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Nitrogen mineralization and immobilization in surface sediments of coastal reclaimed aquaculture ecosystems

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Sediment nitrogen (N) mineralization and immobilization are two crucial processes driven by microorganisms, which may play significant roles in the regulation of water quality in aquaculture ecosystems. However, limited information is available about the quantitative importance of sedimentary N mineralization and immobilization in coastal aquaculture systems. Here, a combination of incubation experiments with a ¹⁵N isotope dilution technique were employed, aiming to quantify N mineralization and immobilization processes in surface sediments (0–5 cm) of three types of aquaculture ecosystems (seabass, white shrimp, and green crab ponds) reclaimed within the western bank of the Pearl River Estuary. Our results showed that no significant difference in sediment N mineralization and immobilization rates, microbial abundances, and organic matter among different aquaculture types on small-scale range. Meanwhile, prolonged pond-drying significantly reduced sediment N mineralization and immobilization rates, bacterial abundances, organic matter, moisture content, ferrous ion (Fe²⁺), Fe²⁺/Fe³⁺, and ammonium (NH₄⁺), while not strongly altered sediment percentage of NH₄⁺ mineralized per day (PAM), relative ammonium immobilization (RAI), fungal abundances, TOC/TN, nitrate (NO₃⁻), and δ¹³C_{org}. N mineralization and immobilization rates were both significantly related to overlying water NO₃⁻, as well as sediment moisture content, bulk density, organic matter, Fe²⁺, and microbial abundances. In addition, the total mineralized and immobilized N in aquaculture surface sediments from the Guangdong-Hong Kong-Macao Greater Bay Area were estimated to be approximately 4.55×10⁴ and 3.68×10⁴ t N yr⁻¹, respectively. Higher N mineralization relative to N immobilized fluxes indicated that the sediment serves as an important source of eutrophication in reclaimed aquaculture system of coastal wetlands.

KEYWORDS

N mineralization and immobilization, available organic carbon fractions, sediment, reclaimed aquaculture ecosystem, ¹⁵N isotope dilution technique

1 Introduction

Since the last century, anthropogenic activities have already become one of the major factors driving the change of global nitrogen (N) cycling and caused more than tripled N-flows higher than those caused by natural processes, and thus resulting in a total globally fixed N of about 413 Tg N y^{-1} (Zilio et al., 2020). The increasing reactive N load disrupts the N balance in both marine and terrestrial ecosystems. Meanwhile, a series of ecological problems, such as the loss of habitats and biodiversity (Galloway et al., 2008), eutrophication and the expansion of periodic or permanent low oxygen zone (Diaz and Rosenberg, 2008), the outbreak of toxic algae (Li et al., 2014), and the strengthened emission of greenhouse gases (Murray et al., 2015; Mao et al., 2022) were caused in the latest century. Therefore, coastal N pollution is one of the global significant and urgent environmental problem. Studies regarding N biogeochemical processes of estuarine and coastal ecosystems have become a cutting-edge scientific issue, which is also a key topic of many international research programs such as International Geosphere-Biosphere Programme/Land-Ocean Interactions in Coastal Zone (IGBP/LOICZ), Integrated Marine Biogeochemistry and Ecosystem Research (IMBER) and Future Earth (Howarth et al., 2011; Buzzelli et al., 2013).

N mineralization and immobilization processes of N biogeochemical cycle in estuarine and coastal sediments. N mineralization is an important N-cycling process by which organic N is converted into inorganic forms available to organisms, which can exacerbate water eutrophication (Mishra et al., 2005). Conversely, microbial N immobilization is an important N-cycling process in which microorganisms convert inorganic N into organic N (e.g., amino acid and proteins) (Zhu et al., 2013). Meanwhile, some previous studies indicated that ammonium (NH_4^+) immobilization is considered as the major process of anaerobic ammonia consumption, which can effectively transform NH_4^+ into organic N and maintain the health of ecosystem (Matheson et al., 2003; Huang et al., 2021; Yang et al., 2022). These two important N biogeochemical processes have important ecological and environmental significance for maintaining N balance in estuarine and coastal ecosystems. Previous studies have reported that sediment N mineralization and immobilization are mainly affected by the physicochemical properties of sediment (e.g., soil texture, total carbon and N, C/N, pH, and water content) (Paul et al., 2003; Rutigliano et al., 2009; Huang et al., 2021), environmental climate factors (e.g., temperature, precipitation) (Gao et al., 2014; Lin et al., 2016a), and microorganisms in complex estuarine and coastal ecosystems (Cufrey and Kemp, 1992; Herbert, 1999; Bai et al., 2012; Qi et al., 2019). Thus, it is necessary to fully understand these two processes and their controlling factors for evaluation of N balance and prediction

of N dynamics with the ever-changing environmental conditions in estuaries and coasts.

A large area of coastal wetlands has been reclaimed for development over the past few decades, and aquaculture use is one of the major reclamation purposes. In China, more than 1,260 km² of coastal wetlands was reclaimed for aquaculture every year (Cao et al., 2011; Zuo et al., 2013). In order to meet the unprecedented demand of aquatic products supply, large amount of nitrogen- and phosphorus-rich bait is added into these ecosystems. When the nutrients in the water column exceed the requirement of plankton and aquacultured organisms, several disadvantages including water quality deterioration, frequent disease occurrence, and aquaculture benefits decline appeared (Amano et al., 2011). Coastal aquaculture ecosystem is generally regarded as a major source of N pollutants, and subsequent environmental problems have attracted widespread concerns nowadays (Wu et al., 2014; Lin and Lin, 2022). Sediment is a vital place where N mineralization and immobilization occurred, representing frequent transformation between inorganic N and organic N, which could aggravate eutrophication to some extent. Coastal wetlands in the Pearl River Estuary are reclamation hotspots for aquaculture and have been significantly affected by anthropogenic activities. Investigation of N mineralization and immobilization around this area could reflect both regional and typical characteristics. In this study, reclaimed brackish aquaculture ecosystems located on the west river bank of the Pearl River Estuary were selected, and a combination of sediment incubation experiment and ¹⁵N stable isotope dilution method was employed. The main objectives of this study are to: (1) quantify sediment N mineralization and immobilization rates; (2) compare the effects of different cultured species (seabass, white shrimp, and green crab) on sediment N mineralization and immobilization, and identify the key controlling factors of these two processes; (3) preliminarily estimate sediment N mineralization and immobilization fluxes in the Guangdong-Hong Kong-Macao Greater Bay Area, and provide scientific basis for ecological effect evaluation as well as policy formulation and management in the area.

2 Materials and method

2.1 Study area and sampling

The Pearl River Delta (PRD, 112.9–114.4°E, 21.82–23.25°N) located on southern China covers an area of ~54,754 km² and is surrounded by several highly populated cities such as Hong Kong, Guangzhou, Macau, and Shenzhen with a total population of over 78.61 million people in 2021 (Li et al., 2000; Liu et al., 2018). It is plain river network area with the slope of 0.1–0.2‰

and the density of river network of $\sim 0.8 \text{ km km}^{-2}$, which is characterized by subtropical humid climate with an annual average temperature of 21.8°C and an annual average rainfall of 1747.4 mm (Wang et al., 2012). A small irregular semidiurnal tide dominates this area with the average of 0.86 to 1.6 m and maximum of 2.29 to 3.36 m (Wang et al., 2012). The shallow aquaculture pools that were created by the removal of original marsh vegetation (mangrove). The PRD is experiencing massive industrialization and urbanization nowadays and makes great contributions to China's economy. Take the year of 2021 as an example, this area provided $\sim 12.3\%$ of the national gross domestic product ($\$1406.18$ billion) according to the China Statistical Yearbook 2022. Meanwhile, the coastal habitats of the PRD have been highly productive ecosystems, being an extremely important aquaculture area of China (Zhou et al., 2019). Coastal wetlands reclamation for aquaculture pond took the largest proportion of the total conversion from natural wetlands (estuarine water, mangrove forest and salt marsh) towards constructed wetlands in this region. For example, the breeding area of the PRD increased about threefold from 256.01 km^2 in 1980 to 826.72 km^2 in 2015 (Zhou et al., 2019). With rapid economic growth and urbanization of the PRD, local environment has suffered from serious environmental problems, especially the high N loading, which has great effects on water quality and N cycling processes (Dai et al., 2008; Li et al., 2013; Wu et al., 2020).

