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Chitosan and its derivatives regulate lactic acid synthesis during milk fermentation

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Introduction: The influence of chitosan's physicochemical characteristics on the functionality of lactic acid bacteria and the production of lactic acid remains very obscure and contradictory to date. While some studies have shown a stimulatory effect of oligochitosans on the growth of Lactobacillus spp, other studies declare a bactericidal effect of chitosan. The lack and contradiction of knowledge prompted us to study the effect of chitosan on the growth and productivity of *L. bulgaricus* in the presence of chitosan and its derivatives.

Methods: We used high molecular weight chitosan (350 kDa) and oligochitosans (25.4 and 45.3 kDa). The experiment was carried out with commercial strain of *L. bulgaricus* and the low fat skim cow milk powder reconstituted with sterile distilled water. After fermentation, dynamic viscosity, titratable acidity, pH, content of lactic acid, colony forming units, chitosan and oligochitosans radii were measured in the samples. Fermented dairy products were also examined using sodium dodecyl sulfate electrophoretic analysis, gas chromatographymass spectrometry and light microscopy.

Results and discussion: The results of the study showed that when *L. bulgaricus* was cultured in the presence of 25.4 kDa oligochitosans at concentrations of 0.0025%, 0.005%, 0.0075% and 0.01%, the average rate of LA synthesis over 24 hours was 11.0×10^{-3} mol/L/h, 8.7×10^{-3} mol/L/h, 6.8×10^{-3} mol/L/h, 5.8imes 10⁻³ mol/L/h, respectively. The 45.3 kDa oligochitosans had a similar effect, while the average rate of lactic acid synthesis in the control sample was only 3.5×10^{-3} mol/L/h. Notably, 350 kDa chitosan did not affect the rate of lactic acid synthesis compared with the control sample. Interestingly, interaction of chitosan with L. bulgaricus led to a slowdown in the synthesis of propanol, an increase in the content of unsaturated and saturated fatty acids, and a change in the composition and content of other secondary metabolites. The quantity of L. bulgaricus in a sample with 0.01% chitosan exceeded their content in the control sample by more than 1,700 times. At the same chitosan concentration, the fermentation process was slowed down, increasing the shelf life of the fermented milk product from 5 to 17 days while maintaining a high content of L. bulgaricus $(6.34 \times 10^{6} \text{ CFU/g}).$

KEYWORDS

oligochitosan, lactic acid, lactobacilli, fatty acids, propanol, benzaldehyde, secondary metabolites, glucosamine monomers

1 Introduction

A significant amount of scientific research has been dedicated to improving the quality and functional value of food products over the last few decades, as functional products can enhance people's quality of life and health compared to conventional foods (1, 2). One of the most common groups of functional food products includes fermented dairy products, which are recognized as the most popular and extensively produced and consumed dairy products worldwide (3, 4). In the production of fermented dairy products, the production of lactic acid by bacteria of the genera Lactobacillus and Bifidobacterium genera is commonly employed (5). Among lactic acid bacteria, L. bulgaricus represents the most widespread in Central, Eastern, and South-Eastern Europe, commonly used in the preparation of fermented dairy products (6). During the growth of L. bulgaricus in milk and dairy products, the bacteria produce beta-β-galactosidase (E.C.3.2.1.23), which catalyzes the fermentative cleavage of milk lactose (β-Dgalactopyranosyl- $(1 \rightarrow 4)$ -D-glucose) into glucose and galactose, followed by the homofermentation of these carbohydrates into lactic acid as the sole or main end product (7, 8).

The cultivation of lactic acid bacteria usually leads to the production of functional additives such as macro- and micronutrients and dietary fiber (9, 10). Among them, natural poly- and oligosaccharides are of particular interest and value, as they can be used not only as thickeners and gelling agents but also as prebiotics that stimulate the growth of beneficial microorganisms and prolongs their activity in the digestive system (11, 12). Considering this, it is important to note that recent works are increasingly studying Chitosan as polysaccharides with specific properties (13–15).

Chitosan is a partially or fully deacetylated chitin—poly- $(1 \rightarrow 4)$ - β -D-N-acetylglucosamine. It is produced industrially by the deacetylation of chitin from crab, shrimp, or insect shells and is widely used in pharmaceuticals, food, and cosmetic compositions due to its wide range of antimicrobial activities against bacteria, molds, and yeasts (16–19). As an artificial biopolymer, chitosan is characterized by its molecular weight and degree of deacetylation (%) (20, 21). Based on molecular weight, chitosan can be divided into three categories: low molecular weight (molecular weight < 150 kDa), medium molecular weight (molecular weight < 700 kDa), and high molecular weight (molecular weight > 700 kDa) (22).

High molecular weight chitosan possesses longer molecular chains with the availability of more hydroxyl groups (23). There is also a higher possibility that there are more amino groups, although the number of amino groups is determined by the degree of deacetylation (24). High molecular weight, high degree of polymerization, and, as a consequence, a high number of inter- and intra-molecular hydrogen bonds inside the polymer chains determine the relatively low solubility of High molecular weight chitosan and limit its application in some products due to high viscosity (25). However, specific properties of high molecular weight chitosan found wide application in development of active films for food packaging application (26). Medium molecular weight chitosan is soluble in weak acid solutions. This limits their usage compared to the antimicrobial capacity of acid- and water-soluble chitosan with different degrees of deacetylation and viscosities (22). Low molecular weight chitosan is a linear amino polysaccharide with high nitrogen content. It is a weak base with deprotonated amino groups as nucleophiles that is able to form hydrogen bonds between molecules and has highly reactive groups for crosslinking and chemical activation (27).

Low molecular weight chitosan forms salts with organic and inorganic acids, has chelating and complex properties, and exhibits ionic conductivity as polyelectrolytes (pH < 7) (28). In contrast to high and medium molecular weight chitosan, low molecular weight chitosan demonstrates considerable solubility in various media. However, it is characterized by an unpredictably wide distribution of molecular weight and degree of deacetylation, complicating the standardization of parameters for industrial applications (29-31). Therefore, to address this challenge in dispersing food systems such as milk, a promising avenue lies in the conversion of chitosan into oligomers with more consistent molecular sizes and improved solubility (32). These oligomers, termed oligochitosans, result from the profound depolymerization of chitosan. It is noteworthy that significant disparities exist in the physicochemical parameters and biological activity between chitosan and oligochitosans, accentuating the scientific interest in their comparative study within the realm of food technology applications (33-35).

Amidst its bioavailability, safety, and antimicrobial attributes, chitosan exhibits immunostimulatory and anti-angiogenic properties, mitigating the risk of neurodegenerative conditions while facilitating the regeneration of articular cartilage in osteoarthritis, and enhancing the bioavailability of glucosamine (36–38). In an acid media, chitosan has a high positive charge density due to the protonation of free amino groups and can interact with negatively charged cell walls (31, 39). Therefore, chitosan can potentially interact with the plasma membrane of lactic acid bacteria causing a stress and perturbation of the membrane walls and the death of the cells (40, 41). As a response, LAB can generate a number of adaptive reactions (42, 43) which may insert desirable or undesirable changes in the functionality of lactic acid bacteria and fermented dairy products.

