

Experimental Investigation on Isothermal Sections at 1273 and 1473K in the Co–Ti–W System

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The microstructures of 25 annealed alloys and XRD patterns of partial critical alloys in the Co-Ti-W system were investigated by using scanning electron microscopy (SEM) with energy dispersive spectrometer (EDS) and X-ray diffraction (XRD) methods. The isothermal sections at 1273 and 1473 K of the Co-Ti-W system were established. Five three-phase regions and five two-phase regions at 1273 K and eight three-phase regions and two twophase regions at 1473 K were experimentally determined. The maximum solubilities of W in Co₃Ti, α Co₂Ti, β Co₂Ti and CoTi were determined to be ~7.9 at%, ~1.5 at.%, ~5.9 at.% and ~1.8 at.% at 1273 K, respectively. The maximum solubilities of Ti in Co₇W₆ and Co₃W were determined to be ~11.9 at.% and ~15.2 at.% at 1273 K, respectively. The compound Co₃Ti with L1₂ crystal structure was found to be stable at 1473 K in the Co–Ti–W system owing to the addition of W element, which confirmed that W can improve the stability of γ' with L1₂ crystal structure in the Co-based superalloys. The composition range of W in Co₃Ti was measured to be ~7.4–10.4 at.% at 1473 K. In addition, the maximum solubilities of Ti in Co_7W_6 and W in βCo_2Ti and CoTi were ~15.4 at.%, ~7.6 at.% and ~3.1 at.% at 1473 K, respectively. No ternary compounds were found in the Co-Ti-W system at 1273 and 1473 K.

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INTRODUCTION

Superalloys are widely adopted in many fields that require materials to serve at elevated temperature such as aircraft engines, gas turbines, chemical plants (Betteridge and Shaw, 1987; Pollock and Tin, 2006; Reed, 2008; Cheng et al., 2012; Qu et al., 2018). Co-based superalloys show better resistance of hot corrosion and oxidation than Ni-based superalloys. They are considered as one of promising candidates for applying at high temperature (Sims et al., 1987). However, compared with Ni-based superalloys, traditional Co-based superalloys are less used due to the inferior high-temperature strength. In Ni-based superalloys, the γ' -Ni₃Al strengthening phase with L1₂ structure precipitates from the disordered γ matrix phase with fcc crystal structure. And the coherent interfaces are formed between γ' -Ni₃Al and γ . This is one of main reasons for the outstanding high-temperature properties of Ni-based superalloys. Traditional Co-based superalloys are strengthened by carbide precipitations and solid solutions of refractory elements. The inefficient strengthening methods make traditional Co-based superalloys only be applied in mechanically low-loaded corrosive environments (Sims et al., 1987).

As early as 1971, Lee (Lee, 1971) discovered precipitation-hardened phenomenon in a 71 wt% Co-4 wt% Al-25 wt% W alloy, in which the ordered γ' phase (L1₂) precipitated in γ matrix phase

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Phase	Strukturbericht Designation	Pearson Symbol	Space Group	Prototype	Refs	
liquid	_	_	_	_	_	
hcp (Co)	A3	hP2	P6 ₃ /mmc	Mg	(Murray, 1982; Sato et al., 2005)	
fcc (Co)	A1	cF4	Fm-3m	Cu	(Murray, 1982; Sato et al., 2005)	
Co ₃ Ti	L12	cP4	Pm-3m	Cu ₃ Au	Murray, (1982)	
βCo ₂ Ti	C15	cF24	Fd-3m	MgCu ₂	Murray, (1982)	
αCo ₂ Ti	C36	hP24	P6 ₃ /mmc	MgNi ₂	Murray, (1982)	
CoTi	B2	cP2	Pm-3m	CsCl	Murray, (1982)	
CoTi ₂	E93	cF96	Fd-3m	NiTi ₂	Murray, (1982)	
bcc (Ti)	A2	cl2	lm-3m	W	Murray, (1982)	
hcp (Ti)	A3	hP2	P6 ₃ /mmc	Mg	Murray, (1982)	
Co ₃ W	D0 ₁₉	hP8	P6 ₃ /mmc	Ni₃Sn	Magneli and Westgren, (1938)	
Co ₇ W ₆	D85	hR13	R-3m	Fe ₇ W ₆	Sato et al. (2005)	
bcc(W)	A2	cl2	lm-3m	W	Sato et al. (2005)	

The italic values refer to Binary Alloy Phase Diagrams Voluem.

