



# Analysis of Groundwater Chemical Characteristics and Spatiotemporal Evolution Trends of Influencing Factors in Southern Beijing Plain

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During the rapid urbanisation of Beijing in the past two decades, the original natural underlying surface has been changed, and the groundwater recharge and discharge processes had undergone great changes. Therefore, whether these natural and human factors affect the evolution of groundwater chemical composition, and how these influences evolve, need to be studied urgently. In this study, the Daxing District was taken as a typical area to understand the evolution rules of the chemical characteristics of shallow groundwater in the plain areas of Beijing. By collecting monitoring data on shallow groundwater from 2006 to 2016, the spatiotemporal evolution trends of the chemical characteristics and influencing factors of shallow groundwater in Daxing District were studied using the hydrochemical analysis methods and principal component analysis–partial connection number method. The results show the following: 1) The cations and anions of groundwater in the study area increased with time, indicating that the groundwater quality tends to deteriorate. 2) The hydrochemical type and anion and cation components changed accordingly with the change in  $\text{NO}_3^-$ , indicating the profound impact of  $\text{NO}_3^-$  on the evolution of groundwater components. 3) The groundwater has always been affected by human factors and natural factors. Natural factors have dominated, but the influence of human factors on hydrochemical components has been increasing. From the spatial distribution of the impact of human factors and natural factors on groundwater composition, regions with higher urbanisation degrees are highly affected by human factors and vice versa. This study can provide a theoretical and methodological reference for regional groundwater management and industrial layout.

**Keywords:** groundwater chemical characteristics, groundwater pollution, human factors, Southern Beijing plain, spatiotemporal evolution trend

## 1 INTRODUCTION

Rapid urbanisation and rapid economic development happened in China in recent years, but these also led to a series of environmental problems. For example, the rapid growth of the urban population and the continuous expansion of urban construction leads to the sharp increase in urban water supply demands, sewage discharge, domestic waste and so on (Wang and He, 2015; Li et al., 2020a). Beijing, the political, economic and cultural centre of China, is located in the northern part of the North China Plain. Because of the lack of surface water in the region, more than 70% of the

water supply is mainly from groundwater (Ma et al., 2015). Long-term overexploitation of groundwater, coupled with the rapid development of cities, has led to the continuous decline of groundwater level (Zhou et al., 2016), deterioration of water quality (Li et al., 2020a) and surface subsidence (Zhou et al., 2015; Qin et al., 2018; Hu et al., 2019). Since the South-to-North water diversion into Beijing in December 2014, the continuous deterioration of the groundwater environment has been partially alleviated (Zhang et al., 2018a; Zhang et al., 2022), but groundwater supply still accounts for more than 40% of the city's total water supply, and the status of groundwater as the main source of the water supply has not changed.

As the uppermost layer of groundwater, shallow groundwater is closely related to the external environment and is easily affected by external natural and human factors (Baojing et al., 2013; Li et al., 2013; Martín et al., 2014; Shang et al., 2016; Zhang et al., 2016), such as atmospheric precipitation, formation lithology, surface pollutant leakage and human activities. In particular, during the rapid urbanisation of Beijing in the past two decades, the original natural underlying surface has been changed, and the groundwater recharge and discharge processes had undergone great changes (Ji et al., 2021; Qin, 2021). Therefore, studying whether these natural and human factors affect the evolution of groundwater chemical composition and changing their influence degrees are urgent. Therefore, studying the dynamic evolution of groundwater chemical characteristics and influencing factors from the perspective of hydrochemistry is necessary to provide theoretical and data support for regional groundwater resource management and protection.

Hydrochemical methods play an important role in understanding the chemical characteristics and formation processes of regional groundwater (Sarkar et al., 2017; Li et al., 2020b). Many scholars have studied the regional hydrogeochemical processes, the main sources of anions and cations, the direction of development and utilisation and the potential risks of groundwater to human health using hydrochemical methods combined with multivariate statistical methods (Li et al., 2019), geostatistics methods (Yin et al., 2019), isotope methods (Li et al., 2016), water quality assessment (Zhai et al., 2015) and other methods (Adimalla and Qian, 2019; Ji et al., 2020; Podgorski and Berg, 2020). For example, to determine the pollution sources of groundwater, hydrochemistry and multivariate statistics are combined to consider human factors in the hydrogeochemical process (Busico et al., 2020). By combining hydrochemical methods with drinking water or irrigation water standards, the applicability of regional groundwater for living and irrigation purposes can be understood, and the impact of toxicity on human health can be further evaluated (Zhang et al., 2018b; Adimalla and Li, 2019; Liu et al., 2019). Hydrochemistry and stable isotope methods can be combined to analyse the hydrochemical evolution of an area to understand the influence of wastewater irrigation on the hydrochemical processes in the surrounding area (Li et al., 2016). Therefore, the hydrochemical method is the basic method used to realise the sustainable utilisation of groundwater. However, at present, there are few studies on the

evolution of groundwater chemical components over time in the process of rapid regional urbanization (Zhai et al., 2015; Xiao et al., 2017; Diamantopoulou and Voudouris, 2008; Arunprakash et al., 2014; Rao et al., 2016; He et al., 2017; Jia et al., 2018). Many studies only analysed the hydrochemistry or individual years from individual components (Martín et al., 2014; Huang et al., 2013; Yang et al., 2020). Because of the close connection of various water quality indexes in groundwater, drawing scientific conclusions from the interaction and evolution of each component in groundwater is difficult. Moreover, identifying the controlling factors causing the evolution of groundwater components based on regional hydrogeological conditions and long-term water quality monitoring data and taking corresponding measures to better protect the underground environment are necessary to further reflect the evolution direction of the groundwater environment.

From 2003 to 2019, the areas of urban land in the east and south of Beijing expanded the most, especially those in the Tongzhou District and Daxing District (Huang et al., 2015; Wu et al., 2019). As the south gate of Beijing, Daxing District is a key area to undertake the population and estrangement functions of the central urban area (Shi et al., 2014). With the construction of the Beijing Daxing International Airport, Daxing District became one of the fastest-growing areas in Beijing. In order to understand the effects of natural and human factors on the chemical composition of groundwater in the process of rapid urbanization. In this paper, the Daxing District, a typical region in the plain area of Beijing, was selected as the research area. By collecting groundwater quality monitoring data from 2006 to 2016, the evolution direction of regional groundwater chemical types was analysed and the sources of main ions were traced using hydrochemical methods such as the Piper ternary diagram, Gibbs diagram and ion combination ratio analysis. The main factors affecting the evolution of the water chemical composition were identified using the principal component–partial relation method. Moreover, the influence degree of each main factor on the groundwater composition and the change rules in time and space were analysed. This study can provide a theoretical and methodological reference for regional groundwater management and industrial layout.

## 2 STUDY AREA

The Daxing District is located in the southern suburbs of Beijing (Figure 1), with an area of 1,036 km<sup>2</sup>. Its geographical coordinates are 116°13'–116°43'E and 39°26'–39°51'N. There are 14 towns in the jurisdiction: the northern region includes Huangcun, Xihongmen, Jiugong, Yizhuang, and Yinghai; the central region includes Beizangcun, Panggezhuang, Weishanzhuang, and Qingyundian; and the southern region includes Yufa, Lixian, Anding, Changziying, and Caiyu. The region experiences a midlatitude warm temperate continental monsoon climate, with four distinct seasons and an average annual temperature of 11.7°C. The inter-annual variation of rainfall is large, and the annual rainfall is mainly concentrated in June to September. The area belongs to the alluvial plain of the