Unfortunately, the influence of chitosan's physicochemical characteristics on the functionality of lactic acid bacteria and the production of lactic acid remains very obscure and contradictory to date. While some studies have shown a stimulatory effect of oligochitosans on the growth of *Lactobacillus* spp., which used added oligochitosans as nutrients (44, 45), others have described a bactericidal effect of chitosan (40, 42, 46). The lack and contradiction of knowledge prompted us to study the effect of chitosan on the growth and productivity of *L. bulgaricus* in the presence of chitosan and its derivatives. The aim of this work was therefore to study the effect of different concentrations of oligochitosans and chitosan on the production of lactic acid and secondary metabolites by *L. bulgaricus* during fermentation and long-term storage of skimmed cow's milk.

2 Materials and methods

2.1 Materials

Commercial strain of *L. bulgaricus* and the low fat skim cow milk powder (solubility–93%, moisture content–5%; fat content–1.5%; proteins–32%, and lactose–50 \pm 3%) were purchased from official store of "Stavropolsky Dairy Plant" (Stavropol, Russia). The milk powder was reconstituted with sterile distilled water (50 ml) at temperature 30–35°C during 24 h at sterile conditions to the total solids concentration not <9.0%. The reconstituted skim milk solution was characterized by the following parameters: proteins–3.2%, lactose–5%, total fat–0.15%, and mineralization–0.7%, pH 5.9.

High molecular weight chitosan with 350 kDa molecular weight and 5.0% DD manufactured by Bioprogress Ltd. (Moscow, Russia). Two types of oligochitosans hydrochlorides with 25.4 kDa molecular weight, 1.0% degree of deacetylation and 45.3 kDa molecular weight, 1.5% degree of deacetylation were obtained from the initial chitosan according to the protocol of Berezin et al. (47). The obtained oligochitosans were analyzed in accordance with the requirements of the EU Pharmacopeia 9.0 for chitosan hydrochloride (48). Oligochitosans with a solubility in demineralized water Milli-Q > 99.95% formed a colorless and untroubled 1% solution in accordance with the tests on transparency and degree of opalescence, and degree of liquids coloring. The initial 1% chitosan solution was prepared by dissolving the sample in 1% lactic acid. The pH of the initial solution of the sample was adjusted to pH 5.0 by adding drops of 1M sodium hydroxide solution.

2.2 Milk sample preparation and fermentation

Reconstituted skim milk (100 mL) was mixed with a fixed amount (0.0025, 0.005, 0.075, and 0.01 g) of chitosan stock solution, and the mixture was pasteurized at 85°C for 5 min. After cooling to 43–45°C, 3 g of the commercial freeze-dried starter culture of *L. bulgaricus* was added to achieve a viable count of 10⁵ CFU/mL in the sample. The sample was thoroughly mixed and incubated at 43–45°C for 17 days until the maximum titratable acidity was reached. Subsequently, the sample was stored at 4°C before being analyzed.

2.3 Fermented milk product analysis

2.3.1 Dynamic viscosity

After the storage, the dynamic viscosity of fermented dairy product was measured at 20°C using a Brookfield digital rotational viscometer DV-II+PRO (Brookfield Engineering Laboratories, Middleboro, MA, USA).

2.3.2 Titratable acidity and pH

During the fermentation, the milk sample (10 mL) was centrifuged at 6000 rpm (30 min) using MicroCL 17R centrifuge

(Thermo FS, Waltham, MA, USA), and pH value of the supernatant was measured with analizator Expert-001 (Econix-Expert, Moscow, Russia). The titratable acidity of fermented dairy product was determined by titration of the supernatant with 0.1 N NaOH using phenolphthalein as indicator (49). Titratable acidity was expressed in percentages of lactic acid content (T, %) or as the volume of 0.1 M of sodium hydroxide consumed for the neutralization of 100 ml of fermented milk product (V_t, mL).

2.3.3 Content of lactic acid

Molar concentrations of lactic acid (mol/L) in the fermented milk supernatant were determined following the method described in the previous work (50). The molar concentrations of lactic acid (mol/L) were calculated following the equation:

$$LA(mol/L) = [V_t \ 0.1]/100; \tag{1}$$

where V_t-titratable acidity of the fermented milk supernatant.

2.3.4 Sodium dodecyl sulfate electrophoretic analysis

The analysis of the protein composition of fermented milk product was carried out after 17 days of storage using the method of sodium dodecyl sulfate (SDS) electrophoretic separation in polyacrylamide gel (20%) under denaturing conditions (DSN electrophoresis) in accordance with the generally accepted protocol (51).

2.3.5 Colony forming units counting

Colony count technique for CFU determination was used in accordance with ISO 15214 (1998-2021) after 10–1000-fold dilution of the fermented milk supernatant solution.

2.3.6 Microscopy

Sample supernatant was centrifuged at 15,000 rpm for 10 min using MicroCL 17R centrifuge (Thermo FS, Waltham, MA, USA) and filtrated using a $0.2 \,\mu$ m PVDF membrane. The morphology of lactic acid bacteria was observed after the membrane filtration. Samples was treated with methylene blue, destained by water washing and dried on air. Photographs were made and recorded using a BIOLAM light microscope (Scopica, Ekaerinburg, Russia).

2.3.7 Measuring the chitosan and oligochitosans radii

The dynamic light scattering method was used to measure the chitosan and oligochitosans molecules radii. Experiments on dynamic light scattering were carried out in a disposable microcuvette with a volume of 4 μ l, using a detector located at an angle of 90° in the DynaPro Nanostar device (Wyatt Technology, Santa Barbara, CA, USA). The samples were filtered through a 0.2 μ m nylon filter. The data was analyzed using the "Regulation fit" (multimodal) analysis method in the Dynamics software (Wyatt Technology, Santa-Barbara, CA, USA).

2.3.8 Analysis of secondary metabolites: gas chromatography-mass spectrometry

After 17 days of storage, a 10 mL sample of the fermented dairy product was centrifuged at 6,000 rpm for 30 min using a MicroCL 17R centrifuge (Thermo FS, Waltham, MA, USA), followed by lyophilization. The residual solid product (1g) was then extracted twice with 70% ethanol (1:10 wt/v). The resulting extracts were combined and filtered through 0.25 µm PVDF membranes. Secondary metabolites were separated and analyzed using an Agilent 5975B gas chromatograph equipped with a mass selective detector (Agilent Technologies, Santa Clara, CA, USA). The separation of secondary metabolites was performed using a capillary column DB-5MS (5% phenyl methyl siloxane, J&W 122-5062). Identification of components in the mass spectra was conducted using a library of mass spectra NIST0.5A. Comparative semi-quantitative analysis of secondary metabolites was performed based on peak areas without using correction factors. The semiquantitative content of secondary metabolites was calculated from the peak area without considering the peak of lactic acid and without using correction factors.

