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## EDITED BY

Jun Sun,  
China University of Geosciences Wuhan,  
China

## REVIEWED BY

Yoshihisa Mino,  
Nagoya University, Japan  
Sohiko Kameyama,  
Hokkaido University, Japan

## \*CORRESPONDENCE

Hong-Hai Zhang

✉ honghaizhang@ouc.edu.cn

Gui-Peng Yang

✉ gpyang@mail.ouc.edu.cn

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# High-resolution distribution and emission of dimethyl sulfide and its relationship with $p\text{CO}_2$ in the Northwest Pacific Ocean

Shi-Bo Yan<sup>1,2,3</sup>, Xiao-Jun Li<sup>3</sup>, Feng Xu<sup>3</sup>, Hong-Hai Zhang<sup>1,3\*</sup>,  
Jian Wang<sup>3</sup>, Yueqi Zhang<sup>4</sup>, Gui-Peng Yang<sup>1,2,3\*</sup>,  
Guang-Chao Zhuang<sup>1,2</sup> and Zhaohui Chen<sup>4</sup>

<sup>1</sup>Frontiers Science Center for Deep Ocean Multispheres and Earth System, Key Laboratory of Marine Chemistry Theory and Technology, Ministry of Education, Qingdao, China, <sup>2</sup>Laboratory of Marine Ecology and Environmental Science, Qingdao National Laboratory for Marine Science and Technology, Qingdao, China, <sup>3</sup>College of Chemistry and Chemical Engineering/Center for Ocean Carbon Neutrality, Ocean University of China, Qingdao, China, <sup>4</sup>Physical Oceanography Laboratory/Institute for advanced Ocean Study, Ocean University of China, Qingdao, China

Ocean-derived dimethyl sulfide (DMS) is widely concerning because of its hypothesized influence on global climate change. This study aims to explore the distribution characteristics and influencing factors of DMS and partial pressure of carbon dioxide ( $p\text{CO}_2$ ) in the Northwest Pacific Ocean, as well as the potential relationship between DMS and  $p\text{CO}_2$ . A high-resolution, underway, shipboard measurement device was used to determine the DMS and  $p\text{CO}_2$  of the surface seawater and atmosphere in the Northwest Pacific and its marginal seas during November 2019. The result show that atmospheric and surface seawater DMS concentrations ranged from 3 to 125 pptv and 0.63 to 2.28  $\text{nmol L}^{-1}$ , respectively, with mean values of  $46 \pm 19$  pptv and  $1.08 \pm 0.34$   $\text{nmol L}^{-1}$ . The average sea surface  $p\text{CO}_2$  was  $371 \pm 16$   $\mu\text{atm}$  (range from 332 to 401  $\mu\text{atm}$ ). The trends in the surface seawater DMS in different current systems were primarily associated with phytoplankton abundance and composition. Biological activity and physical processes such as cooling jointly influenced the sea surface  $p\text{CO}_2$ . A cold eddy along the transect in the Northwest Pacific Ocean increased DMS at the sea surface by 10% and  $\text{CO}_2$  uptake by 3%. We found a significant negative correlation between DMS and  $p\text{CO}_2$  in the Northwest Pacific Ocean at the  $0.1^\circ$  resolution  $[\text{DMS}]_{\text{seawater}} = -0.0161[p\text{CO}_2]_{\text{seawater}} + 7.046$  ( $R^2 = 0.569$ ,  $P < 0.01$ ). The DMS and  $p\text{CO}_2$  sea-air fluxes were estimated to range from 0.04 to 25.3  $\mu\text{mol m}^{-2}\cdot\text{d}^{-1}$  and from -27.0 to 4.22  $\text{mmol m}^{-2}\cdot\text{d}^{-1}$  throughout the survey area. The Northwest Pacific Ocean, especially the Oyashio Current, is an important sink of  $\text{CO}_2$  and a source of DMS.

## KEYWORDS

dimethyl sulfide,  $p\text{CO}_2$ , shipboard underway measurements, sea-air flux, Northwest Pacific Ocean

# 1 Introduction

Dimethyl sulfide (DMS) is the most important volatile biogenic sulfide in the ocean and its sea-air flux is estimated to be  $\sim 27.1 \text{ Tg}\cdot\text{S}\cdot\text{a}^{-1}$  (Hulswar et al., 2022). DMS in the ocean is mainly produced by certain marine algae and by bacteria through the cleavage of the precursor compound dimethylsulfoniopropionate (DMSP) (Stefels et al., 2007; Todd et al., 2007; Alcolombri et al., 2015). More than 95% of the DMS in the atmosphere is released from the ocean (Kettle and Andreae, 2000). Atmospheric DMS can be rapidly oxidized by free radicals (Glasow and Crutzen, 2004). The produced DMS oxidants enhance the production of atmospheric sulfate aerosols, which can act as cloud condensation nuclei (CCN) and regulate surface solar radiation, creating a negative feedback mechanism that counters the global greenhouse effect (Charlson et al., 1987; Gabric et al., 2004). Since the Industrial Revolution, human activities have rapidly increased the atmospheric  $\text{CO}_2$  concentration from 280ppm to 418ppm at present (Global Monitoring Laboratory - Carbon Cycle Greenhouse Gases (noaa.gov)), driving global warming and ocean acidification (Caldeira and Wickett, 2003; Allen et al., 2009; Landman, 2010). DMS concentrations in seawater are sensitive to ocean acidification (Mélangeon et al., 2016; Gao et al., 2021), and because the carbon and sulfur cycles are both heavily influenced by phytoplankton activity, it is likely that the two cycles are closely linked (Suess, 1980; Chisholm, 2000; Riebesell et al., 2007; Mélangeon et al., 2016; Dani and Loreto, 2017).

The Northwest Pacific Ocean is one of the most productive regions in the global ocean and an important global carbon sink (Honda, 2003; Takahashi et al., 2009; Friedlingstein et al., 2022). Under the influence of the strong western boundary current, the Northwest Pacific Ocean contains an abundance of dynamic marine processes, complex ecosystems, active biological processes, and strong exchanges of matter and energy at the air-sea interface (Sakurai, 2007; Hu et al., 2015; Karl and Church, 2017; Schlundt et al., 2017). The southward Oyashio Current of the subpolar gyre transports cold, low salinity, and nutrient-rich water from high latitudes to low latitudes where it, meets the warm, high salinity, and nutrient-deficient Kuroshio Current off the east coast of Japan, forming the Kuroshio-Oyashio Confluence Region (KOCR) (Mitsudera et al., 2004). Furthermore, the Northwest Pacific Ocean, especially the Kuroshio Extension Region, is one of the most active regions in terms of mesoscale eddies in the global ocean (Qiu and Chen, 2010; Xi et al., 2018). Mesoscale eddies are crucial to the vertical transport and horizontal distribution of nutrients (Vaillancourt et al., 2003) and can create quasi-enclosed mesoscale ecosystems with distinct features which have powerful influences on the distributions of dissolved active gases (Zindler et al., 2014).

Shipboard, underway, high-resolution observation methods for  $p\text{CO}_2$  have become reliable and well-established (Pierrot et al., 2009), while the underway analysis methods for DMS are relatively underdeveloped. Despite this, high-resolution, shipboard, underway DMS measurement methods have recently been applied (Asher et al., 2015; Zhang and Chen, 2015; Kim et al., 2017; Zhang et al., 2019). Kim et al. (2017) used a membrane inlet mass spectrometer (MIMS) to analyze the horizontal distribution of DMS in the Amundsen Sea Polynya region. Wohl et al. (2020); Wohl et al. (2022) used a

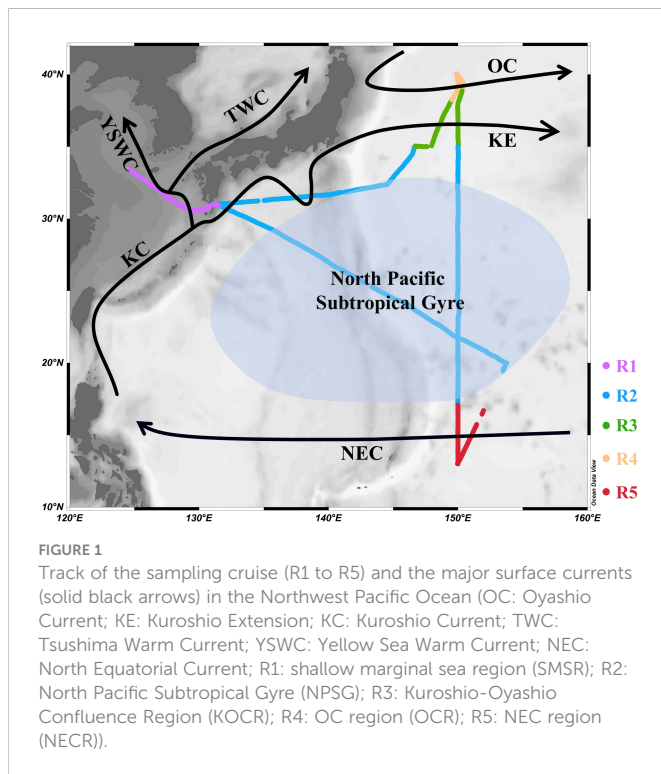
segmented flow coil equilibrator coupled to a proton transfer reaction mass spectrometer (EI-PTR-MS) to measure dissolved gases (e.g., DMS and isoprene) in the Southern Ocean and the Canadian Arctic. However, there were few studies on the DMS of seawater and atmosphere in the Northwest Pacific Ocean at present, and DMS data were relatively discrete (Figure S1) (DMS data of Northwest Pacific surface seawater was obtained from NOAA-PMEL DMS database, <https://saga.pmel.noaa.gov/dms/>). At present, only Zhang et al. (2017) reported the underway data of DMS in the East and South China Seas. In the absence of high-resolution field observation data, the distribution characteristics and control mechanism of DMS at the air-sea interface in the Northwest Pacific Ocean are still not well understood. This is not conducive to accurately estimating the sea-air flux of the Northwest Pacific DMS. This variation in the DMS flux will in turn be used to force models of atmospheric sulfur to derive sulfate aerosols (Chin and Jacob, 1996) and their radiative forcing (Meehl, 1996).

Considering the massive amount of  $\text{CO}_2$  data in the surface Ocean  $\text{CO}_2$  Atlas database (<https://www.socat.info/>) and the inadequacy of DMS data in the Northwest Pacific Ocean from the NOAA-PMEL DMS database, it would be of immense value to establish a reliable indicator for DMS using  $p\text{CO}_2$ . In this study, GC coupled with a custom-made purge and trap device (Zhang et al., 2019) was used to realize the underway observation of DMS. We performed high-resolution surface seawater DMS and  $p\text{CO}_2$  measurements in the Northwest Pacific Ocean in November 2019 to examine the factors driving the surface DMS and  $p\text{CO}_2$  distributions in different oceanic regions. The relationship between DMS and  $p\text{CO}_2$  in the Northwest Pacific Ocean was also examined in detail, providing a convenient method for predicting surface water DMS in the Northwest Pacific Ocean based on  $p\text{CO}_2$  data. We further calculated their sea-air fluxes to determine the source and sink patterns of DMS and  $\text{CO}_2$  in the surface ocean. It is also helpful to evaluate the impact of DMS release in the Northwest Pacific on global climate change.