Our samples were selected from three separate coastal ponds aquaculturing different economic species on the western bank of the Pearl River Estuary, one for fish (*Perca fluviatilis*), one for shrimp (*Penaeus vannamei*), and one for crab (*Scylla serrata*) (Figure 1). The sediments were collected using a core cylinder, a PVC pipe handle and a one-way valve from these sites in winter (January 19, 2019) and summer (July 12, 2019). At each site, triplicate surface sediments ($0\text{--}5 \text{ cm}$) were taken from the cores, and sealed with air-tight, acid-cleaned plastic bags. Overlying water samples of each site were also collected in polyethylene bottles and filtered through $0.2 \mu\text{m}$ filters (Millipore, Bedford, United States). All the samples were transported to the laboratory on ice within 8 h. In the laboratory, filter samples were immediately stored at -20°C . Sediments of each sampling site were mixed thoroughly under a helium condition, and subsequently divided into three parts. One part was stored at 4°C for measuring microbial biomass carbon (MBC) and N-cycling rates, the second part was frozen at -20°C for determination of physicochemical parameters, and the third part was preserved at -80°C for molecular analysis.

2.2 Determination of physicochemical properties

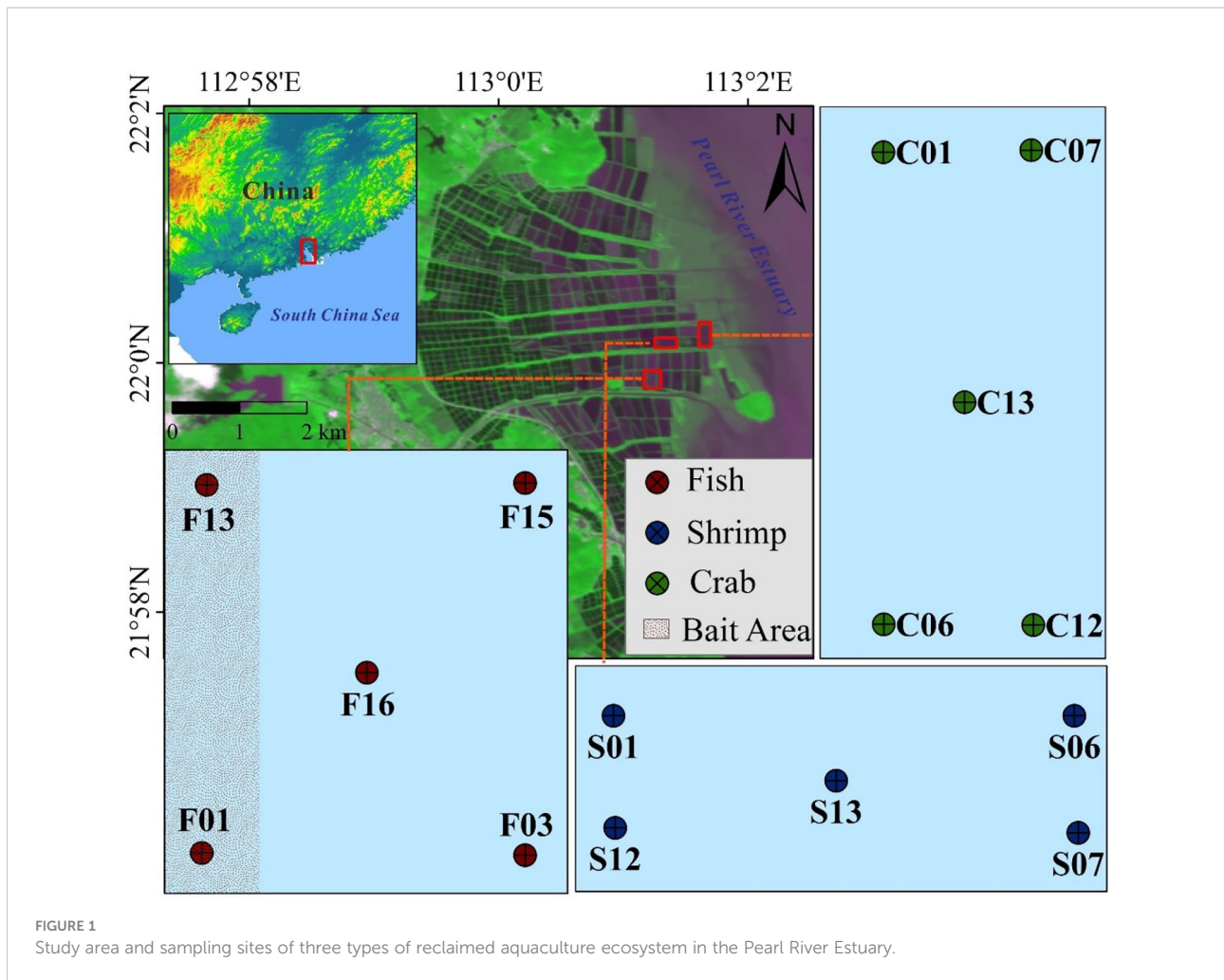
Sediment water content was determined according to the weight loss from fresh sediment after freeze-drying until

constant weight. Sediment pH was determined by measuring the pH of the mixture of fresh sediment and Milli-Q (MQ) water (1:2.5 in v/v) (Zhang et al., 2015). Sediment NH_4^+ , NO_2^- , and NO_3^- were extracted by 2 M KCl purged with N_2 for 30 min and measured using 2 M KCl as standard curve substrate (Hou et al., 2013). Total extractable iron (Fe) and ferrous oxide (Fe^{2+}) were extracted by 0.5 M HCl and 0.25 M hydroxylamine hydrochloride (both purged with N_2 for ~ 15 min) from fresh sediments, and the concentrations were determined using ferrozine-based colorimetric method using the spectrophotometer. The concentrations of ferric iron (Fe^{3+}) were calculated by subtracting Fe^{2+} from Fe (Lovley and Phillips, 1987; Hu et al., 2022). Sediment texture was determined by laser particle size analyzer LS13 320 (Hou et al., 2013). After effective leaching of carbonates with 1 M HCl , sediment total organic carbon (TOC) and total nitrogen (TN) were measured on an elemental analyzer (Vario EL, Elementar, Germany) (Zhang et al., 2015; Wu et al., 2021). Sediment stable organic carbon isotopic composition ($\delta^{13}\text{C}_{\text{org}}$) were analyzed on a Thermo MAT 253Plu isotope ratio mass spectrometer. After oxidation with $333 \text{ mmol L}^{-1} \text{ KMnO}_4$, sediment easily oxidized organic carbon (EOC) was measured on a spectrophotometer colorimetry (Vieira et al., 2007). Sediment dissolved organic carbon (DOC) was extracted with MQ water and determined with a Shimadzu TOC-TN analyzer (Shimadzu Corp., Kyoto, Japan). Sediment MBC were extracted with K_2SO_4 and determined through chloroform fumigation-extraction method (Vance et al., 1987; Beck et al., 1997).