2.4 Statistical analysis

In order to compare the means of factor's levels, a one-, two-, or three-way analysis of variance (ANOVA) with subsequent Dunnett's test (comparing several treatments with a control), Student's *t*-test (matching the means of two groups) or Tukey's Honest Significant Difference (HSD) test (performing multiple pairwise comparisons) were applied. R functions aov, DunnettTest, *t*-test, TukeyHSD and DescTools packages were involved in the statistical analysis. Statistical differences between groups were set as significant at p < 0.05 level with correction for multiple pairwise comparisons. Correlation analysis was performed using Pearson's Criterion. Plots were created in Microsoft Office Excel (MS Corporation, Shadeland, IN, USA). The data in tables and graphs are shown as the mean \pm the half-width of 95% confidence interval (n = 3).

3 Results and discussion

3.1 Cultivation of *L. bulgaricus* in the presence of chitosan and oligochitosans

The effect of oligochitosans of 25.4 kDa molecular weight, 1% degree of deacetylation, and 45.3 kDa molecular weight, 1.5% degree of deacetylation on the lactic acid fermentation process was studied in comparison with CH of 350 kDa molecular weight, 5.0% degree of deacetylation at *L. bulgaricus* cultivation. During cultivation, pH and titratable acidity of the culture liquid were measured in experimental and control samples every 4 h for 24 h. Based on the results of titratable acidity, the molar concentration and the rate of synthesis of lactic acid were calculated. Figure 1 shows changes in the pH and titratable acidity of a culture liquid containing 0.0025, 0.005, 0.0075, and 0.01% oligochitosans of 25.4 and 45.3 kDa. According to the data obtained, pH depends on the concentration of oligochitosans (Figures 1A, C). Notably, the maximum decrease in pH was observed at oligochitosans concentration of 0.0025% (Table 1). At the end of 24-h cultivation, pH of the experimental samples was lower than in control sample by 73% in 25.4 kDa oligochitosans, by 75% in 45.3 kDa oligochitosans, and by 1% in 350 kDa chitosan. At higher concentrations, pH less intense decreased in samples with oligochitosans, but particularly was not changed in samples with chitosan. Inversely dependent, the titratable acidity of the culture liquid decreased with the increase of oligochitosans concentration, but was higher than the titratable acidity of control samples (Figures 1B, D).

According to Table 1, at the same percentage concentration of oligochitosans and chitosan, their molar concentration in fermented dairy products differs significantly. However, the total content of glucosamine monomers included in their composition was the same. In the acidic environment, the oligochitosans amino groups are in a protonated state (52). Being positive charge, they can interact with negatively charged cell walls of lactic acid bacteria (53, 54). With an increase in the concentration of oligosaccharides in the experimental samples, the content of glucosamine monomers also increased. Consequently, the number of positively charged amino groups capable of interacting with lactic acid bacteria increased.

It was revealed that at 0.0025% concentration and 1.39 \times 10⁻⁴ mol/L glucosoamine monomers, oligochitosans increase the synthesis of lactic acid by 3 times (25.4 kDa) and 2.7 times (45.3 kDa) relative to control sample. It should be noted that samples with chitosan, like samples with oligochitosans, contained 1.39 \times 10⁻⁴ mol/L glucosoamine monomers at a concentration of 0.0025%. However, it was surprisingly found that at 0.0025% chitosan and 1.39 \times 10⁻⁴ mol/L glucosoamine monomers, chitosan slowed down the synthesis of lactic acid by 0.98 times relative to the control. Higher concentrations of chitosan more intense affected lactic acid synthesis and consequently, fermented dairy product with chitosan had less lactic acid concentration than control sample.

On the other hand, at the highest concentration of oligochitosans (0.01 %), the content of glucosamine monomers reached 5.56×10^{-4} mol/L and lactic acid synthesis slowed down. However, the resulting lactic acid content in both oligochitosans groups was still higher than in the control sample. The rate of lactic acid synthesis is critically important in milk fermentation (55). Thus, dependence of the average rate of lactic acid synthesis on concentration of glucosamine monomers of oligochitosans and chitosan at 24 h-cultivation was analyzed and presented in Figure 2.

The analysis of the interaction of lactic acid bacteria with the positively charged amino groups of oligochitosans glucosamine monomers shows that with an increase in their concentration, the rate of lactic acid synthesis slows down (Figure 2). Theoretically, at a concentration of oligochitosans glucosamine monomers close to 10^{-3} mol/L, a process of deep inhibition of lactose metabolism and lactic acid synthesis is possible (56, 57). Similarly, an increase in the content of glucosamine monomers of 350 kDa chitosan leads to a reduction in lactic acid, as shown in Figure 2. According to Figure 1, pH and titratable acidity stabilize on the 20th h of cultivation in samples with 0.0025% oligochitosans. This indicates a slowdown in the synthesis of lactic acid by lactic acid bacteria. Therefore, it is critically important to understand the dynamics of lactic acid accumulation in the presence of oligochitosans. For this



purpose, the rate of lactic acid synthesis was analyzed every 4 h of cultivation. The results obtained are presented in Figure 3.

Notably, the maximum rate of lactic acid synthesis was observed at 0.0025% 25.4 kDa oligochitosan on the 12th h of *L. bulgaricus* cultivation, which is 6.44 times higher than in the control sample. Similarly, on the 12th h of *L. bulgaricus* cultivation, the rate of lactic acid synthesis was the most intense, being 5.44 times higher than in the control sample. It should be pointed out that with an increase in oligochitosans concentration, the rate of lactic acid synthesis decreases and achieves the minimal value on the 24th h of *L. bulgaricus* cultivation. Simultaneously, the rate of lactic acid synthesis reached the maximum value only on the 24th h of *L. bulgaricus* cultivation.

Thus, the results obtained show that 25.4 and 45.3 kDa oligochitosans significantly influence the rate of lactic acid synthesis. The presence of 1.39×10^{-4} mol/L oligochitosan glucosamine monomers in the culture liquid leads to a significant decrease in pH, an increase in titratable acidity, and rapid synthesis and accumulation of lactic acid compared to the control sample. Such oligochitosan activity may be explained by the mechanism of exposure to *L. bulgaricus* through extracellular, intracellular, or both extracellular and intracellular effects (58–61).