## 2 Materials and methods

### 2.1 Description of the cruise

Continuous, high-resolution, underway measurements of DMS and  $p\text{CO}_2$  in the surface seawater and atmosphere of the Northwest Pacific Ocean were taken on board the R/V “*Dongfanghong 3*” from 1 November to 29 November 2019. The research vessel crossed the Osumi-kaiyō straight to reach the Northwest Pacific Ocean and then followed a transect along the longitude of approximately  $150^\circ\text{E}$  southwards to the southernmost point ( $13^\circ\text{N}$ ) of the cruise. The sampling transect crossed many dynamic oceanographic features, including the Kuroshio Current, Kuroshio Extension, Oyashio Current, North Pacific Subtropical Gyre, and North Equatorial Current (Figure 1). To further investigate the distributions and influencing factors of DMS and  $p\text{CO}_2$ , more thorough discrete sampling was performed every  $1^\circ$  along the  $150^\circ\text{E}$  transect. Fortunately, the ship passed through a cold eddy on November 3, which clearly had a great impact on the area and provided a convenient example of the influence of mesoscale eddies on active



gases. At the same time, the biochemical parameters such as temperature, salinity, dissolved oxygen (DO), and chlorophyll *a* (Chl-*a*) were measured in the surface waters.

## 2.2 Underway measurements

### 2.2.1 DMS

DMS was determined using a gas chromatograph-flame photometric detector (GC-FPD, 7890B, Agilent Technologies, USA) coupled with a custom-made purge and trap device (Zhang et al., 2019). The analysis process was automated for both seawater and atmospheric samples. Surface seawater samples were collected from the ship's seawater pump system at a depth of 5 m. The air sampling location was located at the vessel's bow approximately 10 m above the sea surface to avoid contamination from the ship's exhaust.

Surface seawater was continuously refreshed through a 5 ml sample loop. Seawater from the sample loop was pushed by flowing high-purity nitrogen ( $35 \text{ ml min}^{-1}$ ) into an extraction chamber and purged for 5 min. The purged seawater DMS samples were dehydrated using Nafion drying tubes (Perma Pure, USA) and captured in a trap tube (below  $-15^\circ\text{C}$ ) filled with Tenax-TA (80 mesh, Supelco, USA). At the same time, atmospheric DMS samples were dehydrated and captured in another trap tube for over 3.5 min at the sampling rate of  $100 \text{ ml min}^{-1}$ . The trap tubes were heated to over  $150^\circ\text{C}$  and the DMS was carried by  $\text{N}_2$  into GC-FPD for measurement. The analysis system was calibrated daily to examine the stability of the instrument during the cruise using a certified gas calibration standard (Dalian Special Gases Co., Ltd, China; nominal volume mixing ratio of 1 ppmv for DMS). The relative deviations of the calibration peak areas were typically within 6% of each other. To prevent biofouling, the water tubes and purge vessels were cleaned

daily. The large volume of nitrogen purging may cause oxygen depletion in the seawater sample and additional DMS emission in the purging bottle (Omori et al., 2013). Therefore, our results of DMS measurements were possibly slightly higher than those in seawater. However, we believe that these errors are acceptable in field observation given the description by Zhang and Chen (2015).

### 2.2.2 $p\text{CO}_2$

An infrared ray detector (LI-COR Model LI-7000, Lincoln, USA) based underway  $p\text{CO}_2$  measuring system (Model 8050, General Oceanics Inc., USA) was installed onboard for the underway automatic measurements of  $p\text{CO}_2$  in surface seawater and atmosphere (Pierrot et al., 2009). Calibration was performed using  $\text{CO}_2$  standard gases from the National Research Center for Certified Reference Materials, China (NRCCRM) with concentrations of 0, 200, 400, 600, and 1000 ppmv ( $\text{CO}_2/\text{N}_2$ ). The sampling frequency of the system was about 3 min and the system was calibrated every 2 ~ 2.5 h. The accuracy was estimated to be better than  $0.1 \mu\text{atm}$  for atmospheric  $p\text{CO}_2$  and  $2 \mu\text{atm}$  for seawater  $p\text{CO}_2$ . The  $p\text{CO}_2$  observations ( $p\text{CO}_2^{\text{obs}}$ ) were corrected for  $p\text{CO}_2$  in the equilibrator ( $p\text{CO}_2^{\text{eq}}$ ) as follows:

$$p\text{CO}_2^{\text{eq}} = p\text{CO}_2^{\text{obs}} \times (p^{\text{eq}} - p\text{H}_2\text{O})$$

where  $p^{\text{eq}}$  represents barometric pressure at equilibration at the measured temperature; and  $p\text{H}_2\text{O}$  represents the saturated water vapor pressure, calculated according to the equilibrium temperature and *in-situ* salinity (Weiss and Price, 1980), the calculation formula is as follows:

$$p\text{H}_2\text{O} = \exp \{24.4543 - 6745.09 / (T^{\text{eq}} + 273.15) - 4.8489 \times \ln [(T^{\text{eq}} + 273.15) / 100] - 0.000544 \times S\}$$

where  $T^{\text{eq}}$  is the temperature in the equilibrator; and  $S$  is the seawater salinity. Finally,  $p\text{CO}_2^{\text{eq}}$  was converted into *in-situ*  $p\text{CO}_2$  ( $p\text{CO}_2^{\text{in-situ}}$ ) by correcting for temperature using the equation from Takahashi et al. (1993):

$$p\text{CO}_2^{\text{in-situ}} = p\text{CO}_2^{\text{eq}} \times \exp [(SST - T^{\text{eq}}) \times 0.0423]$$

where, SST is sea surface temperature measured by the temperature sensor installed at the ship's surface seawater intake. This correction accounts for temperature changes in samples as they transit through the supply lines.

### 2.2.3 Other hydrographic parameters

Sea surface temperature (SST), sea surface salinity (SSS), Chl-*a*, and dissolved oxygen (DO) were monitored continuously with a sampling interval of 1 min by the shipboard underway surface multielement measurement system (Seabird Corporation, USA). Underway Chl-*a* and DO data were linearly validated using the field-measured Chl-*a* and DO in discrete seawater samples taken from the same seawater sampling source as for the DMS and  $p\text{CO}_2$  measurements. Meteorological data, including air temperature, pressure, and wind speed were measured from a meteorological observation instrument (approximately 20 m above sea level) located on the compass deck of the ship. All hydrographic and  $p\text{CO}_2$  data along the cruise track were averaged into 10 min bins and matched with DMS data according to time. Monthly mean sea

level anomaly and geostrophic currents were calculated based on daily satellite-observed sea surface heights data from Copernicus Marine Environment Monitoring Service (CMEMS) accessed at [https://resources.marine.copernicus.eu/product-detail/SEALEVEL\\_GLO\\_PHY\\_L4\\_MY\\_008\\_047/DATA-ACCESS](https://resources.marine.copernicus.eu/product-detail/SEALEVEL_GLO_PHY_L4_MY_008_047/DATA-ACCESS).

## 2.3 Discrete sampling and analysis

Discrete seawater samples were obtained using 12 L Niskin bottles mounted on a Seabird 911-plus conductivity-temperature-depth (CTD) rosette system (Seabird Corporation, USA), which simultaneously monitored the salinity and temperature of seawater. Samples for phytoplankton community analysis and Chl-*a* concentrations were collected, treated, and analyzed according to Zhang et al. (2022). Phytoplankton abundance is expressed as the number of phytoplankton cells per liter of seawater (cells L<sup>-1</sup>). DO samples were measured on board following the Winkler titration method (Hansen and Koroleff, 1999). DO saturation (DO% = [O<sub>2</sub>]<sub>obs</sub>/[O<sub>2</sub>]<sub>eq</sub> × 100%, where [O<sub>2</sub>]<sub>obs</sub> is *in-situ* DO concentration, and [O<sub>2</sub>]<sub>eq</sub> is the concentration at equilibrium with the atmosphere) and apparent oxygen utilization (AOU = [O<sub>2</sub>]<sub>eq</sub> - [O<sub>2</sub>]<sub>obs</sub>) were also calculated (Benson and Krause, 1984). Discrete dissolved inorganic carbon (DIC), *p*CO<sub>2</sub>, and Revelle factor (RF) were calculated from the measured pH<sub>T, 25</sub> and TA data, using CO2SYS v2.1 program (Pierrot et al., 2006), together with the *in-situ* temperature and salinity, and with the equilibrium constants of the carbonate acid K<sub>1</sub> and K<sub>2</sub> from Mehrbach et al. (1973) refit by Dickson and Millero (2007), the K<sub>H2SO4</sub> was from (Dickson, 1990), and the [B]<sub>T</sub> value from Uppström (1974), where pH<sub>T, 25</sub> and TA data were obtained from Mou et al. (2022).

## 2.4 Estimation of sea-air flux

The sea-air fluxes (*F*, μmol m<sup>-2</sup> d<sup>-1</sup>) of gases were generally determined using the bulk flux equation proposed by Liss and Merlivat (1986) as follows:

$$F = k \times (C_w - C_a)$$

where *k* (units: cm h<sup>-1</sup>) is the gas transfer velocity; and C<sub>w</sub> and C<sub>a</sub> are the gas concentrations in surface seawater and marine atmosphere, respectively.

For DMS sea-air flux, the equation proposed by Liss and Merlivat (1986) was used:

$$F_{DMS} = k_{DMS} \times ([DMS]_w - [DMS]_a/H)$$

where *k*<sub>DMS</sub> (units: cm h<sup>-1</sup>) is the gas transfer velocity of DMS; [DMS]<sub>w</sub> and [DMS]<sub>a</sub> are the surface seawater and marine atmosphere DMS concentrations, respectively, [DMS]<sub>w</sub> is usually several orders of magnitude higher than that [DMS]<sub>a</sub> and [DMS]<sub>a</sub>/H is often ignored; and H is the solubility of DMS in seawater, which varies with temperature and can be calculated using the following equation (Dacey et al., 1984):

$$\ln H \text{ (atm L mol}^{-1}\text{)} = -3547/T(K) + 12.64$$

The *k*<sub>DMS</sub> is parameterized using wind speed (*U*, m s<sup>-1</sup>) and the Schmidt number (*S<sub>c</sub>*), according to commonly used equation (Nightingale et al., 2000) as follows:

$$k_{DMS} = (0.222U_{10}^2 + 0.33U_{10}) \times (S_c/660)^{-1/2} \times [DMS]_w$$

where *S<sub>c</sub>* for DMS can be calculated from the SST (°C) according to Saltzman et al. (1993).

