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Formation of low- and high-spatial frequency laser-induced periodic surface structures (LIPSSs) in ALD-deposited MoS₂

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The formation of laser-induced periodic surface structures (LIPSSs) on the atomic layer-deposited (ALD) molybdenum disulfide (MoS₂) upon femtosecond laser processing is studied experimentally. Laser-processing parameters such as average laser power and the scan speed at which the formation of the periodic nanostructures takes place are identified. Optical and scanning electron microscopy are applied to identify the parameter regions for the different LIPSS formations and transitions between them. High- and low-spatial frequency LIPSS (HSFL and LSFL) with two distinct periods $\lambda_{LSFL} \approx 1.1 \,\mu m$ and $\lambda_{HSFL} \approx 83 \,nm$ can be observed. The HSFL are dominating at higher and the LSFL at lower laser average powers. Formation of LIPSS is found to inhibit laser ablation at lower scan speeds.

KEYWORDS

laser-induced periodic surface structures, molybdenum disulfide, femtosecond laser, atomic layer deposition, 2D materials

1 Introduction

Recent growing interest in 2D transition metal dichalcogenide (TMDC) materials such as molybdenum disulfide (MoS₂) is driven by a large number of applications such as photocatalysis, sensing and disinfection (Wang and Mi, 2017), semiconductor electronics (Butler et al., 2013), and energy storage (Acerce et al., 2015). The possibility to modify such thin layers with lasers enables their application for electronic devices (Radisavljevic et al., 2011). Two important types of the MoS₂ layer modifications required for all-MoS₂ electronic and sensing devices are crystallization (Zhang et al., 2020) and patterning. Structuring of thin MoS₂ layers can be conducted by direct laser ablation (Pan et al., 2020), which can be used, e.g., to write periodic stripes with the resolution down to approximately 250 nm (Kim et al., 2016). Such a deterministic way of patterning has certain disadvantages, like slow processing velocity and high requirements for the positioning precision and temperature stability. An alternative way for MoS₂ nanopatterning could be self-organization, i.e., appropriate choice of the processing parameters at which the homogeneously deposited layer undergoes some instability, manifesting itself as a periodic pattern. Using this method, self-organized hexagonal patterns were recently observed in MoS₂ upon fs laser processing (Becher et al., 2023).

LIPSSs are self-organized structures with periods usually comparable to the wavelength of laser radiation (LSFL - lowspatial frequency LIPSS) or approximately one order of magnitude smaller than the wavelength (HSFL — high-spatial frequency LIPSS) (Bonse et al., 2012). The physical mechanism of the LIPSS formation is still under debate. Most probably the LSFL formation is a multistep process (Gurevich et al., 2020), starting with the incident wave interfering with the wave scattered by surface roughness (Emmony et al., 1973; Sipe et al., 1983) and periodically modulating the temperature of the electrons in a thin surface layer. Then, the electrons heat up the lattice, whereupon the amplitude of the relative temperature modulation increases (Gurevich et al., 2017). This periodic lattice temperature modulation forms the LIPSS either by selective ablation or by hydrodynamic flow in the melt. The formation mechanisms of the HSFL are less clear and can involve either near-field amplification of the field on the LSFL structure (Geng et al., 2023) or de-wetting of the grooves formed by the LSFL ribbons (Khare et al., 2007). Such patterns were first discovered in germanium (Birnbaum, 1965) and were recently observed in 2D materials, such as graphene (Beltaos et al., 2014; Kasischke et al., 2018). However, to the best of our knowledge, no high- and lowspatial frequency LIPSS have been observed in MoS₂ simultaneously, and the corresponding laser processing parameters have not been studied systematically. Recently, formation of HSFL-like structures in MoS₂ referred to as *nanoribbons* was reported by Zuo et al. (2019). Zuo et al. (2019) fabricated large arrays of regularly arranged MoS₂ nanowires using the femtosecond laser technique on exfoliated flakes. Based on these nanoribbons, they produced macroscopic FET structures that exhibit rectifying behavior. Salimon et al. (2023) were able to demonstrate significantly improved photo-conductivity and photo-current on 200-nm ribbons fabricated via LIPSS. Furthermore, the nanopatterning of 2D materials improves the performance of switches and sensors (Rahman and Lu, 2022). In addition to the electrotechnical applications, MoS₂ is widely used as a lubricant. For example, MoS2 nanocoatings are used in space applications as a solid-based lubricating material (Mukhtar et al., 2023). Nanostructuring using LIPSS could further improve the tribological properties of these coatings (Bonse et al., 2018). In this paper, we demonstrate direct formation of LSFL and HSFL in ALD-deposited MoS₂ layers irradiated with femtosecond laser pulses and investigate the parameters at which the LIPSS formation occurs.