(fcc_A1). But almost nobody paid attention to his research results at that time. Until recently, Sato et al. (Sato et al., 2006) observed a $\gamma+\gamma'$ two-phase coherent microstructure in an alloy with the composition of 83.5 at.% Co-9 at% Al-7.5 at.% W, which was similar to that in Ni-based superalloys. The γ' -Co₃(Al, W) phase precipitated from γ matrix and formed the coherent interfaces with γ phase, which made Co-based superalloys possess the increased high-temperature strength. Due to higher melting point of Co than Ni, Co-based superalloys might possess

better temperature capabilities. Subsequent studies (Kobayashi et al., 2009; Tsukamoto et al., 2010; Xue et al., 2011; Lass et al., 2014) showed the stability of the γ' phase was still uncertain. According to the experimental results of Lass et al. (Lass et al., 2014), the γ' phase existed for thousands of hours at 1173 K in the Co–Al–W alloys but its volume fraction decreased continuously and slowly with annealing time.

The high temperature performances of superalloys can be improved by adding alloying elements (Yokokawa et al., 2003;

TABLE 2 The constituent phases and their compositions of the Co-Ti-W	alloys
annealed at 1273 K.	

TABLE 3 The constituent phases and their compositions of the Co-Ti-W	alloys
annealed at 1473 K.	

No.	Alloy Composition (at%)			Phases	Phase Composition (at%)		
	Co	Ti	w		Co	Ti	w
a1	81.4	3.9	14.7	fcc (Co)	89.5	2.8	7.7
	_	—	_	Co ₃ W	75.8	5.3	18.9
a2	80.9	8.7	10.4	fcc (Co)	88.9	5.9	5.2
	_	—	_	Co ₃ W	75.6	10.5	13.9
a3	79.9	13.7	6.4	fcc (Co)	88.2	7.6	4.2
	_	_	_	Co ₃ W	75.3	15.2	9.5
a4	73.1	2.7	24.2	Co ₃ W	75.4	3.5	21.1
	_	_	_	Co ₇ W ₆	56.2	0.6	43.2
a5	70.5	6.8	22.7	Co ₃ W	74.1	6.9	19.0
	_	_	_	Co ₇ W ₆	56.6	2.1	41.3
a6	72.3	12.5	15.2	Co ₃ W	73.9	14.2	11.9
	_	_	_	Co ₇ W ₆	58.1	5.3	36.6
a7	70.3	19.2	10.5	Co ₃ Ti	74.1	18.0	7.9
	_	_	_	βCo ₂ Ti	66.5	27.6	5.9
	_	_	_	Co ₇ W ₆	55.4	8.8	35.8
a8	36.9	13.2	49.9	bcc(W)	2.2	1.8	96.0
	_	_	_	Co ₇ W ₆	52.3	11.9	35.8
	_	_	_	βCo ₂ Ti	65.9	28.4	5.7
a9	41.5	30.7	27.8	bcc(W)	3.1	3.9	93.0
	_	_	_	βCo ₂ Ti	65.5	32.3	2.2
	_	_	_	CoTi	53.9	44.3	1.8
a10	22.2	45.4	32.4	bcc(W)	1.1	21.0	77.9
	_	_	_	CoTi	48.3	51.7	0.0
	_	_	_	CoTi ₂	30.3	68.7	1.0
a11	27.9	59.5	12.6	bcc(W)	0.3	23.7	76.0
	_	_	_	CoTi	46.4	53.0	0.6
	_	_	_	CoTia	30.8	68.0	1.2
a12	12.9	68.7	18.4	CoTia	31.3	68.4	0.3
	_	_	_	bcc (Ti)	6.1	86.8	7.1
	_	_	_	bcc(W)	0.7	34.2	65.1
a13	81.3	16.6	2.1	fcc (Co)	86.9	11.2	1.9
	_	_	_	CoaTi	78.2	18.6	3.2
a14	34.4	41.0	24.6	bcc(W)	26	13.7	83.7
	_	_	_	CoTi	49.5	49.5	1.0
a15	70.3	27.3	2.4	CoaTi	74.4	21.2	4.4
	_	_	_	aCo ₂ Ti	68.4	30.1	1.5