2.3 Measurements of N transformation rates

Gross N mineralization (GNM) and NH_4^+ immobilization (GAI) rates were determined using ^{15}N isotope dilution technique (Kirkham and Bartholomew, 1954). Briefly, twelve centrifuge tubes (50 mL) were prepared for each sample, and 5 g fresh sediment was weighed into each centrifuge tube. For determination of the GNM and GAI rates, six centrifuge tubes with a mixture of sediment and saltwater (1:5 in w/w) were vortexed thoroughly. Tubes were sealed and preincubated for 24 h under *in situ* temperature in dark on a shaker table (150 rpm). After the preincubation, $^{15}\text{NH}_4^+$ (99 atom%, $^{15}\text{NH}_4\text{Cl}$) was added into the tubes, the final concentration of ^{15}N was about $2 \mu\text{g g}^{-1}$ and the final percentage of ^{15}N was about 10–15%. Triplicate initial samples were terminated by adding 1 mL 50% Zinc chloride (ZnCl_2) solution and immediately frozen at -20°C , and triplicates incubation samples were sealed and were put back to the incubator. After 24-hour incubation, final samples were terminated and immediately frozen at -20°C .

$$\text{GNM} = \frac{M_i - M_f}{t} \times \frac{\log(H_i M_f / H_f M_i)}{\log(M_i / M_f)} \quad (1)$$



2.4 Microbial analysis

Sediment DNA was extracted by the FastDNA spin kit for sediment (MP Biomedical, United States) based on the manufacturer's instructions. The concentration and purity of extracted DNA were determined by NanoDrop spectrophotometer (NanoDrop 2000C, Thermo Scientific, United States), and the fragments size and quality were evaluated by 1.0% agarose gel electrophoresis. Quantitative PCR (qPCR) assays were used to measure microbial gene abundances on an ABI 7500 Fast real-time qPCR system (Applied Biosystems, United States). For bacterial 16S rRNA gene and fungal ITS gene, primer pairs 341F/519R (Bachar et al., 2010) and SSU081 and 1196R were used (Rousk et al., 2010), respectively. The standard curves for these two genes were created using a 10-fold dilution series (10^2 – 10^9 copies) of the

$$GAI = \frac{M_i - M_f}{t} \times \frac{\log(H_i/H_f)}{\log(M_i/M_f)} \quad (2)$$

M_i and M_f ($\mu\text{g N g}^{-1}$ dry weight) are total NH_4^+ concentrations of initial (before incubation) and final (after incubation) samples; H_i and H_f ($\mu\text{g N g}^{-1}$ dry weight) are $^{15}\text{NH}_4^+$ concentrations of initial and final samples; t (d) is the incubation time (24 h).

The relative NH_4^+ immobilization (RAI) is calculated as the ratio of GAI to GNM rates, and RAI value of ≥ 1 indicates a N-limited sediment, whereas a value of approximately 0.5 indicates N saturation (Aber, 1992). The percentage of NH_4^+ mineralized per day (PAM%) was referred to as the rates of GNM divided by sediment N contents, which can indicate the sediment available N by the internal N cycle.

standard plasmids DNA. Each sample was measured in triplicates.

2.5 Calculations and statistical analyses

Statistical analysis was performed using SPSS 19.0 (SPSS Inc., United States). All variables were tested for normality by Shapiro-Wilk test, and the non-normality variable were normalized by using Blom's formula. Statistically significant difference of N transformation rates and sediment physicochemical properties between three ponds were determined by one-way ANOVA, Turkey test ($p < 0.05$, equal variances assumed) or Dunnett's test ($p < 0.05$, equal variances not assumed). The correlations were analyzed by and linear regression analyses and Pearson test (two-tailed, $p < 0.05$). The graphs were drawn by ArcGIS 10.2 (ArcMap 10.2, ESRI, United States) and Origin 2019 (OriginLab Corporation, United States).

3 Results

3.1 Physicochemical properties of overlying water

Physicochemical properties of overlying water in the three types of reclaimed aquaculture ponds are shown in Table 1 and Table S1. During winter sampling, crab pond was in the early period of pond-drying, and only a litter overlying water was remaining in the crab

ponds, while the shrimp pond was in the late drying period, and the surface sediments were dried and cracked. Therefore, overlying water of the shrimp and crab ponds in winter was not sampled because of that those two ponds were in drying and dredging period. One-way ANOVA showed that overlying water salinity, pH, NH_4^+ , NO_3^- , and NO_2^- of the fish pond showed significant seasonal differences ($p < 0.05$), while no significant seasonal difference was found for the dissolved oxygen (DO) ($p > 0.05$). The salinity and pH in winter was higher than those in summer. Influenced by aquaculture activities, NH_4^+ , NO_2^- , and NO_3^- were all significantly higher in summer than those in winter ($p < 0.05$). In summer, both DO and salinity showed significant differences among three ponds ($p < 0.05$ for all). The highest DO was observed in the shrimp ponds ($10.30 \pm 1.05 \text{ mg L}^{-1}$) followed by the crab ponds ($7.90 \pm 0.52 \text{ mg L}^{-1}$) and fish ponds ($5.11 \pm 0.14 \text{ mg L}^{-1}$), which may potentially result from the most operating waterwheel-type aerators in the shrimp and crab ponds. Consistently, the highest salinity was measured in the crab ponds ($2.85 \pm 0.00 \text{ ‰}$) followed by the crab ponds ($2.25 \pm 0.18 \text{ ‰}$) and fish ponds ($0.9 \pm 0.04 \text{ ‰}$). The overlying water quality of these ponds were relatively well controlled. Overlying water NH_4^+ and NO_2^- of the fish ponds were both significantly higher than those in shrimp and crab ponds ($p < 0.05$).

3.2 Physicochemical properties of sediments

Physicochemical properties of surface sediments in the reclaimed aquaculture ponds are shown in Table 2 and Table S2.

TABLE 1 Physicochemical parameters of overlying water in different reclaimed aquaculture ponds.

Physicochemical parameters		Fish		Shrimp	Crab
		Summer	Winter	Summer	Summer
DO (mg L^{-1})	Range	4.43–5.25	4.57–5.05	9.34–11.18	7.72–8.55
	Mean \pm SD	$4.93 \pm 0.32^{\text{Ac}}$	$4.79 \pm 0.23^{\text{A}}$	$10.22 \pm 0.65^{\text{a}}$	$8.13 \pm 0.30^{\text{b}}$
Salinity (‰)	Range	0.84–0.94	1.53–1.56	2.01–2.57	2.85–2.85
	Mean \pm SD	$0.90 \pm 0.04^{\text{Bc}}$	$1.55 \pm 0.01^{\text{A}}$	$2.25 \pm 0.20^{\text{b}}$	$2.85 \pm 0.00^{\text{a}}$
pH	Range	8.13–8.18	8.37–8.65	8.93–9.24	8.85–8.90
	Mean \pm SD	$8.15 \pm 0.02^{\text{Bc}}$	$8.53 \pm 0.11^{\text{A}}$	$9.10 \pm 0.12^{\text{a}}$	$8.88 \pm 0.02^{\text{b}}$
NH_4^+ (μM)	Range	5.90–6.90	0.25–1.66	0.44–1.90	0.68–3.04
	Mean \pm SD	$6.34 \pm 0.47^{\text{Aa}}$	$0.64 \pm 0.58^{\text{B}}$	$1.03 \pm 0.58^{\text{c}}$	$1.92 \pm 0.95^{\text{b}}$
NO_2^- (μM)	Range	0.59–0.64	0.20–0.27	0.04–0.09	0.02–0.09
	Mean \pm SD	$0.61 \pm 0.02^{\text{Aa}}$	$0.24 \pm 0.03^{\text{B}}$	$0.07 \pm 0.02^{\text{b}}$	$0.04 \pm 0.02^{\text{c}}$
NO_3^- (μM)	Range	36.70–39.09	1.90–4.90	34.86–43.56	34.72–41.27
	Mean \pm SD	$37.91 \pm 0.96^{\text{Aa}}$	$3.89 \pm 1.18^{\text{B}}$	$37.42 \pm 3.53^{\text{a}}$	$37.57 \pm 2.76^{\text{a}}$

Statistically significant difference between seasons ($p < 0.05$ according to the F statistics) is labeled in the different uppercase letters, and the significant difference between aquaculture ponds (Turkey multiple comparison $p < 0.05$) is labeled in the different lowercase letters.

TABLE 2 Spatiotemporal variations of sediment physicochemical properties in different reclaimed aquaculture ponds.

