



## The Stability and Elasticity in Ta-Ti-V Medium-Entropy Alloys Using First-Principles Calculations

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Due to the low neutron activity, the high thermal stability, and the good compression strength at elevated temperatures, medium-entropy Ta-Ti-V alloys have the application potential to be used in future nuclear engineering. According to experimental results, a special quasi-random structure (SQS) model and first-principles calculations are used to calculate the structural stability and the elasticity of the ternary single-phase body-centered cubic (BCC) Ta-Ti-V alloys with the compositions Ta<sub>24</sub>Ti<sub>14</sub>V<sub>16</sub>, Ta<sub>19</sub>Ti<sub>17</sub>V<sub>18</sub>, Ta<sub>16</sub>Ti<sub>21</sub>V<sub>17</sub>, Ta<sub>22</sub>Ti<sub>23</sub>V<sub>9</sub>, and Ta<sub>18</sub>Ti<sub>18</sub>V<sub>18</sub>. The mixing energy of five compositions is positive, which indicates that they are unstable at low temperature but stable at high temperature. Meanwhile, we predict the ability of elasticity in those five compositions, and the sequence of bulk modulus is  $Ta_{24}Ti_{14}V_{16} > Ta_{19}Ti_{17}V_{18} > Ta_{18}Ti_{18}V_{18} > Ta_{16}Ti_{21}V_{17} > Ta_{22}Ti_{23}V_9$  and ductility is  $Ta_{19}Ti_{17}V_{18} > Ta_{18}Ti_{18}V_{18} > Ta_{22}Ti_{23}V_9 > Ta_{24}Ti_{14}V_{16} > Ta_{16}Ti_{21}V_{17}$ . Additionally, the influence of the element content on elastic properties shows that the Ta content will increase the bulk modulus of the alloys, while Ti content decreases the bulk modulus. Higher V and Ti contents are beneficial to ductility, especially the V content. The higher Ta content will sharply reduce the ductility of the alloys. Our calculation has provided further theoretical mechanism for the Ta-Ti-V medium-entropy alloys (MEAs) preparation and development.

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## INTRODUCTION

The recently developed novel concept of high-entropy alloys (HEAs), named by Yeh et al. (2004), has attracted an extensive attention in recent decades from both experimentalists and theoreticians. Initially, HEAs were loosely defined as single-phase solid solutions with at least five principal elements, each with an atomic percentage between 5 and 35% (Gao et al., 2016). These alloys owe their name to the assumption that the solid solution is stabilized by high configurational entropy of mixing. Meanwhile, the definition has been broadened to also include four-component alloys. Alloys with three and four principal elements are sometimes referred to as medium-entropy alloys (MEAs). Like HEAs, MEAs also exhibit good thermal stability (Wu et al., 2014; Sathiyamoorthi et al., 2017), high strength (Wu et al., 2014; Zhao et al., 2017), and excellent ductility (Gludovatz et al., 2016; Fu et al., 2018). Usually, HEAs can form a simple microstructure, e.g.,

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body-centered cubic (BCC) or face-centered cubic (FCC), rather than complex intermetallic phases (Zhang et al., 2014).

In Ta-Ti-V MEAs (Jia et al., 2019), experimental results showed that they all exhibit BCC structures in their homogenized states. The four nominal compositions in this work are  $Ta_{40}Ti_{27}V_{33}$ ,  $Ta_{27}Ti_{40}V_{33}$ ,  $Ta_{35}Ti_{45}V_{20}$ , and the equiatomic Ta33Ti33V34. They have examined their mechanical properties including ductility at room temperature and compression resistance at high temperature up to 800°C, and Ta<sub>35</sub>Ti<sub>45</sub>V<sub>20</sub> exhibits the best compression resistance. Except for Ta<sub>40</sub>Ti<sub>27</sub>V<sub>33</sub>, the other three alloys have an acceptable tensile ductility of 7.6-13% fracture strain at room temperature. However, to date, the other elastic properties including shear modulus, Young's modulus, and the influence of elemental content on elastic properties of Ta-Ti-V systems are still unknown. Fortunately, special quasi-random structures (SQSs) (Zunger et al., 1990) in combination with first-principles calculations (Perdew and Zunger, 1981) offer the possibility to study these properties. Theoretically, an infinite unit cell is needed to model random structures. The larger a unit cell is, the more first-principles calculations are needed. The SQS can effectively reduce the modeling size and provide reliable results (Wen et al., 2018).

