



# Structures, and Thermophysical Properties Characterizations of $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$ Solid Solutions as Thermal Barrier Coatings

Lin Chen, Yitao Wang, Qi Zheng and Jing Feng\*

Faculty of Material Science and Engineering, Kunming University of Science and Technology, Kunming, China

A sequence of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  solid solutions were fabricated in this work, which were studied as candidate for thermal insulation materials. The lattices were identified *via* XRD, when SEM and EDS were used to characterize the microstructures and element distributions. The results showed that the highest modulus, hardness, and toughness of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  were 196 GPa, 9.2 GPa, and  $1.6 \text{ MPa m}^{1/2}$ , respectively, and they accorded with the mechanical property requirements. Also, a low thermal conductivity ( $1.06 \text{ W m}^{-1} \text{ K}^{-1}$ ) and high thermal expansion coefficients (TECs:  $11.3 \times 10^{-6} \text{ K}^{-1}$ ) were simultaneously realized in  $(\text{La}_{3/6}\text{Ho}_{3/6})_3\text{NbO}_7$ , at high temperatures. No phase transition was detected up to  $1,200^\circ\text{C}$ , which proved their good high-temperature lattice stability. The intense anharmonic lattice vibrations might contribute to the outstanding thermal properties of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ceramics. The suitable modulus, high hardness, low thermal conductivity, and high TECs of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  solid solutions proclaimed that they were exceptional thermal insulation ceramics.

**Keywords:** TBCs, oxides, structures, thermal conductivity, mechanical properties, niobates

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### \*Correspondence:

Jing Feng  
jingfeng@kust.edu.cn

### Specialty section:

This article was submitted to  
Ceramics and Glass,  
a section of the journal  
Frontiers in Materials

**Received:** 30 April 2021

**Accepted:** 07 June 2021

**Published:** 01 July 2021

### Citation:

Chen L, Wang Y, Zheng Q and Feng J  
(2021) Structures, and  
Thermophysical Properties  
Characterizations of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$   
Solid Solutions as Thermal  
Barrier Coatings.  
Front. Mater. 8:703098.  
doi: 10.3389/fmats.2021.703098

## INTRODUCTION

Diverse oxide ceramics are studied as candidate for thermal barrier coatings (TBCs), which are used to reduce the temperatures of superalloy parts (Padture et al., 2002; Clarke, 2003; Chen et al., 2011; Zhang et al., 2016; Zhang et al., 2017; Chen et al., 2018a; Liu et al., 2018; Chen et al., 2019a; Chen and Feng, 2019; Liu et al., 2019; Zhang et al., 2019; Zhou et al., 2019; Zhang et al., 2020; Zhou et al., 2020; Zhou et al., 2021). Good thermal insulations and a long service life are researched in TBCs, which require that oxides should possess low thermal conductivity, high thermal expansion coefficients (TECs), high hardness, and so on (Padture et al., 2002; Clarke, 2003; Yang et al., 2016; Zhang et al., 2016; Chen et al., 2018a; Zhou et al., 2019). Ytria-stabilized zirconia (YSZ) ceramics are widely researched and applied as TBCs, and they have good comprehensive thermophysical properties, including high hardness and fracture toughness, comparatively low thermal conductivity, and low density (Padture et al., 2002; Clarke, 2003; Chen et al., 2011). Since the phase transition of YSZ reduces its application temperature below  $1,200^\circ\text{C}$ , new materials are researched to replace it.  $\text{La}_2\text{Ce}_2\text{O}_7$ ,  $\text{LnMgAl}_{11}\text{O}_{19}$ ,  $\text{La}_2\text{Zr}_2\text{O}_7$ , and other oxides are extensively researched as TBCs to replace YSZ, and low thermal conductivity, high hardness, and good lattice stability are found in some oxides (Wang et al., 2012; Yang et al., 2016; Zhang et al., 2016; Zhang et al., 2017; Chen et al., 2019a; Liu et al., 2019). Among various ceramics, the ones with pyrochlore-/fluorite-type structures have lower thermal conductivity and better phase stability than YSZ. Hence, oxides having

pyrochlore-/fluorite-type structures are broadly studied, and thermophysical property improvements are essential for various materials (Ma et al., 2008; Wang et al., 2012; Wright et al., 2020; Adjaoud and Albe, 2021; Irum et al., 2021; Wang et al., 2021; Xiang et al., 2021; Zhang et al., 2021; Zhu et al., 2021).

The excellent thermophysical properties of pyrochlore-/fluorite-type RE<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub> ceramics include low thermal conductivity (1–2 W · m<sup>-1</sup> · K<sup>-1</sup>, 25–1,000°C), high hardness (9–12 GPa), and good high-temperature phase stability (Wang et al., 2012; Yang et al., 2016; Liu et al., 2019; Wright et al., 2020; Xiang et al., 2021). One shortage of RE<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub> is the inadequate fracture toughness (~1 MPa m<sup>1/2</sup>), which stops it from being used as TBCs. Therefore, other oxides possessing pyrochlore-/fluorite-type structures are studied as TBCs. A<sub>3</sub>BO<sub>7</sub>-type rare earth niobates (RE<sub>3</sub>NbO<sub>7</sub>) have been studied in the past several years, and they have orthorhombic weberite and cubic fluorite lattices, which rely on RE<sup>3+</sup> ionic size (Ma et al., 2008; Zhang et al., 2017; Yang et al., 2019; Chen et al., 2021a; Chen et al., 2021b; Xiang et al., 2021). The crystal structures of RE<sub>3</sub>NbO<sub>7</sub> derive from fluorite A<sub>4</sub>O<sub>8</sub>, when the four A<sup>4+</sup> ions are substituted by three RE<sup>3+</sup> and one Nb<sup>5+</sup> cation, and one oxygen vacancy is produced in the lattice to maintain charge balance (Wang et al., 2012; Chen et al., 2018a; Chen et al., 2021a; Chen et al., 2021b). RE<sub>3</sub>NbO<sub>7</sub> is a weberite lattice when RE<sup>3+</sup> ionic size is longer than 1.03 Å, and others are cubic fluorite with RE<sup>3+</sup> size less than 1.03 Å. There are many similarities between crystal structures of RE<sub>3</sub>NbO<sub>7</sub> and RE<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub> ceramics. The lattices of RE<sub>3</sub>NbO<sub>7</sub> and RE<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub> originate from prototype fluorite A<sub>4</sub>O<sub>8</sub>; high concentration of oxygen vacancy (12.5%), [REO] polyhedrons, and excellent lattice stability are found in these two-type ceramics. Therefore, RE<sub>3</sub>NbO<sub>7</sub> and RE<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub> may have similar thermophysical properties.