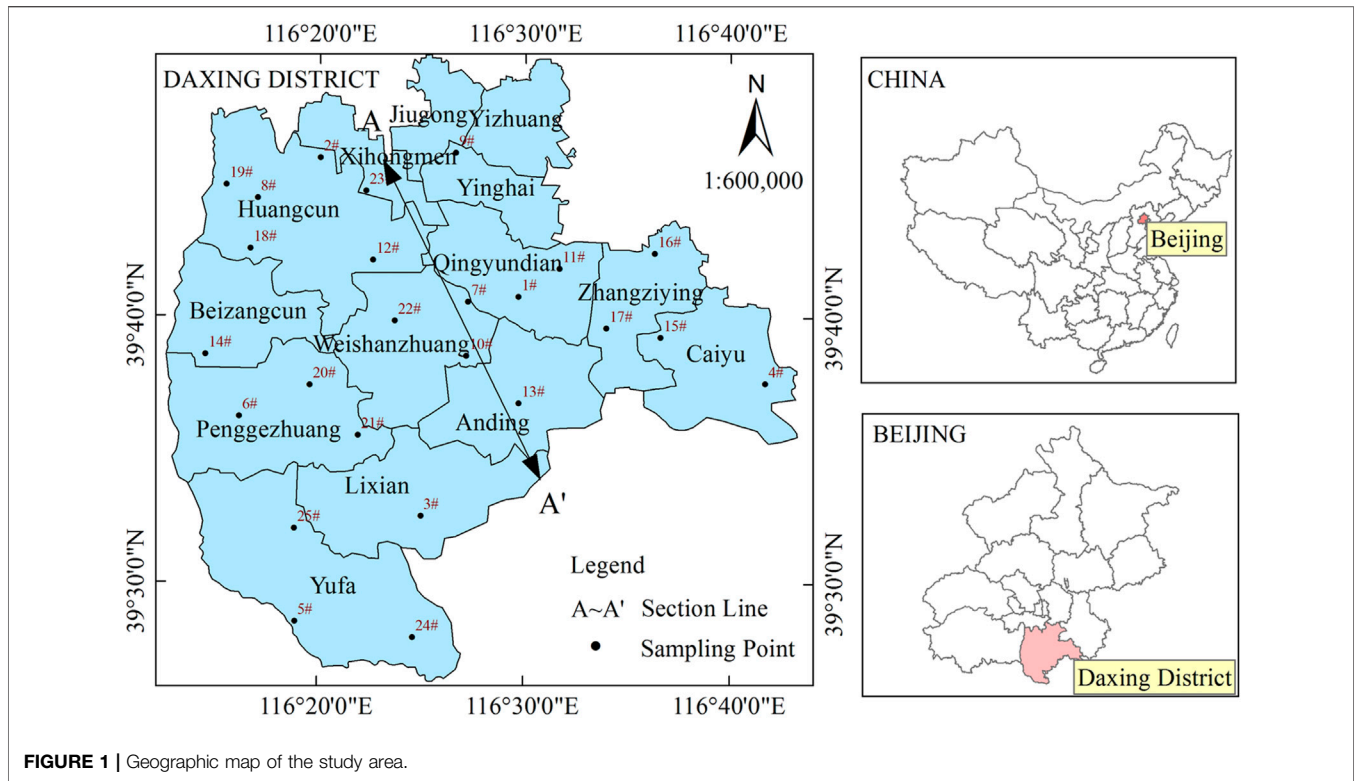


FIGURE 1 | Geographic map of the study area.

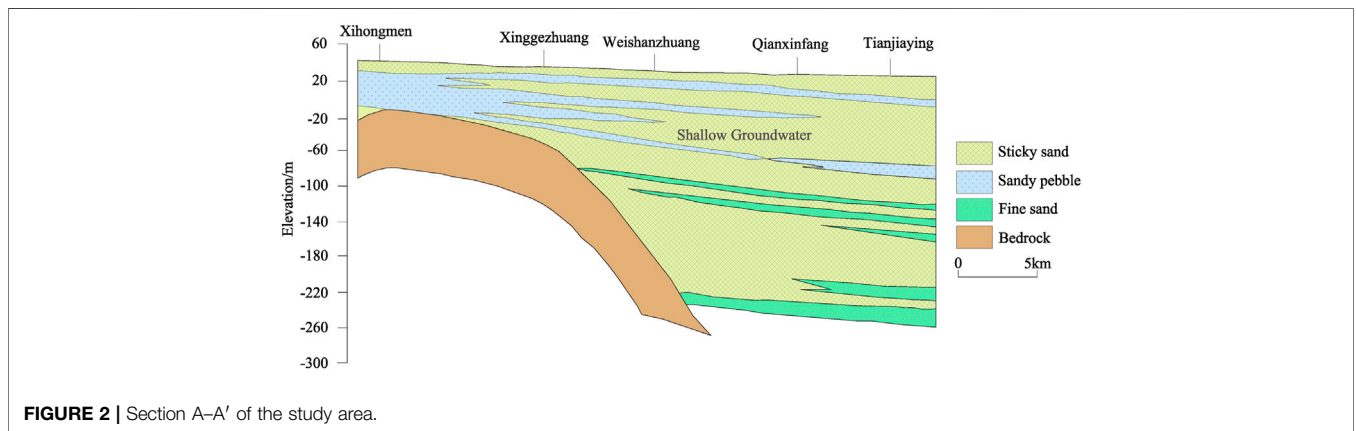


FIGURE 2 | Section A-A' of the study area.

Yongding River, with relatively flat terrain and a slope of 0.5–2.0%. The study area is located in the northeast tectonic belt on the southeast side of the intersection of the Yanshan and Taihang Mountains, called the “Daxing uplift”. The thickness of the Quaternary unconsolidated layer is 200–300 m, and the aquifer is dominated by coarse medium sand and fine sand. Also, some areas have a certain thickness of sand and gravel layer distributions (Figure 2). Controlled by the uplift of the basement, the thickness of the Quaternary sediments varies greatly, from 50 m in the northwest to 300 m in the southeast. The thickness of the Quaternary deposits is 50–60 m in the northwest of study area, and 70–90 m in Huangcun, Jiugong, and Yizhuang in the east of study area. The thickness increases gradually in the south

and reaches 300 m in Yufa and Kangying. Quaternary aquifer particles range from coarse to fine and water-rich from large to small.

### 3 MATERIALS AND METHODS

#### 3.1 Sample Collection and Testing

A total of 275 shallow groundwater samples were collected from the study area from 2006 to 2016 from 25 sampling points, as shown in Figure 1. During the sampling process, water was pumped until the parameters were stable and then sampled. The sampling bottles were 2.5-L PVC plastic bottles. The water

samples were used for wetting and washing three times, and the original samples were sealed and preserved after sampling. Then, the samples were sent to the Daxing District Water Environment Monitoring Center in Beijing for testing within 48 h. The total dissolved solids (TDS) were determined using the gravimetric method (drying at 105°C). The total hardness (TH) was determined using ethylenediaminetetraacetic acid. Na<sup>+</sup>, K<sup>+</sup>, Ca<sup>2+</sup> and Mg<sup>2+</sup> were determined using inductively coupled plasma emission spectrometry. SO<sub>4</sub><sup>2-</sup>, Cl<sup>-</sup>, and NO<sub>3</sub><sup>-</sup> were determined using ion chromatography. The CO<sub>3</sub><sup>2-</sup> and HCO<sub>3</sub><sup>-</sup> contents were determined by titration. The ion balance errors of all samples were less than 5%.

### 3.2 Research Methods

This study was conducted to understand the temporal-spatial evolution of groundwater chemical characteristics and influencing factors in the study area and to provide theoretical support for regional groundwater protection and industrial layout. On the basis of the hydrochemical method, the changes in the main anion and cation components and hydrochemical types in groundwater from 2006 to 2010 and 2011 to 2016 and the main factors causing these changes were analysed. The natural sources of the main ions in the groundwater in the study area were determined using the Gibbs diagram and the ion combination ratio method. The main controlling factors affecting the evolution of groundwater components were extracted using the principal component-partial relation method to further analyse the temporal and spatial evolution trend of the dominant factors affecting the evolution of the groundwater chemistry in the study area. The dynamic evolution law of the main controlling factors was analysed, which provided the basis for the protection measures of regional groundwater resources and the industrial layout scheme.

#### 3.2.1 Hydrochemical Analysis Methods

The chemical characteristics and formation of shallow groundwater were qualitatively and quantitatively analysed using the Piper three-line diagram, Gibbs diagram and ion combination ratio method. The mineral saturation index was calculated using the Phreeqc software.

#### 3.2.2 Principal Component Analysis-Partial Connection Number Method

Principal component analysis is a statistical method that converts multiple variables into a few principal components using dimension reduction technology to reduce the number of predicted variables (Lianne et al., 2017; Maria et al., 2012). Partial connection number is the adjoint function of set pair analysis, which is mainly used to reflect the dynamic evolution between connection numbers (Yang and Zhao, 2019; Yang et al., 2021). At present, the first-order partial connection number is commonly used (Xiang et al., 2021). Since the first-order connection number only reflects the evolution between adjacent connection numbers, the mutual transformation between identity and opposition is not considered (Li

et al., 2022). The second-order connection number not only considers the mutual transformation between identity, difference and opposition but also reflects the mutual transformation between identity and opposition. Therefore, this paper adopted principal component analysis-partial connection number to extract the main control factors affecting the evolution of the shallow groundwater components and analysed the dynamic evolution rules of the main control factors.

The specific model construction steps are as follows:

##### 3.2.2.1 Eliminating the $X_{ij}$ Dimension of the Raw Data

$$X_{ij}^* = \frac{X_{ij} - \bar{X}_j}{S_j} \quad (i = 1, 2, \dots, n; j = 1, 2, \dots, p) \quad (1)$$

In the formula,  $X_{ij}^*$  is the original data  $X_{ij}$  after a standardised processing value.  $\bar{X}_j$  and  $S_j$  are the mean and standard deviation of the  $j$ th index.  $i$  is the evaluation sample, and  $j$  is the  $j$  index in the  $i$  sample.  $n$  is the number of samples, and  $p$  is the number of indicators.