In this study, we calculate the structural parameters, the stability, the elastic properties, and the influence of elemental content on elastic properties of Ta-Ti-V MEAs using first-principles calculations according to the SQS models. A series of useful results have been obtained including the elastic properties of different compositions for Ta-Ti-V MEAs and the influence of single element's content on their ductility and deformation resistance.

#### **COMPUTATION METHODS AND DETAILS**

#### SQS Method

The SQS method (Wei et al., 1990; Zunger et al., 1990) essentially finds small unit cell special structure, which can exactly reproduce the pair and multisite correlation functions of a random alloy between the first few nearest neighbors, deferring periodicity errors to more distant neighbors. In practice, convergence tests need to be performed in order to ensure that long-range interatomic interactions have negligible effects on the property of interest, such that the SQS method remains applicable. In the studies by Jiang et al. (2004) and Jiang (2009), rapid convergence of SQS calculated alloy properties (e.g., formation energy, lattice parameter, and band gap) with respect to the SQS size has been observed, which strongly indicates that those properties are indeed dominated by interactions between near neighbors. In such a case, even small SQSs are already sufficient to provide reliable results. In this study, according to the chemical compositions of the experimental samples (Jia et al., 2019), we generated various SQS-N structures (N is the atom number per SQS unit cell) with a high approximation atom percentage compositions of random BCC alloys. Our search criterion requires that the pair correlation functions of the SQS should be as close as those of the corresponding random alloys; it needs at least up to the 10th nearest neighbor; the second nearest

neighbor for both ternary and quaternary correlation functions. We use the mcsqs code in the Alloy-Theoretic Automated Toolkit (ATAT) (van de Walle et al., 2002) to search SQSs.

## **First-Principles Calculations**

All the first-principles calculations are employed by using VASP (Vienna *ab initio* simulation package) (Kresse and Furthmüller, 1996; Kresse and Furthüller, 1996). Projector-augmented wave (PAW) potentials (Blöchl, 1994) and Perdew–Burke–Ernzerhof generalized gradient approximation (GGA-PBE) (Perdew et al., 1996) for the exchange–correlation function were performed. Plane-wave cutoff energy of 400 eV is used. A  $3 \times 3 \times 3$  BCC supercell containing 54 atoms is used, and the Brillouin zone is sampled with a  $5 \times 5 \times 5$  k-point mesh based on the Monkhorst-Pack scheme. The total energy convergence criterion of the self-consistent field (SCF) is set to be  $1.0 \times 10^{-5}$  eV/atom, and the first-order Methfessel-Paxton method with a width of 0.1 eV is implemented to fully relax the structures. Once the Hellmann–Feynman force acting on each atom is less than 0.01 eV/Å, the atoms are regarded as being fully relaxed.

The mixing and formation energy (Liu et al., 2005) of a random BCC alloy with compositions  $x_{Ti}$  and  $x_V$  are obtained as:

$$\Delta E_{\text{form}} = E (\text{SQS}) - (1 - x_{\text{Ti}} - x_{\text{V}}) E (\text{Ta, BCC})$$
$$-x_{\text{Ti}} E (\text{Ti, HCP}) - x_{\text{V}} E (\text{V, BCC})$$
(1)

$$\Delta E_{\text{mix}} = E (\text{SQS}) - (1 - x_{\text{Ti}} - x_{\text{V}}) E (\text{Ta, BCC})$$
$$-x_{\text{Ti}} E (\text{Ti, BCC}) - x_{\text{V}} E (\text{V, BCC})$$
(2)

where E(Ta), E(Ti), E(V), and E(SQS) are total energies (per atom) of the constituent elements Ta, Ti, and V from firstprinciples calculations and the corresponding SQS result. All the calculations are performed based on the full structural optimization. The pure elements BCC Ta, hexagonal closepacked (HCP) Ti, and BCC V structures are used as reference states in Eq. (1), noting that it takes BCC Ta for the mixing energy calculation in Eq. (2) for the SQS Ta-Ti-V MEAs with the BCC structure.

