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Progress and prospects of recycling technology for carbon fiber reinforced polymer

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This paper provides a comprehensive review of current and prospective technologies for recycling carbon fiber reinforced polymer (CFRP), addressing the growing need for sustainable disposal and resource recovery driven by increasing CFRP applications. Four primary recycling methods are discussed: incineration, physical recycling, chemical recycling, and thermal recycling. Incineration, while offering energy recovery, is unsustainable due to limited material recovery and environmental pollution. Physical recycling, utilizing mechanical processes, is cost-effective but results in significant fiber degradation, restricting applications to low-value fillers. Chemical recycling, involving solvent or acid-based resin decomposition, can yield high-quality fibers but faces challenges regarding environmental impact, process efficiency, and economic viability. Thermal recycling, encompassing pyrolysis in various configurations (conventional, fluidized bed, and microwave), emerges as the most promising approach, offering reasonable recovery rates and acceptable fiber properties, albeit with concerns about energy consumption and potential fiber damage. The review critically evaluates each method based on recovery efficiency, recovered fiber quality, environmental impact, and economic feasibility. It highlights the need for future research focusing on developing greener chemical solvents and catalysts, optimizing thermal processes and exploring product valorization, and investigating novel recycling technologies such as supercritical fluids, bio-based methods, and electrochemical approaches. Furthermore, it emphasizes the importance of establishing comprehensive performance evaluation standards for recycled fibers, exploring surface modification techniques, and expanding application possibilities. Life cycle assessment, economic analysis, and strengthened collaborations among academia, industry, and government are also crucial for advancing CFRP recycling towards industrialization and promoting a circular economy within the composites sector.

KEYWORDS

carbon fiber reinforced composites, recycling technology classification, advantages and disadvantages analysis, current situation analysis, future prospects

Highlights

- Analyzes four CFRP recycling methods: incineration, physical, chemical, and thermal.
- Evaluates CFRP recycling: benefits, drawbacks, applications, and future prospects.

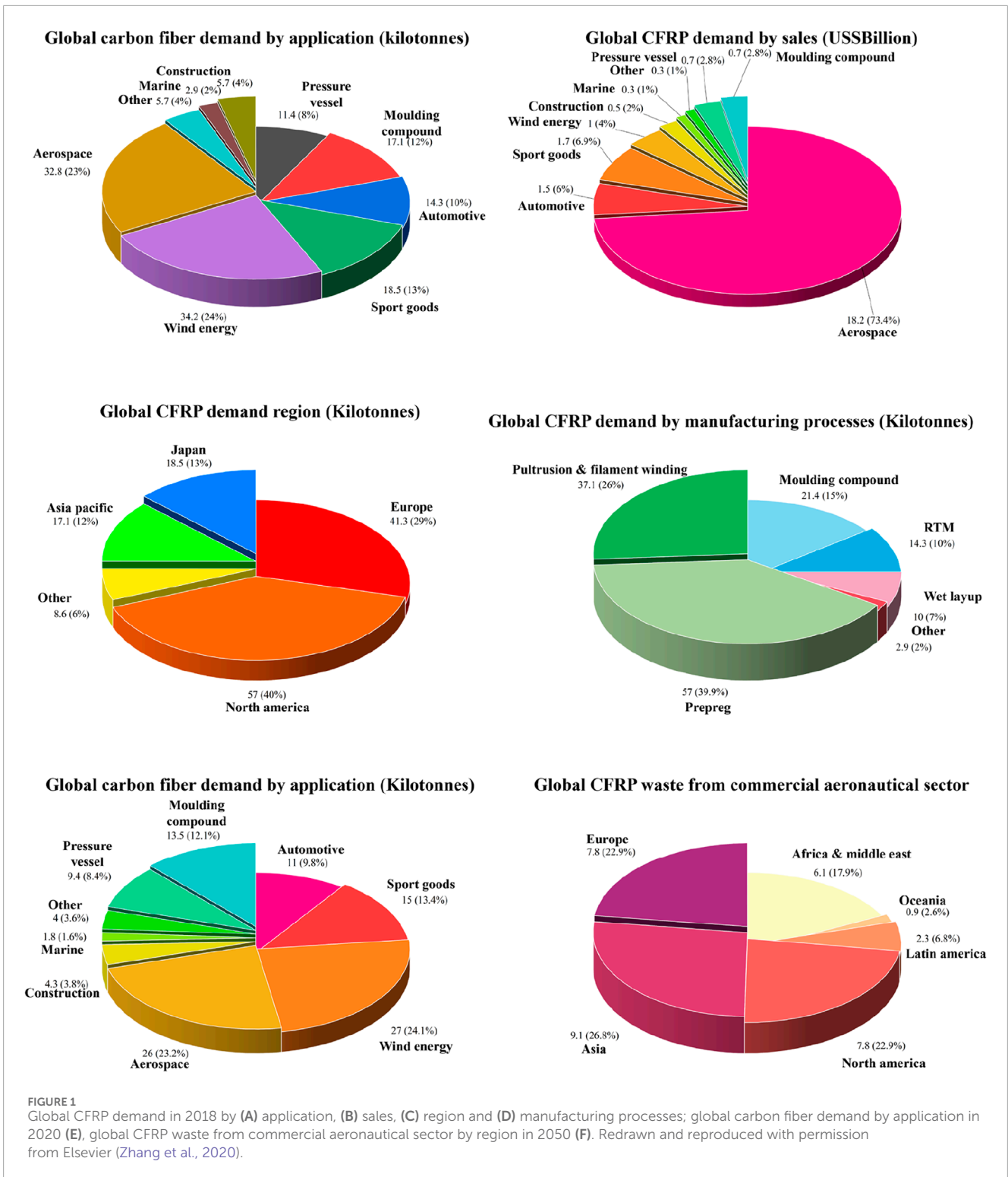
1 Introduction

CFRP, with its exceptional properties like light weight, high strength, corrosion resistance, fatigue resistance, excellent moldability, and design flexibility (Kojima and Furukawa, 1997; He et al., 2007; Liang qian, 2011; Gore and Kandasubramanian, 2018), is widely used in aerospace, new energy, automotive, rail transit, construction, and other fields (Jayalakshmi et al., 2019; Bintai et al., 2016; Xing, 2017; Das et al., 2019). Nevertheless, as with all materials, CFRP is subject to a finite lifespan. The long-term durability issues and eventual failure of CFRP inevitably result in a considerable amount of waste, which presents a significant challenge to the sustainability of the environment and natural resources.

Current waste management strategies for composite materials primarily depend on traditional methods such as incineration and landfilling, with landfilling being the main approach for most fiber-reinforced polymers (FRPs) (Naqvi et al., 2018; Krauklis et al., 2021; Deng et al., 2021). However, landfilling wastes substantial land resources and leads to soil and water contamination, while incineration contributes to greenhouse gas emissions (Li et al., 2016). Neither approach qualifies as recycling, as they fail to enable effective material recovery and reuse (Jang et al., 2020; Dong et al., 2015; Zhou et al., 2021). To underscore the urgency of this issue and the necessity of recycling, we examine global demand distribution, application areas, manufacturing processes, and waste generation patterns for carbon fiber and CFRP (Zhang et al., 2020) (Figure 1). As shown in Figure 1, wind energy (24.1%) and aerospace (23.2%) are the largest consumers of carbon fiber, where lightweight and high-strength materials are essential. However, these sectors also produce substantial waste. The aerospace industry alone accounts for 73.4% of total CFRP sales and generates a large volume of waste, primarily concentrated in Asia (26.8%), North America (22.9%), and Europe (22.9%). Recycling in these high-consumption regions could significantly reduce environmental impact and reliance on virgin materials, resulting in notable economic benefits. Geographical distribution also shapes recycling demand, with North America (40%) and Europe (25%) as primary markets and Asia (12%) following closely. Many countries now incentivize sustainable waste management, such as the UK's landfill taxes and Europe's landfill bans aimed at reducing municipal waste by 10% by 2030 (Shuaib and Mativenga, 2016; Bernatas et al., 2021). With increasingly stringent environmental regulations, the need for CFRP recycling and reuse grows more pressing, driven by high demand and substantial waste production. Recycling not only offers material cost savings but also promotes circular resource utilization, achieving both environmental protection and economic gains. Table 1 summarizes CFRP waste sources, types, and traditional treatment methods. CFRP recycling also faces manufacturing-related challenges. Prepreg (39.9%) and pultrusion/filament winding (26%) are dominant production methods, with prepreg recycling