For estimating sea-air CO<sub>2</sub> flux (*F*<sub>CO<sub>2</sub></sub>, mmol m<sup>-2</sup> d<sup>-1</sup>), the equation is as follows:

$$F_{CO_2} = k_{CO_2} \times K_H \times \Delta pCO_2$$

where *k*<sub>CO<sub>2</sub></sub> (units: cm h<sup>-1</sup>) is the gas transfer velocity of CO<sub>2</sub>; *K<sub>H</sub>* (mol kg<sup>-1</sup> atm<sup>-1</sup>) is the solubility of CO<sub>2</sub> in seawater; and Δ*p*CO<sub>2</sub> is the air-sea *p*CO<sub>2</sub> difference. The *k*<sub>CO<sub>2</sub></sub> was calculated based on the Wanninkhof (2014) empirical function:

$$k_{CO_2} = 0.251U_{10}^2(S_c/660)^{-1/2}$$

where *S<sub>c</sub>* for CO<sub>2</sub> was based on Wanninkhof (2014).

The wind speeds were measured at 20 m above sea level, and corrected for speeds at 10 m (*U*<sub>10</sub>) based on the logarithmic wind profile established by Hsu et al. (1994) and calibrated according to the speed of the ship's passage:

$$U_x/U_{10} = (Z_x/Z_{10})^p$$

where *U<sub>x</sub>* is observed wind speed at 20 m; *Z<sub>x</sub>* and *Z*<sub>10</sub> are heights of 20 m and 10 m, respectively; and *p* is set to 0.11 (Hsu et al., 1994).

## 3 Results and discussion

### 3.1 Oceanographic characteristics

The circulation features of the dominating surface current systems and the horizontal distributions of SST and SSS in the Northwest Pacific Ocean are shown in Figures 1, 2A, B. SST ranged from 13.99 to 29.64 °C, which was a wide range that illustrated the dynamic nature of the study region. The data were divided into five oceanic regions (i.e., R1 to R5) according to the characteristics of the sea areas and water masses using S-T data and the criteria established by Hanawa and Mitsudera (1987) (Figure S2). The marginal sea region with low SST and SSS, which was mainly affected by the Yellow Sea warm current (Yanagi and Takahashi, 1993), was designated R1. The North Pacific Subtropical Gyre (NPSG), designated R2, was oceanographically characterized by high temperatures and high salinity and mainly influenced by Kuroshio Current. The lowest temperature occurred in the Oyashio Current Region (R4, OCR), where the temperature was 16.26 ± 1.75°C on average, while the North Equatorial Current Region (R5, NECR) experienced the highest temperatures of any region, with an average SST of 29.43 ± 0.16°C. Strong latitudinal gradients in SST and SSS were present through the Kuroshio-Oyashio Confluence Region (Figure 3A, R3, KOOCR), which may have been connected to the subarctic front formed by the convergence of the Kuroshio and Oyashio currents (Sugimoto et al., 2014).

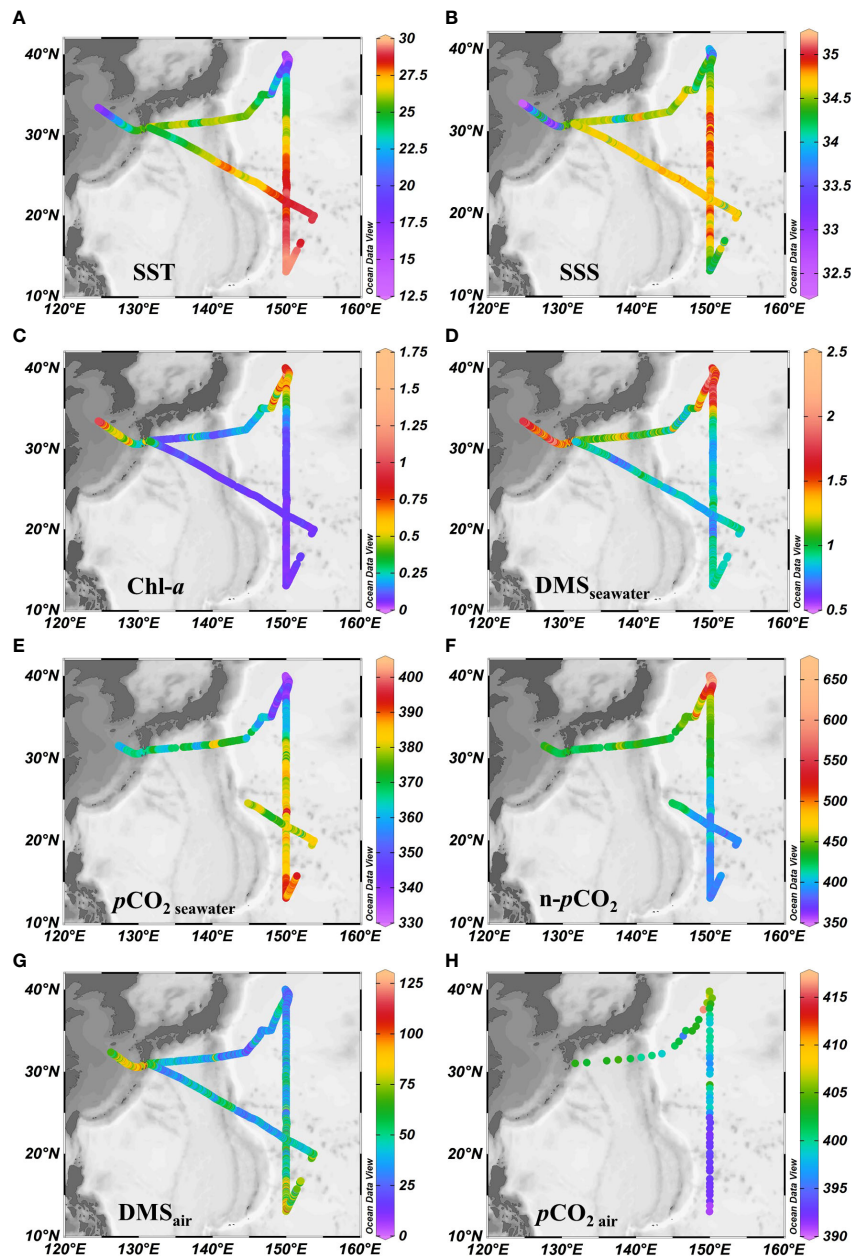


FIGURE 2

Horizontal distributions of underway (A) SST (°C), (B) SSS, (C) Chl-*a* (μg L<sup>-1</sup>), (D) DMS<sub>seawater</sub> (nmol L<sup>-1</sup>), (E) pCO<sub>2</sub> seawater (μatm), (F) n-pCO<sub>2</sub> (μatm), (G) DMS<sub>air</sub> (pptv), and (H) pCO<sub>2</sub> air (μatm). Note that n-pCO<sub>2</sub> was calculated using the formula  $n\text{-}p\text{CO}_2 = p\text{CO}_2 \times \exp[0.0423 \times (29.5 - \text{SST})]$  following Takahashi et al. (2002), where n-pCO<sub>2</sub> is the sea surface pCO<sub>2</sub> when the temperature is normalized to the mean temperature of NECR (29.5°C).

### 3.2 Spatial distribution of DMS in surface seawater and its relationship with phytoplankton

The concentrations of DMS (Figure 2D) followed similar trends to those of Chl-*a* (Figure 2C), they both showed obvious gradients along the ocean front (Figures 3C, F), roughly matching the discrete nutrient measurements which exhibited cliff-like changes (Figure 3B). The mean concentrations of DMS and Chl-*a* were  $1.08 \pm 0.34$  nmol L<sup>-1</sup> (range: 0.63–2.28 nmol L<sup>-1</sup>) and  $0.22 \pm 0.24$  μg L<sup>-1</sup> (range: 0.04–1.60 μg L<sup>-1</sup>), respectively. It is worth noting that the observations from our cruise track filled in the gaps of previous observations in the same area during the same period (Figure S1). The concentrations of Chl-*a*

and DMS in the five regions are summarized in Table 1. The average concentrations of Chl-*a* increased from  $0.10 \pm 0.06$  μg L<sup>-1</sup> and  $0.05 \pm 0.01$  μg L<sup>-1</sup> in the oligotrophic NPSG and NECR to  $0.44 \pm 0.21$ ,  $0.51 \pm 0.15$ , and  $0.86 \pm 0.24$  μg L<sup>-1</sup> in the nutrient rich marginal sea, KOCR, and OCR. Meanwhile, the concentrations of DMS in the marginal sea, KOCR, and OCR were about twice than those in the NPSG and NECR. The highest DMS values occurred alongside the highest Chl-*a* levels in the OCR. The abundant nutrients of the Oyashio Current promoted the growth of phytoplankton in this area, which promoted the production and release of DMS. The concentrations of DMS in the NPSG and NECR were relatively low and did not exhibit significant change, which was related to the limited nutrient availability that restricted phytoplankton growth (Yasunaka et al., 2014; Lin et al.,

