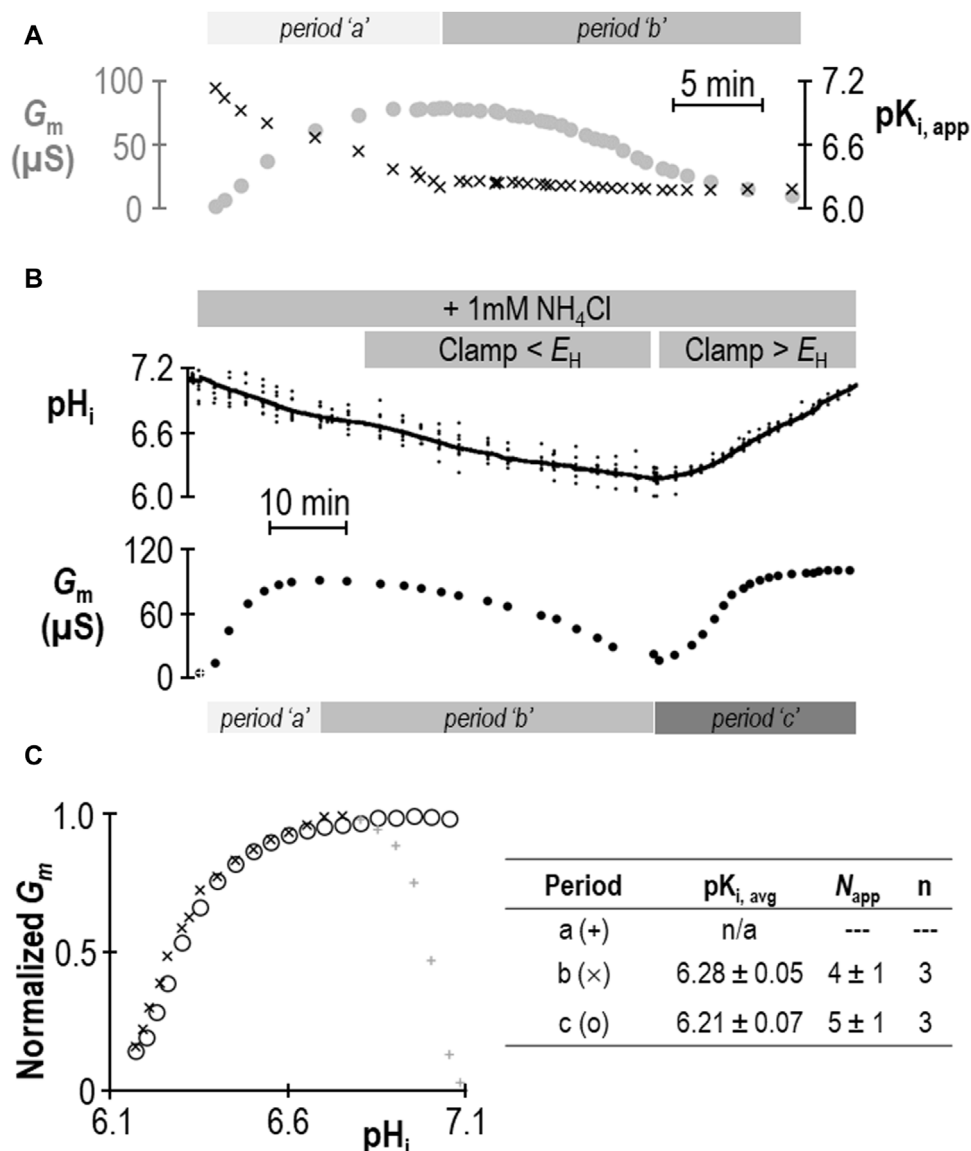
Does the action of SLC4A11 during period “a” represent a second  $pK_i$  that reports acid-activation? We hypothesized that period “a” represented the time during which  $pK_i$  was transitioning between its  $\pm NH_4Cl$  values. If we assume that neither  $G_{m,max}$  nor  $N_{app}$  change in a single SLC4A11-expressing oocyte during the course of



**FIGURE 2** SLC4A11 behavior in the presence of 1 mM  $\text{NH}_4\text{Cl}$  (extracellular  $pH = 8.50$ ). **(A)** A representative example of the  $pH_i$  (top) and  $G_m$  (bottom) response of an SLC4A11-expressing oocyte to the addition of 1 mM  $\text{NH}_4\text{Cl}$ . The first  $G_m$  point in the data series (white cross) was gathered prior to the presence of  $\text{NH}_4\text{Cl}$ . The inset shows a representative  $pH_i$  response to the addition of 1 mM  $\text{NH}_4\text{Cl}$  (point of solution change indicated by gray triangle) of a  $\text{H}_2\text{O}$ -injected cell at  $pH_e = 8.50$ . **(B)** A representative selection of I-V relationships gathered from a single SLC4A11-expressing oocyte as  $pH_i$  falls during period “b”. **(C)** The relationship between  $pH_i$  and  $G_m$  from the full-set of I-V relationships gathered from the SLC4A11-expressing cell shown in panel (B) (black diamonds) and from six other SLC4A11-expressing cells (each represented by its own symbol). The  $pH_i$  versus  $G_m$  relationship for six  $\text{H}_2\text{O}$ -injected oocytes after acidification by HCl injection is represented by crosses. **(D)** Best-fit lines for each cell to the Hill equation are shown in gray. The dashed black line represents the Hill equation generated using the average  $pK_i$  and  $N_{\text{app}}$  of the  $n = 7$  replicates, as shown in the inset table.

an experiment such as that in **Figure 2A**, we can solve the Hill equation to generate an apparent  $pK_i$  ( $pK_{i, \text{app}}$ ) for each point in the experiment at which we have paired values of  $G_m$  and  $pH_i$ . The results of this approach are shown in **Figure 3A** and support the hypothesis that period “a” is dynamic time of  $pK_i$  adjustment, while period “b” represents a time over which SLC4A11 has assumed a new and relatively stable  $pK_i$ . In three of our seven experiments, we extended the protocol to examine a period “c” (example shown in **Figure 3B**) during which we could examine the

behavior of SLC4A11 in the  $pH$ -range of period “a”, but after SLC4A11 has assumed its +  $\text{NH}_4\text{Cl}$   $pK_i$ . As shown in **Figure 3C**, SLC4A11 exhibits a similar  $pK_i$  during period “b” versus period “c”  $pK_i$  ( $P = 0.19$ : two-tailed, paired t-test) and does not revisit the apparent acid-activated behavior exhibited during period “a.” In summary for this section, we find that the presence of 1 mM  $\text{NH}_4\text{Cl}$  results in a significant acidic shift in the value of  $pK_i$  for human SLC4A11 at  $pH_e = 8.50$ .



