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Carbonate system data tracing freshwater inflow into the Ross Sea through the eastern gate and along the Ross Ice Shelf (Antarctica)

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The eastern Ross Sea is a key area to understand the role of the Amundsen Sea inflow of freshwater that can influence the Ross Sea water properties and salt budget. A survey was carried out in the eastern Ross Sea during the austral summer 2019–20 to evaluate the contribution of the Amundsen Sea Water (ASW) to the salinity variability. A total of 248 seawater samples were collected for the analysis of total alkalinity (A_T) and pH. The data collected were used together with temperature and salinity to obtain a full description of the carbonate system properties including total inorganic carbon (C_T), CO_2 partial pressure (pCO_2), calcium carbonate saturation state of aragonite and calcite (Ω), and Revelle factor. Moreover, we estimated the anthropogenic carbon (C_{ant}) throughout the TrOCA method to better understand the carbon cycle, also considering the effect of atmospheric CO_2 uptake on ocean acidification. We used principal component analysis (PCA) to investigate the major controls on the carbonate system parameters with the aim of defining their sensitivity as chemical tracers. The changes in carbonate chemistry in surface waters were mainly due to the physical properties. A_T and pH traced the entry of the ASW showing limited mixing between water masses on the shelf area. Shelf waters were enriched in C_{ant} , which resulted lower than the estimated value for shelf waters produced in western Ross Sea.

KEYWORDS

carbonate system, Ross Sea, Amundsen Sea Water, chemical tracers, anthropogenic carbon, carbonate saturation grade

Introduction

The Ross Sea plays a fundamental role in the global carbon budget and in the air–sea carbon dioxide (CO₂) flux, behaving as an atmospheric CO₂ sink (Dejong and Dunbar, 2017). The Ross Sea is one of the most biologically productive regions in the Southern Ocean (Arrigo et al., 1999; Arrigo and van Dijken, 2004; Smith et al., 2010; Iudicone et al., 2011; Smith et al., 2014; Mangoni et al., 2017) and a key site of Antarctic Bottom Water (AABW) formation (Jacobs et al., 1970; Orsi et al. 1999; Bergamasco et al., 2002; Rivaro et al., 2010a).

The AABW is produced by the tidal mixing of the Dense Shelf Water (DSW) with the Circumpolar Deep Water (CDW) along the continental shelf margins in the western sector (Whitworth and Orsi, 2006; Castagno et al., 2017; Bowen et al., 2021) and it contributes significantly to the ventilation of the deep layer of the Southern Ocean (Orsi and Wiederwohl, 2009). Therefore, it plays an important role in sequestering CO₂ from the atmosphere and transferring it into the deep ocean (Caldeira and Duffy, 2000; Sabine et al., 2004; Sandrini et al., 2007). The magnitude of the CO₂ sink depends on the environmental conditions together with physical and biological processes. Therefore, it is crucial to define the parameters (i.e. environmental, physical, biological) that influence the variability of the properties of the carbonate system.

The properties of the carbonate system in natural waters can be defined by measuring the pH, the total alkalinity (A_T), the total inorganic carbon (C_T), and the CO₂ partial pressure in wet (100% water-saturated) air at equilibrium with the sample (pCO₂), that is proportional to the dissolved CO₂. The addition of freshwater together with the increase of atmospheric CO₂ could affect the carbonate system parameters, such as pH and carbonate saturation grade (Ω) which are key variables for monitoring the state of ocean acidification (OA). In fact, the increase in atmospheric CO₂ concentration displaces the chemical equilibria occurring in seawater between the different species contributing to the C_T (i.e. HCO₃⁻, CO₂, and CO₃²⁻) causing a decrease of the pH for the formation of H₃O⁺ ions. Furthermore, CO₂ addition lowers the Ω of calcium carbonate (Feely et al., 2004).

Polar seas are particularly vulnerable to OA, due to the cold, relatively fresh, and low A_T surface waters that have a great potential for CO₂ uptake and a low buffer capacity (Chierici and Fransson, 2009; Steinacher et al., 2009). Changes in pH may affect the biogeochemical cycles of many metals, including the bioavailability of iron that control the biological activities (Millero et al., 2005). Anthropogenic carbon (C_{ant}) can be estimated using carbonate system properties. Touratier and Goyet defined the semi-conservative tracer TrOCA (Tracer combining Oxygen, inorganic Carbon and total Alkalinity) (Touratier and Goyet, 2004a; Touratier and Goyet, 2004b; Touratier et al., 2007).

In the recent decades, AABW has been subject to warming, freshening, and decreasing in volume and density (Johnson,

2008; Purkey and Johnson, 2012; Purkey and Johnson, 2013). The largest freshening has been observed in the Pacific and Australian Antarctic Basins where bottom water is originated from the Ross Sea DSW. The 50 years of AABW freshening has been linked to the salinity decrease of the HSSW formed in the Ross Sea due to the increased continental ice discharge and melting upstream in the Amundsen Sea (Jacobs et al., 2022). Recent observations show a rebound of both Ross Sea DSW (Castagno et al., 2019) and AABW salinity in the Southern Pacific (Silvano et al., 2020). This rebound has been related to a reduced sea ice import through the Ross Sea eastern gate driven by climate anomalies (Silvano et al., 2020). Furthermore, Guo et al. (2021) related the rapid enhancement of DSW salinity after 2014 to the reduced freshwater input into the Ross Sea due to weakening of the westward zonal flow from the upstream Amundsen Sea. Therefore, the Eastern Sector of the Ross Sea (ESRS) is an area of great importance for the whole Ross Sea shelf water production and variability.

In this paper we consider the ESRS as the area east of the Greenwich antemeridian (according also to the SOOS Science and Implementation Plan 2021-2025, <https://www.soos.aq>). The observations in this sector are very scarce, despite this being the area where the water masses coming from the West Antarctica sector pass through. According Orsi and Wiederwhol (2009) the ESRS is characterized by the presence of Low Salinity Shelf Waters (LSSW) over almost all the continental shelf area deep layer (below the neutral density – γ^n – surface of 28.27 kg m⁻³) and the absence of HSSW and AABW production. A fresher ISW than in the western sector is observed at the edge of the Ross Ice Shelf (RIS). The intermediate layer, defined as the layer between the isopycnal surface of γ^n 28 and 28.27 kg m⁻³, is divided into two sub-sectors: the first occupies the easternmost area of the ESRS where mainly Modified Shelf Waters (MSW) can be observed; the westernmost sub-sector, instead, is characterized by the intrusion of CDW. Then, a strong temperature front occurs in between the Hayes and Houtz Bank (Figure 1) and no intrusion of CDW onto the shelf can be observed in the easternmost sub-sector. The surface layer (the layer above the γ^n surface of 28 kg m⁻³) shows salinity value greater than 34.3 in the ESRS central sector, but close to the coast, near Cape Colbeck, a fresh water mass coming from West Antarctica can be observed. According to the surface circulation scheme (e.g. Sedwick et al., 2011), this fresh water mass moves westward along the RIS edge becoming more saline.

The primary objective of the ESTRO (Effect of the eaSTern inflow of water on the ROSS Sea salinity field variability) project was to collect, for the first time, a comprehensive dataset (physical and chemical observations) to characterize the eastern Ross Sea sector water masses and dynamics focusing on the upstream water inflow coming from western Antarctica and entering the Ross Sea.

Here, we present a unique set water column data of pH and A_T and derived carbonate system parameters (C_T, pCO₂ and Ω) from the nearly unexplored eastern sector of the Ross Sea and the along the Ross Ice Shelf (RIS) collected during the 2019–20