Shinagawa et al., 2009; Bauer et al., 2010). Ti (Xue et al., 2013; Christofidou et al., 2016; Bocchini et al., 2017; Llewelyn et al., 2017), W (Gupta et al., 2008a; Gupta et al., 2008b; Liu et al., 2010; Balam and Paul, 2011), Cr (Gupta et al., 2008b; Liu et al., 2010), Nb (Balam and Paul, 2011; Xue et al., 2013), Mo (Liu et al., 2010; Balam and Paul, 2011), and Ta (Balam and Paul, 2011; Xue et al., 2013) are important alloying elements in superalloys. The addition of Ti can improve the high-temperature stability of γ' phase by increasing of its solvus temperature (Xue et al., 2013; Christofidou et al., 2016; Bocchini et al., 2017; Llewelyn et al., 2017) and promote the strength at high temperature of the Cobased superalloys. Furthermore, only Ti can form stable A₃B type compound (Co₃Ti) with ordered L1₂ crystal structure with Co among the above elements. And a two-phase region $fcc + L1_2$ is formed in the Co-Ti phase diagram. W is one of the most efficient alloying elements for improving high-temperature strength by solid-solution strengthening and increasing the stability of γ' phase (Gupta et al., 2008a; Gupta et al., 2008b; Liu et al., 2010;

No.	Alloy Composition (at%)			Phases	Phase Composition (at%)		
	Co	Ti	w		Co	Ti	w
b1	78.8	15.7	5.5	fcc (Co)	83.4	11.0	5.6
	_	_	_	liquid#1	79.4	19.0	1.6
	—	_	—	Co ₃ Ti	76.5	14.6	8.9
b2	76.7	18.8	4.5	liquid#1	78.4	19.2	2.4
	—	_	—	βCo ₂ Ti	69.8	23.9	6.3
	_	_	_	Co ₃ Ti	75.3	17.3	7.4
b3	70.6	17.3	12.1	Co ₇ W ₆	57.4	7.6	35.0
	_	_	_	Co ₃ Ti	73.5	16.2	10.3
	_	_	_	βCo ₂ Ti	69.1	25.0	5.9
b4	74.0	9.0	17.0	Co ₇ W ₆	56.2	5.4	38.4
	_	_	_	fcc (Co)	81.5	7.7	10.8
	_	_	_	Co ₃ Ti	77.3	12.3	10.4
b5	32.3	10.7	57.0	bcc(W)	0.4	1.5	98.1
	_	_	_	Co ₇ W ₆	53.2	15.4	31.4
	_	_	_	βCo ₂ Ti	65.5	26.9	7.6
b6	60.2	33.2	6.6	bcc(W)	4.3	8.1	87.6
	_	_	_	βCo ₂ Ti	65.6	29.3	5.1
	_	_	_	CoTi	56.0	40.9	3.1
b7	28.0	46.9	25.1	bcc(W)	1.0	15.5	83.5
	_	_	_	CoTi	46.7	52.2	1.1
	_	_	_	liquid#2	29.4	69.6	1.0
b8	10.6	76.4	13.0	bcc(W)	1.3	45.0	53.7
	_	_	_	liquid#2	21.0	78.7	0.3
	_	_	_	bcc (Ti)	6.2	80.8	13.0
b9	73.2	3.9	22.9	fcc (Co)	85.0	2.5	12.5
	_	_	_	Co ₇ W ₆	55.9	2.3	41.8
b10	15.0	66.4	18.6	liquid#2	24.5	74.9	0.6
	_	—	—	bcc(W)	1.3	33.0	65.7

Balam and Paul, 2011). Therefore, the Co-Ti-W system is worth being studied in detail.

The ternary phase diagram is an important basis for the design and application of materials (Li et al., 2009; Zhang et al., 2011; Xu et al., 2016). So far, the phase diagram of the Co–Ti–W ternary system has hardly been studied. In the current work, the phase relationships at 1273 and 1473 K of the Co–Ti–W system were experimentally determined. The purpose is to construct the Co–Ti–W isothermal sections and explore composition range and stability of γ' phase at high temperature.