**FIGURE 3** Validation of change in SLC4A11  $pK_i$  upon addition of  $NH_4Cl$ . (A)  $G_m$  data from Figure 2A replotted with the apparent  $pK_i$  ( $pK_{i,app}$ ) calculated from the  $pH_i$  at each value of  $G_m$ . (B) A representative experiment similar to that shown in Figure 2A, extended into a third experimental period in which  $G_m$  is monitored during a return to starting  $pH_i$ . (C)  $G_m$  values from panel (B) plotted for each of the three experimental periods, normalized to  $G_{m,max}$  for each period. The inset table shows  $pK_i$  and  $N_{app}$  for periods “b” and “c” calculated from best-fit data to the Hill equation for three such experiments.

### 2.3 Determining the ion-selectivity of SLC4A11 in the presence of 1 mM $NH_4Cl$

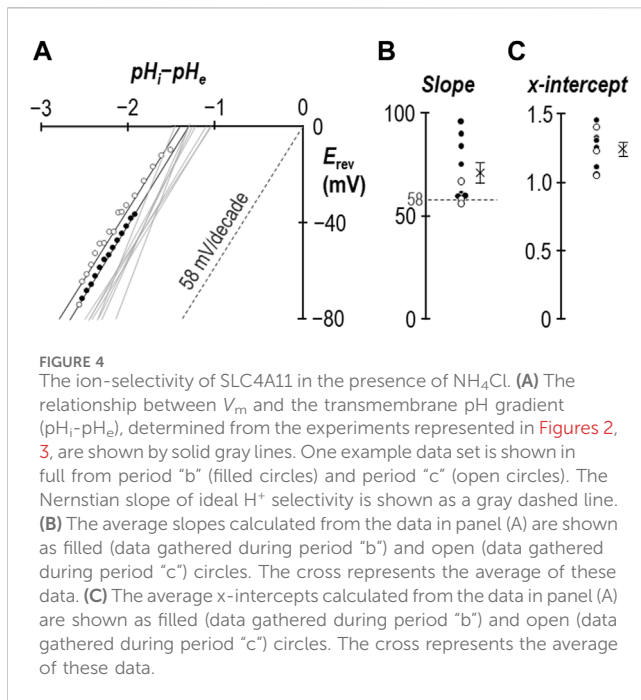
In an earlier study we demonstrated that for SLC4A11-expressing oocytes, the relationship between  $V_m$  and the transmembrane pH gradient (64 mV/pH-unit) was close to Nernstian with respect to  $H^+$  (Myers et al., 2016). Figure 4A shows equivalent data gathered in the presence of 1 mM  $NH_4Cl$  from experiments such as those shown in Figures 2A, 3B. The average slope of the relationship is  $71 \pm 5$  mV/decade (Figure 4B) with an  $x$ -axis intercept of  $-1.25 \pm 0.04$  (Figure 4C). Data from one of the three Figure 3B-style experiments is highlighted in Figure 4A, with black data points taken from period “b” and white data points taken from period “c.” In summary for this section, the slope of the

relationship between  $V_m$  and transmembrane pH gradient does not appear to be greatly disturbed by the presence of 1 mM  $NH_4Cl$ , except for the unusual observation that the relationship does not intersect with the origin. The meaning of this observation is explored in Section 3.3.

### 2.4 Comparing the influence of $pH_e$ and $[NH_3]$ on SLC4A11 $pK_i$

Using a similar work-flow to that described above, we determined the  $pK_i$  of SLC4A11 at  $pH_e = 7.50$  in the presence of 1 mM  $NH_4Cl$  (Figures 5A–C).  $G_{m,max}$  in this cohort of cells was  $104 \pm 7$   $\mu S$ , which is not different from the equivalent range reported





from cells assayed at  $\text{pH}_e = 8.50 + \text{NH}_4\text{Cl}$  ( $p = 0.09$ : two-tailed, unpaired t-test). Because the ratio of  $\text{NH}_3:\text{NH}_4^+$  is pH-sensitive, we also determined the  $\text{pK}_i$  of SLC4A11 at  $\text{pH}_e = 8.50$  in the presence of 0.12 mM  $\text{NH}_4\text{Cl}$  (**Figures 6A–C**). In both conditions, although  $[\text{NH}_4\text{Cl}]$  is different,  $[\text{NH}_3]$  is the same (0.017 mM) **Figure 6D** summarizes the values of  $\text{pK}_i$  that have been determined during this study. We find that  $\text{pK}_i$  at  $\text{pH}_e = 7.50$  in the presence of 1 mM  $\text{NH}_4\text{Cl}$  is  $7.06 \pm 0.05$  (**Figure 5C**), and the  $\text{pK}_i$  at  $\text{pH}_e = 8.50$  in the presence of 0.12 mM  $\text{NH}_4\text{Cl}$  is  $7.04 \pm 0.05$  (**Figure 6C**). These values are not significant different from each other ( $p = 0.68$ : two-tailed, unpaired t-test).

We can make two additional statistical comparisons: [1] at  $\text{pH}_e = 8.50$ ,  $\text{pK}_i$  is not significantly different in the presence or absence of 0.12 mM  $\text{NH}_4\text{Cl}$  ( $P = 0.50$ : two-tailed, unpaired t-test), and [2] in the presence of 1 mM  $\text{NH}_4\text{Cl}$ , raising  $\text{pH}_e$  from 7.50 to 8.50 has a significant acidifying effect on  $\text{pK}_i$  ( $p < 0.01$ : two-tailed, unpaired t-test). The interpretation of these findings are discussed in **Section 3.4**.