Physicochemical properties	Fish pond		Shrimp pond		Crab pond	
	Summer	Winter	Summer	Winter	Summer	Winter
Temperature (°C)	30.64 ± 0.18 ^{Ac}	17.5 ± 0.07 ^{Ba}	32.68 ± 0.40 ^{Aa}	17.26 ± 0.05 ^{Bc}	32.22 ± 0.15 ^{Ab}	17.4 ± 0.07 ^{Bb}
Moisture content	0.36 ± 0.06 ^{Aa}	0.36 ± 0.04 ^{Ab}	0.54 ± 0.19 ^{Aa}	0.29 ± 0.04 ^{Bc}	0.44 ± 0.08 ^{Aa}	0.43 ± 0.05 ^{Aa}
Bulk density (g mL ⁻¹)	1.63 ± 0.14 ^{Aa}	1.69 ± 0.05 ^{Ab}	1.38 ± 0.25 ^{Ba}	1.85 ± 0.15 ^{Aa}	1.51 ± 0.10 ^{Aa}	1.51 ± 0.06 ^{Ac}
NH ₄ ⁺ (μg N g ⁻¹)	12.20 ± 4.12 ^{Aa}	6.81 ± 1.09 ^{Ba}	31.65 ± 18.26 ^{Ba}	6.30 ± 4.08 ^{Ab}	21.69 ± 17.22 ^{Aa}	1.78 ± 0.39 ^{Bc}
NO ₂ ⁻ (μg N g ⁻¹)	0.04 ± 0.01 ^{Bb}	0.15 ± 0.03 ^{Aa}	0.12 ± 0.09 ^{Aa}	0.46 ± 0.64 ^{Aa}	0.04 ± 0.02 ^{Bb}	0.17 ± 0.08 ^{Aa}
NO ₃ ⁻ (μg N g ⁻¹)	3.18 ± 0.40 ^{Aa}	1.58 ± 0.17 ^{Ba}	5.10 ± 2.19 ^{Aa}	7.29 ± 12.10 ^{Aa}	4.01 ± 0.58 ^{Aa}	1.77 ± 1.04 ^{Ba}
Fe ²⁺ (mg Fe g ⁻¹)	8.56 ± 2.94 ^{Ac}	3.09 ± 0.52 ^{Ba}	19.62 ± 8.34 ^{Aa}	0.43 ± 0.28 ^{Ba}	13.04 ± 3.30 ^{Ab}	2.92 ± 0.62 ^{Ba}
Fe ³⁺ (mg Fe g ⁻¹)	5.71 ± 2.20 ^{Ab}	2.20 ± 0.97 ^{Ba}	1.37 ± 0.54 ^{Ac}	3.28 ± 1.60 ^{Aa}	6.62 ± 6.62 ^{Aa}	1.81 ± 0.34 ^{Ba}
Fe ²⁺ /Fe ³⁺	1.95 ± 1.80 ^{Ac}	1.72 ± 0.95 ^{Aa}	14.97 ± 6.62 ^{Aa}	0.13 ± 0.05 ^{Ba}	2.13 ± 0.98 ^{Ab}	1.61 ± 0.20 ^{Aa}
Median grain size (μm)	47.53 ± 11.17 ^{Aa}	24.88 ± 9.73 ^{Ba}	29.94 ± 13.00 ^{Ab}	16.29 ± 10.59 ^{Aa}	16.16 ± 4.02 ^{Ac}	18.27 ± 9.21 ^{Aa}
TOC (mg C g ⁻¹)	10.49 ± 2.10 ^{Aa}	11.03 ± 3.02 ^{Aa}	16.46 ± 7.75 ^{Aa}	12.30 ± 4.61 ^{Aa}	11.04 ± 0.80 ^{Aa}	10.05 ± 0.50 ^{Ba}
TN (mg N g ⁻¹)	1.00 ± 0.15 ^{Aa}	0.89 ± 0.17 ^{Aa}	2.15 ± 1.25 ^{Aa}	1.57 ± 0.66 ^{Aa}	1.28 ± 0.16 ^{Aa}	1.09 ± 0.09 ^{Ba}
TOC/TN	10.47 ± 0.66 ^{Aa}	12.71 ± 4.25 ^{Aa}	8.05 ± 1.16 ^{Ac}	7.93 ± 0.69 ^{Ac}	8.69 ± 0.72 ^{Ab}	9.28 ± 0.49 ^{Ab}
δ ¹³ C _{org} (‰)	-24.65 ± 0.15 ^{Aa}	-24.98 ± 0.38 ^{Ac}	-23.77 ± 0.71 ^{Aa}	-24.02 ± 0.81 ^{Aa}	-24.32 ± 0.66 ^{Aa}	-24.27 ± 0.23 ^{Ab}
EOC (mg C g ⁻¹)	4.77 ± 1.95 ^{Aa}	3.25 ± 0.80 ^{Aa}	8.19 ± 4.12 ^{Aa}	2.52 ± 0.95 ^{Ba}	5.79 ± 0.74 ^{Aa}	2.67 ± 1.35 ^{Ba}
DOC (mg C g ⁻¹)	0.10 ± 0.04 ^{Aa}	0.11 ± 0.04 ^{Aa}	0.14 ± 0.05 ^{Aa}	0.06 ± 0.01 ^{Bc}	0.11 ± 0.01 ^{Aa}	0.10 ± 0.01 ^{Ab}
MBC (mg C g ⁻¹)	0.13 ± 0.05 ^{Aa}	0.13 ± 0.03 ^{Aa}	0.18 ± 0.08 ^{Aa}	0.06 ± 0.02 ^{Bc}	0.15 ± 0.04 ^{Aa}	0.09 ± 0.01 ^{Bb}
Bacterial (×10 ⁹ copies g ⁻¹)	8.94 ± 4.18 ^{Aa}	8.31 ± 2.16 ^{Aa}	8.31 ± 2.65 ^{Aa}	3.26 ± 1.90 ^{Bc}	6.22 ± 2.60 ^{Aa}	3.80 ± 0.94 ^{Ab}
Fungal (×10 ⁷ copies g ⁻¹)	7.11 ± 4.74 ^{Aa}	5.19 ± 2.44 ^{Aa}	6.20 ± 3.07 ^{Aa}	4.21 ± 2.59 ^{Aa}	3.07 ± 1.28 ^{Aa}	3.38 ± 1.23 ^{Aa}

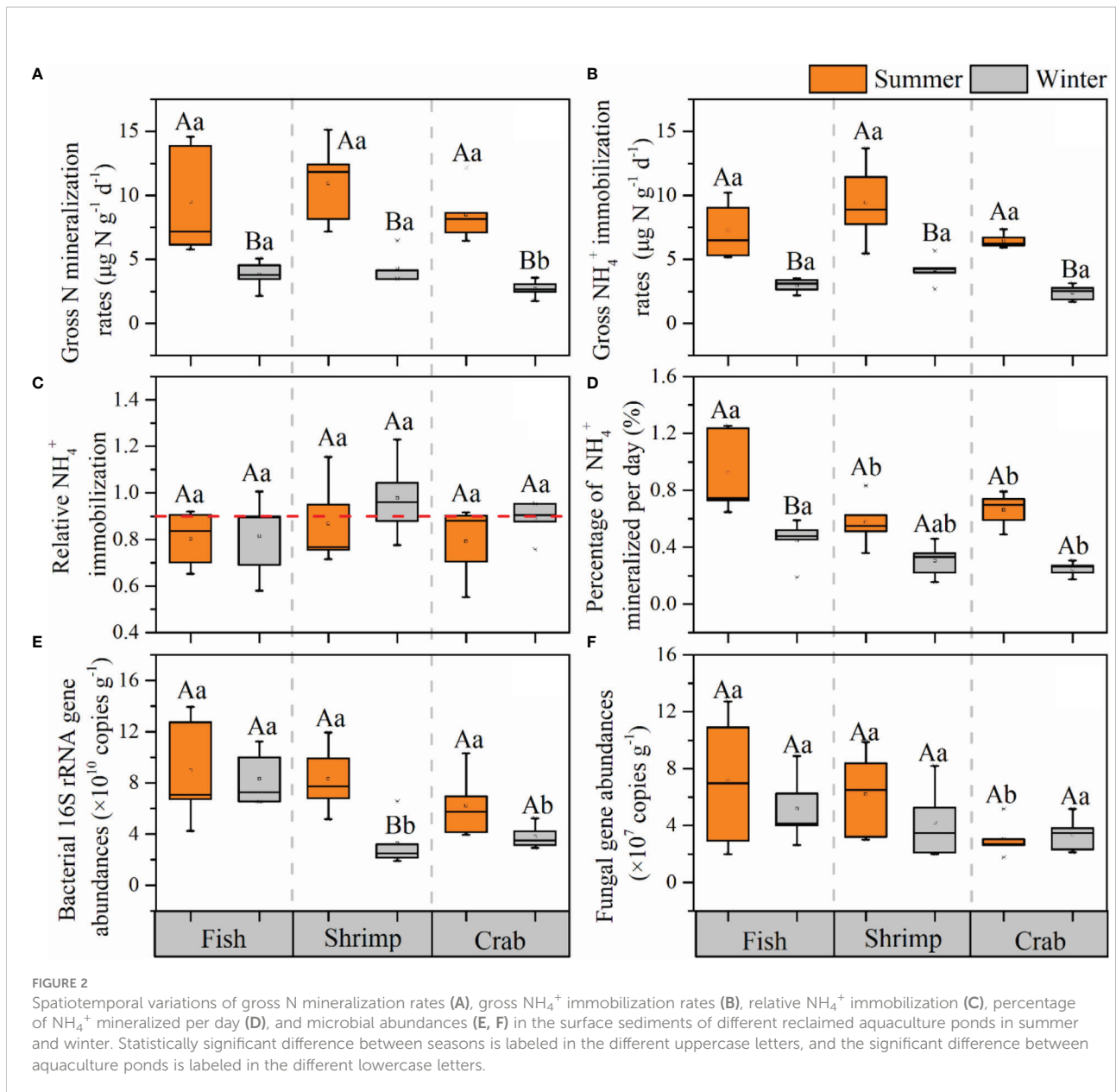
Statistically significant difference between seasons (F statistical variance homogeneity test $p < 0.05$) is labeled in the different uppercase letters, and the significant difference between aquaculture ponds (Turkey multiple comparison $p < 0.05$) is labeled in the different lowercase letters.

