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# Impact of microplastics on the *in situ*, high-resolution of key nutrient dynamics at the soil-water interface in rice fields

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**Introduction:** Microplastics are characterized by their small size, widespread distribution, and durability, present a significant environmental risk. Despite their omnipresence in terrestrial and aquatic systems, the potential consequences on nutrient cycling remain under-investigated. Microplastics have emerged as a focal point of current research, presenting both a challenge and a frontier in environmental science.

**Methods:** This study explores the effects of microplastics on the high-resolution, *in situ* distribution and exchange dynamics of key nutrients, nitrogen (N) and phosphorus (P), at the soil-water interface in rice paddies, utilizing the Diffusive Gradients in Thin-films (DGT) technique.

**Results:** Our results reveal distinct spatial distribution patterns for N and P across the soil-water interface. Labile phosphorus (P) concentrations were significantly higher in the soil than in the overlying water, whereas DGT-NO<sub>3</sub><sup>-</sup> concentrations exhibited the inverse trend. Different microplastic concentrations notably impacted DGT-NO<sub>3</sub><sup>-</sup> ( $P = 0.022$ ) and DGT-NH<sub>4</sub><sup>+</sup> ( $P = 0.033$ ), with an increase between 27.79% and 150.68%. Moreover, different particle sizes significantly influenced NH<sub>4</sub><sup>+</sup>. Interestingly, paddy soil acted as a “source” for labile P and a “sink” for NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup>.

**Discussion:** These insights provide valuable insights into the interactions between microplastics and nutrient cycles at the soil-water interface, and assess the effects on nutrient migration and transformation. The outcomes of this study will contribute to an improved understanding of the broader ecological implications of microplastic pollution in agricultural settings. It will also provide a foundation for the development of strategies to manage and mitigate the impacts of microplastic pollution in agricultural soils, particularly in rice dominated agroecosystems.

## KEYWORDS

soil-water interface, diffusive gradients in thin-films, exchange fluxes, microplastics, source-sink characteristic

# 1 Introduction

Microplastics (MPs), initially termed by Professor Richard Thompson in 2004 (Thompson et al., 2004), are heterogeneous plastic particles less than 5 mm in diameter. Owing to their low density, diminutive size, and lightweight nature (Thompson et al., 2004), MPs have proliferated extensively, permeating diverse environments such as soils, water bodies, and the atmosphere (Fuller and Gautam, 2016). Recent research has detected the presence of microplastics in human tissues, including the lungs, placenta, and blood (Cavaliere and Baulch, 2018; Amato-Lourenço et al., 2021; Leslie et al., 2022), suggesting their widespread human exposure. Additionally, traces of MPs have been detected in seemingly pristine environments like Antarctic snowfall, the abyssal regions of the North Atlantic, and even the summit of Mount Everest (Pabortsava and Lampitt, 2020), revealing their global pervasiveness. Microplastics in the environment are susceptible to self-induced physical adsorption, with potential long-term persistence and accumulation in soil matrices. As accumulated microplastic concentrations escalate to specific thresholds, they could impede the growth, development, and reproductive processes of various organisms (Qiguo et al., 2006), presenting a potential hazard to human health and safety (Lombardi et al., 2022). Consequently, microplastics have emerged as a focal point of current research, presenting both a challenge and a frontier in environmental science. (Tirkey and Upadhyay, 2021; Davtalab et al., 2023).

China, in 2020, accounted for approximately 70% of the world's mulch usage, the largest global share, equating to around  $135.7 \times 10^4$  t (Jing et al., 2023). Agricultural mulch is resistant to natural degradation processes, leading to widespread residual deposits in the soil, and persisting for decades, or even centuries (Qiguo et al., 2006). Microplastic density in soil has been reported to be between 4 and 23 times higher than in marine environments, yet soil microplastic pollution reports represent less than one-third of those related to the ocean (Yang and He, 2021). The infiltration of microplastics into soil significantly impedes the activities of soil microorganisms, disrupting key ecological processes like nitrogen and phosphorus nutrient cycling, thereby affecting seed germination and plant growth (De Silva et al., 2022). Microplastics impact the nitrogen and phosphorus cycling flora function, mainly through the activity of soil microorganisms and associated enzymes, thereby affecting soil nitrogen transformation, as well as the broader environmental nitrogen transformation via effects on nitrification and denitrification processes (Seeley et al., 2020). The influence of microplastics on soil is multifaceted and associated with various factors, such as microplastic particle size and concentration (Zhang et al., 2014).