The cell wall of *L. bulgaricus* can prevent the direct binding of lactic acid to the cell membrane components and impede intracellular effects (62). Transporting molecules through the barrier of the outer layer of the rigid cell wall occurs through several subtle mechanisms or via simple diffusion (63–66). The porosity of the cell wall and the pore size determine whether oligochitosans can pass through the bacterial cell wall (67). Pore sizes vary among different bacteria and fungi, ranging from 2–4 to 8 nm (68, 69). Therefore, the radii of hydrated oligochitosans were measured using the dynamic light scattering method. According to the data obtained, the average hydrodynamic radius was 2.5 nm for 25.4 kDa oligochitosan and 3.5 nm for 45.3 kDa oligochitosan. Consequently, due to their small sizes, oligochitosans can penetrate the pores of the lactic acid bacteria cell walls and interact with plasma membrane proteins. It is likely that oligochitosans induce stress in lactic acid bacteria cells, leading to adaptive reactions that result in deviations in physiological and biochemical processes (70, 71). For instance, based on the results presented in Figure 3, the presence of 0.0025% oligochitosans in the culture liquid stimulates the acceleration of lactose catabolism, leading to increased lactic acid synthesis and accumulation.

Likely, the reason for such changes is the reaction of lactic acid bacteria stress response caused by oligochitosans. The mechanism of stress response entails the metabolic changes necessary for the survival of lactic acid bacteria (72–74). Previous studies have demonstrated that oligochitosans and chitosan, upon interacting with eukaryotic cells, induce abiotic stress, triggering physiological protective reactions (75). These protective responses may include overexpression of genes associated with carbohydrate metabolism (76), among other reactions (77–79).

At high concentrations, oligochitosans cause not only intracellular but also extracellular effects. Gram-positive bacteria have a negative charge due to the presence of phosphate groups associated with teichoic acids in the cell wall structure (80, 81). With an increase in the concentration of oligochitosans molecules, an electrostatic interaction with negatively charged lactic acid

Index	Concentration of OCHs and CH, %							
	0 (control)	0.0025	0.005	0.0075	0.01			
25.4 kDa, 1% DD OCH (glucosamine monomers 140/mol OCH)								
Molar concentration of OCH, mol/L	-	$0.99 imes 10^{-6}$	$1.98 imes 10^{-6}$	$2.98 imes 10^{-6}$	$3.97 imes 10^{-6}$			
Glucosamine monomers content, mol/L	-	$1.39 imes 10^{-4}$	$2.78 imes 10^{-4}$	$4.17 imes 10^{-4}$	5.56×10^{-4}			
рН	$5.00\pm0.20^{\text{A}}$	3.65 ± 0.10^{B}	$3.82\pm0.10^{\ B}$	$4.29\pm0.10~^{\rm C}$	$4.56\pm0.20~^{\rm C}$			
LA content, mol/L	$0.085\pm0.009^{\text{A}}$	$0.255\pm0.02^{\text{B}}$	$0.209\pm0.019^{\text{C}}$	$0.163\pm0.015^{\rm D}$	$0.139\pm\!0.011^E$			
The average rate of LA synthesis, mol/L/h	$3.5\times10^{-3A,I}$	$11.0\times10^{-3B,I}$	$8.7 imes 10^{-3C,I}$	$6.8\times 10^{-3\text{D},\text{I}}$	$5.8 imes 10^{-3E,I}$			
45.3 kDa, 1.5% DD OCH (glucosamine monomers 248/mol OCH)								
Molar concentration of OCH, mol/L	-	$0.56 imes 10^{-6}$	1.12×10^{-6}	1.68×10^{-6}	2.24×10^{-6}			
Glucosamine monomers content, mol/L	-	$1.39 imes 10^{-4}$	$2.78 imes 10^{-4}$	$4.17 imes 10^{-4}$	5.56×10^{-4}			
pН	$5.00\pm0.20^{\rm A}$	3.78 ± 0.10^{B}	3.87 ± 0.10^B	$4.70\pm0.20^{\rm C}$	$4.82\pm0.10^{\rm C}$			
LA content, mol/L	$0.087\pm0.009^{\text{A}}$	0.228 ± 0.025^B	0.202 ± 0.018^B	$0.123\pm0.011^{\text{C}}$	$0.111\pm0.008^{\rm C}$			
The average rate of LA synthesis, mol/L/h	$3.6 imes 10^{-3 A, I}$	$9.5\times10^{-3B,I}$	$8.4 imes 10^{-3B,I}$	$5.1 imes 10^{-3C,II}$	$4.6\times 10^{-3C,II}$			
350 kDa, 5% DD CH (glucosamine monomers 1850/mol CH)								
Molar concentration of CH, mol/L	-	0.75×10^{-7}	1.50×10^{-7}	$2.25 imes 10^{-7}$	3.00×10^{-7}			
Glucosamine monomers content, mol/L	-	$1.39 imes 10^{-4}$	$2.78 imes 10^{-4}$	$4.17 imes 10^{-4}$	5.56×10^{-4}			
рН	$5.00\pm0.20^{\rm A}$	$4.95\pm0.20^{\text{A}}$	$4.99\pm0.10^{\rm A}$	$4.98\pm0.10^{\rm A}$	$5.01\pm0.20^{\rm A}$			
LA content, mol/L	$0.090\pm0.010^{\rm A}$	$0.089\pm0.009^{\rm A}$	0.082 ± 0.009^B	0.083 ± 0.009^B	$0.075\pm0.007^{\rm C}$			
The average rate of LA synthesis, mol/L/h	$3.8 imes 10^{-3 A, I}$	$3.7 imes 10^{-3 A, II}$	$3.4 imes 10^{-3B,II}$	$3.4 imes 10^{-3B,III}$	$3.1 imes 10^{-3C,III}$			

TABLE 1 Effect of OCHs and CH on LA fermentation parameters after 24 h of L. bulgaricus cultivation.

The values represent the mean \pm SD (n = 3). Means without a common letter within the same column (I–III) and row (A–E) indicate significant difference at p < 0.05.

bacteria cell walls can occur (40, 82, 83). Cell walls are dynamic structures that undergo changes during replication, development, and age (84). This flexibility allows different molecules to pass through the cell wall (85). Consequently, ion immobilization of oligochitosans on the lactic acid bacteria cell wall can lead to a loss of flexibility, porosity, and alteration of its permeability (86, 87). The low permeability of the lactic acid bacteria cell walls leads to reduction of nutrient intake, resulting in a slowdown in the lactic acid synthesis and an increase in the pH of the culture liquid. This process, as depicted in Figures 1-3 and Table 1, depends on the molecular weight of oligochitosans and their concentration. Thus, two mechanisms of their action are realized depending on the concentration of oligochitosans. At low concentrations of oligochitosans, the acceleration of lactic acid synthesis occurs due to intracellular processes in L. bulgaricus. However, increasing the concentration of oligochitosans by four times leads to the realization of their extracellular effects, which slow down the metabolism of lactose and decrease the accumulation of lactic acid by 1.6 times at 25.4 kDa oligochitosan, or by 1.28 times at 45.3 kDa oligochitosan. The implementation of these two mechanisms of oligochitosans activity is multidirectional. As observed in Figure 2 and Table 1, the dominance of the mechanism of extracellular action of oligochitosans increases with an increase in the content of glucosamine monomers.