Different structural and thermophysical property features are documented in weberite and fluorite RE<sub>3</sub>NbO<sub>7</sub> oxides, and they display good thermal insulation performance (Chen et al., 2018b; Yang et al., 2019; Chen et al., 2021a; Chen et al., 2021b). The advantages of RE<sub>3</sub>NbO<sub>7</sub> as TBCs include ultralow thermal conductivity, high TECs, and competitive modulus. The features of weberite and fluorite RE<sub>3</sub>NbO<sub>7</sub> have been reported, and the differences of structures and properties between these two types of niobates are distinct. To study the variations of properties of RE<sub>3</sub>NbO<sub>7</sub> along with the changes of lattice structures and to provide good understanding on these series niobates, the relevant study should be performed. The lowest thermal conductivity of weberite niobates is detected in La<sub>3</sub>NbO<sub>7</sub>, and the minimum thermal conductivity of fluorite niobates is found in Ho<sub>3</sub>NbO<sub>7</sub> oxides (Chen et al., 2018b; Chen et al., 2019b; Yang et al., 2019; Chen et al., 2021a; Chen et al., 2021b). Dense (La<sub>1-x</sub>Ho<sub>x</sub>)<sub>3</sub>NbO<sub>7</sub> solid solutions are fabricated in this work, and their lattices, thermal conductivity, hardness, and other properties are characterized. The variation trends of structures and thermophysical properties of (La<sub>1-x</sub>Ho<sub>x</sub>)<sub>3</sub>NbO<sub>7</sub> are documented, which enables scholars to study their inner dominated mechanism from a good aspect.

## EXPERIMENTS AND METHODS

The bulk samples of (La<sub>1-x</sub>Ho<sub>x</sub>)<sub>3</sub>NbO<sub>7</sub> (x = 0/6, 1/6, 2/6, 3/6, 4/6, 5/6, and 6/6) solid solutions were fabricated *via* a high-

temperature sintering process using La<sub>2</sub>O<sub>3</sub>, Ho<sub>2</sub>O<sub>3</sub>, and Nb<sub>2</sub>O<sub>5</sub> powders (purity beyond 99.9% and a particle size smaller than 20 μm). The RE<sub>2</sub>O<sub>3</sub> and Nb<sub>2</sub>O<sub>5</sub> powders were weighted and then mixed in alcohol *via* a ball mill at a speed of 300r/min for 10 h, and then they were heated at 90°C for 6 h to remove alcohol. Approximately 1–2 g dry mixtures were taken and compressed *via* a tablet press under a condition of 300 MPa for 2 min. The green bodies of (La<sub>1-x</sub>Ho<sub>x</sub>)<sub>3</sub>NbO<sub>7</sub> were finally sintered at 1,600–1700°C for 5–10 h to obtain their dense bulk samples. The lattice structures were identified *via* XRD (X-ray diffraction), and their surface microstructures were observed *via* a SEM (scanning electron microscope) and an EDS (energy dispersive spectrometer). The microhardness (*H*) and toughness (*K<sub>IC</sub>*) were tested by an indentation test (DHV-1000Z-CCD, China), when Young's modulus (*E*) and mean acoustic velocity (*V<sub>a</sub>*) were obtained by using an ultrasonic pulser/receiver equipment (UMS-100, France). The hardness and toughness of oxides were affected by bonding strengths, which were reflected *via* Young's modulus and Debye temperature ( $\Theta_D$ ) (Anderson, 1963; Schlichting et al., 2001):

$$\theta_D = \frac{h}{k_B} \left( \frac{3m}{4\pi V} \right)^{1/3} V_a, \quad (1)$$

where *h*, *k<sub>B</sub>*, and *m* represented Planck's constant, the Boltzmann constant, and the lattice weight, respectively. As for thermal properties, TECs were measured *via* a TMA (thermal mechanical analysis), and thermal diffusivity ( $\lambda$ ) was tested *via* a LFA to calculate thermal conductivity (*k*) (Schlichting et al., 2001):

$$k = \frac{\lambda \cdot \rho \cdot C_p}{1 - 4\phi/3}, \quad (2)$$

where  $\rho$ , *C<sub>p</sub>*, and  $\phi$  represented density, heat capacity, and porosity, respectively. Porosity was tested *via* an Archimedes balance, the weight (*w<sub>1</sub>*) of sample was measured first, and then the sample was put in water to obtain its weight (*w<sub>2</sub>*) under the action of water buoyancy; finally, the sample was taken out from water and wiped up to measure its weight (*w<sub>3</sub>*). The porosity was calculated *via* a formula:  $\phi = 1 - w_1/(w_3 - w_2)$ . Thermal conductivity was calculated *via* phonons, and the corresponding phonon mean free path (*l*) was obtained *via* thermal diffusivity and average acoustic velocity (Kingery, 1955; Anderson, 1963; Kittle, 1996; Schlichting et al., 2001; Qu et al., 2012; Zhao et al., 2016; Wang et al., 2018; Ye et al., 2019):