#### **Elastic Properties**

We can apply small strains to the equilibrium lattice, determine the resulting change in the total energy, and, from this information, deduce the elastic constant  $C_{ij}$ . The elastic constants are identified as proportional to the second-order coefficient in a polynomial fit of the total energy as a function of the distortion parameter  $\delta$  (Li, 2014). By using the elastic constant  $C_{ij}$ , the mechanical stability of the cubic crystal can be estimated by the Born stable criterion [as Eq. (3)] (Ameri et al., 2014). Additionally, the polycrystalline elastic properties, including bulk modulus *B*, shear modulus *G*, and Young's modulus *E*, are averaged by the Voigt-Reuss-Hill schemes [as Eq. (4)] (Hill, 1952). Here, *B* value is used to represent the ability of resistance of elastic deformation of the material, while it also uses Pugh ratio *B/G* value to represent the brittleness or ductility of the material (Push, 1954). According to Pugh's classification, the critical value of *B*/*G* is 1.75. Meanwhile, the higher the *B*/*G* ratio, the better the ductility, otherwise more brittle.

$$C_{11} + 2C_{12} > 0, C_{11} - C_{12} > 0, C_{44} > 0;$$
 (3)

$$\begin{cases} B_{\rm V} = \frac{1}{3} \times (C_{11} + 2C_{12}); \ B_{\rm R} = B_{\rm V}; \\ G_{\rm V} = \frac{1}{5} \times (C_{11} - C_{12} + 3C_{44}); \ G_{\rm R} = \frac{5 \times (C_{11} - C_{12}) \times C_{44}}{4C_{44} + 3 \times (C_{11} - C_{12})} \\ B = \frac{B_{\rm V} + B_{\rm R}}{2}; \\ G = \frac{G_{\rm V} + G_{\rm R}}{2}; \\ E = 9B \times \frac{G}{3B + G}; \end{cases}$$

$$(4)$$

#### **RESULTS AND DISCUSSION**

#### **Convergence Tests**

In order to test the appropriate amount of atoms for the SQS-N structure, N is set as 24, 36, 54, and 120. To highly reflect the same proportion of concentration of each element in experiment (Jia et al., 2019), the same chemical compositions (in at.%) are set for Ta, Ti, and V, respectively, in this study. In **Figure 1**, the SQS calculated mixing energy  $\Delta E_{mix}$  (i.e., mixing enthalpies at 0 K) and equilibrium lattice constant (*a*) of BCC Ta-Ti-V alloys with equiatomic compositions, which are plotted as a function of N. This shows a rapid convergence of the SQS calculation with respect to N. Remarkably, our SQS-24 structure gives identical lattice constants as those given by the SQS-120 structure. For mixing energies, the difference between SQS-24 structure and SQS-120 structure is less than 0.5 kJ/mol, and the results from the SQS-54 structure is nearly equal to those obtained using SQS-120. In order to perform an efficient and economical calculation process, we considered SQS-54 in further elastic properties calculation, and it resulted in an accepted comparison for compositions with experiments. For example, the lattice constant *a* is from 3.184(9) Å when N = 120, which agrees with the parameter with 3.208  $\pm$  0.005 Å of equiatomic Ta<sub>34</sub>Ti<sub>33</sub>V<sub>33</sub> in experiment (Jia et al., 2019). The predicted structures with 54 atoms from SQS are given as Ta<sub>24</sub>Ti<sub>14</sub>V<sub>16</sub>, Ta<sub>19</sub>Ti<sub>17</sub>V<sub>18</sub>, Ta<sub>16</sub>Ti<sub>21</sub>V<sub>17</sub>, Ta<sub>22</sub>Ti<sub>23</sub>V<sub>9</sub>, and Ta<sub>18</sub>Ti<sub>18</sub>V<sub>18</sub>. Their crystalline structures are illustrated in **Figure 2**.

#### **Structural and Stable Properties**

The calculated equilibrium lattice constants for pure BCC Ta, HCP Ti, and BCC V are given in **Table 1**. The relative error of the calculation value of the lattice constants of Ta, Ti, and V elements compared with experimental data (Martienssen and Warkimont, 2005) is less than 1.7%. The results reveal that our lattice constants calculations agree well with experimental values.

