



Electronic-Nose as Non-destructive Tool to Discriminate “Ferrovia” Sweet Cherries Cold Stored in Air or Packed in High CO₂ Modified Atmospheres

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This study aimed to explore the applicability of electronic-nose (E-nose) as a rapid method in discriminating samples of sweet cherry cv “Ferrovia” stored in high-CO₂ (16% O₂ + 20% CO₂ + 64% N₂) or air (control) up to 21 days. Projection to Latent Structures (PLS) methods applied to E-nose data showed that fresh fruit and the packaged or unpackaged samples can be distinguished, according to both the storage condition and the storage days. Moreover, a correlation analysis between E-nose sensors and 45 volatile compounds were overall, obtained from all the investigated sweet cherry samples by Headspace Solid-Phase Microextraction (HS SPME) coupled to Gas Chromatography-Mass Spectrometry (GC-MS). These methods allowed to associate samples with a specific flavour profile to one or more E-nose sensors. Finally, quality attributes (visual quality, colour, firmness, antioxidant activity, total phenols, and sugar content) were assessed during storage. Among these, visual quality and berry deformation resulted affected by storage conditions, showing that high-CO₂ treatment better preserved the fruit quality than control.

Keywords: sweet cherry cv “Ferrovia,” electronic-nose (E-nose), volatile profiles, projection to latent structure, correlation analysis, firmness

INTRODUCTION

Cold temperatures storage and modified atmosphere (MA) packaging are well-established practises for delaying the senescence of fresh fruit and vegetables, extending their postharvest life.

Aroma and flavour, which are crucial sensory attributes for consumer acceptability, are directly influenced by the organic volatile compounds (VOCs) profiles (1). Volatile metabolites represent the final products of fruit metabolism and changes in their profiles during storage can suggest the development of spoilage (2). In the last decades, the identification and quantification of the aroma compounds by Headspace Solid-Phase Microextraction (HS SPME) sampling followed by Gas Chromatography-Mass Spectrometry (GC-MS) became a well-established method for the evaluation of the most suitable postharvest conditions to preserve the sensory quality of freshly harvested fruit and vegetables during the storage (3). However, the practical application in the food industry of HS SPME and GC-MS is limited because these techniques involve expensive instrumentations and need considerable analytical skilled manpower to handle the

complex mixture and the high number of volatiles that comprise the VOCs pattern of crops (4). In recent years, electronic nose (E-nose) technology has been demonstrated to overcome some of the drawbacks associated with the methods traditionally used for the evaluation of VOCs profile (3). Electronic nose is a device equipped with an array of partial specific and broad-spectrum electronic chemical sensors which simulate human olfactory perception and offers a digital fingerprint of the volatile components that can be investigated with appropriate statistical tools. The use of the olfactometric method for the investigation of volatiles of fruit and vegetable presents several advantages, such as low cost, easy-to-handle, fast and non-destructive analysis, no preliminary sample preparation steps, and automatic data management (5). However, there are some disadvantages of using different E-nose sensors, including their response and recovery times, sensitivities, detection range, operating restrictions, physical size, inactivation by certain poisoning agents, and other limitations that are specific to individual sensor types. As clearly reported by the study of Wilson and Baietto (6), the varieties of advantages and limitations related to each E-nose sensor type are strictly connected with the nature of the technology that regulates the principle for detection and the types of analytes that may be detected by each sensor type. This technique has already an application in the food industry, including food quality control, authenticity, and traceability studies of fresh fruit (4).

Sweet cherries are greatly appreciated worldwide because of their appealing colour, aroma, taste, and health beneficial nutritional properties. This fruit is very perishable, and for the various cultivars, different MA treatments have been reported to preserve sweet cherry quality for the fresh market (7–10). A typical Italian sweet cherry cultivar “Ferrovia” is very appreciated for its big and bright skin berries, characterised by agreeable and middle sweetness flavour that allows this cultivar to be suitable for fresh consumption (11). In a very recent study of Cozzolino et

al. (12), the changes of VOCs profiles were detected by HS SPME and GC-MS of sweet cherries cv “Ferrovia,” cold-stored in high CO₂ modified atmospheres with three different compositions up to 21 days (11). Among the MA treatments, the authors observed that storage in high-CO₂ (16% O₂ + 20% CO₂ + 64% N₂) resulted in the most suitable packaging condition in preserving the sensory quality of the fruit till the end of storage (12).

The objective of the present study was to explore, for the first time, the applicability of E-nose as a rapid methodology in differentiating samples of sweet cherry cv “Ferrovia” stored in high-CO₂ or air (control) for up to 21 days. Multivariate and univariate statistical data analysis have been applied to investigate the effects of storage time and storage condition on the E-nose fingerprint and VOCs profile [previously studied in (12)], while correlation analysis was performed to discover the relationships between sensors and VOCs. Moreover, some quality attributes, including visual quality (VQ), colour, deformation, antioxidant activity, total phenols, and sugar content, have also been investigated.

MATERIALS AND METHODS

Chemicals and Reagents

Sodium carbonate was purchased from Merck (Germany), while 2,2-Diphenyl-1-picryldazyl (DPPH), hydrochloric acid, potassium chloride, sodium acetate, and Folin-Ciocalteu’s phenol reagent were obtained from Fluka (Buchs, Switzerland).