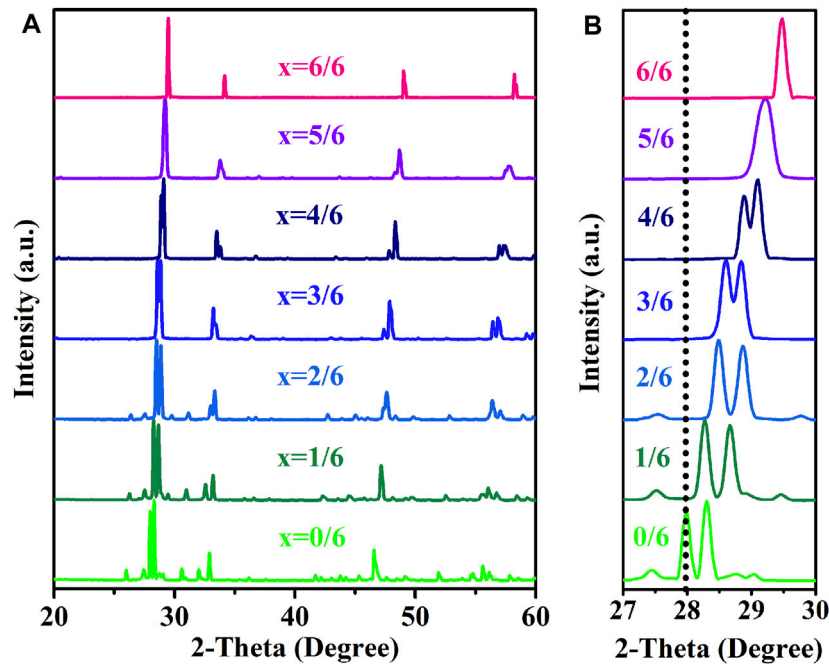
$$l = 3\lambda/V_a, \quad (3)$$

where *V<sub>a</sub>* was a constant, and it indicated that the temperature dependence of *l* relied on  $\lambda$  (Bruls et al., 2005). More details about sample preparations, structures identifications, and properties measurements could be found in our previous articles (Chen et al., 2018b; Chen et al., 2019b; Chen et al., 2021a).

## RESULTS AND DISCUSSIONS

### Structures Analysis

In the current work, the variations of properties along with the changes of lattices in (La<sub>1-x</sub>Ho<sub>x</sub>)<sub>3</sub>NbO<sub>7</sub> oxides are studied. The



**FIGURE 1** | The normalized XRD peaks of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ( $x = 0/6, 1/6, 2/6, 3/6, 4/6, 5/6,$  and  $6/6$ ) ceramics: **(A)**  $20^\circ \leq 2\text{-Theta} \leq 60^\circ$  and **(B)**  $27^\circ \leq 2\text{-Theta} \leq 30^\circ$ .

**TABLE 1** | The effective  $\text{RE}^{3+}$  size ( $r_{\text{RE}}/\text{\AA}$ ), space group (SG), lattice constants ( $a, b, c/\text{\AA}$ ), unit cell volume ( $V/\text{\AA}^3$ ), theoretical density ( $\rho_0/\text{g} \cdot \text{cm}^{-3}$ ), and porosity ( $\phi/\%$ ) of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ( $x = 0/6, 1/6, 2/6, 3/6, 4/6, 5/6,$  and  $6/6$ ) ceramics.

$x$	$r_{\text{RE}}$	SG	$a$	$b$	$c$	$V$	$\rho_0$	$\phi$
0/6	1.160	Cmcm	11.176	7.639	7.761	662.58	6.25	2.7
1/6	1.135	Cmcm	10.990	7.702	7.511	635.86	6.63	2.9
2/6	1.110	Cmcm	10.842	7.672	7.525	625.91	6.87	3.8
3/6	1.085	Cmcm	10.783	7.601	7.468	612.09	7.17	1.1
4/6	1.060	Cmcm	10.691	7.524	7.602	611.52	7.31	3.2
5/6	1.039	C222 <sub>1</sub>	10.612	7.535	7.477	597.88	7.63	3.7
6/6	1.015	Fm $\bar{3}$ m	$a = b = c = 5.260$			145.53	7.98	2.2

normalized XRD peaks of prepared specimens are depicted in **Figure 1**, and it is found that three types of lattices are formed. When  $r_{\text{RE}}$  is 1.060–1.160 Å, the prepared samples ( $0/6 \leq x \leq 4/6$ ) are in an orthorhombic phase with the space group (SG) of Cmcm; when  $r_{\text{RE}}$  is 1.039 Å, the prepared sample ( $x = 5/6$ ) is in an orthorhombic phase with the SG of C222<sub>1</sub>; and Ho<sub>3</sub>NbO<sub>7</sub> is a cubic fluorite phase with Fm $\bar{3}$ m SG. The main XRD peaks shift to the right along with the decline of La content displayed in **Figure 1B**, which indicates the shrinkage of the lattice. **Table 1** exhibits the exact information of lattices of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  solid solutions. Some certain relationships can be constructed between cubic and orthorhombic  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  lattice parameters. In cubic fluorite Ho<sub>3</sub>NbO<sub>7</sub>, it is found that  $a_c = b_c = c_c$ , when the lattice constants of orthorhombic  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  are derived from the relationships of  $a_o \approx 2a_c$ ,  $b_o \approx 2^{0.5}b_c$ , and  $c_o \approx 2^{0.5}c_c$ . The shrinkages of lattices and increments of unit cell mass lead to their theoretical density increases with the increasing Ho content. Also, **Table 1** exhibits

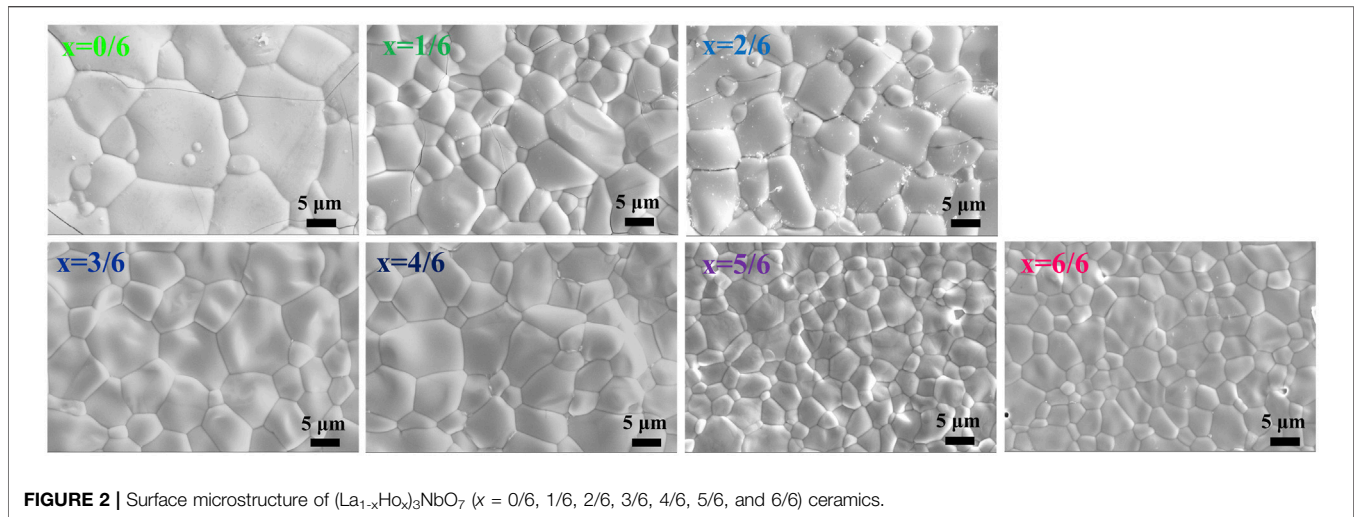
that the porosity of each sample is 1–4%, and dense bulk samples are made in this work. The impacts of porosity on thermophysical properties are not taken into account.