##### 3.2.2.2 Calculating the Correlation Coefficient Matrix and Corresponding Eigenvector

$$R = [r_{ij}]_{p \times p} \quad (2)$$

$$r_{ij} = \frac{\text{cov}(X_i^*, X_j^*)}{\sqrt{D(X_i^*)} \cdot \sqrt{D(X_j^*)}} \quad (3)$$

$$|\lambda I - R| = 0 \quad (4)$$

In the above formulas,  $R$  is the correlation coefficient matrix of the processed data,  $r_{ij}$  is the correlation coefficient, and  $\text{cov}(X_i^*, X_j^*)$  is the covariance of the index data  $X_i^*$  and  $X_j^*$  after processing.  $D(X_i^*)$  and  $D(X_j^*)$  are the variances of the processed index data  $X_i^*$  and  $X_j^*$ , respectively. By solving **Formula (4)**, the eigenvectors  $\lambda_1, \lambda_2, \dots, \lambda_p$  and the corresponding eigenvectors  $\gamma_1, \gamma_2, \dots, \gamma_p$  are obtained.

##### 3.2.2.3 Determining Principal Components by Calculating the Principal Component Contribution Rate and the Cumulative Contribution Rate

$$P_t = \frac{\lambda_t}{\sum_{j=1}^p \lambda_j} \quad (t = 1, 2, \dots, p) \quad (5)$$

$$P = \sum_{t=1}^m P_t \quad (6)$$

In these formulas,  $P_t$  is the contribution rate of the main components,  $P$  is the cumulative contribution rate, and  $\lambda_t$  is the eigenvalue corresponding to the first  $m$  principal components when the cumulative contribution rate reaches the requirement.

**3.2.2.4 Calculating the Score of the Groundwater Chemical Composition Evolution Factors**

$$x = \sum_{t=1}^m \sum_{j=1}^p \gamma_{tj} X_{tj}^* P_t \tag{7}$$

In the formula,  $x$  is the score of the factors affecting the evolution of the groundwater chemical composition and  $\gamma_{tj}$  is the principal component load, which is the  $t$ th component of the eigenvector  $r_j$ .

**3.2.2.5 Determining the Evaluation Level of Each Main Control Factor Score**

The principal component scores of the two main factors affecting the groundwater chemical composition were evaluated. According to the data characteristics of the score values and referring to existing data, the evaluation grade standards of the score values of the impact factors were determined as follows:

$$\{s_{gj} | g = 1, 2, 3; j = 1, 2\} \tag{8}$$

In the formula,  $s_{gi}$  represents the critical value of the evaluation level of each impact factor.  $g$  is the evaluation level (Levels 1–3) and  $j$  represents the  $j$ th influencing factor in sample  $i$ , representing natural and human factors, respectively.

$$\{x_{ij} | i = 1, 2, 3, \dots, n; j = 1, 2\} \tag{9}$$

In the formula,  $x_{ij}$  represents the score value of the  $j$ th index of the  $i$ th sample.  $i$  is the evaluation sample and  $n$  is the number of samples.

**3.2.2.6 Calculating the Connection Number of Each Main Control Factor Score About Each Sample**

The relationship between the principal component score of the  $j$ th index about the evaluation sample  $i$  and the index value of the evaluation standard was calculated.

If the indicator is a positive indicator  $x_{ij} \leq s_{1j}$  or a reverse indicator  $x_{ij} \geq s_{1j}$ ,

$$u_{ij1} = \begin{cases} 1 \\ 1 - 2 \frac{x_{ij} - s_{1j}}{s_{2j} - s_{1j}} \\ -1 \end{cases} \tag{10}$$

If the indicator is a positive indicator  $s_{1j} < x_{ij} \leq s_{2j}$  or a reverse indicator  $s_{1j} > x_{ij} \geq s_{2j}$ ,

$$u_{ij2} = \begin{cases} 1 - 2 \frac{s_{1j} - x_{ij}}{s_{1j} - s_{0j}} \\ 1 \\ 1 - 2 \frac{x_{ij} - s_{2j}}{s_{3j} - s_{2j}} \end{cases} \tag{11}$$

If the indicator is a positive indicator  $s_{2j} < x_{ikj} \leq s_{3j}$  or a reverse indicator  $s_{2j} > x_{ikj} \geq s_{3j}$ ,

$$u_{ij3} = \begin{cases} -1 \\ 1 - 2 \frac{s_{2j} - x_{ij}}{s_{2j} - s_{1j}} \\ 1 \end{cases} \tag{12}$$

In the above formulas,  $u_{ij1}$ ,  $u_{ij2}$ ,  $u_{ij3}$  represent the number of index connections within each level for the  $j$ th index of sample  $i$ , respectively.  $s_{1j}$  and  $s_{2j}$  are the limit values of Grades 1 and 2 and Grades 2 and 3, respectively.  $s_{0j}$  is the minimum value of Grade 1, and  $s_{3j}$  is the maximum value of Grade 3.

The identity, difference and opposition of the  $j$ th principal component score index about sample  $i$  are transformed into relative membership. The calculation process is as follows:

$$v''_{ijg} = 0.5 + 0.5u_{ijg} \tag{13}$$

$$v'_{ijg} = v''_{ijg} / \sum_{g=1}^3 v''_{ijg} \tag{14}$$

$$u_{ij} = v'_{ij1} + v'_{ij2}I + v'_{ij3}J \tag{15}$$

In the above formulas,  $u_{ij}$  represents the degree of identity ( $v'_{ij1}$ ), degree of difference ( $v'_{ij2}$ ) and degree of opposition ( $v'_{ij3}$ ) of the  $j$ th indicator about sample  $i$ .  $v'_{ijg}$  is the calculation process.

**3.2.2.7 Calculating the Partial Linkages of Influencing Factors**

The support degree of each grade was calculated using the second-order partial connection number. The calculation formulas are shown in (16)–(18):

$$S_1 = a \cdot (1 + \partial^+ a + \partial^{2+} a) \tag{16}$$

$$S_2 = b \cdot (\partial^- b + 1 + \partial^+ b) \tag{17}$$

$$S_3 = c \cdot (\partial^{2-} c + \partial^- c + 1) \tag{18}$$

In the above formulas,  $S_1$ ,  $S_2$ , and  $S_3$  represent Level 1, Level 2, and Level 3 support, respectively.

By comparing the sizes of  $S_1$ ,  $S_2$ , and  $S_3$ , the depth evolution trend of the evaluation level of the research object can be determined.  $\partial^+ a$  and  $\partial^{2+} a$  represent the positive evolution rates from  $b$  and  $c$  to  $a$ , respectively.  $\partial^- b$  represents the negative evolution rate of  $a$  to  $b$ , and  $\partial^+ b$  represents the positive evolution rate of  $c$  to  $b$ .  $\partial^{2-} c$  and  $\partial^- c$  represent the negative evolution rates from  $a$  and  $b$  to  $c$ , respectively.

The calculation formulas of the evolution rate of the connection number are shown in Formulas (19)–(24):

$$\partial^+ a = \frac{a}{a + b} \tag{19}$$

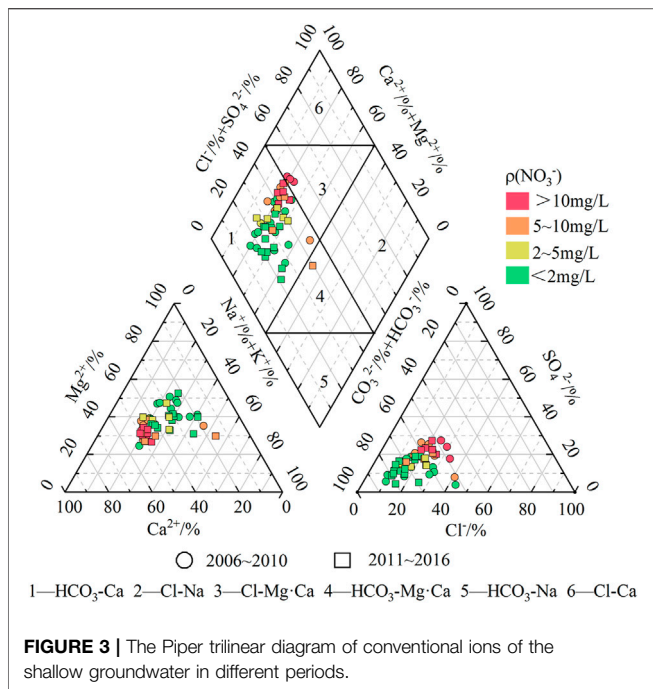
$$\partial^{2+} a = \frac{a}{a + b} / \left( \frac{a}{a + b} + \frac{b}{b + c} \right) \tag{20}$$

$$\partial^- b = \frac{b}{a + b} \tag{21}$$

$$\partial^+ b = \frac{b}{b + c} \tag{22}$$

$$\partial^{2-} c = \frac{c}{b + c} / \left( \frac{b}{a + b} + \frac{c}{b + c} \right) \tag{23}$$