Sweet Cherry Samples

Sweet cherries (*Prunus avium* L., cv Ferrovia) were provided by a local farm (Ermes snc, Noicattaro, Italy) in May 2017 and immediately transported to the Postharvest Laboratory of ISPA CNR. Fruits weighing 200 g were closed in 30 cm × 40 cm polyamide/polyethylene (PA/PE, 90 μm thick, Orved, Musile di Piave, Italy) plastic bags (Boxer 50, Lavezzini Vacuum Packaging

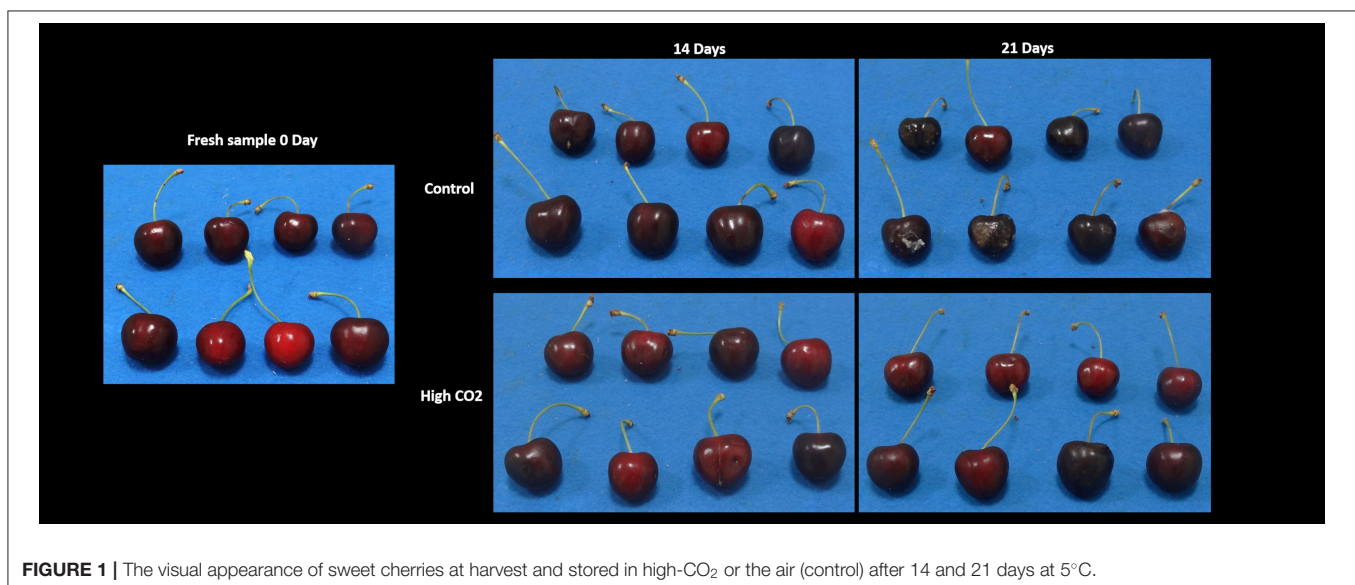


FIGURE 1 | The visual appearance of sweet cherries at harvest and stored in high-CO₂ or the air (control) after 14 and 21 days at 5°C.

TABLE 1 | Main effects of storage days and storage condition on visual quality score, ΔE^* , fruit deformation and antioxidant activity of sweet cherries (cv. *Ferrovía*).

	Visual quality (score 5–1)		ΔE^*		Fruit deformation (%)		Antioxidant activity (mg Trolox/100 g fw)	
Days	**		*		**		*	
14	3.0	b	5.7	ab	12.4	b	69.74	b
21	2.5	c	7.4	a	14.1	a	85.29	a
Storage condition	***		ns		***		ns	
Control	3.0	b	6.3		13.9	a	76.35	
High-CO ₂	4.5	a	5.0		12.3	b	78.56	

For each parameter, different letters indicate a statistical difference for $p = 0.05$ (LSD test). Ns, not significant; * $p \leq 0.05$; ** $p \leq 0.01$; *** $p \leq 0.001$. Data are mean values of 3 replicates for each storage condition at each storage day.

System, Fiorenzuola d'Arda, Italy) in a modified atmosphere of high-CO₂ (16% O₂ + 20% CO₂) or were stored in open bags in air (control). All samples in triplicate were stored at 5°C and were analysed, as reported below, at harvest (fresh sample) and after 14 and 21 days. The headspace gas composition (O₂ and CO₂) within each high-CO₂ bag was measured every day using a gas analyser (CheckPoint, PBI Dansensor, Ringsted, Denmark) and a gas chromatograph (p200 micro-GC, Agilent, CA, USA).

Quality Analysis

In order to evaluate the effect of storage on the visual quality, the fruits were scored by 10 trained panellists, using a hedonic scale of 5–1, with 5 = excellent, no defects; 4 = very good, minor defects; 3 = fair, moderate defects (limit of marketability); 2 = poor, major defects (limit of edibility) and 1 = inedible (13).

The L^* (lightness), a^* (redness), and b^* (yellowness) parameters were measured at 3 random points on 10 fruits for each replication, using a colorimeter (CR-400, Konica Minolta, Osaka, Japan), as previously described (14). The ΔE^* was then calculated from primary a^* and b^* readings, using the following formula $\Delta E^* = [(L_0^* - L^*)^2 + (a_0^* - a^*)^2 + (b_0^* - b^*)^2]^{1/2}$ (15).

Sweet cherry deformation was measured, on 10 fruits from each replication, as the force required to obtain a 3 mm deformation of each berry, using a texture analyser (ZwickLine Z0.5-Zwick/Roell, Ulm, Germany) equipped with a 100 mm diameter plate (16). The measure was expressed in %, normalising the diameter of the berry.

The total soluble phenolic compounds were analysed on fresh weight according to the Folin-Ciocalteu colourimetric method (17) and the concentrations were expressed in mg g⁻¹ of gallic acid equivalents. The antioxidant activity was assessed on fresh weight by using the DPPH assay (18).