The grain sizes of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  are a micron scale, and most grains have sizes of 1–20 μm shown in **Figure 2**. The good combinations among neighboring grains and a small grain size may result in good mechanical properties. The sintering temperatures of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  increase from 1,600 to 1700°C in conjunction with the increasing Ho content. Navrotsky's research showed that the formation enthalpy of RE<sub>3</sub>NbO<sub>7</sub> became more exothermic with the increases of RE<sup>3+</sup> ionic radius, which indicated that the sintering temperature decreased with the increasing RE<sup>3+</sup> ionic size (Mielewczyk and Navrotsky, 2015; Chen et al., 2018b). Furthermore, the crystal structure is affected by RE<sup>3+</sup> ionic radius, and fluorite RE<sub>3</sub>NbO<sub>7</sub> has a smaller grain size than weberite RE<sub>3</sub>NbO<sub>7</sub>. Therefore, the grain size of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  is affected by the sintering temperatures and crystal structures, and they are dominated by RE<sup>3+</sup> ionic radius. The increments of sintering temperature and order–disorder (weberite–fluorite) transition of crystal structure lead to a decrease of the grain size. **Figure 3** exhibits the backscattered electron (BSE) photo and the corresponding element mappings of  $(\text{La}_{3/6}\text{Ho}_{3/6})_3\text{NbO}_7$ . No precipitated phase is found, and each element is evenly distributed in this sample. The XRD, SEM, and EDS results prove that dense and high-purity  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  samples are made via current methods.

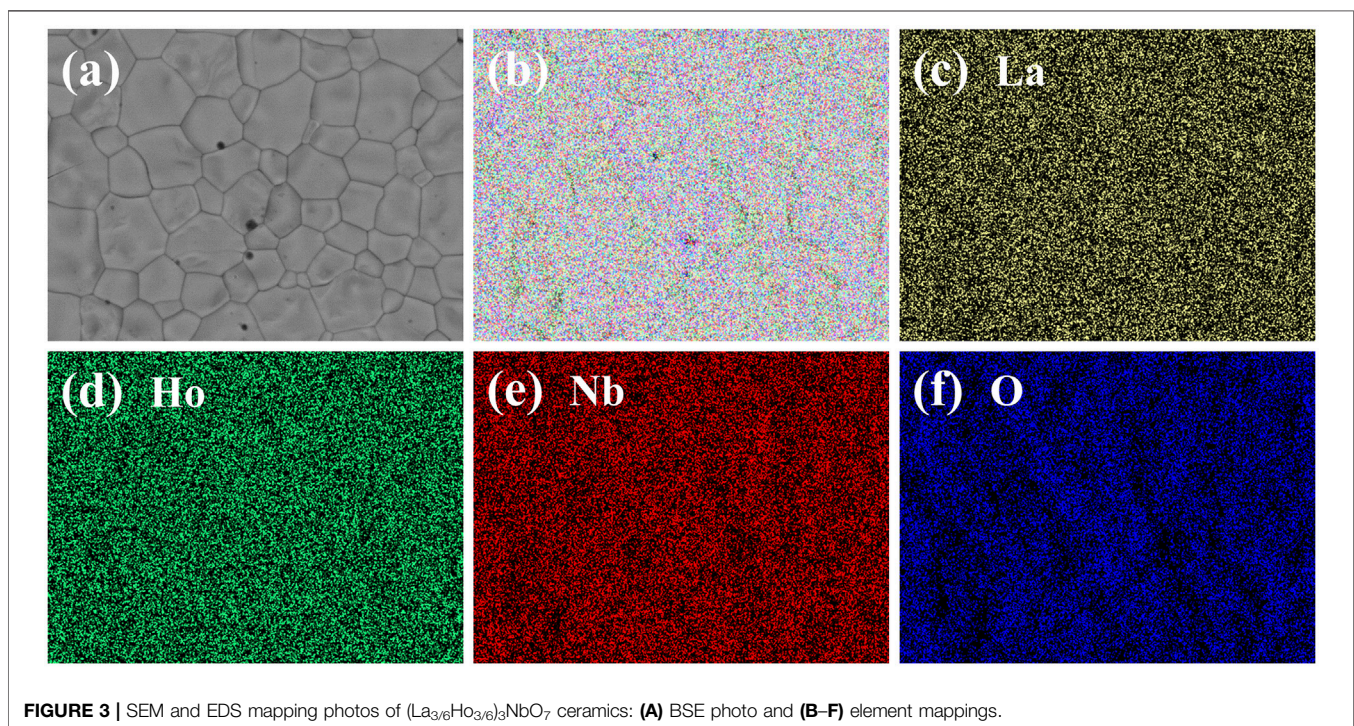
## Mechanical Properties

**Equation 1** expresses that Debye temperature is proportional to the average acoustic velocity. **Figures 4A,B** exhibit that  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ( $0/6 \leq x \leq 2/6$ ) oxides have similar Young's