LITERATURE REVIEW

Murray (Murray, 1982) and Cacciamani et al. (Cacciamani et al., 2000) reviewed the experimental data of the Co–Ti system that were published before 1998 in detail. Davydov et al. (Davydov et al., 2001) determined the congruent melting point of the CoTi phase using differential thermal analysis (DTA) and a visual observation of melting (VOM) and optimized the phase diagram of the Co–Ti system. Recently, Wu et al. (Wu et al., 2020) experimentally confirmed that the invariant reaction among liquid, Co₃Ti and α Co₂Ti was a eutectic reaction at 1427 K. The phase diagram of the Co–Ti system in Ref (Wu et al.,



2020). was adopted in this work and shown in **Figure 1A**. In the Co–Ti phase diagram, five solution phases liquid, fcc (Co), hcp (Co), bcc (Ti) and hcp (Ti), and five intermetallic compounds Co₃Ti, α Co₂Ti, β Co₂Ti, CoTi and CoTi₂ are stable.

In the Co-W phase diagram calculated by Kaufman et al. (Kaufman and Nesor, 1978), fcc (Co), hcp (Co), Co₃W, Co₇W₆ and bcc(W) phases were stable. Subsequently, Gabriel et al. (Gabriel et al., 1985) experimentally determined solid-liquid phase equilibria on the Co-rich side of the Co-W system. The temperatures of the liquidus and invariant reactions were determined using DTA with a heating rate of 5 K/min. Sato et al. (Sato et al., 2005) investigated phase equilibria among solid phases using equilibrated alloys and the diffusion couple technique. And the magnetic and martensitic transition temperatures were systematically studied by differential scanning calorimetry (DSC), vibrating sample magnetometer (VSM) and dilatometric measurement. Dmitrieva et al. (Dmitrieva et al., 2005) determined the homogeneity ranges of Co₃W and Co₇W₆ at 1473 K using equilibrated alloys. Ravi et al. (Ravi and Paul, 2011) precisely measured the solubility of W in fcc (Co) by the diffusion couple technique. Recently, Wang et al. (Wang et al., 2019) evaluated the Co-W phase diagram based to the above experimental data. The Co-W phase diagram in Ref (Wang et al., 2019). was adopted in the current work and shown in Figure 1B. The Co-W phase diagram contains two intermediate phases, Co₇W₆ and Co₃W, which are formed by peritectic reaction liquid

+ $bcc(W) \rightarrow Co_7W_6$ and peritectoid reaction fcc (Co)+ $Co_7W_6 \rightarrow Co_3W$, respectively.

The Ti-W phase diagram is relatively simple, and liquid phase and two solution phases bcc (Ti, W) and hcp (Ti) are included. Murry (Murray, 1981) reviewed previous experimental information and evaluated the Ti-W phase diagram. Jin et al. (Jin and Qiu, 1993) optimized the Ti-W system on the basis of the modified Gibbs energy descriptions of pure Ti and W from the Scientific Group Thermodata Europe (SGTE) database. And all relevant experimental data were taken into account simultaneously. Jonsson (Jonsson, 1996) re-optimized the Ti-W system using simpler models and all the experimental data were well reproduced. The Ti-W phase diagram optimized by Jonsson (Jonsson, 1996) was adopted in present work and shown in Figure 1C. In the Ti-W phase diagram, there is a miscibility gap of bcc (Ti, W) below 1523 K, which makes the high temperature bcc (Ti, W) phase decompose into the Ti-rich and W-rich solid solutions.

Although the Co-Ti-W system is one of the most important Co-based superalloy subsystems, only very limited literature on this ternary system (König et al., 2014; Naujoks et al., 2017) was found. König et al. (König et al., 2014) fabricated a Co-Ti-W thin film materials library by magnetron sputtering. The compositions and resistances of various points on the thin film materials library were measured. Since the resistances of single-phase regions are



lower than that of mixed phase regions, an unknown ternary phase region was revealed at 1223 K in the Co–Ti–W system according to sudden changes of resistance values at the phase region boundaries. The composition of the new ternary phase was about 60 at% Co–15 at% Ti–25 at% W. However, the new phase was not directly observed in the microstructure of the annealed bulk alloy at 1373 K. Naujoks et al. (Naujoks et al., 2017) proved that the solubility of Ti in Co_7W_6 phase was 18 at% at 1223 K and determined a three-phase region bcc(W)+ Co_7W_6 + αCo_2Ti in the 1273 K isothermal section of the Co–Ti–W system using thin-film and bulk materials. Besides, no experimental phase diagram data of the Co–Ti–W system is available.