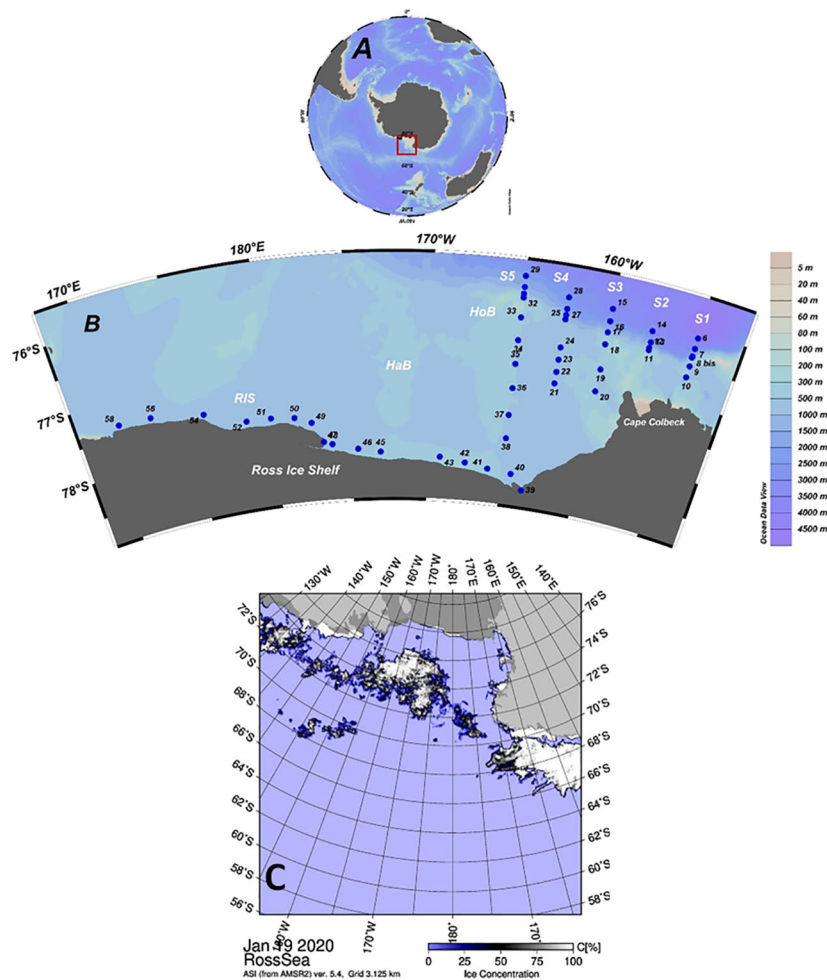


FIGURE 1

(A) Position of the Ross Sea in the Antarctic continent. (B) Location of ESTRO project sampling stations. S1: stations 6–10; S2: stations 11–14; S3: stations 15–20; S4: stations 21–28; S5: stations 29–38; RIS: stations 39–58. HaB: Hayes Bank; HoB: Houtz Bank. (C) Advanced Microwave Scanning Radiometer 2 (AMSR2) sea ice concentration map, by the University of Bremen provided for the start day of the ESTRO project (19 January 2020) (<http://iup.physik.uni-bremen.de/iuppage/psa/2001/amsrop.html>).

austral summer. We will discuss the distribution of carbonate system parameters by examining the drivers of their variability in surface waters, the possible use as chemical tracers in intermediate and deep waters. We will also do some considerations regarding the sensitivity of the area to ocean acidification and estimates of anthropogenic carbon.

Materials and methods

Sample collection and processing

Sea water samples were collected on board the R.V. Laura Bassi in the Ross Sea (Antarctica) during the 2020 austral summer,

as part of ESTRO project of the Italian National Program of Research in Antarctica (PNRA) from 19 to 24 of January 2020.

A Sea Bird Electronics SBE9/11plus CTD profiler with two pairs of temperature-conductivity sensors was employed to acquire conductivity, temperature, and depth data. The CTD was coupled to an SBE 23 O₂ sensor and to a Chelsea Aquatrack III fluorometer for measuring the oxygen concentration and the fluorescence, respectively.

An SBE 32 plastic coated carousel sampler was used to collect water samples from 24 12-L Niskin bottles. A total of 248 seawater samples for carbonate system analyses were collected at 52 stations at selected depths based on the CTD profiles.

The ESTRO cruise covered the shelf region and the eastern slope area of the Ross Sea. The stations made up five transects in

the eastern sector (S1-S5) and one transect along the RIS (Figure 1). The samples were stored in 500 ml borosilicate glass bottles following standard operating procedures (Dickson et al., 2007). The samples were poisoned in the container with saturated HgCl₂ to stop biological activity and were then stored in dark, cold (+4°C) conditions.

Hydrographic data

The CTD calibrations were performed before and after both cruises. Data were acquired at the maximum frequency (24 Hz) with a descent speed of 1 m s⁻¹. The data were corrected and processed according to international procedures (SCOR Working Group, 1988). θ and S were computed using standard algorithms (Fofonoff Jr. and Millard, 1983). The data were processed, and quality was checked using the CTD Sea Bird Electronics data processing software. Finally, the profiles were vertically averaged over 1-m-depth bins.

Total alkalinity and pH analysis

A_T analysis was carried out by potentiometric titration system with a Metrohm 806 Exchange Unit and a 702 SM titrino dosimat following the method described in Rivaro et al. (2010b); Rivaro et al. (2017). Certified Reference Material (batch 191 and 196, provided by A. G. Dickson, Scripps Institution of Oceanography) was analyzed once a week. The precision was $\pm 4.0 \mu\text{mol kg}^{-1}$ and the recovery was 99.8%.

Potentiometric pH measures were carried out by a Metrohm 827 pH meter employing a combination glass/reference electrode with an NTC temperature sensor. pH was expressed on the pH total scale (i.e. [H⁺] as moles per kilogram of seawater, pH_T).

The Tris(hydroxymethyl)aminomethane (TRIS) buffer, provided by A. G. Dickson (Batch #T33) was used to standardize the pH electrode. Both the TRIS buffer and the seawater samples were brought to the same temperature (25°C \pm 0.1°C) using a thermostatic water bath before the measurements were completed. The pH_T values at 25°C were then recalculated at *in situ* temperature and pressure conditions (pH *in situ*). The precision of the pH measurement was ± 0.007 U and it was evaluated by repeated analysis of the A_T certified material.

Carbonate system data processing

The CO₂SYS program was used to recalculate at *in situ* conditions pH and A_T together with C_T, pCO₂, and Ω calcite and aragonite (Ω Ca and Ar, respectively) from the pH and A_T data measured at 25°C (Pierrot et al., 2006). The Revelle factor (RF) was computed for surface samples only. Equilibrium constants of CO₂ (K1 and K2) of Millero, 2006 and the pH_T scale were used for the calculations (Millero, 2007) together with CTD data (temperature, salinity and pressure).

Estimating air-sea CO₂ flux

The instantaneous air-sea CO₂ flux (F, mmol m⁻²d⁻¹) was calculated using the formula:

$$1) F = ks(\Delta p\text{CO}_2)$$

where k is the CO₂ gas transfer velocity (cm h⁻¹), s is the solubility of CO₂ (mol kg⁻¹ atm⁻¹) and $\Delta p\text{CO}_2$ is the difference between the pCO₂ of the seawater (pCO_{2SW}) and the atmosphere (pCO_{2atm}). We calculated k following the Wanninkhof formulation (2014):

$$2) k = 0.251u^2(660/Sc)^{0.5}$$

where u is the wind speed at 10 m height and Sc is the temperature dependent Schmidt number. The shipboard wind speed data, measured at 10 m height, were used. Negative $\Delta p\text{CO}_2$ values indicate absorption of the gas by the sea surface, while positive values correspond to the release of the gas into the atmosphere (Wanninkhof, 2014).

The $\Delta p\text{CO}_2$ values were calculated as the difference between the pCO_{2SW} computed by CO₂SYS and the mean pCO_{2atm} value of 409 μatm available from the South Pole Observatory (<https://gml.noaa.gov/obop/spo/>). The pCO_{2atm} was corrected to 100% humidity at *in situ* sea surface temperature and salinity using the standard formula as given in Dickson (2007).

Estimating anthropogenic CO₂ (C_{ant})

The C_{ant} was estimated throughout the TrOCA method. This approach is based on the definition given by Touratier et al. (2007) of TrOCA and TrOCA⁰, where TrOCA⁰ is defined as the pre-industrial TrOCA.

$$3) \text{TrOCA} = O_2 + a \left(C_T - \frac{1}{2} A_T \right)$$

$$4) \text{TrOCA}^0 = O_2^0 + a \left(C_T^0 - \frac{1}{2} A_T^0 \right)$$

$$5) \text{TrOCA}^0 = e^{\left[\frac{b+c\theta+d}{AT^2} \right]}$$

Since in Equation 3) only C_T is significantly affected by the atmospheric increase of CO₂, C_{ant} concentration can be computed as follows:

$$6) C_{\text{ant}} = C_T - C_T^0 = \text{TrOCA} - \text{TrOCA}^0$$

where a represents the Redfield coefficient. Since TrOCA⁰ has conservative property, the four parameters (a , b , c , d) are determined by minimizing the fit standard error of Eq. 5). Hence the C_{ant} is estimated using the following equation (Sandrini et al., 2007):

$$7) \text{ Cant} = \text{O}_2 + 1.279[\text{CT} - 12\text{AT}] - e[7.511 - (1.087 \times 10^{-2})\theta - 7.81 \times 10^5/\text{AT}^{21.279}]$$

Principal component analysis

Principal component analysis (PCA) was applied to the dataset to explore the correlations between carbonate system data and the measured environmental parameters (temperature, salinity, fluorescence, and O_2). First, data were normalized by log-transformation, then the data matrix was processed after autoscaling the data using the R based software CAT (Leardi et al., 2017).