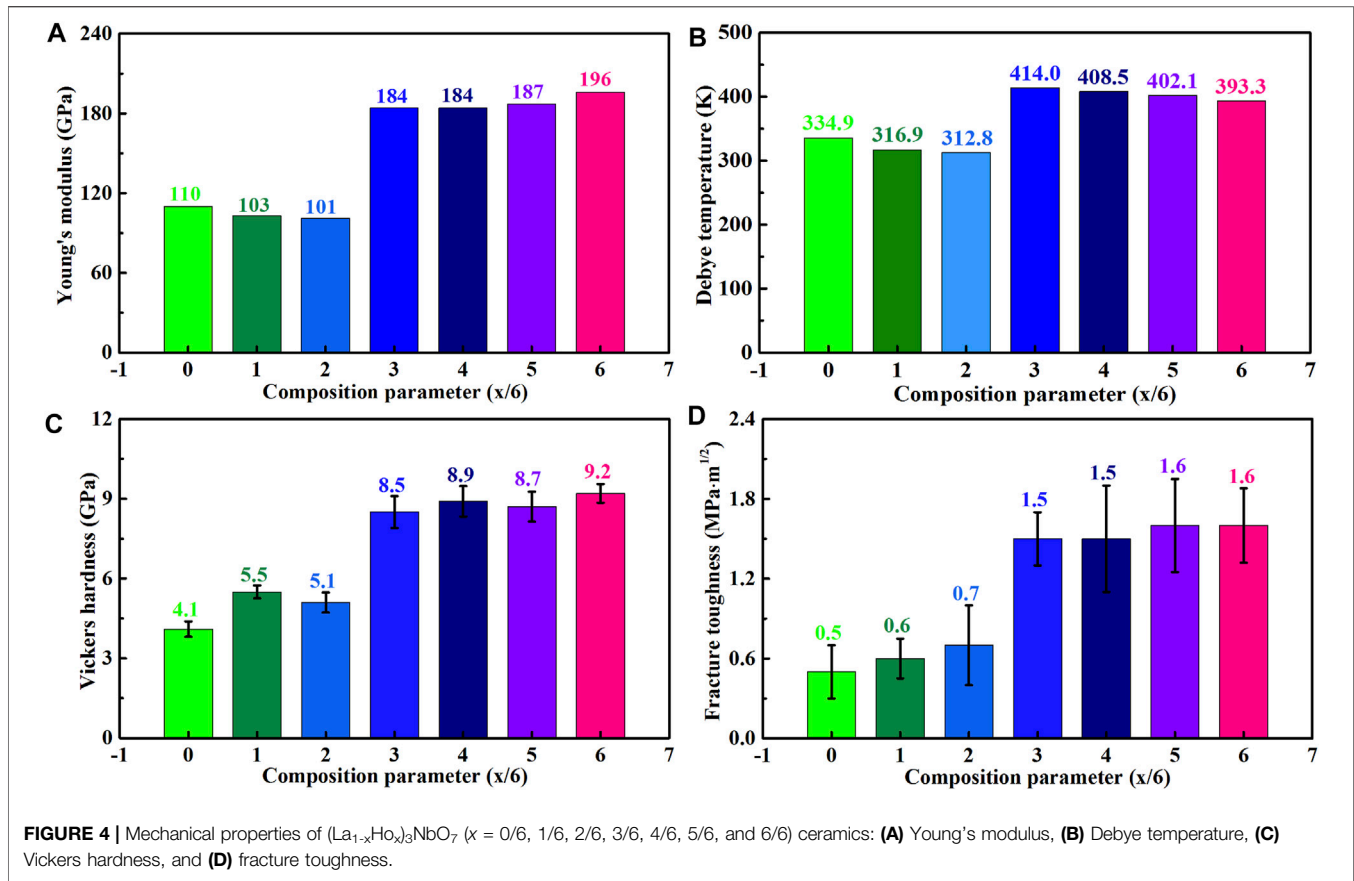
**FIGURE 2** | Surface microstructure of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ( $x = 0/6, 1/6, 2/6, 3/6, 4/6, 5/6,$  and  $6/6$ ) ceramics.



**FIGURE 3** | SEM and EDS mapping photos of  $(\text{La}_{3/6}\text{Ho}_{3/6})_3\text{NbO}_7$  ceramics: **(A)** BSE photo and **(B–F)** element mappings.

modulus (100–110 GPa) and Debye temperatures (310–340 K), implying that their bonding strengths are approximate. With the further increments of Ho content,  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ( $3/6 \leq x \leq 6/6$ ) oxides have far higher Young's modulus (180–200 GPa) and Debye temperatures (390–420 K) than those of the aforementioned samples. Similar variation trends are found in hardness and toughness of the prepared samples, as shown in **Figures 4C,D**.  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ( $3/6 \leq x \leq 6/6$ ) oxides have higher hardness and toughness than the remaining samples and their maximums are  $9.2 \pm 0.35$  GPa and  $1.6 \pm 0.28$  MPa  $\text{m}^{1/2}$ , respectively. The detail values of various mechanical properties are listed in **Table 2**. One can see that  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ceramics have higher hardness

than  $\text{LaNbO}_4$  ( $\sim 3$  GPa), while they have a lower value than YSZ (12–14 GPa),  $\text{RE}_2\text{Zr}_2\text{O}_7$  (9–12 GPa), and  $\text{LaMgAl}_{11}\text{O}_{19}$  ( $\sim 14$  GPa) (Wang et al., 2012; Chen et al., 2019a; Liu et al., 2019). Furthermore,  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ceramics have lower Young's modulus than YSZ ( $\sim 250$  GPa),  $\text{RE}_2\text{Zr}_2\text{O}_7$  (240 GPa), and  $\text{LnMgAl}_{11}\text{O}_{19}$  (300 GPa), while they have higher values than  $\text{LaNbO}_4$  (60 GPa),  $\text{Ho}_2\text{SiO}_5$  (150 GPa), and  $\text{REPO}_4$  (140 GPa) (Vassen et al., 2000; Wang et al., 2012; Feng et al., 2013; Li et al., 2018; Chen et al., 2019a; Liu et al., 2019; Wu et al., 2020). The high modulus produces poor thermal stress fracture resistance, which is indicated by a so-called *TSR* parameter (Kingery, 1955):



**TABLE 2 |** The average acoustic velocity ( $V_a/m \cdot s^{-1}$ ), Young's modulus ( $E/GPa$ ), hardness ( $H/GPa$ ), Debye temperature ( $\Theta_D/K$ ), Poisson's ratio ( $\nu$ ), and fracture toughness ( $K_{IC}/MPa \cdot m^{1/2}$ ) of (La<sub>1-x</sub>Ho<sub>x</sub>)<sub>3</sub>NbO<sub>7</sub> ( $x = 0/6, 1/6, 2/6, 3/6, 4/6, 5/6,$  and  $6/6$ ) ceramics.