The crystallographic data of all phases in the three binary systems are listed in **Table 1**(Magneli and Westgren, 1938; Murray, 1982; Sato et al., 2005).

MATERIALS AND METHODS

Pure Cobalt (99.99 wt%), Titanium (99.99 wt%) and Tungsten (99.99 wt%) (China New Material Technology Company, Ltd.) were used as raw materials in the present work. The weight of each alloy sample is about 3–4 g. These samples were melted by using the arc-furnace (MTI MSM20-7) in an argon atmosphere with a non-consumable tungsten electrode. Several titanium rods firstly were

melted in order to absorb residual oxygen before melting samples. To improve the homogeneity, every sample was turned around and remelted at least 4 times. The samples which weight loss did not exceed 1% during arc melting were used in the following experiments. Then, each sample was cut into several pieces for different purposes.

In order to construct the isothermal sections, the Co–Ti–W alloys were annealed to obtain phase equilibria at 1273 and 1473 K. At 1273 K, the samples were annealed for 1000 h. At 1473 K, the time of the heat treatment varied from 5 h to 3 days for the samples with liquid phase. Others were annealed for about 720 h. In the process of annealing, the samples were sealed in evacuated quartz capsules (6×10^{-2} Pa), and the chamber furnace (MTI KSL-1400) with a temperature accuracy of ±1 K was used. After heat treatment, the annealed samples were quenched into water to keep the phase constituents at 1273 and 1473 K.

Before microstructural characterization, all samples were ground and polished. The phases and element distributions of the all samples were studied using SEM (Carl Zeiss LEO 1450) in combination with EDS (Thermo Scientific UltraDry EDS). In order to increase the precision of the composition measurements, 3–5 points or areas were analyzed for each phase.

XRD analyses were carried out using X-ray diffractometer (Rigaku Ultima IV) with Cu-K α radiation at 40 kV and 40 mA to identify the crystal structures of the constituent phases. Diffraction patterns were collected over the 2 θ range from 20°



to 90° with a scan step 0.02° . The software Jade 6.0 was used to assist phase identifications in this process.

RESULTS AND DISCUSSIONS

Isothermal Section at 1273 K

For the investigation of the isothermal section at 1273 K of the Co-Ti-W system, 15 alloys were prepared. The constituent phases

and their compositions of the above alloys are summarized in **Table 2**. The analysis and discussion of each alloy is as follows.

The BSE micrographs of alloys a1–a3 and XRD pattern of alloy a3 are shown in **Figure 2**. The similar microstructures were observed in **Figure 2A–C**. Combined the phase compositions of alloys a1–a3 obtained by EDS with the phase constituent of alloy a3 obtained by XRD, it could be determined that three alloys were located in two-phase region fcc (Co)+Co₃W, and gray and dark phases were Co₃W and fcc (Co), respectively.



The two-phase microstructures $Co_3W + Co_7W_6$ were observed in alloys a4–a6. Figure 3 shows the BSE images of alloys a4–a6 and XRD pattern of alloy a6. The white and gray phases were determined to be Co_7W_6 and Co_3W combining the phase compositions with XRD pattern, respectively.

The BSE micrographs and XRD patterns of alloys a7-a9 annealed at 1273 K are shown in Figure 4. As shown in Figure 4A, three phases were observed in the BSE micrograph of alloy a7. According to the results of EDS measurements, white phase was Co₇W₆ (55.4 at.% Co-8.8 at.% Ti-35.8 at.% W), gray phase was Co3Ti (74.1 at.% Co-18.0 at.% Ti-7.9 at.% W) and dark gray phase was Co₂Ti (66.5 at.% Co-27.6 at.% Ti-5.9 at.% W). In the Co-Ti phase diagram, the Co₂Ti phase has two crystal structures which are C36 for α Co₂Ti and C15 for β Co₂Ti. In present work, the Co₂Ti phase in alloy a7 was regarded as βCo₂Ti on the basis of XRD pattern in Figure 4B. The same conclusion could be obtained in alloys a8 and a9. Due to the extremely slow atomic mobility of W element, it was difficult to reach an equilibrium state in short annealing time for the alloy a8 with high W content. The microstructure in alloy a8 seemed to be out of complete equilibrium. However, the microstructure was very different from that of as-cast sample. The similar microstructure was also found in Figure 2J in Ref (Liu et al., 2012). when the isothermal section at 1473 K of the Co-Nb-W ternary system was experimentally determined. Therefore, on the basis of the results of EDS and XRD of alloy a8, the alloy a8 was