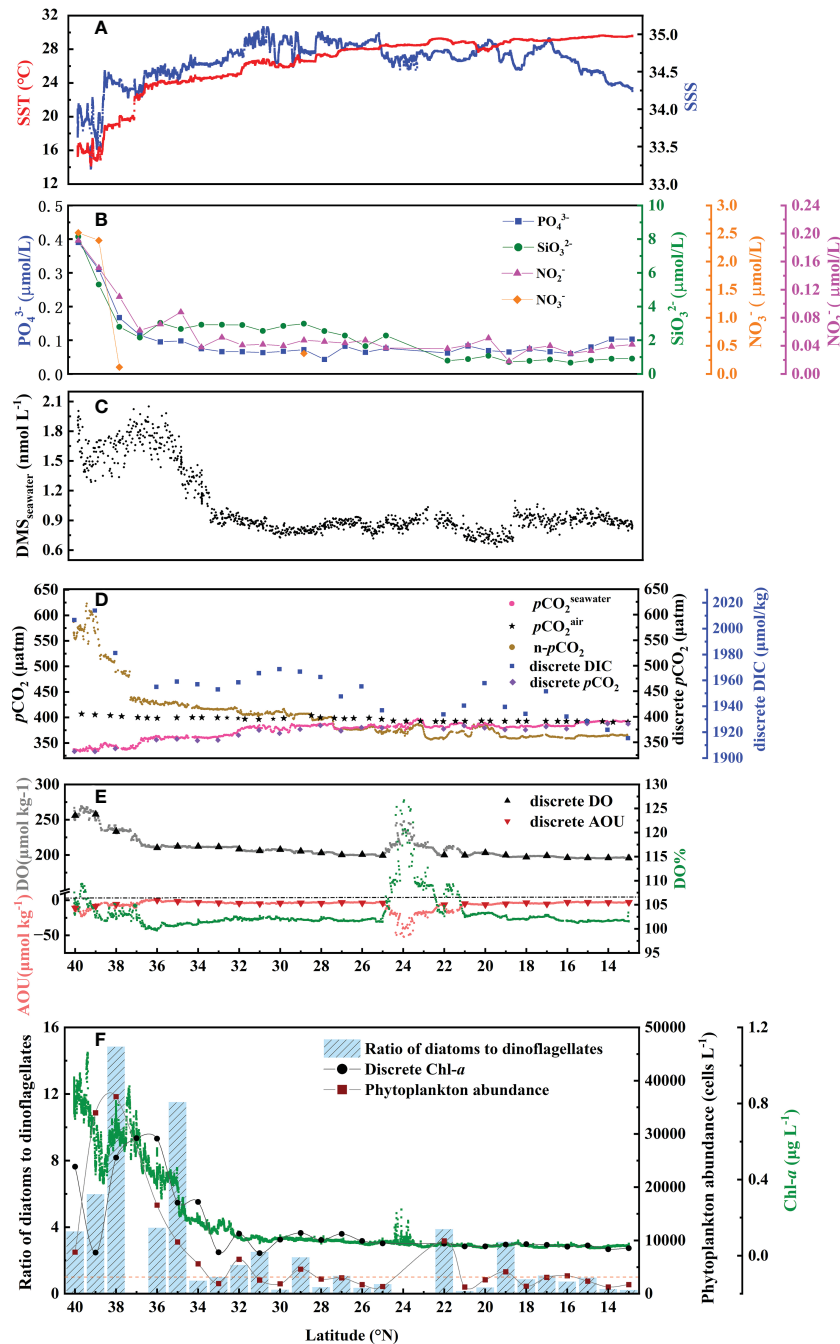


FIGURE 3

Latitudinal change along the 150°E transect in November 2019 in (A) underway SST and SSS, (B) discrete  $\text{PO}_4^{3-}$ ,  $\text{SiO}_3^{2-}$ ,  $\text{NO}_2^-$ , and  $\text{NO}_3^-$ , (C) underway  $\text{DMS}_{\text{seawater}}$ , (D) underway  $p\text{CO}_2^{\text{seawater}}$ ,  $p\text{CO}_2^{\text{air}}$ , and  $n\text{-}p\text{CO}_2$ ; discrete DIC and  $p\text{CO}_2$ , (E) underway DO, AOU and DO%; discrete DO and AOU, and (F) underway Chl-*a*, discrete Chl-*a*, ratio of diatoms to dinoflagellates, and phytoplankton abundance. Samples for discrete parameters were collected at each station.

2020; Wang et al., 2022). The Northwest Pacific Ocean is a significant contributor to the global DMS.

We explored the relationship between the DMS and Chl-*a* in the Northwest Pacific Ocean (Figure 4B) and showed that there was a strong correlation between DMS and Chl-*a* in the surface seawater of the Northwest Pacific Ocean (Figure 4B),  $[\text{DMS}]_{\text{seawater}} = 0.604[\text{Chl-}a]_{\text{seawater}} + 0.834$  ( $R^2 = 0.658$ ,  $P < 0.01$ ). The discrete phytoplankton abundance of the three stations (P1: 40° N, P2: 39° N, and P3: 38° N) in the OCR were very high, reaching 7800, 33950, and 37000 cells  $\text{L}^{-1}$ ,

respectively. However, the concentrations of DMS (P1: 1.91  $\text{nmol L}^{-1}$ , P2: 1.67  $\text{nmol L}^{-1}$ , and P3: 1.82  $\text{nmol L}^{-1}$ ) did not change according to the increase of phytoplankton abundance (Figures 3C, F). In the OCR region, the dominance of diatoms increased from north to south (Figure 3F), with the ratios of diatoms to dinoflagellates at P1, P2, and P3 stations being 3.7, 6.0, and 14.8, respectively. However, diatoms are generally considered to be inefficient producers of DMS, the dinoflagellates contributed much more to DMS production (Liss et al., 1994; Stefels et al., 2007). This is why the high phytoplankton

TABLE 1 SST, SSS, DO, Chl-*a*, DMS<sub>seawater</sub>, DMS<sub>air</sub>, pCO<sub>2</sub> seawater, and pCO<sub>2</sub> air levels in different regions.

	SST (°C)	SSS	Chl- <i>a</i> (μg L <sup>-1</sup> )	DMS <sub>seawater</sub> (nmol L <sup>-1</sup> )	DMS <sub>air</sub> (pptv)	pCO <sub>2</sub> seawater (μatm)	pCO <sub>2</sub> air (μatm)	ΔpCO <sub>2</sub> (μatm)	Wind speed (m s <sup>-1</sup> )	Flux <sub>DMS</sub> (μmol m <sup>-2</sup> d <sup>-1</sup> )	Flux <sub>CO<sub>2</sub></sub> (mmol m <sup>-2</sup> d <sup>-1</sup> )
marginal sea	23.01 ± 2.68	33.82 ± 0.66	0.44 ± 0.21	1.43 ± 0.30	71 ± 29	364 ± 3	-	-	10.6 ± 2.9	10.04 ± 4.87	-7.07 ± 2.59
NPSG	26.76 ± 1.63	34.67 ± 0.15	0.10 ± 0.06	0.91 ± 0.15	42 ± 3	376 ± 9	398 ± 4	-22 ± 12	8.9 ± 3.1	4.67 ± 2.89	-4.16 ± 3.54
KOCR	22.12 ± 2.06	34.36 ± 0.14	0.51 ± 0.15	1.52 ± 0.27	35 ± 7	350 ± 9	401 ± 5	-51 ± 13	4.6 ± 2.1	2.36 ± 2.02	-2.70 ± 2.31
OCR	16.25 ± 1.75	33.81 ± 0.42	0.86 ± 0.24	1.78 ± 0.21	30 ± 7	341 ± 6	406 ± 1	-65 ± 5	7.8 ± 3.6	6.38 ± 5.11	-9.59 ± 7.60
NECR	29.42 ± 0.16	34.94 ± 0.18	0.05 ± 0.01	0.90 ± 0.05	74 ± 30	389 ± 4	392 ± 1	-3 ± 4	11.9 ± 2.4	7.38 ± 3.21	-0.46 ± 0.71

abundance in P2 and P3 stations did not correspond to extremely high DMS values. These results indicated that, in addition to phytoplankton abundance, the composition of the phytoplankton community also affects the distribution of DMS in the Northwest Pacific Ocean. The difference in phytoplankton community only existed between the OCR and other regions, which had little effect on the relationship between DMS and Chl-*a* in the whole survey area.

A strong cold eddy, which was accompanied by low SST and SSS, was observed where the cruise track crossed the Kuroshio large meander (Figures 5A, B). This indicated that the cold eddy might be mainly composed of the coastal water mass in southern Japan. The maximum differences in SST and SSS between the center of the cold eddy and the surrounding sea area were 2.9 °C and 0.83, respectively.

Correspondingly, the DO increased from 205 ± 1 μmol Kg<sup>-1</sup> outside the cold eddy to 215 ± 1 μmol Kg<sup>-1</sup> inside the cold eddy. The seawater DMS concentrations increased slightly with the elevation of Chl-*a* along the path through the cold eddy (Figure 5B). In contrast to the Kuroshio waters, the Japanese coastal waters north of the Kuroshio have abundant nutrients that promote phytoplankton reproduction and DMS release. Meanwhile, surface seawater pCO<sub>2</sub> decreased from 370 ± 1 ppm to 360 ± 3 ppm (Figure 5B). Coastal waters generally have lower DIC concentrations than the Kuroshio, which may be the cause of the low pCO<sub>2</sub> (Ishii et al., 2001). Overall, the cold eddy increased DMS levels by 10% and the carbon sink intensity by 3%. These rapid changes in DMS levels occurred over a short time (under 0.5 h) and distance (about 9 km), and could not be observed by

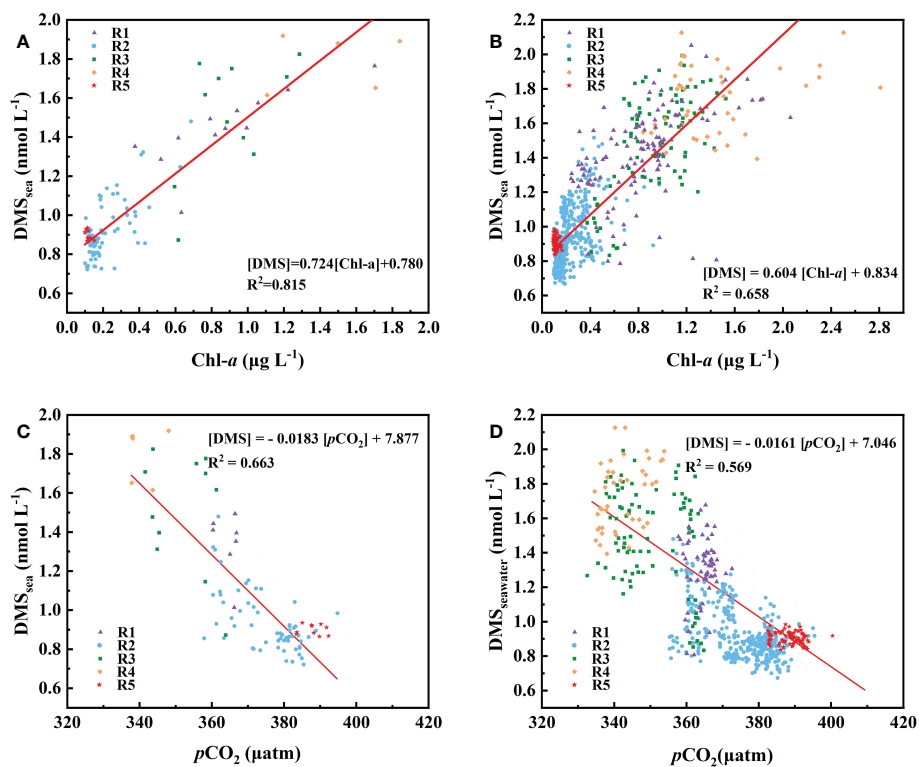


FIGURE 4

Relationship between Chl-*a* (A, B) and pCO<sub>2</sub> (C, D) with observed DMS concentrations. The data presented are means of (A, C) 1° and (B, D) 0.1° grids.

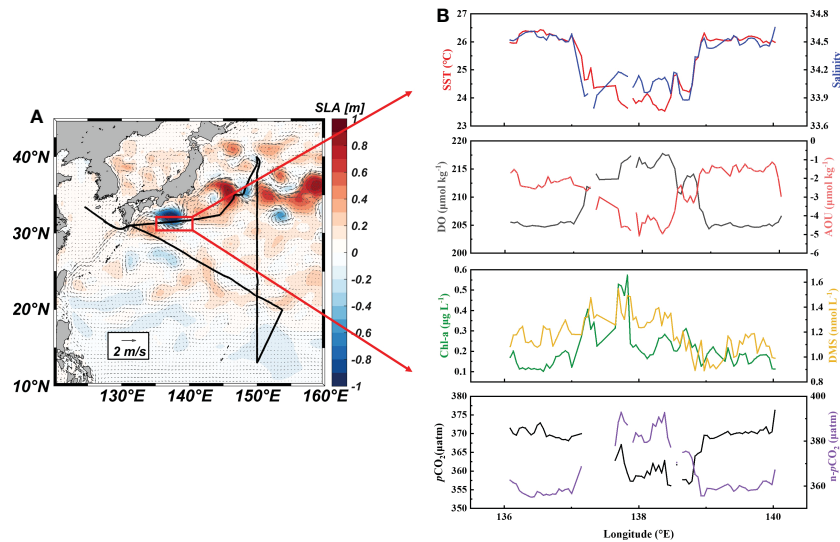


FIGURE 5 Monthly mean sea level anomaly and geostrophic currents in the Northwest Pacific Ocean (A) and underway parameters (SST, SSS, DO, AOU, Chl-a, DMS,  $p\text{CO}_2$  and  $n-p\text{CO}_2$ ) through the cold eddy (B).

traditional sampling methods. The high-resolution underway measurements enhance the synchronous changes of DMS with various physical and biological parameters on small spatial scale features. These observations showed that, when estimating the DMS and  $\text{CO}_2$  air-sea budgets in the Northwest Pacific, the influence of eddies should be taken into consideration.

