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Biodegradation of insecticides: oligonucleotide insecticides and double-stranded RNA biocontrols paving the way for eco-innovation

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Each new class of insecticides that emerged during the development of plant protection gradually found the most suitable group of insect pests for application. At the same time, for each individual insecticide, a balance was sought between its effectiveness, on the one hand, and its safety for non-target organisms and the ecosystem as a whole, on the other hand. Neonicotinoids, diamides and pyrethroids, as effective control agents, dominate the insecticide market, but do not have outstanding performance in selectivity and biodegradation. The biodegradation of insecticides is one of the most important indicators, representing what will be said about the hidden costs for the resulting harvest paid by the environment and human health. Oligonucleotide insecticides (contact unmodified antisense DNA (CUAD) biotechnology, or 'genetic zipper' method) and RNA biocontrols (double-stranded RNA technology) as natural polymers and the next-generation classes of insecticides possess unique characteristics in fast biodegradation and high selectivity in action. While current chemical insecticides require days, months and even years for biodegradation by bacteria and fungi, oligonucleotide insecticides and RNA biocontrols are substantially biodegraded within hours in the presence of nucleases. Nucleic acid-based insecticides have the potential to complement the existing insecticide market and set an eco-precedent for crop protection products where the effectiveness of the insecticide will be determined by its safety for non-target organisms, and other factors being equal, the choice of a particular control agent will be determined by its biodegradability. It should be noted that not a single class of insecticides that once appeared has completely disappeared; rather, it has occupied its niche, gradually declining under the pressure of new classes of insecticides. At the same time, the common trend in plant protection is towards use of insecticides with higher biodegradability, which gives hope for a safer future of the planet.

KEYWORDS

chemical insecticides, oligonucleotide insecticides, contact unmodified antisense DNA (CUAD) biotechnology, RNA biocontrols, RNAi, biodegradability of insecticides

1 Introduction

Insecticides, like any other substances that enter insect cell, are subject to biodegradation and disposal. Substances of natural origin undergo rapid biodegradation, while xenobiotics degrade slowly. The rate of decomposition of xenobiotics depends on how many functional groups they contain found in natural molecules, how they are arranged and depends on activity of enzymes that can catalyze the transformation reactions with the participation of these functional groups (Migliani et al., 2022). Microorganisms due to their short life cycle are able to quickly adapt to an environment that accumulated xenobiotics (Migliani et al., 2022; Mishra et al., 2021). During their rapid microevolution microorganisms form strains that can quickly degrade xenobiotics reducing the ecotoxicological load in ecosystems (Huang et al., 2018). Should also be noted that omics approaches (genomics, transcriptomics, metabolomics, proteomics) addressing microbial metabolism and physiology could be of immense help towards unravelling the mechanism behind the phenomenon and even developing novel biotechnological tools (Rodríguez et al., 2020). Genera *Pseudomonas*, *Bacillus*, *Alcaligenes*, *Cladosporium*, *Actinomyces*, *Nocardia*, *Aspergillus*, *Fusarium*, *Trichoderma*, *Mortierella*, etc., are considered as perfect “biodegradation micro machines” for many classes of insecticides. Just like a hundred years ago, when creating insecticides, scientists first of all focus on effectiveness and only then consider the possibilities of minimizing the negative effect of new insecticides on non-target organisms (Sánchez-Bayo, 2021). Of course, it is impossible to calculate all the environmental risks of an insecticide being created, otherwise it will never reach the market. However, it is the effectiveness of an insecticide that most often brings it to the market (Kalyabina et al., 2021), while the safety of the insecticide and its low potential for biodegradation most often influences its ban (Damalas and Eleftherohorinos, 2011). Majority of insecticides were banned in one way or another after decades or years of use in plant protection, when effective competitors with proven or supposed greater safety appeared (Donley, 2019).

Any insecticide can be evaluated in terms of its effectiveness, affordability, half-life (biodegradation) and safety. Biodegradation and the safety of an insecticide are almost always related: the safer the insecticide, the shorter its half-life (Pathak et al., 2022). Nature does not slow down the degradation of compounds that have natural origin; they are broken down to water, carbon dioxide, ammonia, acetic acid, phosphoric acid, and other simple substances serving as precursors for more complex natural compounds. Fortunately, later classes of insecticides reduced their hazard for non-target organisms and increased their biodegradability potential (Araújo et al., 2023). However, using the example of neonicotinoids, as one of the later class of insecticides, and their dangerous effect on pollinators and birds, we see that the achieved level is still insufficient (Muth and Leonard, 2019). Today, a new generation of plant protection products is being developed, based on biopolymers, DNA (contact unmodified antisense DNA (CUAD) biotechnology) and RNA (double-stranded RNA technology) (Oberemok et al., 2019; Oberemok et al., 2022; Oberemok et al., 2023; Gal'chinsky et al., 2023; Puzanova et al., 2023; Rank and Koch, 2021; De Schutter et al., 2022). The function of these natural molecules in the cell primarily emphasizes the exceptional selectivity of the action of such

compounds, which depends on the combination of nitrogenous bases. Both CUAD biotechnology and double-stranded RNA technology are essentially based on the mechanism of gene silencing using antisense DNA and RNA fragments and DNA- or RNA-guided nucleases (Oberemok et al., 2018). Both DNA and RNA are natural molecules that eventually are biodegraded by ubiquitous nucleases in ecosystems (Yang, 2011). If in the near future a balance is found between the effectiveness and cost of such pest control agents, then the insecticide market will be replenished in abundance with two new classes of insecticides possessing high level of biodegradability. Obviously, such insecticides will not become a panacea, but will be effective against certain orders of insect pests. In the medium term (5–10 years) insecticides based on nucleic acids will not be able to dominate the insecticide market, but their appearance on the market will be another quantum leap towards safe and sustainable plant protection.

In this article we will consider biodegradability of chemical insecticides including novel pest control agents based on CUAD biotechnology (oligonucleotide insecticides) and double-stranded RNA technology (RNA biocontrols). Priority of consideration of classes of insecticides in this article will follow rank of total sales worldwide of a particular class of pest control agents.

1.1 Neonicotinoids

The age of neonicotinoids (neonics) began with nitromethylene nithiazine, but really was inspired by Shinzo Kagabu's important discoveries (Casida, 2018), culminating in 1990s with the introduction of chloropyridinyl and nitroimine moieties to generate imidacloprid, the best-selling insecticide for many years (Nugnes et al., 2023). Today, the neonicotinoid class accounts for 20% of total pesticide sales worldwide (King, 2022).

Neonicotinoids are used on a wide range of crops, such as horticultural and industrial crops, grapes, citrus fruits, pome fruits, stone fruits, flower and ornamental plants (Kundoo et al., 2018). Neonicotinoids can control a wide range of insect pests (Matsuda et al., 2020), including representatives of orders Hemiptera (Li et al., 2018; Li L. et al., 2020; Rizwan et al., 2022; Barbosa et al., 2018) and Lepidoptera (Krishnan et al., 2021); soil insects (Labrie et al., 2020) and Colorado potato beetle (CPB) *Leptinotarsa decemlineata* (Jeschke et al., 2011) are included among the target pests (Kundoo et al., 2018).

