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RECEIVED 29 March 2024 ACCEPTED 30 May 2024 PUBLISHED 11 June 2024

CITATION

Kim Y, Kim HW, Sung J and Kim Y (2024) Optimal extraction conditions and quantification of lignan phytoestrogens in cereal grains using targeted LC-MS/MS. *Front. Nutr.* 11:1409309. doi: 10.3389/fnut.2024.1409309

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Optimal extraction conditions and quantification of lignan phytoestrogens in cereal grains using targeted LC-MS/MS

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Lignans are phytoestrogens found in various forms such as glycosides, esterlinked oligomers, and aglycones in a variety of foods, including soy products, legumes, grains, nuts, vegetables, and fruits. This study aimed to optimize the extraction of lignans from cereal grains using response surface methodology (RSM). Lignans, including secoisolariciresinol (Seco), matairesinol (Mat), pinoresinol (Pin), lariciresinol (Lar), and syringaresinol (Syr), were quantified using high-performance liquid chromatography-tandem mass spectrometry. A Box-Behnken design was employed to determine the optimal values for three extraction parameters: temperature (X1: 20°C-60°C), methanol concentration (X2: 60%-100%), and extraction time (X3: 30-90 min). The highest lignan contents were obtained at X1 = 44.24°C, X2 = 84.64%, and X3 = 53.63 min. To apply these experimental conditions to the actual experiment, the optimal conditions were slightly adjusted to X1 = 40°C, X2 = 80%, and X3 = 60 min. The predicted results closely matched the experimental results obtained using the modified optimal extraction conditions. The highest lignan content found in barley sprouts (85.930 µg/100 g), however, most grains exhibited relatively low concentrations of lignans. These findings provide valuable insights into the lignan content of grains and contribute to the generation of reliable data in this field.

KEYWORDS

lignans, LC-MS/MS, grains, RSM, optimal extraction

1 Introduction

Lignans are phenylpropanoid dimers, where the phenylpropane units are linked by the central carbon of their side chains (1). Lignans, including secoisolariciresinol (Seco), matairesinol (Mat), pinoresinol (Pin), medioresinol (Med), lariciresinol (Lar), syringaresinol (Syr), sesamin (Ses), 7'-hydroxymatairesinol (HMR), and isolariciresinol, are widely distributed in the bark, bulbs, leaves, seeds, and stems of plants (2, 3). The compounds occur mainly in the glycoside, ester-linked oligomer, and aglycone forms (4, 5). Lignans are known as potential antioxidants along with phenolic compounds such as flavonoids and phenolic acids (6, 7). Some of the plant lignans including secoisolariciresinol diglucoside, Mat, Pin, and Lar are deglycosylated, dehydroxylated, demethylated, and converted by intestinal bacteria to the mammalian lignans enterodiol and enterolactone and then absorbed through the colon (8–10). Moreover, lignans, referred to as phytoestrogens, bind to estrogen receptors and may exhibit both estrogenic and anti-estrogenic

effects, affecting conditions such as menopause, cardiovascular disease, and cancer (11, 12). Recent studies demonstrated the bioaccessibility and bioavailability of plant lignans during digestion and fermentation in foodstuffs (13, 14). Grain such as wheat, corn, rice, oats, millet, barley, spelt, and rye, are commonly consumed worldwide (15, 16). Various studies have investigated the absorption and role of dietary lignans from grains (17-19). Moreover, Nørskov et al. (20) have investigated the LC-MS/MS method of the free and bound form of plant lignans and enterolignans after consumption of cereal-based diets. Previous studies have shown the lignan content in seeds, vegetables, and fruits (21). Smeds et al. (22) evaluated that lignans, including HMR, Seco, Mat, Lar, Pin, Med, and Syr, were quantified in wheat, oat, and rye. Thompson et al. (23) reported the isoflavone, lignan, and coumestan contents in various foods (e.g., soy products, legumes, nuts, grains, vegetables, and fruits) commonly consumed in Canadian diets. Lignan contents in seeds, such as sesame seeds and flaxseeds have been reported (24-26), but research on lignan contents in cereal grain and their products has been limited.

Various extraction methods of lignans in plants have been reported, such as ultrasound-assisted extraction, Soxhlet extraction, reflux extraction, and microwave-assisted extraction, using organic solvents including methanol, ethanol, and ethyl acetate (27-29). Additionally, hydrolysis methods have been studied to break the glycoside linkages of lignans, such as enzymatic, acidic, and alkaline hydrolysis (24, 30, 31). Liquid chromatography combined with mass spectrometry is preferred for metabolite identification and quantitative research in foods and drugs due to its sensitivity, fast analysis times, simplicity of sample preparation, and mass accuracy (32-34). López et al. (35) reported that Pin, hydroxy-Pin, and acetoxy-Pin, found in olive oil, were quantitatively analyzed by high-performance liquid chromatography mass spectrometry (HPLC-MS/MS). Moreover, Kuhnle et al. (36) reported that Seco and Mat were simultaneously analyzed in vegetables using HPLC-MS/MS. Response Surface Methodology (RSM) is a mathematical and statistical technique for determining the optimal values to describe the relationship between independent variables and responses (37). Currently, RSM is applied in the food and pharmacological industries to optimize compound extraction from various foods (38, 39). Zhang et al. (40) used RSM to optimize the extraction of lignans from flaxseed, and Zhao et al. (41) used RSM to optimize schizandrin and schisandrol B extraction from Schisandra chinensis using accelerated solvent extraction. A previous study evaluated the influence of antioxidants, including polyphenols and vitamin B2, in ultrasound-assisted extraction using the response surface method with Box-Behnken design (42). However, only a limited number of studies have reported on extraction methods and simultaneous analysis of lignans such as Lar, Mat, Pin, Seco, and Syr found in cereal grain. This study aimed to determine the content of five lignans (Figure 1; Lar, Mat, Pin, Seco, and Syr) in cereal grains and their products, and to assess the optimal extraction conditions of lignans using RSM.

2 Materials and methods

2.1 Standards and reagents

All the reagents were of analytical and HPLC grade. Syr (\geq 98%) was obtained from ChemFaces (Wuhan, China). Seco (\geq 95%), Mat (\geq 85%), Lar (\geq 95%), and Pin (\geq 95%) were purchased from Sigma-Aldrich (St. Louis, Mo, United States). Acetonitrile (ACN) was purchased from Merck (Darmstadt, Germany). Methanol was from

Honeywell Burdick & Jackson (Muskegon, MI, USA). Water was deionized from a Milli-Q system (Millipore, Bedford, United States).

2.2 Samples

The oat (*Avena sativa* L.) powder used in this study was commercial product from Natural Hill (Youngin, Korea). A total of 50 grain products were provided from Rural Development Administration (Jeonju, Korea) in 2022 or 2023. All samples were freeze-dried to achieve less than 3% water content. All samples were stored at -80° C until further analysis.

2.3 Parameters on the extraction of lignan

To select the parameters on the extraction of lignan, we select.

2.3.1 Extraction temperature

An amount of 0.1 g of oat powder was weighed in a 2-mL centrifuge tube, and 1 mL of 80% methanol was added and vortexed. Ultrasound sonication was performed at 0°C, 20°C, 40°C, 60°C, and 80°C for 60 min. The extract was then cooled and centrifuged at 22,250×g at 4°C for 10 min to yield a clear supernatant. The supernatant was filtered through a 0.2 μ m nylon filter (Whatman Inc., Maidstone, United Kingdom) into a vial for HPLC-MS/MS analysis.