**FIGURE 3 |** The Piper trilinear diagram of conventional ions of the shallow groundwater in different periods.

$$\bar{d}^-c = \frac{c}{b+c} \quad (24)$$

## 4 RESULTS AND DISCUSSION

### 4.1 Evolution of Groundwater Chemical Characteristics

#### 4.1.1 Content and Evolution of Chemical Components in Water

The collected shallow groundwater samples were divided into two groups according to time: 2006–2010 and 2011–2016. The hydrochemical data of each group were averaged for each period. Overall, the total concentration of cations and anions in the shallow groundwater increased over time. From 2006 to 2010 and 2011 to 2016, the order of the average cation concentration was  $\text{Ca}^{2+} > \text{Na}^+ > \text{Mg}^{2+} > \text{K}^+$ , and  $\text{Ca}^{2+}$ ,  $\text{Na}^+$  and  $\text{Mg}^{2+}$  increased by 7.94, 20.88, and 2.69%, respectively. In contrast,  $\text{K}^+$  decreased by 2.56%. For the anions, the order of the anion average concentration in the two periods changed from  $\text{HCO}_3^- > \text{Cl}^- > \text{SO}_4^{2-}$  to  $\text{HCO}_3^- > \text{SO}_4^{2-} > \text{Cl}^-$ , which was mainly due to the decrease in the  $\text{Cl}^-$  concentration from 95.75 to 88.77 mg/L and the increase in that of  $\text{SO}_4^{2-}$  from 88.52 to 99.62 mg/L. Because of the increase in cation and anion concentrations, the TDS and TH increased. TDS increased from 760.96 to 786.36 mg/L, and TH increased from 489.73 to 509.84 mg/L. The  $\text{NO}_3^-$  concentration increased from 4.86 to 4.89 mg/L. Although the average concentration of  $\text{NO}_3^-$  in 2011–2016 only increased by 0.03 mg/L compared with that in 2006–2010, the number of water quality monitoring cases with  $\text{NO}_3^-$  concentrations over 10 mg/L increased from three to five, indicating that the range of the  $\text{NO}_3^-$  pollution of the

groundwater tended to increase. The pH decreased slightly from 7.5 to 7.4, indicating an alkaline-to-acidic trend.

The variation coefficients of the main components from 2011 to 2016 were lower than those from 2006 to 2010, and the largest reduction was that of  $\text{Cl}^-$ . From the average coefficient of the variation of each ion, the maximum coefficient of variation was that of  $\text{NO}_3^-$ , with an average variation coefficient of 1.48, indicating that  $\text{NO}_3^-$  is mainly affected by human factors (Zhai et al., 2015; Xiao et al., 2017; Yin et al., 2019). It is followed by  $\text{Cl}^-$  and  $\text{SO}_4^{2-}$ , with an average variation coefficient of about 0.5. The smaller coefficients of variation were those of  $\text{HCO}_3^-$ , TDS, TH, and  $\text{Mg}^{2+}$ , and the average coefficient of variation was about 0.2. The smallest coefficient of variation was that of  $\text{HCO}_3^-$ , with an average of 0.19, indicating that  $\text{HCO}_3^-$  was mainly affected by natural factors (Zhai et al., 2015; Yin et al., 2019).

#### 4.1.2 Evolution of Groundwater Chemical Types

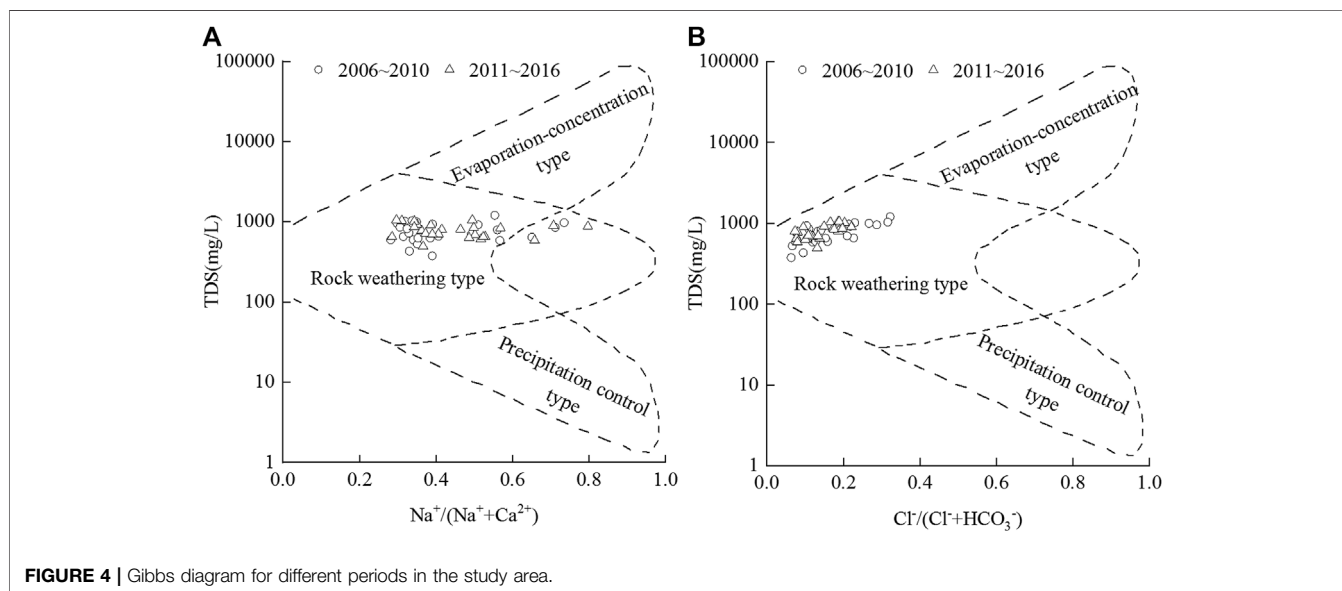
The groundwater chemical types are affected by natural conditions and human activities. The Piper three-line diagram was used to analyse the hydrochemical types in 2006–2010 and 2011–2016 to analyse the evolution characteristics of the hydrochemical types in the two periods.

As shown in **Figure 3**, 2006–2010 is similar to 2011–2016. Most water quality monitoring sites were located in Area 1, indicating that alkaline earth metals ( $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$ ) exceeded alkaline metals ( $\text{K}^+$  and  $\text{Na}^+$ ) and weak acids ( $\text{HCO}_3^-$ ) exceeded strong acids ( $\text{Cl}^-$  and  $\text{SO}_4^{2-}$ ) in most water quality monitoring sites in the study area. Therefore, the water of this region is mainly  $\text{HCO}_3\text{-Ca}$ -type water. In 2011–2016, compared with 2006–2010, part of the  $\text{Cl-Mg-Ca}$ -type water changed into  $\text{HCO}_3\text{-Ca}$ -type water, whereas a small amount of  $\text{HCO}_3\text{-Ca}$ -type water changed into  $\text{HCO}_3\text{-Na-Mg}$ -type water. The change of  $\text{Cl-Mg-Ca}$ -type water into  $\text{HCO}_3\text{-Ca}$ -type water may have been caused by the migration of  $\text{Cl}^-$  to the deep groundwater and thus the decrease of the concentration of  $\text{Cl}^-$  in the shallow groundwater. The conversion of  $\text{HCO}_3\text{-Ca}$  to  $\text{HCO}_3\text{-Na-Mg}$  may be related to the dissolution of silicate on the one hand. According to the monitoring data of the groundwater level in the study area, the average groundwater level decreased from 11.89 to 17.60 m from 2000 to 2016. The recharge path of groundwater was lengthened, which led to the increase in  $\text{Na}^+$  dissolution in silicate during the recharge process. This resulted in the transformation of  $\text{HCO}_3\text{-Ca}$  into  $\text{HCO}_3\text{-Na-Mg}$ . On the other hand, it may be related to human inputs of  $\text{Na}^+$ .