### 3.3 Distribution characteristics and controlling factors of $p\text{CO}_2$

The surface seawater  $p\text{CO}_2$  in the study region ranged from undersaturated at  $332 \mu\text{atm}$  to highly supersaturated at  $401 \mu\text{atm}$  with an average of  $371 \pm 16 \mu\text{atm}$ , and showed a spatial distribution pattern similar to SST, with an abrupt drop occurring at  $37^\circ \text{N}$  at the Kuroshio front (Figure 2E). In addition, sea surface  $p\text{CO}_2$  ( $390 \pm 4 \mu\text{atm}$ ) was at or close to equilibrium with the atmosphere ( $392 \pm 1 \mu\text{atm}$ ) in the NECR. The sea surface  $p\text{CO}_2$  levels of the NPSG, KOCC, and OCR were  $376 \pm 9 \mu\text{atm}$ ,  $350 \pm 9 \mu\text{atm}$ , and  $341 \pm 6 \mu\text{atm}$ . Along the  $150^\circ \text{E}$  transect, the  $p\text{CO}_2$  difference between seawater and air ( $\Delta p\text{CO}_2^{\text{sea-air}}$ ) ranged from  $-3 \pm 12 \mu\text{atm}$ ,  $-22 \pm 12 \mu\text{atm}$ ,  $-51 \pm 13 \mu\text{atm}$ , and  $-65 \pm 5 \mu\text{atm}$  in the NECR, NPSG, KOCC, and OCR, respectively (Table 1). The surface seawater  $p\text{CO}_2$  was normalized to the mean temperature of the NECR ( $29.5^\circ \text{C}$ ), and the changes in normalized  $p\text{CO}_2$  ( $n-p\text{CO}_2$ ,  $n-p\text{CO}_2 = p\text{CO}_2 \times \exp [0.0423 \times (29.5 - \text{SST})]$ ) (Takahashi et al., 1993) was symmetric with the changes in *in-situ* surface  $p\text{CO}_2$  (Figures 2F, 3D) (Note that the average temperature in the NECR was selected as the normalized temperature, which was consistent with the NECR as the “starting area” when discussing the effects of  $p\text{CO}_2$  distribution below). The latitudinal variation characteristic of  $n-p\text{CO}_2$  was consistent with that of discrete DIC (Figure 3D), and the difference in  $n-p\text{CO}_2$  between NECR and OCR might be due to the DIC regional variation. In KOCC and OCR, low surface  $p\text{CO}_2$  corresponded to relatively high Chl-*a* and oversaturated DO (Figures 2C, E, 3D and E), indicating that biological activities

(photosynthesis) partly induced DO addition and depleted sea surface  $\text{CO}_2$  (reduce  $p\text{CO}_2$ ).

The SST of the survey area had a  $15^\circ \text{C}$  change, which may have influenced the spatial distribution of  $p\text{CO}_2$ . Therefore, to further reveal the controlling factors of surface seawater  $p\text{CO}_2$  distribution in the study area, we plotted the relationship between surface  $p\text{CO}_2$  and SST using the equation from Takahashi et al. (1993):  $p\text{CO}_2 = 390 \times \exp [0.0423 \times (\text{SST} - 29.5)]$ , where  $390 \mu\text{atm}$  is the mean surface  $p\text{CO}_2$  in the “starting zone”. In general, the latitudinal variation in surface  $p\text{CO}_2$  was roughly controlled by SST change, especially in the NECR where the surface  $p\text{CO}_2$  had a strong relationship with temperature (Figure S3A). In the NPSG, KOCC, and OCR, the surface  $p\text{CO}_2$  values were higher than the values forecasted based on SST (Figure S3A), demonstrating that non-thermal processes were elevating the surface  $p\text{CO}_2$ . In other words, DIC concentrations were higher in those areas compared to NECR. In addition, the surface  $p\text{CO}_2$  in NPSG, KOCC, and OCR was lower than the atmospheric equilibrium  $p\text{CO}_2$ , the process of sea-air exchange also increased their surface  $p\text{CO}_2$  (Figure S3B).

To quantify the regulators of latitudinal variation in  $\Delta p\text{CO}_2^{\text{sea-air}}$ , from the “starting area” to the OCR, the decomposition of  $\Delta p\text{CO}_2^{\text{sea-air}}$  can be expressed as follows according to Li et al. (2022):

$$\Delta p\text{CO}_2^{\text{sea-air}} = (p\text{CO}_2^{\text{sea}0} + \delta p\text{CO}_2^{\text{cool}} + \delta p\text{CO}_2^{\text{bio}} + \delta p\text{CO}_2^{\text{residual}}) - (p\text{CO}_2^{\text{air}0} + \delta p\text{CO}_2^{\text{air}})$$

where  $p\text{CO}_2^{\text{sea}0}$  and  $p\text{CO}_2^{\text{air}0}$  were both  $390 \mu\text{atm}$  in the starting zone, respectively. The  $\delta p\text{CO}_2^{\text{cool}}$ ,  $\delta p\text{CO}_2^{\text{bio}}$ , and  $\delta p\text{CO}_2^{\text{air}}$  represent the contributions of cooling, biological activities, and atmospheric  $p\text{CO}_2$  changes to  $\Delta p\text{CO}_2^{\text{sea-air}}$ , respectively, and  $\delta p\text{CO}_2^{\text{residual}}$  primarily indicates contributions of vertical mixing and air-sea exchange on  $\Delta p\text{CO}_2^{\text{sea-air}}$ . Note that the effect of air-sea exchange on  $p\text{CO}_2$  in the Northwest Pacific is weak (Ishii et al., 2001; Li et al., 2022), and  $\delta p\text{CO}_2^{\text{residual}}$  can be understood to be mainly contributed by vertical mixing which brings  $\text{CO}_2$  (DIC)-rich waters from depth to the surface



layer. The calculation method of each process contribution is shown in the supporting material. The results showed that  $\delta p\text{CO}_2^{\text{ool}}$  changed from  $0 \pm 2 \mu\text{atm}$  in NECR to  $-32 \pm 22 \mu\text{atm}$  in NPSG to  $-97 \pm 25 \mu\text{atm}$  in KO CR, and  $-173 \pm 3 \mu\text{atm}$  in OCR, while the changes of  $\delta p\text{CO}_2^{\text{bio}}$  and  $\delta p\text{CO}_2^{\text{air}}$  were not obvious (Figure S4 and Table S2). However,  $\delta p\text{CO}_2^{\text{residual}}$  values were  $0 \pm 2 \mu\text{atm}$  in NECR,  $23 \pm 17 \mu\text{atm}$  in NPSG,  $62 \pm 18 \mu\text{atm}$  in KO CR, and  $132 \pm 2 \mu\text{atm}$  in OCR, respectively. In summary, physical processes such as cooling and vertical mixing mostly influenced the latitudinal changes in  $\Delta p\text{CO}_2^{\text{sea-air}}$ . In the Northwest Pacific, the OCR and KO CR acted as significant carbon sinks.

### 3.4 Relationship between surface DMS and $p\text{CO}_2$

Although previous simulation of surface seawater DMS presented relatively high  $R^2$  (0.68 to 0.84) by using the mixed layer depth and Chl-*a* (Simó and Dachs, 2002), there are few practical data to construct this model, especially the data from the Northwest Pacific Ocean. In addition, the relative uncertainty in quantifying the annual mean concentrations of surface seawater DMS in temperate regions was up to 50% (Belviso et al., 2004). The empirical equation of sea surface DMS in the North Pacific Ocean was constructed using SST, sea surface nitrate (SSN), and latitude (Watanabe et al., 2007). However, due to the lack of observation time series of DMS and other hydrological parameters (especially SSN), the algorithm may be difficult to be applied to future DMS prediction. A comparison of the time series for DMS and  $p\text{CO}_2$  showed that they tended to exhibit opposing trends (Figures 2D, E). Tortell et al. (2012) and Zhang et al. (2017) compared DMS concentrations and  $p\text{CO}_2$  in surface seawater and observed a large range of negative correlations ( $r$  value ranged from -0.03 to -0.73) in the Southern Ocean, but they were never tested in the Northwest Pacific Ocean. The  $n\text{-}p\text{CO}_2$  could be used as an indicator of net community production (NCP) when the effects of vertical mixing and air-sea gas exchange were negligible (Kameyama et al., 2014), and previous studies also found a good correlation between DMS and NCP (Kameyama et al., 2013). However, our results indicate that vertical mixing was an important factor affecting  $p\text{CO}_2$  distribution in the Northwest Pacific Ocean (Section 3.3). The NCP calculation process limited the application of this method in a wide range, especially in areas with deep water upwelling, such as OCR.