### 3 Discussion

#### 3.1 $\text{NH}_4\text{Cl}$ increases SLC4A11 $G_m$ by shifting $\text{pK}_i$ in the acidic direction

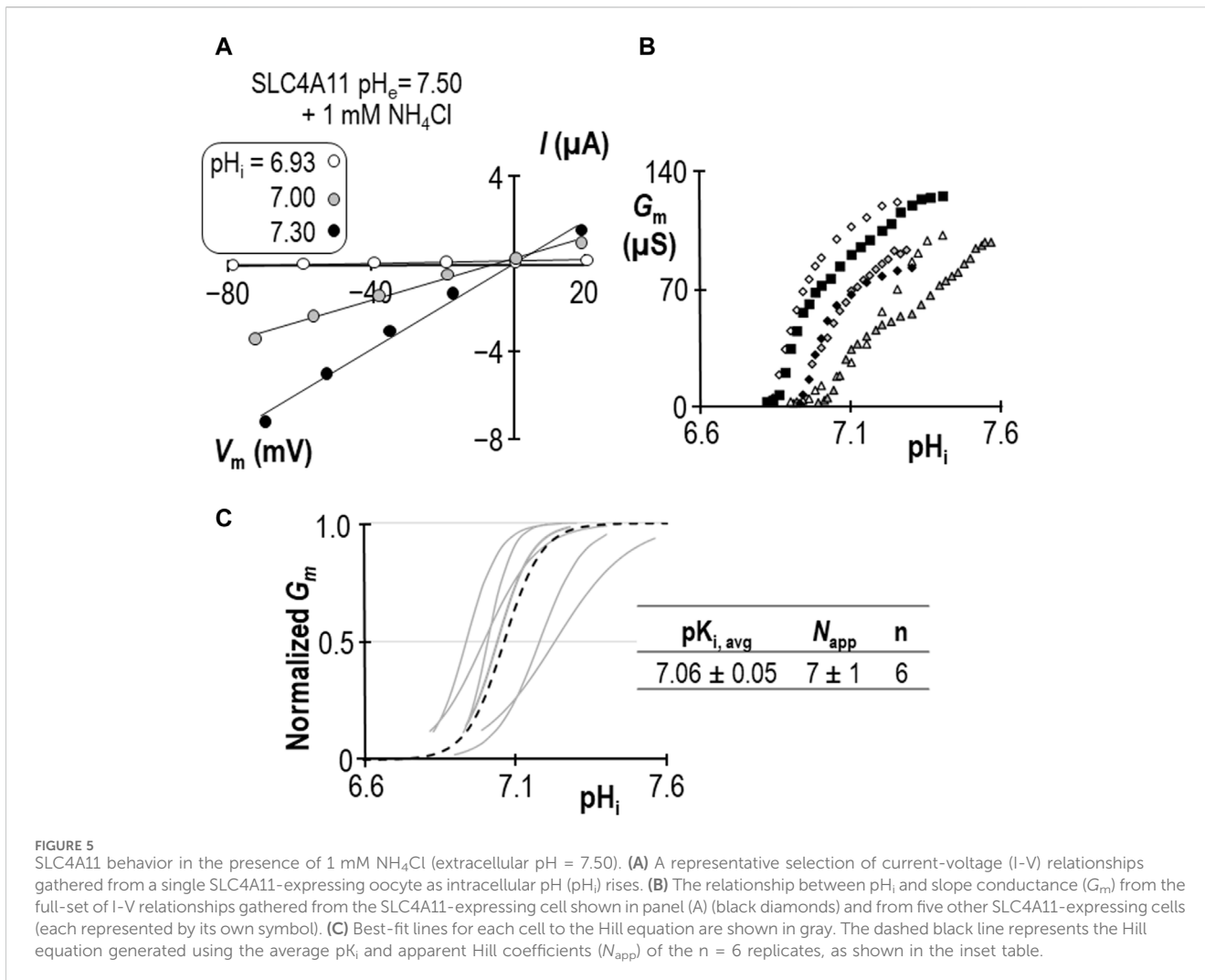
Our data show that, at  $\text{pH}_e = 8.50$ , the extracellular presence of 1 mM  $\text{NH}_4\text{Cl}$  results in a significant acidic shift in  $\text{pK}_i$  such that, at almost any value of  $\text{pH}_i$  between 6.0 and 7.3 (i.e., the range bounded by the two  $\pm \text{NH}_4\text{Cl}$  traces: solid and dashed black-lines in **Figure 7A**),  $G_m$  would be increased. A formal determination of  $\text{pK}_i$  at physiological  $\text{pH}_e$  in the absence of  $\text{NH}_4\text{Cl}$  has been precluded by a practical limitation on how high we can raise oocyte  $\text{pH}_i$ , but we have previously estimated that the value must be more alkaline than 7.6 (**Quade et al., 2020**). The ability of  $\text{NH}_4\text{Cl}$  to cause an acidic shift

in  $\text{pK}_i$ , enables us to determine a  $\text{pK}_i$  range of  $7.06 \pm 0.05$  at  $\text{pH}_e = 7.50$  in the present study. Thus, we can imagine that, even with the conservative estimate of the  $\text{NH}_4\text{Cl}$ -free  $\text{pK}_i$ , the implication is similar: the presence of  $\text{NH}_4\text{Cl}$  would cause an increase in  $G_m$  at typical physiological values of  $\text{pH}_i$  (e.g., 7.0–7.3, which is included in the range bounded by the solid and dashed gray-lines in **Figure 7A**). Critically, this stimulatory effect of  $\text{NH}_4\text{Cl}$  represents only an increase in  $G_m$  caused by a redefinition of the  $\text{pH}_i$  versus  $G_m$  relationship. Neither  $G_{m,\text{max}}$  nor  $N_{\text{app}}$  are significantly altered by  $\text{NH}_4\text{Cl}$  (at least as determined at  $\text{pH}_e = 8.50$ ) so the redefinition appears to represent a direct acidic-translation of the relationship. Although we are wary of assigning any meaning to the numerical value of  $N_{\text{app}}$  (which is a function of the number of titratable moieties within SLC4A11), the observation of an unchanging  $N_{\text{app}}$  at least implies that mechanism by which SLC4A11 responds to  $\text{pH}_i$  is similarly complex in the absence and presence of  $\text{NH}_4\text{Cl}$ .

We had once before investigated the role of  $\text{NH}_4\text{Cl}$  in stimulating SLC4A11 (**Myers et al., 2016**). We found, as others had before us (**Zhang et al., 2015**), that the presence of 5 mM  $\text{NH}_4\text{Cl}$  causes an increase in SLC4A11  $G_m$  (or  $I_m$  at fixed  $V_m$  in the case of those other studies). However, although we appreciated at that time that SLC4A11 was  $\text{pH}_i$ -dependent, we did not consider the possibility that the relationship between  $\text{pH}_i$  and  $G_m$  could be modulated. The addition of 5 mM  $\text{NH}_4\text{Cl}$  to (even  $\text{H}_2\text{O}$ -injected) oocytes causes a rapid and robust depolarization (**Musa-Aziz et al., 2009**). This action in itself is sufficient to drive SLC4A11-mediated  $\text{H}^+$  efflux, resulting in a cellular alkalization that further activates SLC4A11 (**Myers et al., 2016**). Thus, in that study, when we increased  $G_m$  to  $G_{m,\text{max}}$  at  $\text{pH}_e = 8.50$  by adding 5 mM  $\text{NH}_4\text{Cl}$  and saw no change in that value upon  $\text{NH}_4\text{Cl}$  removal (maintaining a voltage clamp at 0 mV to mimic the depolarizing effect of  $\text{NH}_4\text{Cl}$  presence), we assumed that the presence of  $\text{NH}_4\text{Cl}$  was merely causing  $G_m$  to rise according to the prescribed  $\text{pH}_i$  vs.  $G_m$  relationship. In light of our new data, we reinterpret those data as likely to have been gathered at a  $\text{pH}_i$  value that was greater than the  $\text{NH}_4\text{Cl}$ -free  $\text{pK}_i$ , where  $G_m$  values for both  $\pm \text{NH}_4\text{Cl}$  relationships are close to  $G_{m,\text{max}}$ . That is to say that we reinterpret our data in favor of a model in which  $\text{NH}_4\text{Cl}$  causes a direct rather than indirect allosteric activation of SLC4A11.