In addition, the Piper trilinear diagram (**Figure 3**) shows that with the increase in  $\text{NO}_3^-$  concentration, the main trend of the water chemistry type evolved from  $\text{HCO}_3\text{-Ca}$ -type water to  $\text{Cl-Mg-Ca}$ -type water. At the same time, with the increase in  $\text{NO}_3^-$  concentration, the main cation and anion components also change accordingly. For the cations, with the increase in  $\text{NO}_3^-$  concentration, the ratio of  $\text{Ca}^{2+}$  and  $\text{Na}^+$  increased, whereas that of  $\text{Mg}^{2+}$  decreased. For the anions, with the increase in  $\text{NO}_3^-$  concentration, the proportions of  $\text{Cl}^-$  and  $\text{SO}_4^{2-}$  increased, whereas the proportion of  $\text{HCO}_3^-$  decreased. This indicates that the change in the  $\text{NO}_3^-$  concentration profoundly affected the component evolution of the groundwater. **Table 1** shows that

**TABLE 1** | Calculation results of the chemical components and coefficients of variation of the shallow groundwater in the study area (unit: mg/L).

| Time      | Name                      | K <sup>+</sup> | Na <sup>+</sup> | Ca <sup>2+</sup> | Mg <sup>2+</sup> | Cl <sup>-</sup> | SO <sub>4</sub> <sup>2-</sup> | HCO <sub>3</sub> <sup>-</sup> | TDS    | TH     | NO <sub>3</sub> <sup>-</sup> | PH   |
|-----------|---------------------------|----------------|-----------------|------------------|------------------|-----------------|-------------------------------|-------------------------------|--------|--------|------------------------------|------|
| 2006–2010 | Content of each component | 2.03           | 71.58           | 95.37            | 59.61            | 95.75           | 88.52                         | 448.29                        | 760.96 | 489.73 | 4.86                         | 7.5  |
|           | Coefficient of variation  | 0.4            | 0.45            | 0.35             | 0.3              | 0.67            | 0.53                          | 0.21                          | 0.27   | 0.24   | 1.6                          | 0.02 |
| 2011–2016 | Content of each component | 1.97           | 86.53           | 102.94           | 61.21            | 88.77           | 99.62                         | 552.07                        | 786.36 | 509.84 | 4.89                         | 7.4  |
|           | Coefficient of variation  | 0.35           | 0.38            | 0.32             | 0.23             | 0.38            | 0.45                          | 0.16                          | 0.19   | 0.19   | 1.36                         | 0.02 |
| 2006–2016 | Content of each component | 2.00           | 79.06           | 99.16            | 60.41            | 92.26           | 94.07                         | 500.18                        | 773.66 | 499.79 | 4.88                         | 7.45 |
|           | Coefficient of variation  | 0.38           | 0.42            | 0.34             | 0.27             | 0.53            | 0.49                          | 0.19                          | 0.23   | 0.22   | 1.48                         | 0.02 |



**FIGURE 4** | Gibbs diagram for different periods in the study area.

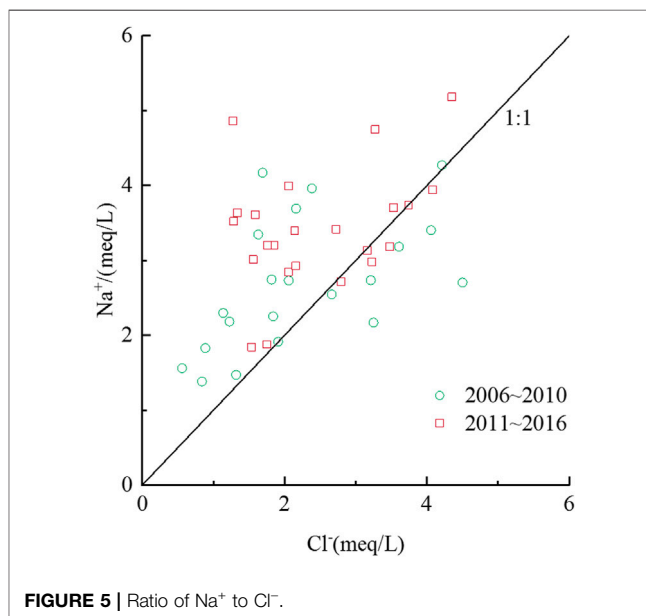
the variation coefficient of NO<sub>3</sub><sup>-</sup> is very large compared with those of the other main ions. Thus, it can be considered that NO<sub>3</sub><sup>-</sup> is mainly caused by human factors; that is, human factors have a great impact on groundwater components.

## 4.2 Source Evolution of Groundwater Chemical Components

### 4.2.1 Evolution of Water–Rock Interaction

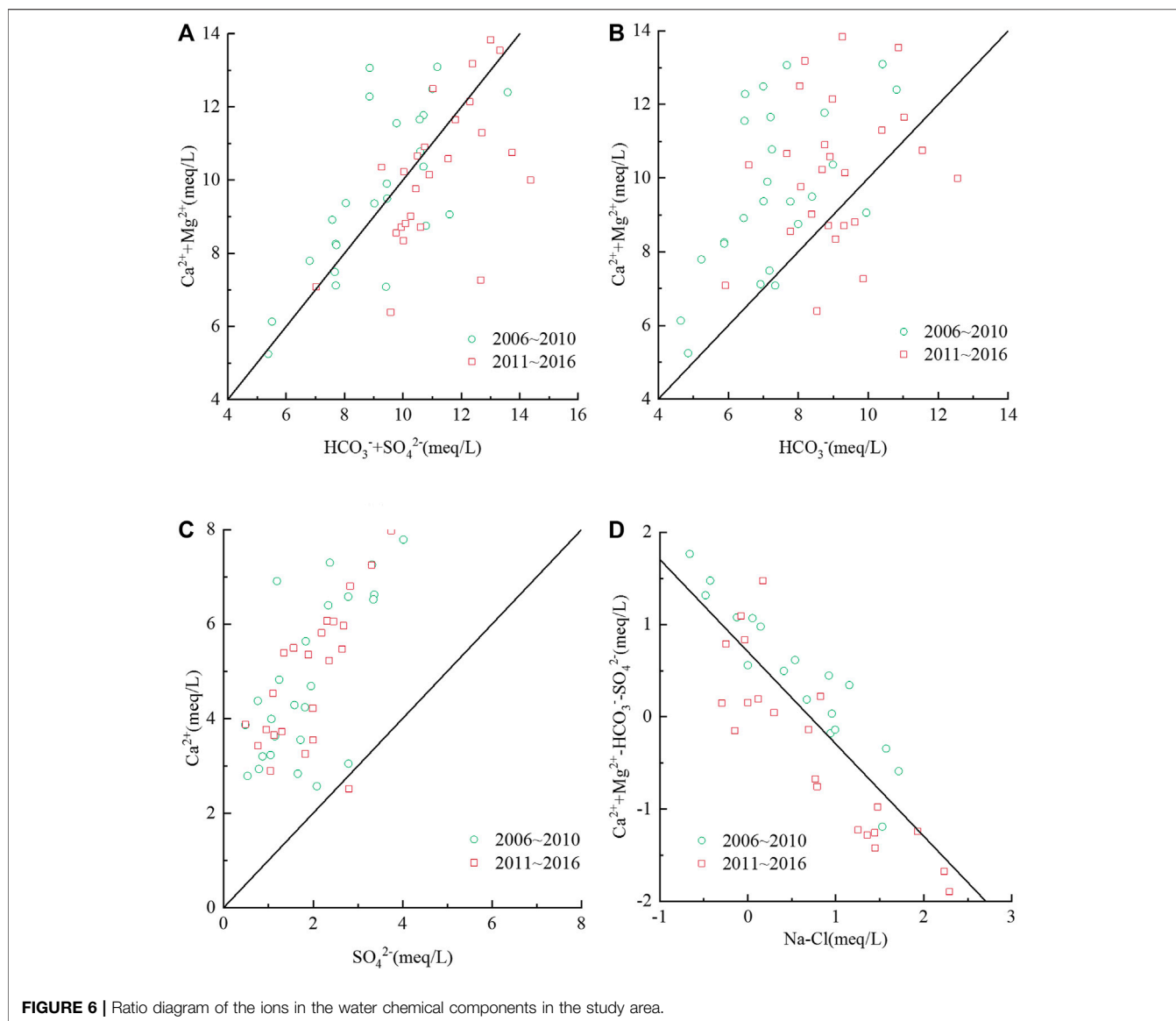
The Gibbs graphical method can be used to identify the sources of chemical components in natural water. The natural origin control mode of water chemical compositions can be divided into three types: evaporation–concentration type, rock weathering type and precipitation control type. The relationships between TDS and Na<sup>+</sup>/(Na<sup>+</sup> + Ca<sup>2+</sup>), Cl<sup>-</sup>/(Cl<sup>-</sup> + HCO<sub>3</sub><sup>-</sup>) in 2006–2010 and 2011–2016 were drawn using the Gibbs diagram method. **Figure 4** shows that the natural origin control mode of the shallow groundwater chemical composition in the study area is mainly rock weathering type.

From 2006 to 2010, the average value of Na<sup>+</sup>/(Na<sup>+</sup> + Ca<sup>2+</sup>) was 0.42, and the TDS was 760.96 mg/L. From 2011 to 2016, the average value of Na<sup>+</sup>/(Na<sup>+</sup> + Ca<sup>2+</sup>) was 0.45, and the TDS was 786.36 mg/L. Both Na<sup>+</sup>/(Na<sup>+</sup> + Ca<sup>2+</sup>) and the TDS increased. However, the TDS values of the two periods showed a stable trend with increasing Na<sup>+</sup>/(Na<sup>+</sup> + Ca<sup>2+</sup>), indicating that the cation components were mainly



**FIGURE 5** | Ratio of Na<sup>+</sup> to Cl<sup>-</sup>.

controlled by rock weathering and that the other natural types had little effect on the cation components. At the same time, the Na<sup>+</sup>/(Na<sup>+</sup> + Ca<sup>2+</sup>) values greater than 0.6 of some water quality



monitoring points indicate that the cation components in these regions may have been affected by human factors.