The soil-water interface in paddy fields is an active ecological junction where nitrogen and phosphorus cycling frequently occurs and is influenced by environmental factors like pH, redox potential, and dissolved oxygen (Fan et al., 2004). It has been found that microplastic addition in soils disrupts the nitrogen and phosphorus cycles (Hu et al., 2021; Yin et al., 2023). Contrary to the prevailing notion that microplastics negatively impact the environment, some suggest that microplastics could contribute to soil carbon stocks, thus influencing the biogeochemical cycling of nitrogen (Lu et al., 2021). Yet, the addition of microplastics disrupts the ecological health of paddy

soils, reduces soil microbial diversity, and may negatively impact the migration and release of nitrogen and phosphorus (Wang et al., 2022). Therefore, a more thorough examination of the effects of microplastics on nitrogen and phosphorus migration and release at the soil-water interface in rice fields is necessary.

Previous studies have found that microplastic addition reduced microbial activity and altered enzymatic activities associated with the nitrogen cycle with increased Ammonia monooxygenase activity and decreased nitrate reductase activity (Huang et al., 2020), reducing the release of nitrogen and phosphorus from the sediment to the overlying water (Yu et al., 2022). Microplastics can alter the dispersion and surface properties of sedimental minerals (Lu et al., 2023), thereby may influence nutrient accumulation and transportation at the sediment-water interface. Moreover, microplastic addition restrained worm bioturbation activities and thus decreased the contribution of tubificid worms on organic matter mineralization and nutrient fluxes at the sediment-water interface (Wazne et al., 2023). However, current research primarily focuses on the impact of microplastics on nitrogen and phosphorus migration and transformation in laboratory experimentations, with limited *in situ* studies evaluating the effect of microplastics on nitrogen and phosphorus concentration and distribution in the soil-water interface in rice fields.

Thin Film Diffusion Gradient Technology (DGT), an *in situ* passive sampling technique, is based on the Fick's first law for measuring non-stable components that can pass the diffusion layer and be accumulated by the binding layer, namely the labile concentrations (Zhang et al., 2014). It has been effectively used for *in situ* quantification of labile concentrations of elements in the water-sediment and soil interface zones (Li et al., 2020). Recently developed multi-phase DGTs (e.g., ZrO-Chelex and ZrO-AgI) enhance simultaneous acquisition of multiple target analytes, making it a powerful tool for understanding crucial elemental biogeochemical processes (Han et al., 2015; Wang et al., 2017).

Northeast China, known as an important commercial grain base and one of four major black soil areas in the world, has been reported to be increasing in the usage of agricultural mulch, leading to issues of land quality degradation and microplastic pollution (Yan et al., 2022). In this study, we propose to utilize DGT technology to investigate the concentration and distribution of labile nitrogen and phosphorus at the soil-water interface in rice fields. We will consider black soil regions with different sizes of microplastic particles (10  $\mu$ m and 80  $\mu$ m) and varying concentrations, and aim to elucidate the distribution pattern and source-sink characteristics of microplastics in the soil-water interface of these rice fields.

The goals of this study were to provide valuable insights into the interactions between microplastics and nutrient cycles at the soil-water interface, and assess the effects on nutrient migration and transformation. The outcomes of this study will contribute to an improved understanding of the broader ecological implications of microplastic pollution in agricultural settings. It will also provide a foundation for the development of strategies to manage and mitigate the impacts of microplastic pollution in agricultural soils, particularly in rice-dominated agroecosystems. Given the scale of rice farming in China and the increasing reliance on agricultural mulch, our findings will hold significant importance, not only for the ecological health of our farmlands, but also for the sustainability of our food production systems.