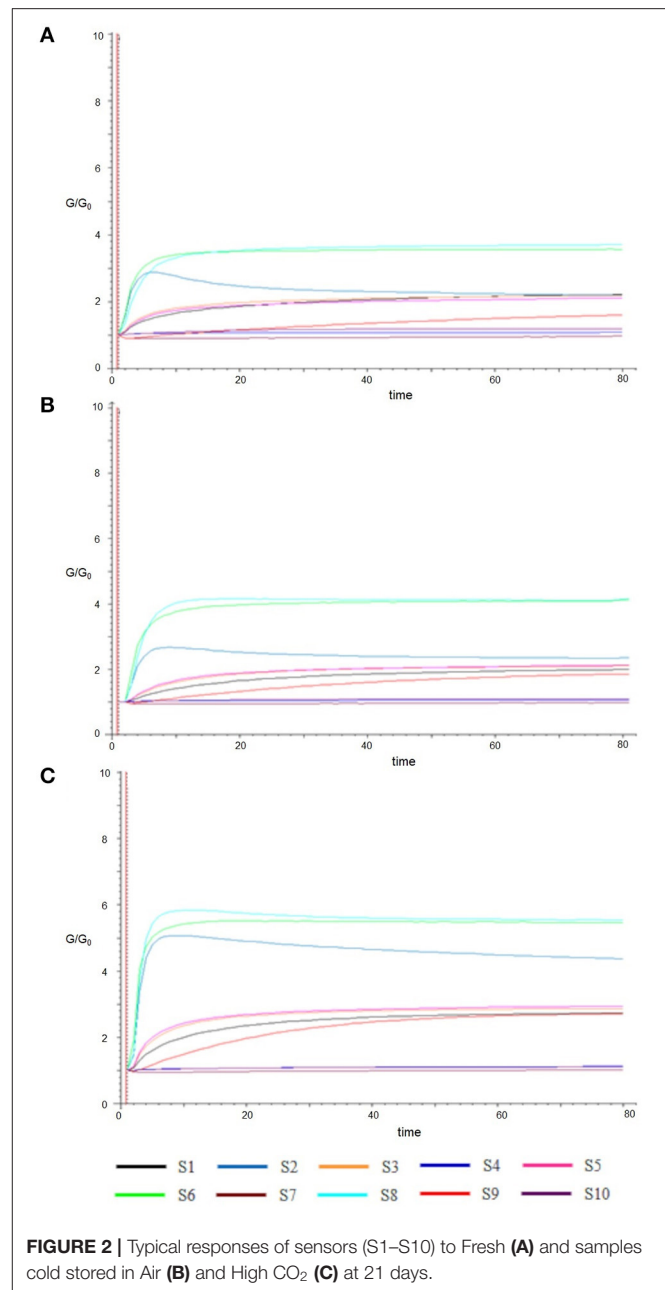
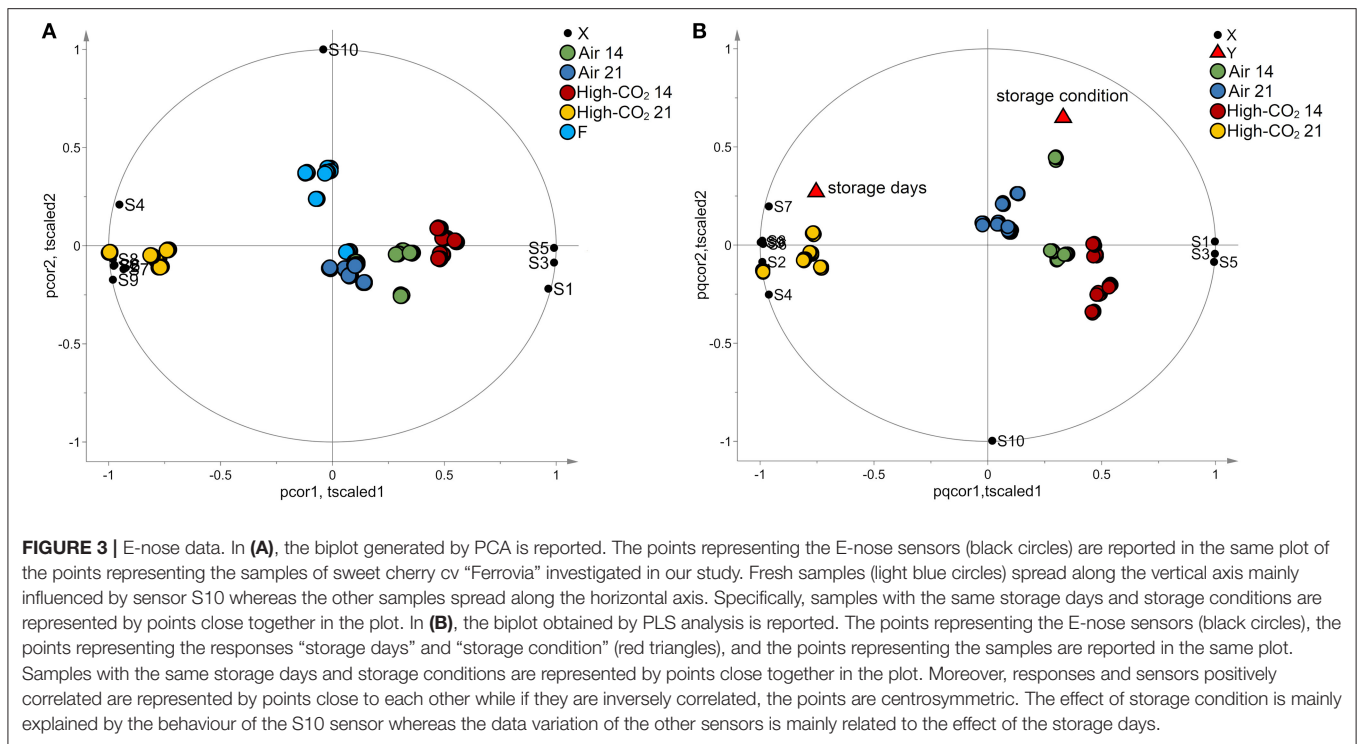


FIGURE 2 | Typical responses of sensors (S1–S10) to Fresh (A) and samples cold stored in Air (B) and High CO₂ (C) at 21 days.