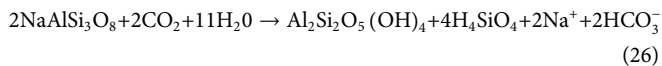
The average value of  $\text{Cl}^-/(\text{Cl}^- + \text{HCO}_3^-)$  was 0.14, slightly lower than 0.18 in 2006–2010. The  $\text{Cl}^-/(\text{Cl}^- + \text{HCO}_3^-)$  ratio was less than 0.4 in both periods, indicating that water–rock interactions play a major role in the determination of the anion components. In addition, **Figure 4B** shows that the TDS value increased with increasing  $\text{Cl}^-/(\text{Cl}^- + \text{HCO}_3^-)$ , indicating that, in addition to water–rock interactions, natural and human factors also affected the anion components and TDS.

#### 4.2.2 Analysis Based on the ion Combination Ratio

The ion ratio relationship was used to reveal the source of the main ions to further analyse the chemical evolution of the groundwater. The  $\text{Na}^+/\text{Cl}^-$  ratio was used to reflect the sources of  $\text{Na}^+$  and  $\text{Cl}^-$ . **Figure 5** shows that the water quality monitoring

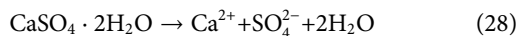
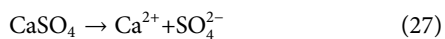
points are scattered on both sides of the 1:1 line, and the points above the 1:1 line are significantly more than those below. Compared with that in 2006–2010, the average ratio of  $\text{Na}^+/\text{Cl}^-$  increased from 1.44 to 1.67 in 2011–2016, resulting in more points above the 1:1 line in 2011–2016 than those in 2006–2010. On the one hand, this indicates that the source of  $\text{Na}^+$  is not only rock salt dissolution but also silicate dissolution. On the other hand, the source of  $\text{Na}^+$  may also be related to ion exchange reactions or artificial inputs of  $\text{Na}^+$ . According to the simulated value SI of the mineral saturation index (**Supplementary Table S1**), the values of rock salt mineral saturation in 2006–2010 and 2011–2016 were –6.70 and –6.73, respectively, which are less than 0. This indicates that the rock salt minerals were unsaturated and that part of  $\text{Na}^+$  came from rock salt. The dissolution formulas of rock salt and silicate minerals are shown in **Eqs 25, 26**, respectively:





In addition, a small number of water quality monitoring points are located below 1:1, deviating from the 1:1 line, indicating that there may be an increase in the  $\text{Cl}^-$  concentration caused by anthropogenic inputs of chloride.

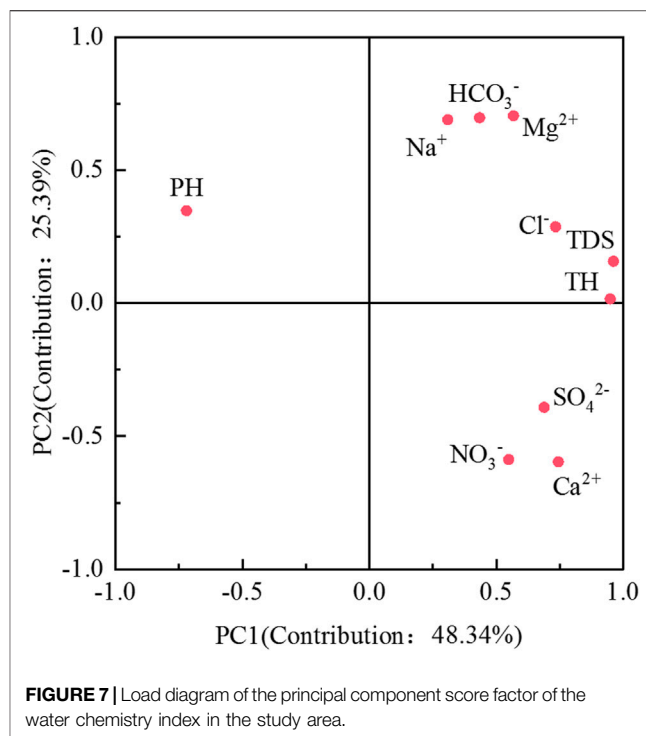
In the analysis of the sources of  $\text{Ca}^{2+}$ , the relationship between  $\text{Ca}^{2+} + \text{Mg}^{2+}$  and  $\text{HCO}_3^- + \text{SO}_4^{2-}$  (Figure 6A) shows that the water quality monitoring points are located on both sides of the 1:1 line, indicating that the sources of  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  are mainly carbonate and sulphate. At the same time, in further tracing the sources of  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$ , it can be known from the ratio of  $\text{Ca}^{2+} + \text{Mg}^{2+}$  to  $\text{HCO}_3^-$  (Figure 6B) that most of the water quality monitoring points are above the 1:1 line. According to the simulation results in Supplementary Table S1, the mineral saturation values of calcite and dolomite in 2006–2010 and 2011–2016 were 0.14, 0.25 and 0.43, 0.62, respectively. These values are greater than 0, indicating that the two carbonate minerals are in a saturated state. Therefore, the sources of  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  are not carbonates. In further tracing the source of  $\text{Ca}^{2+}$ , the relationship between  $\text{Ca}^{2+}$  and  $\text{SO}_4^{2-}$  was plotted. Figure 6C shows that the ratio of  $\text{Ca}^{2+}$  to  $\text{SO}_4^{2-}$  is basically greater than one. At the same time, according to the simulation results in Supplementary Table S1, the mineral saturation values of anhydrite and gypsum were  $-1.92$  and  $-1.70$  from 2006 to 2010 and  $-1.86$  and  $-1.64$  from 2011 to 2016, respectively. These values are all less than 0, indicating that sulphate dissolution was an important source of  $\text{Ca}^{2+}$ . However, the  $\text{Ca}^{2+}/\text{SO}_4^{2-}$  ratio is higher than the 1:1 line, indicating that, in addition to sulphate,  $\text{Ca}^{2+}$  was derived from other exogenous inputs. The dissolution equations of anhydrite and gypsum are shown in Eqs 27, 28, respectively.



In exploring whether ion exchange reactions occurred between  $\text{Na}^+$  and  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ , the ratio of  $\text{Ca}^{2+} + \text{Mg}^{2+} - \text{HCO}_3^- - \text{SO}_4^{2-}$  to  $\text{Na}^+ - \text{Cl}^-$  (Figure 6D) is around  $-1$ , indicating that ion exchange reactions occurred. Meanwhile, the relationship between  $\text{Ca}^{2+} + \text{Mg}^{2+}$  and  $\text{HCO}_3^- + \text{SO}_4^{2-}$  shows that the average ratios of  $\text{Ca}^{2+} + \text{Mg}^{2+}$  and  $\text{HCO}_3^- + \text{SO}_4^{2-}$  in 2006–2010 and 2011–2016 are 1.05 and 0.92, respectively. The water quality monitoring points have a downward trend to 1:1, indicating that the exchange reaction of  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  in 2011–2016 has an increasing trend compared with that in 2006–2010.