considered to be located in a three-phase region in the present work, in which white, gray and dark phases were determined to be bcc(W) (2.2 at.% Co-1.8 at.% Ti-96.0 at.% W), Co₇W₆ (52.3 at.% Co-11.9 at.% Ti-35.8 at.% W) and βCo₂Ti (65.9 at.% Co-28.4 at.% Ti-5.7 at.% W), respectively. In the BSE micrograph of alloy a9, the compositions of white, gray and dark phases were 3.1 at.% Co-3.9 at.% Ti-93.0 at.% W, 65.5 at.% Co-32.3 at.% Ti-2.2 at.% W and 53.9 at.% Co-44.3 at.% Ti-1.8 at.% W, respectively. Combined with the XRD pattern shown in Figure 4F, white phase was bcc(W), gray phase was βCo₂Ti and dark phase was CoTi. Based on the above analysis, alloys a7-a9 were located in three-phase regions Co₇W₆+Co₃Ti+βCo₂Ti, bcc(W)+Co₇W₆+βCo₂Ti and $bcc(W)+CoTi+\beta Co_2Ti$, respectively.

The alloys a10 and a11 were both located in three-phase region $CoTi + CoTi_2+bcc(W)$. **Figure 5A** shows the microstructure of alloy a11 annealed at 1273 K. According to the composition of each phase obtained by EDS, white phase was bcc(W) (0.3 at.% Co-23.7 at.% Ti-76.0 at.% W), light-gray phase was CoTi (46.4 at.% Co-53.0 at.% Ti-0.6 at.% W), and dark-gray phase was $CoTi_2$ (30.8 at.% Co-68.0 at.% Ti-1.2 at.% W). The XRD pattern of the alloy is shown in **Figure 5B**, where the characteristic peaks of the bcc(W), CoTi and $CoTi_2$ phases were well distinguished and marked by different symbols. The boundary of three-phase region $bcc(W)+CoTi + CoTi_2$ was determined by the average values of phase compositions in alloys a10 and a11. The BSE micrograph and XRD pattern of alloy a12 are presented in

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FIGURE 8 | BSE micrographs and XRD patterns of alloys (A) and (B) for b3 (C) and (D) for b4.







Figure 5C,D. According to the phase compositions and XRD result, alloy a12 was located in three-phase region bcc $(Ti)+bcc(W)+CoTi_2$.

Figure 6 shows the microstructures of alloys a13–a15 and the XRD pattern of alloy a15. According to the BSE micrographs and the composition of each phase, dark phase was fcc (Co) (86.9 at.% Co–11.2 at.% Ti–1.9 at.% W), gray phase was Co₃Ti (78.2 at.% Co–18.6 at.% Ti–3.2 at.% W) in alloy a13 and white phase was bcc(W) (2.6 at.% Co–13.7 at.% Ti–83.7 at.% W), dark phase was CoTi (49.5 at.% Co–49.5 at.% Ti–1.0 at.% W) in alloy a14. For alloy a15, the XRD result in **Figure 6D** indicated that Co₃Ti and aCo₂Ti co-existed in this alloy. The BSE micrograph and EDS measurement confirmed that a two-phase microstructure Co₃Ti+aCo₂Ti occurred in alloy a15.

Based on the present experimental results, the isothermal section of the Co–Ti–W system at 1273 K was derived. Five three-phase regions and five two-phase regions were experimentally determined. The maximum solubility of W in Co₃Ti phase was measured to be ~7.9 at.%. In addition, the maximum solubilities of Ti in Co_7W_6 and Co_3W were ~11.9 at.% and ~15.2 at.%, respectively. And the maximum solubilities of W in αCo_2Ti , βCo_2Ti and CoTi were ~1.5 at.%, ~5.9 at.% and ~1.8 at.%, respectively.