## 2 Materials and methods

### 2.1 Experimental design and sample analysis

The experiments were conducted in the experimental greenhouse base from April to October 2021. Dry soil was thoroughly mixed and placed in flowerpots of 25 cm in length, 20 cm in width, and 28 cm in height with 8 kg for each pot. Five full-height seedlings of the Longjing 31 rice variety were planted in each pot. The experiment included four microplastics (MPs; polyvinyl chloride, PVC) treatments and a control group (CK), with each group consisting of four replicates. Microplastics of differing concentrations and particle sizes were added, as follows: 5 g/kg of 10  $\mu\text{m}$  (high concentration addition with small size, HS), 0.5 g/kg of 10  $\mu\text{m}$  (low concentration addition with small size, LS), 0.5 g/kg of 80  $\mu\text{m}$  (low concentration addition with large size, LL), and 5 g/kg of 80  $\mu\text{m}$  (high concentration addition with large size, HL). The pots were randomly arranged in the greenhouse, and the rice plants were watered following standard management practices.

All the ZrO-AT DGT devices were provided by EasySensor Ltd (Nanjing, China). After rice maturation in October, the ZrO-AT binding gel devices were carefully inserted vertically into the soil-water interface within each treatment, allowing for a 72-h deployment period. During this deployment period, water temperature measurements were collected in triplicate. Subsequently, all the DGT probes were retrieved from the soil and rinsed meticulously with deionized water. The ZrO-AT binding gel pieces were then precisely sliced into 2 mm intervals using a ceramic cutter, following the established protocol as described in previous studies (Menzies et al., 2005; Ding et al., 2015). Each slice was subjected to elution using 1.0M NaOH, facilitating the subsequent analysis of phosphorus (P) concentrations through the application of the phenanthroline colorimetric method. Similarly, the concentrations of DGT-ammonium ions ( $\text{NH}_4^+$ ) and nitrate ions ( $\text{NO}_3^-$ ) were determined using a segmented flow analyzer.

### 2.2 Data analysis

The concentrations of labile nitrogen and phosphorus at the soil-water interface of the paddy field were calculated according to the following formula  $C_{DGT}$  (mg/L) (Heidari et al., 2017).

$$C_{DGT} = \frac{M \times \Delta g}{D \times A \times T}$$

where  $M$  represents the total amount of target analyte adsorbed by DGT (mg);  $\Delta g$  is the total thickness between the diffusion and the filter membrane;  $D$  is the diffusion coefficient of the target substance within the diffusion film ( $\text{cm}^2 \cdot \text{s}^{-1}$ );  $A$  is the opening area of DGT device ( $\text{cm}^2$ ); and  $T$  represents the placement time (s) of the DGT device.

Exchange fluxes ( $F$ ) at the sediment-water interface were estimated according to Fick's first Law, as follows (Xu et al., 2012):

$$F = F_s + F_w = \left( -\varphi D_s \frac{\partial c_s}{\partial x_s} \right) + \left( -D_w \frac{\partial c_w}{\partial x_w} \right)$$

where  $F$  is the material exchange flux at the interface ( $\text{mg}/(\text{m}^2 \cdot \text{d})$ );  $F_s$  represents the exchange flux in the sediment;  $F_w$  is the exchange flux in the overlying water;  $\partial c_s / \partial x_s$  denotes the concentration gradient in the sediment;  $\partial c_w / \partial x_w$  denotes the concentration gradient in the overlying

water;  $x_s$  is the sediment depth (cm);  $c_s$  is the ion concentration in the sediment (mg/L);  $x_w$  is the depth of overlying water (cm);  $c_w$  is the ion concentration in the overlying water (mg/L); and  $D_s$  is the diffusion coefficient in the sediment ( $\text{cm}^2/\text{s}$ ), which can be calculated from the molecular diffusion coefficient in the water ( $D_w$ ) and the porosity of the surface sediment ( $\varphi$ ).

The Origin software (Origin 2021) was employed to construct the distribution profiles of labile nitrogen and phosphorus concentrations at the soil-water interface. Two-way ANOVA was performed using SPSS Statistics 26 software (IBM Corporation, USA) to assess the effects of different concentrations and particle sizes of microplastics, and their combined effects on the labile nitrogen and phosphorus concentrations. The effects of microplastic addition were further analyzed by the Duncan's multiple comparison ( $p < 0.05$ ).