being particularly complex due to its intricate structure. Given its extensive use in aerospace and wind energy, overcoming prepreg recycling obstacles could reduce waste and offer significant economic gains. Regulations like the European End-of-Life Vehicle Directive, which mandates the recovery or reuse of 85% of vehicle components by weight (Petraikli et al., 2020), and strict European Commission regulations on construction, vehicles, and electronics (Correia et al., 2011; Gharde and Kandasubramanian, 2019) are further driving the composites industry to adopt innovative recycling strategies. China's 2021 guidelines for establishing a green, low-carbon, and circular economy underscore the importance of CFRP recycling technology and reuse. These factors highlight the need for sustainable practices in the CFRP industry, with recycling as a key solution. As global demand for lightweight materials rises, CFRP production expands, while traditional recycling methods remain inefficient and costly. Therefore, developing an efficient, economical, and environmentally friendly recycling system is crucial. This will require increased investment in R&D for innovative recycling technologies, supportive government policies, and enhanced industry collaboration to create a robust CFRP recycling ecosystem. This comprehensive approach is essential not only for environmental protection but also for the long-term viability and sustainability of the CFRP industry. Compared to other fiber-reinforced composites, CFRP recycling poses greater challenges but also presents higher economic value. CFRP's superior strength, stiffness, lightweight characteristics, and corrosion resistance make it valuable in demanding applications like aerospace and automotive. While GFRP and BFRP perform well, CFRP's advantages in strength-to-weight and stiffness-to-weight ratios are significant. However, this performance also increases recycling complexity and potential rewards. CFRP's high initial cost limits widespread adoption, and its unique properties make it susceptible to performance degradation during recycling, impacting the market demand and value of recovered materials. Efficient CFRP recycling not only reduces pollution but also conserves raw materials, lowers production costs, and supports sustainable resource utilization, critical for high-end manufacturing. Effective CFRP recycling requires careful separation and processing of fibers and matrix to preserve fiber performance. Chemical and thermal recycling surpass traditional mechanical methods in achieving high recovery efficiency and minimizing fiber degradation. Advances in recycling technology could enable recycled CFRP use not only in non-structural applications but potentially in high-performance sectors. For high-end manufacturing, CFRP recycling significantly cuts production costs and improves resource utilization, fostering sustainable development. Research into CFRP recycling technologies that enhance efficiency and reduce performance degradation is essential to maximize resource circularity and environmental protection. Process optimization, cost reduction, and performance improvements in recovered materials will expand CFRP applications and contribute to sustainable high-performance materials.

GFRP, despite its valuable properties and broad applications, also faces sustainability challenges due to durability issues and waste accumulation. Research by Xian et al. (2024a), Xian et al. (2024b) on the long-term performance of GFRP rebars and plates in various environments reveals declines in mechanical properties due to water absorption, temperature, stress, and



hydrothermal conditions, predicting finite service life and eventual waste generation. Xian et al. (2024c) also notes the impact of manufacturing processes on GFRP rebar performance, where fiber twisting, interfacial debonding, and stress concentrations during bending reduce strength and increase failure risks, emphasizing the need for recycling. These studies affirm the inevitability of GFRP failure and waste generation, underscoring

the importance of efficient recycling strategies. Recycling GFRP promotes circularity and environmental protection by reducing virgin material reliance, minimizing pollution, and recovering valuable fiber and resin resources, aligning with circular economy principles. However, efficient GFRP recycling requires careful consideration of the resin matrix type's complexity and feasibility. Thermoset GFRPs, with their highly cross-linked structures (e.g.,

TABLE 1 Sources, types and conventional treatment of CFRP waste.

Point	Source (of information, etc.)	Form	Traditional methods
Composite moulding process	Test Material Non-conforming products Composite Edge Material	composite material	landfill incineration
Fabric manufacturing process	Edge material reject sth.	woven material	
Prepreg manufacturing process	Edge material Non-conforming products Deteriorated Failed Products	prepreg	
Carbon fiber composite products or products	Products and components that have reached the end of their useful life or have been scrapped	component	

epoxy), are challenging to re-melt and reprocess, posing significant recycling obstacles. Existing methods, such as mechanical grinding, pyrolysis, and chemical recycling, suffer from low efficiency, high energy consumption, and secondary pollution. Research on advanced recycling technologies like selective resin dissolution, supercritical fluid technology, efficient glass fiber separation, large-scale reuse, and new recyclable/degradable thermoset resins is crucial. In contrast, thermoplastic GFRPs (e.g., polypropylene-based) offer easier recycling through melt reprocessing and injection molding, retaining material properties and providing economic and environmental benefits. Optimizing mechanical recycling and reuse processes for thermoplastic GFRPs, improving recycled product performance, expanding applications, and developing high-performance recyclable thermoplastic resin matrices are key to enhancing circularity. Developing efficient, economical, and environmentally friendly GFRP recycling technologies with tailored strategies for thermoset and thermoplastic GFRPs is essential for sustainable development in the composites industry.

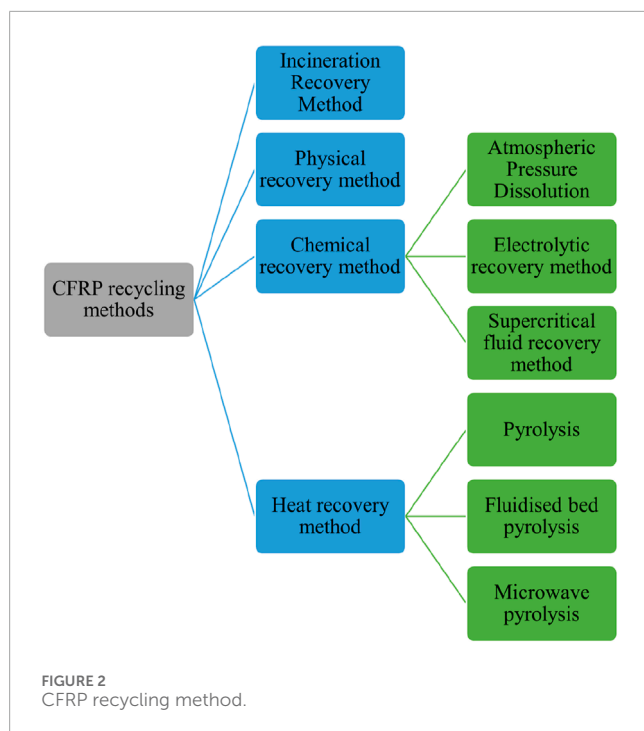
The novelty of this study lies in its systematic comparison of CFRP recycling methods and forward-looking perspective on future directions. We analyze four primary CFRP recycling methods (mechanical, pyrolysis, chemical, and solvent recycling) alongside recent advancements. Beyond traditional reviews, we detail each method's core principles, efficiency, cost-effectiveness, and environmental impact. This multi-faceted evaluation clarifies each method's suitability for specific applications and identifies bottlenecks. Additionally, we examine the integrated operational pathways of these recycling methods, providing guidance for future technology selection and optimization. This comprehensive comparison offers a unique perspective on CFRP recycling, setting the stage for future innovation and practical applications. Achieving these advancements requires collaboration among academia, industry, and government to support technological R&D, policy development, and industry synergy, establishing a comprehensive recycling ecosystem for both CFRP and GFRP.

2 Classification of recycling methods

The recycling of carbon fiber reinforced polymers (CFRPs) is closely linked to their distinct physicochemical properties. CFRPs,

typically produced through thermo-compression molding, consist of carbon fibers embedded in a resin matrix. This composite material combines excellent mechanical properties with significant anisotropy, low interlaminar strength, and notable differences in physicochemical properties between the fibers and the matrix. These characteristics pose challenges in processing, repair, and especially recycling. Epoxy resins, which make up approximately 80% of CFRP matrices, are favored for their heat, corrosion, and aging resistance, strong adhesion, and excellent mechanical properties. However, the irreversible cross-linking structure formed during curing renders epoxy resin insoluble and non-fusible, complicating CFRP recycling and reuse (Xuan-Jun et al., 2019; Wu et al., 2021; Mishnaevsky et al., 2017; Liu et al., 2022).

Currently, CFRP recycling methods are categorized into three main approaches: pyrolytic, mechanical, and chemical recycling (Figure 2). Table 2 outlines the process characteristics of these methods, comparing their working conditions, recovered fiber quality, and recycling benefits. The pyrolysis method degrades the resin matrix by heating it in various atmospheres to recover relatively pure carbon fibers. While simple and cost-effective, it is difficult to fully prevent oxidative damage to the fibers, even under inert conditions. Incomplete resin degradation and residual pyrolytic carbon on the fiber surface can also reduce the mechanical properties of recycled carbon fiber (rCF). The mechanical recycling method involves shredding and grinding CFRP waste into powder, followed by a separation process to isolate fiber powder from resin powder. Though easy to operate, this method significantly weakens the mechanical properties of the fibers. Moreover, the rCF produced is typically in the form of short fibers or powder, limiting its applications primarily to sheet or block molding compounds or as fillers in building materials. The chemical recovery method employs solvents or acids, along with catalysts, temperature, and pressure, to dissolve the resin matrix and recover purer carbon fibers. This approach better preserves the mechanical properties of the fibers, and some studies have even achieved resin matrix recycling. However, it has notable drawbacks: (1) environmental impact, as the process often generates chemical pollutants harmful to the environment; (2) low recycling efficiency, since the resin dissolves from the surface inward, prolonging the process; and (3) high economic cost, due to the need for corrosion-resistant vessels and pollutant treatment.



In conclusion, each CFRP recycling method has its advantages and limitations, requiring selection based on specific circumstances. Future research should focus on developing more efficient, environmentally friendly, and cost-effective recycling technologies to promote resource recycling and sustainable development.