2.3.2 Solvent concentration

For the extraction, 0.1 g of samples was weighed in a 2-mL microcentrifuge tube, and six different concentrations (water, 20%, 40%, 60%, 80%, and 100% methanol) of methanol were prepared using an ultrasound sonicator at 40°C for 60 min. The sample was then centrifuged at $22,250 \times \text{g}$ for 10 min, and the supernatant was filtered through a $0.2 \,\mu\text{m}$ syringe filter for HPLC-MS/MS analysis.

2.3.3 Extraction time

The sample (0.1g) was weighed and transferred to a 2-mL microcentrifuge tube. Then, 1 mL of 80% methanol was added and vortexed for 3 min. Subsequently, the extraction was carried out at 40°C for four different durations (30, 60, 90, and 120 min) using an ultrasonic extractor. Afterward, the extract was filtered through a 0.2 μ m nylon syringe filter following centrifugation (22,250 × g, 4°C, 10 min) before HPLC–MS/MS analysis.

2.4 Experimental design and statistical model

The RSM was employed to assess the impacts of extraction parameters and optimize conditions for multiple responses. The Box– Behnken design (BBD) was used to determine the extraction parameters for lignans from oats. The experimental design comprised 15 experimental points with three levels (-1, 0, 1) of each factor. The independent variables selected for optimizing lignan extraction were extraction temperature (°C) (X1), methanol concentration (% (v/v)) (X2), and extraction time (min) (X3). The natural and coded values of the factors used in the experimental design are presented in Table 1. The experimental data were fitted to a second-order polynomial model to determine the regression coefficients. The optimum



TABLE 1 Coded and actual levels of three variables in oat.

Independent	Coded levels					
variables	-1	0	1			
Temperature (°C) (X_1)	20	40	60			
Methanol concentration (%) (X ₂)	60	80	100			
Time (min) (X ₃)	30	60	90			

conditions were conducted by 3D response surface plots with the relationship between the independent variables and responses. The second-order polynomial model expressed total lignan contents (μ g/100 g) using the following equation:

$$Y = \beta_0 + \sum_{i=1}^{2} \beta_i X_i + \sum_{i=1}^{2} \beta_{ii} X_i^2 + \sum_{i=1}^{2} \sum_{j=i+1}^{2} \beta_{ij} X_i X_j$$

where Y represents the response variable (total lignan contents $\mu g/100 \text{ g}$), β_0 is the intercept, β_i is the linear regression coefficient, β_{ii} and β_{ij} are the quadratic and interaction terms, respectively. X_i and X_j are the actual levels of the independent variables. To assess the predictive performance of the model on the response variable, an analysis of variance (ANOVA) was conducted with a confidence level of 95%. The regression coefficient (\mathbb{R}^2) and the *p*-value of the lack of fit were calculated using Minitab ver. 14 software (Minitab Inc., State College, PA, United States).

2.5 HPLC-MS/MS quantitative analysis of total lignans

The HPLC-MS/MS conditions for total lignans were followed as described in a previous study (43). HPLC-MS/MS analysis was

performed using an Agilent 1,260 Infinity II HPLC (Agilent Technologies, California, United States) coupled with an AB Sciex Triple Quad 4,500 LC-MS/MS system (AB Sciex, Framingham, MA, United States) equipped with an electrospray ionization source operating in negative mode. Samples were separated on an Agilent Poroshell C18 column (2.1×50 mm, 1.9 µm; Agilent Technologies, United States). The mobile phase A consisted of water, and mobile phase B consisted of acetonitrile, with an injection volume of $2\,\mu\text{L}.$ The separation gradient elution program was as follows: 0-2 min, 15% B; 2-4 min, 50% B; 4-4.1 min, 50% B; 4.1-15 min, 15% B. The flow rate was maintained at 0.4 mL/min, and the column temperature was set at 30°C. The Turbo-V source parameters were configured as follows: curtain gas, 30 psi; collision gas, 9 psi; ion source gas 1, 80 psi; ion source gas 2, 40 psi; ionspray voltage, -4,500 V; ion source temperature, 450°C. The multiple reaction monitoring (MRM) conditions are listed in Table 2. Data acquisition and processing were performed using the AB SCIEX Analyst 1.7.1 Software and MultiQuant (ver 3.0.3).

2.6 Statistical analysis

Statistical analysis of data was carried out using SAS 9.4 software (Statistical Analysis System, SAS Institute Inc., Cary, NC, United States). The data were expressed as mean \pm standard deviation of three experiments. The data were analyzed with one-way analysis of variance followed by Duncan's multiple range test (p < 0.05).

3 Results and discussion

3.1 Effects of the parameters on the extraction of lignan

The effects of various parameters (extraction temperature, methanol concentration, and extraction time) on the contents of

lignans in oats are presented in Table 3. Among the extraction temperatures, the total lignan content of oats ranged from 37.714µg/100g to 56.326µg/100g, with Syr showing the highest content ranging from 24.454 to 39.592 µg/100 g, while Mat and Seco were not detected. Additionally, the total lignan contents of oats were lowest at 0°C, increased from 20°C to 60°C, and then decreased at 80°C. The total lignan content of oats, using different methanol concentrations, ranged from 18.166 to 56.326 µg/100 g. Among these conditions, the total lignan content was elevated with the use of 80% methanol as the extraction solvent. However, Mat and Seco were not detected. The content of Syr was 18.166 µg/100 g in 0% methanol, while Lar, Mat, Pin, and Seco were not detected. Moreover, Syr was the most abundant in 80% methanol $(39.592 \,\mu g/100 \,g)$, followed by Pin and Lar at 11.252 and 5.482 µg/100 g, respectively. Previous studies reported highly variable ultrasonic extraction times for lignans, ranging from 5 min to 4h depending on the sample type (25, 44–46). Therefore, this study used extraction times from 30 to 120 min to investigate the effect of extraction time on the lignan content of oats. In this study, most of the lignans showed higher content at 60 min, with a decrease observed at 90 min. The highest total lignan content in oats was observed after 60 min of extraction, with values of 56.326 µg/100 g. However, a significant decrease was observed at 90 and 120 min. Among oats, Syr was the most abundant with 39.592 µg/100 g, while Mat and Seco were not detected. The lignan content depends on the variety of the plant and/or the geographical location of oats. Durazzo et al. (47) studied that Seco and Mat were not detected in oats, which aligns with our results. However, the lignan contents of various oat varieties, with Seco and Mat ranging from 6 to $19\mu g/100 g$ and 0 to $104\mu g/100 g$, respectively (25). The thermal stability of lignans is known to depend on the structure of the compound and its interactions with other compounds in the plant matrix (44, 48, 49). Gerstenmeyer et al. (49) evaluated that the Lar content in sesame seeds and wheat was stable at 100°C but decreased rapidly above 200°C. These findings indicate that the effect of heat treatment on lignan concentration depends on the plant species, cultivation, extraction, and processing methods (50, 51). Therefore, the results of this study suggest that the differences in oat lignan content depending on temperature are due to differences in oat

TABLE 2	MRM	transitions	and	optimized	parameters	for lignans.	