$x$	$V_a$	$E$	$H$	$K_{IC}$	$\Theta_D$	$\nu$
0/6	2703.8	110.9	4.1 ± 0.29	0.5 ± 0.20	334.9	0.267
1/6	2591.3	103.7	5.5 ± 0.24	0.6 ± 0.16	316.9	0.244
2/6	2544.7	101.8	5.1 ± 0.38	0.7 ± 0.31	312.8	0.283
3/6	3343.1	184.3	8.5 ± 0.60	1.5 ± 0.20	414.0	0.249
4/6	3297.1	184.1	8.9 ± 0.57	1.5 ± 0.40	408.5	0.256
5/6	3221.9	187.0	8.7 ± 0.56	1.6 ± 0.38	402.1	0.263
6/6	3105.4	196.2	9.2 ± 0.35	1.6 ± 0.28	393.3	0.282

$$TSR = \frac{\sigma_f(1-\nu)}{E \cdot TECs}, \quad (4)$$

where  $\sigma_f$  is the flexural strength and  $\nu$  is Poisson's ratio. Evidently, comparatively low modulus is necessary for maintaining high hardness and good thermal stress fracture resistance, which is essential for the lifetime of TBCs. Besides, the highest fracture toughness of (La<sub>1-x</sub>Ho<sub>x</sub>)<sub>3</sub>NbO<sub>7</sub> is lower than that of YSZ (3.5 MPa·m<sup>1/2</sup>), but it is better than that of La<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub> (1.0 MPa·m<sup>1/2</sup>) (Schlichting et al., 2001; Yang et al., 2016; Liu et al., 2019).

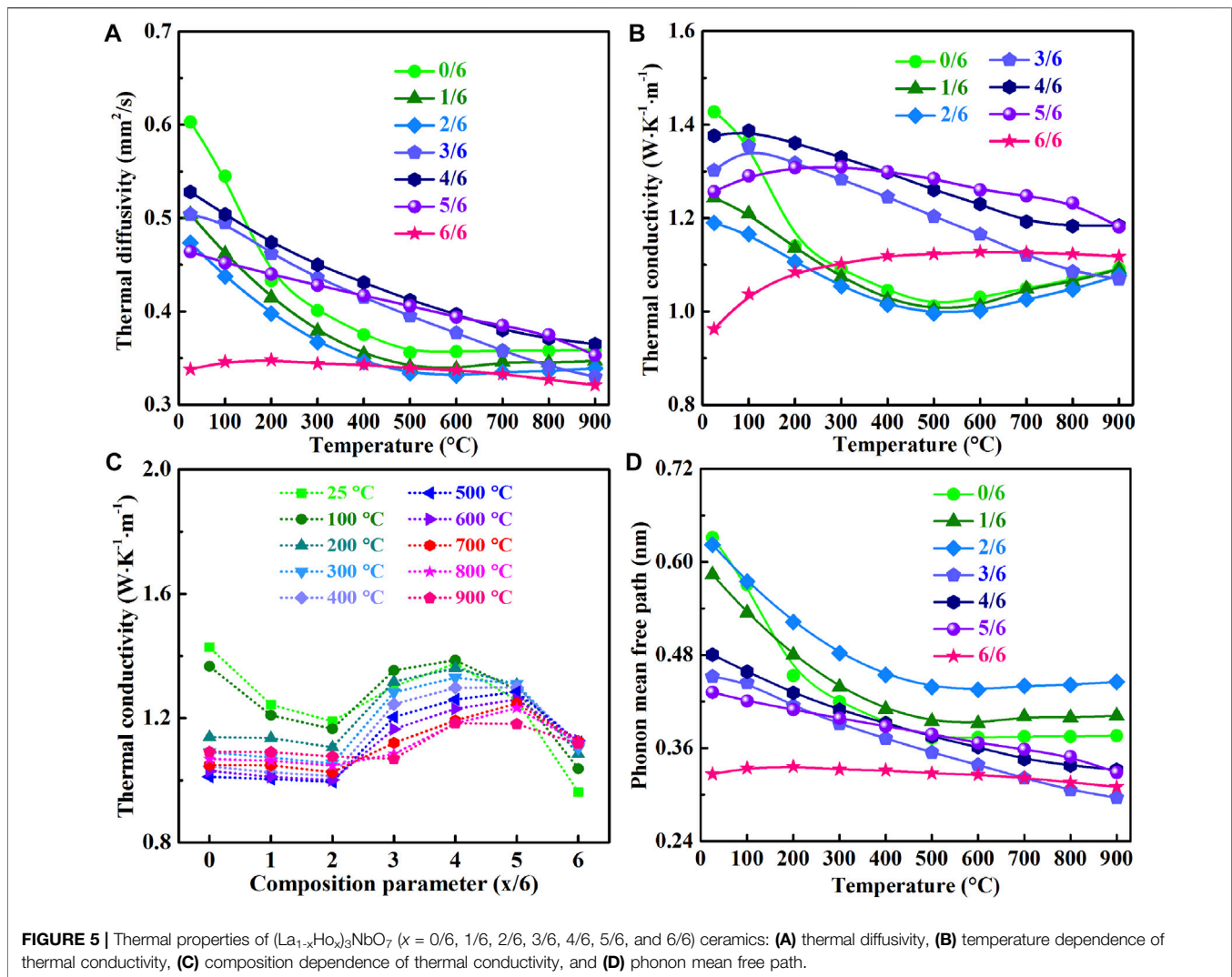
The variations of aforementioned properties are affected by bonding strengths, which are dominated by the following factors.

First, lattice shrinkages caused by the decrements of  $r_{RE}$  result in the enhancements of bonding strength as that bonding strengths increase with the decrease of bonding lengths (Zhao et al., 2016; Wang et al., 2018). Second, the Ho–O bonds have higher strengths than La–O bonds, which is reflected by their modulus and Debye temperatures. Third, **Figure 3** shows that the grain size of (La<sub>1-x</sub>Ho<sub>x</sub>)<sub>3</sub>NbO<sub>7</sub> is decreased by the increased Ho content. The decrements of grain size will lead to increases of grain boundary density, fracture toughness, and hardness.