It was found that insecticides of this group are able to bioaccumulate and persist in the environment after period of their use, which is due to the low bioavailability and ability of their residues to bind to the soil matrix (Hussain et al., 2016; Zamule et al., 2021). The persisting chemicals in this group, clothianidin and imidacloprid, have the highest half-lives of about 1,250–6,931 days in sandy loam soil, having slower rates of decomposition in temperate climates (Farahy et al., 2021; Goulson, 2013). In general, the half-life of neonicotinoids in soil typically ranges from 200 to more than 1,000 days. The moderate half-life (7–353 days) is shown for thiamethoxam in soil under dry conditions (Goulson, 2013; Bonmatin et al., 2015). Relatively lower persistence in soil was observed for thiacloprid and dinotefuran with half-lives of approximately 90 days (Gautam and Dubey, 2023). Half-life appears to be shorter for the

N-cyanoamidines (for cetamiprid it comprises ca. 450 days) (Goulson, 2013). Traces of hazardous neonicotinoid residues have been found dispersed in parts per million and parts per billion amounts in open environments (Bonmatin et al., 2015). Clothianidin, imidacloprid and thiamethoxam in water have half-life of 365–408 days (Borsuah et al., 2020).

Epidemiological studies are sparse, but they do relate neonicotinoid exposure to oxidative stress, neurological symptoms, metabolic alterations, osteoporosis, and liver cancer (Li A. J. et al., 2020; Marfo et al., 2015; Vuong et al., 2022; Carmichael et al., 2014), as well as imidacloprid exposure to developmental, congenital, hematologic, hepatic, and renal consequences (Carmichael et al., 2014; Khan et al., 2010; Shaw et al., 2014; Yang et al., 2014). Neonicotinoids have also caused huge losses in bee populations around the world (colony collapse syndrome) (Zee Van Der et al., 2012). Research has shown that trace amounts of neonicotinoids used in agriculture usually do not kill bees directly (lethal effect), but have an indirect (sublethal) effect, as bees exhibit all the symptoms of insecticide poisoning—uncoordinated movements, tremors and seizures (Buszewski et al., 2019). In 2019, the European Union prohibited two main neonicotinoids, imidacloprid and clothianidin, later thiamethoxam and thiacloprid were banned in 2020 because they were used on crops attractive to bees. Also due to side effects on bees, the sulfoxaflor was banned in 2022 (Strouhova et al., 2023).

Neonicotinoid insecticides are cleared from the environment through various processes of biotic (microbe- and plant-mediated) and abiotic (physicochemical) transformation (Gautam and Dubey, 2023). Microbes play an indispensable role in ultimate fate of neonicotinoid pollutants. The biocatalytic activity of microbes enables them to act as powerful degraders of neonicotinoids (Bilal et al., 2019). As an example, there are some powerful microorganisms such as genera *Ensifer*, *Phanerochaete*, *Bacillus*, *Ochrobactrum*, *Trametes*, *Rhodococcus* and *Pseudomonas*, which have enormous potential in degradation of neonicotinoids (Gautam and Dubey, 2023). Cytochrome P450 enzymes have been shown to actively mediate the degradation of neonicotinoids in various organisms. In bacteria, P450s were found to closely relate to hydroxylation (Wei et al., 2023). For instance, a study by Guo et al. (2020) using *Hymenobacter latericoloratu* CGMCC 16346 to reduce imidacloprid from surface water has shown that the process of the hydroxylation of imidacloprid to 5-hydroxy imidacloprid and olefin imidacloprid was inhibited by the presence of piperonyl butoxide, an inhibitor of cytochrome P450 monooxygenases (Guo et al., 2020). In a report, Sun et al. (2018) isolated *Ensifer adhaerens* CGMCC 6315 whose resting cells achieved ~95% of 200 mg·L⁻¹ acetamiprid degradation within 12 h and quickly eliminated ~88% of 5 mg/kg of residual soil acetamiprid within 2 days (Sun et al., 2018). *Phanerochaete chrysosporium* (CCTCC AF 96007) can degrade ~98% of thiamethoxam (10 mg·L⁻¹) within 25 days (Gautam and Dubey, 2023; Sharma et al., 2014). Within 2 days, strain *Ochrobactrum* sp. D-12 was able to degrade acetamiprid with initial concentrations of 250–3,000 mg·L⁻¹ (Wang et al., 2013). Specifically, *Trametes versicolor* could remove 20% and 64.7% of acetamiprid and imidacloprid (4 mg·L⁻¹), respectively, in 7 days (Hu et al., 2022). Actinomycete *Rhodococcus ruber* (CGMCC 17550) can degrade neonicotinoid insecticide nitenpyram (25 mg·L⁻¹) by 71.85% and 87.11% after

3 and 8 days of incubation, respectively (Dai et al., 2021). *Bacillus brevis*, *Pseudomonas* sp. F1, *Bacillus subtilis* and *Rhizobia* degraded 25%–45% of imidacloprid (25 g·L⁻¹) in a carbon-limited minimum salt environment in 25 days (Sabourmoghaddam et al., 2015).

1.2 Pyrethroids

The next class we will consider is pyrethroids, its world market share accounts for 19% of total pesticide sales worldwide, worth approximately US\$ 2.8 billion (King, 2022). Pyrethroids, like neonicotinoids, are effectively degraded by many bacteria and fungi. Historically, pyrethroids are a group of derivatives of naturally occurring compounds, pyrethrins, isolated from the flowers of the plant *Chrysanthemum cinerariaefolium* alias *Tanacetum cinerariaefolium* (Dalmatian pyrethrum) (Hołyńska-Iwan and Szewczyk-Golec, 2020; Hodoşan et al., 2023). Plant materials containing pyrethrins have been sold in Europe since the mid-18th century (Ahmed et al., 2021). The early research that laid the foundation for the discovery of pyrethroids has a significant historical trajectory dating back to the mid-20th century and characterized by a number of important advances in the development of synthetic insecticides.

Currently, pyrethroids are widely used in crop protection; in the forestry, woodworking, and textile industries; in medicine and veterinary medicine for the treatment of infections by parasitic crustaceans (Chrutek et al., 2018; Banks et al., 2014; Barlow et al., 2001; Orsborne et al., 2016; WHO, 2016). Pyrethroids are effective against a wide range of insect pests of the order Coleoptera (Athanssiou et al., 2015; Arthur et al., 2018; Stará and Kocourek, 2019), Hemiptera (Anderson and Cowles, 2012; Menger et al., 2020), Diptera (Önder and Huseyin, 2019; Santamaría and Marceló, 2019), Hymenoptera (Pereira Costa et al., 2020), Lepidoptera (Rabelo et al., 2020; Moreno et al., 2017), Orthoptera (Mullié et al., 2023), and Thysanoptera (Ahmad and Kumar, 2023a). As a rule, they are also used for personal protection against insects in the form of impregnated mosquito nets, sprays, or gels (Banks et al., 2014; Wylie et al., 2016; Ranjkesh et al., 2013).

Based on the vast amount of published research regarding the toxic effects of pyrethroids on non-target organisms and humans, some of these pyrethroids are banned in developed countries. Some prohibited pyrethroid insecticides such as fenvalerate, permethrin and other compounds, including certain metabolites, cause DNA damage in human (Jurewicz et al., 2015) and animal sperm, including decreased sperm count and concentration, decreased sperm motility, and increased abnormal sperm morphology (Cremonese et al., 2017; Orlu, 2014; Shi et al., 2011). Pyrethroids are neurotoxic pesticides and affect neurotransmitters (Shafer et al., 2005; Gammon et al., 2019). Additionally, a cohort study documented that some exposure to pyrethroids in pregnant women resulted in autism spectrum disorder in their children (Barkoski et al., 2021). At the animal experimental level, it has been reported that certain pyrethroids, i.e., deltamethrin (Gasmi, 2020) and cypermethrin (Abd El-Moneim Ibrahim et al., 2020), caused a decrease in serotonin and dopamine levels in rats (Bayoumi, 2022). Unfortunately, it should be emphasized that, despite the proven negative impact of pyrethroids on human health, at the current stage of knowledge it is not possible to

offer a safer measure of personal protection against insects (WHO, 2014; WHO, 2015). However, for example, fenvalerate is banned in Sweden due to side effects in workers caused by occupational poisoning (Hadnagy et al., 2003; Muller-Mohnssen and Hahn, 1995).