Surface samples were considered separately from those collected at depths deeper than 100 m. Fluorescence data were not considered for deep samples, as the values were close to zero.

Results

Physical characteristics of the water masses

To identify the water masses present in the study area we have used hydrographic ranges defined by Orsi and Wiederwohl (2009) and Budillon et al. (2011) for the water originated in the Ross Sea

and by Randall-Goodwin et al. (2015) for the upstream water that arrives with the westward coastal current at the eastern gate.

Based on the thermohaline properties and neutral density γ^n (Jackett and McDougall, 1997), we have identified the following water masses (Figure 2 and Table 1):

Antarctic Surface Water (AASW) characterized by a γ^n lower than 28.00 kg m^{-3} and by a wide range of temperatures and salinities;

The saltiest High Salinity Shelf Water (HSSW) defined by $S > 34.7$, $\theta < -1.85^\circ\text{C}$ and $\gamma^n > 28.27 \text{ kg m}^{-3}$. It is produced in the Terra Nova Bay (TNB) polynya during wintertime by brine rejection during sea ice formation. In the study area HSSW is found only on the western sector of the RIS;

Ice Shelf Water (ISW) is formed by basal melting of the Ross Ice Shelf (RIS) by the dense shelf waters and is characterized by temperature below the surface freezing point ($\theta < -1.92^\circ\text{C}$). This is found mainly at intermediate depth along the RIS with two cores in the central and eastern Ross Sea. The colder and saltier core is found in the central Ross Sea at about 180°E ;

Low Salinity Shelf Water (LSSW) is defined by $S < 34.62$, $\theta < -1.85^\circ\text{C}$ and $\gamma^n > 28.27$. This water is formed by the interaction of surface water with the subsurface layer during freezing-melting cycles (Jacobs et al., 1985). In the study area LSSW is found at the bottom and at intermediate depths in the eastern continental shelf;

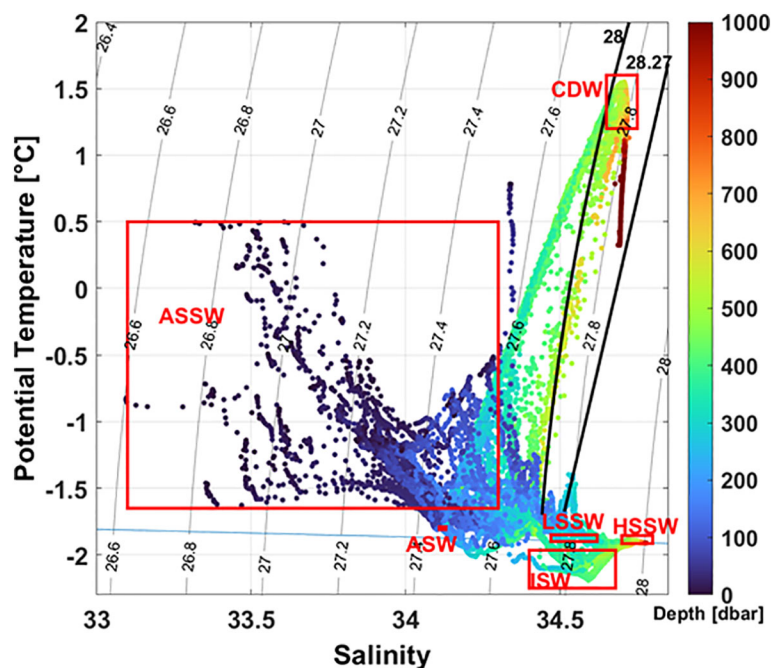


FIGURE 2

Potential temperature salinity (θ/S) plot. Solid lines show the 28.00 and 28.27 kg m^{-3} neutral density γ^n surfaces. The blue horizontal line shows the surface freezing point of seawater. Major water masses are labeled: Antarctic Surface Water (AASW), Circumpolar Deep Water (CDW), High Salinity Shelf Water (HSSW), Low Salinity Shelf Water (LSSW), Ice Shelf Water (ISW), and Amundsen Sea Water (ASW).

TABLE 1 Bounding potential temperature (θ), salinity and neutral density (γ^n) values that define the major water masses in the area investigated by the ESTRO cruise.

Water mass	θ (°C)	CT (°C)	S	SA	γ^n (kg m ⁻³)
AASW	-1.65 > θ < 0.50	-1.65>CT<0.50	33.10–34.31	33.26–34.48	<28.00
HSSW	<-1.85	<-1.85	>34.7	>34.87	>28.27
ISW	<-1.92	<-1.92	>34.4	>34.57	>28.27
LSSW	<-1.85	<-1.85	<34.62	<34.79	>28.27
CDW	>1.20	>1.20	–	–	28.00> γ^n <28.27
ASW	-1.79 to -1.81	-1.79 to -1.81	34.11–34.13	34.28–34.30	<28.27

Water mass properties are also reported in conservative temperature (CT) and absolute salinity (SA).

Circumpolar Deep Water (CDW) is a relatively warm, salty, low-oxygenated, and nutrient-rich water mass and is the primary source of heat and nutrients over the continental shelf. CDW is defined by $\gamma^n > 28.00$ kg m⁻³ and $\theta > 1.2^\circ$ C. In the study area, CDW is mainly confined on the continental slope and does not intrude onto the shelf.

Amundsen Sea Water (ASW) is formed in the polynya area of the Amundsen Sea during winter and is characterized by $\theta \approx -1.81^\circ$ C and $S = 34.13$. We identified this water in the subsurface layer between 100 and 300 m in northeastern stations of the study area, both on the slope and over the continental shelf area. The ASW found in the study area has potential temperature and salinity slightly modified from the source region with values of θ between -1.79° and -1.81° C and S between 34.11 and 34.13.

Carbonate system parameters

The carbonate system parameters data were constrained by the physical data, allowing us to define mean values for each water mass. The mean values and relative standard deviations of A_T , pH, C_T , pCO_2 , Ω_{Ca} , and Ω_{Ar} are summarized in Table 2. Our results fall within the ranges of previously collected data in the Ross Sea (Sandrini et al., 2007; Mattsdotter Björk et al., 2014; Rivaro et al., 2014; Dejong et al., 2015; Rivaro et al., 2019). All parameters showed greater variability in the AASW than in deep waters, where they varied within a relatively narrow range. A general increase with depth was observed for A_T , C_T , and pCO_2 . In particular, the highest concentration of A_T and C_T (2365

$\mu\text{mol kg}^{-1}$ and 2272 $\mu\text{mol kg}^{-1}$, respectively) was measured in the HSSW in the western RIS section and the highest pCO_2 (503 μatm) in the CDW. The pH decreased with depth with the absolute minimum (7.83) measured at stations 14 and 15 (Section S2) at a 2500-m depth, where CDW was identified.

The computed surface water pCO_2 showed undersaturation compared to atmospheric value at all stations except at stations 12 and 13 (409 and 452 μatm , respectively) collected in the eastern sector (section S2). High spatial variability was observed, with the lowest pCO_2 at station 56 (259 μatm) matching the minimum of C_T (2160 $\mu\text{mol kg}^{-1}$) and the maximum of pH (8.21). Negative ΔpCO_2 values found, in general, suggested absorption of the gas by the sea surface, particularly along RIS (Table 3).

Carbonate dissolution is primarily controlled by the saturation state of seawater with respect to carbonate minerals (Ω). The solubility of calcite and aragonite increases with depth. All surface and sub-surface samples showed supersaturation conditions ($\Omega >$) with respect to calcite and aragonite. Undersaturation conditions of aragonite ($\Omega < 1$) were only found at a few offshore stations at deep depths, with the absolute minimum being 0.7 at station 14, at 2500 m. The values found agree with what reported by Manno et al. (2007) in the western Ross Sea, where only the deepest layers (approximately below 1000 m) were undersaturated with respect to aragonite.

Due to the large seasonal variability in the surface waters, the TrOCA method is not applicable; therefore, we only computed C_{ant} for depths below 100 m. An uncertainty of 8 $\mu\text{mol kg sw}^{-1}$ associated with the mean C_{ant} value was estimated. The

TABLE 2 Mean and standard deviation of values of carbonate system parameters in water masses. .