## 3 Results

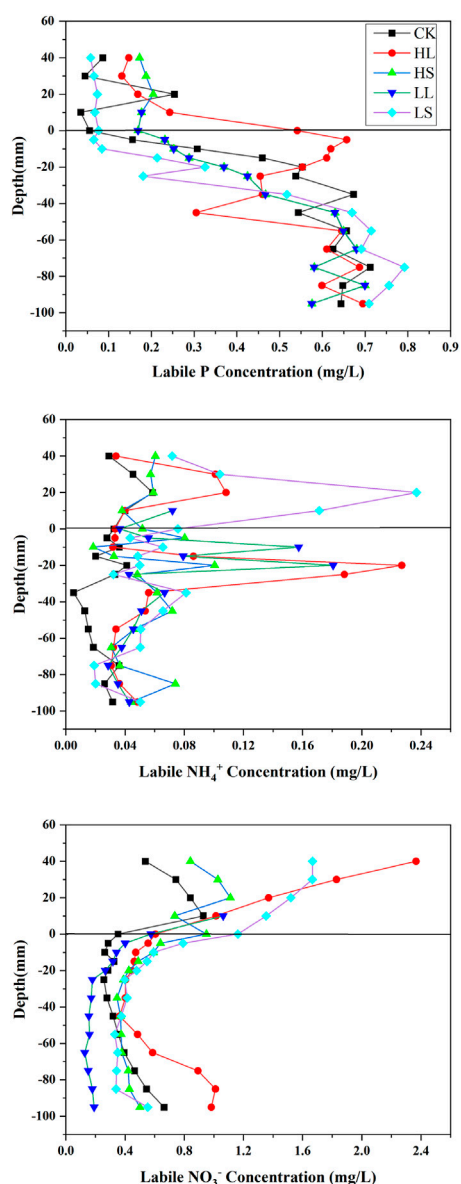
### 3.1 Distribution patterns of labile nitrogen and phosphorus at the soil-water interface under different microplastic treatments

The distribution patterns of nitrogen (N) and phosphorus (P) concentrations at the soil-water interface in rice fields under different microplastic treatments are shown in Figure 1. The concentrations of labile P ranged from 0.03 to 0.79 mg/L,  $\text{NH}_4^+$  from 0.01 to 0.24 mg/L, and  $\text{NO}_3^-$  from 0.13 to 2.37 mg/L (Figure 1). The overall trend of labile P remained consistent across different microplastic addition treatments, showing low concentrations in the overlying water but increasing with soil depth with a peak at a soil depth of 75–85 mm. The concentration of  $\text{NH}_4^+$  at the soil-water interface exhibited relative stability, while the concentration of  $\text{NO}_3^-$  in the overlying water decreased with depth, reaching a minimum value at the soil-water interface, and then exhibited an increasing trend with further soil depth.

Our results reveal complex relationships between the presence of microplastics and the labile N and P concentrations in our experimental conditions (Figure 2). The addition of different concentrations and particle sizes of microplastics had significant effects on  $\text{NH}_4^+$  and  $\text{NO}_3^-$  concentrations (Table 1), with no significant effect on the labile P ( $p > 0.05$ ) (Table 1). In comparison with CK, the labile  $\text{NH}_4^+$  concentration increased from 84.83% to 150.68%, and  $\text{NO}_3^-$  increased from 27.79% to 81.80% under different treatments (Figure 2). Furthermore, the interaction effects of microplastic concentration and particle sizes had a significant effect on the labile  $\text{NO}_3^-$  concentration ( $p < 0.001$ ).