These observations are confirmed by the study of the effect of various concentrations of chitosan on the cultivation of L. bulgaricus. According to Table 1, the chitosan molecule contained 1850 glucosamine monomers and had an average hydrodynamic radius of 133.2 nm. Due to the large size of the chitosan molecules and the positively charged amino groups of glucosamine residues, this polysaccharide can interact with the cell walls of lactic acid bacteria (23). With an increase in pH and chitosan concentration, the density of its positive charge increases, and it effectively interacts with the negatively charged cell walls of L. bulgaricus. As can be seen from Table 1, at 0.0025% chitosan and oligochitosans, their molar concentrations differ, but the total content of glucosamine monomers remains the same (1.39 \times 10⁻⁴ mol/L). However, at 0.0025% oligochitosans, interaction with L. bulgaricus leads to the realization of the intracellular mechanism of their action, which leads to an acceleration of lactic acid synthesis. At the same time, 0.0025% chitosan causes opposite effects in lactic acid bacteria, which slow down the catabolism of lactose and the lactic acid synthesis due to extracellular interaction. At 0.01% chitosan and oligochitosans, the total content of glucosamine monomers increases by 4 times and reaches 5.56×10^{-4} mol/L.

As a result, the total number of protonated amino groups of these polysaccharides increases, which leads to effective interaction with the lactic acid bacteria cell wall and inhibition of lactose catabolism. Moreover, the biochemical effects of oligochitosans and chitosan are unidirectional, which may indicate the same mechanism of their action due to extracellular effects. Thus, comparative studies of the effect of different concentrations of oligochitosans and chitosan on the cultivation of *L. bulgaricus* revealed two mechanisms of their action. At low concentrations, oligochitosans interact with the plasma membrane, which leads to an acceleration of metabolic processes in *L. bulgaricus*. At 0.01% oligochitosans and chitosan, they interact with the *L. bulgaricus* cell wall and affect the rate of lactic acid synthesis. This observation confirms the previously obtained results (88).

Thus, the results obtained expand the understanding of the mechanism of oligochitosans and chitosan effect on lactic acid bacteria. However, regarding fermented milk products, it is also critical to understand the oligochitosans and chitosan behavior toward lactic acid bacteria not only at cultivation but also at storage. Therefore, the next stage of the experiments represents the effect of these cationic polysaccharides on the metabolic processes of *L. bulgaricus* during long-term storage of fermented dairy products.

3.2 Effect of oligochitosans and chitosan on the metabolic processes of *L. bulgaricus* during long-term storage of fermented milk product

During a 17-day storage period of fermented dairy products containing oligochitosans, a slowdown in lactic acid fermentation was observed alongside a decrease in oligochitosans concentration. The interaction of oligochitosans with the plasma membrane of *Lactobacillus bulgaricus* affects intracellular processes related to lactose catabolism and cell division. Table 2 illustrates that at low concentrations of oligochitosans, the CFU value of *L. bulgaricus* were lower than in the control sample. This decrease in viable cells in the fermented dairy product led to a slowdown in the average rate of lactic acid synthesis after 17 days of storage. Consequently, there was a slight increase in LA content, maintaining low pH values compared to the control sample. At a concentration of 0.01% oligochitosans, where they interact with the *L. bulgaricus* cell wall, the synthesis of LA slows down, but a significant number of viable *L. bulgaricus* cells are preserved.

Based on the data presented in Table 2, it is evident that an increase in the concentration of chitosan leads to a reduction in the intensity of lactic acid fermentation. The most notable deceleration of lactose catabolism via homofermentative lactic acid fermentation occurred in the sample containing 0.01% chitosan. This decrease in lactose catabolism intensity in *L. bulgaricus* could be attributed to the interaction between chitosan and the bacterial cell wall. This external influence may disrupt the permeability of the cell wall to the substrate and the enzyme β -galactosidase (EC 3.2.1.23), responsible for lactose hydrolysis into glucose and galactose (89). Additionally, structural alterations in the cell wall might impede the active transport of lactose hydrolysis products into Lactobacillus cells (90). On the 17th day of storage, the *L. bulgaricus* content in the control sample of fermented milk product was measured at

Index	Concentration of OCHs and CH, %							
	0 (control)	0.0025	0.005	0.0075	0.01			
25.4 kDa, 1% DD OCH (glucosamine monomers 140/mol OCH)								
рН	$3.80\pm0.10^{\text{A}}$	$3.30\pm0.10^{\text{B}}$	$3.66\pm0.10^{\rm A}$	$3.88\pm0.10^{\rm A}$	4.35 ± 0.10^B			
LA content, mol/L	$0.121\pm0.011^{\text{A}}$	$0{,}037\pm0{.}001^{\text{B}}$	$0{,}017\pm0{.}001^{\rm C}$	$0{,}034\pm0{.}002^{\text{B}}$	$0{,}015\pm0{.}002^{\rm C}$			
The average rate of LA synthesis, mol/L/h	$7.3\times10^{-3\text{A},\text{I}}$	$2.2\times10^{-3B,I}$	$1.0 imes 10^{-3C,I}$	$2.0\times 10^{-3B,I}$	$0.9 imes 10^{-3C,I}$			
L. bulgaricus content, CFU/g	$3.73\times 10^{3A_{,\alpha}}$	$2.28 imes 10^{2B,lpha}$	$2.46 imes 10^{2B,lpha}$	$3.89 imes 10^{3A, lpha}$	$4.65 imes 10^{4C,lpha}$			
45.3 kDa, 1.5% DD OCH (glucosamine monomers 248/mol OCH)								
рН	$3.82\pm0.10^{\text{A}}$	3.48 ± 0.10^{B}	$3.61\pm0.10^{\text{B}}$	$3.91\pm0.10^{\rm A}$	$3.96\pm0.10^{\rm A}$			
LA content, mol/L	$0.116\pm0.01^{\rm A}$	0.043 ± 0.001^B	0.048 ± 0.001^B	$0.070\pm0.001^{\rm C}$	$0.063\pm0.001^{\text{C}}$			
The average rate of LA synthesis, mol/L/h	$6.8\times10^{-3\text{A},\text{I}}$	$2.5\times10^{-3B,I}$	$2.8\times 10^{-3B,II}$	$4.1\times10^{-3C,II}$	$3.7\times10^{-3C,II}$			
L. bulgaricus content, CFU/g	$3.71\times 10^{3A,\alpha}$	$2.6 imes 10^{2B,lpha}$	$2.56 imes 10^{2B,\beta}$	$3.49 imes 10^{3A,lpha}$	$4.89 imes 10^{4\mathrm{C},lpha}$			
350 kDa, 5% DD CH (glucosamine monomers 1850/mol CH)								
рН	$3.82\pm0.1^{\rm A}$	$3.85\pm0.1^{\rm A}$	$3.87\pm0.1^{\rm A}$	$3.96\pm0.1^{\rm A}$	$4.04\pm0.1^{\rm A}$			
LA content, mol/L	$0.113\pm0.01^{\rm A}$	$0.102\pm0.009^{\text{B}}$	$0.104\pm0.009^{\text{B}}$	$0.072\pm0.007^{\rm C}$	$0.063\pm0.005^{\rm C}$			
The average rate of LA synthesis, mol/L/h	$7.0 imes 10^{-3 A, I}$	$6.0 imes 10^{-3B,II}$	$6.0 imes 10^{-3B,III}$	$4.2 \times 10^{-3C,II}$	$3.7 imes 10^{-3C,II}$			
L. bulgaricus content, CFU/g	$3.69\times 10^{3A,\alpha}$	$3.65\times10^{3A,\beta}$	$3.59\times 10^{3A,\gamma}$	$5.3\times10^{5B,\beta}$	$6.34 imes 10^{6C,\beta}$			

TABLE 2 Effect of OCHs and CH on pH and metabolism and content of L. bulgaricus in the fermented milk product at 17-days storing.