Analytes	RT (min)	Q1 (m/z)	Q3 (m/z)	DP (V)	EP (V)	CE (V)	CEP (V)
Lar	3.39	359.100	329.0	-90	-10	-17	-9
Mat	4.11	356.892	83.0	-70	-10	-50	-7
Pin	3.86	357.000	150.9	-95	-10	-24	-9
Seco	3.25	360.892	164.9	-95	-10	-34	-9
Syr	3.78	416.956	181.0	-80	-10	-26	-9

RT, Retention time; Q1, Parent ions; Q3, Product ions; DP, Declustering potential; EP, Entrance potential; CE, Collision energy; CEP, Cell exit potential; Lar, Lariciresinol; Mat, Matairesinol; Pin, Pinoresinol; Seco, Secoisolariciresinol; Syr, Syringaresinol.

TABLE 3	The effect	of different	parameters	of lignan	contents in oats.
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Parameters		Lignan contents (µg/100 g)								
		Lar	Mat	Pin	Seco	Syr	Total lignan			
	0°C	$3.471 \pm 0.175^{\circ}$	ND	$9.394 \pm 0.293^{\rm b}$	ND	$24.848 \pm 0.326^{\circ}$	$37.714 \pm 0.208^{\rm d}$			
_	20°C	4.995 ± 0.380^{ab}	ND	11.376 ± 0.238^{a}	ND	$25.156 \pm 0.873^{\circ}$	$41.527 \pm 0.255^{\circ}$			
Extraction	40°C	5.482 ± 0.046^{a}	ND	11.252 ± 0.249^{a}	ND	39.592 ± 0.863^{a}	56.326 ± 1.157^{a}			
temperature	60°C	4.788 ± 0.126^{b}	ND	$9.702 \pm 0.258^{\rm b}$	ND	$33.772 \pm 1.710^{\rm b}$	48.262 ± 2.093^{b}			
	80°C	4.730 ± 0.208^{b}	ND	$9.811\pm0.189^{\mathrm{b}}$	ND	$24.454 \pm 0.760^{\circ}$	38.995 ± 1.157^{cd}			
	0%	ND ¹	ND	ND	ND	$18.166 \pm 0.209^{\rm d}$	$18.166 \pm 0.209^{\rm d}$			
	20%	$1.869 \pm 0.007^{\rm d}$	ND	$9.670 \pm 0.805^{\rm b}$	ND	$23.380 \pm 0.481^{\circ}$	$34.919 \pm 1.293^{\circ}$			
Methanol	40%	$2.952 \pm 0.326^{\circ}$	ND	$9.264 \pm 0.235^{\rm bc}$	ND	$22.394 \pm 0.004^{\circ}$	$34.610 \pm 0.088^{\circ}$			
concentration	60%	4.090 ± 0.253^{b}	ND	$8.779 \pm 0.157^{\rm bc}$	ND	26.516 ± 0.313^{b}	$39.385 \pm 0.722^{\rm b}$			
	80%	$5.482 \pm 0.046^{\rm a}$	ND	11.252 ± 0.249^{a}	ND	39.592 ± 0.863^{a}	56.326 ± 1.157^{a}			
	100%	$4.037 \pm 0.071^{\rm b}$	ND	$8.587 \pm 0.114^{\circ}$	ND	$26.168 \pm 1.054^{\rm b}$	$38.791 \pm 1.011^{\rm b}$			
	30 min	$3.963 \pm 0.012^{\circ}$	ND	$9.060 \pm 0.286^{\circ}$	ND	$25.031 \pm 1.020^{\rm b}$	$38.054 \pm 0.722^{\circ}$			
	60 min	$5.482 \pm 0.046^{\rm a}$	ND	11.252 ± 0.249^{a}	ND	$39.592 \pm 0.863^{\rm a}$	56.326 ± 1.157^{a}			
Extraction time	90 min	$4.710 \pm 0.135^{\rm b}$	ND	$10.491 \pm 0.470^{\rm ab}$	ND	$25.380\pm0.059^{\text{b}}$	$40.581 \pm 0.393^{\rm b}$			
	120 min	$4.492 \pm 0.484^{\rm bc}$	ND	$10.168 \pm 0.081^{\rm b}$	ND	$22.154 \pm 0.036^{\circ}$	$36.814 \pm 0.600^{\circ}$			

All values are mean ± SD. *-dMeans with different letters in the same column of each sample are significantly different by Duncan's multiple range test at *p* < 0.05. Lar, Lariciresinol; Mat, Matairesinol; Pin, Pinoresinol; Seco, Secoisolariciresinol; Syr, Syringaresinol.

variety, growing conditions, and extraction methods. Most lignans in grains and seeds are commonly extracted using polar organic solvents (methanol, ethanol, acetone), while less polar lignans can be extracted using non-polar organic solvents such as n-hexane, dichloromethane, and chloroform (5, 18, 52). Additionally, adding 5%-10% water in organic solvents promotes solvent penetration into the plant matrix and facilitates the extraction of polar lignans (5). A previous study examined Seco in flaxseed hulls at various ethanol and methanol concentrations (53). The results showed higher contents of Seco in methanol compared to ethanol, with the highest extraction observed at 70 and 80% methanol concentrations, while extraction yields decreased with ethanol concentrations above 80%. The lignans examined in this study were generally not detected or showed low quantity when extracted with water; however, lignan content increased at 80% methanol compared to 100% methanol. These findings are consistent with previous research, indicating that the five lignans analyzed in this study are more easily extracted in organic solvents than water. Additionally, adding 30% water to methanol improves the polarity of the solvent, facilitating the extraction of polar lignans. Ultrasound-assisted extraction is an approach to create cavitation bubbles within plant cell walls to facilitate solvent penetration and increase the release of various organic compounds (54, 55). The efficiency of the method can be affected by the extraction time, solvent-to-sample ratio, and extraction temperature, which should be taken into consideration (56-58). Previous studies have shown that the lignan secoisolariciresinol diglucoside and phenolic compounds, such as ferulic acid glucoside and caffeic acid glucoside in flaxseed increased with extraction time, specifically more than a 2-fold increase at 60 min compared to 15 min extraction (59). Guo et al. (28) conducted a study to optimize ultrasonic extraction conditions for

three lignans ((-)-fargesin, Ses, L-asarinin) from Zanthoxylum armatum roots using RSM. The highest lignan content was observed at 55° C with an extraction time of 40 min, but prolonged sonication time led to its degradation. This study showed that lignan content was highest after 60 min of ultrasonic extraction, with the content decreasing after 90 min. This indicates that the lignans present in the sample matrix were extracted within 60 min, but longer extraction time resulted in ultrasonic degradation of released lignans.

3.2 Optimization of total lignan extraction conditions

Fifteen experiments were designed with independent variables of temperature, methanol concentration, and extraction time coded in three levels. The results for the lignan content of oats and the response of the surface are presented in Table 4 and Figure 2, respectively. The lignan content of oats ranged from 38.777 to 60.709 µg/100 g. Among the five lignans, Syr had the highest content, ranging from 23.003 to $41.812 \,\mu\text{g}/100 \,\text{g}$. Pin and Lar were present in the range of 10.494 to 13.025 µg/100 g and 4.041 to 6.346 µg/100 g, respectively, while Mat and Pin were not detected. The experimental results for oats, designed by the BBD, were analyzed for significance and adequacy using ANOVA and are presented in Table 5. The *p*-values of the linear, quadratic, and interaction terms of the model ranged from 0.031 to 0.553, with the term X_2 acting as the significant term. The linear term, methanol concentration, has a p-value of 0.031, indicating a significant effect on the response variable. However, the extraction temperature (X_1) and extraction time (X_3) demonstrated p > 0.05 and were not influenced by the response variables. Similarly, the quadratic term was

TABLE 4 Response surface design and experimental data for the total lignan content in oat.