The calculated equilibrium lattice constant *a*, mixing energy  $\Delta E_{\text{mix}}$ , and formation energy  $\Delta E_{\text{form}}$  of Ta-Ti-V MEAs are summarized in **Table 2**. We note that Ta content can increase, but the V content can decrease the lattice constant of the alloy. This is mainly due to the higher Ta and the lower V lattice constant. The necessary requirement to form a single-phase solid solution is that the mixing energy should satisfy  $-16.25 \text{ kJ/mol} \leq \Delta E_{\text{mix}} \leq 5 \text{ kJ/mol}$  (Gao et al., 2017), and the mixing energy of all the predicted alloys' compositions satisfies this condition. From





**Table 2**, we can see that their mixing energies are all positive, which suggest they are unstable at a low temperature and stable at a high temperature. This phenomenon has also been observed from an experiment of the equiatomic composition (Jia et al., 2019). By using the formula  $G = H_{SQS} - TS_{conf}$  [G is Gibbs free energy and  $H_{SQS}$  is formation enthalpy, which is equal to the formation energy at 0 K; in case of ternary equiatomic composition,  $S_{conf} = 1.1R$ ,  $R = 8.314 \text{ J/(mol} \cdot \text{K})$  (Yeh et al., 2004)], it can be roughly estimated that BCC Ta<sub>18</sub>Ti<sub>18</sub>V<sub>18</sub> can exist stably at T = 760 K, which is similar to the testing temperature (Jia et al., 2019).

#### **Elastic Properties**

The calculated single-crystal elastic constants, bulk modulus, shear modulus, Young's modulus, and B/G ratio for pure BCC Ta, Ti, and V are listed in **Table 3**.

TABLE 1 | Calculated equilibrium lattice constants (Å) for pure elements. Lattice constants Ti ۷ Та a = 2.92, c = 4.63This work a = 2.98a = 3.31Expt.\* (Martienssen and a = 2.95, c = 4.69a = 3.03a = 3.30Warkimont, 2005) \*The experimental values temperature are at room (Martienssen and Warkimont, 2005).

It can be obtained that the prediction of  $C_{44}$ , B, G, and E of Ta and Ti elements are in line with the experimental values<sup>1</sup> (Shang et al., 2010). However, for V atom, the difference between the calculated elastic constant values  $C_{44}$ , B, G, E, and B/G and its experimental values<sup>1</sup>, (Shang et al., 2010) is relatively large, especially  $C_{44}$ , which may be due to a deviation between V calculated lattice constant and its experimental value (Martienssen and Warkimont, 2005). This problem can be solved by treating semi-core 3p electrons of V as valence. We can see that Ta has the best material stiffness characterized by bulk modulus B = 200.97 GPa and Ti has the highest brittleness characterized by B/G ratio among the three elements. The calculated single-crystal elastic constants, bulk modulus, shear modulus, Young's modulus, and B/G ratio for Ta-Ti-V MEAs are listed in **Table 4**.

<sup>1</sup>https://periodictable.com/Elements/073/data.html

TABLE 2	Calculated equilibrium lattice constant a (Å), mixing energy
$\Delta E_{\rm mix}$ (kJ	mol), and formation energy $\Delta E_{\text{form}}$ (kJ/mol) for Ta-Ti-V MEAs.

Compositions	Lattice constants a	$\Delta E_{mix}$	$\Delta E_{form}$
Ta <sub>24</sub> Ti <sub>14</sub> V <sub>16</sub>	3.203	4.48	6.98
Ta <sub>19</sub> Ti <sub>17</sub> V <sub>18</sub>	3.185	3.80	6.84
Ta <sub>16</sub> Ti <sub>21</sub> V <sub>17</sub>	3.185	2.98	6.73
Ta <sub>22</sub> Ti <sub>23</sub> V <sub>9</sub>	3.229	2.27	6.38
Ta <sub>18</sub> Ti <sub>18</sub> V <sub>18</sub>	3.184	3.76	6.97

TABLE 3 | Calculated elastic constant C<sub>ij</sub> (GPa), bulk modulus B (GPa), shear modulus G (GPa), Young's modulus E (GPa), and B/G ratio for pure elements, combined with experiment values\*.