Isothermal Section at 1473 K

10 alloys were used to investigate the isothermal section at 1473 K of the Co–Ti–W system. The compositions and phase constituents of the alloys are summarized in **Table 3**. In the Co–Ti phase diagram, two liquid phase regions exist in the Co-rich side and Ti-rich side at 1473 K, namely liquid#1 and liquid#2 in current work, respectively. The eutectic microstructures from the solidification of liquid could be observed in alloys b1, b2, b7, b8 and b10.

Figure 7A shows BSE micrograph of alloy b1. Three phases, fcc (Co), liquid#1 and Co_3Ti , were identified based on marked characteristic peaks of the XRD pattern in **Figure 7B**. According to the EDS results, the eutectic microstructure was the solidification microstructure from liquid#1 (79.4 at.% Co-19.0

Isothermal Sections of Co-Ti-W System

at.% Ti–1.6 at.% W), white and gray phases were Co₃Ti (76.5 at.% Co–14.6 at.% Ti–8.9 at.% W) and fcc (Co) (83.4 at.% Co–11.0 at.% Ti–5.6 at.% W), respectively. A three-phase region fcc (Co)+liquid#1 + Co₃Ti was determined. In the microstructure of alloy b2, liquid#1 (78.4 at.% Co–19.2 at.% Ti–2.4 at.% W) was also observed. The other two phases were determined to be Co₃Ti (75.3 at.% Co–17.3 at.% Ti–7.4 at.% W) which was gray phase and β Co₂Ti (69.8 at.% Co–23.9 at.% Ti–6.3 at.% W) which was white phase according to the results of EDS and XRD.

The BSE micrograph and XRD pattern of alloy b3 annealed at 1473 K are shown in **Figure 8A,B**. A three-phase microstructure could be observed in **Figure 8A.** According to the results of the EDS measurements, the compositions of the dark, gray and white phases were determined to be 69.1 at.% Co–25.0 at.% Ti–5.9 at.% W, 73.5 at.% Co–16.2 at.% Ti–10.3 at.% W and 57.4 at.% Co–7.6 at.% Ti–35.0 at.% W, respectively. Combined with XRD pattern shown in **Figure 8B**, dark, gray, and white phases were β Co₂Ti, Co₃Ti and Co₇W₆, respectively. As shown in **Figure 8D**, alloy b4 was composed of three phases, Co₇W₆, fcc (Co) and Co₃Ti, based on the distinguished characteristic peaks of the XRD pattern. The compositions of phases indicated that gray phase was Co₃Ti (77.3 at.% Co–12.3 at.% Ti–10.4 at.% W), white phase was Co₇W₆ (56.2 at.% Co–5.4 at.% Ti–38.4 at.% W) and dark phase was fcc (Co) (81.5 at.% Co–7.7 at.% Ti–10.8 at.% W).

All of the equilibrium microstructures in the above four alloys (b1, b2, b3 and b4) contained Co₃Ti phase. However, the Co₃Ti phase was not stable at 1473 K in the Co–Ti system. It was demonstrated that Co₃Ti phase became a stable compound at 1473 K owing to the addition of W. According to the above experimental results, the homogeneity range of W in Co₃Ti was about ~7.4–10.4 at.%. It could be inferred that the new ternary compound proposed by König et al. (König et al., 2014) may be Co₃Ti due to the relatively similar composition ranges of the two compounds.