Total sugars were analysed on ethanol extracts by a phenol-sulfuric colourimetric method (19) with colour development at 490 nm and glucose as a standard.

Analysis of Volatile Compounds

Volatile compounds identification and semi-quantification by HS SPME and GC-MS were previously reported in the study of Cozzolino et al. (12) which used DVB/CAR/PDMS (50/30 μm) fibre at 45°C and 20 min as extraction temperature and time, respectively. Volatiles were analysed by using the GC device, model GC 7890A, Agilent (Agilent Technologies, CA, USA),



coupled to the mass spectrometer 5975 C (Agilent). Semi-quantitative data of each metabolite (Relative Peak Area, RPA%) were calculated with respect to the peak area of 4-methyl-2-pentanol, used as IS (12). Areas of the identified metabolites were measured from the total ion chromatogram (TIC).

Evaluation of Cherry Samples by E-Nose

The aroma profiles from the headspace of cherry samples were carried out using a commercial portable E-nose (E-nose, PEN 3, Airsense Analytics Inc., Schwerin, Germany, including the Win Muster software). The electronic nose system consists of a sampling unit and a gas detection system equipped with an array of 10 metal oxide semiconductors (MOS) sensors with different thicknesses and chemical compositions, to offer selectivity toward various volatile classes, as reported in the study of Shi et al. (20). Due to the high operative temperatures (200–500°C), VOCs transported to the surface of the sensors were completely combusted to carbon dioxide and water, causing a change in the resistance. The response of the MOS sensors, expressed as resistivity (O), was based on the variations in conductivity, due to the adsorption of gas molecules, and on the following surface reactions. For sample preparation, 4 g of each sample were put in 45 ml airtight glass vials and sealed with a screw cap with Poly (1,1,2,2-tetrafluoroethylene) (PTFE)/silicone septum. To reach the headspace equilibrium, each vial was kept at 30°C for 30 min and analysed at $22 \pm 2^\circ\text{C}$ and $50 \pm 5\%$ relative humidity (RH). In the course of the measurement time, the gas headspace was injected into the E-nose for 80 s at 400 ml/min. Analyses were conducted in six technical replicates for each biological sample and data were collected at each second.

When the volatiles reached the measurement unit, the sensor conductivity modifies, first growing and then stabilising reaching a steady-state, causing changes of the ratio G/G_0 (G and G_0 are conductance of the sensors exposed to sample gas and zero gas, respectively) of each sensor. Subsequently, a second pump carries the filtered air to the sensor array for 400 s with a flow rate of 600 ml/min to clean the system between two consecutive analyses.

Electronic nose data were collected by the pattern recognition software (WinMuster, v.1.6., Airsense Analytics GmbH, Germany) and the average of each sensor response in the range from 70 to 75 s (area under the curve) was used for statistical data analysis.

Statistical Data Analysis

Quality parameters were analysed by ANOVA (Statistica 6 software, SatSoft Inc, Tulsa Oklahoma), considering factors such as storage condition (high- CO_2 or air), storage days (14 and 21 days), and their interaction.

The electronic nose data and the VOCs have been investigated by ANOVA and multivariate data analysis based on projection methods. Specifically, the factors storage condition, storage days, and their interaction term have been included in the ANOVA model controlling the false discovery rate by the Benjamini-Hochberg procedure at the level $\delta = 0.05$. Principal Component Analysis (PCA) has been applied for exploratory data analysis whereas Projection to Latent Structure (PLS) by partial least squares regression were conducted to investigate the effects of storage condition and storage days on the collected data (21).

The relationships between E-nose responses and VOCs have been explored by correlation analysis. Specifically, the Pearson

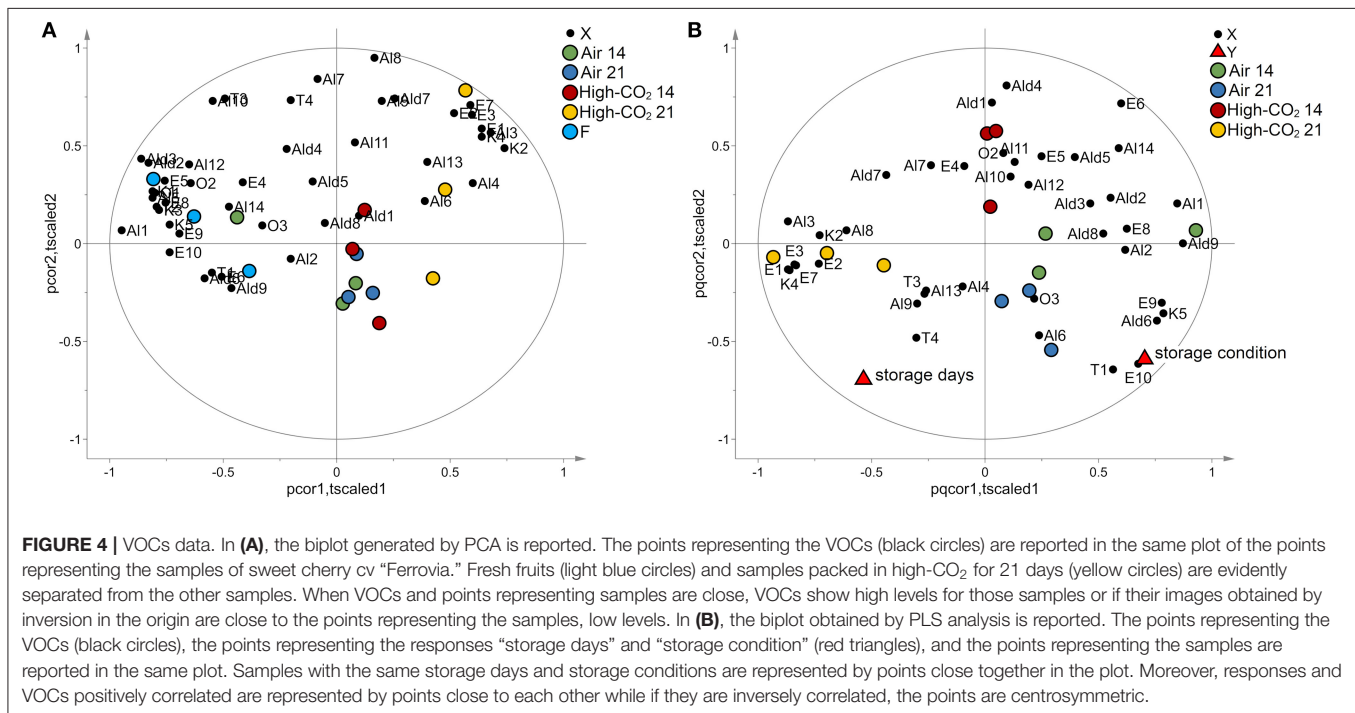
TABLE 2 | ANOVA: *p*-value (*p*) of the effects of storage condition (High-CO₂ and Air), storage days (14 and 21 days at 5°C), and their interaction on VOCs and E-nose data. Effects selected by Benjamini-Hochberg procedure assuming $\delta = 0.05$ are indicated with “*”.