Due to geographic proximity, most surface sediment physicochemical properties (excluding NO₂⁻, Fe²⁺, and TOC/TN) were found no significant spatial variation among those three ponds in summer. In winter, the shrimp and crab ponds were in the drying and dredging period, and the surface sediments of the shrimp ponds were dried and cracked. Thus we found that surface sediment moisture content, Fe²⁺, EOC, DOC, and MBC of shrimp pond in winter were significantly lower than those in summer ($p < 0.05$ for all), and the sediment bulk density in winter were significantly higher than those in summer ($p < 0.05$). Meanwhile, several sediment physicochemical properties in shrimp pond were significantly higher (including bulk density and δ¹³C_{org}) or lower (including moisture content, TOC/TN, DOC, MBC, and bacterial 16S rRNA abundances) than those in other ponds in winter ($p < 0.05$ for all). In crab pond, sediment temperature, NH₄⁺, NO₃⁻, Fe²⁺, Fe³⁺, TOC, TN, EOC, and MBC in summer were significantly higher than those in winter ($p < 0.05$ for all). In addition, all the surface sediments in those three ponds were mainly composed of fine silt and clay with low median grain size (7.69–57.77 μm).

3.3 Spatiotemporal variations of N transformation rates and microbial abundances

The spatiotemporal distributions of sediment GNM, GAI, RAI, PAM, and microbial abundances in three reclaimed aquaculture ponds were shown in Figure 2. GNM rates varied from 5.77 to 14.58 μg N g⁻¹ d⁻¹ with an average value of 9.64 ± 3.31 μg N g⁻¹ d⁻¹ in summer and from 1.75 to 6.45 μg N g⁻¹ d⁻¹ with an average value of 3.60 ± 1.19 μg N g⁻¹ d⁻¹ in winter. Significant seasonal variations in GNM rates were observed in those three ponds, with significantly higher values in summer than in winter ($p < 0.05$ for all, Figure 2A). No significant difference in GNM rates occurred among three ponds in summer ($p > 0.05$ for all, Figure 2A). In comparison, in summer, the GNM rates in crab pond were significantly lower than those in the other two ponds ($p < 0.05$ for both, Figure 2A).

The GAI rates ranged from 5.15 to 13.68 μg N g⁻¹ d⁻¹ with an average of 7.71 ± 2.49 μg N g⁻¹ d⁻¹ in summer and from 1.67 to 5.66 μg N g⁻¹ d⁻¹ with an average of 3.17 ± 1.06 μg N g⁻¹ d⁻¹ in



winter. Similarly, GAI rates were higher in summer than in winter with a remarkable seasonal difference within these three ponds ($p < 0.05$ for all, Figure 2B). Spatially, no significant spatial difference in GAI rates occurred among three ponds in both summer and winter ($p > 0.05$ for all, Figure 2B). In addition, GNM rates were significantly positively correlated with GAI rates in the whole study area in both summer and winter ($p < 0.05$ for both, Figure S1).

The relative NH_4^+ immobilization (RAI) values ranged between 0.55 and 1.15 in summer and between 0.60 and 1.23 in winter (Figure 2C). There was no significant seasonal variation in RAI in those three ponds, and no significant

spatial difference among three ponds ($p > 0.05$ for all; Figure 2C). A previous study has confirmed that $\text{RAI} \leq 0.5$ and $\text{RAI} > 0.9$ indicated N-saturated environment and N-limited environment, respectively (Aber, 1992). Here, the RAI values in most sampling sites were between 0.5 and 0.9, indicating these reclaimed aquaculture ecosystems were able to maintain N balance.

PAM values ranged from 0.36% to 1.25% in summer and from 0.15% to 0.59% in winter (Figure 2D). Spatially, PAM in fish pond were significantly higher than those in other ponds ($p < 0.05$ for all; Figure 2D). The PAM in summer ($0.92 \pm 0.30\%$) was significantly higher than that in winter ($0.45 \pm 0.15\%$) in fish

pond ($p < 0.05$; [Figure 2D](#)). In shrimp and crab ponds, PAM in summer were also higher than those in winter, but with no significant seasonal difference ($p > 0.05$; [Figure 2D](#)).

The bacterial 16S rRNA gene abundances were in a range of $3.93\text{--}13.94 \times 10^9$ copies g^{-1} with an average of $7.82 \pm 3.22 \times 10^9$ copies g^{-1} in summer. In winter, the range and average value decreased to $1.91\text{--}11.22 \times 10^9$ copies g^{-1} and $5.12 \pm 2.85 \times 10^9$ copies g^{-1} . The bacterial 16S rRNA abundances in summer were significantly higher than those in winter in shrimp pond ($p < 0.05$; [Figure 2E](#)). The values in sediment of fish pond were significantly higher than those in other two ponds in winter ($p < 0.05$ for both, [Figure 2E](#)). The fungal ITS gene abundances were in a range of $1.76\text{--}12.73 \times 10^7$ copies g^{-1} with an average of $5.46 \pm 3.57 \times 10^7$ copies g^{-1} in summer and in a range of $2.00\text{--}8.88 \times 10^7$ copies g^{-1} with an average of $4.26 \pm 2.15 \times 10^7$ copies g^{-1} in winter, showing no significant temporal differences in those three ponds and no spatial differences among three ponds ($p < 0.05$ for all; [Figure 2F](#)).

3.4 Effects of physicochemical properties on N-cycling processes

Considering all sampling sites, GNM and GAI rates showed significant positive correlations with sediment temperature, organic matter (TOC, TN, EOC, DOC, and MBC), microbial abundances (bacterial 16S rRNA and fungal abundances), Fe^{2+} , NH_4^+ , moisture content, as well as overlying water DO and NO_3^- , while they were correlated negatively with sediment bulk density ($p < 0.05$; [Table 3](#)). PAM were significantly positively correlated with sediment MBC, microbial abundances, Fe^{2+} , Fe^{3+} , as well as overlying water DIN ($p < 0.05$; [Table 3](#)). RAI were positively correlated with sediment $\delta^{13}\text{C}_{\text{org}}$ and correlated negatively with sediment microbial abundances, MBC, DOC, and TOC/TN ($p < 0.05$; [Table 3](#)).