Pyrethroids are currently emerging pollutants due to their ecotoxicological effects on aquatic systems even at extremely low concentrations (Geissen et al., 2015). Frequent use of pyrethroids in agriculture and forestry leads to the accumulation of their residues in the environment, especially in topsoil and water (Bragança et al., 2019; DeMars et al., 2021; Deng et al., 2020; Budd et al., 2020). These pyrethroids can enter rivers through a variety of routes, such as agricultural drainage, atmospheric deposition, surface runoff, and wastewater discharges (Alonso et al., 2012; Prusty et al., 2015; Luo and Zhang, 2011). In the natural environment, pyrethroids are degraded through biotic and abiotic pathways, including photooxidation, chemical oxidation, and biodegradation (Abraham and Silambarasan, 2014; Abraham and Silambarasan, 2016).

Today, there is a fairly large number of studies confirming that bacteria and fungi can quite effectively decompose the traces of pyrethroids in water and soil (Cycoń and Piotrowska-Seget, 2016; Huang et al., 2022; Huang et al., 2023). Microbiological degradation is considered a reliable and inexpensive method of recovery compared to common physicochemical methods (adsorption, photolysis and ozonation). It is important to note that microorganisms can use pyrethroids as a sole source of energy or with the help of other nutrients to remove pyrethroids through co-metabolism (Zhao et al., 2019). The primary way that pyrethroids are degraded by microorganisms is *via* ester-bond hydrolysis by carboxylesterases (carboxylic-ester hydrolase, EC 3.1.1.1), which yields carboxylate and alcohol (Sogorb and Vilanova, 2002; Aranda et al., 2014). A variety of potent pyrethroid-degrading bacteria have been reported. *Bacillus cereus* GW-01 (Jiang et al., 2023), *Pseudomonas aeruginosa* CH7 (Zhang et al., 2011), and *Brevibacillus parabrevis* BCP-09 (Tang et al., 2018a) can degrade beta-cypermethrin; *Bacillus* sp. DG-02 can degrade 61% of extremely high concentration of bifenthrin within 7 days (Chen et al., 2012); *Ochrobactrum anthropi* strain JcM1 (Akbar et al., 2015), *Ochrobactrum lupini* strain DG-S-01 (Chen et al., 2011a), *Ochrobactrum tritici* strain pyd-1 (Wang et al., 2011), and *Serratia nematodiphila* strain CB2 (Tyagi and Prashar, 2015) was found to be able to directly utilize more than 90% of the initial dose of cypermethrin within 5–10 days; *Bacillus subtilis* 1D has been reported to hydrolyze 95% of cypermethrin within 15 days; *Paracoccus acridae* SCU-M53 can metabolize 79.84% of cyhalothrin in 2 days (Tian et al., 2018); *Photobacterium ganghwense* can grow with cyfluthrin as the sole carbon source and degrade 60% of cyfluthrin in 3 days (Wang et al., 2019). *Serratia marcescens* DeI-1 (Cycoń and Piotrowska-Seget, 2016) and *Lysinibacillus* sp. ZJ6 (Jiang et al., 2023) can degrade deltamethrin; and *Bacillus flexus* XJU-4 (Mulla et al., 2017) and *Bacillus licheniformis* CY-012 (Tang et al., 2018a) can degrade fenvalerate. Also, fungi have been found as the effective degraders of pyrethroids, for example, *Cladosporium* sp. strain HU can degrade fenvalerate within 5 days (Chen et al., 2011b). Most pyrethroids are similar in structure, which opens the possibility for one strain to degrade many different pyrethroids (Bhatt et al., 2020).

1.3 Diamides

Next, we will look at the increasingly popular class of insecticides, diamides, with a global market share of 16% (King, 2022). Diamides like neonicotinoids and pyrethroids are degraded by many microorganisms. In 1993, the Japanese company Nihon Nohyaku began an insecticide program that led to a breakthrough in 1998 with the development of phthalimides, which have the aniline moiety replaced by a perfluoroalkyl side chain (Seo et al., 2008; Hamaguchi et al., 2012).

The spectrum of action of diamides is wide: flubendiamide mainly affects Lepidoptera, chlorantraniliprole controls whiteflies, leaf miners, beetles, and termite species in addition to lepidopterans, while cyantraniliprole is effective in controlling a large number of Lepidoptera, Hemiptera, Coleoptera, Diptera and Thysanoptera species (Teixeira and Andaloro, 2013; Li W. et al., 2019; Kadala et al., 2020). The insecticidal activities of some synthesized chlorantraniliprole derivatives were evaluated using a lepidopteran harmful pest of crops worldwide, diamondback moth *Plutella xylostella* (El-Sheikh and Ashour, 2021). Recent studies have found that broflanilide poses a potential risk to lady beetles *Coccinella septempunctata* (Cong et al., 2023).

When used extensively and incorrectly, diamides have a significant impact on the ecological environment (plant development and non-target organisms) (Ma et al., 2021). Phase I and phase II metabolism of xenobiotic compounds is thought to be mediated by three enzyme classes, cytochrome P450s (P450), carboxylesterases (CE) and glutathione S-transferases (GST) (Li et al., 2007). Metabolism of diamides in mammals has been shown to occur by oxidation, e.g., hydroxylation of alkyl groups (Yoshida, 2014). Although diamide metabolism in insects is not well elucidated, several studies have found upregulation of P450, CE or GST genes as well as enzyme activity in response to insecticide exposure, suggesting that the mechanism is potentially similar (Richardson et al., 2020). Reports have shown that the insecticide flubendiamide severely affects the antennal neurons of honey bees by disrupting calcium homeostasis (Kadala et al., 2020). Chlorantraniliprole and cyantraniliprole inhibit the growth rate, weight and reproduction of the earthworm *Eisenia fetida* (Qiao et al., 2019). It was also found that chronic exposure to chlorantraniliprole 20SC in pollen suppressed bumblebee reproduction and increased lethargy in the worker population for the duration of exposure (Smagghe et al., 2013). In addition, it was found that bees exposed to topical chlorantraniliprole had a decrease in locomotor activity (Kadala et al., 2019). It has also been reported that flubendiamide has been restricted or banned in some countries due to its toxicity to aquatic invertebrates (Zhao et al., 2020; Zhang H. et al., 2022).