		<b>C</b> <sub>11</sub>	C <sub>12</sub>	<b>C</b> <sub>44</sub>	В	G	E	B/G
Та	This work	278.35	162.28	70.48	200.97	65.21	176.53	3.08
	Expt. https://periodictable.com/Elements/073/data.html (Shang et al., 2010)	260.23	154.46	82.55	200	69	186	2.90
Ti	This work	175.54	92.52	39.41	117.91	46.49	123.28	2.54
	Expt. https://periodictable.com/Elements/073/data.html (Shang et al., 2010)	162.4	92	46.7	110	44	116	2.50
V	This work	281.62	141.01	15.57	187.88	30.04	85.55	6.25
	Expt. https://periodictable.com/Elements/073/data.html (Shang et al., 2010)	228.7	119	43.2	160	47	128	3.40

\*The experimental values are at room temperature.

TABLE 4 | Calculated elastic constant C<sub>ij</sub> (GPa), bulk modulus B (GPa), shear modulus G (GPa), Young's modulus E (GPa), and B/G ratio for Ta-Ti-V MEAs.

Compositions	C11	C12	Caa	В	G	E	B/G
	228 53	134.19	37.78	165.64	41 29	114 38	4 01
Ta <sub>19</sub> Ti <sub>17</sub> V <sub>18</sub>	219.17	129.9	31.73	159.66	36.39	101.45	4.39
Ta <sub>16</sub> Ti <sub>21</sub> V <sub>17</sub>	201.8	126	37.95	151.27	37.93	105.02	3.99
Ta <sub>22</sub> Ti <sub>23</sub> V <sub>9</sub>	197.53	127.76	37	151.01	36.14	100.41	4.18
Ta <sub>18</sub> Ti <sub>18</sub> V <sub>18</sub>	207.73	128.7	33.11	155.04	35.54	99.05	4.36

The mechanical stability of the MEAs structure has a great significance in the manufacture and applications of the material. Regarding the Born stability criteria, the stability of cubic crystal is represented in Eq. (3). As the elastic strain energy is required to be positive, it can be seen from **Table 4** that  $Ta_{24}Ti_{14}V_{16}$ ,  $Ta_{19}Ti_{17}V_{18}$ ,  $Ta_{16}Ti_{21}V_{17}$ ,  $Ta_{22}Ti_{23}V_9$ , and  $Ta_{18}Ti_{18}V_{18}$  all meet the conditions of mechanical stability.

It is also known from Table 4 that  $Ta_{24}Ti_{14}V_{16}$  has the best bulk modulus and Ta19Ti17V18 has the best ductility. Meanwhile, B and B/G value of  $Ta_{18}Ti_{18}V_{18}$  are both smaller than Ta<sub>19</sub>Ti<sub>17</sub>V<sub>18</sub>, indicating that Ta<sub>19</sub>Ti<sub>17</sub>V<sub>18</sub> has higher bulk stiffness and ductility than Ta<sub>18</sub>Ti<sub>18</sub>V<sub>18</sub>, which means that the equiatomic ratio composition may have the good thermal stability, while it is not necessary for the best mechanical performance. The elasticity for all predicted compositions has been compared, such as bulk modulus:  $Ta_{24}Ti_{14}V_{16} >$  $Ta_{19}Ti_{17}V_{18} > Ta_{18}Ti_{18}V_{18} > Ta_{16}Ti_{21}V_{17} > Ta_{22}Ti_{23}V_9$  and ductility:  $Ta_{19}Ti_{17}V_{18} > Ta_{18}Ti_{18}V_{18} > Ta_{22}Ti_{23}V_9 > Ta_{24}Ti_{14}V_{16}$  $> Ta_{16}Ti_{21}V_{17}$ . The trend is similar from the behavior of compression resistance in experiment (Jia et al., 2019), for example, increasing the Ta content can enhance the material strength. For the ductility, it can also be concluded that Ta is not favorable for the material ductility, while a relatively higher Ti can increase the ductility of the alloy, and the effect of V is between Ti and Ta.

# Influence of Element Content on Elastic Properties

The influence of element content on the elastic properties of Ta-Ti-V MEAs is also investigated. Considering that elastic properties are binary functions of the element content, we assume that the first variable is the ratio of random two elements' contents and the other variable is the remaining one element's content. The calculation is carried out by keeping one element's content unchanged, then adjusting the ratio of the remaining two elements' contents, and calculating the elastic constants of the system. Here, taking equal composition  $Ta_{18}Ti_{18}V_{18}$  as the benchmark, for example, the Ti content is fixed to 33.3 at.%, then the ratio of Ta and V content is adjusted from 0.385 to 2.6, and the elastic constants of the system are calculated. In this case, the relationship between the bulk modulus, shear modulus, Young's modulus, and *B/G* with the ratio of different elements' contents is shown in **Figure 3**.