The values represent the mean \pm SD (n = 3). Means without a common letter within the same column (I–III; α - γ) and row (A–C) indicate significant difference at p < 0.05.

 3.69×10^3 CFU/g. Table 2 illustrates that the inclusion of chitosan resulted in an increase in *L. bulgaricus* content, reaching 3.65×10^3 CFU/g at 0.0025% chitosan, 3.59×10^3 CFU/g at 0.005% chitosan, 5.3×10^5 CFU/g at 0.0075% CH, and 6.34×10^6 CFU/g at 0.01% chitosan. Consequently, the fermented milk product sample containing 0.01% CH contained 1,700 times more *L. bulgaricus* than the control sample.

As can be seen from Tables 1, 2, the effect of oligochitosans and chitosan on the metabolic processes of L. bulgaricus is multidirectional. The acceleration of metabolic processes is realized due to intracellular processes of interaction of oligochitosans with lactic acid bacteria cells. Chitosan interacting with the lactic acid bacteria cell wall slows down metabolic processes. Based on the results obtained, it is logical to assume that due to the differences in the mechanism of action of oligochitosans and chitosan, they should significantly affect the composition and content of secondary metabolites during long-term storage of fermented milk product. In this regard, the relative content of secondary metabolites in the control and experimental samples was determined using GC-MS. A typical GC-MS chromatogram of a fermented milk product extract on the 17th day of storage is shown in Supplementary Figure 1. The chromatogram contains a profile of the metabolites that were analyzed and included in Table 3. Samples of fermented dairy products with 45.3 kDa oligochitosan were chosen due to the higher values of rate of lactic acid synthesis and lactic acid content in comparison with ones with addition of 25.4 kDa oligochitosan.

The analysis of the results shows that 0.0025% of 45.3 kDa oligochitosan accelerates the synthesis of propanol, leading to a 230% increase in its relative content compared to the control sample. Propanol metabolism is closely related to the synthesis

of lactic acid (91), the content of which was 260% higher in this sample than in the control sample. The acceleration of lactic acid synthesis was accompanied by an increase in propanol synthesis, consistent with previously obtained results (92, 93). With an increase in oligochitosan concentration, the process of lactic acid synthesis slows down and the relative propanol content decreases. Another important component of secondary metabolites presented in Table 3 is benzoic acid. It was revealed, that different concentrations of oligochitosan had practically no effect on its synthesis compared to the control sample.

Fermented dairy products also contained a significant number of phenolic compounds: 3,4-dihydro-4,4-dimethyl-2H-1,2-benzisothiazine; 2-furanmethanol; 3,5-dimethyl-4deuteroxymethyl-isoxazole; 4H-pyran-4-one, 3-hydroxy-2-methyl; 2,3-dihydro-3,5-dihydroxy-6-methyl-4H-pyran-4-one. Their total content in the control sample reached 54.68%. In the experimental samples of fermented dairy products with oligochitosan, the content of phenolic compounds ranged from 31.81% at a concentration of 0.0025% 45.3 kDa oligochitosans to 45.65% at a concentration of 0.005% 45.3 kDa oligochitosans. It is important to note that phenolic substances exhibit antioxidant properties and affect the sensory characteristics of fermented dairy products (94, 95).

Fatty acids have a significant role in shaping the sensory attributes of fermented dairy products. Saturated fatty acids made up 1.81% of the control sample, including methyl stearate (0.44%), heptadecanoic acid, methyl ester (0.89%), and heptadecanoic acid, 16-methyl-, and methyl ester (0.49%). Furthermore, the control sample had two unsaturated fatty acids, 8-octadecanoic acid, methyl ester, (E)—(0.66%) and 12-octadecenoic acid, methyl ester (0.63%), that were absent from the other samples. The control

Retention time	Substance, CAS, molecular formula	The relative content, %						
		Control	Concentration of 45.3 kDa OCH, %		Concentration of 350 kDa CH, %		350	
			0.0025	0.0075	0.01	0.0025	0.0075	0.01
5.64	2-Propanol, 000067-63-0, C ₃ H ₈ O	8.35	19.50	2.45	1.63	4.84	2.08	n.d.
5.66	3,4-Dihydro-4,4-dimethyl-2H-1,2- benzisothiazine, 2000114-14-7, C ₄₉ H ₆₆ N ₁₀ O ₁₂ S	2.55	3.95	n.d.	n.d.	n.d.	1.69	n.d.
5.72	2-Furanmethanol, 000098-00-0, $C_5C_6C_2$	n.d.	n.d.	n.d.	n.d.	n.d.	2.48	2.05
9.03	3,5-dimethyl-4-deuteroxymethyl- isoxazole, 2000024-72-1, $C_{12}H_{12}N_2O$	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	2.77
9.90	4H-Pyran-4-one, 3-hydro-xy-2-methyl, 000118-71-8, C ₆ H ₆ O ₃	10.17	0.91	1.21	1.22	0.99	2.35	8.85
10.33	Carbamic acid, butyl-, ethyl ester, 000591-62-8, C 7 $\rm H_{15}NO_2$	4.13	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
10.46	2,3-Dihydro-3,5-dihydroxy-6- methyl-4H-pyran-4-one, 028564-83-2, C ₆ H ₈ O ₄	37.86	26.95	44.44	37.24	37.90	35.23	20.70
10.67	Benzoic acid, 000065-85-0, C ₇ H ₆ O ₂	8.87	9.22	7.60	8.68	12.40	12.77	9.62
13.25	Pentadecanoic acid, 001002-84-2, $C_{15}H_{30}O_2$	n.d.	n.d.	n.d.	n.d.	n.d.	1.31	1.28
19.60	Hexadecanoic acid, methyl ester, 000112-39-0, $C_{\rm 17}H_{\rm 34}O_{\rm 2}$	0.89	0.69	0.73	1.26	0.60	1.14	0.90
19.95	n-Hexadecanoic acid, 000057-10-3, $C_{16}\mathrm{H}_{32}\mathrm{O}_2$	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	7.91
21.28	8-Octadecenoic acid, methyl ester, (E)-, 026528-50-7, $C_{19}H_{36}O_2$	0.66	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
21.29	12-Octa decenoic acid, methyl ester, 056554-46-2, $\rm C_{19}H_{36}O_2$	0.63	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
21.39	Methyl stearate, 000112-61-8, C ₁₉ H ₃₈ O ₂	0.44	0.37	n.d.	n.d.	0.37	n.d.	n.d.
21.49	Heptadecanoic acid, 16-methyl-, methyl ester, 005129-61-3, $C_{19}H_{38}O_2$	0.49	n.d.	0.39	n.d.	n.d.	0.74	0.55
21.64	Cis-11-octadecenoic acid, 000506-17-2, C ₁₈ H ₃₄ O ₂	n.d.	n.d.	n.d.	n.d.	n.d.	5.79	21.58
21.82	Octadecanoic acid, 000057-11-4, C ₁₈ H ₃₆ O ₂	n.d.	n.d.	n.d.	n.d.	1.07	2.09	2.93
29.91	Cholest-5-en-3-ol, 2000683-53-6,	1.34	0.98	1.22	2.67	1.25	2.47	1.35