## Thermal Properties

**Figures 5A,B,D** exhibit that thermal diffusivity and conductivity, and phonon mean free path of (La<sub>1-x</sub>Ho<sub>x</sub>)<sub>3</sub>NbO<sub>7</sub> solid solutions have the similar temperature dependences because they are dominated by phonons. At room temperature, Ho<sub>3</sub>NbO<sub>7</sub> has the lowest thermal diffusivity (0.34 mm<sup>2</sup>/s), and conductivity (0.96 W·m<sup>-1</sup>·K<sup>-1</sup>) contributed to its disorder cubic fluorite lattice (Chen et al., 2021a; Yang et al., 2019; Chen et al., 2018b). Amorphous thermal conductivity is found in Ho<sub>3</sub>NbO<sub>7</sub>, when the remaining prepared samples exhibit different temperature dependences of thermal conductivity. The low thermal conductivity originates from the short phonon mean free path shown in **Figure 5D**. The lowest value of  $k$  (1.06 W·m<sup>-1</sup>·K<sup>-1</sup>) at 900°C is found in (La<sub>3/6</sub>Ho<sub>3/6</sub>)<sub>3</sub>NbO<sub>7</sub>, and it has the shortest  $l$  (0.29 nm). It is evident that (La<sub>1-x</sub>Ho<sub>x</sub>)<sub>3</sub>NbO<sub>7</sub> has much lower thermal conductivity (0.96–1.42 W·m<sup>-1</sup>·K<sup>-1</sup>) than YSZ



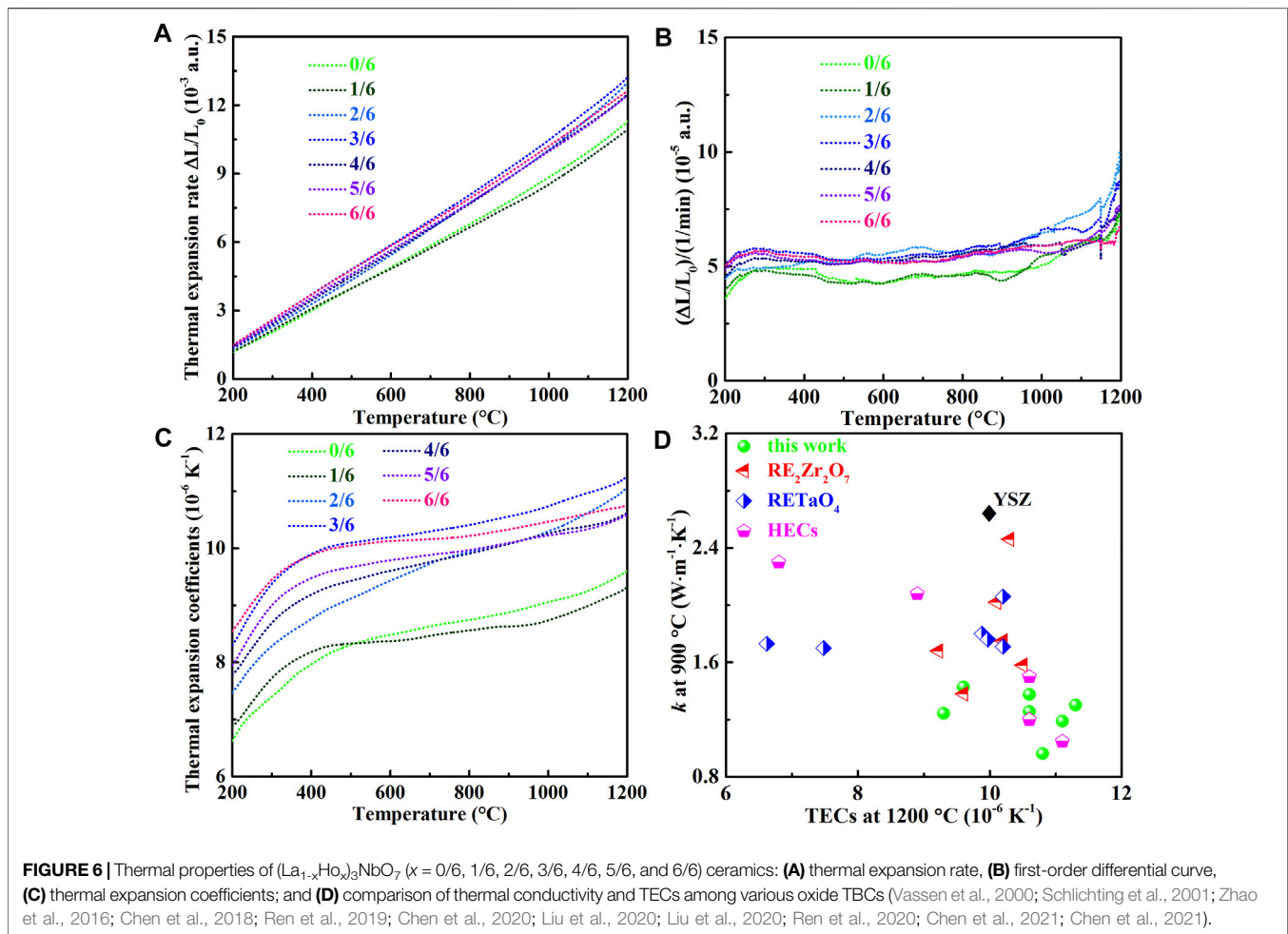


( $2.50\text{--}3.45 \text{ W} \cdot \text{m}^{-1} \cdot \text{K}^{-1}$ ),  $\text{La}_2\text{Zr}_2\text{O}_7$  ( $1.45\text{--}2.58 \text{ W} \cdot \text{m}^{-1} \cdot \text{K}^{-1}$ ),  $\text{LaNbO}_4$  ( $1.43\text{--}3.18 \text{ W} \cdot \text{m}^{-1} \cdot \text{K}^{-1}$ ), and other candidate oxides (Vassen et al., 2000; Feng et al., 2013; Zhao et al., 2016; Chen et al., 2018b; Li et al., 2018; Liu et al., 2019; Ren et al., 2019; Liu et al., 2020a; Liu et al., 2020b; Chen et al., 2020; Ren et al., 2020; Wu et al., 2020; Chen et al., 2021a; Chen et al., 2021b).