Run	Variables				Responses in oat (µg/100 g)						
	Temperature X ₁ (°C)	MeOH concentration X ₂ (%)	Time X₃ (min)	Lar	Mat	Pin	Seco	Syr	Total lignan		
1	−1 (20°C)	-1 (60%)	0 (60 min)	5.106	0	10.872	0	26.485	42.464		
2	1 (60°C)	-1 (60%)	0 (60 min)	4.733	0	10.610	0	28.044	43.386		
3	−1 (20°C)	1 (100%)	0 (60 min)	4.041	0	10.494	0	27.602	42.137		
4	1 (60°C)	1 (100%)	0 (60 min)	5.449	0	11.068	0	36.116	52.632		
5	−1 (20°C)	0 (80%)	-1 (30 min)	5.416	0	10.859	0	32.977	49.253		
6	1 (60°C)	0 (80%)	-1 (30 min)	5.902	0	10.917	0	39.785	56.603		
7	−1 (20°C)	0 (80%)	1 (90 min)	6.012	0	11.696	0	30.937	48.645		
8	1 (60°C)	0 (80%)	1 (90 min)	4.665	0	10.632	0	25.408	40.705		
9	0 (40°C)	-1 (60%)	-1 (30 min)	4.700	0	11.074	0	23.003	38.777		
10	0 (40°C)	1 (100%)	-1 (30 min)	5.324	0	13.025	0	31.409	49.759		
11	0 (40°C)	-1 (60%)	1 (90 min)	4.817	0	11.922	0	30.984	47.722		
12	0 (40°C)	1 (100%)	1 (90 min)	5.364	0	11.563	0	35.385	52.312		
13	0 (40°C)	0 (80%)	0 (60 min)	5.854	0	11.526	0	40.444	57.824		
14	0 (40°C)	0 (80%)	0 (60 min)	6.346	0	12.220	0	41.812	60.378		
15	0 (40°C)	0 (80%)	0 (60 min)	6.608	0	11.703	0	39.203	60.709		

Lar, Lariciresinol; Mat, Matairesinol; Pin, Pinoresinol; Seco, Secoisolariciresinol; Syr, Syringaresinol.



found to be a significant variable with a *p*-value of 0.027 in term X_2 , whereas X_1 and X_3 did not have a significant effect on the response variable (*p* > 0.05). Furthermore, the interaction term did not show significance in all terms (*p* > 0.05). The results indicated that methanol concentration has a significant effect on total lignan content, whereas extraction time and extraction temperature do not appear to be significant variables. The *p*-value for lack of fit was 0.059, indicating no significant difference, suggesting that the response surface model adequately describes the variation in lignan content. Furthermore, the R^2 value for the response surface model was 82.39%, indicating a good fit.

The regression equation obtained for total lignan is expressed as equation given below:

$$\begin{aligned} \mathbf{Y}_{\text{Total lignan}} &= -135.465 + 1.254 \mathbf{X}_1 + 3.302 \mathbf{X}_2 + 1.037 \mathbf{X}_3 \\ &\quad -0.016 \boldsymbol{X}_1^2 - 0.020 \boldsymbol{X}_2^2 - 0.005 \boldsymbol{X}_3^2 + 0.006 \mathbf{X}_1 \mathbf{X}_2 \\ &\quad -0.006 \mathbf{X}_1 \mathbf{X}_3 - 0.003 \mathbf{X}_2 \mathbf{X}_3 \end{aligned}$$

Table 6 displays the optimized extraction conditions for oat lignans, determined by setting the lignan content to its maximum value. The predicted optimal extraction conditions for oat lignans are a temperature of 44.24°C, a methanol concentration of 84.64%, and an extraction time of 53.53 min. However, in this study, the optimal extraction conditions were modified to include a temperature of 40°C, a methanol concentration of 84%, and an extraction time of 60 min. These parameters were selected for experimental convenience in evaluating the total lignan content of oats. The results showed that the lignan content of the oats was 59.564 µg/100 g, which was similar to the predicted value of 60.213 µg/100 g. Several factors, including solvent-to-solid ratio, solvent concentration, extraction time, and temperature, may significantly impact the extraction efficiency of phenolic compounds from plant sources (60-62). Liyana-Pathirana and Shahidi (63) reported that the concentration of solvent significantly influenced the total antioxidant activity of wheat extracts using RSM. Fombang et al. (64) also reported that solvent concentration had the greatest effect on the total phenol content of Moringa oleifera Lam. leaves compared to extraction temperature and extraction time. Additionally, several studies have demonstrated that solvent concentration is an important factor in the extraction of phenolic compounds from various natural products (65-67), consistent with our findings.

3.3 Lignan contents of cereal grains

To validate the LC-MS/MS analytical method of lignans, the linearity, limits of detection (LOD), limit of quantitation (LOQ), and intra-day and inter-day precision were evaluated. The linearity was evaluated with six concentration levels (15.625-500 ng/mL) for each analyte, showing acceptable correlation coefficients (R2>0.999) (Supplementary Figure 1). The LOD is in the range of $0.041-0.877 \,\mu\text{g}/100 \,\text{g}$, and the LOQ is in the range of 0.118-1.831 µg/100 g (Supplementary Table 1). The intra-day and inter-day precision of the lignans in oats ranged from 0.075%-3.480% and 0.749%-13.735% RSD, respectively (Supplementary Table 2). In this study, a total 53 samples of grains and their products using optimal lignan extraction conditions based on RSM results was analyzed (Table 7). The total lignan contents of the grain samples ranged from 0 to $85.930 \,\mu\text{g}/100 \,\text{g}$. Most of the samples exhibited low lignan content, except for Syr. Barley sprout had the highest total lignan content among the samples with 85.930 µg/100 g. The most abundant lignan in barley sprout was Syr with 41.713 µg/100 g. Lar and Pin were present at 36.491 and 7.726 µg/100 g, respectively, while Mat and Seco were not detected. In wheat, all lignans except Syr were not detected. In addition, the total lignan content of rice increased after cooking

Source	DF	Seq SS	Adj SS	Adj MS	F value	<i>p</i> -value
Model	9	592.302	592.302	65.811	2.60	0.153
X ₁ ¹⁾	1	14.658	66.464	66.464	2.63	0.166
X ₂ ²⁾	1	74.971	224.108	224.108	8.85	0.031
X ₃ ³⁾	1	3.137	102.309	102.309	4.04	0.101
x ₁ ²	1	113.727	151.794	151.794	6.00	0.058
x_2^2	1	221.962	240.504	240.504	9.50	0.027
x ₃ ²	1	72.270	72.270	72.270	2.85	0.152
X_1X_2	1	22.912	22.912	22.912	0.91	0.385
X_1X_3	1	58.452	58.452	58.452	2.31	0.189
X ₂ X ₃	1	10.213	10.213	10.213	0.40	0.553
Lack of fit	3					0.059
R ²	82.39%					

TABLE 5 Analysis of variance (ANOVA) of BBD model for total lignan from oat.