The half-life of chlorantraniliprole was found to be 1.36 days after its use at recommended doses (Kar et al., 2013). The half-life of chlorantraniliprole was 51.3 and 62.5 days for low (1 mg kg⁻¹) and high (10 mg kg⁻¹) application dose, respectively, in a silty-loam paddy soil in subtropical China (Meng et al., 2018). Cyantraniliprole was found to be a relatively persistent compound in the soil, with a half-life of 16–89 days (EFSA, 2014). Half-life of tetraniliprole in maize plants and soil is ~9 days (Ma et al., 2021). Flubendiamide degrades to des-iodoflubendiamide by laboratory photolysis of soil with a half-life of 135.5 days (Das et al., 2012). Flubendiamide degradation in soils was reported to follow first-order kinetics, and

its average half-life in three types of soil ranged from 37.62 to 60.21 days. The persistence of flubendiamide in soils significantly increased in the following order: coastal soil > red and lateritic soil > new alluvial soils (Paramasivam and Banerjee, 2012). The half-life of photolysis of tetrachlorantraniliprole in natural water and pH buffers was 1.4–2.8 h, compared with 1.2–231 days of the half-life for hydrolysis (Zhou et al., 2022). Although diamide insecticides have been promising insecticides, their excessive use will result in their deposition into the soil (Zhang et al., 2020). Soil bioremediation is an environmentally friendly, cost-effective and efficient method compared to physical and chemical methods. Bhatt et al. (2021) showed that soil microorganisms can help degrade pesticides and be a source of energy, carbon and other nutrients (Bhatt et al., 2021). Recent results from Fahmy et al. (2022) showed that *P. aeruginosa* strain KZFS4 (LC599404.1) produced the highest CO₂ content, about 1.226 mg CO₂/16 days, with an efficiency in the biodegradation of flubendiamide-chlorantraniliprole (78.6%). Additionally, the lowest average CO₂ production rate was recorded at 0.81 mg CO₂/16 days (51.9%) for *Bacillus mojavensis* AZFS15 (Fahmy et al., 2022), while for *B. subtilis* strain AZFS3 produced 0.901 mg CO₂/16 days (57.7%). Isolated *Chryseobacterium* sp. strain SSJ1 removed 89.06% of 1,000 mg·L⁻¹ flubendiamide at optimum temperature of 35°C and pH 7.0 within 5 days of incubation period (Jadhav and David, 2016).

1.4 Organophosphates

Next, we will look at the organophosphate class of insecticides, which ranks fourth in popularity with a niche of 14% in the global market (King, 2022). Like diamides, organophosphate pesticides are degraded relatively quickly through hydrolysis on exposure to photolysis and microbial consortia. The basics of obtaining organophosphates (OP) started in the mid-19th century (Gammon et al., 2019; Laskowski, 2002), and they began to be used as insecticides in the first half of the 20th century (Abraham and Silambarasan, 2014). Among the characteristic representatives of organophosphates, the following insecticides can be distinguished: chlorpyrifos (Bragança et al., 2019; DeMars et al., 2021), monocrotophos (Li W. G. et al., 2019), malathion (Barkoski et al., 2021; Gasmī, 2020), dimethoate (Orlu, 2014; Shi et al., 2011), (Luo and Zhang, 2011; Gajendiran and Abraham, 2018) and quinalphos (Cycoń and Piotrowska-Seget, 2016). Organophosphate compounds target mostly pests such as termites (Paul et al., 2018), lice (Sanhueza-Guevara et al., 2018), flies (Galil et al., 2021; Van Laerhoven, 2008), mosquitoes (Rowland et al., 2013), and roundworms (de Albuquerque et al., 2020).

Organophosphates are capable of bioaccumulation and biomagnification, which leads to ecotoxicological effects (Jurewicz et al., 2015; Shafer et al., 2005). Chlorpyrifos was previously shown to be resistant to biodegradation and has now been shown to undergo enhanced microbial breakdown to produce less harmful and non-toxic metabolites. To date, research activity in this area has shown that a wide range of microorganisms are responsible for the degradation of chlorpyrifos (DeMars et al., 2021). The half-life of chlorpyrifos in soil is typically 60–120 days but can vary from 2 weeks to more than 1 year, depending on soil type, climate, and other conditions (Budd et al., 2020). For example, monocrotophos

has a half-life of 17–96 days and results in surface and groundwater contamination, neurotoxicity, genotoxicity, and stress effects on various organisms (Alonso et al., 2012).

The insecticide dimethoate is decomposed by microbes under anaerobic conditions. It is known that the main product of its decomposition, omethoate, has been identified. Dimethoate has been found to have adverse effects on many organisms. In plants, photosynthesis and growth are severely affected, while in birds, brain enzyme activity is suppressed (Bhatt et al., 2019). Residues of organophosphates are washed out, accumulate in the soil, and groundwater contaminates terrestrial and aquatic food webs (Geissen et al., 2015). The insecticide quinalphos acts as a neurotoxicant that inhibits acetylcholinesterase, affecting the respiratory system, skin and eyes. Quinalphos affects aquatic organisms, especially freshwater fish (Ahmad and Kumar, 2023b). To date, there are many countries which have banned the usage of greater than 10 organophosphates: Mauritania (12), Thailand (12), Guinea (12), China (15), Saudi Arabia (15), Cambodia (15) and USA (26) (Hertz-Picciotto et al., 2018). In the US, the nation of Hawaii currently banned the distribution, sale, transportation and use of all insecticides containing chlorpyrifos (Reese, 2018).

Most microorganisms can degrade one OP or a narrow range of OP compounds (Singh, 2009). The biochemistry of organophosphorus compound degradation by most of the bacteria seems to be identical, in which a structurally similar enzyme called organophosphate hydrolase or phosphotriesterase catalyzes the first step of the degradation (Singh and Walker, 2006). Microbes such as *Bacillus*, *Pseudomonas*, *Aspergillus*, *Anabaena* and *Nostoc* at 25°C–37°C use monocrotophos as a source of nutrients and cause its complete or partial decomposition to dimethyl phosphate, phosphoric acid, and acetic acid (Alonso et al., 2012). Reported that *Arthrobacter* sp. can degrade chlorpyrifos and showed 99% (100 mg·L⁻¹) decomposition within 10 h in modified mineral salt media (Mali et al., 2022). Strain *Enterobacter aerogenes* CP2 and *Streptococcus pyogenes* CP11 could remove 77% and 74% of chlorpyrifos, respectively (Lourthuraj et al., 2022). Strain *Pseudomonas* Ch1D degraded 98% of chlorpyrifos within 120 h of incubation at an initial pesticide concentration of 10 mg·L⁻¹ (Singh, 2009). Two bacterial strains, *Achromobacter xylosoxidans* JCp4 and *Ochrobactrum* sp. FCp1 were capable to degrade 84.4% and 78.6% of the initial concentration of chlorpyrifos (100 mg·L⁻¹) within 10 days, respectively (Akbar and Sultan, 2016). Moreover, it is also reported that 84.61% decomposition of quinalphos by *Ochrobactrum* sp. HZM strain can be achieved under optimum pH of seven at 27°C using response surface methodology (Mulla et al., 2020). In another study, three bacterial strains were isolated: *P. aeruginosa* PF3, *P. aeruginosa* PF2, and *Pseudomonas plecoglossicida* PF1, which have the ability to degrade profenofos. These bacterial strains individually decreased concentration of profenofos (20 mg·L⁻¹) by 95.3%, 93.1% and 95.0% within 96 h, respectively (Siripattanakul-Ratpukdi et al., 2015).

Today, organophosphates are an integral part of agriculture; their excessive use leads to contamination of ground and surface waters. That is why the search for analogues and other types of insecticides that have more gentle effects and will not cause great harm to ecosystems continues.

1.5 Carbamates

The fifth place in popularity with a niche of 6% in the global insecticide market is occupied by carbamates (King, 2022), which will be discussed further. The first carbamic acid derivatives that have insecticidal properties were synthesized in 1947. These are N-methylcarbamates, which are derived from aminofuric acid (Tian et al., 2018; Wang et al., 2019).

Among carbamate insecticides, there are about 50 chemicals (Zhao et al., 2019; Chen et al., 2012; Bhatt et al., 2020). Some striking examples include the following insecticides: carbaryl is used for control of bark beetles Scolytidae (*Dendroctonus ponderosae*, *D. adjunctus*, *D. rufipennis*) (Akbar et al., 2015; Chen et al., 2011a); aldicarb is used as a systemic insecticide, acaricide and nematocide (Zhang et al., 2011; Tang et al., 2018a; Hao et al., 2018; Mulla et al., 2017); carbofuran is used as an insecticide and nematocide (Yan et al., 2018), exhibits broad-spectrum properties, affecting various pests such as soil insects, sap-feeding insects, chewing insects, and nematodes (Rizwan et al., 2022); tefluthrin can control lepidopteran and coleopteran pests (Wen et al., 2019).