2 Materials and methods

2.1 Atomic layer deposition of MoS₂

A plasma-enhanced ALD (PE-ALD) process (Sharma et al., 2018; Jagosz et al., 2022) was used to deposit a MoS₂ film on a silicon dioxide (d = 200 nm)/silicon substrate. The substrate temperature was $T_{sub} \approx 230$ °C due to limited thermal contact between the substrate and table (T = 300 °C). The MoS₂ film was deposited using a SENTECH Instruments GmbH SI ALD reactor equipped with a remote capacitive coupled plasma source (f = 13.56 MHz). The precursor Bis(t-butylimido) bis(dimethylamino)molybdenum (Strem, 98%) was heated in a stainless-steel canister to T = 50 °C. A 100-sccm nitrogen

bubbling flow rate was applied to facilitate precursor delivery into the reactor. A plasma mixture of 5 sccm hydrogen sulfide (99.5%) and 45 sccm argon was used as the co-reactant. The plasma was operated at a power of 200 W at a pressure of 10 Pa. After 400 cycles, a 34-nm-thick polycrystalline MoS_2 film (verified by ellipsometric measurements) was deposited.

2.2 Generation of LIPSS

A femtosecond laser system from Jenoptik (JenLas D2.fs, $\lambda = 1025 \text{ nm} \pm 10 \text{ nm}$, and $\tau_P = 400 \text{ fs}$) with a repetition rate of $f_{rep} = 200 \text{ kHz}$ was used for the processing of the MoS₂ samples. The laser power was adjusted by a half—wave plate and a polarization beam splitter. The laser beam is led to a galvanometer scanner and focused on the sample surface with an F- Θ lens, with a spot size of $d \ge 27 \mu \text{m}$. Rectangular areas ($250 \mu \text{m} \times 250 \mu \text{m}$) were processed by scanning the spot with a line distance of 5 μm .

2.3 Characterization

Optical micrograph images were captured with a camera [mvBlueFOX-ML/IGC, (Matrix Vision GmbH)] in combination with a $\times 100$ NA 0.85 objective. Scanning electron microscope (SEM) images were captured with a Schottky field emission electron microscope Leo Gemini 982 at 2 kV.

3 Results and discussion

For the generation of LIPSS, square areas of $250 \ \mu m \times 250 \ \mu m$ were processed on the MoS₂ film. Depending on the chosen parameters, seven different topographies could be determined in the processed areas. Starting at P_L = 20 mW and v_{scan} = 0.1 mm/s, an unstable generation of LIPSS appears. With the increasing laser power (P_L < 50 mW) and scan speed (v_{scan} < 100 mm/s) stable, areal LIPSS starts to form. Further increase of the laser power results in modification of the film without formation of LSFL, while further increase of the scan speed still results in a formation of LSFL, but MoS₂ starts to get ablated. A possible explanation of the latter effect will be given at the end of this section. Figure 1 gives an overview about the different regions, depending on the laser power and scan speed.

A more detailed view of the MoS₂ LIPSS is shown in Figure 2. Here, a processed area ($P_L = 40 \text{ mW}$ and $v_{scan} = 0.5 \text{ mm/s}$) is displayed, with different magnifications. Figure 2A shows a continuous pattern of LSFL with a period of $\Lambda \approx 1.1 \mu \text{m}$, which is close to the laser wavelength of $\lambda = 1.03 \mu \text{m}$. Upon closer examination at higher magnifications (Figure 2B,C), it becomes apparent that the HSFL also forms in an orthogonal pattern to the LSFL. The period of the HSFL is around $\Lambda = 83 \text{nm}$. These "nanoribbons" were described by Zuo et al. (2019) in the past but without the appearance of LSFL. Figure 3 demonstrates the transition between the appearance of LSFL and HSFL, by adjusting the laser power at a constant scan speed ($P_L = 30-90 \text{ mW}$ and $v_{scan} = 0.5 \text{ mm/s}$). Processing the MoS₂ film with low laser powers ($P_L = 30 \text{ mW}$), an LSFL pattern forms, whose stripes consist of