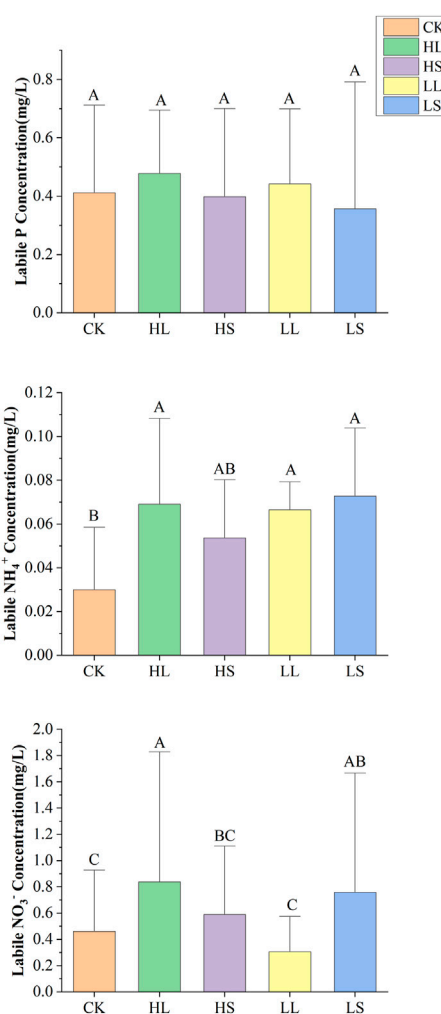
### 3.2 Sink-source dynamics of labile N and P fluxes under microplastic treatments

The  $\text{NH}_4^+$ ,  $\text{NO}_3^-$  and P fluxes were determined based on Fick's first law (Figure 3). Positive flux values imply that the labile N and P are being released from the soil to the overlying water, hence acting as a source. Conversely, negative values suggest that nutrients migrate from the overlying water into the soil, acting as a sink. The diffusion flux of labile P varied from  $-0.09$ – $0.64$   $\text{mg}/\text{d} \cdot \text{m}^2$  in all



**FIGURE 1**  
Trends in labile N and P distributions at the soil-water interface in rice paddy across under different microplastic treatments. The x-axis represents the depth of overlying water (positive values) and the depth of sediments (negative values). CK, control group; HS, high concentration of microplastics, 5 g/kg, small size–10 μm; LS, low concentration of microplastics, 0.5 g/kg, small size–10 μm; LL, low concentration of microplastics, 0.5 g/kg, large size–80 μm; and HL, high concentration of microplastics, 5 g/kg, large size–80 μm. Note: The depth for treatment LL was shorter 25 mm due to accidentally broken.

treatments, and the mean value for different treatments was 0.32 mg/d·m<sup>2</sup>, indicating that the labile P source came from soil. The diffusion fluxes of labile NH<sub>4</sub><sup>+</sup> ranged from –9.52 to 0.75 mg/d·m<sup>2</sup>, and the mean value for different treatments was –2.17 mg/d·m<sup>2</sup>. The diffusion fluxes of labile NO<sub>3</sub><sup>–</sup> ranged from –2.82 to 1.22 mg/d·m<sup>2</sup>, and the mean value for different treatments was –0.85 mg/d·m<sup>2</sup>. Thus, both of NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>–</sup> exhibited the characteristics of “sink” in soil.



**FIGURE 2**  
Concentrations of labile N and P at the soil-water interface in rice fields under microplastic treatment. Different uppercase letters represent significant differences.

The mean values of diffusion fluxes of labile N and P fluxes in the CK treatment were positive, while the addition treatments of different concentrations and particle sizes of microplastics changed the direction of fluxes (Figure 3). Especially, the diffusion flux of NH<sub>4</sub><sup>+</sup> with a low concentration of 80 μm (LL group) changed from 0.03 to –9.52 mg/d·m<sup>2</sup>, indicating a shift from source to sink. The addition of microplastics also enlarged the NO<sub>3</sub><sup>–</sup> flux from 0.72 to 1.22 mg/d·m<sup>2</sup> in the LL treatment.

## 4 Discussion

### 4.1 Distribution patterns of vertical gradient of nutrients in overlying water and soil

Distinct spatial heterogeneity in the concentrations of labile P, NH<sub>4</sub><sup>+</sup>-N, and NO<sub>3</sub><sup>–</sup>-N at the soil-water interface were found via the DGT technique. The fact that concentrations of labile P increased with soil depth is primarily attributed to the process of leaching in

TABLE 1 Two-way ANOVA for labile N and P at the soil-water interface in rice fields under microplastic addition treatments. \*:  $p < 0.05$ ; \*\*:  $p < 0.01$ ; \*\*\*:  $p < 0.001$ .