TABLE 3 Relative content of the main secondary metabolites in extracts from fermented dairy products containing various concentrations of OCHs and CH on the 17th day of storage.

sample of fermented dairy product had a total fatty acid level of 3.11%.

Variations in the content of 45.3 kDa oligochitosan led to notable alterations in the fatty acid synthesis in fermented dairy products. Table 3 shows that samples with an oligochitosan of 45.3 kDa were devoid of unsaturated fatty acids. Heptadecanoic acid, methyl ester ($C_{17}H_{34}O_2$), methyl stearate ($C_{19}H_{38}O_2$), and hepta-decanoic acid, 16-methyl-,

methyl ester ($C_{19}H_{38}O_2$) are examples of saturated fatty acids. Their overall content was lower than that of the control sample (3.11%), although it did grow significantly with increasing OCH concentration, reaching 1.06% at 0.0025% oligochitosan, 1.15% at 0.0075% oligochitosan, and 1.26% at 0.01% oligochitosan. Saturated fatty acids have been demonstrated in the past to be a crucial growth factor for several lactic acid bacteria (9, 96).

Storage time, days	Titrated acidity, $^\circ T$							
	Concentration of 350 kDa CH, %							
	0 (control)	0.0025	0.005	0.0075	0.01			
1	$90\pm3^{A,I}$	$89\pm2^{A,I}$	$82\pm2^{B,I}$	$83\pm1^{B,I}$	$75\pm1^{C,I}$			
3	$117\pm2^{A,II}$	$110\pm3^{B,II}$	$106\pm1^{B,II}$	$104\pm3^{B,C,II}$	$100\pm4^{\text{C,II}}$			
5	$140\pm4^{ m A,III}$	$132\pm3^{B,III}$	$118\pm3^{\text{C,III}}$	$110\pm2^{D,II,III}$	$103\pm2^{\text{E,II}}$			
7	$157\pm3^{A,IV}$	$145\pm2^{\mathrm{B,IV}}$	$132\pm3^{\text{C,IV}}$	$116\pm4^{D,III}$	$114\pm1^{\rm D,III}$			
9	$180\pm2^{A,V}$	$164\pm1^{B,V}$	$148\pm3^{\mathrm{C,V}}$	$128\pm4^{D,IV}$	$126\pm4^{D,\rm IV}$			
11	$185\pm4^{A,V,VI}$	$170\pm3^{B,\rm VI}$	$155\pm2^{D,VI}$	$138\pm2^{\text{D,V}}$	$129\pm4^{E,IV,V}$			
13	$192\pm5^{A,VI,VII}$	$179\pm2^{B,VII}$	$161\pm4^{\text{C,VI,VII}}$	$142\pm5^{D,V,VI}$	$131\pm3^{\text{E,IV,V}}$			
15	$198\pm4^{\text{A,VII,VIII}}$	$185\pm4^{B,VII,VIII}$	$164\pm2^{C,VII}$	$145\pm3^{\mathrm{D,VI}}$	$132\pm4^{E,IV,V}$			
17	$203\pm2^{A,VIII}$	$191\pm5^{B,VIII}$	$186\pm2^{B,VIII}$	$155\pm3^{C,VII}$	$138\pm4^{\mathrm{D,V}}$			

TABLE 4 Effect of 350 kDa CH on titratable acidity of fermented milk product during storage.

The values represent the mean \pm SD (n = 3). Means without a common letter within the same column (I–VIII) and row (A–E) indicate significant difference at p < 0.05.

The main and secondary metabolite composition of the fermented milk product underwent considerable alterations as a result of the 45.3 kDa oligochitosan, according to the data shown in Tables 1–3. This is because the 45.3 kDa oligochitosan's intracellular mechanism of action causes biochemical alterations, which are responsible for the fermented milk product's low pH, high propanol and LA content, minimal saturated fatty acid content, and low *L. bulgaricus* content. Consequently, the consumer qualities of the fermented dairy product with 45.3 kDa oligochitosan are lost. Nevertheless, in light of the effects that have been found, oligochitosans may be utilized as a stimulant for lactic acid synthesis during lactic acid bacteria culture.

Unlike the 45.3 kDa oligochitosan, chitosan's effect on L. bulgaricus occurs via extracellular interaction. According to Tables 2, 3, with an increase in the chitosan concentration, the lactic acid synthesis slows down and the synthesis of propanol is prevented. Fermented dairy product containing 0.0025 and 0.0075% has a 140% higher content of benzoic acid relative to the control sample. It is well-known that a relatively high content of benzoic acid can prevent contamination of fermented milk products by yeast and other microorganisms (97-99). Table 3 shows that chitosan causes additional synthesis of some saturated fatty acids: pentadecanoic acid; hexadecanoic acid, methyl ester; nhexadecanoic acid; methyl stearate; heptadecanoic acid, 16-methyl-, methyl ester; octadecanoic acid; as well as one unsaturated fatty acid-cis-11-octadecenoic acid. At 0.0025% chitosan, fermented milk product contained 2.04% saturated fatty acids and did not contain unsaturated acids. However, 0.0075% chitosan caused an increase in the content of saturated fatty acids to 5.28% and unsaturated by 5.79%. Fermented milk product with 0.01% chitosan had the highest content of fatty acids after 17 days of storage: 13.57% saturated fatty acids and 21.58% unsaturated fatty acid (cis-11-octadecanoic acid). As presented in Table 2, with an increase in the chitosan concentration, the content of L. bulgaricus increased as well. At the same time, according to Table 3, there was a significant increase in the content of fatty acids, which confirms their importance for L. bulgaricus growth. Thus, chitosan in interaction with L. bulgaricus causes metabolic shifts in the synthesis of secondary metabolites. As a result, lactic acid bacteria produce antifungal metabolites such as organic acids, phenolic compounds and a wide range of carboxylic acids and their esters (100, 101). These substances determine the high sensory characteristics of fermented milk product (102).