	A_T ($\mu\text{mol kg sw}^{-1}$)	pH <i>in situ</i>	pCO_2 (μatm)	C_T ($\mu\text{mol kg sw}^{-1}$)	Ω_{Ca}	Ω_{Ar}
AASW	2295 \pm 18	8.09 \pm 0.05	353 \pm 41	2167 \pm 15	2.4 \pm 0.4	1.5 \pm 0.2
ASW	2310 \pm 5	8.00 \pm 0.02	435 \pm 23	2219 \pm 8	1.9 \pm 0.1	1.2 \pm 0.1
LSSW	2328 \pm 6	7.98 \pm 0.03	438 \pm 30	2235 \pm 9	1.8 \pm 0.1	1.1 \pm 0.1
HSSW	2365 \pm 2	7.97 \pm 0.02	439 \pm 5	2272 \pm 3	1.7 \pm 0.1	1.1 \pm 0.1
DISW	2338 \pm 17	8.02 \pm 0.01	404 \pm 11	2234 \pm 17	1.9 \pm 0.1	1.2 \pm 0.1
CDW	2355 \pm 9	7.90 \pm 0.04	503 \pm 29	2262 \pm 10	1.6 \pm 0.3	1.0 \pm 0.2

TABLE 3 Air–sea CO₂ gradient ($\Delta p\text{CO}_2$), wind speed and estimated CO₂ flux (F).

Station	$\Delta p\text{CO}_2$	Wind speed (m s ⁻¹)	F (mmol m ⁻² d ⁻¹)
7	-59	5.61	-1.9
8	-64	10.29	-6.9
10	-79	7.72	-4.8
12	0.3	7.20	-0.02
13	43	8.23	3.0
15	-20	7.72	-1.2
16	-46	4.89	-1.1
17	-44	7.20	-2.3
20	-35	11.52	-4.7
23	-41	8.95	-3.4
24	-11	4.22	-0.2
25	-10	3.86	-0.1
26	-38	4.94	-1.0
27	-33	2.57	-0.2
28	-83	1.54	-0.2
29	-68	4.22	-1.3
30	-48	2.42	-0.3
31	-60	1.85	-0.2
33	-40	0.72	0.0
36	-87	3.60	-1.1
37	-97	4.12	-1.7
38	-94	4.12	-1.6
40	-104	7.77	-6.4
41	-93	9.41	-8.5
42	-105	8.08	-7.0
43	-92	8.08	-6.2
45	-102	6.94	-5.1
46	-91	4.48	-1.9
47	-84	5.25	-2.4
48	-88	6.17	-3.4
49	-79	4.27	-1.5
51	-111	3.86	-1.7
52	-21	5.81	-0.7
54	-104	6.02	-3.8
56	-150	7.41	-8.4
58	-141	6.17	-5.4

calculated C_{ant} concentration ranged from 21 $\mu\text{mol kg sw}^{-1}$ in CDW to 65 $\mu\text{mol kg sw}^{-1}$ in HSSW. The water masses observed over the shelf in the eastern sector (ASW and LSSW) were also enriched in C_{ant} , albeit at slightly lower concentrations than the HSSW (54 $\mu\text{mol kg sw}^{-1}$).

Principal component analysis

Principal component analysis applied to surface samples led to the identification of two principal components (PC) explaining about 82% of the total variance: PC1 explained

62.6% and PC2 explained a further 19.7%, which is valuable for the exploratory data analysis performed in this study and comparable to previous studies (Rivarolo et al., 2019).

As regards the loading plot (Figure 3A), O₂, pH, fluorescence, Ω Ca, and Ω Ar loaded at negative values of PC1 and positive values of PC2. They were negatively correlated with the A_T and S that loaded at negative values of PC1 and PC2, and with pCO₂ and RF that are loaded at positive values of PC1 and negative values of PC2. Finally, T and C_T were negatively correlated, with the temperature loading at positive values of both PC1 and PC2 and the C_T loading at negative values of both PC1 and PC2.

The score plot (Figure 3B) highlights that the samples were ordered in three groups depending on the sampling area. In general, the samples were mainly distributed along PC1, moving from the samples collected in the western part of the RIS (station 56 and 58), characterized by high values of O₂, pH, fluorescence and Ω , to the stations belonging to the eastern Ross Sea, characterized by higher values of pCO₂ and of RF, in between the group composed by the stations collected at the easternmost section (S1), characterized by the highest temperature, lowest C_T, and intermediate values of O₂ and pH.

Two principal components were identified for the deeper samples which explained 69.6% and 16.8% of the total variance. pH, Ω Ar, and Ω Ca loaded at positive values of PC1 and negative values of PC2 and were negatively correlated with C_T, T, S, and O₂. C_T and T loaded at negative values of PC1 and PC2, whereas O₂ loaded at positive values of PC1 and PC2. C_T, T, S, and A_T loaded at negative values of PC1 and PC2 and were negatively correlated with pCO₂, which was placed in the opposite side of the PCA plot (Figure 4A). From the score plot, it was possible to identify different groups of samples, principally distributed along PC1, moving from samples characterized by high values of T and C_T to samples with high values of pH, Ω , and O₂ (Figure 4B).

Discussion

Variability of the carbonate system in the surface waters

The distribution of the carbonate system parameters in AASW is controlled by physical properties together with the circulation pattern and by biological activities.

Few A_T measurements have been previously reported for the eastern Ross Sea and for the RIS area (Mattsdotter Björk et al., 2014; Dejong et al., 2015; Rivarolo et al., 2019). Both the PCA results (Figure 3) and the longitudinal distribution outlined the link of A_T variability with the salinity. Since A_T strongly covaried with salinity (n = 52, r = 0.911, p = 0.01), the lowest A_T values were in the eastern Ross Sea, where the salinity was lowest and temperature highest (Figures 5A–C). In fact, the Amundsen Sea

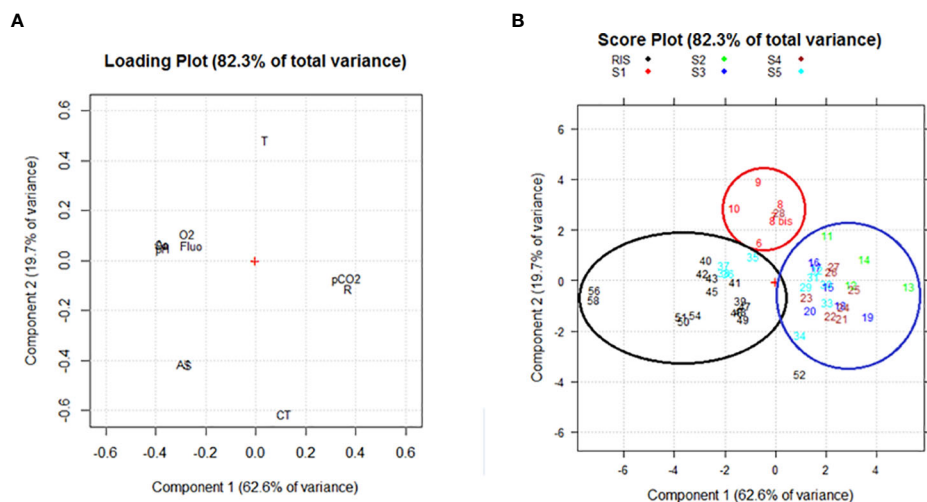


FIGURE 3

(A) Loading plot and (B) Score plot obtained from the PCA for the surface samples. The following abridgements were used for the variables in the loading plot: dissolved oxygen (O₂), total inorganic carbon (C_T), total alkalinity (A_T), partial pressure of CO₂ (pCO₂), Revelle factor (R), saturation grade of calcite and aragonite (Ca and Ar), fluorescence (Fluo), temperature (T), and salinity (S).

has lower A_T than the Ross Sea, which coincides with lower salinity. Mattsdotter Björk et al. (2014) reported for the Amundsen Sea surface waters a mean value and relative standard deviation of A_T of 2995 ± 6 μmol kg⁻¹ that are close to our data (2286 ± 11 μmol kg⁻¹) measured in eastern Ross Sea (S1-S5). The slight increase in A_T along the RIS (2313 ± 13 μmol kg⁻¹) was statistically significant compared to the S1-S5 area (*t*-test, 95% confidence level). We can state that the A_T distribution found in our survey traced near Cape Colbeck a freshwater mass coming from West Antarctica, which, moving westward along the RIS edge, became more saline with a higher A_T. Therefore, our finding is consistent with the general Ross Sea surface circulation scheme (Sedwick et al., 2011).

The distribution of pH and C_T also showed differences between the various sections examined (Figures 5D, E). Our CT data (2166 ± 16 μmol kg⁻¹) at those stations sampled in S1-S5 sections agreed to the mean value and relative standard deviation reported by Mattsdotter Björk et al. (2014) for the Amundsen Sea (2179 ± 9 μmol kg⁻¹). Differently to A_T, the C_T values found along the RIS (2170 ± 17 μmol kg⁻¹) were not statistically different from those measured in the S1-S5 area. Moreover PCA results evidenced that the C_T correlation with salinity, although positive, was less significant than that shown by the A_T (n = 52, r = 0.350, p = 0.01), similar to the C_T-A_T correlation (n = 52, r = 0.375, p = 0.01), outlining the C_T's less conservative behavior in tracing the fresh water mass from west Antarctica.