¹⁾Temperature (°C); ²⁾Methanol concentration (%); ³⁾Extraction time (min). DF, Degrees of freedom; Seq SS, Sequential sum of squares; Adj SS, Adjusted sum of square; Adj MS, Adjusted mean squares.

TABLE 6 Optimum conditions, predicted and experimental values of responses of oat.

Responses in oat	X1	X ₂	X ₃	Total lignan contents (µg/100 g)	
				Predicted	Experimental
Optimum conditions	44.24°C	84.64%	53.63 min	60.213	59.277
Modified conditions	40°C	84%	60 min	-	59.564

X1, Temperature (°C); X2, Methanol concentration (%); X3, Extraction time (min).

process. However, considering that the moisture content of cooked rice is about 65% (68), the content of lignans consumed from cooked rice will be smaller than the value presented in Table 7. For example, the lignan intake from cooked Saeilmi brown rice would be 17.648 µg per serving of cooked rice (210 g). Smeds et al. (25) conducted a study on the lignan content of various cereals and found that wheat and barley accounted for approximately 80% of the total lignan content, while oats contained 42% Syr, with higher proportions of Lar and Pin compared to other varieties. The study also showed that Mat was not detected in some wheat and oat varieties, and Lar content varied widely, ranging from 50.3 to 291 µg/100 g depending on the variety. Peñalvo et al. (10) reported that wheat, oat, barley, and rice had the highest levels of Syr, with wheat accounting for over 50% of the total lignan content, three times more than barley and oat. Barley, wheat, and barley had low levels of Mat, ranging from $1 \mu g/100 g$ to $3 \mu g/100 g$. The lignan content of barley husks was found to be higher compared to other grains, particularly Lar, which was more than nine times higher compared to other grains. Makowska et al. (18) found that Syr had the highest lignan content among germinated grains, with some varieties showing increased contents of Lar and Syr after germination. Furthermore, Katina et al. (69) reported that germination approximately doubled the lignan levels in wheat. It is known that lignans are more easily released from the structure of germinated grains, leading to an increase in their content. In addition, the breakdown of some starch during germination can increase the content of bioactive compounds such as phenolic compounds. In a study conducted by Gerstenmeyer et al. (49), the content of Lar and Syr increased over time when wheat was steamed at 100°C. The increase in lignan content may be attributed to the facilitation of lignan extraction resulting from changes in the matrix after heating. Therefore, it is considered that the high content of lignan observed in the cooked samples was due to the release of lignans resulting from the gelatinization of starch during cooking.

In conclusion, optimal extraction conditions of lignans from cereal grains using RSM was investigated. A Box–Behnken design was employed to develop optimal extraction condition of lignan using the three extracting parameters (temperature, methanol concentration, and extraction time). The optimum extract conditions for lignans from grain were obtained at $X1 = 44.24^{\circ}$ C, X2 = 84.64%, and X3 = 53.63 min. To implement these experimental conditions in the actual experiment, slight modifications were made to the optimal conditions, resulting in $X1 = 40^{\circ}$ C, X2 = 80%, and X3 = 60 min. The predicted results matched well with the experimental results obtained using the modified optimal extraction conditions. The highest total lignan content of grain was found in barley sprout. These results may be useful for providing reliable data about the lignan contents in cereal grain.

TABLE 7 The contents of lignan in grains and grain products.

Sample		Lignan contents (μ g/100 g dry weight)						
		Lar	Mat	Pin	Seco	Syr	Total lignan	
	Baegokchal, white rice, raw	0.398 ± 0.023	ND	ND	ND	3.336 ± 0.006	3.734 ± 0.029	
	Baegokchal, white rice, cooked	0.931 ± 0.005	ND	ND	ND	3.577 ± 0.036	4.508 ± 0.041	
	Baekjinju, white rice, raw	0.527 ± 0.035	ND	ND	ND	ND	0.527 ± 0.035	
	Baekjinju, white rice, cooked	0.682 ± 0.021	ND	ND	ND	ND	0.682 ± 0.021	
	Dodamssal, white rice, raw	ND	ND	ND	ND	8.527 ± 0.164	8.527 ± 0.164	
	Dodamssal, white rice, cooked	ND	ND	ND	ND	10.124 ± 0.195	10.124 ± 0.195	
	Haedam, white rice, raw	ND	ND	ND	ND	7.892 ± 0.253	7.892 ± 0.253	
	Haedam, white rice, cooked	ND	ND	ND	ND	9.444 ± 0.609	9.444 ± 0.609	
	Heukjinju, brown rice, cooked	ND	ND	ND	ND	18.642 ± 0.503	18.642 ± 0.503	
	Heukjinju, brown rice, raw	ND	ND	ND	ND	10.538 ± 0.409	10.538 ± 0.409	
	Odae, brown rice, raw	ND	ND	ND	ND	9.189 ± 0.026	9.189 ± 0.026	
Rice (Oryza	Odae, brown rice, cooked	ND	ND	ND	ND	19.943 ± 1.139	19.943±1.139	
sativa L.)	Odae, white rice, raw	ND	ND	ND	ND	6.825 ± 0.723	6.825 ± 0.723	
	Odae, white rice, cooked	ND	ND	ND	ND	8.729 ± 0.197	8.729 ± 0.197	
	Saeilmi, brown rice, raw	ND	ND	ND	ND	7.805 ± 0.184	7.805 ± 0.184	
	Saeilmi, brown rice, cooked	ND	ND	ND	ND	24.011 ± 0.683	24.011 ± 0.683	
	Saelimi, white rice, raw	ND	ND	ND	ND	8.243 ± 0.209	8.243 ± 0.209	
	Saelimi, white rice, cooked	ND	ND	ND	ND	7.558 ± 0.095	7.558 ± 0.095	
	Samgwang, brown rice, raw	ND	ND	ND	ND	ND	ND	
	Samgwang, brown rice, cooked	1.012 ± 0.058	ND	ND	ND	9.080 ± 0.199	10.092 ± 0.141	
	Samgwang, white rice, raw	1.766 ± 0.095	ND	ND	ND	ND	1.766 ± 0.095	
	Samgwang, white rice, cooked	1.899 ± 0.028	ND	ND	ND	ND	1.899 ± 0.028	
	Shindongjin, brown rice, raw	0.980 ± 0.110	ND	ND	ND	6.621 ± 0.112	7.601 ± 0.222	
	Shindongjin, brown rice, cooked	0.466 ± 0.025	ND	ND	ND	5.858 ± 0.076	6.324 ± 0.051	
	Instant cooked brown rice	ND	ND	ND	ND	11.342 ± 0.407	11.342 ± 0.407	
	Instant cooked white rice	1.859 ± 0.089	ND	ND	ND	1.850 ± 0.016	3.709 ± 0.105	
Rice (Oryza	Instant scorched rice	2.059 ± 0.130	ND	ND	ND	2.895 ± 0.062	4.954 ± 0.191	
sativa L.)	Instant rice porridge	1.735 ± 0.037	ND	ND	ND	2.980 ± 0.023	4.715 ± 0.060	
products	Rice noodle, dried	2.428 ± 0.082	ND	ND	ND	ND	2.428 ± 0.082	
	Rice noodle, boiled	3.121 ± 0.147	ND	ND	ND	ND	3.121 ± 0.147	
	Chewy noodle, boiled	ND	ND	ND	ND	8.778 ± 0.091	8.778 ± 0.091	
	Chewy noodle, dried	ND	ND	ND	ND	8.452 ± 0.138	8.452 ± 0.138	
	Chopped noodle, raw	ND	ND	ND	ND	10.258 ± 0.143	10.258 ± 0.143	
	Chopped noodle, boiled	ND	ND	ND	ND	7.293 ± 0.385	7.293 ± 0.385	
	Ramyeon noodle, boiled	ND	ND	ND	ND	8.638 ± 0.361	8.638 ± 0.361	
Wheat (Triticum	Ramyeon noodle, dried	ND	ND	ND	ND	10.407 ± 0.452	10.407 ± 0.452	
aestivum L.)	Fine wheat noodle, boiled	ND	ND	ND	ND	7.329 ± 0.072	7.329 ± 0.072	
products	Fine wheat noodle, dried	ND	ND	ND	ND	8.063±0.313	8.063±0.313	
	Spaghetti noodle, boiled	ND	ND	ND	ND	8.266 ± 0.600	8.266 ± 0.600	
	Spaghetti noodle, dried	ND	ND	ND	ND	10.006 ± 0.469	10.006 ± 0.469	
	Udon noodle, raw	ND	ND	ND	ND	9.437 ± 0.075	9.437 ± 0.075	
	Udon noodle, boiled	ND	ND	ND	ND	9.263 ± 0.194	9.263 ± 0.194	
	Wheat flour, all-purpose A	ND	ND	ND	ND	11.532 ± 0.267	11.532 ± 0.267	