2.1 Incineration recovery method

The incineration recycling method is mainly to incinerate CFRP directly and then recover the remaining carbon fiber. Incineration was a common means of disposal of waste carbon fiber reinforced polymer (CFRP). In the early days, incineration was a common method for disposing of CFRP waste (Palola et al., 2022). However, incinerating 1 ton of CFRP releases up to 2011 kg of CO₂ (Chen et al., 2023), contributing significantly to greenhouse gas emissions and negatively impacting air quality (Shen et al., 2023; Höhne et al., 2023; Termine et al., 2023; Yuyan et al., 2009; Deng et al., 2023). As environmental awareness has grown, the incineration of CFRP waste has increasingly been restricted. The carbon fiber production process is energy-intensive and costly (Limburg et al., 2019; Deng et al., 2021), and large amounts of carbon fiber waste are generated annually. Therefore, recycling carbon fiber waste offers both environmental and economic benefits. Recycling can reduce pollution, alleviate waste disposal burdens, decrease reliance on new resources, conserve energy, and support the development of a circular economy. It is important to note that incineration, while providing energy recovery, does not involve material recycling. Although some inorganic residues from incineration may be used in the cement industry, this process does not qualify as material recycling. Nevertheless, certain municipal solid waste incineration facilities with high thermal efficiencies are often classified as “recycling” installations. However, a clear

distinction between “recycling” and “recovery” is made in the relevant European recycling directives. Currently, heat recovery technologies for CFRP waste are primarily limited to incineration and fluidized bed processes, with the latter showing promise for both incineration and pyrolysis applications. As a result, the prospects for using incineration for carbon fiber recovery are limited, and relevant studies remain scarce. Nanchang University of Aviation Yi can (Yi et al., 2013) and other use of incineration recycling method to recover carbon fiber, recycled carbon fiber combined with a new type of resin through the hand-lay-up molding to get the recycled composites, testing its tensile strength and other properties, found that only the raw material of about 35%–40%. It can be seen that this method is simple and low cost. But the recycling process will produce a large number of harmful substances, while the recycled carbon fiber is only 35%–40% of the raw material. The carbon fiber recovered by this method can only be used to produce a lower level of composite materials, and the strength and toughness of the composite materials produced are lower.

2.2 Physical recovery method

The main principle of the physical recycling method is through CFRP composites are first crushed into small pieces of 50–100 mm size and then ground by ball milling or fluidized bed reactor (Yang et al., 2015). The milled product consists of recyclates of different sizes, which can be classified into fibrous and powdery resin materials after sieving. The recycled fibrous product, which can be used as an additive or reinforcing component of low value-added composites (Ogi et al., 2007), is widely used in the field of infrastructure in materials such as concrete, synthetic panels, iron-making reductions, asphalt and so on (Verma et al., 2018; Meng et al., 2018). Figure 3 shows a schematic diagram of the physical recycling method. This method is only applicable to uncontaminated CFRP waste, and has the advantages of low cost, simple process and no pollutants, but the disadvantages are that the strength of the treated fibers will be severely reduced and the reuse value is low, and they can only be used as some fillers. Mechanical recovery currently faces limitations in fully separating reinforcing fibers from the polymer matrix, resulting in particles containing short fibers and residual matrix material (Rahimizadeh et al., 2020). While this process is cost-effective and has low carbon emissions, the degradation of fiber properties restricts its use in highly loaded structural applications (Rani et al., 2021). In the case of glass fiber reinforced plastic (GFRP) composites, the resin-rich fine powder obtained from recycling is often used as a filler in sheet molding compounds (SMC) or block molding compounds (BMC), replacing virgin fillers like calcium carbonate. Recycled materials can replace up to 10% of the virgin fillers without significantly affecting mechanical properties. The relatively low density of polymers compared to virgin fillers allows the use of recyclates to reduce the weight of the composition by about 5%. However, the low cost of virgin fillers makes this alternative economically uncompetitive (Pickering, 2006). SMC and BMC technologies are widely used in the automotive industry and comply with the 85% reuse criteria for end-of-life vehicles set by Directive 2000/53/EC (Fonte and Xydis, 2021). The automotive industry’s self-organized recycling system is expected to support recycling efforts, though this may limit the use of recycled materials in the wind power

TABLE 2 Process characteristics and parameters of current recovery methods.

Parameters	Method						
	Thermal (Howarth et al., 2014; Wu et al., 2022; Liu et al., 2021; Wang et al., 2019; Pickering et al., 2000)	Mechanical recovery method (Shuaib and Mativenga, 2016; Bernatas et al., 2021)	Chemical (Petrakli et al., 2020; Correia et al., 2011; Gharde and Kandasubramanian, 2019; Xuan-Jun et al., 2019; Wu et al., 2021; Mishnaevsky et al., 2017; LIU et al., 2022; Steven Montagna et al., 2023; Butenegro et al., 2023)				
	Pyrolysis	Microwave pyrolysis	Supercritical fluid recovery method				
	Atmospheric pressure dissolution						
Fiber properties	Size (mm)	1–100	5–10	1–100	0.001–50	1–50	unlimited
	Tensile strength (%)	6–25	73–99	10–82	50–65	95–100	81–90
	Tensile modulus (%)	50–80	87–119	100	Same as original	100	93–121
	Temperature (°C)	80–100	450–600	450–500	22–25	25–170	Normal
Recycling parameters	Pressure (MPa)	400–1,000	Normal pressure	Normal pressure	22–35	Normal pressure	Normal pressure
	Poisonous	No	No	No	No	Yes	Yes
	Products	Oil, carbon, gas	Oil, carbon, gas	Gas, Dust particles	Dust particles	Chemical waste liquids	Chemical waste liquids
Energy efficiency	Energy consumption (MJ/kg)	1–35	1–35	—	60–90	60–90	60–90
	Greenhouse gas emissions (kg/kg)	About 2	About 2	About 0	About 1.5	About 1.5	About 1.5