Besides the circulation pattern, A_T and C_T variability in the AASW can also be controlled by freshwater addition from melting sea ice, which also acts to change salinity (Lee et al., 2006). Both field and laboratory studies have demonstrated that

ikaite, a hydrous calcium carbonate mineral (CaCO₃·6H₂O), precipitates out within brines during sea ice formation (Marion et al., 2001; Papadimitriou et al., 2007). The net effect of ikaite precipitation in brine is to reduce the concentration of A_T and C_T, whilst increasing the pCO₂. Any brine released during ice melt transfers these inorganic carbon characteristics to the underlying water. A relationship between normalized inorganic carbon and total normalized alkalinity (C_TN and A_TN, respectively) of 1:2 is indicative of ikaite precipitation during sea ice formation (Zeebe and Wolf-Gladrow, 2001; Jones et al., 2010).

Thus, surface A_T and C_T were normalized (A_TN and C_TN, respectively) to a constant salinity of 34.50 (roughly the average salinity of the Ross Sea upper water column) to remove the effects of dilution from the melting sea ice (Dunbar et al., 2003). A_TN and C_TN ranged from 2336 to 2370 μmol kg⁻¹ and from 2177 to 2240 μmol kg⁻¹, respectively (Figure 6). No 1:2 ratios were observed from the distribution of C_TN and A_TN except for a few stations sampled in S2, S3, and S4 (stations 12, 13, 15, 18, 22, and 24) where pCO₂ higher than 398 μatm and pH lower than 8.05 were measured.

The ESTRO cruise was characterized by ice-free conditions over most of the study area, except for the shelf area of section 2 which, for this reason, was not possible to complete. Therefore, when the sampling took place, the possible effect of melting sea ice on the distribution of C_T and A_T was no longer visible, apart at the abovementioned stations. Potentially, the contribution from melting sea ice in surface waters can be estimated by a combination of salinity and δ¹⁸O (Dini and Stenni, 2000). The measurement of δ¹⁸O was not included in ESTRO's sampling

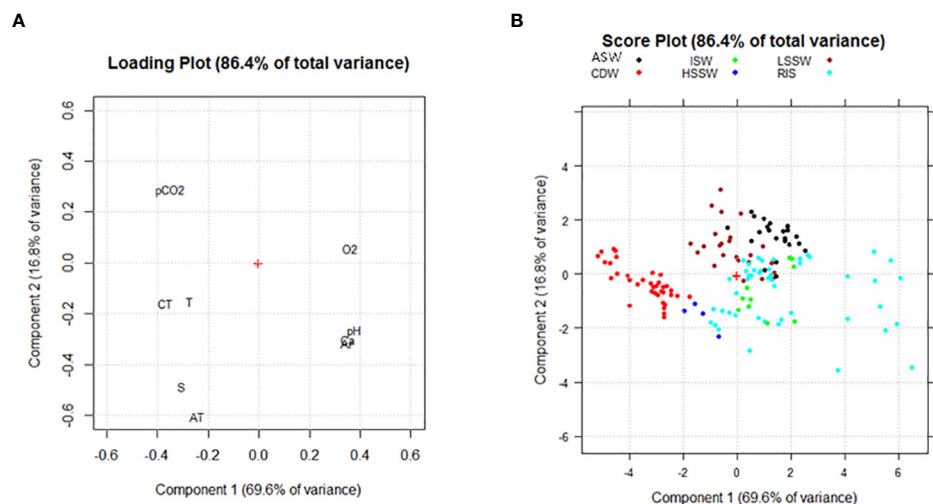


FIGURE 4

(A) Loading plot and (B) Score plot obtained from the PCA for samples collected at depths deeper than 100 m. The following abridgements were used for the variables in the loading plot: dissolved oxygen (O_2), total inorganic carbon (C_T), total alkalinity (A_T), partial pressure of CO_2 (pCO_2), saturation grade of calcite and aragonite (Ca and Ar), temperature (T) and salinity (S). The following colors were used to identify water masses: black = Amundsen Sea Water (ASW); brown = Low Salinity Shelf Water (LSSW); red = Circumpolar Deep Water (CDW); green = Ice Shelf Water (ISW); blue = High Salinity Shelf Water (HSSW); cyan = Shelf Waters sampled along the RIS that belong to neither HSSW nor ISW (RIS).

plan, so no further conclusions can be drawn from this. Moreover, the high C_T , CO_2 -rich, and low pH surface water explained the low Ω_{Ca} and Ω_{Ar} found at those stations (Figures 5F–H).

Ocean pH is affected by biological primary production and respiration, physical mixing, air-sea CO_2 exchange, and sea water CO_2 chemistry. The melting of sea ice, besides influencing the chemical properties, may promote the development of phytoplankton since it affects the depth of the Upper Mixed Layer (UML) and the stability of the water column (Mangoni et al., 2017).

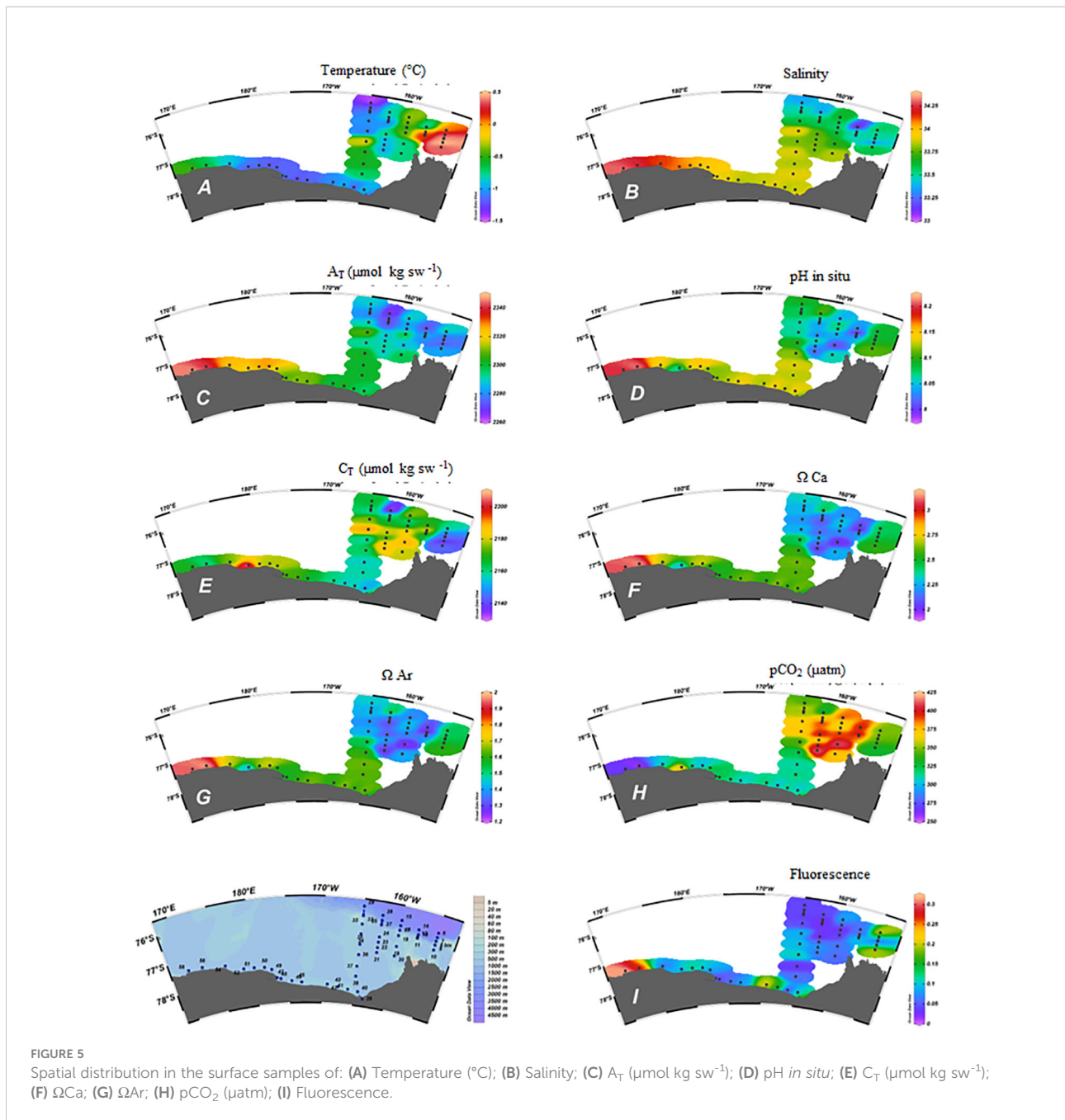
The Ross Sea is a key region for global carbon uptake and is characterized by a strong seasonality with the annual CO_2 uptake being mediated by biological carbon drawdown in summer that increases pH, Ω , and O_2 concentration (Arrigo et al., 2008; Smith et al., 2014; Rivarolo et al., 2019). During biological production, C_T is consumed (increase in pH), which increases $[CO_3^{2-}]$ and the saturation state for aragonite and calcite (Chierici and Frasson, 2009).