	MPS concentration	MPS particle size	Interaction items
DGT-P	0.233	0.236	0.873
DGT-NH <sub>4</sub> <sup>+</sup>	0.033*	0.041*	0.090
DGT-NO <sub>3</sub> <sup>-</sup>	0.022*	0.146	0.000***

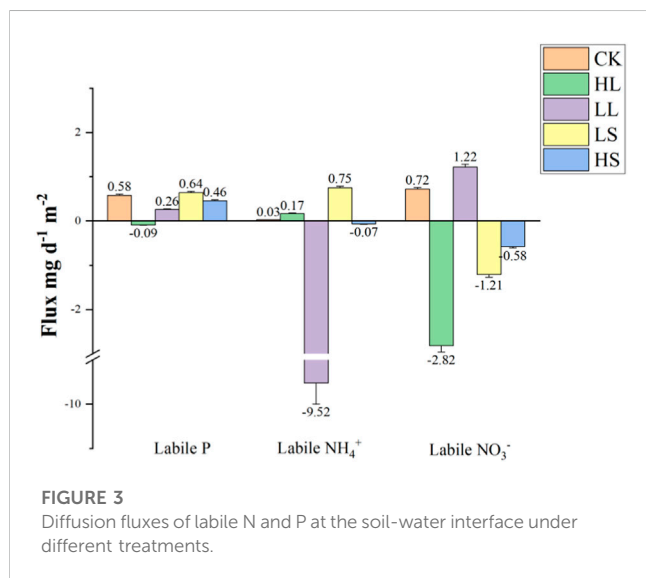


FIGURE 3  
Diffusion fluxes of labile N and P at the soil-water interface under different treatments.

the soil profile and the uptake of rice. Our results indicated there were no significant effects of microplastic addition on labile P concentrations, which is consistent with previous study that PVC MPs have no significant short-term effect on phosphorus cycling in soil-water systems (Yin et al., 2023).

The concentration trend of NH<sub>4</sub><sup>+</sup> at the soil-water interface remained relatively stable, while NO<sub>3</sub><sup>-</sup> concentration in the overlying water was significantly higher than those in the soil. Because the surface of soil particles is capable of absorbing NH<sub>4</sub><sup>+</sup>, this adsorption can limit NH<sub>4</sub><sup>+</sup> migration and leaching in the soil, so as to keep NH<sub>4</sub><sup>+</sup> concentration at the interface relatively stable (Liu et al., 2008). In addition, negatively charged ion-exchange complexes such as clay minerals and organic matter were present in the soil. These negative charges are capable of binding NH<sub>4</sub><sup>+</sup>. The addition of microplastics increased soil porosity, increased oxygen flux, and increased nitrification (Lu et al., 2021). Besides, oxygen is usually sufficient in the overlying water, which is conducive to the nitrification process and increases the concentration of NO<sub>3</sub><sup>-</sup> in the overlying water (Wang et al., 2002). Microplastics also lead to increased nitrate reductase (NR) activity in soil, which affects nitrogen utilization and promotes labile NO<sub>3</sub><sup>-</sup> to labile NH<sub>4</sub><sup>+</sup> conversion (Yu et al., 2022). Moreover, MPs used in our study can promote the abundance of microbial denitrification genes in the soil (Yin et al., 2023), leading to low NO<sub>3</sub><sup>-</sup> concentrations in soil.

The penetration depth of dissolved oxygen in soil is usually limited to a few centimeters of the sediment depth, and the environment becomes more reduced as depth increased and oxygen is depleted. Under anoxic conditions in deep soil, the need for an electron acceptor for the decomposition of organic matter by the obligate anaerobic

bacteria and anaerobic bacteria may lead to the reduction of labile NO<sub>3</sub><sup>-</sup> as an alternative electron acceptor (Pabortsava and Lampitt, 2020). The concentration of labile NO<sub>3</sub><sup>-</sup> in overlying water was significantly higher than that in the soil.