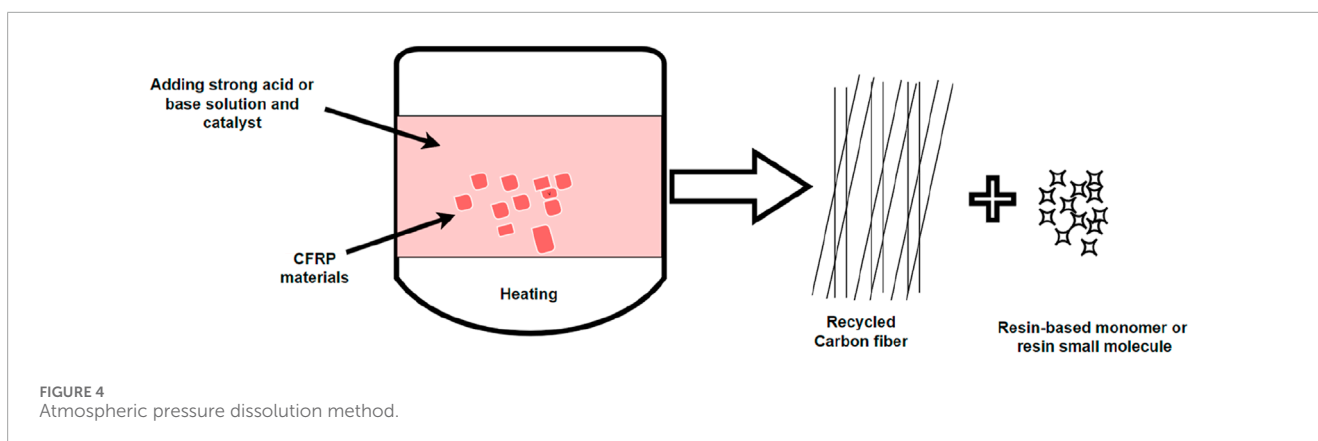
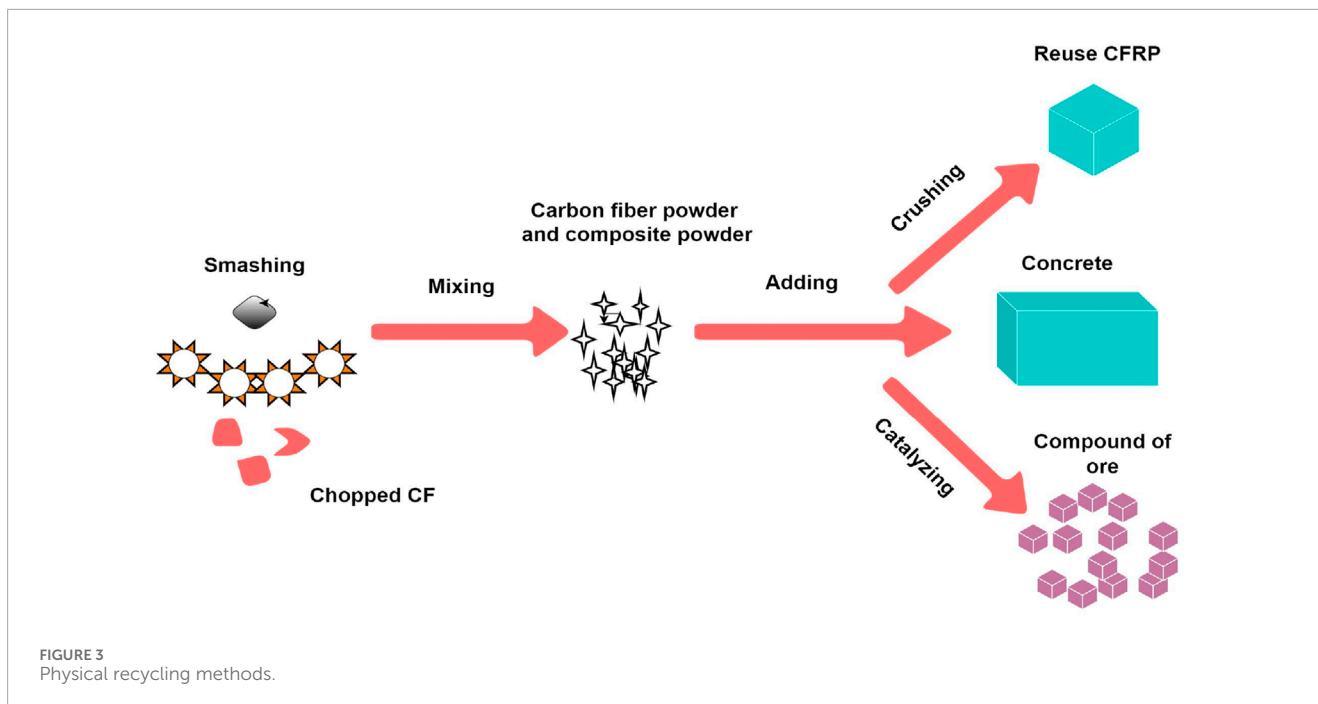
sector by other industries. For the wind power industry, recycling is a key goal, but mechanical recycling remains challenging due to the degradation of material properties. For example, recycled fiber-rich materials are of lower quality than virgin reinforcements, and their adhesion to the polymer matrix is poor. Larger particles can lead to stress concentrations, increasing the risk of failure (Pickering, 2006). Beauson et al. explored using recycled glass fiber composites from wind turbine blade load-bearing beams as reinforcing components, creating novel composites through a vacuum injection process. However, the resulting composites exhibited low breaking strength and deformation capacity, which was attributed to insufficient adhesion between the recycled material and the new matrix (Beauson et al., 2016). Furthermore, Palmer et al. emphasized that fine separation of recyclates and precise control during reformulation and production are essential for producing high-quality novel doughy molding compounds (DMCs). Examples of successful mechanical recycling applications include noise barriers made from debris materials by MILJØSKÆRM[®] (Miljoskarm). Studies have shown that replacing virgin reinforcement with recycled fibers in products can maintain reasonable mechanical properties, even with a high percentage of recycled fibers (Palmer et al., 2009). Additionally, mechanically recycled wind turbine blade waste has been used to enhance the mechanical properties of 3D-printed PLA materials (Rahimizadeh et al., 2019), improve UV resistance in wood coatings (Jensen and Skelton, 2018), and demonstrate superior mechanical properties in building materials and concrete applications (Ribeiro et al., 2015; García et al., 2014; Zhang et al., 2021; Asokan et al., 2010; Yazdanbakhsh et al., 2018). Although mechanical recycling has primarily focused on GFRP materials, some studies have also explored the recycling potential of CFRP. The high cost of carbon fiber and the degradation of recycled fiber properties present challenges for the application of mechanically recycled CFRP. Mamanpush et al. developed a composite panel combining recycled CFRP with thermoset and thermoplastic materials (Mamanpush et al., 2021), while Nassiri et al. found that adding chopped CFRP waste improved the durability of pervious concrete (Nassiri et al., 2021). Furthermore, adding GFRP and CFRP waste to pervious concrete pavements has been shown to improve flexural strength and enhance resource utilization of large quantities of composite waste, providing an effective solution to reduce landfill use (Singh et al., 2022).

2.3 Chemical recovery method

The chemical recovery method uses chemical degradation or chemical dissolution to disrupt the crosslinked structure in the CFRP to form low molecular polymers or partially dissolve in reagents, thus removing the resin matrix around the fibers and achieving the separation of the fibers from the resin matrix (Yi-Feng, 2013). Compared to the thermal recovery method, the chemical recovery method only needs to complete the separation of resin and fiber at a relatively mild temperature, with less resin residue on the surface of the recovered fiber and less reduction in the mechanical properties of the fiber. Therefore, chemical recycling method is the most popular and studied method for CFRP recycling. Chemical recovery methods mainly include atmospheric pressure

dissolution method, electrolytic chemical method, and supercritical fluid recovery method.

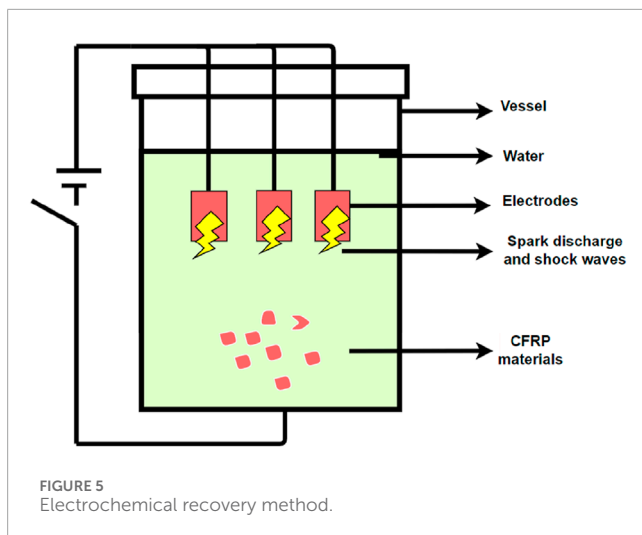
Atmospheric pressure dissolution method refers to the use of suitable strong acid and alkali solutions and catalysts to degrade the waste CFRP matrix into small molecules or monomers under heated conditions, so as to achieve the effective separation of the resin matrix and CF, and this method can even achieve the complete recycling of the two materials and then be used, Figure 4 shows the schematic diagram of atmospheric pressure dissolution method. Dissolution treatment at normal ambient pressure has the unique advantage of reducing damage to CF. Liu et al. utilized a mild chemical recovery method to recover CFRP at high T_g (>200 °C) using a ZnCl₂/ethanol catalyst (Liu et al., 2017). The lower treatment temperature (<200 °C) caused negligible damage to CF and the recovered CFRP maintained high strength and modulus at loadings up to 15 wt% after addition of other new epoxy resins (Lo et al., 2018). Wang's group developed an effective chemical recovery method for the selective cleavage of tertiary carbon-nitrogen bonds by using AlCl₃/CH₃COOH as a catalyst (Wang et al., 2015). By applying optimal recycling conditions (catalyst mass fraction of 15 wt%, 180°C, 6 h), up to 97% of the cured epoxy matrix was recycled and the recycled carbon fibers retained 98% of the tensile strength of virgin carbon fibers. Hou also investigated the recovery of CFRP from supercritical 1-propanol by adding 1 wt% of KOH additive to 1-propanol, and the final results showed higher recovery efficiency, but with a slight loss of mechanical properties of the carbon fibers (Yan et al., 2016). Ma and Nutt (2018) demonstrated that a mixture of acetic acid and hydrogen peroxide effectively degrades the GFRP matrix at 110°C under atmospheric pressure. However, this method has limitations, as it degrades the properties of the recovered glass fibers. In contrast Dang et al. (2005), Yuyan et al. (2006), achieved high-quality glass fibers by treating GFRP with nitric acid solution at lower temperatures, highlighting the superiority of nitric acid in low-temperature solvent cracking of GFRP. For CFRP recovery, various low-temperature solvent cracking methods have been explored. Among these, sulfuric acid and nitric acid solutions have been extensively studied and shown to yield significant results. Several studies (Feraboli et al., 2012; Liu et al., 2004; Lee et al., 2011) have demonstrated that sulfuric and nitric acids can effectively degrade the CFRP matrix and recover high-quality carbon fibers by controlling reaction conditions. Current research on low-temperature recycling focuses on developing environmentally friendly chemical solutions that minimize environmental impact while preserving fiber performance. For example, Zhao et al. (2022) proposed a low-temperature solvent cracking method using a mixture of ethanolamine and potassium hydroxide, successfully recovering carbon fibers with excellent performance under mild conditions. This approach offers a promising new direction for environmentally friendly chemical recovery technology. In addition to solvent cracking, oxidative degradation is another important chemical recycling route, utilizing oxidants to break down the polymer matrix and recover fibers. Xu et al. (2013) showed that hydrogen peroxide effectively degrades the CFRP matrix at low temperatures, with the recovered carbon fibers retaining more than 95% of their original tensile strength, demonstrating hydrogen peroxide's potential for low-temperature CFRP recycling. Furthermore Das et al. (2018), developed an advanced oxidative



degradation method that combines hydrogen peroxide with pure acetic acid. This method not only recovers carbon fibers from CFRP but also enables the multiple recycling of the polymer matrix and solvent, offering significant environmental and sustainability benefits. Above all, the atmospheric pressure dissolution method is simple to operate and although it can achieve the recovery of resin and carbon fibers, the whole reaction process is slow and the degradation mechanism is complicated. Moreover, the degradation solvent mainly uses very corrosive reagents such as strong acids, strong bases, strong oxidants and catalysts, which can easily cause environmental pollution. The core advantages of this technology are its low energy consumption and effective protection of fiber properties, making it a research hotspot in the field of composite recycling. But the reaction process is difficult to control. At present, the atmospheric pressure dissolution method is still in the experimental and small trial stage.