No chlorophyll data were collected during the ESTRO cruise; however, information about primary production can be obtained by looking at the *in situ* fluorescence data (Figure 5I) or at satellite maps that provide an indication of the abundance of phytoplankton. Both the low fluorescence data acquired during the cruise and the MODIS map referred to the investigated period (available at <http://giovanni.gsfc.nasa.gov>) outlined the low biological activities occurring in the eastern sector.

We used the A_T - C_T relationship to further confirm the scarce role of biological activities in determining the distribution

and variability of carbonate system species in the eastern Ross Sea surface waters. In fact, photosynthesis and respiration can significantly influence C_T while A_T is only slightly influenced by photosynthesis (Zeebe and Wolf-Gladrow, 2001). When considering the whole ASSW dataset, A_T N variability was smaller ($34 \mu\text{mol kg}^{-1}$) than C_T N variability ($63 \mu\text{mol kg}^{-1}$), but their low correlation ($n = 39$, $r = 0.6$, $p = 0.01$) confirmed the general low role of the biological processes. Consistent with this observation, the O_2 saturation (i.e., the ratio of the measured concentration to the concentration at the equilibrium, which is a function of temperature and salinity) was $\approx 100\%$ or lower, supporting the hypothesis that low photosynthetic activity was occurring during the ESTRO survey. Surface waters of the investigated area were supersaturated with respect to $CaCO_3$, but the range of both Ω_{Ca} and Ω_{Ar} even if comparable to those of Mattsdotter Björk et al. (2014) for the Amundsen Sea was lower than that reported for the western Ross Sea in those areas characterized by high rates of primary production during the summer (Dejong et al., 2015; Rivarolo et al., 2019). We found a little effect of phytoplankton activity on carbonate system chemistry at S1 and along the RIS at stations 56 and 58 only, as also highlighted by the PCA score plot (Cf. Figure 3B). Thus, our observations highlight differences between the eastern and western Ross Sea, where distribution of the carbonate system parameters was largely controlled by phytoplankton activity.

The combination of low-speed winds and low productivity resulted in low CO_2 flux rates into surface waters. The eastern Ross Sea seems a low atmospheric CO_2 sink area probably due to the sea ice that prevented air-sea CO_2 exchange during the winter coupled



with low wind speeds and high surface water pCO_2 . We found a lower averaged flux in the eastern sector than along RIS (-1.4 ± 2.7 $\text{mmol m}^{-2} \text{d}^{-1}$ and -4.5 ± 2.6 $\text{C mmol m}^{-2} \text{d}^{-1}$, respectively), that agrees with mean CO_2 fluxes for January estimated by Dejong and Dunbar (2017).

The estimated fluxes in the eastern Ross Sea were particularly lower than those found in previous surveys at TNB polynya (-12.7 – 15.4 $\text{mmol C m}^{-2} \text{d}^{-1}$, Rivaro et al., 2014; Rivaro et al., 2017). Although TNB is smaller than the Ross Sea polynya, it is characterized by high rates of primary production during the summer (Tremblay and Smith, 2007; Mangoni et al.,

2017; Bolinesi et al., 2020) and a high air–sea CO_2 disequilibrium is frequently observed in summer, with the sea surface undersaturated with respect to atmospheric CO_2 corresponding to high measured chlorophyll-*a* concentration (Dejong and Dunbar, 2017; Rivaro et al., 2017; Rivaro et al., 2019). Moreover, the high katabatic winds speed blowing in TNB polynya, and the absence of sea ice facilitates gas exchange with the atmosphere.

The higher mean flux observed along the RIS section can be ascribed to both the higher air–sea CO_2 disequilibrium and wind speed than the eastern sector. However, the effect of biological

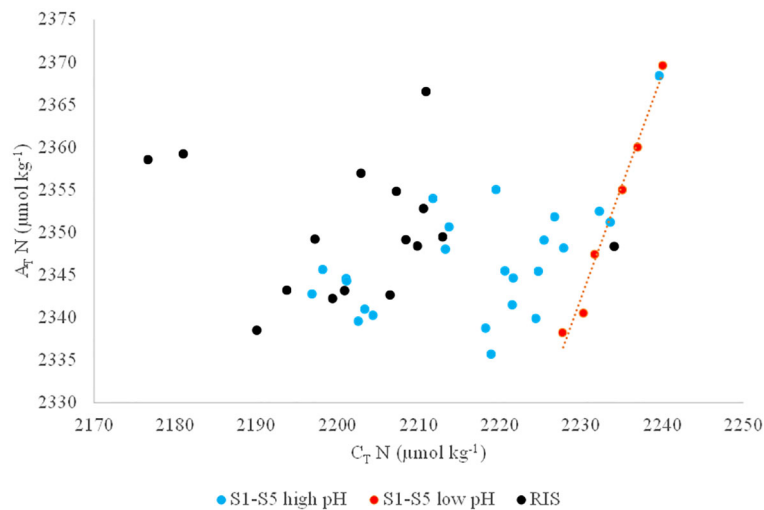


FIGURE 6

Normalized total alkalinity ($A_T N$) as a function of normalized total inorganic carbon ($C_T N$). Black dots (RIS) refer to samples taken along the RIS, blue dots (S1-S5 high pH) refer to samples taken in sections S1-S5 having a pH higher than 8.05 and pCO_2 lower than $398 \mu atm$ and red dots (S1-S5 low pH) refer to samples taken in sections S1-S5 having a pH lower than 8.05 and pCO_2 higher than $398 \mu atm$.

drawdown of CO_2 seems to be low even in this area, except at stations 56 and 58, where the biological carbon uptake could have reduced seawater pCO_2 and C_T , favoring CO_2 sinks and increased Ω_{Ar} and Ω_{Ca} values.

Carbonate system properties as chemical tracers of the main water masses and of the intrusion of Amundsen Sea waters

The PCA loading plot showed that the chemical properties in samples collected below 100 m were not uniformly distributed; thus, we focused on the relationships linking the distribution of the measured carbonate system properties (i.e. A_T and pH) to those of S and θ . The A_T significantly and positively correlated with both S (Spearman, $n = 122$, $r = 0.892$, $p = 0$) and θ (Spearman, $n = 122$, $r = 0.590$, $p = 0$). pH significantly and negatively correlated with both S (Spearman, $n = 122$, $r = -0.650$, $p = 0$) and θ (Spearman, $n = 122$, $r = -0.653$, $p = 0$).

Based on these observations, we can therefore assume that both A_T and pH had conservative behavior and could be used as tracers to chemically identify water masses.

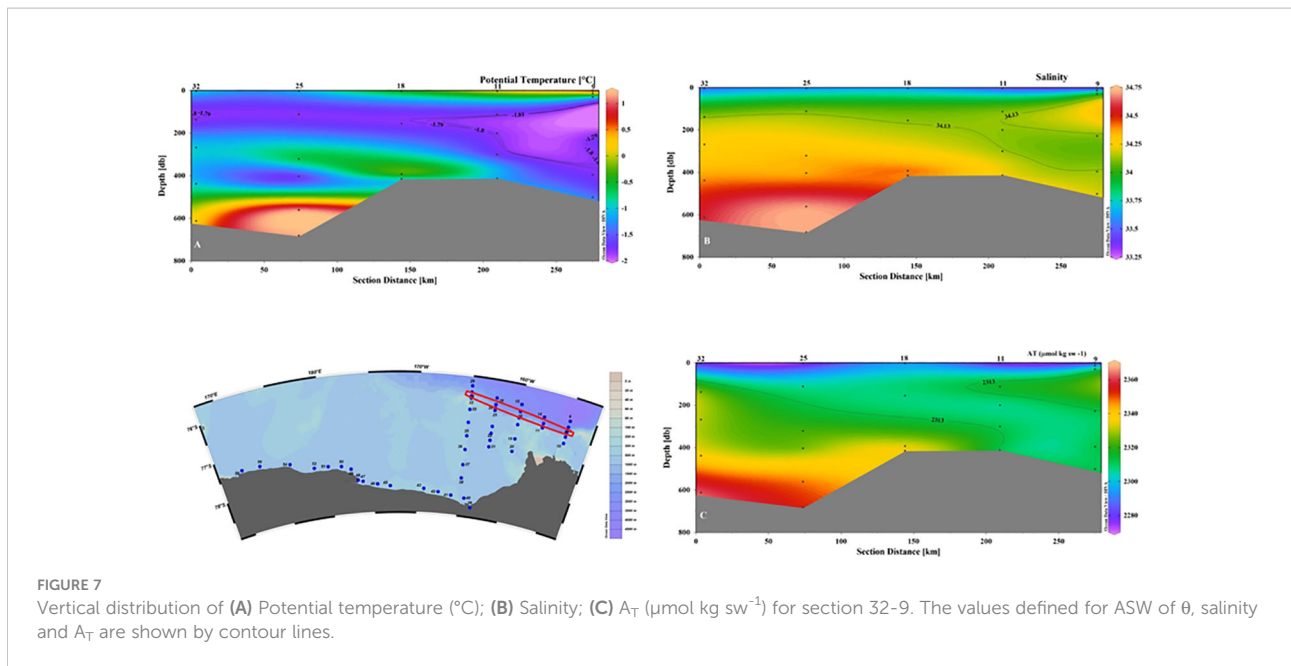
The eastern sector is the entrance gate of glacial melt water coming from the Amundsen and Bellingshausen Sea that alters the freshwater budget of the downstream Ross Sea (Rignot et al., 2013; Shepherd et al., 2018). The physical and chemical properties of the ASW (indicated as Winter Water in the Amundsen Sea) were defined as a result of the activities of the Amundsen Sea Polynya International Research Expedition