4 Discussion

4.1 Effects of physicochemical properties on sediment N mineralization and immobilization

Temperature could have directly impacts on microbial physiological activities through the regulation of enzyme activities and is considered as a crucial factor of N mineralization and immobilization ([Yang et al., 2010](#)). In this study, GNM and GAI rates were both significantly higher in summer than in winter among those three ponds ($p < 0.05$ for all). The seasonal variations are mainly due to temperature differences between summer ($31.81 \pm 0.86^\circ\text{C}$) and winter ($17.54 \pm 0.05^\circ\text{C}$). Also, the ratios of N-cycling rates in summer

to rates in winter were used to characterize the temperature sensitivity of N mineralization and immobilization rates. A previous study found that total N immobilization rate is very sensitive to temperature while N mineralization rate shows relatively less sensitivity under low temperature condition ($5\text{--}15^\circ\text{C}$) ([Andersen and Jensen, 2001](#)). However, we found that no significant difference between the temperature sensitivity of GNM and GAI ([Figure 3A](#)). The major reason is that the winter temperature in our study is over 15°C , and the temperature sensitivity of N mineralization has been found to decrease with increasing temperature ([Kirschbaum, 1995](#)). In addition, no significant spatial difference in temperature sensitivity of N mineralization and immobilization rates among those three ponds ([Figure 3A](#)). This is mainly due to similar physicochemical properties (including sediment grain size, organic matter, DIN, and TOC : TN) among those three ponds, and previous studies have found that temperature sensitivity of N mineralization and immobilization rates mainly depended on the above physicochemical properties in agricultural ecosystems ([Liu et al., 2017](#); [Miller and Geisseler, 2018](#)).

Organic matter of aquaculture sediments originates from both autochthonous sources (microalgae, sediment and particulate organic matter) and bait inputs ([Chen et al., 2016](#)). According to the values of TOC/N ratio and $\delta^{13}\text{C}_{\text{org}}$, we found the surface sediment organic matter were mainly from marine algae, marine DOC, and marine POC ([Figure 3B](#)). This is caused mainly by the fact that the aquaculture ponds were located on the estuary, where a significant proportion of overlying water is derived from marine. More importantly, the main ingredients ($\sim 60\%$) of bait are fish meal and fish oil ([Park et al., 2021](#)), and they were made from marine fish (e.g. *Engraulis japonicus*).

As we all know, organic matter acting as energy source of microorganisms and substrates of N mineralization, which plays an important role in microbial-mediated N-cycling processes. To clarify how labile C impacts on N mineralization and immobilization in the reclaimed aquaculture sediments, several labile organic carbons including sediment EOC, DOC, and MBC, as well as potential N mineralization and immobilization rates, were measured. When considering all sites, the correlation strengths between labile organic carbons (EOC, DOC, and MBC) and N mineralization and immobilization rates were much higher than those between TOC and the rates. The order of above correlations was $\text{MBC} > \text{EOC} > \text{DOC} > \text{TOC}$ ([Table 3](#)). When considering seasonal effects, we found that this regularity and trend was only found in summer ([Figure 4](#)), and the order of above correlations was changed slightly ($\text{MBC} > \text{DOC} > \text{EOC} > \text{TOC}$). This finding agreed well with several previous studies, which found that high availability of organic matter is favorable for soil/sediment N mineralization process in coastal wetland ([Li et al., 2020](#); [Yang et al., 2022](#)), estuary ([Huang et al., 2022](#)) and forest ([Jones and](#)

TABLE 3 Correlations between physicochemical properties and N mineralization and immobilization rates.

Physicochemical properties		GNM	GAI	RAI	PAM
Overlying water	DO	0.49*	0.58**	0.06	-0.16
	Salinity	0.11	0.14	0.00	-0.33
	pH	0.27	0.35	0.03	-0.39
	NH ₄ ⁺	0.35	0.25	-0.17	0.70**
	NO ₂ ⁻	-0.02	-0.07	-0.01	0.49*
	NO ₃ ⁻	0.67**	0.74**	0.07	0.48*
Sediment	Temperature	0.79**	0.79**	-0.23	0.68**
	Moisture content	0.56**	0.52**	-0.27	0.00
	Bulk density	-0.55**	-0.53**	0.24	-0.14
	NH ₄ ⁺	0.65**	0.67**	-0.18	0.24
	NO ₂ ⁻	-0.19	-0.15	0.17	-0.27
	NO ₃ ⁻	0.14	0.16	0.05	-0.01
	Fe ²⁺	0.73**	0.76**	-0.17	0.39*
	Fe ³⁺	0.30	0.19	-0.28	0.52**
	Fe ²⁺ /Fe ³⁺	0.27	0.31	-0.02	-0.02
	Medium grain size	0.20	0.18	-0.01	0.25
	SSA	0.11	0.12	-0.10	-0.24
	TOC	0.54**	0.52**	-0.20	-0.06
	TN	0.49**	0.50**	-0.04	-0.20
	TOC/TN	-0.13	-0.22	-0.37*	0.27
	δ ¹³ C _{Org}	0.01	0.22	0.45*	-0.02
	EOC	0.79**	0.76**	-0.25	0.30
	DOC	0.62**	0.53**	-0.44*	0.32
	MBC	0.77**	0.64**	-0.52**	0.38*
	Bacterial	0.71**	0.53**	-0.61**	0.63**
	Fungal	0.63**	0.53**	-0.38*	0.43*

*, significant correlation (two-tail test, p<0.05); ** highly significant correlation (two-tail test, p<0.01).

Kielland, 2012). Thus, our results suggest that organic quality rather than quantity was more crucial to the control of N mineralization and immobilization in aquaculture sediments.

No significant correlations existed between organic matter and N mineralization and immobilization rates in winter, this may be probably due to the profound effect of human activities (pond-drying) on these different aquaculture ponds. As mentioned in the preceding text, fish ponds are still being farmed, and shrimp and crab ponds are in a resting state in winter. Namely, shrimp and crab ponds were in the drying and dredging period, and surface sediment of the shrimp ponds was dried and cracked. Firstly, pond-drying can significantly reduce the sediment moisture content in shrimp ponds in winter,

previous studies have found that N mineralization increased with the increasing soil/sediment moisture (Jia et al., 2019; Li et al., 2020; Huang et al., 2022). Sediment moisture is involved in the metabolism of microbes and is also the irreplaceable medium linking substrates supply and microbial activity (Greaver et al., 2016). Elevated sediment moisture has also been reported to increase the decomposition and leaching of organic matter and extracellular enzyme activities, promoting soil N transformations (Mooshammer et al., 2014; Greaver et al., 2016; Jia et al., 2017). Also, increased soil moisture could limit oxygen to penetrate soils and further lead to more strongly anoxic and reducing environments, thereby facilitating denitrification and N mineralization in coastal wetlands

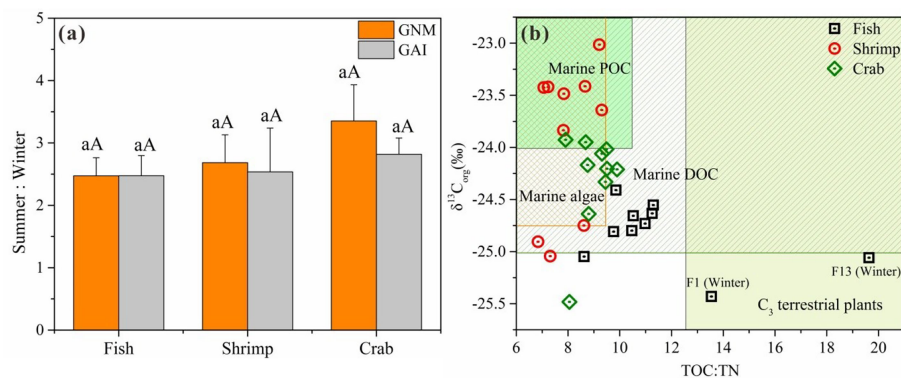


FIGURE 3

The temperature sensitivity of sediment GNM and GAI rates to seasonal change in those three ponds (A); Determination of organic matter sources from autochthonous or allochthonous (B). The background ranges of organic matter difference sources were obtained by Lamb et al. (2006). Statistically significant difference between seasons is labeled in the different uppercase letters, and the significant difference between aquaculture ponds is labeled in the different lowercase letters. The error bar represents the standard error.