Carbamates are typically transformed into different reaction products through various pathways, such as biotransformation, biodegradation, hydrolysis, bioaugmentation, oxidation, photolysis, and metabolic reactions in living organisms (Cai et al., 2015). There are few studies on the stability or migration of aldicarb in air. However, based on their research, scientists conclude that aldicarb or its decomposition products enter the vapor phase (Tang et al., 2018b). There are also early studies that indicate that aldicarb accumulates after its use in ground and drinking water (Nugnes et al., 2023; Buszewski et al., 2019; Kurwadkar and Evans, 2016). The half-life of aldicarb in soil is about 30 days; this may vary depending on microbial populations, soil composition, humidity, temperature (Casida, 2018; Mustapha et al., 2019). It should be noted that the half-life of a few carbamates in soil lasts from 1 to 210 days (for example, carbaryl – 4–72 days, carbofuran – 1–2 days, propoxur – 80–210 days, pirimicarb – 53 days, and oxamyl – 2.5–4 days) (Mustapha et al., 2019). The insecticide carbofuran has a half-life of 16 years in acidic soil, whilst in normal, alkaline or alkaline soils, the half-life of carbofuran is 149 days (Krishnan et al., 2021). In addition, few microorganisms stay in acidic soil, so biodegradation does not longer arise naturally (Jeschke et al., 2011; Gautam and Dubey, 2023).

Cases of carbamate poisoning are most often associated with intentional oral ingestion or occupational dermal exposure. Large outbreaks due to contaminated food and crops have been reported in developing countries (Tian et al., 2018). Long-term contamination from these pesticides is unlikely, but use of the runoff by marine animals may cause harm to them (Jiang et al., 2023). For example, carbamate aldicarb was banned in the European Union since 2003, plant protection products containing this compound benefited from exemptions in France until 2007 for use on beet and vines, but ultimately aldicarb was banned in Europe in 2007 (Berny et al., 2015; Boucaud-Maitre et al., 2019).

Today, quite a lot of isolated bacteria and fungi are known to be capable of degrading a wide range of carbamates in soil and water environments (Mishra et al., 2021; Malhotra et al., 2021). Various species of microorganisms involved in the transformation or

degradation of carbamate pesticides have been reported to belong to the genera *Pseudomonas*, *Stenotropomonas*, *Micrococcus*, *Enterobacter*, *Nocardioideis*, *Pseudaminobacter*, *Serratia*, *Mucor*, *Trametes*, *Trichoderma*, *Pichia* and *Aspergillus*, amongst others. The first step in the metabolic degradation of pesticide carbamates is their hydrolysis catalysed by carboxyl ester hydrolases (Ufarté et al., 2017). The isolated *Sphingobacterium multiorum* showed high carbofuran (50 mg·L⁻¹) degradation ~74% at pH 7°C and 25°C (Tien et al., 2017). Several *Trichoderma* spp., such as *T. viride*, *T. harzianum* can metabolize carbamate insecticide (Vydat®) 200 mg·Kg⁻¹ up to 92% and 89.5%, respectively, within 14 days of incubation in soil (Helal and Abo-El-Seoud, 2015). The immobilized cells of *Enterobacter cloacae* strain TA7 showed high (100%) efficient in degradation for three carbamates: carbaryl (8.9 mg·L⁻¹), carbofuran (17.87 mg·L⁻¹), and aldicarb (22.6 mg·L⁻¹) N-methylated carbamates (Fareed et al., 2017). Recent results showed that *Sedum alfredii* associated with 20.5 mg·Kg⁻¹ of carbendazim-degrading bacterial strains enhanced carbendazim degradation up to 83.3% (Xiao et al., 2013).

However, the toxicity of carbamates entails consequences worse than people could imagine. That is why there is a search for different directions in the development and re-research of insecticides, thus their new forms and types appear from year to year. Carbamates are among the most toxic known pest control agents (Matsuda et al., 2020).

Although biodegradation of neonicotinoids, pyrethroids, diamides, OPs and carbamates are successfully performed by bacteria and fungi (*Ensifer*, *Phanerochaete*, *Bacillus*, *Trichoderma*, *Trametes*, *Aspergillus*, etc.), their fast biodegradation frequently depends on many factors (UV, pH, type of soil, etc.). Necessity of presence of multiple factors for successful biodegradation decreases speed of decomposition of carbamates and results in significant toxic load on ecosystems requiring months and even years for complete biodegradation. Obviously, use of natural molecules as insecticides can improve this situation. DNA and RNA insecticides in recent years made a huge progress in direction of creation of efficient end-products for plant protection whose biodegradation is limited to days and even hours.

1.6 Contact unmodified antisense DNA (CUAD) biotechnology (oligonucleotide insecticides)

In 2008 it was discovered that unmodified antisense DNA may serve as contact insecticide on spongy moth *Lymantria dispar* (Oberemok, 2008). This surprising discovery revealed an entirely new dimension – DNA-programmable plant protection based on tiny antisense DNAs with profound insecticidal potential. Oligonucleotide insecticides (briefly, olinscides or DNA insecticides) as a new class of insecticides has its own peculiar characteristics. Olinscides are short unmodified antisense DNA fragments that use pre-rRNA and rRNA of pests as target. A target rRNA and an olinscide interlock and resemble zipper mechanism performed by DNA-RNA duplex (“genetic zipper” method). Oligonucleotide insecticides act through DNA containment mechanism that consists of two steps: at first step antisense DNA oligonucleotide (oligonucleotide insecticide) complementarily interacts with target rRNA (in other words,

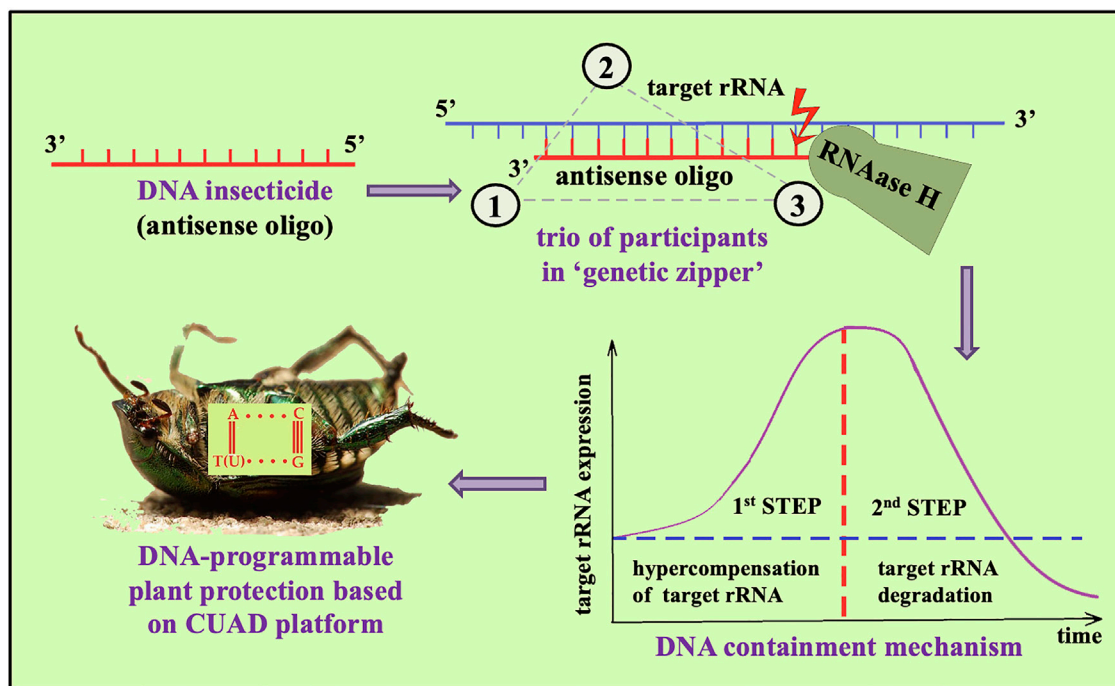


FIGURE 1
CUAD biotechnology provided by DNA insecticides.