We explored the relationship between the DMS and  $p\text{CO}_2$  in the Northwest Pacific Ocean, and plotted DMS concentrations against  $p\text{CO}_2$  for the  $1^\circ$ -grid and  $0.1^\circ$ -grid mean data sets (Figure 4). There was a significant negative correlation between DMS and  $p\text{CO}_2$  for both the  $1^\circ$ -grid data set ( $[\text{DMS}]_{\text{seawater}} = -0.0183 \times + 7.877$ ,  $R^2 = 0.663$ ,  $P < 0.01$ , Figure 4C) and  $0.1^\circ$ -grid data set ( $[\text{DMS}]_{\text{seawater}} = -0.0161 \times p\text{CO}_2^{\text{seawater}} + 7.0$ ,  $R^2 = 0.569$ ,  $P < 0.01$ , Figure 4D). The consistency of the slopes and intercepts between data sets at different spatial resolutions suggested that the influences of controlling factors such as SST, sea-air exchange, and biological activity were consistent over different spatial scales. In the surface seawater of the Northwest Pacific Ocean, the DMS concentration and SST exhibited a significant negative correlation ( $r = -0.744$ ,  $p < 0.01$ , Table S1), while there was a significant positive correlation between  $p\text{CO}_2$  and SST ( $r = 0.847$ ,  $p < 0.01$ , Table S1). SST

inversely affected the surface ocean DMS and  $p\text{CO}_2$  values. The surface seawater of the Northwest Pacific Ocean released DMS to the atmosphere and absorbed  $\text{CO}_2$  through air-sea exchange, which contributed to the negative correlation between DMS and  $p\text{CO}_2$ . High phytoplankton abundance and strong biological activity can enhance the uptake of  $\text{CO}_2$  from seawater while simultaneously synthesizing more DMS precursor. A significant correlation was found between DMS and Chl-*a* in the surface seawater of the Northwest Pacific Ocean (Figures 4A, B). Although the variation of its slope (16.5%) with sampling frequency is slightly greater than that between DMS and  $p\text{CO}_2$  (12%), the correlation between DMS and Chl-*a* ( $R^2 = 0.658$ , Figure 4B) is slightly higher than that between DMS and  $p\text{CO}_2$  ( $R^2 = 0.569$ , Figure 4C). In the Northwest Pacific Ocean, Chl-*a*, as an indicator of phytoplankton standing stock, seems to have a more directly effect on DMS concentration than  $p\text{CO}_2$ . Hence, the DMS-vs-Chl-*a* correlation appears to be useful for estimating the distribution of DMS in the Northwest Pacific Ocean, given the easy availability of remote sensing Chl-*a* data. On the other hand, direct measurements of  $p\text{CO}_2$  are relatively abundant (usually below  $0.1^\circ$ ) in the SOCAT database and the data were strictly quality controlled, which were better than the Chl-*a* data obtained from the satellite remote sensing. In addition, our algorithm can explain 56.9% of surface seawater DMS variance in the Northwest Pacific Ocean. Although the use of data with high temporal resolution in the algorithm leads to partial dispersion and underestimation (less than 20%) of the predicted values (Figure S5A), the standard deviation (SD) of the observed DMS value of 0.34 is 1.8 times that of the algorithm the root mean square error (RMSE=0.19), suggesting that the algorithm is also satisfactory in predicting DMS. (Ritter and Muñoz-Carpena, 2013). Note that the RMSE is used to quantify the prediction error of the variable unit calculated by the model, and its definition is  $RMSE = \sqrt{\frac{\sum_{i=1}^N (O_i - P_i)^2}{N}}$ , where  $O_i$  and  $P_i$  represent the sample (of size  $N$ ) containing the observations and the model estimates, respectively. Therefore,  $p\text{CO}_2$  has great potential as an indicator for the distribution of surface seawater DMS in the Northwest Pacific Ocean and may be simpler than previously used DMS parameterization methods. This algorithm may not be suitable for reconstructing more productive spring and summer surface water DMS in the Northwest Pacific ( $R^2 = 0.14$ , May-June 2021, unpublished). In spring and summer, phytoplankton blooms and changes in phytoplankton community structure led to latitude decoupling of DMS, Chl-*a* and  $p\text{CO}_2$ . However, in OCR and KO CR with the highest biological activity, the relationship between DMS and  $p\text{CO}_2$  will still be applicable due to the shallower mixing layer in summer, the reduced vertical mixing effect, and the greater influence of thermodynamic effects and biological activities on  $p\text{CO}_2$ .

### 3.5 Estimation of DMS and $\text{CO}_2$ sea-air fluxes in the Northwest Pacific Ocean

The mean atmospheric DMS mixing ratios (range) were  $46 \pm 19$  (3- 125) pptv throughout the survey area (Figure 2G). The KO CR region had the lowest mean wind speed (mean:  $5 \pm 2 \text{ m s}^{-1}$ ), which was accompanied by a relatively low DMS mixing ratio, despite the DMS concentration being relatively high in this region (Table 1). The concentrations of sea surface DMS did not correlate with atmospheric DMS mixing ratios ( $r = -0.017$ ,  $P > 0.05$ ) (Table S1), likely because, the

transport of air mass was much faster than that of seawater, leading to a decoupling between atmospheric and seawater DMS concentrations (Aranami and Tsunogai, 2004; Wohl et al., 2020). The surface seawater DMS concentrations were roughly uniform and generally low in the NECR, but the atmospheric DMS mixing ratios were high (mean:  $74 \pm 30$  pptv), due to the high DMS sea-air flux caused by the high wind speeds (mean:  $\sim 12$  m/s).

The continuous underway DMS sea-air flux,  $\text{CO}_2$  sea-air flux, and wind speed measurements during the cruise are shown in Figure 6. The DMS sea-air flux varied over a wide range, from 0.04 to  $25.3 \mu\text{mol m}^{-2} \text{d}^{-1}$ , with an average flux of  $5.37 \pm 3.86 \mu\text{mol m}^{-2} \text{d}^{-1}$ . In addition, based on the DMS concentration in the Northwest Pacific Ocean derived from the algorithm in Section 3.4, the sea-air flux of DMS was predicted ( $\text{Flux}_{\text{pre}}$ ). The values of  $\text{Flux}_{\text{pre}}$  correspond well to the Flux values (Figure S5B), and the SD (3.86) of Flux is 3.6 times that of RMES (1.07) of this algorithm, that is, the algorithm provides a very good prediction of DMS air-sea flux in the Northwest Pacific (Ritter and Muñoz-Carpena, 2013). The DMS sea-air flux was high in the OCR region, which might have been because this area had the highest DMS levels (mean:  $1.78 \pm 0.21 \text{ nmol L}^{-1}$ ) and relatively high wind speeds (mean:  $7.8 \pm 3.6 \text{ m s}^{-1}$ ). The marginal sea region accounted for only 0.21% of the global ocean area (based on the area of the East China Sea), but contributed to 0.64% of global annual DMS emissions due to the highest sea-air flux of DMS ( $10.0 \pm 4.87 \mu\text{mol m}^{-2} \text{d}^{-1}$ ), making it an important source of global DMS emissions. Although the mean DMS sea-air flux in the open waters of the Northwest Pacific Ocean was only half that in the marginal sea, it remained 5.2% of global DMS emissions given its vast area (3.2% of the global area). Although seasonal variations were not taken into account in the calculation of DMS sea-air fluxes based on one-shot survey data, the season of this observation was at the end of autumn, when biological activity and DMS concentrations were at lower all the year round, which may underestimate the contribution to global DMS emissions. The survey area was more productive and released more DMS in spring and summer, we speculated that the sea-air flux of DMS in spring and summer in the Northwest Pacific Ocean would be higher.

The atmospheric  $p\text{CO}_2$  in the Northwest Pacific Ocean ranged from 391 to  $416 \mu\text{atm}$  and showed a decreasing trend from high latitudes to low latitudes (Figures 2H, 3D), and the mean  $\text{CO}_2$  sea-air flux was  $-4.20 \pm 4.21 \text{ mmol m}^{-2} \text{d}^{-1}$  (range:  $-27.0$  to  $4.22 \text{ mmol m}^{-2} \text{d}^{-1}$ ). In the NECR, atmospheric  $p\text{CO}_2$  ( $392 \pm 1 \mu\text{atm}$ ) was close to the

seawater  $p\text{CO}_2$  ( $389 \pm 4 \mu\text{atm}$ ). The OCR had a much higher flux (mean:  $-9.59 \pm 7.60 \text{ mmol m}^{-2} \text{d}^{-1}$ ) compared to the NPSG (mean:  $-4.16 \pm 3.54 \text{ mmol m}^{-2} \text{d}^{-1}$ ) and NECR (mean:  $-0.46 \pm 0.71 \text{ mmol m}^{-2} \text{d}^{-1}$ ), because the OCR had both higher wind speed (mean:  $7.8 \pm 3.6 \text{ m s}^{-1}$ ) and a much larger  $\Delta p\text{CO}_2$  (mean:  $-65 \pm 5 \mu\text{atm}$ ). Although  $\Delta p\text{CO}_2$  was large in the KOCR (mean:  $-51 \pm 13 \mu\text{atm}$ , Table 1), the  $p\text{CO}_2$  air-sea flux in this area was only  $-2.70 \pm 2.31 \text{ mmol m}^{-2} \text{d}^{-1}$  because it had the lowest wind speed (mean:  $4.6 \text{ m s}^{-1}$ ). The carbon sink intensity was weak in the low latitude region (NECR), which was even a carbon source in some areas (Figure 6). However, the Northwest Pacific Ocean remains one of the most important carbon sinks in the world, especially in the Oyashio Current Region.

## 4 Summary

Continuous underway measurements of DMS and  $p\text{CO}_2$  in the surface water and air of the Northwest Pacific Ocean were taken between 1 November to 29 November 2019. There was much larger variability in the DMS and  $p\text{CO}_2$  in the surface seawater than in the atmosphere. The highest surface seawater DMS concentration occurred in the region co-influence by the Kuroshio-Oyashio Currents, which had the highest Chl-*a* levels and the lowest  $p\text{CO}_2$ . The trends in DMS of surface seawater in the Northwest Pacific Ocean under different current systems were primarily associated with phytoplankton abundance and community composition. The distribution of surface seawater  $p\text{CO}_2$  in the Northwest Pacific Ocean was influenced by both temperature and biological activity. The observed cold eddy in the Northwest Pacific Ocean promoted the growth of phytoplankton, resulting in the elevation of DMS and reduction of  $p\text{CO}_2$  in the surface layer. By comparing biophysical and chemical parameters, we found a significant negative relationship between the distributions of DMS and  $p\text{CO}_2$  in surface seawater, which may be helpful in reconstructing the distributions of DMS in surface seawater of the Northwest Pacific Ocean and the sea-air flux of DMS in the Northwest Pacific Ocean. The Northwest Pacific Ocean, especially the OCR, is an important source of DMS and an important sink of  $\text{CO}_2$ .

## Data availability statement

The original contributions presented in the study are included in the article/Supplementary Material. Further inquiries can be directed to the corresponding authors.

## Author contributions

H-HZ and G-PY designed the study. S-BY performed the experiments, with assistance from X-JL, FX and JW. X-JL provided  $p\text{CO}_2$  data. S-BY organized and analyzed the database, wrote the manuscript and prepared the tables and figures. H-HZ, FX, G-PY, G-CZ, YZ, and ZC provided comments on data analysis and revised the manuscript. All authors contributed to the article and approved the submitted version.

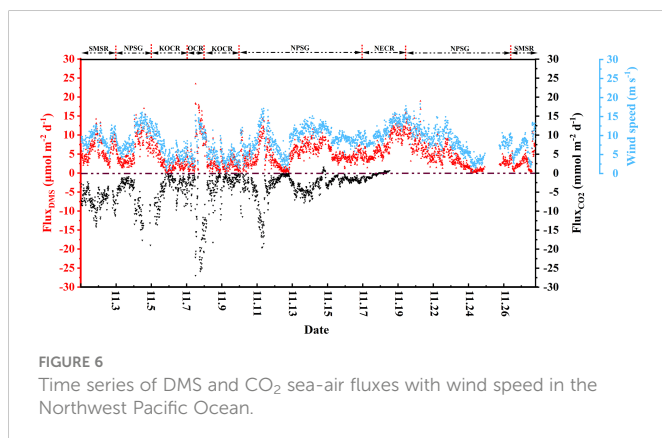


FIGURE 6  
Time series of DMS and  $\text{CO}_2$  sea-air fluxes with wind speed in the Northwest Pacific Ocean.