Furthermore, the shelf life of a fermented dairy product is a crucial indicator. Titratable acidity is seen to rise with an increase in storage duration. The titratable acidity of fermented dairy products, which should not be more than $140^{\circ}T$ (49), determines their maximum shelf life. This value was attained in the control sample on the 5th day of storage and surpassed by 145% on the 17th day of storage (Table 4).

Experimental samples containing 0.0025, 0.005, 0.0075, and 0.01% chitosan reached the required titratable acidity levels on the 7, 9, 13, and 17th days of storage, respectively. Surprisingly, the optimal concentration of 0.01% chitosan maintained the titratable acidity of the fermented dairy product at $138^{\circ}T$ on 17th day of storage. In general, this concentration increased the shelf life of the fermented dairy product by three times compared to the control sample. Additionally, lactic acid fermentation process was slowed down, while maintaining a high content of lactic acid bacteria and a low concentration of lactic acid.

An important sensory indicator of fermented milk product during storage is the clot density of the clot (103, 104). On the 17th day of storage, the viscosity of the control sample was 168 MPa×s, while the viscosity of the sample with 0.01% chitosan reached to 176 MPa×s. Thus, in the presence of chitosan, the liquefaction of the clot slowed down during storage. This is attributed to the fact that the experimental sample contained less lactic acid and maintained a relatively high pH value. Consequently, hydrolysis of milk proteins and destruction of casein micelles does not occur in experimental samples, as confirmed by SDS-electrophoretic examination (Supplementary Figure 2).

Further, in order to clarify the effect of chitosan on the form of LAB and confirm the absence of extraneous microflora in the samples, fixed preparations were obtained and studied after 17 days of storage. The results of microscopic studies are depicted in Figure 4.

Analysis of the microscopic preparations showed that interaction of *L. bulgaricus* with chitosan at concentrations of 0.0075 and 0.01% did not lead to a change in their rod shape. There were no foreign microflora in the samples examined. This



may be due to the high content of secondary metabolites, such as benzoic acid, which prevent contamination of the product by other microorganisms (105).

Thus, the effective interaction of *L. bulgaricus* with chitosan is due to the presence in its 1850/mol structure of glucosamine monomers which, in an acidic environment, perform a multipoint ionic interaction with negatively charged teichoic acid molecules of the lactic acid bacteria cell wall. With increasing chitosan concentration, the process of lactose catabolism by *L. bulgaricus* and the accumulation of the lactic acid slows down. As a result of this process, the shelf life of the fermented milk product is significantly increased. At the same time 0.01% chitosan provided a high level of lactic acid bacteria and improved sensory characteristics of the product during 17 days storage. The concentrations of chitosan used did not give astringent taste to the fermented dairy product.

4 Conclusions

This study presents the results of a comparative study of the effect of oligochitosans and chitosan on the lactic acid fermentation process during cultivation of L. bulgaricus and longterm storage. The foundation of this study was the potential utilization of the ionic interaction between cationic polysaccharides and negatively charged L. bulgaricus cells to regulate lactic acid synthesis. Depending on the molecular weight, concentration, and molecules size of oligochitosans and chitosan effect on lactic acid synthesis due to intracellular or extracellular interactions during cultivation of L. bulgaricus and long-term storage of the fermented dairy product. When L. bulgaricus was cultured in the presence of 0.0025% 25.4 kDa oligochitosan, 45.3 kDa oligochitosan, and 350 kDa chitosan, their molar concentrations were 0.99×10^{-6} , $0.56\,\times\,10^{-6},$ and $0.75\,\times\,10^{-7}$ mol/L, respectively. At the same time, the total content of glucosamine monomers in these samples was the same and amounted to 1.39×10^{-4} mol/L. After 24 h of cultivation, the lactic acid content in the fermented milk samples was 0.087 mol/L in the control sample, 0.255 mol/L in sample with 25.4 kDa oligochitosan, 0.228 mol/L in sample with 45.3 kDa oligochitosan, and 0.089 mol/L in sample with 350 kDa chitosan. At equal glucosamine monomers content, oligochitosans accelerate the synthesis of lactic acid, while chitosan has no such an effect compared to the control sample. These results indicate different mechanisms of action of oligochitosans and chitosan on the synthesis of lactic acid. With increasing concentrations of oligochitosans, a slowdown in the synthesis of lactic acid is observed. At the same time, its content remained higher than in the control sample. During long-term storage (17 days) of dairy product fermented with oligochitosan, further accumulation of lactic acid, decrease in pH and *L. bulgaricus* content occurred. The results obtained confirm that oligochitosans can be used as stimulators of lactic acid synthesis based on lactose containing substrates industrial fermentation using *L. bulgaricus* starter cultures.

Chitosan with molecular weight 350 kDa had the opposite effect on metabolic processes in fermented dairy product. With increasing chitosan concentration, lactic acid and propanol synthesis slowed down and the content of saturated and unsaturated fatty acids significantly increased. As a result, the titratable acidity decreased and the L. bulgaricus content increased. The result was an increase in the shelf life of the fermented dairy product. Chitosan provided low acidity of the product, high content of biologically active substances, increased L. bulgaricus content and better sensory characteristics compared to samples with addition of oligochitosans. The concentrations of chitosan used did not give astringent taste to the fermented dairy product. Further prospects are related to the study of the effect of 350 kDa chitosan on the technological processes of lactic acid bacteria various strains cultivation used in the production of commercial fermented dairy products.

Data availability statement

The original contributions presented in the study are included in the article/Supplementary material, further inquiries can be directed to the corresponding authors.

Author contributions

VK: Conceptualization, Data curation, Methodology, Project administration, Supervision, Writing – original draft. TH: Conceptualization, Formal analysis, Investigation,

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Methodology, Resources, Validation, Visualization, Writing – original draft. AY: Methodology, Writing – original draft. MS: Investigation, Methodology, Validation, Writing – original draft. LA: Investigation, Writing – review & editing. IE: Conceptualization, Data curation, Resources, Writing – review & editing. AL: Conceptualization, Funding acquisition, Investigation, Methodology, Project administration, Writing – original draft, Writing – review & editing. VT: Methodology, Resources, Writing – original draft. AN: Validation, Visualization, Writing – original draft. FA: Validation, Visualization, Writing – original draft. FA: Validation, Visualization, Writing – review & editing. AA-F: Formal analysis, Writing – review & editing. MS: Writing – review & editing.

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Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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Supplementary material

The Supplementary Material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/fnut.2024. 1441355/full#supplementary-material

A typical GC-MS chromatogram of an extract from fermented dairy product on the 17th day of storage.

SUPPLEMENTARY FIGURE 2

SDS-electrophoregram of proteins of fermented dairy product on the 17th day of storage: control sample (5), 0.01% CH (4), 0.0075% CH (3), 0.0025% CH (2), original milk (1).

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