### 4.3 The Spatiotemporal Evolution Trend of the Influence Degrees

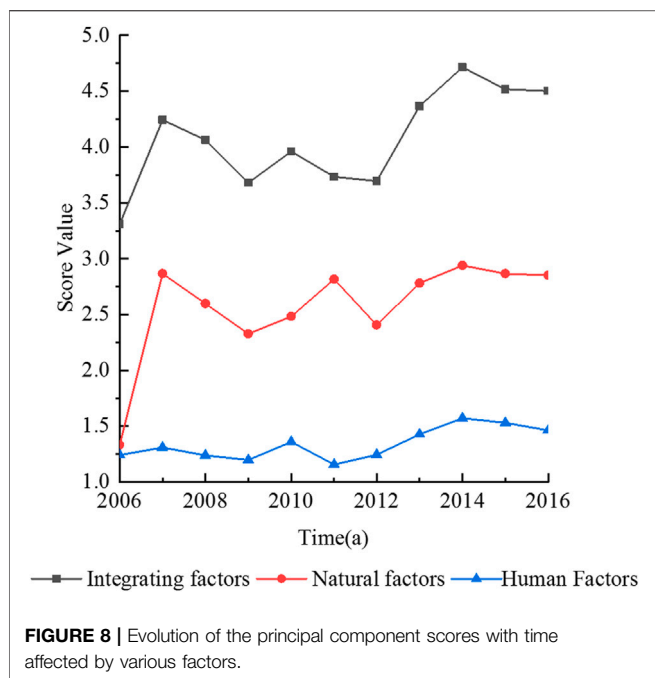
A principal component analysis of the shallow groundwater in the study area was conducted to further analyse the dominant



factors of the water chemistry evolution in the study area. On the basis of Formulas (1)–(3), the correlation coefficient matrix of the shallow groundwater hydrochemical data was obtained using the SPSS software. As shown in Supplementary Table S2,  $\text{Ca}^{2+}$  and  $\text{SO}_4^{2-}$  had high positive correlations with  $\text{NO}_3^-$ ; the correlation coefficients were more than 0.6, indicating that the concentrations of  $\text{NO}_3^-$  and  $\text{Ca}^{2+}$ ,  $\text{SO}_4^{2-}$  increased or decreased. The higher correlation with  $\text{HCO}_3^-$ ,  $\text{Na}^+$ , and  $\text{Mg}^{2+}$  featured correlation coefficients greater than 0.6, indicating that the  $\text{NO}_3^-$  and  $\text{Ca}^{2+}$ ,  $\text{SO}_4^{2-}$ ,  $\text{HCO}_3^-$  and  $\text{Na}^+$ ,  $\text{Mg}^{2+}$  concentrations increased or decreased.

The principal component analysis of the shallow groundwater in the study area was performed using Formulas (4)–(7). Two principal components (PC1 and PC2) with eigenvalues greater than one were extracted, and their contribution rates were 48.34 and 25.39%, respectively, and the cumulative contribution rate was 73.73%. The load of the principal component score factor is shown in Figure 7. The Piper plot (Figure 1) shows that the anions in the study area are mainly  $\text{HCO}_3^-$ , accounting for more than 70% of the anion concentration equivalent, and the minimum spatial variation coefficient is 0.19. Therefore,  $\text{HCO}_3^-$  can be used as the dominant factor for identifying natural factors. The spatial variation coefficient of  $\text{NO}_3^-$  was the largest relative to those of other indicators, reaching 1.48. With increasing  $\text{NO}_3^-$ , the groundwater hydrochemical type and anion and cation components also changed accordingly. So,  $\text{NO}_3^-$  can be used as the dominant factor in identifying human factors.

According to the principal component score factor load diagram (Figure 7), the maximum positive load scores of PC1 are TDS, TH and  $\text{Ca}^{2+}$ . Given the two indicators  $\text{HCO}_3^-$  and  $\text{NO}_3^-$ , the positive load score of PC1 can be used as a



comprehensive index of natural and human factors.  $\text{HCO}_3^-$ ,  $\text{Mg}^{2+}$ , and  $\text{Na}^+$  with the highest score of the PC2 positive load chart can be used as the evaluation index for the influence degrees of natural factors. The maximum scores of the PC2 negative load diagram are  $\text{NO}_3^-$  and  $\text{Ca}^{2+}$ , which can be used as evaluation indices for the influence degree of human factors.

According to the identification results of the influencing factors of the hydrochemical evolution, **Formula (7)** was used to calculate the scores of the human, natural and natural-human factors. The PC2 load was processed positively and the evolution process of the principal component scores with time under the guidance of such three factors was drawn (**Figure 8**) to facilitate the comparison of data. The score of the comprehensive effect was always greater than that of the natural and human factors, indicating that the evolution process of the water's chemical components in the study area was always affected by human and natural factors. The scores of the natural and human influences were 2.56 and 1.34, respectively, indicating that the evolution of the groundwater components was always dominated by natural influences. At the same time, the score of the comprehensive effect increased from 3.73 in 2011 to 4.50 in 2016. From 2011 to 2016, the role of natural factors was basically stable. Moreover, the score always fluctuated around 2.80. The score under human influence rose from 1.16 in 2011 to 1.46 in

2016, indicating that the impact of human factors on groundwater chemical composition increased.

The partial connection number method was used to calculate the support degree of the principal component scores of the natural and human factors for each grade to reflect the deep evolution trend of natural and human factors. The classification of the score evaluation grades is shown in **Table 2**.

According to the anthropogenic impact on the spatial distribution of the human influence degree on groundwater chemical components (**Figure 9**), the northern region with greater influence is mainly located in the Huangcun-Xihongmen-Jiugong-Yizhuang-Yinghai area, which is close to the central urban area of Beijing. In addition, the study area had high degrees of urbanisation and a dense population (Zhai et al., 2015). Furthermore, regional sewage channels such as the Xinfeng River Liangshui River and other rivers in the northern region may have caused groundwater pollution. In contrast, the central and southern regions mainly involve agriculture, so they were relatively less affected by humans.

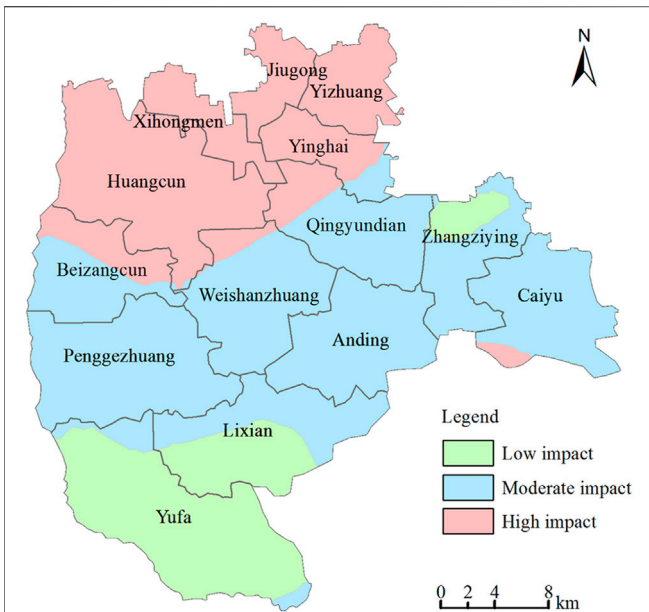
According to the calculations of the support degree of each grade under the influence of the partial connection number on human factors (**Figure 10**), the support degree of Grade 2 was always in a dominant position compared with Grades 1 and 3, indicating that the influence degree of human factors on the groundwater chemical components in the region was basically stable. Besides, the support of Grade 1 is higher than that of Grade 3, but the distance between the support curve of Grade 3 and that of Grade 1 had a decreasing trend since 2012, indicating that the degree of human influence may be further enhanced.

From a spatial perspective (**Figure 11**), the influence of natural factors on the chemical composition of the groundwater in the Yufa, Lixian, Anding, and Caiyu areas in the southern region is greater than those in the northern region and part of the central region. According to the support degree of the partial connection number under the natural influence (**Figure 12**), the support degree of Grade 2 was much higher than those of Grades 1 and 3, indicating that the influence of natural factors is the same as that of human factors, and it is basically stable. From 2007 to 2012, the Grades 1 and 3 support were basically in a balanced state. Since 2012, the support degree of Grade 3 has been significantly greater than that of Grade 1, indicating that the natural impact has increased. It may have been caused by the continuous overexploitation of groundwater in the study area and the continuous decline of the water level, which has changed the regional groundwater environment.

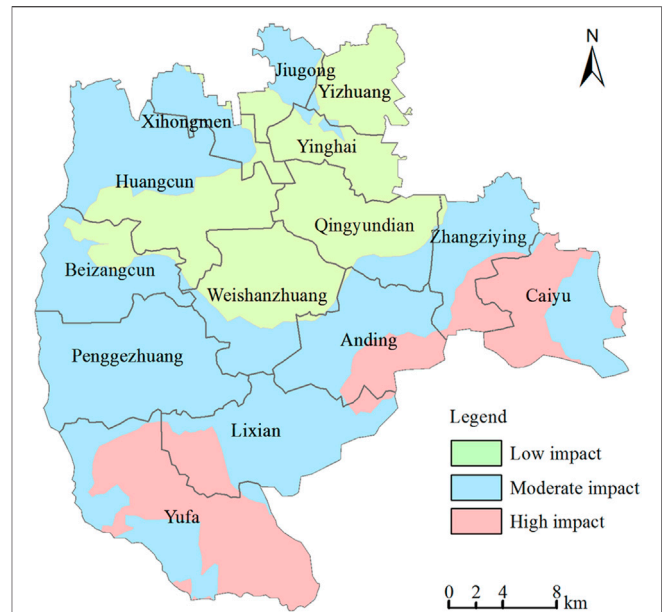
**TABLE 2 |** Principal component score evaluation level under the guidance of various factors.