'arrests' target rRNA) and interferes with normal functioning of ribosomes (protein machinery and precise architecture of this step remains a mystery); at second step, RNase H cleaves target rRNA and substantial decrease in its concentration occurs (Gal'chinsky et al., 2024). Species-specific oligonucleotide insecticides are basic substances of DNA-programmed plant protection and CUAD (contact unmodified antisense DNA) platform (Figure 1). Our investigations suggest that short unmodified DNA sequences can play a crucial role in regulation of rRNA expression by complementary genomic DNA (direct rDNA master regulation) and replication of DNA viruses (rRNA switchboard mechanism) (Oberemok et al., 2024) and can turn to be essential in innate immunity system (Gal'chinsky et al., 2024) against ssDNA viruses for which hemipteran insects serve as major vectors (Wang and Blanc, 2021; Wu et al., 2022) and also against DNA viruses that normally infect them (Guo et al., 2022).

Oligonucleotide insecticides represent new principle of action of pest control agents and have proven themselves most successful in the control of insect pests from the order Hemiptera; in average, mortality rate comprises 80%–90% in 3–14 days after single treatment (Gal'chinsky et al., 2024; Oberemok V. et al., 2024). The first successful experiment with olinscides within suborder Sternorrhyncha (Hemiptera) was carried out on the scale insect *Unaspis euonymi* in 2019 and pest 28S rRNA as a target for the oligonucleotide insecticide UE-11 was used (Gal'chinsky et al., 2020). Soft scale insects, armored scale insects, whiteflies, mealybugs, aphids, psyllids and other representatives of the suborder Sternorrhyncha show high mortality in response to the use of contact oligonucleotide insecticides targeting pest rRNAs (Oberemok et al., 2022; Gal'chinsky et al., 2023; Puzanova et al., 2023; Gal'chinsky et al., 2020; Useinov et al., 2020). Oligonucleotide

insecticides and DNA-programmable plant protection is developing intensively (Oberemok V. et al., 2024; Oberemok V. V. et al., 2024) and it is not unrealistic to believe that ASO-based pesticides may become a reality sooner rather than later, advancing more quickly by leapfrogging past existing technical solutions (TriLink, 2024).

Degradation of nucleic acids mainly depends on abiotic factors (water temperature, pH, salinity, and ultraviolet (UV) radiation) and biotic factors (microbes and activity of extracellular enzymes) (Barnes and Turner, 2016). It is well known that most bacteria ingest environmental DNA (eDNA) through extracellular enzymes and ectoenzymes in search of nutrients in aquatic ecosystems. Active enzymes that degrade eDNA have been found in fractions of filtered water containing bacteria, algae, cyanobacteria, fungi, as well as unicellular and multicellular planktonic animals. (Siuda and Chróst, 2001; Mauvisseau et al., 2022). For example, extracellular DNase produced by *P. aeruginosa*, ensures the degradation of extracellular DNA into an available source of carbon, nitrogen and phosphate (Mulcahy et al., 2010); all members of the group A *Streptococcus* have been shown to produce at least one extracellular DNase, and most strains make more than one distinct enzyme (Cunningham, 2000; Allan et al., 2020). Additionally, degradation is likely to be related to genomic characteristics: target region, fragment length, genomic origin (nuclear or mitochondrial), and nucleic acid composition (DNA or RNA) (Barnes and Turner, 2016; Siuda and Chróst, 2001). However, studies of the nuclease activity of target insect pests (*L. dispar*, *Icerya purchasi*, *L. decemlineata*) and their host plants (*Quercus pubescens*, *Pittosporum tobira*, *Solanum tuberosum*) have shown that most of the used olinscides degrade within 24 h at 27°C (Oberemok et al., 2019; Oberemok et al., 2022; Oberemok et al., 2023; Gal'chinsky et al., 2023); and even faster,

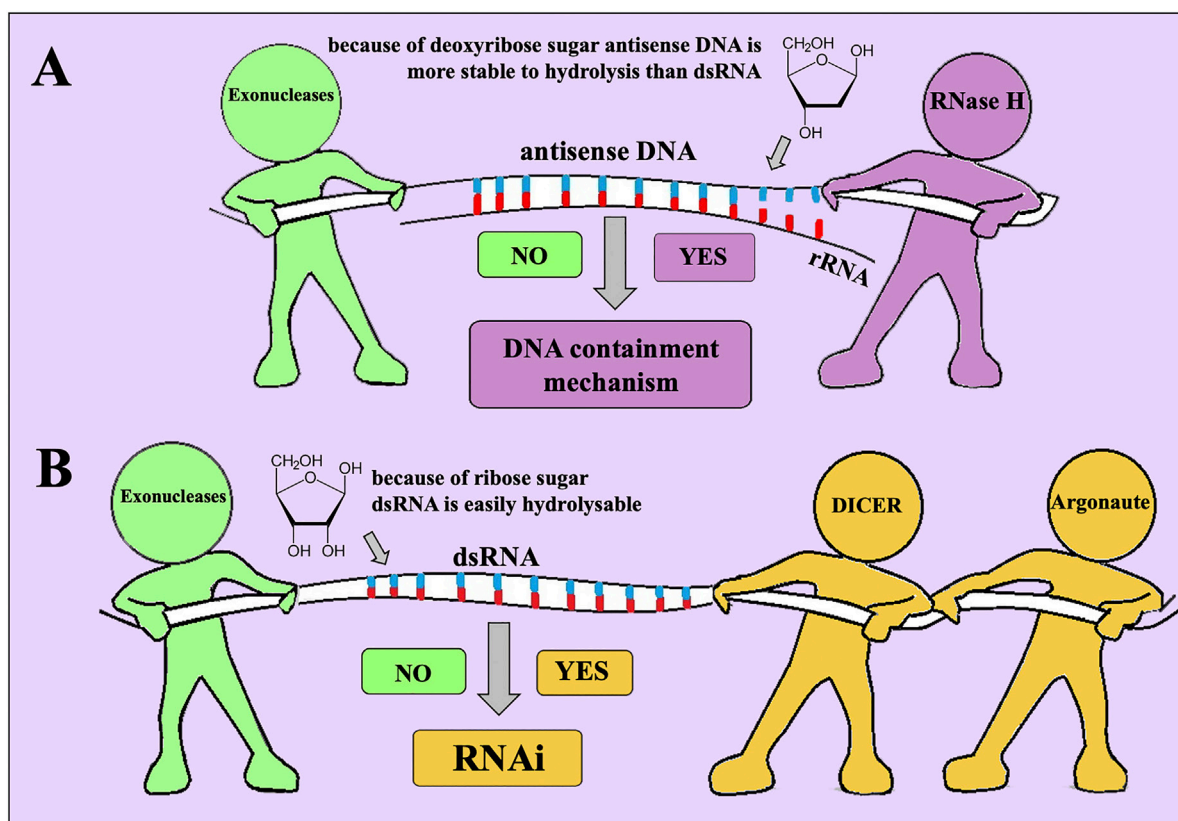


FIGURE 2
Competition between exonucleases and endonucleases together with nucleic acid-guided nucleases for their substrates: (A) exonucleases vs. DNA-guided RNase H; (B) exonucleases vs. endonuclease DICER and RNA-guided Argonaute.

within 1 h, by DNases of *Macrosiphoniella sanborni* (Puzanova et al., 2023).