(Jia et al., 2019). Secondly, our measured sediment labile organic matter contents (including EOC, DOC, and MBC) in shrimp pond were significantly less in winter compared with those in summer (Table 2, $p < 0.05$ for all). Thus, pond-drying can accelerate the decomposition of surface sediment organic matter, resulting in decreased the supply of energy source and substrate for microbial growth and mineralization. In addition, pond-drying can significantly reduce microbial biomass (MBC) and bacterial abundances in shrimp pond (Table 2, $p < 0.05$ for both), thereby regulating the N mineralization and immobilization. Therefore, most physicochemical properties were significantly changed with aquaculture management activities (pond-drying), resulting in the profound reduction of sediment N mineralization and immobilization in aquaculture ecosystems.

No significant seasonal and spatial difference in microbial abundances was observed in our study excluding those in shrimp ponds in winter (Table 2), this is largely due to small summer-winter temperature differences and low spatial heterogeneity in sediment physicochemical properties. When the surface sediment was dried and cracked after long-term pond-drying, this aquaculture management activity can significantly reduce the microbial biomass and bacterial abundances, but not for fungal abundances (Table 2). This is mainly due to fungi can be remarkably drought tolerant, and they can remain active and even grow under extremely dry conditions (Treseder et al., 2010; Yuste et al., 2011). However, our study lack of data about microbial community structure and diversity. A previous study showed that pond-drying can change the sediment microbial community structure, and the abundance of *Proteobacteria*,

Nitrospirae and *Bacteroidetes* were increased, while the pernicious microbe such as *Cyanobacteria* was decreased after pond-drying in aquaculture ecosystem (Wang et al., 2020). In addition, natural sunlight has effect on the abundance and microbial community structure in the sediments during pond-drying. Thus, we boldly speculated that pond-drying can alter microbial community composition from anaerobic, fast-growing, and copiotrophic to aerobic, slow-growing, and oligotrophic, thus promote the ecosystem reparation effectively.

Many previous studies have reported that N mineralization and immobilization rates were affected by the microbial biomass and activities, and thus they were generally closely related to the microbial respiration rates, enzymatic activities (including extracellular and intracellular enzymes), and ATP content (Bengtsson et al., 2003; Silva et al., 2005). N mineralization and immobilization rates were also controlled by the microbial biomass (MBC) and abundances (including bacterial 16S rRNA and fungal abundances) in this study (Table 4), which is consistent with previous studies in many estuarine and coastal zone (Lin et al., 2016a; Li et al., 2020; Huang et al., 2022; Yang et al., 2022). Thus, microbial abundances could be considered as important indicators of N mineralization and immobilization rates in aquaculture sediments.

In addition, GNM and GAI rates were also positively correlated with overlying water NO_3^- (Table 3, $p < 0.05$). Previous studies have revealed significant influence of environmental effective electron acceptors on N mineralization. For example, NO_3^- , as an electron acceptor of oxidized organic matter, could promote N mineralization (equation 3) and make high contribution (15–35%) to the mineralization of organic matter especially under

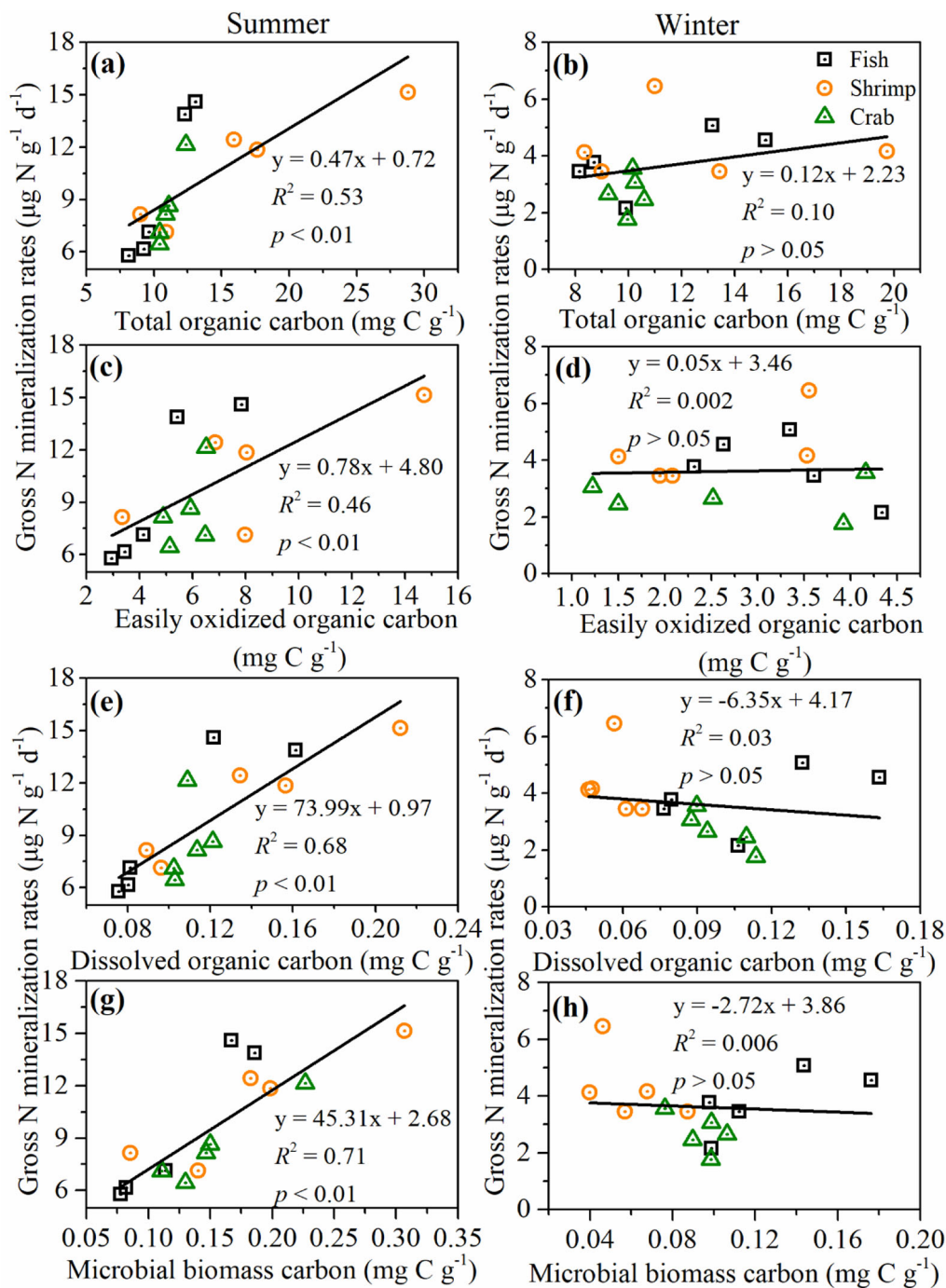
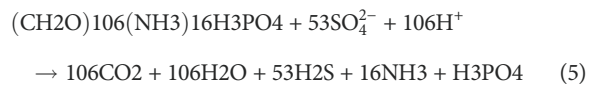
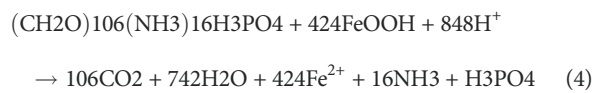
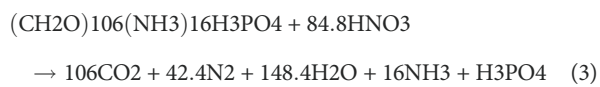


FIGURE 4
 Correlations between different organic matter compounds and gross N mineralization rates in the surface sediments of different reclaimed aquaculture ponds. (A, B) Total organic carbon; (C, D) easily oxidized organic carbon; (E, F) dissolved organic carbon; and (G, H) microbial biomass carbon.

anoxic conditions (Khalil et al., 2018). Under anaerobic conditions, NO₃⁻, Fe³⁺, SO₄²⁻ as well as iron and manganese ions often work as active electron acceptors for coupling redox reactions with sediment organic matter (equation 4 and 5) (Pena et al., 2010). As sediment is flooded for long time and the vertical penetration

ability of oxygen from overlying water into sediment is relatively weak, anaerobic conditions for the aforementioned reactions could be created in aquaculture ecosystem. Additional work regarding aerobic N mineralization can be done to compare the mechanisms between anoxic environments and aerobic environments.