| Serial number | Evaluating indicator                | Limits and levels of the evaluation index |              |              |              |              |              |
|---------------|-------------------------------------|---|--------------|--------------|--------------|--------------|--------------|
|               |                                     | Index limits                              | Index levels | Index limits | Index levels | Index limits | Index levels |
| 1             | PC2 positive load (natural factors) | ≤2.10                                     | Level I      | ≤3.02        | Level II     | ≤7.02        | Level III    |
| 2             | PC2 negative Load* (human Factors)  | ≤0.77                                     | Level I      | ≤1.55        | Level II     | ≤5.21        | Level III    |

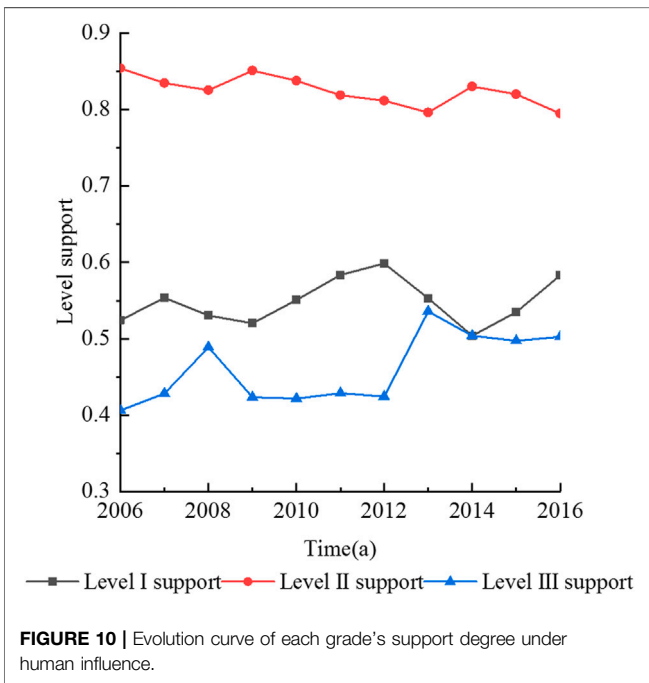
\*The PC2 negative load is the index value after the positive treatment.



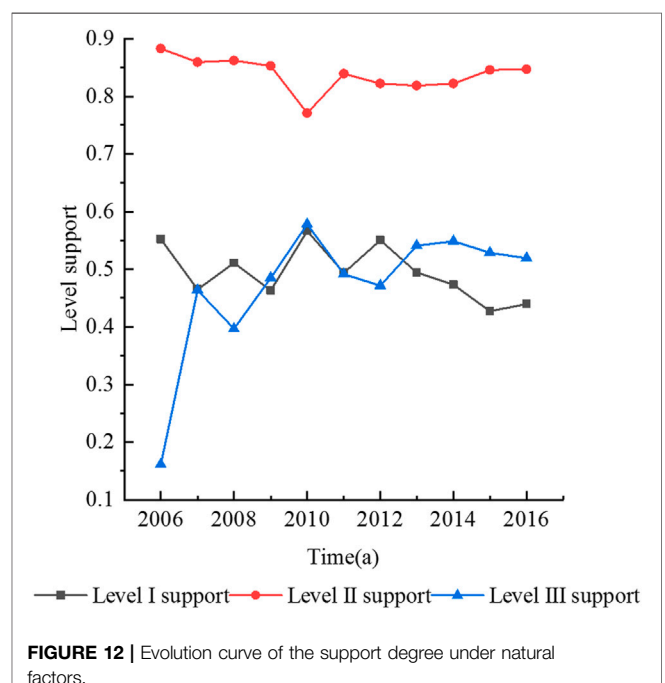
**FIGURE 9** | Spatial distribution of the influence degree of human factors on the groundwater chemical composition.



**FIGURE 11** | Spatial distribution of the natural factors' influence degree on the groundwater chemical composition.



**FIGURE 10** | Evolution curve of each grade's support degree under human influence.



**FIGURE 12** | Evolution curve of the support degree under natural factors.

On the one hand, the longer supply path of atmospheric precipitation leads to mineral intensification leaching in the vadose zone. On the other hand, with the decline of the water level, some aquifers become aeration zones, and the redox environment in the strata changes greatly. Some insoluble substances become soluble substances because of the increase in oxygen content, which leads to the deterioration of the groundwater environment.

### 5 CONCLUSION

In order to understand the effects of natural and human factors on the chemical composition of groundwater in the process of rapid urbanization. The Daxing District, a typical area in the Beijing Plain, was taken as the research object in this paper. By

collecting monitoring data on the shallow groundwater quality from 2006 to 2016, the spatiotemporal evolution trend of the groundwater's chemical characteristics and influencing factors were studied using the Piper three-line diagram, Gibbs diagram, ion combination ratio method, variation coefficient method, principal component-partial correlation method and other methods. The main conclusions are as follows:

- 1) From 2006 to 2016, the total cation concentration, total anion concentration, TDS and TH of the shallow groundwater in the study area showed an upward trend over time. The average concentration of  $\text{NO}_3^-$  had a little change, but  $\text{NO}_3^-$  in the groundwater pollution range had an increasing trend. At the same time, according to the calculation results of each ion variation coefficient, the maximum coefficient of variation was that of  $\text{NO}_3^-$  (1.48), indicating that it was mainly affected by human factors. The variation coefficient of  $\text{HCO}_3^-$  was 0.19, which indicated that it was mainly affected by natural factors.
- 2) The water chemical types in the study area from 2006 to 2016 were still mainly  $\text{HCO}_3\text{-Ca}$  water. However, with the increased  $\text{NO}_3^-$  concentrations, the main cation and anion components changed accordingly, showing an evolution trend from  $\text{HCO}_3\text{-Ca}$ -type to  $\text{Cl-Mg-Ca}$ -type. The change in the  $\text{NO}_3^-$  concentration profoundly affected the component evolution of the groundwater.
- 3) The regions greatly affected by natural factors were those in the south, whereas the northern and central regions were relatively less affected by natural factors. The areas most affected by human factors were those in the north, mainly in areas with higher urbanisation, followed by the central and southern regions. The influence of human and natural factors on the groundwater chemical composition was highly correlated with regional urbanisation and industrial distribution. The southern region was mainly dominated by agricultural planting, whereas the northern region was relatively developed in industry and commerce with a high degree of urbanisation.

In the process of urbanisation, human factors greatly impact the groundwater environment, especially regarding the input of  $\text{NO}_3^-$  into shallow groundwater. This leads to the increase in  $\text{NO}_3^-$  concentrations in aquifers, which profoundly affects the evolution of water chemical components and types. Therefore, corresponding measures should be taken to ensure the safety of the regional water supply.

- 1) Aiming at areas with high regional permeability or shallow groundwater depth, engaging in polluting industries must be strictly forbidden. The regional municipal sewage drainage facilities should be further improved. For example, domestic sewage should be collected and treated. Meanwhile, the direct discharge of domestic sewage into surface water must be prohibited. Because of the lack of surface water in the southern plains of Beijing, the water in the existing rivers is basically from treated reclaimed water in the central urban area. The concentration of  $\text{NO}_3^-$  and salt in the

reclaimed water is higher than that in natural water, which also affects the groundwater in the region (Xiao et al., 2017; Yin et al., 2019). Therefore, relevant departments must strictly follow relevant standards in treating urban sewage to reduce its impact on groundwater aquifers.

- 2) According to the spatial distribution of the main controlling factors affecting the evolution of the groundwater components, corresponding groundwater protection measures and industrial layout schemes should be formulated. The groundwater quality in the region is affected by both domestic sewage and industrial wastewater due to the high degree of urbanisation. Domestic pollution sources show an increasing trend with increasing population, but the main pollution source is still industrial pollution. Thus, heavy-groundwater-pollution industries and high-water-consumption industries should be gradually withdrawn to give way to the construction, expansion and technical transformation of industrial industries that meet the requirements for the prevention and control of groundwater pollution to prevent enterprises with serious groundwater pollution from moving to rural areas and control groundwater pollution from the source. On another note, agricultural planting is the main factor for regions where natural factors have great influence. Therefore, the efficient utilisation and recycling of agricultural resources such as chemical fertilisers, pesticides, straws and livestock manure should be comprehensively promoted to encourage cleaner agricultural production and reduce the impacts on groundwater quality.

## DATA AVAILABILITY STATEMENT

The original contributions presented in the study are included in the article/**Supplementary Materials**, further inquiries can be directed to the corresponding author.

## AUTHOR CONTRIBUTIONS

Conceptualization, S-YY and B-HM; methodology, CL; software, CL; validation, CL; for-mal analysis, CL; investigation, CL; resources, CL; data curation, CL; writing—original draft preparation, CL; writing—review and editing, CL; visualization, CL; supervision, B-HM and S-YY; project administration, S-YY and B-HM; funding acquisition, S-YY; All authors have read and agreed to the published version of the manuscript.

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## SUPPLEMENTARY MATERIAL

The Supplementary Material for this article can be found online at: <https://www.frontiersin.org/articles/10.3389/fenvs.2022.913542/full#supplementary-material>



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