#### 3.2 Data do not conclusively support a more important role for $\text{NH}_3$ versus $\text{NH}_4^+$ for SLC4A11 action

Models in which SLC4A11 transports  $\text{NH}_3/\text{NH}_4^+$  favor a  $\text{NH}_3:n\text{H}^+$  cotransport mechanism over  $\text{NH}_4^+$  transport because  $\text{NH}_3$  increases in concentration with rising pH, thereby providing an explanation for the greater stimulation of SLC4A11 currents/conductance at  $\text{pH}_e = 8.50$  than  $\text{pH}_e = 7.50$ . We might use the same logic to conclude that  $\text{NH}_3$  is more likely than  $\text{NH}_4^+$  to be the allosterically activating species. However, our data provide an alternative explanation: as shown in **Figures 7A,B**, at the typical resting  $\text{pH}_i$  range for SLC4A11-expressing oocytes [6.9–7.0: (**Quade et al., 2020**)], the shift in  $\text{pK}_i$  makes less difference to  $G_m$  at  $\text{pH}_e =$



**FIGURE 5** SLC4A11 behavior in the presence of 1 mM  $NH_4Cl$  (extracellular  $pH = 7.50$ ). **(A)** A representative selection of current-voltage ( $I$ - $V$ ) relationships gathered from a single SLC4A11-expressing oocyte as intracellular  $pH$  ( $pH_i$ ) rises. **(B)** The relationship between  $pH_i$  and slope conductance ( $G_m$ ) from the full-set of  $I$ - $V$  relationships gathered from the SLC4A11-expressing cell shown in panel (A) (black diamonds) and from five other SLC4A11-expressing cells (each represented by its own symbol). **(C)** Best-fit lines for each cell to the Hill equation are shown in gray. The dashed black line represents the Hill equation generated using the average  $pK_i$  and apparent Hill coefficients ( $N_{app}$ ) of the  $n = 6$  replicates, as shown in the inset table.

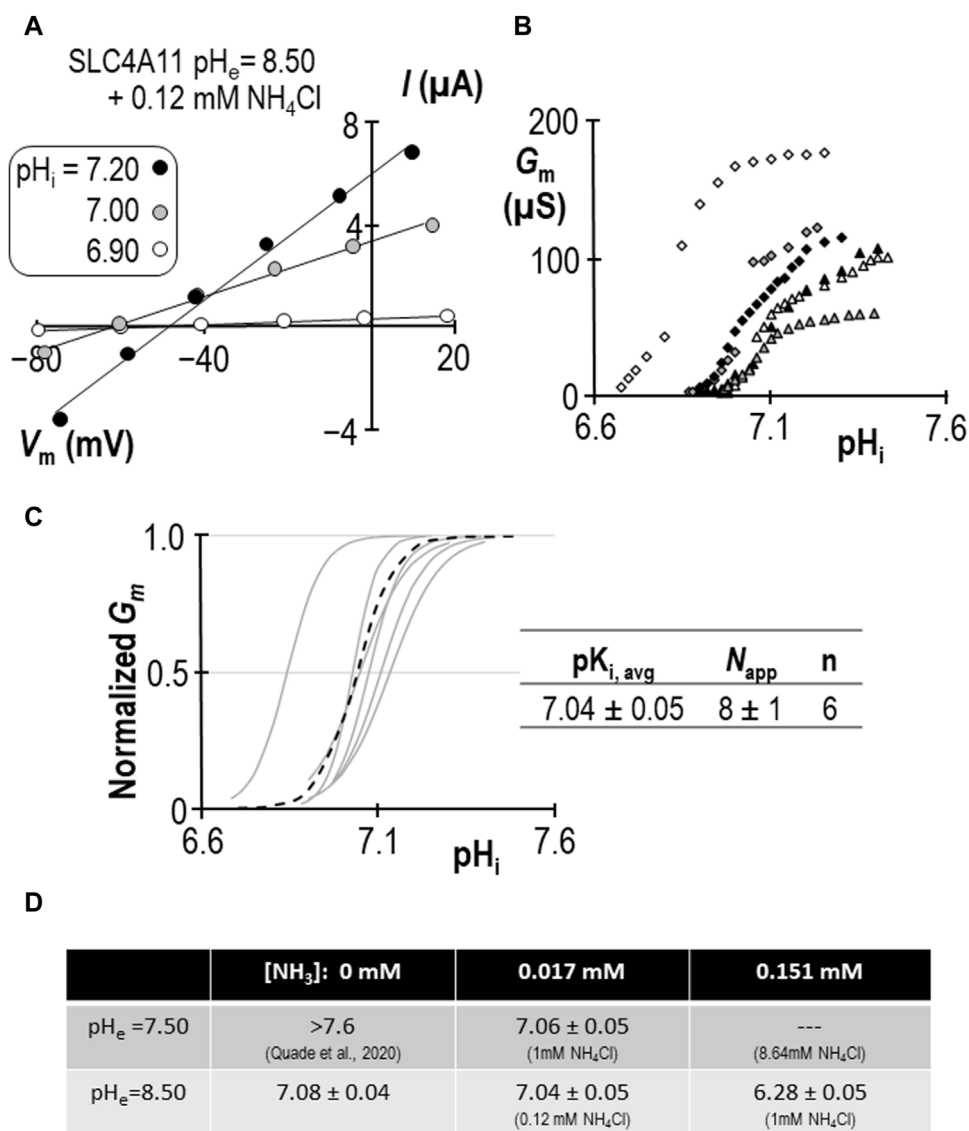
7.50 (light gray arrow in Figures 7A,B/upper panel) than at  $pH_e = 8.50$  (dark gray arrow in Figures 7A,B/lower panel). In this case it is not necessary to invoke a model in which the abundance of the substrate/activator is  $pH$  dependent;  $NH_4^+$  (whose fractional abundance is relatively unchanged between  $pH_e 7.50$  and  $8.50$ ) could be considered equally likely to be the species responsible for the observed stimulation.