FIGURE 1

Overview of the generated MoS_2 surface, depending on the laser power and the scan speed. The first appearance of unstable LIPSS (green rectangles) is visible at $P_L = 20$ mW and $v_{scan} = 0.1$ mm/s. With higher laser powers and scan speeds, a region is formed, where stable and areal LIPSS are generated (red stars). By increasing only the scan speed, partial ablation of the MoS_2 film and the LIPSS is induced (blue crosses). At lower scan speeds but higher laser powers, the formation of LIPSS disappears in the optical micrograph and a different modification of the MoS_2 film appears (orange triangles).



FIGURE 2

SEM images of LIPSS generated with P_L = 40 mW and v_{scan} = 0.5 mm/s with (A) 10.000x, (B) 50.000x, and (C) 100.000x magnification. In (A) and (B), the LSFL are seen, with a period of approximated $\Lambda \approx 1.1 \,\mu$ m. Images (B) and (C) show that the LSFL consists of HSFL with a period of approximate $\Lambda \approx 83$ nm.

the HSFL pattern (seen in Figures 2, 3A). With the rise of the laser power above $P_L = 50$ mW, the LSFL pattern fades away and only the HSFL patterns are formed. Figures 3B,C show that the degree of ablation and a resulting covering of the MoS₂ HSFL can be adjusted, based on the laser power. More SEM images of the laser power-depending generation of HSFL and LSFL can be found in Supplementary Figure S1.

The reason why the HSFL dominates at higher laser power is not fully understood but can be linked to the theory of LIPSS formation. The commonly used theory suggests that the periodic pattern appears due to interference between the incident and the surface-scattered waves (Emmony et al., 1973; Sipe et al., 1983). In the later steps of this multi-step formation process (Gurevich et al., 2020), the interference between these waves modulates first the electron and then the lattice temperatures, and causes surface modification. This is shown in Figure 3A, in which two perpendicular periodic patterns can be seen: a modified (as HSFL) MoS₂ film in the nodes and an unmodified MoS₂ film in the anti-nodes of the temperature profile. Why the modified MoS₂ layer transforms to HSFL is not clear and out of scope of this paper. However, as the laser power and the pulse energy increase, the amplitude of the field in the anti-nodes grows and the area at which the HSFL formation is reduced grows too. This leads to an increased area with thinner HSFL and a decreased area with thicker HSFL-patterned MoS₂ (see Figure 3B). As the pulse energy



FIGURE 3

SEM images of the evolution of HSFL and LSFL depending on the average laser power with a constant scan velocity $v_{scan} = 0.5$ mm/s. (A) With a laser power of $P_L = 30$ mW, both HSFL and LSFL start to build up. The inset image (50.000x magnification) shows that the LSFL consists of HSFL with different thicknesses. (B) When the laser power is risen to $P_L = 60$ mW, no LSFL is built and only the HSFL is formed in the MoS₂ film. (C) An even further rise of the laser power to $P_L = 90$ mW leads to partial ablation of the MoS₂ HSFL, resulting in a more island-like HSFL appearance.



FIGURE 4

SEM images of the evolution of HSFL and LSFL depending on scan speed with a constant average laser power $P_L = 50 \text{ mW}$. (A) A slow scan speed ($v_{scan} = 0.5 \text{ mm/s}$) forms a continuous LSFL pattern with an almost closed MoS₂ film consisting of HSFL. (B) An increase in the scan speed to $v_{scan} = 10 \text{ mm/s}$, and the LSFL and HSFL are still formed. The increase in the scan speed starts to separate the HSFL region from each other through ablation of material along the LSFL. (C) Even higher scanning speeds ($v_{scan} = 125 \text{ mm/s}$) leads to more ablation of the LSFL and HSFL patterns in the MoS₂ film, leading to partial MoS₂ free space in the film.

and the field in the anti-nodes grow and reach the ablation threshold in the crests, the HSFL can be still formed between these crests, where the field of the interference pattern is between the ablation and the HSFL-formation thresholds. This makes the HSFL structure less regular and the LSFL can hardly be observed (see Figure 3C).