The electrochemical method uses the principle of electrocatalysis to achieve the degradation and recovery of CFRP

in an electrolytic solution, Figure 5 shows the schematic diagram of the electrochemical recovery method. Chen (2017) optimized the current and sodium chloride parameters and added a catalyst, and greatly shortened the time of CFRP recovery, and also improved the amount of carbon fiber recovery and tensile strength. Zhu et al. (2019) proposed an electrically driven non-homogeneous catalytic decomposition method, in which CFRP was immersed in a sodium chloride electrolyte solution with potassium hydroxide catalyst fused in it, and the CFRP was used as the anode and the stainless steel bar was used as the cathode, with the CFRP being the anode and the stainless steel bar being the cathode. The reaction between CFRP and the solution was catalysed by applying an electric current to the electrolyte under test conditions at temperatures ranging from 40°C to 75°C. Approximately 100% removal of the resin matrix was achieved, and the mechanical properties of the carbon fibers were maintained at more than 90%. The advantages of the electrochemical method lie in the overall simplicity of the procedure, the low difficulty in recovering CFRP, the high recovery efficiency



and the good strength of the recovered carbon fibers, but its disadvantages are also relatively significant, as the reaction process consumes a large amount of energy, which is mostly consumed by the electrolyte, resulting in a large amount of energy waste. Therefore, high energy consumption becomes the biggest limiting factor for the commercialization of the electrochemical method.

The supercritical fluid recovery method refers to the use of solvents such as water, methanol, or propanol as the fluid, under conditions exceeding its inherent critical temperature and pressure, so that the fluid is between liquid and gas and has the low viscosity, high mass transfer coefficient, and high diffusion coefficient of both gases and liquids, which can improve the reaction speed and efficiency of the degradation of the resin matrix as shown in Figure 6. Kim et al. (2019) used supercritical water to recover carbon fibers from commercial composites and achieved 99.5% degradation of the resin matrix by optimizing the recovery conditions. Supercritical water to recover carbon fibers from commercial composites, and the resin matrix degradation rate was up to 99.5% by optimizing the recycling conditions. Okajima et al. (2011) achieved the recovery of carbon fibers from CFRP as well as epoxy monomers in subcritical water at a temperature of 400°C and a pressure of 20 Mpa by using a concentration of potassium carbonate of 2.5% as a catalyst in the reaction, and the carbon fiber The mechanical properties were 85% of the original, and the recovery rate of epoxy monomer was 70.9%. Morales Ibarra et al. (2015) used supercritical water and subcritical benzyl alcohol to recover the resin matrix on the surface of CFRP, and the resin recovery rate reached 89.1% and 93.7%, respectively. Oliveux et al. (2012) observed a significant reduction in glass fiber tensile strength, up to 65%, when subcritical water hydrolysis was applied at 350°C. Kao et al. (2012) confirmed this result, attributing the strength loss to reactions between water and alkali metal oxides on the fiber surface, which led to microcrack formation. To address this issue and improve the quality of recovered fibers, researchers have investigated alternative supercritical solvents. Bai et al. (2010) showed that treating glass fiber reinforced composites with supercritical acetone retained up to 89% of the tensile strength, significantly higher than that achieved with near-critical water. This highlights the importance of solvent selection in preserving fiber properties during recycling.

Catalysts are often used in both supercritical and subcritical solvent hydrolysis to enhance recovery efficiency, but their use complicates downstream processing by requiring the separation of catalysts from the recovered fibers and matrix. To simplify the process and improve environmental compatibility Das et al. (2018), proposed a catalyst-free subcritical recycling method using D-limonene, a relatively benign solvent, for polyester-based GFRP composites. Their results showed that subcritical D-limonene treatment enabled the recovery of fibers retaining approximately 85% of their original strength. For epoxy-based composites Morales Ibarra et al. (2015) treated CFRP with supercritical water and subcritical benzyl alcohol, achieving decomposition rates of 89.1% and 93.7%, respectively, demonstrating the effectiveness of both techniques in epoxy resin degradation. Kim et al. (2019) used catalyst-free supercritical water to remove epoxy resin from CFRP, achieving over 99% cleanliness. Okajima et al. (2017) employed subcritical acetone for CFRP recycling, successfully recovering intact, high-quality carbon fiber fabric while preserving the structural integrity of the original fibers. Their results indicated near-identical tensile strength between the recovered and virgin fibers, further demonstrating the efficacy of subcritical acetone in maintaining fiber properties. Different chemical recycling methods offer various advantages and disadvantages depending on material type, solvent choice, and process parameters. Low-temperature solvent lysis and oxidative degradation are energy-efficient and cost-effective, making them suitable for recovering both glass and carbon fibers. While supercritical and subcritical solvent hydrolysis require higher capital investment in equipment, they provide superior performance in matrix decomposition and fiber strength preservation. Future research should focus on developing energy-efficient, environmentally benign solvents for a broader range of composite materials and further optimizing recovered fiber properties to support circular economy principles and sustainable material utilization. In summary, the supercritical fluid recovery method is more efficient, with low contamination and high resin removal rate, but the test conditions are very harsh, requiring reactions at high temperature and pressure, and the overall safety factor is low, therefore, at this stage, the recovery method is still immature and stays only in the laboratory stage. All these factors have caused a certain degree of hindrance to the bulk recovery of CFRP. In summary, the chemical recovery method can better achieve the separation of resin matrix and carbon fiber, with less resin residue on the surface of the recovered fiber and less reduction of the mechanical properties of the fiber. However, the reasonably efficient and pollution-free chemical recycling method has not been explored yet, so the chemical recycling method is the most concerned and researched method for CFRP recycling. The advantages and disadvantages of chemical recovery methods are summarized in detail in Table 3.

2.4 Heat recovery method

Thermal treatment is a method of high temperature degradation of waste composites using the high temperature resistance of carbon fibers to decompose the thermosetting resin into hydrocarbons, gases such as methane and low molecular weight carbon-containing materials, to obtain degradation products as oils, gases, and solid

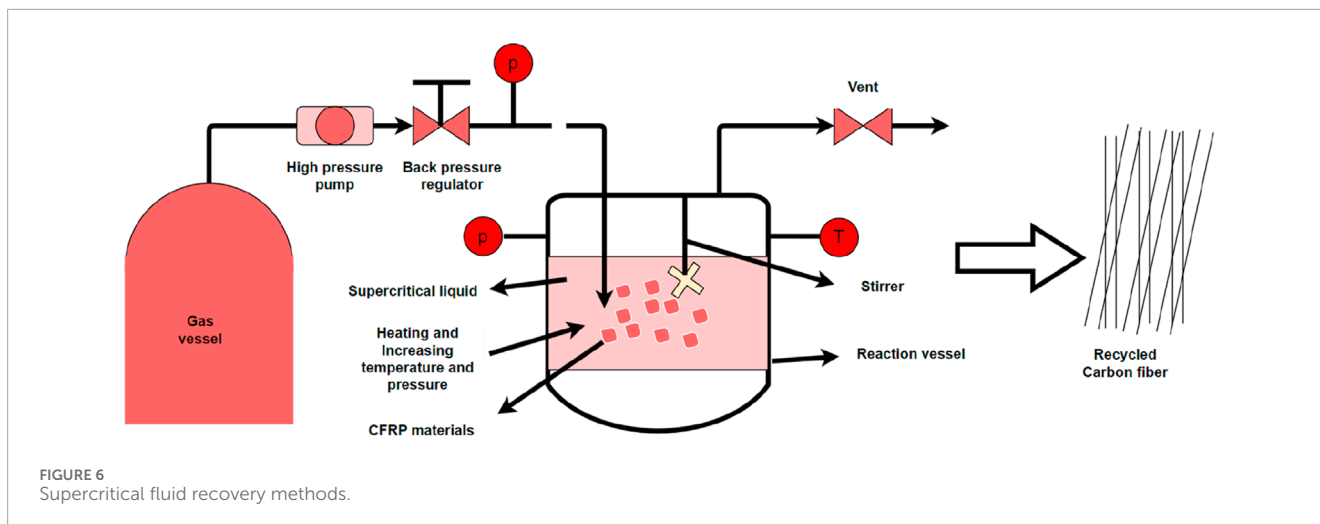


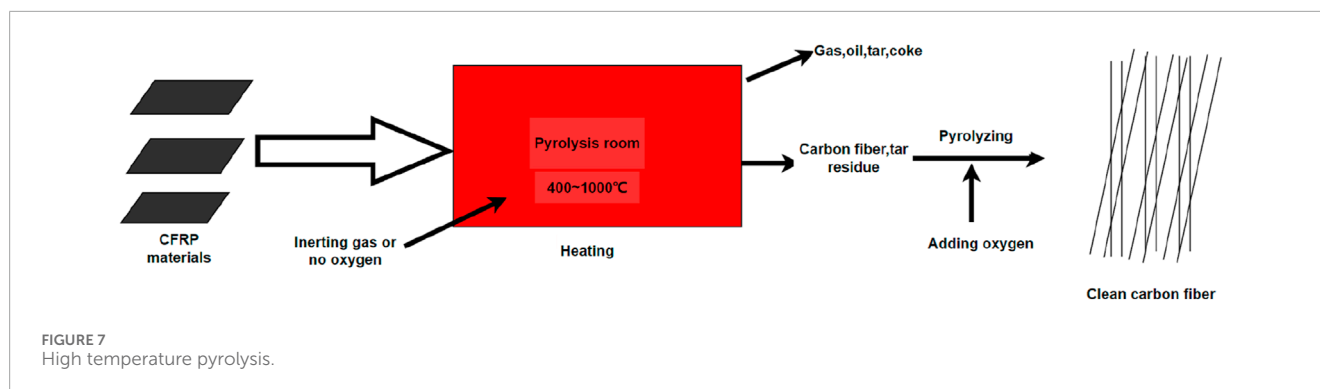
TABLE 3 Comparison of advantages and disadvantages of chemical recovery methods.