During the substitution process of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$ , misfits of atomic mass and ionic size are introduced to the lattice to enhance phonon scattering strength, and thermal conductivity is reduced (Wang et al., 2012; Feng et al., 2013; Zhao et al., 2016; Liu et al., 2019; Liu et al., 2020b). Normally, thermal conductivity decreases with the increasing phonon scattering strength caused by point defects, which can be reflected by the phonon scattering coefficient. It is believed that the phonon scattering coefficient of  $(\text{La}_{3/6}\text{Ho}_{3/6})_3\text{NbO}_7$  reaches the maximum based on phonon point defect scattering theory. However, besides phonon point defect scattering, influence of phonon propagation speed on thermal conductivity should be considered according to the evident variations of sound speed. The average acoustic velocities of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ( $x = 3/6, 4/6, 5/6, 6/6$ ) ( $3,100\text{--}3,400 \text{ ms}^{-1}$ )

are far higher than those of rest samples ( $2,500\text{--}2,700 \text{ ms}^{-1}$ ), indicating that  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ( $x = 3/6, 4/6, 5/6, 6/6$ ) oxides have far higher phonon propagation speed than the others. Therefore, the thermal conductivity of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ( $x = 0/6, 1/6, 2/6$ ) decreases with the increasing Ho content, and it increases with the further increments of Ho content in  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ( $x = 3/6, 4/6, 5/6$ ) (Figure 5C). Both point defect phonon scattering strength and changes in phonon propagation speed play essential roles on regulations of thermal conductivity of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ceramics.

Besides thermal conductivity, high TECs are essential for the service of TBCs. Figure 6A displays that the deformation of each sample increases with the increasing temperature, and their slopes are almost constant, which are further proved *via* the first-order differential curves depicted in Figure 6B. The phase transition and corresponding temperature can be detected by TMA, which have been proven in our previous research studies on  $\text{RE}_3\text{NbO}_7$  and  $\text{RENbO}_4$  ceramics (Chen et al., 2018b; Wu et al., 2020). It is believed that  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ceramics have a stable lattice up to  $1,200^\circ\text{C}$ . The TECs of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  increase with



the increments of temperature, as shown in **Figure 6C**, and the maximal value ( $11.3 \times 10^{-6} \text{K}^{-1}$ ) at  $1,200^\circ\text{C}$  is found in  $(\text{La}_{3/6}\text{Ho}_{3/6})_3\text{NbO}_7$ . One drawback of YSZ (TECs:  $10 \times 10^{-6} \text{K}^{-1}$ ) and  $\text{RE}_2\text{Zr}_2\text{O}_7$  (TECs:  $9 \times 10^{-6} \text{K}^{-1}$ ) is the low TECs, which produce high thermal stress during high-temperature applications (Yang et al., 2016; Schlichting et al., 2001; Kingery, 1955; Vassen et al., 2000). The lowest high-temperature thermal conductivity and highest TECs are simultaneously achieved in  $(\text{La}_{3/6}\text{Ho}_{3/6})_3\text{NbO}_7$  because of the strongest anharmonic lattice vibrations. Normally, the anharmonic lattice vibrations increase with the increasing temperature, which leads to decrease of thermal conductivity and increase of TECs. The TECs and thermal conductivity of  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  are shown in **Figure 6D** to compare with other TBC oxides. The prepared  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  samples have lower thermal conductivity and higher TECs than current YSZ,  $\text{RE}_2\text{Zr}_2\text{O}_7$ ,  $\text{RETaO}_4$ , and distinct high-entropy ceramics (HECs) (Vassen et al., 2000; Schlichting et al., 2001; Zhao et al., 2016; Chen et al., 2018b; Ren et al., 2019; Liu et al., 2020a; Liu et al., 2020b; Chen et al., 2020; Ren et al., 2020; Chen et al., 2021a; Chen et al., 2021b). It is believed that good thermal insulation performances and low thermal stress can be provided, when  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ceramics are applied as TBCs.

## CONCLUSION

Dense  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  oxide bulks have been fabricated in this work, and their structures and thermophysical properties are characterized. The excellent thermophysical properties indicate that the prepared samples are candidate TBCs, and following conclusions are obtained:

- 1) Dense and high-purity  $(\text{La}_{1-x}\text{Ho}_x)_3\text{NbO}_7$  ceramics are obtained by a solid-state sintering process, and phase transitions of  $\text{Cmcm} \rightarrow \text{C222}_1 \rightarrow \text{Fm}\bar{3}\text{m}$  SG are detected along with the increments of Ho content. The grain sizes are a micron scale, and each element is evenly distributed in these samples.
- 2) The mechanical properties are affected by their bonding strengths and grain sizes, and the changes of modulus and Debye temperatures prove that their bonding strengths enhance with the increments of Ho content, which is caused by the shortening of bonding length and addition of stiff Ho–O bonds. The highest modulus, hardness, and toughness are 196 GPa, 9.2 GPa, and  $1.6 \text{MPa m}^{1/2}$ , respectively.

- 3) Good thermal insulations ( $k$ : 1.06 W m<sup>-1</sup> K<sup>-1</sup> at 900°C), high TECs (11.3 × 10<sup>-6</sup> K<sup>-1</sup> at 1,200°C), and excellent high-temperature phase stability are simultaneously realized in (La<sub>3/6</sub>Ho<sub>3/6</sub>)<sub>3</sub>NbO<sub>7</sub> ceramics, which are far better than those of YSZ, RE<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub>, RETaO<sub>4</sub>, and other TBCs. The intense anharmonic lattice vibrations lead to obvious thermal property improvements.

## DATA AVAILABILITY STATEMENT

The original contributions presented in the study are included in the article/Supplementary Material; further inquiries can be directed to the corresponding author.

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## AUTHOR CONTRIBUTIONS

LC conducted the experiments and wrote this paper, YW and QZ conducted partial experiments on thermal conductivity and thermal expansions, and JF designed the detailed experiments, and discussed and improved this article.

## FUNDING

The current work was supported by the National Natural Science Foundation of China (No. 91960103 and 51762028) and Yunnan Province Materials Genome Engineering (No. 2018ZE019).

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**Conflict of Interest:** The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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