The addition of microplastic significantly increased the concentrations of NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup> at the soil-water interface (Figure 2), because microplastics can act as organic substrates and release of additives in microplastics (Seeley et al., 2020). The surfaces on the microplastic particles provide many available biofilms that are ideal environments for the attachment and growth of nitrifying bacteria. This process of attachment and growth increases the number of nitrifying bacteria and further catalyzes the nitrification reaction, this leads to an increase in labile NO<sub>3</sub><sup>-</sup> concentration (Chen et al., 2020). PVC microplastics can reduce the abundance amoA, nirS and nirK, which leads to an increase in labile NH<sub>4</sub><sup>+</sup> concentrations (Seeley et al., 2020). It was reported that microplastics affect microbial community composition, enzyme activity and gene abundance associated with nitrogen cycling, which in turn affects nitrogen cycling processes and increases NH<sub>4</sub><sup>+</sup> concentrations (Zhang et al., 2014; Liu et al., 2017; Ren et al., 2018). PVC microplastics reduce the concentration of denitrifying bacteria and inhibit the removal of NO<sub>3</sub><sup>-</sup> (Ding et al., 2019).

## 4.2 Exchange fluxes of labile nitrogen and phosphorus under different treatments

The average phosphorus exchange flux at the soil-water interface presented a positive value, implying that phosphorus diffused from soil to water. Notably, the presence of microplastics did not significantly influence the flux of phosphorus, regardless of the differences in microplastic concentration and particle size. Conversely, the average exchange fluxes of NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup> were negative, demonstrating a “sink” characteristic, where these substances migrated from water to soil. The translocation of labile nitrogen from the water to the soil was dictated not only by the concentration gradient at the soil-water interface in the rice fields, but also by microplastic adsorption (Sun et al., 2022). PVC microplastic adsorption in the soil obstructed the transfer of labile nitrogen from the soil to the water surface (Sun et al., 2022), and thus promoted the soil as a sink of nitrogen.

Smaller microplastic particles are more likely to impact soil porosity and aeration (Kumar et al., 2022). This not only accelerates the flux rate of labile nitrogen, but also enhances oxygen diffusion, thereby promoting nitrification and reducing NH<sub>4</sub><sup>+</sup> concentrations. High concentrations of microplastics can alter enzymatic activities related to nitrification and

denitrification, facilitating the conversion of  $\text{NO}_3^-$  to  $\text{N}_2$ . Besides, compared with low MPs concentration treatments, high MPs concentrations can strongly affect  $\text{NH}_4^+$  metabolism via altering microbial denitrification genes abundance, and influence  $\text{NO}_3^-$  metabolism via regulating the abundance of *nirS* and *nirK* genes and Nitrospirae related to nitrogen cycle (Sun et al., 2022). Moreover, the migration of active nitrogen from the water column to the soil is driven by both oxygen enrichment and concentration gradients in the water (Zhu et al., 2022). Overall, microplastic particle size and concentrations strongly impacted nutrient cycles at the soil-water interface, particular in nitrogen.

## 5 Conclusion

In this study, the thin-film diffusion gradient technique was employed to obtain *in situ*, high-resolution information about key nutrients at the soil-water interface in the rice paddy. There were distinct spatial distribution patterns for labile N and labile P across the soil-water interface under different treatments. The addition of microplastic significantly increased the concentrations of  $\text{NO}_3^-$  and  $\text{NH}_4^+$  by affecting the nitrification and denitrification processes. Additionally, the presence of microplastics significantly changed the source and sink characteristics of  $\text{NO}_3^-$  and  $\text{NH}_4^+$  fluxes. Our study offers a theoretical foundation for understanding the *in situ* distribution patterns and source-sink characteristics of nutrients impacted by microplastics at the soil-water interface in rice fields.

## Data availability statement

The raw data supporting the conclusion of this article will be made available by the authors, without undue reservation.

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## Author contributions

YJ and XZ wrote the manuscript. XZ proposed and structured the study. YJ, SZ, JY, and YW performed the experiments. SZ, JY, and YW made the figures. XZ and YL reviewed and improved the language of the manuscript and secured funding acquisition. All authors contributed to the article and approved the submitted version.

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## Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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