(Continued)

TABLE 7 (Continued)

Sample		Lignan contents (μ g/100 g dry weight)							
		Lar	Mat	Pin	Seco	Syr	Total lignan		
	Wheat flour, all-purpose B	ND	ND	ND	ND	18.519 ± 1.125	18.519 ± 1.125		
	Wheat flour, pan-frying	ND	ND	ND	ND	11.594 ± 0.479	11.594 ± 0.479		
Miscellaneous	Barley sprout, dried	36.491 ± 1.928	ND	7.726 ± 0.725	ND	41.713 ± 3.174	85.930 ± 1.970		
	Job's tears (Coix lacryma-jobi), raw	ND	ND	ND	ND	8.404 ± 0.390	8.404 ± 0.390		
	Job's tears (Coix lacryma-jobi), steamed	ND	ND	ND	ND	10.996 ± 0.363	10.996 ± 0.363		
	Oat (Avena sativa), Daeyang, raw	ND	ND	ND	ND	27.371 ± 0.261	27.371 ± 0.261		
	Oat (Avena sativa), Daeyang, cooked	ND	ND	ND	ND	44.255 ± 1.449	44.255 ± 1.449		

All values are mean ± SD. ND, Not detected; Lar, Lariciresinol; Mat, Matairesinol; Pin, Pinoresinol; Seco, Secoisolariciresinol; Syr, Syringaresinol.

Data availability statement

The original contributions presented in the study are included in the article/Supplementary material, further inquiries can be directed to the corresponding authors.

Author contributions

YooK: Formal analysis, Investigation, Writing – original draft. H-WK: Conceptualization, Writing – review & editing. JS: Conceptualization, Supervision, Writing – original draft. YouK: Conceptualization, Funding acquisition, Writing – review & editing.

Funding

The author(s) declare that financial support was received for the research, authorship, and/or publication of this article. This study was carried out with the support of "Cooperative Research Program for Agriculture Science and Technology Development (project no. RS-2022-RD009982)" Rural Development Administration, Republic

References

1. Umezawa T. Phylogenetic distribution of lignan producing plants. *Wood Res.* (2003) 90:27–110.

2. Landete JM. Plant and mammalian lignans: a review of source, intake, metabolism, intestinal bacteria and health. *Food Res Int.* (2012) 46:410–24. doi: 10.1016/j. foodres.2011.12.023

3. Xu WH, Zhao P, Wang M, Liang Q. Naturally occurring furofuran lignans: structural diversity and biological activities. *Nat Prod Res.* (2019) 33:1357–73. doi: 10.1080/14786419.2018.1474467

4. Penalvo JL, Adlercreutz H, Uehara M, Ristimaki A, Watanabe S. Lignan content of selected foods from Japan. *J Agric Food Chem.* (2008) 56:401–9. doi: 10.1021/jf072695u

5. Willfor SM, Smeds AI, Holmbom BR. Chromatographic analysis of lignans. J Chromatogr A. (2006) 1112:64–77. doi: 10.1016/j.chroma.2005.11.054

6. Yashin AY, Yashunskii DB, Vedenin AN, Nifant'ev NE, Nemzer BV, Yashin YI. Chromatographic determination of lignans (antioxidants) in food products. *J Aanal Chem.* (2018) 73:399–406. doi: 10.1134/S106193481805012X

7. Tardugno R, Cicero N, Costa R, Nava V, Vadalà R. Exploring lignans, a class of health promoting compounds, in a variety of edible oils from Brazil. *Food Secur.* (2022) 11:1386. doi: 10.3390/foods11101386

8. Niemeyer HB, Honig DM, Kulling SE, Metzler M. Studies on the metabolism of the plant lignans secoisolariciresinol and matairesinol. *J Agric Food Chem.* (2003) 51:6317–25. doi: 10.1021/jf030263n

of Korea and the BB21plus funded by Busan metropolitan City and Busan Techno Park.

Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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Supplementary material

The Supplementary material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/fnut.2024.1409309/ full#supplementary-material

9. Hu C, Yuan YV, Kitts DD. Antioxidant activities of the flaxseed lignan secoisolariciresinol diglucoside, its aglycone secoisolariciresinol and the mammalian lignans enterodiol and enterolactone in vitro. *Food Chem Toxicol.* (2007) 45:2219–27. doi: 10.1016/j.fct.2007.05.017

10. Peñalvo JL, Haajanen KM, Botting N, Adlercreutz H. Quantification of lignans in food using isotope dilution gas chromatography/mass spectrometry. *J Agric Food Chem.* (2005) 53:9342–7. doi: 10.1021/jf051488w

11. Cornwell T, Cohick W, Raskin I. Dietary phytoestrogens and health. *Phytochemistry*. (2004) 65:995–1016. doi: 10.1016/j.phytochem.2004.03.005

12. Wang LQ. Mammalian phytoestrogens: Enterodiol and Enterolactone. J Chromatogr B Analyt Technol Biomed Life Sci. (2002) 777:289–309. doi: 10.1016/s1570-0232(02)00281-7

13. Zaki UKH, Fryganas C, Trijsburg L, Feskens EJ, Capuano E. In vitro gastrointestinal bioaccessibility and colonic fermentation of lignans from fresh, fermented, and germinated flaxseed. *Food Funct.* (2022) 13:10737–47. doi: 10.1039/D2FO02559K

14. Berenshtein L, Okun Z, Shpigelman A. Stability and bioaccessibility of Lignans in food products. ACS Omega. (2024) 9:2022–31. doi: 10.1021/acsomega.3c07636

15. Cui S, Liu RH. Health benefits of oat phytochemicals. Oats Nutr Technol. (2013) 2013:171-94. doi: 10.1002/9781118354100.ch7

16. Singh R, De S, Belkheir A. Avena sativa (oat), a potential neutraceutical and therapeutic agent: an overview. Crit Rev Food Sci Nutr. (2013) 53:126-44. doi: 10.1080/10408398.2010.526725