(ASPIRE) project (Yager et al., 2012; Randall-Goodwin et al., 2015).

An average value of $2215 \pm 5 \mu mol kg^{-1}$ for C_T and of $2295 \pm 5 \mu mol kg^{-1}$ for A_T were defined (Yager et al., 2016). Our data seem to suggest a limited effect of mixing between water masses and circulation in general in changing these chemical properties on a relatively short spatial scale, as it can be seen by comparing the distribution of salinity and A_T in sections 9–32 (Figure 7). In fact, the properties of the intruding ASW remain almost unchanged for tens of kilometers before the mixing of ASW with the shelf's resident waters takes place. Even considering the average pH values, we can assume that the ASW input does not appreciably affect the distribution of the carbonate system in the entire eastern sector of the Ross Sea.

The contribution of the CDW could be important in determining a change in the carbonate equilibrium system of the Ross Sea since it is a more acidic water mass than the shelf waters. It has been predicted that the Southern Ocean could become under-saturated with aragonite by 2050, with a lysocline at 730 m (Orr et al., 2005; Mcneil et al., 2010). The transport of CDW on the continental shelf, although sporadic in time and space, could accelerate the ocean acidification of Ross Sea relative to these predictions.

The low pH measured in the CDW is dependent on the less recent ventilation time compared to recently formed shelf waters (Rivarolo et al., 2015). Prolonged processes of remineralization of organic matter cause an increase in pCO_2 and in C_T as well, decreasing O_2 concentration, pH, and Ω . CDW is the only water mass found in this study with $\Omega_{Ar} < 1$. The signal of CDW is very



clear on the slope, traced by both θ and S values, as well as by chemical parameters (Figure 8). The core CDW data are comparable to those collected in the western Ross Sea (Rivaró et al., 2014), but they are higher than the observations by Yager et al. (2016) in the Amundsen Sea Polynya where the CDW did not exhibit the undiluted characteristics found during the ESTRO survey.

However, compared to what was observed in the western Ross Sea (Budillon et al., 2011; Castagno et al., 2017), the circumpolar waters remain confined on the slope, without intruding to the shelf and changing to Modified Circumpolar Deep Water (MCDW). Iron budgets constructed for the Ross Sea showed CDW accounts for most of the nutrients and of iron that supports primary production (McGillicuddy et al., 2015; Castagno et al., 2017). Other studies showed that MCDW, which is cooler and saltier than CDW, provides dissolved iron that is important for supporting phytoplankton blooms (Kustka et al., 2015) and biological CO_2 drawdown in the western Ross Sea.

We can therefore assume that the non-intrusion of CDW in the shelf area reduces biological activities by not providing nutrients and micronutrients. This would also explain the low levels of chlorophyll a that we observed in the AASW. However, the non-intrusion of CDW into the shelf area would reduce the risk of accelerated acidification of intermediate waters in the eastern Ross Sea.

In contrast to the other water masses, an increasing longitudinal gradient of mean A_T , pH, and Ω_{Ar} values was found for the LSSW. This could be due to local mixing phenomena, which, however, cannot be confirmed from the available data. LSSW is mainly produced in the eastern sector of the Ross Sea where the absence of polynyas does not allow

for the high salinities observed in the western sector. Several mechanisms of LSSW formation have been proposed but, to date, due to the limited number of measurements available, there is no fully agreed theory (Orsi and Wiederwohl, 2009). However, it is believed, based on climatology, that the volume occupied by LSSW in the shelf area is smaller than that occupied by HSSW, and that 90% of this volume is confined to the sector investigated by the ESTRO project (Orsi and Wiederwohl, 2009).

The section sampled along RIS (Figure 9) can be considered the link between the eastern and western sectors of the southern Ross Sea. A sharp horizontal salinity gradient along RIS was inferred within the shelf waters near 175°E , which led to their classification into HSSW and LSSW.

Also in this section, as might be expected from the positive correlation and the values attributed to the water masses, the A_T proved to be a good conservative chemical tracer, appearing the same as the salinity of HSSW and LSSW. However, it proved less effective in tracing the presence of the ISW core. Smethie and Jacobs (2005) suggested that the source water that melts the ice and produces ISW is likely to be a mixture of HSSW and LSSW. The proportion of contributing water masses can vary, and there is therefore no fixed make-up that can be applied. Thus, depending on the percentage with which each end member participates in the formation of the ISW, there may be variations in the chemical signature of the ISW that the sensitivity of the A_T is not able to highlight. This limitation is even greater in the case of pH, since any differences in the oxidative degradation processes of the organic substances and the ventilation times must also be considered in addition to the considerations made above.

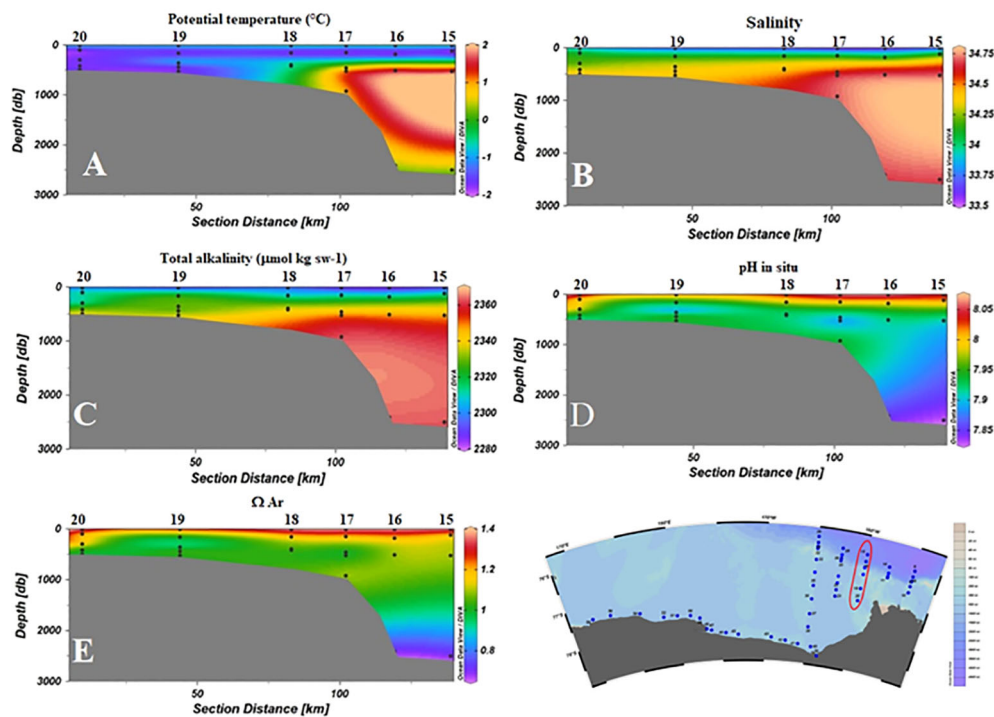


FIGURE 8

Distribution of physical and chemical parameters as a function of depth for S3 (stations 15–20). (A) Potential temperature (°C); (B) Salinity; (C) A_T ($\mu\text{mol kg sw}^{-1}$); (D) pH *in situ*; (E) Ω_{Ar} .