Recent results demonstrated remarkable specificity of oligonucleotide insecticides in action (Oberemok et al., 2019; Oberemok et al., 2018) and showed their safety for several non-target organisms: *Quercus robur*, *Malus domestica* (Zaitsev et al., 2015), *Triticum aestivum* (Oberemok et al., 2013; Nyadar et al., 2019), *Manduca sexta*, *Agrotis ipsilon* (Oberemok et al., 2015), *Galleria mellonella* (Oberemok et al., 2019). Use of insect pest pre-rRNA and rRNA as target leads to high efficiency of oligonucleotide insecticides, since pre-rRNA and rRNA comprise 80% of all RNA in the cell (Oberemok V. et al., 2024). Thousands of different mRNAs make up only 5% of all RNA and use of pre-rRNA and rRNA for targeting substantially increases signal-to-noise ratio, ca. $10^5:1$ (rRNA vs. random mRNA). In this situation, even 11 nt long oligonucleotide insecticide provides sufficient level of selectivity in action, uniqueness frequency comprises ca. $1/4.19 \cdot 10^{-6}$ (Oberemok et al., 2022). Moreover, recent research articles have shown that it is enough to replace one nucleotide in the sequence of an oligonucleotide insecticide to substantially reduce its effectiveness, in turn, this indicates high selectivity in action (Oberemok et al., 2019; Puzanova et al., 2023; Gal'chinsky et al., 2024). The advantage of using natural oligomers, unmodified antisense oligonucleotides, seems to be the safest way, since the cells of all living organisms contain ubiquitous nucleases that can neutralize them (Oberemok et al., 2019). Consequently, for

oligonucleotide insecticides there is no need to look for methods of accelerated biodegradation. The principle of using oligonucleotide insecticides is that they must have enough time to act in the right place and on the right organism before their rapid biodegradation. In contrast, conventional chemical insecticides have too much time for their action not only in the right place and not only on the right organism. Thus, when using unmodified antisense DNA fragments, as well as double-stranded RNA fragments, in insect cells there will be competition between different types of nucleases for the substrate, mainly between exonucleases and nucleic acid-guided nucleases (Figure 2). Obviously, in the case of using DNA insecticides on sternorrhynchan pests, RNase H manages to successfully degrade the target rRNA in the DNA insecticide/rRNA duplex, which is ensured by the high concentration of the target rRNA in the insect cell (Oberemok V. et al., 2024). In addition, in the absence of nucleases, DNA is less susceptible to hydrolysis in the environment and insect cells than RNA molecules, due to the absence of an oxygen atom in the deoxyribose in the 2' position (Thorp, 2000). This chemical parameter gives antisense DNA, rather than dsRNA, the opportunity to act before being degraded.

The innovative strategy of managing pests using custom-made short oligonucleotides opens up a new horizon of "oligo-insecticides" (Patil et al., 2024) and DNA-programmable plant protection as an advanced technology. It should be noted that our research group has discovered a 50–100 nucleotides long DNA (mainly single-stranded) fraction on *P. tobira* leaves that

has significant insecticidal activity against *C. hesperidum* (Oberemok et al., 2022). This fraction has been detected for four consecutive years (2021–2024) and is an example of DNA insecticides formed in nature. Thus, under natural conditions, plant DNA can be partially biodegraded and appear in the form of previously unknown nanolayer of single-stranded DNA insecticides on the surface of plant leaves and can be an example of adaptation to sap-feeding insects having sedentary lifestyle. Natural DNA insecticides act via DNA containment mechanism blocking expression of mRNAs and non-coding RNAs that have complementary sequences to them. It is amazing that ‘genetic zipper’ method, elaborated by scientists in the lab to benefit of mankind, has its previously unknown and strikingly similar analogue in nature. At the same time, there is no example in nature of insect pest control using plant dsRNA, at least none is known until now.

1.7 RNA biocontrols (double-stranded RNA technology)

In 1998, A. Fire and C. Mello discovered that the main antiviral immune system of the roundworm *Caenorhabditis elegans* is a mechanism of post-transcriptional gene silencing, known today as RNA interference (RNAi) (Fire et al., 1998). RNA interference is a mechanism for post-transcriptional gene silencing using a double-stranded RNA copy (usually >200 bp) of the target gene (Tomoyasu et al., 2008; Svoboda, 2020).

Recent studies on plant protection by RNAi have shown that in insects such as the tephritid fruit fly (Li et al., 2011) and the beet armyworm (Tian et al., 2009), feeding dsRNA produced in bacteria causes effective knockdown of target genes and mortality. dsRNA produced by yeast (Murphy et al., 2016) and algae (Fei et al., 2020; Fei et al., 2021; Fei et al., 2023) has also been reported to trigger RNAi in mosquitoes. Calantha™ is the first dsRNA biopesticide and currently the only one that was commercially available by GreenLight Biosciences in 2023. Calantha™ is a sprayable double-stranded ribonucleic acid (dsRNA) product that targets CPB (Rodrigues et al., 2021; Pallis et al., 2023; GreenLight Biosciences, 2023).

dsRNA is degraded quickly both in soil and in the aquatic environment (Dubelman et al., 2014; Bachman et al., 2020; Mogilicherla and Roy, 2023). For example, Fischer et al. (2017), demonstrated that unformulated dsRNA generally degrades quickly once in aquatic systems due to microbial activity and hydrolysis. Within the aquatic study performed by Fischer et al. (2017) the authors note that the degradation time for 50% (DT₅₀) and 90% (DT₉₀) values for aquatic system are less than 3 days and 4 days, respectively. The aerobic aquatic degradation study for Calantha™ calculated a DT₉₀ of 6.18 and 4.2 days and DT₅₀ values of 1.87 and 1.27 days for two representative aquatic environments (Fischer et al., 2016). The aerobic representative agricultural soils degradation study for Calantha™ showed that degraded 80% dsRNA of the initial concentration for all soils tested (EPA, 2023). Rapid degradation is possible due to microbial nucleases and ultraviolet radiation present in the soil and on the leaves. Moreover, runoff from dew and rain can also significantly limit the availability of dsRNA for pests (Bachman et al., 2020; Parker et al., 2019). For example, *Staphylococcus aureus* uses environmental RNA as a building

material in specific polysaccharide-dependent biofilms (Chiba et al., 2022); *Streptococcus pneumoniae* have extracellular nuclease EndA which degrading both DNA and RNA (Gonzalez and Hernandez, 2022); *S. pyogenes* can destroy both ssDNA and dsDNA, as well as RNA, due to the extracellular nucleases they actively produce (Remington and Turner, 2018).