Type	Name	Code	p[storage condition]	p[storage days]	p[interaction]
VOCs					
Esters					
	Ethyl acetate	E1	3.5E-05*	3.5E-05*	3.5E-05*
	Ethyl 2-butenoate	E2	1.1E-01	1.1E-01	1.1E-01
	Ethyl Hexanoate	E3	1.2E-02	1.2E-02	1.2E-02
	1-Hexyl acetate	E4	5.3E-01	6.8E-01	6.3E-01
	2-Hexen-1-ol acetate	E5	8.5E-01	2.6E-01	6.8E-01
	2-Hexenyl butyrate	E6	6.8E-01	1.0E-03*	3.4E-02
	Ethyl benzoate	E7	2.4E-02	2.4E-02	2.4E-02
	trans 2-hexenyl hexenoate	E8	2.4E-01	2.4E-01	2.4E-01
	2-Hexenyl tiglate	E9	1.7E-02	5.1E-01	5.1E-01
	Isopropyl Laurate	E10	2.5E-03*	4.8E-01	4.8E-01
Alcohols					
	1-Penten-3-ol	Al1	2.8E-02	2.7E-02	7.4E-02
	3 Hexanol	Al2	1.6E-02	1.6E-02	1.6E-02
	1 Pentanol	Al3	1.2E-04*	3.3E-03*	3.3E-03*
	3-Methyl-3-buten-1-ol	Al4	8.0E-01	4.3E-01	8.2E-01
	3-methyl-2-Buten-1-ol	Al6	1.8E-01	4.4E-01	9.6E-01
	1-Hexanol	Al7	3.0E-01	8.4E-01	8.8E-01
	trans 3-Hexen-1-ol	Al8	1.7E-01	3.1E-01	3.9E-01
	cis 3-Hexen-1-ol	Al9	1.0E+00	1.2E-01	8.7E-01
	cis 2-Hexen-1ol	Al10	8.5E-01	4.7E-01	9.8E-01
	1-Octanol	Al11	6.6E-01	3.0E-01	7.0E-01
	Nonanol	Al12	8.1E-01	1.3E-01	1.8E-01
	Benzene methanol	Al13	7.4E-01	3.2E-01	6.4E-01
	1-Dodecanol	Al14	7.8E-01	5.4E-02	7.9E-01
Aldehydes					
	Butanal 3-methyl	Ald1	1.7E-01	1.7E-01	1.7E-01
	Hexanal	Ald2	4.1E-01	3.6E-01	8.3E-01
	2-Hexenal	Ald3	4.0E-01	4.4E-01	8.0E-01
	Octanal	Ald4	1.1E-01	3.9E-02	3.9E-01
	Nonanal	Ald5	7.4E-01	7.4E-02	8.0E-01
	Decanal	Ald6	3.0E-02	7.1E-01	7.1E-01
	Benzaldehyde	Ald7	1.4E-01	8.3E-01	8.6E-01
	Dodecanal	Ald8	4.8E-01	5.6E-01	8.8E-01
	Tetradecanal	Ald9	4.5E-02	1.8E-01	1.2E-01
Ketones					
	2 pentanone 4 methyl	K2	4.9E-02	1.6E-01	5.8E-01
	γ Butyrolactone	K4	6.4E-07*	6.4E-07*	6.4E-07*
	2-Dodecanone	K5	7.4E-03*	5.6E-01	5.6E-01
Terpenes					
	dl-Limonene	T1	3.5E-03*	8.9E-01	8.9E-01
	Linalool	T3	8.3E-01	2.9E-01	3.1E-01
	α Terpineol	T4	6.0E-01	4.7E-02	2.1E-02
Others					
	Formamide N,N-dibutyl	O2	7.2E-01	5.5E-01	7.4E-01
	Benzothiazole	O3	3.3E-01	5.5E-01	8.9E-02

(Continued)

TABLE 2 | Continued

Type	Name	Code	p[storage condition]	p[storage days]	p[interaction]
E-nose sensors					
	S1		2.3E-07*	2.3E-12*	4.7E-11*
	S2		8.5E-10*	9.8E-13*	1.2E-11*
	S3		1.9E-06*	9.2E-13*	3.5E-11*
	S4		4.4E-09*	3.2E-10*	5.2E-09*
	S5		1.4E-05*	8.2E-13*	1.1E-10*
	S6		2.4E-07*	1.7E-13*	5.2E-09*
	S7		7.3E-03*	1.3E-09*	2.4E-09*
	S8		8.5E-07*	2.9E-13*	2.1E-09*
	S9		1.9E-08*	1.4E-13*	3.3E-11*
	S10		5.3E-03*	1.6E-01	8.2E-01



correlation coefficient has been calculated for all pairs of E-nose data and VOCs and the resulting correlation matrix has been investigated by heatmap using a clustering procedure based on Euclidean distance and Ward's method.