Figure 9 shows the BSE micrographs of alloys b5-b8 annealed at 1473 K. The three-phase microstructures in four alloys were observed in the BSE images. Based on the EDS measurements, the white, lightgray and dark-gray phases in alloy b5 were bcc(W) (0.4 at.% Co-1.5 at.% Ti-98.1 at.% W), Co₇W₆ (53.2 at.% Co-15.4 at.% Ti-31.4 at.% W) and BCo2Ti (65.5 at.% Co-26.9 at.% Ti-7.6 at.% W), respectively. A three-phase equilibrium bcc(W)+CoTi+βCo₂Ti was determined in the microstructure of alloy b6. According to the results of EDS analysis, white phase was bcc(W) (4.3 at.% Co-8.1 at.% Ti-87.6 at.% W), gray phase was βCo2Ti (65.6 at.% Co-29.3 at.% Ti-5.1 at.% W) and black phase was CoTi (56.0 at.% Co-40.9 at.% Ti-3.1 at.% W). As shown in Figure 9C,D, both alloys b7 and b8 contained liquid#2 in the annealed microstructures. The eutectic structure from solidification of liquid#2 in alloy b7 was not obvious due to the low volume fraction of liquid#2. According to the results of SEM and EDS, alloys b7 and b8 were located in the three-phase regions of liquid#2 + bcc(W)+CoTi and liquid#2 + bcc(W)+bcc (Ti), respectively.

Two two-phase equilibrium microstructures were observed in the alloys b9 and b10. According to the BSE micrograph of alloy b9 in **Figure 10A** and phase compositions from EDS measurements, alloy b9 was located in the two-phase region fcc (Co)+Co₇W₆. The white and gray phases were Co_7W_6 (55.9 at.% Co–2.3 at.% Ti–41.8 at.% W) and fcc (Co) (85.0 at.% Co–2.5 at.% Ti–12.5 at.% W), respectively. The alloy b10 consisted of bcc(W) (1.3 at.% Co–33.0 at.% Ti–65.7 at.% W) and liquid#2 (24.5 at.% Co–74.9 at.% Ti–0.6 at.% W) as shown in Figure 10B.

Based on the above experimental results, the isothermal section of the Co–Ti–W system at 1473 K was constructed. Eight three-phase regions and two two-phase regions were determined in the isothermal section. The Co₃Ti phase was proved to be a stable phase and the composition range of W in Co₃Ti phase was measured to be ~7.4–10.4 at.% at the 1473 K. The maximum solubilities of Ti in Co₇W₆ and W in β Co₂Ti and CoTi at 1473 K were ~15.4 at.%, ~7.6 at.% and ~3.1 at.%, respectively.

The constructed isothermal sections at 1273 and 1473 K of the Co–Ti–W system on the basis of the above experimental data are shown in **Figure 11A,B**.

CONCLUSION

The isothermal sections at 1273 and 1473 K of the Co-Ti-W system were constructed by phase equilibrium relationships obtained using SEM/EDS, XRD methods in the current work.

In the isothermal section at 1273 K of the Co–Ti–W system, five three-phase regions, $Co_7W_6+Co_3Ti+\beta Co_2Ti$, $Co_7W_6+\beta Co_2Ti$ + bcc(W), βCo_2Ti + CoTi + bcc(W), CoTi + CoTi_2+bcc(W) and CoTi_2+bcc(W)+bcc (Ti), were experimentally determined, and three three-phase regions, fcc (Co)+Co₃W + Co₃Ti, Co₃W + Co₇W_6+Co₃Ti and Co₃Ti+\beta Co₂Ti+aCo₂Ti, were derived according to the phase relations. The experimental results showed that Co₃Ti phase could dissolve ~7.9 at.% W at 1273 K.

The isothermal section at 1473 K of the Co–Ti–W system included eight three-phase regions, fcc (Co)+Co₇W₆+Co₃Ti, fcc (Co)+Co₃Ti + liquid#1, Co₃Ti + liquid#1+ β Co₂Ti, Co₇W₆+Co₃Ti+ β Co₂Ti, bcc(W)+Co₇W₆+ β Co₂Ti, bcc(W)+ β Co₂Ti + CoTi, bcc(W)+CoTi + liquid#2 and bcc(W)+liquid#2 + bcc (Ti). The Co₃Ti phase became a stable phase at 1473 K owing to the addition of W, and the homogeneity range of W in the Co₃Ti phase was determined to be ~7.4–10.4 at.% at 1473 K.

DATA AVAILABILITY STATEMENT

The original contributions presented in the study are included in the article/Supplementary Material, further inquiries can be directed to the corresponding authors.

AUTHOR CONTRIBUTIONS

YS was responsible for conduct of the experiments, performance tests and data analysis. CG was responsible for experimental design, data analysis and paper revision. CL, ZD and DH were responsible for paper revision.

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