Studies of *Spodoptera litura* nuclease activity has shown that naked dsRNA (414 bp in the final concentration 0.05 µg/µL) is degraded by serum nucleases in 30 min, and saliva nucleases almost completely degrade dsRNA in 2 h (Peng et al., 2018). A high rate of dsRNA degradation (within 2–3 h) was also observed in the aphid *Acyrtosiphon pisum* (Christiaens et al., 2014) and the tobacco hornworm *M. sexta* (Garbutt et al., 2013). Scientists are trying to solve this problem with the help of improvements and recent discoveries in the field of chemistry and nanotechnology to hinder dsRNA biodegradation by non-specific RNases in insect cells (for example, use of dsRNA-protective agents, surfactants and diluents and nanoparticle) (Lucena-Leandro et al., 2022; Ma et al., 2024) (Figure 2). Unfortunately, such decisions can lead to more unexpected negative consequences for the environment. Some researchers showed that co-silencing of gene Zc dsRNase1 (dsRNase) along with gene ZcCOPI-alpha which plays a major role in insect development and survival, improved RNAi efficacy in *Zeugodaus cucurbitae* (led to 84% larval mortality) (Ahmad et al., 2024). A critical issue in assessing dsRNA's persistence in environmental samples is the method of dsRNA detection. So far, the methods employed are based on probe assays (Quantigene) which are not very sensitive or accurate. More sensitive methods such as RT-qPCR for dsRNA detection coupled with small RNA sequencing for the detection of small RNAs (around 20 nt) which may accumulate as intermediate degradation products upon dsRNA degradation (but remain as such undetectable by conventional dsRNA detection methods while retaining their biological potential for RNAi) need to be developed for this purpose (Dalakouras et al., 2024).

There is a general expectation that by careful selection of dsRNA length and sequence (Chen et al.), RNAi can be limited to target insect pests without any direct harmful effects on non-target species, including beneficial insects such as pollinators and biological control agents (Cagliari et al., 2019; Zotti et al., 2018). However, the expectation that dsRNA will have high specificity can be refuted by two arguments (Arora et al., 2021). First, RNAi susceptibility to heterologous dsRNA can be obtained with a single matching 21-mer sequence, require several or many matching 21-mers, be achieved with as few as 15 contiguously matching bases and be tolerant of some sequence mismatches that vary with species, gene and dsRNA concentration (Chen et al., 2021; Powell et al., 2017). Second, animals, including insects, can respond to dsRNA in a sequence-non-specific manner, i.e., independent of the sequence of the dsRNA. In particular, various studies have shown that dsRNA with no matching 15–21-mer sequences in an insect genome can have substantial effects on antiviral immunity, gene expression and performance in insects (Chen et al., 2021; Brutscher et al., 2017; Flenniken and Andino, 2013; Hirai et al., 2004; Chen and De Schutter, 2024; Joga et al., 2021). Moreover, Organisation for Economic Cooperation and Development has published two guideline documents on the biodegradability of dsRNA pesticides

and their overall risk assessment on the environment (OECD, 2020) and human health (OECD, 2023). Besides off-target effects to non-target organisms and concerns for the environment, an unexplored risk of dsRNA pesticides is that they may lead to epigenetic effects in the crops of applications, with the induced epimutations being trans-generationally inheritable (Dalakouras and Papadopoulou, 2020; Dalakouras and Ganopoulos, 2021). Therefore, it is obvious that further research is needed in this direction.

1.7.1 Biosafety of CUAD and RNAi

The double-stranded RNA molecules, like single-stranded DNA, biodegrade very fast and do not require any additional measures. Minimalist formulation containing only water and a target sequence of nucleic acids, unmodified antisense DNA or dsRNA, will have minimal negative effect on environment because of ubiquitous nucleases. Obviously, additional substances may enhance efficiency of nucleic acid-based insecticides on resistant species but can also pose additional environmental risks. To date, dsRNA-based technology does not have an easy algorithm for creation insecticides providing selectivity in action and high efficiency like CUAD biotechnology does on several groups of pests. Compared to short antisense DNA, long and fragile dsRNAs are more unpredictable and complex that is why it is not easy to use them as selective and efficient practical tool for plant protection. In addition to better target specificity, antisense oligonucleotides are easier and cheaper to synthesize than siRNAs or dsRNAs, and in the human system, antisense oligonucleotides have been shown to have lower immunoreactivity, which is also of significant importance for potential applications in edible plants that could accumulate this kind of means of insect pest control. It is important to note, from the point of view of environmental risks, there are no findings in the numerous human clinical studies that prove, for example, genomic integration events attributable to the use of antisense oligonucleotides (Gruber et al., 2023).

1.7.2 Limitations and prospects of CUAD and RNAi in plant protection

At present, there are no absolute rules for improving dsRNA efficiency (Zhao et al., 2024). For dsRNA design, the three-dimensional structures of RNA molecules that are critical to their function should be considered. Despite decades of intense effort, few RNA structures are known, and predicting the structure of RNAs remains a great challenge (Townshend et al., 2021). Until today, DNA-programmable plant protection based on oligonucleotide pesticides is the only simple and successful approach for pest control based on nucleic acids. It is easy and efficient algorithm for plant protection at least against many sternorrhynchans and mites (Oberemok V. et al., 2024; Oberemok V. V. et al., 2024). Advances in the direction of CUAD biotechnology ('genetic zipper' method) will help to expand this approach on other groups of pests providing easy biodegradation for the pest control agents after efficient and selective action. Today, dsRNA insecticides are perceived as 'difficult' insecticides, whilst oligonucleotide insecticides (ssDNA insecticides) are considered as 'easy' insecticides. While dsRNA insecticides need easy and efficient algorithm and more

groups of pests to show their effectiveness on, ssDNA insecticides require only the latter. Both technologies individually can become successful for insect pest control on distinct groups of pests and also can complement each other's action in complex pest control formulations. In our opinion, in short-term perspective (5–10 years) insecticides based on CUAD and RNAi are supposed to occupy not more than 5% of global insecticide market but will help make a quantum leap towards eco-friendly plant protection.

2 Conclusion

It should be noted that in eukaryotic and prokaryotic cells, DNA and RNA consist about 1% of the mass. However, studies show that they do not accumulate in ecosystems and amount to only 1 millionth of the mass of soil or marine water. This suggests that nucleic acids do not accumulate in nature but are biodegraded to the simplest monomers and are then used to construct new sequences of DNA and RNA within living organisms. From this point of view, the use of DNA and RNA as insecticides fits into the framework of natural processes and is a safe strategy for controlling insect pests. It is obvious that the need to use insecticides will not decrease but will only increase against the backdrop of an increase in population and a reduction in areas for agriculture. Biodegradability, or in other words neutralization, of insect pest control agents introduced into ecosystems must be at a high level so as not to jeopardize non-target organisms and human health (Guerrero Ramírez et al., 2023). Back in the 20th century, the possibility of creating insecticides based on natural polymers (such as DNA and RNA) to control a specific insect pest with minimal negative effects on non-target organisms seemed fantastic. This was impossible either conceptually or economically. Today this is reality. Time will tell how far we can progress in this direction, but the prospect looks encouraging: natural molecules act on natural systems through natural mechanisms and are utilized by fast "biodegradation micro machines" (bacteria and fungi). The share of modern chemical insecticides (neonicotinoids, pyrethroids, diamides, organophosphates, etc.) will obviously decrease against the background of the expected increase in the popularity of insecticides based on CUAD biotechnology and double-stranded RNA technology. However, from the point of view of increasing the level of biodegradability of modern chemical insecticides, improvements are still possible by creating molecules with a biomimetic structure, as well as obtaining bacterial strains with a high potential for biodegradation of certain substances. Thus, it is obvious that all conventional classes of insecticides will continue to exist in the medium term (5–10 years), and one would like to hope that ways will be found for them to compete with insecticides based on DNA and RNA in the level of biodegradability. We are on the verge of creating such plant protection products, the effectiveness of which will be determined by their safety, and CUAD biotechnology and double-stranded RNA technology are on the way to show an eco-precedent.

Author contributions

VO: Conceptualization, Funding acquisition, Resources, Supervision, Validation, Writing–original draft, Writing–review and editing. KL: Writing–original draft, Writing–review and editing. OA: Writing–original draft, Writing–review and editing. NG: Supervision, Validation, Writing–original draft, Writing–review and editing.

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Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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