Data analysis has been performed by in house R-function implemented using the R 4.0.4 platform (R Foundation for Statistical Computing, Vienna, Austria).

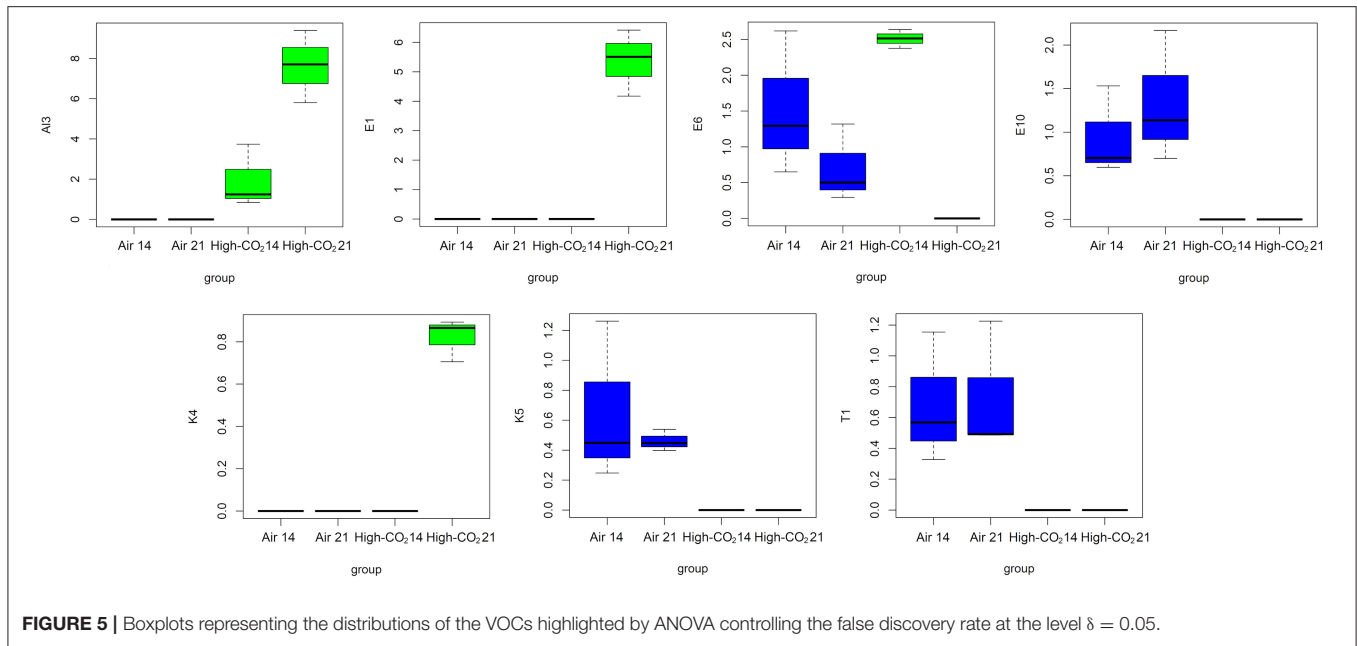
RESULTS AND DISCUSSION

Changes in Quality Parameters of Sweet Cherries Cold Stored in High-CO₂ or Air

Fruit stored in high-CO₂ reported a mean value of the visual quality significantly higher than control, with a quality

decrease during storage from 5 to roughly 2.5 at the end of the storage (Figure 1, Table 1). Changes in visual quality during storage can be mainly related to the browning of the skin and a significant increase in berry deformation (Table 1). This last parameter resulted higher in air than in high-CO₂ fruit, influencing the sensory evaluation of the visual quality. This result confirmed the positive effect of high-CO₂ MA storage in the preservation of quality of sweet cherry cv. "Ferrovia," as previously reported in the study by Cozzolino et al. (12).

For fresh fruit the initial mean values of antioxidant activity, total phenols and total sugar content were 63.6 ± 7.2 mg Trolox 100 g^{-1} fw, 130.3 ± 6.2 mg gallic acid 100 g^{-1} , and 21.04 ± 2.0 g glucose 100 g^{-1} , respectively. Except for the antioxidant activity that significantly increased during the storage of about



25% in both storage condition, the other parameters were kept unchanged during the experiment and were not affected by the storage condition (Table 1).

Changes in E-Nose Responses of Sweet Cherries Cold Stored in High-CO₂ or Air

Five biological samples have been collected for fresh produce and each combination of storage condition (air or high-CO₂) and storage days (14 and 21 days at 5°C). Since six technical replicates have been considered for each biological sample, a data set composed of 150 observations was obtained.

In Figure 2, the typical E-nose responses to Fresh (A) and cold-stored sweet cherry samples in air (B) or high-CO₂ (C) at 21 days are represented. Each curve displayed the response values, expressed as the ratio G/G_0 (vertical axis) of the relative sensor which changed with time (horizontal axis). The signals trend in all samples was almost similar. The response values of S8 (sensitive to broad-alcohol) appeared significantly increased, followed by S6 (sensitive broad-methane) and S2 (broad range). On the contrary, the signals of the sensors S4, S6, and S10 were sensitive to hydrogen, broad-methane, and methane-aliph, respectively. They did not show significant variations, keeping the value of G/G_0 around or less than one. Finally, the sensors S1, S3, S5, and S9, which were all responsive to aromatic components, regularly declined slightly. Comparing the three panels of Figure 2, the values of S8, S6, and S2 were different and indicate that sweet cherries stored in high-CO₂ (C) could produce more aroma components than fresh (A) and air (B) samples.

The data set composed of 150 observations and 10 features has been autoscaled and investigated by PCA. A model with two principal components explaining 97% of the total variance has been obtained. The biplot of the model is reported in Figure 3A.