Recycling methods	Advantages	Disadvantages
Atmospheric Pressure Dissolution	Simple operation and little difficulty in recycling Good mechanical properties of the recovered carbon fiber	The reaction process is slow and the degradation mechanism is complicated; it is easy to cause environmental pollution; the reaction process is difficult to control
Electrolytic chemical method	The overall procedure is simple, with low recycling difficulty and high recycling efficiency; the recycled carbon fiber has good mechanical properties	Consumes a lot of energy Easily cause environmental pollution
Supercritical fluid recovery method	Recycling method is more efficient, low pollution and high resin removal rate Fiber mechanical properties are maintained better	The test conditions were very harsh; the overall safety margin was low

products (fibers, final fillers, and coke, etc.). Currently, thermal recovery method is the most commonly used method for CFRP waste recovery and has become commercially available (Verma et al., 2018). Thermal recovery methods mainly include high temperature pyrolysis, fluidized bed pyrolysis, and microwave pyrolysis.

High-temperature pyrolysis (Zhang et al., 2020), shown in Figure 7, refers to a thermal decomposition process carried out in an oxygen-free inert gas atmosphere, where the CFRP is heated up to between 400°C and 1,000°C to recover long carbon fibers with high modulus (Machado et al., 2018). The recovered carbon fibers have a tensile strength of w 50%–85% of virgin carbon fibers.

During pyrolysis, the decomposed resin matrix is separated from the CFRP to produce gas, oil, tar and coke, and the pyrolyzed carbon fibers also require oxidation reaction in oxygen to remove the coke residue on the surface of the carbon fibers and to obtain clean fibers for reuse (Overcash et al., 2018; Gastelu et al., 2018). Typically, polyester resins require lower temperatures for complete conversion, while epoxy resins require higher temperatures for decomposition. The selection criteria for the processing temperature is determined by the degree of conversion of the resin, which leads to a significant loss of mechanical properties of the carbon fiber. Giorgini et al. studied the pyrolysis of fiber-reinforced polyester composite waste in their 70 kg pilot plant and found that pyrolysis at 500°C–600 °C produced 20 wt% of oil, 40 wt% of gases, and 40 wt% of solid residue. Then further oxidation at 500°C for 50 min or at 600°C for 20 min was sufficient to obtain clean fibers (Giorgini et al., 2016). Kim used a new super-heated steam method (temperature 550°C) to recover high-quality carbon fibers with no carbon residue on the surface of the fibers, and the recovered fibers had better mechanical properties (Kim et al., 2017). Guo et al. (2019) pyrolyzed CFRERC to produce 2D rCF-reinforced carbon composites, observing minimal damage to the rCF and properties comparable to vCF, with flexural strengths of 90.3 MPa (rCF) and 95.5 MPa (vCF). Residual pyrolytic carbon did not significantly affect mechanical performance. In another study Guo et al. (2021), used pyrolysis on CFRERC to produce high-value-added rCF composites, finding minimal oxidation or damage to the rCF, which retained similar interfacial bonding (12.6 MPa vs. 13.0 MPa) and flexural strengths (106.4 MPa vs. 111.5 MPa) compared to vCF. Additionally Guo et al. (2022), examined the effects of oxidation conditions on rCF properties derived from CFRERC. Pyrolysis at 600°C for 30 min, followed by oxidation at 450°C for 15–20 min, yielded optimal results, with over 90.2% tensile strength retention, over 84.4% tensile modulus retention, over 70% conductivity, and over 95.2% recovery. Ren et al. (2023) achieved a high rCF tensile strength of 3,229.3 MPa (96.2% of vCF) with similar surface morphology. Wei et al. (Wei and Hadigheh, 2023) combined low-temperature pyrolysis with solvent pretreatment for CFRERC, reducing energy consumption and improving rCF mechanical properties, achieving 90.5% strength retention, 10.2% higher than



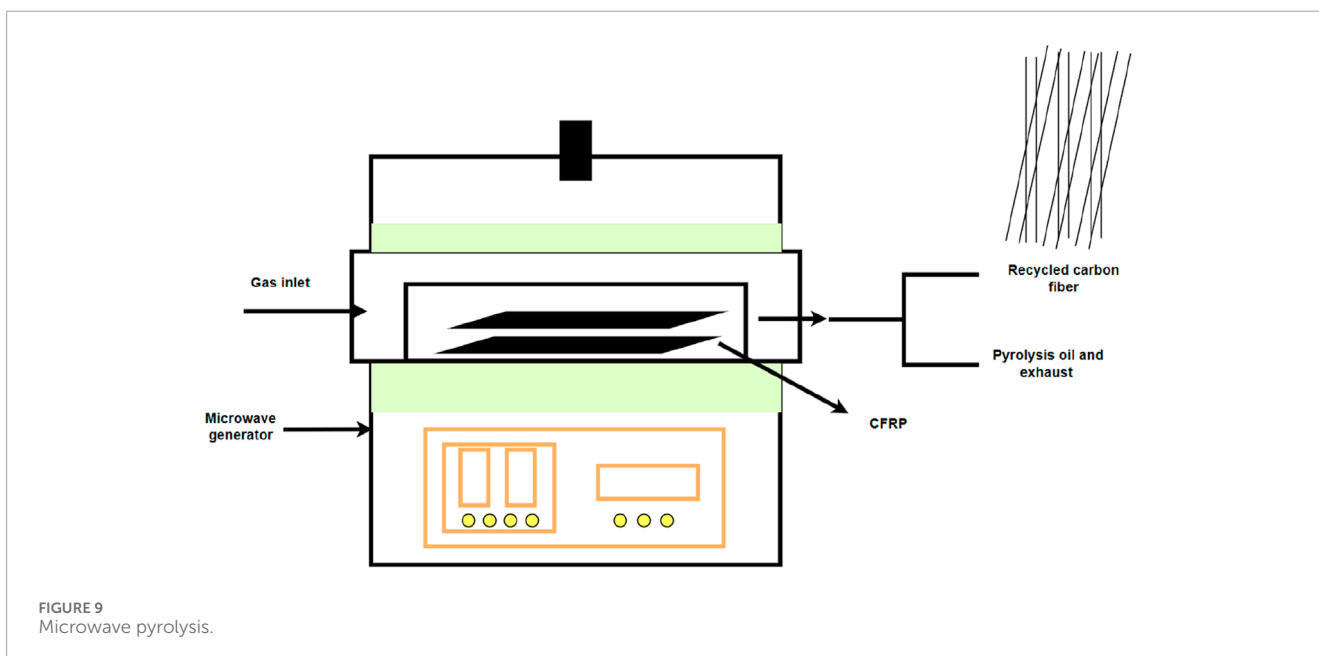
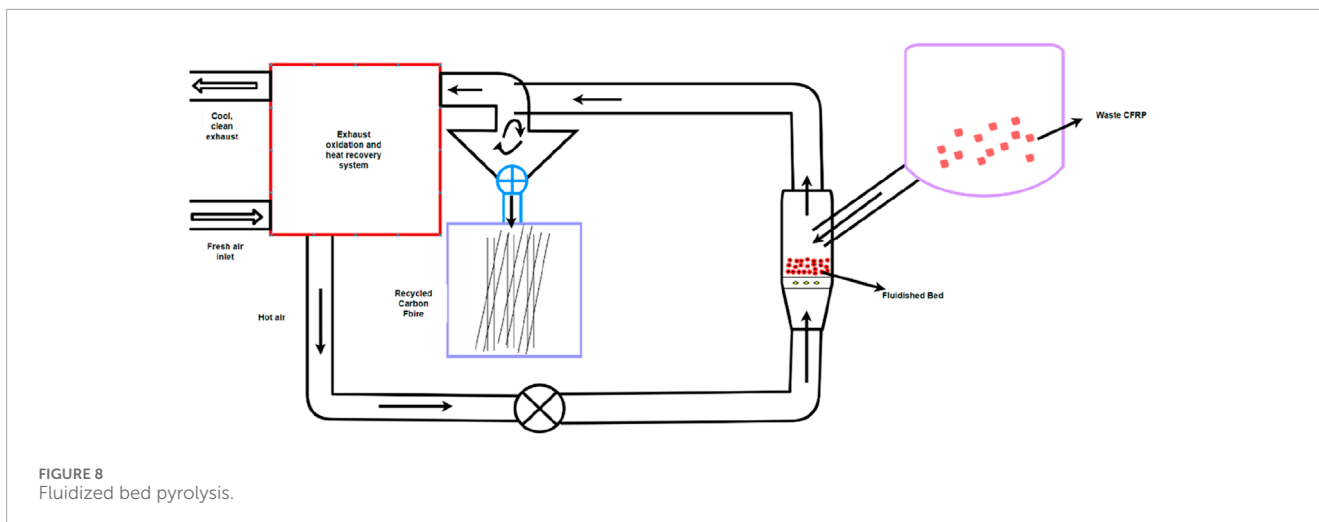
pyrolysis alone. Lopez-Urionabarrenechea et al. (2021) recovered rCF from CFRERC while generating hydrogen gas through a two-stage thermal process, retaining 80% tensile modulus, 50% interlaminar shear strength, and 70% flexural strength of vCF, while achieving over 66% hydrogen yield. Nahil and Williams (2011) recovered rCF from carbon fiber reinforced benzoxazine resin composites at 500°C, retaining 90% of vCF mechanical properties. In a similar approach Lopez-Urionabarrenechea et al. (2020) used pyrolysis to recover clean, mechanically sound rCF and valuable gas fractions from benzoxazine resin-based waste composites. Yousef et al. (2024) recovered carbon fiber from unsaturated polyester-based wind turbine blade waste, demonstrating that pyrolysis at 500°C effectively decomposed polyester into oil and gas products, with a 78 wt% solid residue containing carbon fiber. A life cycle assessment showed the environmental benefits of oxidative pyrolysis. Wei and Hadigheh (2024) analyzed the pyrolysis kinetics of CFRC to optimize parameters for clean, undamaged rCF. A nitrogen flow rate of 20 mL/min, a heating rate of 15 C/min, and a temperature of 425°C resulted in higher conversion rates. Isothermal oxidation at 550°C for 45 min produced clean rCF, retaining 87.6% tensile modulus and 80.3% tensile strength of vCF.