17. Bolvig AK, Adlercreutz H, Theil PK, Jørgensen H, Knudsen KEB. Absorption of plant lignans from cereals in an experimental pig model. *Br J Nutr*. (2016) 115:1711–20. doi: 10.1017/S0007114516000829

18. Makowska A, Waśkiewicz A, Chudy S. Lignans in triticale grain and triticale products. J Cereal Sci. (2020) 93:102939. doi: 10.1016/j.jcs.2020.102939

19. Bleidere M, Zute S, Gailāne N, Kantāne I, Jākobsone I, Bartkevičs V. Lignans in oat and barley grains depending on genotype and farming practice. *Proc Latv Acad Sci Section B Nat Exact Appl Sci.* (2022) 76:124–30. doi: 10.2478/prolas-2022-0019

20. Nørskov NP, Knudsen KEB. Validated LC-MS/MS method for the quantification of free and bound lignans in cereal-based diets and feces. J Agric Food Chem. (2016) 64:8343–51. doi: 10.1021/acs.jafc.6b03452

21. Schwartz H, Sontag G. Determination of secoisolariciresinol, lariciresinol and isolariciresinol in plant foods by high performance liquid chromatography coupled with coulometric electrode array detection. *J Chromatogr B Analyt Technol Biomed Life Sci.* (2006) 838:78–85. doi: 10.1016/j.jchromb.2006.03.058

22. Smeds AI, Jauhiainen L, Tuomola E, Peltonen-Sainio P. Characterization of variation in the lignan content and composition of winter rye, spring wheat, and spring oat. *J Agric Food Chem.* (2009) 57:5837–42. doi: 10.1021/jf9004274

23. Thompson LU, Boucher BA, Liu Z, Cotterchio M, Kreiger N. Phytoestrogen content of foods consumed in Canada, including isoflavones, lignans, and coumestan. *Nutr Cancer.* (2006) 54:184–201. doi: 10.1207/s15327914nc5402_5

24. Moazzami AA, Haese SL, Kamal-Eldin A. Lignan contents in sesame seeds and products. *Eur J Lipid Sci Technol.* (2007) 109:1022–7. doi: 10.1002/ejlt.200700057

25. Pilar B, Gullich A, Oliveira P, Stroher D, Piccoli J, Manfredini V. Protective role of flaxseed oil and flaxseed lignan secoisolariciresinol diglucoside against oxidative stress in rats with metabolic syndrome. *J Food Sci.* (2017) 82:3029–36. doi: 10.1111/1750-3841.13964

26. Smeds AI, Eklund PC, Willfor SM. Content, composition, and stereochemical characterisation of lignans in berries and seeds. *Food Chem.* (2012) 134:1991–8. doi: 10.1016/j.foodchem.2012.03.133

27. Cheng Z, Song H, Yang Y, Zhou H, Liu Y, Liu Z. Smashing tissue extraction of five lignans from the fruit of *Schisandra chinensis*. *J Chromatogr Sci*. (2016) 54:bmv116. doi: 10.1093/chromsci/bmv116

28. Guo T, Su D, Huang Y, Wang Y, Li YH. Ultrasound-assisted aqueous two-phase system for extraction and enrichment of *Zanthoxylum armatum* lignans. *Molecules*. (2015) 20:15273–86. doi: 10.3390/molecules200815273

29. Nemes SM, Orsat V. Evaluation of a microwave-assisted extraction method for lignan quantification in flaxseed cultivars and selected oil seeds. *Food Anal Methods*. (2012) 5:551–63. doi: 10.1007/s12161-011-9281-6

30. Angeloni S, Navarini L, Khamitova G, Maggi F, Sagratini G, Vittori S, et al. A new analytical method for the simultaneous quantification of isoflavones and lignans in 25 green coffee samples by HPLC-MS/MS. *Food Chem.* (2020) 325:126924. doi: 10.1016/j. foodchem.2020.126924

31. Smeds AI, Eklund PC, Sjoholm RE, Willfor SM, Nishibe S, Deyama T, et al. Quantification of a broad spectrum of lignans in cereals, oilseeds, and nuts. J Agric Food Chem. (2007) 55:1337–46. doi: 10.1021/jf0629134

32. Castro-Perez JM. Current and future trends in the application of HPLC-MS to metabolite-identification studies. *Drug Discov Today*. (2007) 12:249–56. doi: 10.1016/j. drudis.2007.01.007

33. Gallien S, Duriez E, Demeure K, Domon B. Selectivity of LC-MS/MS analysis: implication for proteomics experiments. *J Proteome*. (2013) 81:148–58. doi: 10.1016/j. jprot.2012.11.005

34. Lucci P, Saurina J, Núñez O. Trends in LC-MS and LC-HRMS analysis and characterization of polyphenols in food. *TrAC Trends Anal Chem*. (2017) 88:1–24. doi: 10.1016/j.trac.2016.12.006

35. López MJO, Innocenti M, Ieri F, Giaccherini C, Romani A, Mulinacci N. HPLC/ DAD/ESI/MS detection of lignans from Spanish and Italian Olea europaea L. fruits. J Food Compos Anal. (2008) 21:62–70. doi: 10.1016/j.jfca.2007.04.012

36. Kuhnle GG, Dell'Aquila C, Runswick SA, Bingham SA. Variability of phytoestrogen content in foods from different sources. *Food Chem.* (2009) 113:1184–7. doi: 10.1016/j. foodchem.2008.08.004

37. Khuri AI, Mukhopadhyay S. Response surface methodology. Wiley Interdisciplin Rev Comput Stat. (2010) 2:128–49. doi: 10.1002/wics.73

38. Adki KM, Kulkarni YA. Chemistry, pharmacokinetics, pharmacology and recent novel drug delivery systems of paeonol. *Life Sci.* (2020) 250:117544. doi: 10.1016/j. lfs.2020.117544

39. Tirado-Kulieva VA, Sánchez-Chero M, Yarlequé MV, Aguilar GFV, Carrión-Barco G, Santa Cruz AGY. An overview on the use of response surface methodology to model and optimize extraction processes in the food industry. *Curr Res Nutr Food Sci.* (2021) 9:745–54. doi: 10.12944/CRNFSJ.9.3.03

40. Zhang ZS, Li D, Wang LJ, Ozkan N, Chen XD, Mao Z-H, et al. Optimization of ethanol-water extraction of lignans from flaxseed. *Sep Purif Technol.* (2007) 57:17–24. doi: 10.1016/j.seppur.2007.03.006

41. Zhao LC, He Y, Deng X, Yang GL, Li W, Liang J, et al. Response surface modeling and optimization of accelerated solvent extraction of four lignans from fructus schisandrae. *Molecules*. (2012) 17:3618–29. doi: 10.3390/molecules17043618

42. Wang D, Zhang L, Xu Y, Qi X, Wang X, Wang X, et al. Optimization of an ultrasoundassisted extraction for simultaneous determination of antioxidants in sesame with response surface methodology. *Antioxidants*. (2019) 8:321. doi: 10.3390/antiox8080321

43. Kim Y, Pyeon J, Baek I, Kim Y. Development of simultaneous analytical method for five lignans in edible seeds. *Korean J Food Nutr.* (2022) 35:543–51. doi: 10.9799/ksfan.2022.35.6.543