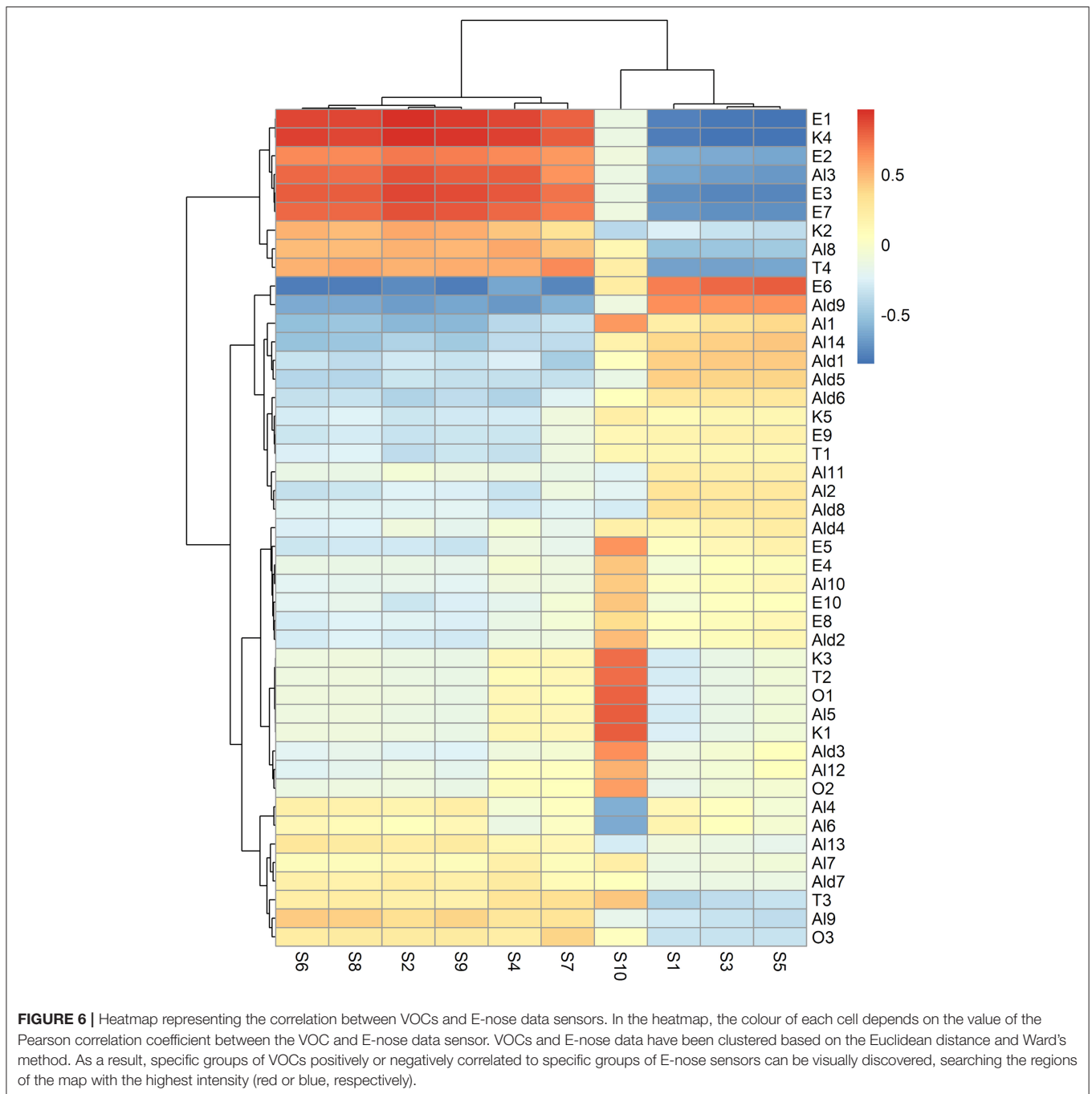
Fresh fruits were spread along the second component being characterised by high responses of the sensor S10, whereas the other observations were spread along with the first component. The responses of the sensors S1, S3, and S5 had decreased from 14 to 21 days, while the responses of sensors S2, S4, S6, S7, S8, and S9 showed an opposite trend. Observations with the same storage days and storage conditions were represented by points close together in the plot.

To better investigate the effect of the storage condition and the storage days on the E-nose responses, a PLS model has been built considering the storage condition and the time of storage as responses and the E-nose data as predictors. In the analysis, the fresh fruits were excluded. Considering autoscaled data, a PLS model with 3 latent variables, $R^2Y = 0.75$ and $Q^2Y = 0.59$ has been obtained. The model passed the permutation test on the responses (1,000 random permutations). The biplot of the model is reported in Figure 3B. The effect of storage condition was mainly explained by the behaviour of the S10 sensor whereas the data variation of the other sensors was mainly related to the effect of the storage days. The trend discovered by PCA for the responses of sensors S1, S2, S3, S4, S5, S6, S6, S7, S8, and S9 was confirmed by PLS analysis.

Projection to latent structure findings were also confirmed by ANOVA performed on the E-nose data, which highlighted that all the sensors are significantly affected by storage condition and storage days except for S10 that is influenced only by storage condition (Table 2).

Changes in VOCs of Sweet Cherries Cold Stored in High-CO₂ or Air

Three biological samples have been collected for fresh fruit and each combination of storage condition (air or high-CO₂)



and storage days (14 and 21 days at 5°C) obtaining a data set composed of 12 observations and 45 VOCs.

Applying autoscaling to the data, a PCA model with two principal components has been obtained. The model explained 54% of the total variance. The biplot is reported in **Figure 4A**. Fresh fruits and samples packed in high-CO₂ for 21 days were evidently separated from the other samples being characterised by specific VOCs. Sweet cherries stored in air or Hhgh-CO₂ for 14 days were located

in the centre of the plot and are not characterised by specific VOCs.