The influence of the scan speed at a constant laser power ($P_L = 50 \text{ mW}$) on the formation of LSFL and HSFL is displayed in Figure 4. The SEM image in Figure 4A shows that a scan speed of $v_{scan} = 0.5 \text{ mm/s}$ produces an almost homogeneous MoS₂ film consisting of HSFL with a superimposed LSFL pattern. Increased scan speeds of $v_{scan} = 10 \text{ mm/s}$ lead to a more pronounced LSFL pattern and a partial separation of the HSFL stripes in the MoS₂ film. Even higher scan speeds (up to $v_{scan} = 500 \text{ mm/s}$) still produce both HSFL and LSFL patterns but with partial ablation of the MoS₂ film, resulting in the formation of islands of the LIPSS pattern (see Figure 4C; Supplementary Figure S2). Interestingly, for fixed average laser power, ablation occurs at higher scan speeds (i.e., at lower energy deposition), the layer tends to form LIPSS without or with less ablation. The accumulation effect is well known in laser material

processing, but it usually facilitates the ablation when the number of pulses per cite grows (Mannion et al., 2004). The explanation of the negative accumulation effect (i.e., the incubation coefficient S > 1(Mannion et al., 2004)) observed in our experiments is not completely clear to us. The distance between the consequent laser pulses on the sample surface upon scanning can be calculated as $\Delta x = v_{scan}/f_{rep}$; thus, Δx corresponding to the critical scan speed separating the ablation ($v_{\textit{scan}} \gtrsim$ 100 mm/s) and the LIPSS formation $(v_{scan} \leq 50 \text{ mm/s})$ is comparable to the period on the LSFL observed in our experiments. However, this might be a coincidence since we cannot suggest any physical mechanism for such a negative accumulation effect. Another possible explanation may be partial recrystallization of the MoS₂ layer (Sládek et al., 2022; Becher et al., 2023) near to the center of the laser spot, which increases either the ablation threshold or the reflectance of the surface. For low scan speeds, the following pulse interacts with a partly recrystallized area and remains below the ablation threshold, whereas at higher speeds, the following pulse meets a pristine surface. Due to fluctuations either in the pulse energy or in the layer properties, the ablation threshold may be overcome so that randomly distributed ablation spots or completely

ablated areas appear; see Figure 1. A similar explanation has been given by Geng et al. (2022), who observed a competition between laser ablation and the chemical reaction (oxidation) in thin Si layers. The formation of silicon oxide protects the surface from ablation if the oxide layer is strong enough, i.e., at a lower scanning speed.

4 Conclusion

This work shows the formation of LSFL and HSFL in ALDdeposited MoS₂. We determine the parameters for the laser power and the scan speed of the galvo scanner for the LIPSS formation and observe different modes of LIPSS formation. Depending on the laser power, it is possible to choose between the formation of only HSFL or a combination of HSFL and LSFL. With the selection of the scanning speed, a degree of ablation/separation of the HSFL through the LSFL can be set. Previously, simultaneous formation of HSFL and LSFL in 2D materials was observed in laser-reduced graphene oxide (Kasischke et al., 2018). Further research has to be conducted to characterize the electrical and optical properties (Calvani et al., 2014; Maragkaki et al., 2020; Skoulas et al., 2021) of the LSFL and HSFL for integration in highly sensitive photo-detectors, innovative rectifier structures, complementary inverters, or thin-film transistors and the tribological properties for applications in nanostructured thin film coatings.

Data availability statement

The original contributions presented in the study are included in the article/Supplementary Material; further inquiries can be directed to the corresponding author.

Author contributions

MB, EG, and AO contributed to the conception and design of the study. MB and JJ performed the experiments. CB guided the

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Conflict of interest

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Supplementary material

The Supplementary Material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/fnano.2023.1227025/ full#supplementary-material

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