Our data do not speak definitively to the nature of the activating species. We believe, because the time course of  $pK_i$  shift ( $>10$  min: Figure 3A) is much slower than the rate of solution turnover in the bath ( $<1$  min) that the allosteric activation likely requires the activating species to accumulate intracellularly in order to exert its effect on SLC4A11. Because the handling of  $NH_3/NH_4^+$  by oocytes is unusually complex (Musa-Aziz et al., 2009), it is difficult to specifically relate this time course of activation to the accumulation of either species. On the one hand  $NH_3$  is presumed to be the more membrane permeable of the two species. On the other hand, in contrast to the expected alkalinization observed in mammalian cells exposed to  $NH_4Cl$ , (even  $H_2O$ -injected) oocyte  $pH_i$  paradoxically acidifies as  $NH_4^+$  is accumulating faster, perhaps entering via non-selective cation channels as  $NH_3$  is sequestered in sub-membranous granules (Burckhardt and Frömter, 1992). For this

reason, it is not clear whether there is a diagnostically useful differential in their rate of accumulation that could point to one species over the other.

### 3.3 SLC4A11 retains $H^+$ selectivity in the presence of $NH_4Cl$

Another critical parameter that is unchanged in the presence of  $NH_4Cl$  is its  $H^+$  selectivity, because the  $V_m$  of SLC4A11-expressing cells exhibits a close-to-Nernstian response to changes in the transmembrane  $pH$  gradient (Figure 4). This suggests that SLC4A11 remains a selective  $H^+$  conductor rather than assuming a novel  $NH_3$ -coupling mode. Although previous studies have claimed to provide evidence of  $NH_3$ -coupled  $H^+$  transport using a similar approach (Zhang et al., 2015; Kao et al., 2020), we believe that these findings should be interpreted with caution as the approach violates necessary assumptions of reversal potential calculations: chiefly that substrates can only cross the membrane via SLC4A11 (not true for  $NH_3$ ) and that  $V_m$  is dominated by the action of SLC4A11 (not true due to the depolarizing action of  $NH_3/NH_4^+$ ). In the absence of a specific inhibitor for SLC4A11, to distinguish the



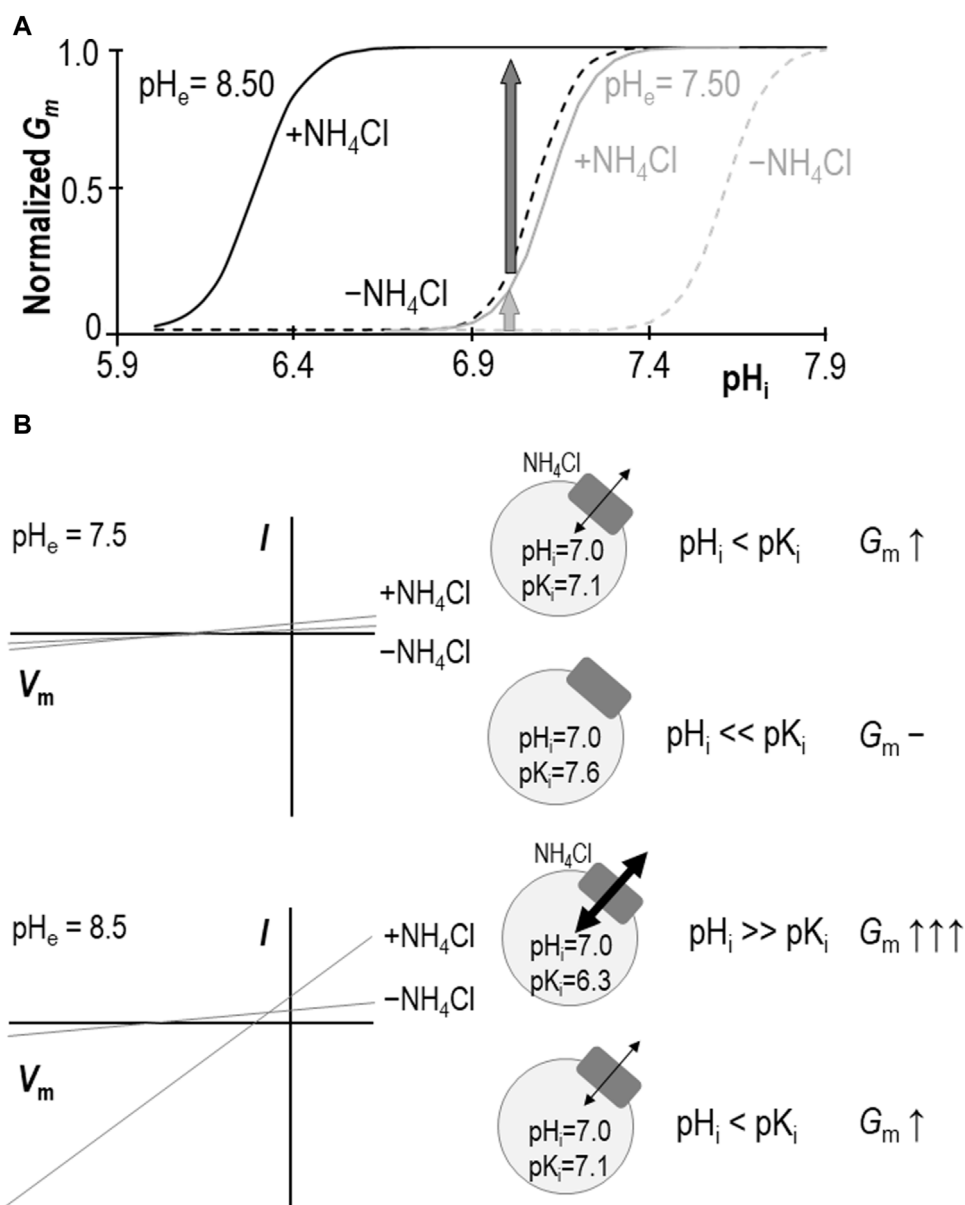
**FIGURE 6** SLC4A11 behavior in the presence of 0.12 mM  $NH_4Cl$  (extracellular pH = 8.50). (A) A representative selection of current-voltage (I-V) relationships gathered from a single SLC4A11-expressing oocyte as intracellular pH ( $pH_i$ ) rises. (B) The relationship between  $pH_i$  and slope conductance ( $G_m$ ) from the full-set of I-V relationships gathered from the SLC4A11-expressing cell shown in panel (A) (black diamonds) and from five other SLC4A11-expressing cells (each represented by its own symbol). (C) Best-fit lines for each cell to the Hill equation are shown in gray. The dashed black line represents the Hill equation generated using the average  $pK_i$  and apparent Hill coefficients ( $N_{app}$ ) of the  $n = 6$  replicates, as shown in the inset table. (D) A summary of the  $pK_i$  of SLC4A11 determined under the various conditions tested in this study.

behavior of the protein from that of the system, the best that such calculations may achieve is a description of the permeability of the system (i.e., SLC4A11 and its membrane environment). One such study concluded that SLC4A11 was capable of both  $NH_3:H^+$  cotransport and  $H^+$  transport (Kao et al., 2020). If so, this is not the typical action of an obligatorily coupled cotransport protein and could equally describe the behavior of a  $H^+$  conductor in an  $NH_3$ -permeable membrane.