In summary, it can be seen that the mechanical properties of the recovered fibers are retained better and the size is controllable, the disadvantages are the generation of harmful gases and residues on the recovered surface. Currently, the high temperature pyrolysis process is considered to be the most feasible and sustainable CFRP recycling process to achieve process and resource efficiency.

Secondly, fluidized bed pyrolysis is a process in which waste CFRPs fragments, crushed to less than 25 mm, are placed in the charging hopper of a fluidized bed reactor (with air acting as the fluidizing gas), and under certain conditions of temperature and pressure (450°C–550°C, 10–25 kPa), the matrix resin undergoes oxidative decomposition, and the fibers are separated from the matrix and transported to the fiber tanks under the action of the air flow, as shown in Figure 8. The main work health hazards of this process are the presence of contaminant gases, organic solvents and high energy use, depending on the recovery conditions. The reaction temperature is chosen to ensure decomposition of the thermoset matrix without significant fiber degradation. Following rapid heating, fibers are released through abrasion and thermal decomposition degradation of the matrix, which is then separated and collected. The gases released from the matrix are combusted to oxidize the by-products. The recovered fibers are

5–10 mm in length and retain 10%–75% of the original carbon fiber tensile strength (Pickering et al., 2015). The process is suitable for EOL composite parts as rivets, bolts and other fittings can be collected in the bed and the carbon fibers of the composite can be recovered. Overall, the reuse of fluidized bed recycled fibers is competitive as they have a relatively low environmental impact (Meng et al., 2017). Fluidized bed recycling has been developed at the University of Nottingham since 2000 and is now operating on a pilot scale (Melendi-Espina et al., 2016). Wong et al. (2010) used a fluidized bed process to convert rCF into nonwoven veils for electromagnetic interference (EMI) shielding materials. The performance of the resulting composites was comparable to those made from virgin carbon fiber (vCF). Shielding effectiveness increased with veil density, while fiber length had little impact on shielding efficiency, provided the fiber distribution was uniform. Jiang et al. (2008) studied the surface properties of carbon fibers recovered through high-temperature fluidized bed processing. Their results showed that surface hydroxyl groups were oxidized during recovery due to exposure to the thermos-oxidative atmosphere, but the oxidative decomposition of the epoxy matrix did not adversely affect the interfacial bonding properties of the rCF. Yip et al. (2002) recovered rCF with average lengths up to 10 mm from fluidized bed degradation of CFRERC. The recovered fibers retained 75% of the tensile strength of vCF, maintained their Young's modulus, and exhibited similar surface conditions. Song et al. (2017) fabricated rCF-reinforced polyvinylidene fluoride (PVDF) composites using carbon fibers recovered via a fluidized bed process. They found that the rCF had higher chemical activity and surface roughness, resulting in stronger interfacial bonding with PVDF compared to vCF.

Microwave pyrolysis refers to the decomposition of resin matrix in composites by microwave radiation heating under inert gas atmosphere. Compared with the traditional pyrolysis method, microwave pyrolysis makes use of its unique heat and mass transfer law, heating uniformity and temperature controllability, so that the resin can be heated internally by the microwave energy absorbed by the CF, realising uniform heating and rapid decomposition of the resin, reducing the overall processing time and saving energy, and achieving the complete recycling of CF, as shown in Figure 9. This method was initially proposed by Lester (Binner et al., 2014) at the University of Nottingham, United Kingdom in 2004 for the recovery of glass fibers, and its main principle is to use microwaves to gasify the resin into gas or grease under the protection of



inert gas. Ren et al. (2022) proposed a new method for rapid recovery of carbon fibers by microwave pyrolysis and oxidation processes. The resin matrix was rapidly pyrolyzed by directly heating CFRP with microwave radiation, and then the residual carbon and organic matter on the surface of the carbon fibers were removed by oxidation to obtain recycled carbon fibers (RCFs).

The recovery rate of the recycled carbon fibers was measured and their mechanical properties were evaluated by tensile strength and tensile modulus tests. The results showed that after microwave pyrolysis at 500°C for 15 min and oxidation at 550°C for 30 min, the maximum tensile strength of the RCFs was 3,042.90 MPa (about 99.42% of that of the virgin carbon fibers), and the tensile modulus was 239.39 GPa, with a recovery rate of about 96.5%. The microstructure and chemical composition of RCF were characterized by scanning electron microscopy, X-ray diffraction, Raman spectroscopy, Fourier transform infrared

(FTIR) spectroscopy and X-ray photoelectron spectroscopy, and the compositions of pyrolysis by-products were detected by gas chromatography-mass spectrometry (GC-MS). These results indicate that the method is suitable for the effective recovery of high quality carbon fibers from CFRP. Hao et al. (2021), in their study of carbon fiber recovery from prepregs by microwave pyrolysis, successfully recovered carbon fibers from cured carbon fiber/epoxy resin (CF/EP) prepregs by microwave pyrolysis at 450, 550, and 650°C. The tensile properties, surface morphology and elements/functional groups on the surface of the recovered carbon fibers were also investigated, and the results showed that the distribution of elements and functional groups on the surface of the recovered carbon fibers was similar to that of the original carbon fibers. Li et al., (2021) used microwave pyrolysis to remove resin from discarded bicycle frames and recover carbon fiber, which was then incorporated into carbon fiber reinforced concrete

composites. Their study found that composites containing 1 wt% rCF exhibited optimal impact resistance compared to those made with vCF or without coupling agents. Lester et al. (2004) also employed microwave pyrolysis to recover rCF from CFRERC and compared its properties to vCF using single-fiber tensile tests. The rCF retained 72.4% of the tensile strength and 89.7% of the tensile modulus of vCF. Chen et al. (2022) successfully recovered carbon fiber from CFRC used in vehicles via microwave pyrolysis. Under controlled microwave conditions (300–1,000 W for 30 min), the resin was nearly completely pyrolyzed without damaging the carbon fiber structure. Thermoplastic composites made from this rCF showed superior thermal properties compared to those reinforced with vCF. However, the effects of microwave pyrolysis on fiber properties can vary. Jensen and Skelton (2018) compared two-step pyrolysis with microwave-assisted pyrolysis for carbon fiber recovery. Two-step pyrolysis resulted in less surface and structural damage, with rCF retaining over 95% of the tensile strength of vCF. In contrast, microwave pyrolysis caused significant surface defects, reducing tensile strength to 60% and altering the graphite structure. Despite this, microwave pyrolysis offered a 70% faster recovery rate and produced rCF with superior surface activity. Obunai et al. (2015) investigated the influence of atmospheric conditions (air, nitrogen, and argon) during microwave pyrolysis of CFRERC. Microwave heating under an argon atmosphere proved most effective, yielding rCF with strength comparable to that of virgin carbon fiber. Deng et al. (2019) compared microwave and conventional pyrolysis for CFRERC, finding that microwave pyrolysis reduced reaction time by 56.7%, increased the recovery rate by 15%, and produced cleaner carbon fibers with greater efficiency and lower energy consumption.