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## References

- Arcolombi, U., Ben-Dor, S., Feldmesser, E., Levin, Y., Tawfik, D. S., and Vardi, A. (2015). Identification of the algal dimethyl sulfide-releasing enzyme: A missing link in the marine sulfur cycle. *Science* 348, 1466–1469. doi: 10.1126/science.aab1586
- Allen, M. R., Frame, D. J., Huntingford, C., Jones, C. D., Lowe, J. A., Meinshausen, M., et al. (2009). Warming caused by cumulative carbon emissions towards the trillionth tonne. *Nature* 458, 1163–1166. doi: 10.1038/nature08019
- Aranami, K., and Tsunogai, S. (2004). Seasonal and regional comparison of oceanic and atmospheric dimethylsulfide in the northern north pacific: Dilution effects on its concentration during winter. *J. Geophys. Res. D: Atmos.* 109, 1–15. doi: 10.1029/2003JD004288
- Asher, E. C., Dacey, J. W. H., Jarniková, T., and Tortell, P. D. (2015). Measurement of DMS, DMSO, and DMSP in natural waters by automated sequential chemical analysis. *Limnol. Oceanography: Methods* 13, 451–462. doi: 10.1002/lom3.10039
- Belviso, S., Bopp, L., Moulin, C., Orr, J. C., Anderson, T. R., Aumont, O., et al. (2004). Comparison of global climatological maps of sea surface dimethyl sulfide. *Global Biogeochem. Cycles* 18, GB3013. doi: 10.1029/2003GB002193
- Benson, B. B., and Krause, D. (1984). The concentration and isotopic fractionation of oxygen dissolved in freshwater and seawater in equilibrium with the atmosphere 1. *Limnol. Oceanography* 29, 620–632. doi: 10.4319/lo.1984.29.3.0620
- Caldeira, K., and Wickett, M. E. (2003). Oceanography: Anthropogenic carbon and ocean pH. *Nature* 425, 365. doi: 10.1038/425365a
- Charlson, R. J., Lovelock, J. E., Andreae, M. O., and Warren, S. G. (1987). Ocean phytoplankton, atmospheric sulfur, cloud albedo and climate. *Nature* 326, 655–661. doi: 10.1038/326655a0
- Chin, M., and Jacob, D. J. (1996). Anthropogenic and natural contributions to tropospheric sulfate: A global model analysis primarily by dry deposition wet is relatively more anthropogenic influence on  $\text{SO}_4^{2-}$  decreases rapidly with altitude because of efficient scavenging of  $\text{SO}$ : and in de. *J. Geophys. Res.* 101, 18691–18699. doi: 10.1029/96JD01222
- Chisholm, S. W. (2000). Stirring times in the southern ocean. *Nature* 407, 685–687. doi: 10.1038/35037696
- Dacey, J. W. H., Wakeham, S. G., and Howes, B. L. (1984). Henry's law constants for dimethylsulfide in freshwater and seawater. *Geophys. Res. Lett.* 11, 991–994. doi: 10.1029/GL011i010p000991
- Dani, K. G. S., and Loreto, F. (2017). Trade-off between dimethyl sulfide and isoprene emissions from marine phytoplankton. *Trends Plant Sci.* 22, 361–372. doi: 10.1016/j.tplants.2017.01.006
- Dickson, A. G. (1990). Standard potential of the reaction:  $\text{AgCl}(s) + 1/2\text{H}_2(g) = \text{ag}(s) + \text{HCl}(aq)$ , and the standard acidity constant of the ion  $\text{HSO}_4^-$  in synthetic sea water from 273.15 to 318.15 K. *J. Chem. Thermodyn.* 22 (2), 113–127. doi: 10.1007/978-1-61779-117-8\_26
- Friedlingstein, P., Jones, M. W., Sullivan, M. O., Andrew, R. M., Bakker, D. C. E., Hauck, J., et al. (2022). Global carbon budget 2021. *Earth system science data.* 14, 1917–2005. doi: 10.48350/168229
- Gabric, A. J., Simó, R., Cropp, R. A., Hirst, A. C., and Dachs, J. (2004). Modeling estimates of the global emission of dimethylsulfide under enhanced greenhouse conditions. *Global Biogeochem. Cycles* 18, GB2014. doi: 10.1029/2003GB002183
- Gao, X. X., Zhang, H. H., Mao, S. H., and Yang, G. P. (2021). Responses of biogenic sulfur compound concentrations to dust aerosol enrichment and ocean acidification in the Western pacific ocean. *Geophys. Res. Lett.* 48, 1–10. doi: 10.1029/2021GL095527
- Glasow, R. V., and Crutzen, P. J. (2004). Model study of multiphase DMS oxidation with a focus on halogens. *Atmos Chem. Phys.* 4, 589–608. doi: 10.5194/acp-4-589-2004
- Hanawa, K., and Mitsudera, H. (1987). Variation of water system distribution in the sanriku coastal area. *J. Oceanography* 42, 435–446. doi: 10.1007/BF02110194
- Hansen, H. P., and Koroleff, F. (1999). “Determination of nutrients,” in *Methods of seawater analysis*. Eds. K. Grasshoff, K. Kremling and M. Ehrhardt (Hoboken: Wiley), 159–228. doi: 10.1002/9783527613984.ch10
- Honda, M. C. (2003). Biological pump in northwestern north pacific. *J. Oceanography* 59, 671–684. doi: 10.1023/B:JOCE.0000009596.57705.0c
- Hsu, S. A., Meindl, E. A., and Gilhousen, D. B. (1994). Determining the power-law wind-profile exponent under near-neutral stability conditions at Sea. *J. Appl. Meteorol. Climatol.* 33, 757–765. doi: 10.1175/1520-0450(1994)033<0757:DTPLWP.2.0.CO;2
- Hulswar, S., Simó, R., Galí, M., Bell, T. G., Lana, A., Inamdar, S., et al. (2022). Third revision of the global surface seawater dimethyl sulfide climatology (DMS-Rev3). *Earth Syst. Sci. Data* 14, 2963–2987. doi: 10.5194/essd-14-2963-2022
- Hu, D., Wu, L., Cai, W., Gupta, A. S., Ganachaud, A., Qiu, B., et al. (2015). Pacific western boundary currents and their roles in climate. *Nature* 522, 299–308. doi: 10.1038/nature14504
- Ishii, M., Inoue, H. Y., Matsueda, H., Saito, S., Fushimi, K., Nemoto, K., et al. (2001). Seasonal variation in total inorganic carbon and its controlling processes in surface waters of the western north pacific subtropical gyre. *Mar. Chem.* 75, 17–32. doi: 10.1016/S0304-4203(01)00023-8
- Kameyama, S., Tanimoto, H., Inomata, S., Yoshikawa-Inoue, H., Tsunogai, U., Tsuda, A., et al. (2013). Strong relationship between dimethyl sulfide and net community production in the western subarctic pacific. *Geophys. Res. Lett.* 40, 3986–3990. doi: 10.1002/grl.50654
- Kameyama, S., Yoshida, S., Tanimoto, H., Inomata, S., and Suzuki, K. Yoshikawa-Inoue H. (2014). High-resolution observations of dissolved isoprene in surface seawater in the Southern Ocean during austral summer 2010–2011. *J. Oceanogr.* 70, 225–239. doi: 10.1007/s10872-014-0226-8

## Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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## Supplementary material

The Supplementary Material for this article can be found online at: <https://www.frontiersin.org/articles/10.3389/fmars.2023.1074474/full#supplementary-material>