There is one unusual aspect to our reversal potential data that requires further explanation. In previous studies we have found that the relationship between the transmembrane gradient and  $V_m$  for wild-type SLC4A11 crosses the  $x$ -axis at a value close to zero as expected for a  $H^+$  conductor. In the present study

performed in the presence of  $NH_4Cl$ , we find that the relationship is substantially offset from the origin and intersects the  $x$ -axis at  $-1.25$  pH-units as if we had underestimated  $pH_i$  by 1.25 units. This is not impossible, as seven of the ten data sets were gathered while SLC4A11 was mediating  $H^+$  influx and thus pH immediately below the membrane may have been more acidic than bulk  $pH_i$  being measured at the tip of our microelectrode, which is impaled deeper into the cell. However, three of these data sets were gathered while SLC4A11 was mediating  $H^+$  efflux, and exhibited the same offset, so we do not believe that this can be the correct explanation. An alternate explanation is illustrated in Figure 8. Here we show that, if we consider these data as being

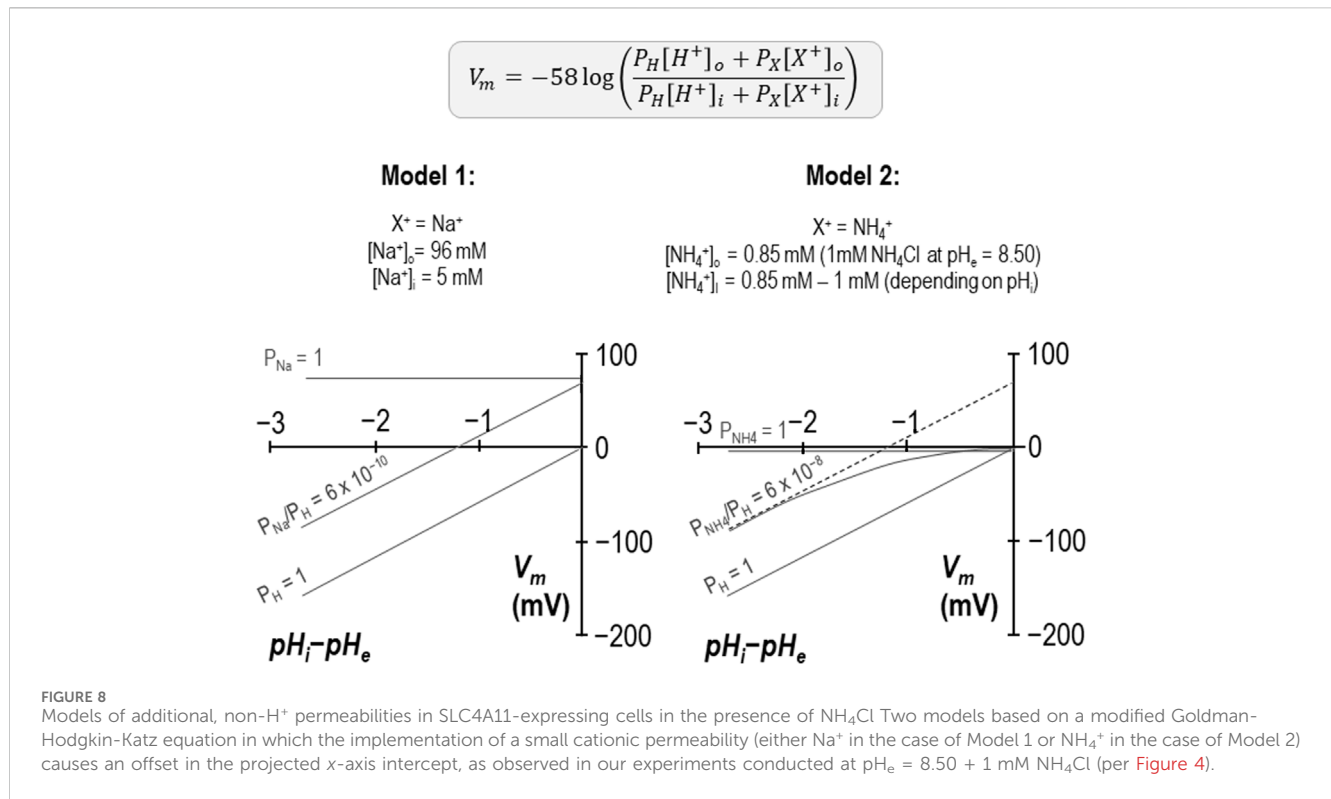




**FIGURE 7** Models of SLC4A11 action in the presence and absence of  $NH_4Cl$ . **(A)** Cartoon representing the  $pH_i$  versus  $G_m$  relationships for SLC4A11 at  $pH_e$  7.50 (gray lines) and 8.50 (black lines) in the presence (solid lines) and absence (dashed lines) of 1 mM  $NH_4Cl$ . **(B)** Cartoon showing how the scheme in panel (A) could explain how  $NH_4Cl$  can achieve greater potency with respect to enhancing SLC4A11-mediated conductance at  $pH_e = 8.5$  (lower panel) versus 7.5 (upper panel), as observed in prior studies such as Zhang et al., 2015 or Myers et al., 2016. I-V plots are cartoons, gray circles represent cells expressing SLC4A11 (gray boxes) with black arrows indicating relative magnitudes of conductance.

representative of the system and interpret them using a modified Goldman-Hodgkin-Katz equation, we can reproduce the offset by implementing a small permeability to a depolarizing cation. Two examples are provided: in Model 1, we add a sodium permeability to the system. Because the abundance of  $H^+$  is so small compared to the abundance of  $Na^+$ , the relative permeability of  $Na^+$  to  $H^+$  must be very small not to completely overwhelm  $V_m$ . In this instance,  $P_{Na}/P_H = 6 \times 10^{-10}$  provides a good fit to our observations. In Model 2 we add an  $NH_4^+$  permeability to the system. We assume that  $[NH_4Cl]$  is 1 mM on both sides of the membrane and calculate  $[NH_4^+]_i$  for each value of  $pH_i$ . In this instance  $P_{NH_4}/$