44. Chen Y, Lin H, Lin M, Zheng Y, Chen J. Effect of roasting and *in vitro* digestion on phenolic profiles and antioxidant activity of water-soluble extracts from sesame. *Food Chem Toxicol.* (2020) 139:111239. doi: 10.1016/j.fct.2020.111239

45. Li X, Yuan JP, Xu SP, Wang JH, Liu X. Separation and determination of secoisolariciresinol diglucoside oligomers and their hydrolysates in the flaxseed extract by high-performance liquid chromatography. *J Chromatogr A*. (2008) 1185:223–32. doi: 10.1016/j.chroma.2008.01.066

46. Mekky RH, Abdel-Sattar E, Segura-Carretero A, Contreras MDM. Phenolic compounds from sesame cake and antioxidant activity: a new insight for Agri-food residues' significance for sustainable development. *Food Secur.* (2019) 8:432. doi: 10.3390/foods8100432

47. Durazzo A, Raguzzini A, Azzini E, Foddai M, Narducci V, Maiani G, et al. Bioactive molecules in cereal grains. *Tec Molitoria*. (2009) 60:130–41.

48. Durazzo A, Zaccaria M, Polito A, Maiani G, Carcea M. Lignan content in cereals, buckwheat and derived foods. *Food Secur.* (2013) 2:53–63. doi: 10.3390/foods2010053

49. Gerstenmeyer E, Reimer S, Berghofer E, Schwartz H, Sontag G. Effect of thermal heating on some lignans in flax seeds, sesame seeds and rye. *Food Chem.* (2013) 138:1847–55. doi: 10.1016/j.foodchem.2012.11.117

50. Akhila H, Beevy SS. Quantification of seed oil and evaluation of antioxidant properties in the wild and cultivated species of *Sesamum* L. (Pedaliaceae). *Int J Pharm Pharm Sci.* (2015) 7:136–42.

51. Chen Y, Lin H, Lin M, Lin P, Chen J. Effects of thermal preparation and in vitro digestion on lignan profiles and antioxidant activity in defatted-sesame meal. *Food Chem Toxicol.* (2019) 128:89–96. doi: 10.1016/j.fct.2019.03.054

52. Grougnet R, Magiatis P, Laborie H, Lazarou D, Papadopoulos A, Skaltsounis AL. Sesamolinol glucoside, disaminyl ether, and other lignans from sesame seeds. *J Agric Food Chem.* (2012) 60:108–11. doi: 10.1021/jf2040687

53. Renouard S, Hano C, Corbin C, Fliniaux O, Lopez T, Montguillon J, et al. Cellulase-assisted release of secoisolariciresinol from extracts of flax (*Linum usitatissimum*) hulls and whole seeds. *Food Chem*. (2010) 122:679–87. doi: 10.1016/j. foodchem.2010.03.036

54. Sharmila G, Nikitha V, Ilaiyarasi S, Dhivya K, Rajasekar V, Kumar NM, et al. Ultrasound assisted extraction of total phenolics from *Cassia auriculata* leaves and evaluation of its antioxidant activities. *Ind Crop Prod.* (2016) 84:13–21. doi: 10.1016/j. indcrop.2016.01.010

55. Wang J, Sun B, Cao Y, Tian Y, Li X. Optimisation of ultrasound-assisted extraction of phenolic compounds from wheat bran. *Food Chem.* (2008) 106:804–10. doi: 10.1016/j. foodchem.2007.06.062

56. Medina-Torres N, Ayora-Talavera T, Espinosa-Andrews H, Sánchez-Contreras A, Pacheco N. Ultrasound assisted extraction for the recovery of phenolic compounds from vegetable sources. *Agronomy*. (2017) 7:47. doi: 10.3390/agronomy7030047

57. Paniwnyk L, Beaufoy E, Lorimer JP, Mason TJ. The extraction of rutin from flower buds of *Sophora japonica*. *Ultrason Sonochem*. (2001) 8:299–301. doi: 10.1016/s1350-4177(00)00075-4

58. Rostagno MA, Palma M, Barroso CG. Ultrasound-assisted extraction of soy isoflavones. J Chromatogr A. (2003) 1012:119–28. doi: 10.1016/s0021-9673(03)01184-1

59. Corbin C, Fidel T, Leclerc EA, Barakzoy E, Sagot N, Falguieres A, et al. Development and validation of an efficient ultrasound assisted extraction of phenolic compounds from flax (*Linum usitatissimum* L.) seeds. *Ultrason Sonochem*. (2015) 26:176–85. doi: 10.1016/j.ultsonch.2015.02.008

60. Majeed M, Hussain AI, Chatha SA, Khosa MK, Kamal GM, Kamal MA, et al. Optimization protocol for the extraction of antioxidant components from *Origanum vulgare* leaves using response surface methodology. *Saudi J Biol Sci.* (2016) 23:389–96. doi: 10.1016/j.sjbs.2015.04.010

61. Martínez-Ramos T, Benedito-Fort J, Watson NJ, Ruiz-López II, Che-Galicia G, Corona-Jiménez E. Effect of solvent composition and its interaction with ultrasonic energy on the ultrasound-assisted extraction of phenolic compounds from mango peels (*Mangifera indica L.*). *Food Bioprod Process.* (2020) 122:41–54. doi: 10.1016/j. fbp.2020.03.011

62. Antony A, Farid M. Effect of temperatures on polyphenols during extraction. *Appl Sci.* (2022) 12:2107. doi: 10.3390/app12042107

63. Liyanapathirana C, Shahidi F. Optimization of extraction of phenolic compounds from wheat using response surface methodology. *Food Chem.* (2005) 93:47–56. doi: 10.1016/j.foodchem.2004.08.050

64. Fombang EN, Nobossé P, Mbofung CM, Singh D. Optimising extraction of antioxidants from roasted *Moringa oleifera* lam. Leaves using response surface methodology. *J Food Process Preserv*. (2020) 44:e14482. doi: 10.1111/jfpp.14482

65. Bilgin M, Elhussein EAA, Ozyurek M, Guclu K, Sahin S. Optimizing the extraction of polyphenols from *Sideritis montana* L. using response surface methodology. *J Pharm Biomed Anal.* (2018) 158:137–43. doi: 10.1016/j.jpba.2018.05.039

66. Kwon JH, Belanger JM, Pare JR. Optimization of microwave-assisted extraction (MAP) for ginseng components by response surface methodology. *J Agric Food Chem.* (2003) 51:1807–10. doi: 10.1021/jf026068a

67. Vongsak B, Sithisarn P, Mangmool S, Thongpraditchote S, Wongkrajang Y, Gritsanapan W. Maximizing total phenolics, total flavonoids contents and antioxidant activity of *Moringa oleifera* leaf extract by the appropriate extraction method. *Ind Crop Prod.* (2013) 44:566–71. doi: 10.1016/j.indcrop.2012.09.021

68. Dibba D, Weaver LT, Hudson GJ. Water content of cooked rice is related to variety and cooking method. *Nutr Res.* (1991) 1991:1397–402.

69. Katina K, Liukkonen KH, Kaukovirta-Norja A, Adlercreutz H, Heinonen SM, Lampi AM, et al. Fermentation-induced changes in the nutritional value of native or germinated rye. *J Cereal Sci.* (2007) 46:348–55. doi: 10.1016/j.jcs.2007.07.006