The effects of storage condition and storage days on the VOCs have been investigated by PLS analogously to the case of E-nose data. Considering autoscaled data, a PLS model with two latent variables, $R^2Y = 0.81$ and $Q^2Y = 0.62$, has been obtained. The effects of storage condition resulted better explained than in the case of E-nose ($R^2Y = 0.84$ and $Q^2Y = 0.67$ for VOCs and $R^2Y = 0.55$ and $Q^2Y = 0.43$ for E-nose data). The model passed the

permutation test on the responses (1,000 random permutations). The biplot of the model is reported in **Figure 4B**. The storage condition affects mainly E1, E3, E7, E9, E10, T1, K4, K5, Ald6, and Al3, whereas E6 is influenced mainly by the storage days.

Analysis of variance performed on VOCs highlighted that E1, Al3, and K4 were significantly affected both by storage condition and by storage days. Moreover, E10, K5, and T1 were significantly affected by storage conditions, and that E6 is significantly affected by storage days (**Table 2**). The distributions of the selected VOCs are reported in **Figure 5**.

Correlation Analysis of E-Nose Data and VOCs

Forty-five VOCs, which were overall identified in the study by Cozzolino et al. (12), were determined in the headspace of fresh, air, and high-CO₂ samples during the storage (**Table S1**). These VOCs were correlated with the E-nose sensors data. The heatmap, presented in **Figure 6**, showed that the data of the sensors S2, S4, S6, S8, S7, and S9 had positive correlations with ethyl acetate (E1), ethyl 2-butenate (E2), ethyl hexanoate (E3), ethyl benzoate (E7), 1-pentanol (Al3), *trans* 3-Hexen-1-ol (Al8), Terpineol (T4), and γ -Butyrolactone (K4), respectively. The response of the sensors S2, S4, S6, S8, S7, and S9 were in agreement with the VOCs pattern analysis which indicated that high-CO₂ treatment seems to preserve the quality of the packaged fruit, since ethyl esters and γ -butyrolactone produced by the fermentative metabolism were observed in high-CO₂ fruit only at the end of the storage (21 days), as reported in **Table S1**.

The sensor S1 was negatively correlated to *trans* 3-Hexen-1-ol (Al8), while the signals of S2 and S4 were indirectly associated to tetradecanal (Ald9) (**Figure 6**). These findings demonstrated that S1, S2, and S4 were found to correlate aliphatic compounds, such as esters (E1, E2, E3, and E7), alcohols (Al8), and aldehydes (Ald9).

The signal of the sensor S10 was directly correlated to 14 volatiles, including 1-hexyl acetate (E4), 2-hexen-1-ol acetate (E5), *trans* 2-Hexenyl hexanoate (E8), Isopropyl laurate (E10) *cis* 2-penten-1-ol (Al5), *cis* 2-hexen-1-ol (Al10), Nonanol (Al12), hexanal (Ald2), 2-Hexenal (Ald3) 3-pentanone (K1), 1-penten-3-one (K3), ocymene (T2), 2-methylfuran (O1), and formamide N,N-dibutyl (O2) (**Figure 6**). These results could suggest that the S10 signal could discriminate potential odorant markers of freshness in “Ferrovia” sweet cherries since all these volatiles were present in the fresh sample. In addition, most of them (Al5, Al10, Ald12, K1, K3, Ald2, Ald3, T2, and O1) have been previously found to be directly correlated to the sensory attribute of *Herbaceous Smell* (11).

On the contrary, the S10 signal exhibited a negative correlation with 3-methyl-3-buten-1-ol (Al4) and 3-methyl-2-buten-1-ol (Al6) (**Figure 2**). This result supported the above finding, as both the alcohols which are always absent in the fresh fruit, are by-products of alcoholic fermentation and have been previously reported to be inversely related to the sensory trait of *Herbaceous Smell* (12).

The ester 2-hexenyl butyrate (E6) was directly associated with S5 but negatively correlated to S2, S4, S6, S7, S8, and S9. This

is in agreement with the fact that E6, in contrast to the volatiles positively correlated to S2, S4, S6, S7, S8, and S9, was detected in the fresh sample, but was completely absent at the end of the storage in high-CO₂ (**Table S1**, **Figure 2**).

The results of the correlation between HS SPME and GC-MS and E-nose data showed that E-nose signal responses were in agreement with the VOCs profile analysis. Additionally, our findings demonstrated that the E-nose could discriminate fresh, air, and high-CO₂ sweet cherry samples by responding to specific flavour profiles of the samples.

CONCLUSION

The present study showed that E-nose is a promising tool in distinguishing sweet cherries cv. “Ferrovia” stored in high-CO₂ or air (control) for up to 21 days. Indeed, the variations of the response of the E-nose sensors depended both on the condition and the day of storage. The relationships between volatile profiles and E-nose signals have been investigated and the E-nose response has been associated with well-defined sets of VOCs. In particular, the signal of the sensor S10, resulting in association with VOCs which are considered putative markers of freshness, might be used for a rapid assessment of the product quality. Finally, since E-nose data were in agreement with the results previously obtained by HS SPME and GC-MS, commercial E-nose may be used in monitoring changes in aroma fingerprint during fruit storage instead of the expensive laboratory equipment currently used. These features and the non-destructive character of the analysis make E-nose useful for real-time physiological evaluations and quality control for industries in which rapid analysis is required.

DATA AVAILABILITY STATEMENT

The raw data supporting the conclusions of this article will be made available by the authors, without undue reservation.

AUTHOR CONTRIBUTIONS

RC and MC designed experiments. BP, CL, and MP carried out the experiments. RC, MC, and MP analysed the experimental results and wrote the manuscript. MC and MS performed the statistical analysis. All authors contributed to the article and approved the submitted version.

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SUPPLEMENTARY MATERIAL

The Supplementary Material for this article can be found online at: <https://www.frontiersin.org/articles/10.3389/fnut.2021.720092/full#supplementary-material>