$P_H = 6 \times 10^{-8}$  provides a good fit for our data. Although the relationship curves off to an asymptote, the initial slope is Nernstian with respect to  $H^+$ , and its projection (dashed gray line) crosses the  $x$ -axis at  $-1.25$  pH units. We note that the  $x$ -intercept of our data is also a projection and we do not know whether our data would also curve in a similar way if extended towards the  $x$ -axis. In any case this is not a critical issue as the relationship can be made to conform to the dashed line if we lower our estimate of  $[NH_4Cl]_i$  to 0.1 mM. As either permeability is trivial compared to that of  $H^+$ , it does not appear to be a major confounding factor to our hypothesis. In the absence of a specific blocker, we cannot know whether the additional permeability is



intrinsic to SLC4A11 or to the system in general. However, as we have only observed this in the presence of NH<sub>4</sub><sup>+</sup>, and because NH<sub>4</sub><sup>+</sup> is a depolarizing influence in even H<sub>2</sub>O-injected cells, we tentatively suggest that the x-axis offset represents the previously described endogenous NH<sub>4</sub><sup>+</sup> conductance.

### 3.4 The relationship between pK<sub>i</sub> shifts caused by extracellular alkalization versus NH<sub>4</sub>Cl addition

As previously observed, in the absence of NH<sub>4</sub>Cl, a shift of pH<sub>e</sub> from 7.50 to 8.50, is itself sufficient to acid-shift SLC4A11 pK<sub>i</sub> by more than 0.5 pH-units (Quade et al., 2020). If we compare SLC4A11 pK<sub>i</sub> determined at pH<sub>e</sub> = 7.50 + 1 mM NH<sub>4</sub>Cl to pK<sub>i</sub> determined at pH<sub>e</sub> = 8.50 + 0.12 mM NH<sub>4</sub>Cl (conditions in which [NH<sub>3</sub>] is the same and pK<sub>i</sub> is not different) we may conclude that, in the presence of NH<sub>4</sub>Cl, pK<sub>i</sub> has become pH<sub>e</sub>-independent and is determined by [NH<sub>3</sub>] alone. In that case we could consider [NH<sub>3</sub>] and pH<sub>e</sub> as activators that share a common mechanism. If SLC4A11 is less pH<sub>e</sub> dependent in the presence of NH<sub>4</sub>Cl, that would imply that the stimulatory effect of pH<sub>e</sub> on SLC4A11 may have limited relevance to its *in vivo* action.

On the other hand, if we compare SLC4A11 pK<sub>i</sub> determined at pH<sub>e</sub> = 7.50 + 1 mM NH<sub>4</sub>Cl to pK<sub>i</sub> determined at pH<sub>e</sub> = 8.50 + 1 mM NH<sub>4</sub>Cl, values which are ~0.8 units apart, we may conclude that the phenomenon of pH<sub>e</sub>-dependence is preserved in the presence of 1 mM NH<sub>4</sub>Cl and that the actions of NH<sub>4</sub>Cl and pH<sub>e</sub> are, at least in part, mechanistically independent and additive. The common ground between these two ways of looking at the

data are that the presence of NH<sub>4</sub>Cl alters the relationship between pH<sub>e</sub> and pK<sub>i</sub> such that a more acidic pK<sub>i</sub> can be achieved at a given pH<sub>e</sub>. Interestingly, this is the opposite to a phenomenon that we have observed in relation to certain pathological SLC4A11 mutants such as R125H, in which the mutation causes pK<sub>i</sub> to become more alkaline at a given pH<sub>e</sub> (Quade et al., 2022). Unfortunately, the technical limitations on how far we can extend pH<sub>i</sub> and the low resolution of our pK<sub>i</sub> assay (due to the large standard error intrinsic to the data sets) currently preclude us from more detailed exploration of the relationship between these activating parameters.

### 3.5 Implications for the physiological role of SLC4A11

The fluid pumping action of corneal endothelial cells requires the action of a basolateral Na/K-ATPase to generate the transmembrane sodium gradient that drives the Na<sup>+</sup>/CO<sub>3</sub><sup>=</sup> cotransporter, drawing osmolytes from the stromal fluid to discourage fluid accumulation. The pump is energized by glutaminolysis that feeds α-ketoglutarate into the TCA cycle to generate ATP; a side product of this reaction is NH<sub>3</sub> (Zhang et al., 2017). We have previously hypothesized that acid-loading by SLC4A11 could be useful to stabilize pH<sub>i</sub> during robust bicarbonate pump function, responding to Na<sup>+</sup>/CO<sub>3</sub><sup>=</sup> cotransporter action by sensing a local rise in pH<sub>i</sub> (Myers et al., 2016; Nehrke, 2016). Our new data indicate that SLC4A11 is a H<sup>+</sup> conductor both in the presence and absence of NH<sub>4</sub>Cl and thus that NH<sub>3</sub>/NH<sub>4</sub><sup>+</sup> is an allosteric activator rather than a cotransported substrate of SLC4A11. This action is incompatible with a role of

corneal endothelial SLC4A11 in mediating the export of excess  $\text{NH}_3$  from glutaminolysis, and incompatible with SLC4A11 being able to harness an outwardly-directed  $\text{NH}_3$  gradient to mediate  $\text{H}^+$  efflux. Because the electrochemical gradient for  $\text{H}^+$  is typically inwardly directed, we predict that a rise in intracellular  $\text{NH}_3/\text{NH}_4^+$  promotes  $\text{H}^+$  influx independent of any rise in  $\text{pH}_i$ . The ability of  $\text{NH}_3/\text{NH}_4^+$  to acid shift  $\text{pK}_i$  implies that the ability of SLC4A11 to support pump function is potentiated by  $\text{NH}_3/\text{NH}_4^+$ , the generation of which could be considered to be a proxy for the energetic requirements of the pump.

### 3.6 Summary

The presence of  $\text{NH}_4\text{Cl}$  causes an acidic shift in the  $\text{pK}_i$  of human SLC4A11, which translates to an increase in  $G_m$  at physiological values of  $\text{pH}_i$ . The presence of  $\text{NH}_4\text{Cl}$  does not affect the  $\text{H}^+$  selectivity of SLC4A11, thus we conclude that  $\text{NH}_4\text{Cl}$  is an allosteric activator of SLC4A11-mediated  $\text{H}^+$  conductance. The influence of increasing  $[\text{NH}_4\text{Cl}]$  upon SLC4A11 activity is reminiscent of the influence of increasing  $\text{pH}_e$ , but further work will be required to determine whether these share a common mechanism.