Anthropogenic carbon in the eastern Ross Sea sector

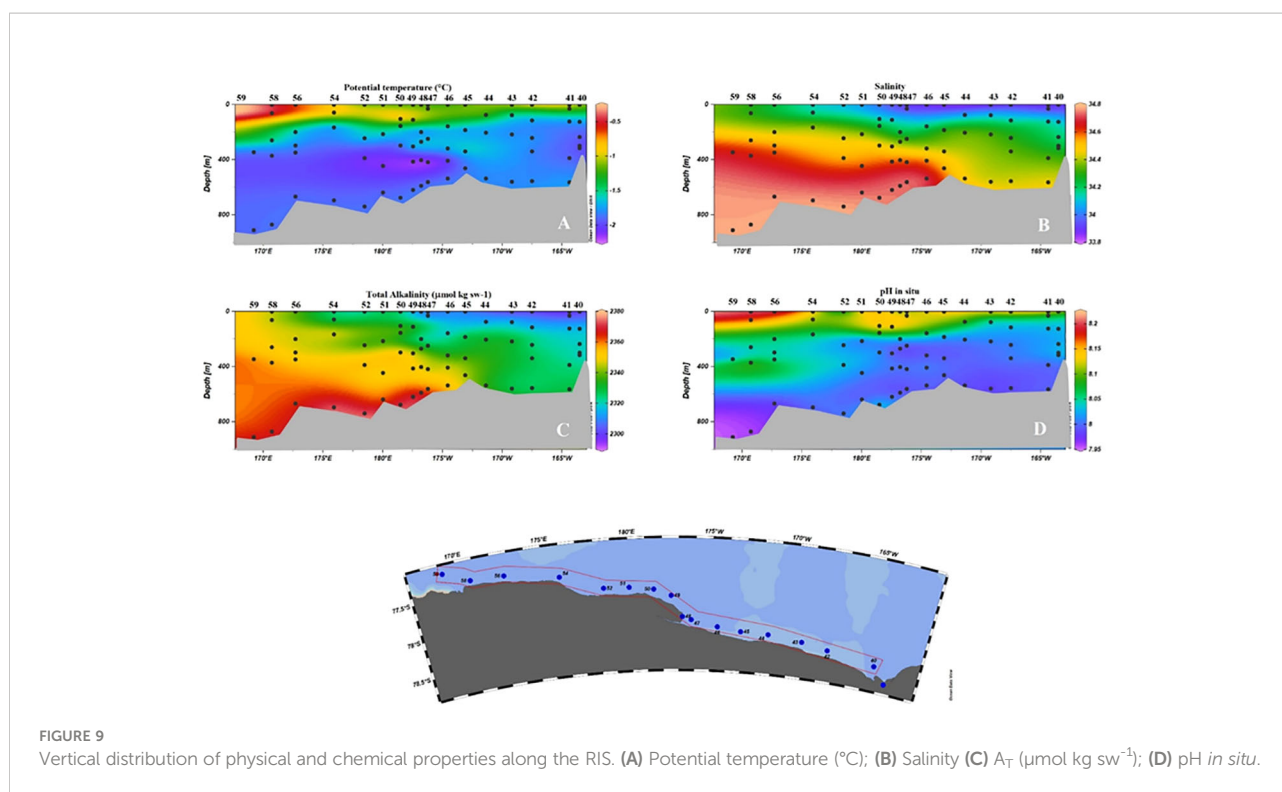
Estimating the distribution of C_{ant} is important to better understand the carbon cycle, also considering the effect of atmospheric CO_2 uptake on OA. The chemical buffering of C_{ant} is the quantitatively most important oceanic process acting as a carbon sink. RF is a measure of the ocean's buffer capacity for the carbonate system in terms of pH changes also in seawater or freshwater. The higher the RF, the lower the ocean's buffer capacity, and the faster the change of pCO_2 in the ocean at a given C_T change (Jiang et al., 2019).

At the surface we found RF values greater than 15 in surface waters, which suggests a low efficient C_{ant} uptake ability. Alkalinity (i.e. the excess of bases), governs the efficiency at which this occurs and provides buffering capacity toward acidification. Thus, the low measured A_T values justify the low buffer ability of the surface waters. Biological processes could contribute to the C_{ant} sequestration in surface waters too. However, our data show that the biological role was moderate, underscoring the importance of chemical and physical drivers in carbon regulation in this region of the Ross Sea.

Different approaches have been proposed to evaluate the C_{ant} in intermediate and deep waters (Sabine and Tanhua, 2010; Pardo et al., 2014). The choice of using the TrOCA method in our study is due to the following reasons: relative simplicity of application, good agreement of the results with those obtained with other more complex methods using, e.g., CFC measurements (Matear et al., 2003), and its application that has already been tested in the western Ross Sea (Sandrini et al., 2007).

ASW and LSSW were enriched in C_{ant} suggesting a recent ventilation of these water masses. Given the calculated RF for AASW, the pCO_2 , and the mean C_T (409 μatm , and 2167 $\mu\text{mol kg sw}^{-1}$, respectively), a change in C_T on the order of 46 $\mu\text{mol kg sw}^{-1}$, between the preindustrial time and the time of sampling, was expected. Thus, our data (54 $\mu\text{mol kg sw}^{-1}$) are slightly higher than expected, but the TrOCA method can lead to some overestimates of C_{ant} in deep and bottom waters (Pardo et al., 2014).

Since our data are the first C_{ant} estimates in this area of the Ross Sea, we cannot make any comparison on trends derived from data sets collected several years apart. There will be an opportunity to verify our results and to get an indication of the



ventilation times of the water masses since the CFCs were sampled as part of the ESTRO project.

We have, however, compared our estimates in the HSSW and in the CDW with those collected in 2001 by the CLIMA project in the western Ross Sea (Sandrini et al., 2007). A change in C_T in the order of $14 \mu\text{mol kg sw}^{-1}$, between 2001 and 2020, was calculated. This increase was observed in the CDW (from 6 to $21 \mu\text{mol kg sw}^{-1}$), whereas the C_{ant} in HSSW resulted higher than expected (65 instead of $44 \mu\text{mol kg sw}^{-1}$). Apart from differences associated with variations of the used stoichiometric ratios, the explanation for a high C_{ant} value could be related to an HSSW salinity rebound. HSSW salinity decreased between 1995 and 2014 (consistent with freshening observed between 1958 and 2008) and it rebounded sharply after 2014 (Castagno et al., 2019). This shift to saltier HSSW may have caused increased ventilation and a higher C_{ant} sequestration rate. Thus, our data hint that the production of HSSW could be a more efficient mechanism for the transfer of anthropogenic CO_2 from the atmosphere to the deep ocean than in the past decades.

Conclusions

Our measurements represent the first detailed contribution on a large spatial scale of carbonate chemistry in the eastern sector of the Ross Sea.

A_T resulted in the carbonate system parameter that mixes conservatively and was well-correlated with salinity in surface waters. A_T distribution traced a freshwater mass coming from West Antarctica near Cape Colbeck, which moved westward along the RIS edge becoming more saline and with a higher A_T .

In contrast to A_T , the other parameters of the carbonate system revealed a more varying distribution associated with sea ice melting and primary production, organic matter respiration, and the formation or dissolution of calcium carbonate. Nevertheless, their distribution appeared more largely controlled by physical properties together with the circulation pattern than by biological activities, underlining the difference in factors affecting the surface variability in the eastern and western Ross Sea.

A_T and pH resulted in conservative chemical properties below the surface waters, allowing the chemical tracing of the input of the Amundsen Sea Water and suggesting a limited effect of mixing between water masses. The study of the mixing evolution processes is mandatory to estimate also the volume reaching the Ross Sea shelf area. Moreover, A_T and pH displayed an increasing longitudinal gradient in the LSSW, which needs further investigation to be fully understood.

Both physical and chemical data confirmed that CDW remains confined off the shelf, thus reducing the risk of accelerated acidification in the eastern Ross Sea, but at the same time not supplying the shelf water with nutrients and trace elements.

Shelf waters in the eastern Ross Sea were enriched in C_{ant} , which resulted lower than the estimated value for HSSW

produced in western Ross sea. The absence of polynya in the eastern Ross Sea may reduce the ability of this sector to transfer C_{ant} from the surface to the deep waters compared to the western one. More extensive observations are needed to study the CO_2 penetration and the air–sea CO_2 exchange, considering that a change in hydrological conditions driven by climate variability could affect the C_{ant} invasion throughout the water column.

Data availability statement

The raw data supporting the conclusions of this article will be made available by the authors, without undue reservation.

Author contributions

EZ, PF, and PR designed the study. PR, DV, PC, and PF carried out fieldwork. DV, PR, PC, PF, EZ, and CI contributed to the laboratory measurements, data analysis, and interpreted the data. DV performed the statistical analysis. PR and DV wrote the manuscript with input and approval from all co-authors. All authors contributed to the article and approved the submitted version.

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Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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