- Karl, D. M., and Church, M. J. (2017). Ecosystem structure and dynamics in the north pacific subtropical gyre: New views of an old ocean. *Ecosystems* 20, 433–457. doi: 10.1007/s10021-017-0117-0
- Kettle, A. J., and Andreae, M. O. (2000). Flux of dimethylsulfide from the oceans: A comparison of updated data sets and flux models. *J. Geophys. Res. Atmos.* 105, 26793–26808. doi: 10.1029/2000JD900252
- Kim, I., Hahn, D., Park, K., Lee, Y., Choi, J. O., Zhang, M., et al. (2017). Characteristics of the horizontal and vertical distributions of dimethyl sulfide throughout the amundsen Sea polynya. *Sci. Total Environ.* 584, 154–163. doi: 10.1016/j.scitotenv.2017.01.165
- Landman, W. (2010). Climate change 2007: the physical science basis. *South Afr. Geograph. J.* 92, 86–87. doi: 10.1080/03736245.2010.480842
- Lin, G., Chen, Y., Huang, J., Wang, Y., Ye, Y., and Yang, Q. (2020). Regional disparities of phytoplankton in relation to different water masses in the Northwest Pacific ocean during the spring and summer of 2017. *Acta Oceanol. Sin.* 39, 107–118. doi: 10.1007/s13131-019-1511-6
- Liss, P. S., Malin, G., Turner, S. M., and Holligan, P. M. (1994). Dimethyl sulphide and *Phaeocystis*: A review. *J. Mar. Syst.* 5, 41–53. doi: 10.1016/0924-7963(94)90015-9
- Liss, P. S., and Merlivat, L. (1986). “Air-Sea gas exchange rates: Introduction and synthesis,” in *The Role of Air-Sea Exchange in Geochemical Cycling*, vol. 185. Ed. P. Buat-Ménard (Dordrecht: Springer), 113–127. doi: 10.1007/978-94-009-4738-2\_5
- Li, C., Zhai, W., and Qi, D. (2022). Unveiling controls of the latitudinal gradient of surface pCO<sub>2</sub> in the kuroshio extension and its recirculation regions (northwestern north pacific) in late spring. *Acta Oceanol. Sin.* 41, 1–37. doi: 10.1007/s13131-021-1949-1
- Meehl, G. A. (1996). Climate change from increased CO<sub>2</sub> and direct and indirect effects of sulfate aerosols. *Geophys. Res. Lett.* 23, 3755–3758. doi: 10.1029/96GL03478
- Mehrbach, C., Culbertson, C. H., Hawley, J. E., and Pytkowicz, R. M. (1973). Measurement of the apparent dissociation constants of carbonic acid in seawater at atmospheric pressure. *Limnol. Oceanography* 18 (6), 897–907. doi: 10.4319/lo.1973.18.6.0897
- Mélançon, J., Lévassieur, M., Lizotte, M., Scarratt, M., Tremblay, J.É., Tortell, P., et al. (2016). Impact of ocean acidification on phytoplankton assemblage, growth, and DMS production following Fe-dust additions in the NE Pacific high-nutrient, low-chlorophyll waters. *Biogeosciences* 13, 1677–1692. doi: 10.5194/bg-13-1677-2016
- Millero, F. J. (2007). The marine inorganic carbon cycle. *Chem. Rev.* 107 (2), 308–341. doi: 10.1021/cr0503557
- Mitsudera, H., Taguchi, B., Yoshikawa, Y., Nakamura, H., Waseda, T., and Qu, T. (2004). Numerical study on the oyashio water pathways in the kuroshio-oyashio confluence. *J. Phys. Oceanography* 34, 1174–1196. doi: 10.1175/1520-0485(2004)034<1174:NSOTOW>2.0.CO;2
- Mou, L., Zhang, H. H., Chen, Z. H., and Hu, Y. B. (2022). Processes controlling the carbonate chemistry of surface seawater along the 150°E transect in the Northwest Pacific ocean. *J. Ocean Univ. China* 21, 1672–5182. doi: 10.1007/s11802-022-5207-
- Nightingale, P. D., Malin, G., Law, C. S., Watson, A. J., Liss, S., Liddicoat, M. I., et al. (2000). In situ evaluation of air-sea gas exchange parameterizations using novel conservative and volatile tracers. *Global Biogeochem. Cycles* 14, 373–387. doi: 10.1029/1999GB900091
- Omori, Y., Tanimoto, H., Inomata, S., Kameyama, S., Takao, S., and Suzuki, K. (2013). Evaluation of using unfiltered seawater for underway measurement of dimethyl sulfide in the ocean by online mass spectrometry. *Limnol. Oceanogr. Methods* 11, 549–560. doi: 10.4319/lom.2013.11.549
- Pierrot, D., Lewis, E., and Wallace, D. W. R. (2006). “MS excel program developed for CO<sub>2</sub> system calculations. ORNL/CDIAC105a,” in *Carbon dioxide information analysis center* (Oak Ridge, Tennessee: Oak Ridge National Laboratory, U.S. Department of Energy). doi: 10.3334/CDIAC/otg.CO2SYS\_XLS\_CDIAC105a
- Pierrot, D., Neill, C., Sullivan, K., Castle, R., Wanninkhof, R., Lüger, H., et al. (2009). Recommendations for autonomous underway pCO<sub>2</sub> measuring systems and data-reduction routines. *Deep-Sea Res. Part II: Topic. Stud. Oceanography* 56, 512–522. doi: 10.1016/j.dsr2.2008.12.005
- Qiu, B., and Chen, S. (2010). Eddy-mean flow interaction in the decadal modulating kuroshio extension system. *Deep-Sea Res. Part II: Topic. Stud. Oceanography* 57, 1098–1110. doi: 10.1016/j.dsr2.2008.11.036
- Riebesell, U., Schulz, K. G., Bellerby, R. G. J., Botros, M., Fritsche, P., Meyerhöfer, M., et al. (2007). Enhanced biological carbon consumption in a high CO<sub>2</sub> ocean. *Nature* 450, 545–548. doi: 10.1038/nature06267
- Ritter, A., and Muñoz-Carpena, R. (2013). Performance evaluation of hydrological models: Statistical significance for reducing subjectivity in goodness-of-fit assessments. *J. Hydrol.* 480, 33–45. doi: 10.1016/j.jhydrol.2012.12.004
- Sakurai, Y. (2007). An overview of the oyashio ecosystem. *Deep-Sea Res. Part II: Topic. Stud. Oceanography* 54, 2526–2542. doi: 10.1016/j.dsr2.2007.02.007
- Saltzman, E. S., King, D. B., Holmen, K., and Leck, C. (1993). Experimental determination of the diffusion coefficient of dimethylsulfide in water. *J. Geophys. Res.* 98, 481–486. doi: 10.1029/93jc01858
- Schlundt, C., Tegmeier, S., Lennartz, S. T., Bracher, A., Cheah, W., Krüger, K., et al. (2017). Oxygenated volatile organic carbon in the western Pacific convective center: Ocean cycling, air-sea gas exchange and atmospheric transport. *Atmos. Chem. Phys.* 17, 10837–10854. doi: 10.5194/acp-17-10837-2017
- Simó, R., and Dachs, J. (2002). Global ocean emission of dimethylsulfide predicted from biogeochemical data. *Global Biogeochem. Cycles* 16, 26-1-26-10. doi: 10.1029/2001gb001829
- Stefels, J., Steinke, M., Turner, S., Malin, G., and Belviso, S. (2007). Environmental constraints on the production and removal of the climatically active gas dimethylsulphide (DMS) and implications for ecosystem modelling. *Biogeochemistry* 83, 245–275. doi: 10.1007/s10533-007-9091-5
- Suess, E. (1980). Particulate organic carbon flux in the oceans - surface productivity and oxygen utilization. *Nature* 288, 260–263. doi: 10.1038/288260a0
- Sugimoto, S., Kobayashi, N., and Hanawa, K. (2014). Quasi-decadal variation in intensity of the western part of the winter subarctic SST front in the Western North Pacific: The influence of kuroshio extension path state. *J. Phys. Oceanography* 44, 2753–2762. doi: 10.1175/JPO-D-13-0265.1
- Takahashi, T., Olafsson, J., Goddard, J. G., Chipman, D. W., and Sutherland, S. C. (1993). Seasonal variation of CO<sub>2</sub> and nutrients in the high-latitude oceans: A comparative study. *Global Biogeochem. Cycles* 7, 843–878. doi: 10.1029/93GB02263
- Takahashi, T., Sutherland, S. C., Wanninkhof, R., Sweeney, C., Feely, R. A., Chipman, D. W., et al. (2009). Climatological mean and decadal change in surface ocean pCO<sub>2</sub>, and net sea-air CO<sub>2</sub> flux over the global oceans. *Deep-Sea Res. Part II: Topic. Stud. Oceanography* 56, 554–577. doi: 10.1016/j.dsr2.2008.12.009
- Todd, J. D., Rogers, R., Li, Y. G., Wexler, M., Bond, P. L., Sun, L., et al. (2007). Structural and regulatory genes required to make the gas dimethyl sulfide in bacteria. *Science* 315, 666–669. doi: 10.1126/science.1135370
- Tortell, P. D., Long, M. C., Payne, C. D., Alderkamp, A. C., Dutrieux, P., and Arrigo, K. R. (2012). Spatial distribution of pCO<sub>2</sub>, ΔO<sub>2</sub>/Ar and dimethylsulfide (DMS) in polynya waters and the sea ice zone of the amundsen Sea, Antarctica. *Deep-Sea Res. Part II: Topic. Stud. Oceanography* 71–76, 77–93. doi: 10.1016/j.dsr2.2012.03.010
- Uppström, L. R. (1974). Boron/chlorinity ratio of deep-sea water from Pacific ocean. *Deep-Sea Res.* 21 (2), 161–162. doi: 10.1016/0011-7471(74)90074-6
- Vaillancourt, R. D., Marra, J., Seki, M. P., Parsons, M. L., and Bidigare, R. R. (2003). Impact of a cyclonic eddy on phytoplankton community structure and photosynthetic competency in the subtropical North Pacific ocean. *Deep-Sea Res. Part I: Oceanogr. Res. Pap.* 50, 829–847. doi: 10.1016/S0967-0637(03)00059-1
- Wang, Y., Bi, R., Zhang, J., Gao, J., Takeda, S., Kondo, Y., et al. (2022). Phytoplankton distributions in the kuroshio-oyashio region of the Northwest Pacific ocean: Implications for marine ecology and carbon cycle. *Front. Mar. Sci.* 9. doi: 10.3389/fmars.2022.865142
- Wanninkhof, R. (2014). Relationship between wind speed and gas exchange over the ocean revisited. *Limnol. Oceanography: Methods* 12, 351–362. doi: 10.4319/lom.2014.12.351
- Watanabe, Y. W., Yoshinari, H., Sakamoto, A., Nakano, Y., Kasamatsu, N., Midorikawa, T., et al. (2007). Reconstruction of sea surface dimethylsulfide in the north Pacific during 1970s to 2000s. *Mar. Chem.* 103, 347–358. doi: 10.1016/j.marchem.2006.10.004
- Weiss, R. F., and Price, B. A. (1980). Nitrous oxide solubility in water and seawater. *Mar. Chem.* 8, 347–359. doi: 10.1016/0304-4203(80)90024-9
- Wohl, C., Brown, I., Kitidis, V., Jones, A. E., Sturges, W. T., Nightingale, P. D., et al. (2020). Underway seawater and atmospheric measurements of volatile organic compounds in the southern ocean. *Biogeosciences* 17, 2593–2619. doi: 10.5194/bg-17-2593-2020
- Wohl, C., Jones, A. E., Sturges, W. T., Nightingale, P. D., Else, B., Butterworth, B. J., et al. (2022). Sea Ice concentration impacts dissolved organic gases in the Canadian Arctic. 1021–1045. *Biogeosciences* 19, 1021–1045. doi: 10.5194/bg-19-1021-2022
- Xi, C., Dong, H., Ke-feng, M., and Yan, L. (2018). Detailed investigation of the three-dimensional structure of a mesoscale cold eddy in the kuroshio extension region. *J. Oper. Oceanography* 11, 87–99. doi: 10.1080/1755876X.2018.1505069
- Yanagi, T., and Takahashi, S. (1993). Seasonal variation of circulations in the East China Sea and the Yellow Sea. *J. Oceanography* 49, 503–520. doi: 10.1007/BF02237458
- Yasunaka, S., Nojiri, Y., Nakaoka, S., Ono, T., Whitney, F. A., and Telszewski, M. (2014). Mapping of sea surface nutrients in the north Pacific: Basin-wide distribution and seasonal to interannual variability. *J. Geophys. Res.: Oceans* 119, 7756–7771. doi: 10.1002/2014JC010318
- Zhang, M., and Chen, L. (2015). Continuous underway measurements of dimethyl sulfide in seawater by purge and trap gas chromatography coupled with pulsed flame photometric detection. *Mar. Chem.* 174, 67–72. doi: 10.1016/j.marchem.2015.05.006
- Zhang, J., Chen, Y., Ren, X., Patil, V., Sun, L., Li, X., et al. (2022). Distribution of phytoplankton in the East China Sea and the southern yellow Sea in spring in relation to environmental variables and dimethylsulfide compounds. *Acta Oceanol. Sin.* 41, 41–53. doi: 10.1007/s13131-021-1913-0
- Zhang, M., Gao, W., Yan, J., Wu, Y., Marandino, C. A., Park, K., et al. (2019). An integrated sampler for shipboard underway measurement of dimethyl sulfide in surface seawater and air. *Atmos. Environ.* 209, 86–91. doi: 10.1016/j.atmosenv.2019.04.022
- Zhang, M., Marandino, C. A., Chen, L., Sun, H., Gao, Z., Park, K., et al. (2017). Characteristics of the surface water DMS and pCO<sub>2</sub> distributions and their relationships in the southern ocean, southeast Indian ocean, and northwest Pacific ocean. *Global Biogeochem. Cycles* 31, 1318–1331. doi: 10.1002/2017GB005637
- Zindler, C., Marandino, C. A., Bange, H. W., Schütté, F., and Saltzman, E. S. (2014). Nutrient availability determines dimethyl sulfide and isoprene distribution in the eastern Atlantic ocean. *Geophys. Res. Lett.* 41, 6298–6305. doi: 10.1002/2014GL059547