## 4 Materials and methods

### 4.1 Oocyte preparation and culture

Ovaries were harvested from female *Xenopus laevis* (*Xenopus* Express, Brooksville, FL) in accordance with the protocol approved by the University at Buffalo Institutional Animal Care and Use Committee. Frogs were anesthetized in 0.2% tricaine solution, ovariectomized, and euthanized by exsanguination. Extracted tissue was cut into  $\sim 1 \text{ cm}^2$  pieces and washed in a  $\text{Ca}^{2+}$ -free solution (82 mM NaCl, 2 mM KCl, 20 mM  $\text{MgCl}_2$ , 5 mM HEPES, pH 7.50). Oocytes were liberated by digestion in 2 mg/mL type 1A collagenase solution, and the isolated cells were washed further in the  $\text{Ca}^{2+}$ -free solution to remove the collagenase prior to resuspension in a physiological buffer (ND96: 96 mM NaCl, 2 mM KCl, 1.8 mM  $\text{CaCl}_2$ , 1 mM  $\text{MgCl}_2$ , 5 mM HEPES, pH 7.50, 200 mOsmol/kg  $\text{H}_2\text{O}$ ). Until experimental use, oocytes were cultured at  $18^\circ\text{C}$  in OR3 medium (14 g/L Leiboviz's L-15 medium powder, 5 mM HEPES, 20 mL/L 100x penicillin-streptomycin, pH 7.50, 200 mOsmol/kg  $\text{H}_2\text{O}$ ).

### 4.2 cRNA preparation and injection

Our starting material was a clone of human SLC4A11-B in pBSXG4 vector. The construct was linearized with HindIII, which cuts at a site downstream of the open reading frame providing a termination point for transcription. The linearized DNA was purified using a MinElute PCR Purification Kit (QIAGEN, Germantown, MD) used as a template for cRNA synthesis using the T7 mMESSAGE mMACHINE kit (Invitrogen, Carlsbad, CA). cRNA was purified using an RNeasy MinElute Cleanup Kit

(QIAGEN). 25 ng of cRNA or  $\text{H}_2\text{O}$  was injected into each oocyte using a Nanoject programmable injector (Drummond Scientific, Broomall, PA).

### 4.3 Electrophysiology

Oocytes were placed into chamber (RC-3Z: Warner Instruments, Hamden, CT) on an anti-vibration table (Vision IsoStation; Newport Corp., Irvine, CA) and were superfused at 2 mL/min with solutions fed from syringe pumps (Harvard Apparatus, Holliston, MA). Borosilicate glass capillaries (BF200-156-10; Sutter Instrument, Novato, CA) were pulled into microelectrodes (such that they exhibited a tip resistance of 0.1–2 M $\Omega$  when filled with saturated KCl solution) using a micropipette puller (P-1000; Sutter Instrument). Oocytes were impaled with two such KCl-filled microelectrodes (one current-passing and one voltage-sensing) connected to an oocyte clamp (OC275; Warner Instruments, Hamden, CT). A bath clamp (725I; Warner Instruments) was used to hold the potential of the chamber fluid at 0 mV. Current-voltage (I-V) plots were gathered in 20 mV, 100 ms steps, returning to the spontaneous membrane potential for 100 ms between each step.  $\text{H}^+$ -selective microelectrodes were pulled in the same manner as voltage electrodes but the tips were filled with hydrogen ionophore I/cocktail B (Sigma Aldrich) and backfilled with a solution composed of 40 mM  $\text{KH}_2\text{PO}_4$ , 15 mM NaCl, pH 7.0. These electrodes were connected to a dual-channel electrometer (HiZ-223; Warner Instruments). Complete technical details can be found in the 2013 review by Lee, Boron, and Parker (Lee et al., 2013). Signals were digitized via a Digidata 1550 unit and captured using Clampex 10.4 software (Molecular Devices LLC, San Jose) and custom continuous acquisition software (written by Mr. Dale Huffman for Walter Boron's laboratory at Case Western Reserve University, Cleveland, OH).

### 4.4 Electrophysiology solutions

pH 7.50 solutions contained 96 mM NaCl, 2 mM KCl, 1.8 mM  $\text{CaCl}_2$ , 1 mM  $\text{MgCl}_2$ , 5 mM HEPES, 200 mOsmol/kg  $\text{H}_2\text{O}$ . pH 8.50 solutions had the same composition but were buffered with 5 mM Bicine in place of HEPES.  $\text{NH}_4\text{Cl}$  was added to  $\text{NH}_4\text{Cl}$ -containing solutions as a powder, and pH was readjusted as necessary.

### 4.5 Data analysis

Data are presented as means  $\pm$  standard error of the mean. Slope conductance ( $G_m$ ) was determined from the slope of a linear trendline fit to I-V data in Microsoft Excel. Normalized  $G_m$  data was plotted against  $\text{pH}_i$  (expressed as  $[\text{OH}^-]$ ) and fit to the Hill equation using the solver add-in of Excel to determine the values of  $\text{EC}_{50}$  and  $N_{app}$  that would result in the minimum root square difference between the observed data and the outcome of the Hill equation:

$$\frac{G_m}{G_{m,max}} = \frac{1}{1 + (\text{EC}_{50}/[\text{OH}^-])^{N_{app}}}$$

$EC_{50}$  was converted into a value of  $pK_i$  using the following equation:

$$pK_i = 14 - [-\log(EC_{50})]$$

For Figure 3A, we calculated  $pK_{i,app}$  by solving the Hill equation for  $EC_{50}$  at each pair of  $G_m$  and  $[OH^-]$  (i.e.,  $pH_i$ ) data points. We assumed that  $G_{m,max}$  and  $N_{app}$  were constants: 79  $\mu S$  and 7 respectively, corresponding to the data gathered from this cell during period “b.” Calculations of  $[NH_3]$  at a given pH value and  $[NH_4Cl]$  assume a  $pK_a$  for the  $NH_3/NH_4^+$  equilibrium of 9.25.

Statistical analysis was performed in Excel using unpaired t-tests, one- or two-tailed as necessary. For multiple comparison, ANOVA was performed using MiniTab software.

## Data availability statement

The datasets presented in this study can be found in online repositories. The names of the repository/repositories and accession number(s) can be found below: <https://doi.org/10.6084/m9.figshare.25928233.v1>.

## Ethics statement

The animal study was approved by IACUC at University at Buffalo Jacobs School of Medicine. The study was conducted in accordance with the local legislation and institutional requirements.

## Author contributions

RP: Conceptualization, Data curation, Formal Analysis, Investigation, Methodology, Validation, Visualization, Writing–original draft, Writing–review and editing. BQ: Conceptualization, Investigation, Methodology, Visualization, Writing–review and editing. AM: Methodology, Project

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## Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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