In summary, the thermal recycling process is simple and is the only waste CFRP recycling technology that has achieved pre-commercial application. However, the resin matrix is mainly recycled in the form of gas or liquid, with low overall recycling efficiency and high energy consumption, and most of the fibers in the whole pyrolysis process are severely damaged and the fiber size is too low, which is not conducive to the recycling of high-value CFs. Therefore, under the current economic cycle system, it is necessary to further develop new thermal recovery processes to achieve maximum recovery of carbon fibers and resins from waste CFRP. Finally we have also summarized the advantages and disadvantages of the pyrolysis method in detail in Table 4.

3 Summary and outlook

At present, CFRP recycling technology is still in the process of continuous research and improvement and has also achieved carbon fiber recycling of various CFRP wastes, which is of great significance to the sustainable development of carbon fiber. First of all, due to the incineration method in the recycling process will produce a lot of harmful substances, at the same time, the recycled carbon fiber is only 35%–40% of the raw material, the strength of the fiber after physical recycling method will be seriously reduced and the value of reuse is also low, it can only be used as some fillers. It can be seen that the disadvantages of the two recycling methods are so great that it is unlikely that there will be room for progress in the future development of these two methods.

TABLE 4 Comparison of advantages and disadvantages of heat recovery methods.

Recycling methods	Advantages	Disadvantages
pyrolysis	Better retention of mechanical properties of fibers Controlled size	Hazardous fumes; recycled fibers Residues on the surface of the fiber
Fluidized bed pyrolysis	Treatment of highly contaminated materials; no resin or carbon residue on the fiber surface; resin matrix can be recycled as an energy source	Fiber size too low; mechanical properties Severe loss
Microwave pyrolysis	Direct heating of the inner part of the material Heating of the material directly from the inside, saving energy	Higher loss of mechanical properties of fibers

Secondly, the chemical recovery method of atmospheric pressure dissolution method, electrolytic chemical method, supercritical fluid recovery method has its own advantages and disadvantages, but after a comprehensive analysis and analysis of the current situation, due to the atmospheric pressure dissolution method and electrolytic chemical method to recover the CFRP are harmful to the environment after the production of substances and electrolytic chemical method of energy consumption is huge and contrary to the advocate of energy in the future development of the two methods of further use of the chances are not great. However, with the development of production technology and production process, supercritical fluid recovery method is expected to be commercialized in the future due to the excellent performance of the recovered carbon fiber, high recovery efficiency and low pollution. Third, thermal recovery methods mainly include high temperature pyrolysis, fluidized bed pyrolysis and microwave pyrolysis. Each of these methods has its own characteristics. The mechanical properties of fibers recovered by high-temperature pyrolysis are better preserved and the size is controllable, but the disadvantages are the generation of harmful gases and residues on the recovered surface. Currently, high temperature pyrolysis is considered to be the most viable and sustainable CFRP recycling process for achieving process and resource efficiency. The main drawbacks of fluidized bed pyrolysis are the generation of pollutant gases, the presence of organic solvents and the high energy use, although commercialization of fluidized bed pyrolysis is very much in the future according to current industrial methods. Microwave pyrolysis is an emerging technology developed in the last few years, with the advantage of excellent performance in recovering carbon fibers and relatively simple experimental equipment, but with the disadvantage of generating toxic gases that pollute the environment. Despite the shortcomings of the current recycling methods, with the development of science and technology and social progress, recycling technology is bound to

flourish. Combined with the development of the current recycling technology, the future development of research work mainly includes the following points: 1) **Green and Efficient Chemical Recovery Processes:** Future chemical recovery research needs to break through the limitations of traditional processes and develop more environmentally friendly and efficient solvent systems and recyclable catalysts. For example, explore the use of low-toxicity, bio-based or deep eutectic solvents (DES) for the selective dissolution of different resin matrices (e.g., epoxy resins, benzoxazine resins), so as to efficiently separate resins from carbon fibers. Meanwhile, the effects of process parameters, such as temperature, pressure, and solvent concentration, on the recovery efficiency and fiber properties should be systematically investigated, and databases and prediction models of optimal process parameters should be constructed in order to achieve a stable and high-quality recovery process. This kind of research will bring both environmental protection and economic benefits to the chemical recovery method, which is in line with the needs of sustainable development; 2) **Optimization and Multiple Integration of Heat Recovery Processes:** Heat recovery technologies such as microwave pyrolysis and fluid bed pyrolysis have high recovery potentials, but energy consumption and fiber damage are still pressing issues. Future research can reduce energy consumption and improve the quality of recovered fibers by optimizing the reactor design and improving the heat transfer performance of fluidized bed media. In addition, attempts can be made to combine the pyrolysis process with chemical recycling and physical recycling methods to form an efficient multistage composite process. For example, low-temperature chemical pretreatment is carried out first to remove part of the resin, and then complete recovery of carbon fibers is achieved through pyrolysis, which enhances the resource utilization rate and reduces the energy consumption of the process. At the same time, pathways for high value-added utilization of pyrolysis products (e.g., oil, gas, coke, etc.) should be explored, e.g., converting pyrolysis oil into chemical products or biofuels through refining, in order to improve the overall economic benefits of recycling; 3) **Exploration of new cutting-edge recycling technologies:** In addition to traditional chemical and thermal recycling methods, supercritical fluid recycling technology, bio-enzyme-based biorecycling technology and low-energy electrochemical recycling technology can be further studied in the future. Supercritical fluid technology, with its high diffusivity and low viscosity, can achieve efficient resin decomposition at low temperatures while maintaining fiber properties to the maximum extent possible; bio-enzyme catalytic technology, with its advantages of low pollution and low energy consumption, can be explored for application in CFRP recycling in the future, especially for low-pollution and sustainable recycling systems; electrochemical recycling can reduce energy consumption while further enhancing fiber purity through selective electrolysis of degraded resins; and electrochemical recycling can reduce energy consumption while further enhancing fiber purity. Electrochemical recycling can further improve fiber purity while reducing energy consumption through selective electrolytic degradation of resin. The successful application of these emerging technologies will open up even more possibilities for CFRP recycling; 4) **Quality Evaluation and Application Expansion of Recycled Fibers:** In order to make recycled carbon fibers widely used in high value-added fields, it is necessary to establish a systematic performance evaluation standard,

including mechanical properties, surface chemical composition and microstructure of fibers. At the same time, surface modification technologies (such as nano-coating, plasma surface treatment, etc.) should be studied to improve the interfacial bonding properties between recycled fibers and the matrix, so as to expand their applications in the fields of composites, cementitious materials, adsorbent materials, etc., and they can even be used in aerospace, automotive and other industries with high requirements for performance, so as to realize value-added utilization of recycled fibers; 5) **Intelligent and data-driven process optimization:** The future CFRP recycling process can be intelligently and data-driven optimized through artificial intelligence (AI) and big data technologies. By building data models, changes in process parameters can be monitored in real time and intelligently adjusted to optimize the recycling effect and improve process stability. Additionally, variables such as temperature, pressure, solvent concentration, etc. can be monitored in real time in combination with sensors and IoT technology to achieve precise process control, thereby significantly improving production efficiency and ensuring the quality of recovered fibers; 6) **Comprehensive Life Cycle Assessment (LCA) and Economic Analysis:** In order to ensure the sustainability of the CFRP recycling process, it is necessary to systematically carry out the Life Cycle Assessment (LCA) and economic analysis of different recycling technologies to quantify the carbon footprint, energy consumption and cost of each method, and to identify recycling pathways with the highest environmental and economic benefits. Based on the LCA results, data support can be provided for policy formulation to encourage the development of green recycling processes and to promote the process of carbon neutrality in the recycling process. Such evaluation will provide scientific basis for enterprises and policymakers to promote the development of the industry in the direction of green and low carbon; 7) **Industry chain synergy and policy support:** Achieving the industrialization of CFRP recycling technology requires the collaboration of all parties, especially the synergy between academia, industry and government. The government can promote the application of green recycling technology by formulating standardized recycling specifications, providing tax incentives, etc., and encouraging enterprises to give priority to environmentally friendly recycling technology in production; while the industry should strengthen cooperation with academic institutions, promote the transformation of technological achievements, and form an efficient recycling industry chain. Meanwhile, the establishment of a sound industry standard and policy framework will help the scale promotion of CFRP recycling technology. These directions cover the optimization and innovation of chemical recycling, thermal recycling and new recycling technologies, as well as key aspects such as performance characterization, application expansion and life cycle evaluation of recycled carbon fibers. In the future, the in-depth development of these research directions will not only promote the comprehensive improvement of CFRP recycling technology in terms of efficiency, economy and environmental protection, but also help to realize the sustainable recycling of carbon fiber materials, promote the green transformation and sustainable development of the composite materials industry, and ultimately provide strong support for achieving the strategic goal of carbon neutrality.

Author contributions

MY: Conceptualization, Data curation, Formal Analysis, Investigation, Methodology, Resources, Software, Supervision, Validation, Visualization, Writing–original draft, Writing–review and editing. ZL: Writing–review and editing, Data curation, Methodology, Supervision, Conceptualization, Investigation. ZT: Writing–review and editing, Methodology, Supervision, Conceptualization, Formal analysis, Investigation.

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