4.2 Environmental implications for N mineralization and immobilization in reclaimed aquaculture ecosystems

The average N mineralization and immobilization rates of reclaimed aquaculture ponds in this study area were lower than the rates of other aquaculture, urban river network, and mangrove wetland sediments, and higher than the rates of estuarine and coastal sediments (Table 4). The daily mineralization percentage of NH_4^+ was 0.55–1.23%, which is comparable with the value in coastal wetlands (Table 4). In

addition, the value of RAI in this study (average: 0.86 ± 0.15 ; range: 0.58–1.23) implied that the ecosystem might be located between N limitation condition and N saturation condition (Aber, 1992). Significantly positive correlations between GNM and GAI identified in this study showed consistency with the observations in offshore seawater (Lin et al., 2016), wetlands (Jin et al., 2012), grassland (Corre et al., 2002) and forests (Bengtsson et al., 2003). It's also noteworthy that a combination of sediment incubation experiment and ^{15}N stable isotope dilution method was adopted in this study. As described in Lin et al. (2017), the added $^{15}\text{NH}_4^+$ may stimulate microbial responses, along with the dilution ratio of 1:5 (fresh sediment: water) and aerobic incubation conditions, the measured GNM and GAI might not reflect *in situ* activities. Nevertheless, it can still imply potential activities of N mineralization and immobilization in aquaculture surface sediments.

Sediment N mineralization and immobilization are important N-cycling processes in coastal wetland reclamation and aquaculture ecosystems, which plays vital roles in the reactive N balance. However, very few systematic studies investigated environmental properties, N mineralization and immobilization, and their relationships, especially in reclaimed aquaculture ecosystems. Here, based on GNM and GNI measured in this study, annual surface sediment (0–5 cm) N

TABLE 4 Comparison of N mineralization and immobilization in coastal, estuarine, and aquacultural ecosystems and our study area.

Study area	Sample type	NH_4^+ / NO_3^- ($\mu\text{g N g}^{-1}$)		GNM / GAI ($\mu\text{g N g}^{-1} \text{d}^{-1}$)		RAI	PAM %	References
Limfjorden, Denmark	Marine	–	–	2.55	–	–	–	(Blackburn, 1979)
Chesapeake Bay, United States	Marine	–	–	1.6–5.73	–	–	–	(Cufrey and Kemp, 1992)
Tilapia fish ponds, Netherlands	Aquacultural	1.15–9.33	–	9.30	–	–	–	(Jiménez-Montealegre et al., 2005)
Native coastal wetland, China	Coastal	–	–	1.71 –10.56	0.89–8.11	–	–	(Jin et al., 2012)
Yangtze Estuary, China	Coastal	–	–	0.02–5.13	–	–	0.01 –2.14	(Lin et al., 2016a)
East China Sea, China	Marine	0.45–8.58	0.62	0.11–6.10	0–9.82	–	0.01 –2.89	(Lin et al., 2016b)
River network in Shanghai, China	River	4.19 –383.25	0.11 –5.02	0.25 –25.83	0.24 –26.27	–	–	(Lin et al., 2017)
Min River Estuary, China	Estuarine	3.53–11.35	–	2.00–5.90	1.10–5.10	–	–	(Li et al., 2020)
Pearl River Estuary, China	Estuarine	3.10–6.91	0.89– 2.85	0–3.43	0.02–2.60	–	–	(Huang et al., 2021)
Pearl River Estuary, China	Estuarine	3.48–9.70	0.76 –2.11	0.15–1.99	0–1.96	0–1.81	0–0.423	(Huang et al., 2022)
Qi'ao Island, China	Coastal	10.43– 21.58	0.76– 3.14	2.69– 17.53	2.29– 21.38	0.79– 1.54	0.24– 0.86	(Yang et al., 2022)
Aquaculture ponds, China	Aquacultural	1.31–58.10	1.11 –8.72	1.75 –15.13	1.67 –13.68	0.55 –1.23	0.15 –1.25	This study

mineralization and immobilization rates in reclaimed aquaculture ecosystems of the Great Bay Area were calculated (equation 6).

$$F = \frac{1}{30} \left(\sum_{i=1}^{15} mi \cdot di + \sum_{i=1}^{15} mj \cdot dj \right) \cdot \alpha \cdot s \cdot h \cdot t \quad (6)$$

In which, F ($t \text{ N yr}^{-1}$) is annual N mineralization or immobilization flux; mi and mj ($\mu\text{g N g}^{-1} \text{ d}^{-1}$) is GNM or GAI of surface sediment (0–5 cm) in summer and winter, respectively; di and dj (g cm^{-3}) is bulk density of surface sediment (0–5 cm) in summer and winter, respectively; a is a unit conversion factor; s (m^2) is the area of reclaimed aquaculture ecosystems in the Great Bay Area ($\sim 395.57 \text{ km}^2$ according to remote sensing data in 2018 followed by the calculation using ArcGIS 10.2) (Figure S2); h (cm) is sampling depth (5 cm); t (d) is time (365 days).

In total, annual sediment N mineralization and immobilization rates in reclaimed aquaculture ecosystems are approximately $4.55 \times 10^4 \text{ t N yr}^{-1}$ and $3.68 \times 10^4 \text{ t N yr}^{-1}$, respectively in the Guangdong-Hong Kong-Macao Greater Bay Area. GNM flux shows higher level than GAI flux, indicating that sediment is a nonnegligible DIN source in the aquaculture ecosystem, which plays a non-negligible role in the exacerbation of eutrophication in the estuarine and coastal ecosystems.

5 Conclusions

This study investigated N mineralization and immobilization rates in surface sediments of the reclaimed aquaculture ecosystems. GNM and GAI rates ranged from 1.75 to $15.13 \mu\text{g N g}^{-1} \text{ d}^{-1}$ and from 1.67 to $13.68 \mu\text{g N g}^{-1} \text{ d}^{-1}$, respectively. Both GNM and GAI rates were significantly higher in summer than their counterparts in winter ($p < 0.05$). No significant differences among three types of aquaculture ecosystems were observed ($p > 0.05$). N mineralization and immobilization rates were significantly correlated with overlying water NO_3^- , as well as sediment moisture content, bulk density, organic matter, Fe^{2+} , and microbial abundances. Furthermore, this study estimated the total mineralized and immobilized N in aquaculture surface sediments from the Guangdong-Hong Kong-Macao Greater Bay Area, with the estimation of approximately $4.55 \times 10^4 \text{ t N yr}^{-1}$ and $3.68 \times 10^4 \text{ t N yr}^{-1}$, respectively. GNM flux shows higher level than GAI flux, indicating that sediment is a crucial DIN source in the aquaculture ecosystem, which plays a non-negligible role in the exacerbation of eutrophication. Overall, this study improves the understanding of sediment N transformation processes and relevant regulation mechanisms in the reclaimed aquaculture ecosystems.

Data availability statement

The original contributions presented in the study are included in the article/Supplementary Material. Further inquiries can be directed to the corresponding author.

Author contributions

XL: Methodology, Validation, Formal analysis, Investigation, Data curation, Writing-original draft, Writing-review & editing, Funding acquisition. GL: Writing-original draft, Formal analysis, Writing-review & editing. YZ: Writing-original draft, Formal analysis, Visualization, Funding acquisition. WL: Investigation, Data curation, Writing - review & editing. PG: Data curation, Formal analysis, Writing - review & editing. SF: Writing-original draft, Formal analysis. TK: Data curation, Formal analysis. DT: Formal analysis, Data curation, Supervision. DS: Investigation, Data curation. ZS: Writing-original draft, Writing - review & editing, Funding acquisition. All authors contributed to the article and approved the submitted version.

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Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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Supplementary material

The Supplementary Material for this article can be found online at: <https://www.frontiersin.org/articles/10.3389/fmars.2022.1093279/full#supplementary-material>

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