It can be seen that the elastic properties vary with the ratio of different elements' contents. In **Figure 3A**, the bulk modulus shows a monotonous change. Specifically, the ratio of Ta and V content to V and Ti content can improve the material stiffness, while the ratio of Ti and Ta content is opposite. Consequently, Ta is the most favorable element to improve the material stiffness, while Ti content is the most unfavorable, which also confirms the data from the experimental. For the partial reason, Ta has the highest melting temperature of 3287 K, when compared to the temperatures 1943 K and 2199 K of Ti and V, respectively. Additionally, Ta has the highest elastic module among the three elements; it could make the alloy stiffness enhanced when the alloy reaches a high percentage of Ta atom.

Furthermore, in **Figures 3B,C**, *G* and *E* of the system firstly decrease and then increase with the increase of the Ta/V ratio, indicating that both higher Ta and V content will rise the material shear and the Young's modulus. Additionally, both *G* and *E* obviously increase and then slightly decrease with the increase of V/Ti ratio, suggesting that the higher Ti content can reduce the shear and Young's modulus. Finally, with the increase of Ti/Ta ratio, the trend of *G* and *E* decreases significantly, which is also consistent with the previous analysis. In general, the most favorable element for the material shear and Young's modulus is Ta element, while the Ti content is the opposite. In the same way, according to **Figure 3D**, higher V and Ti content can improve the ductility, but the effect of V content is more obvious than



the effect of Ti. The higher Ta content will sharply reduce the ductility of the alloys, which is proved by the fact in Reference (Jia et al., 2019), in which a lower lever Ti composition  $(Ta_{27}Ti_{33}V_{40})$  fractured before the elastic deformation stage.

However, the relationship between G, E, and B/G, and the effect of element content on the elastic constants of Ta-Ti-V MEAs are still need further investigations. For example, the study of more compounds with different atomic ratios, and the key factors that determine the mechanical properties of these compounds.

## CONCLUSION

The first-principles calculations in combination with the SQS method are applied to calculate structural parameters, stability, elastic constants, elastic modulus, and the influence of element content on elastic properties for BCC Ta-Ti-V MEAs. According to experimental results, we used the mcsqs code to search SQS and obtained a series of compositions as  $Ta_{24}Ti_{14}V_{16}$ ,  $Ta_{19}Ti_{17}V_{18}$ ,  $Ta_{16}Ti_{21}V_{17}$ ,  $Ta_{22}Ti_{23}V_9$ , and  $Ta_{18}Ti_{18}V_{18}$ . Due to the mixing energy calculation, all compositions were estimated with a thermal stability at high temperature, which implies the application potentials for the high-temperature environment. The predicted Ta-Ti-V MEAs are mechanically stable since their elastic constants satisfy the mechanical stability conditions, and  $Ta_{24}Ti_{14}V_{16}$  has the best bulk module and  $Ta_{19}Ti_{17}V_{18}$  has the

best ductility. Additionally, we calculated the elastic properties under the ratios of different elements' contents. The influence of the element content on the elastic properties of the system shows that the enhancement of the Ta content will increase the bulk modulus of the alloys, while a higher Ti content can decrease the bulk modulus and their material stiffness as well. The higher V and Ti content are beneficial to ductility, especially V content. However, the higher Ta content will sharply reduce the ductility of the alloys. Our calculation has provided further theoretical mechanism basis for the Ta-Ti-V MEAs preparation and development.

## DATA AVAILABILITY STATEMENT

The original contributions presented in the study are included in the article/supplementary material, further inquiries can be directed to the corresponding author.

## **AUTHOR CONTRIBUTIONS**

MP and WH: conception and design of the study. WH: acquisition of data and drafting the manuscript. MP, WH, ZH, HY, HW, LP, ZC, and YZ: analysis and interpretation of data. MP: revising the manuscript critically for important intellectual content. All authors contributed to the article and approved the submitted version.

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**Conflict of Interest:** The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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