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# Surrogate-assisted reliability-based design optimization of PEMFC serpentine flow channel

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In a fuel cell, flow channels are crucial components responsible for various essential functions that enable the system to operate effectively. The design of a directly coupled flow channel in a Proton Exchange Membrane Fuel Cell (PEMFC) system, assuming deterministic parameters, has been extensively studied. However, this deterministic approach neglects the inherent uncertainties in system performance during real-life operation, resulting in potentially unreliable and suboptimal performance. To address this issue, we propose a reliability-based design optimization (RBDO) of the PEMFC's channel structure, considering uncertainties in operating parameters. This paper presents a numerical model of the PEMFC in COMSOL, deterministic designs, reliability-based designs and a global sensitivity analysis on the PEMFC cell's potential output and average water activity on the membrane. Although the RBDO approach shows a reduction in cell efficiency compared to the deterministic design, it significantly improves reliability, with increases from 60.92% to 95.10% for cell potential and from 79.31% to 96.85% for water activity.

## KEYWORDS

proton exchange membrane fuel cell, reliability-based design optimization, surrogate model, sobol sensitivity analysis, flow channel

## 1 Introduction

Fuel cells are a type of alternative energy technology that generates electricity by utilizing the reaction between hydrogen and oxygen. Because of their great efficiency in comparison to conventional combustion engines and their negligible emissions, these devices are particularly intriguing. The only waste products that they generate are heat and water, however, several challenging issues stand in the way of the extensive implementation of the PEMFC (Raj and Shamim, 2014; Hu et al., 2014). This calls for an improvement in PEMFC performance. A significant number of scholars have been working to improve the performance of PEMFCs throughout the last few decades. Two different approaches for PEMFC optimization were suggested such as the operating situation and the geometric structure of the device considering a single or multiple performance of the PEMFC.

Considering a single objective optimization (single performance), Peng et al. (2017) suggested a deterministic optimization algorithm to find the best operating configuration

for PEMFC that would maximize its power density using a support vector machine surrogate model. Li et al. (2020) proposed an optimization approach that pairs a genetic algorithm with a three-dimensional, two-phase PEMFC model to optimize the partially blocked channel design for a PEMFC with a parallel flow field. The purpose of this study was to explore the impact that a blocked channel design with an increasing height arrangement has on the overall performance of the PEMFC as well as the local distributions. Fathy et al. (2021) suggested a new maximum power tracker configuration for PEMFC based on a proportional-integral-derivative controller tuned using a recent metaheuristic method of the salp swarm algorithm. Furthermore, they proposed a new formulation for the constraint objective function of the error between the fuel cell's real input voltage and the voltage at maximum power. Al-Baghdadi and Al-Janabi (2007) introduced an algorithm that accurately computes local activation overpotentials, leading to an improved prediction of the local current density distribution. Wang et al. (2021) also offers a real-time power optimization technique based on active temperature control to maximize PEMFC output power under varying ambient circumstances. They also develop an enhanced temperature perturb and observe (P&O) system to consistently obtain maximum power while maintaining an appropriate temperature reference.

Furthermore, in various responses to system performance evaluations, researchers proposed a multi-objective optimization of the PEMFC system. Such as, Li et al. (2021), developed a framework for multi-objective optimization that can improve three PEMFC performance indicators at the same time: power density, system efficiency, and O<sub>2</sub> distribution uniformity on the catalyst layer. A multi-objective genetic algorithm was also proposed by Liu et al. (2017) to optimize the operating conditions and channel structure of PEMFC. Liu et al. (2023) also used NSGA-II to find the optimal three PEMFC performance indicators: reactant flow uniformity, diffusion flux, and ohmic resistance, while considering both operating and geometric parameters. This study also developed a neural network as a surrogate model to replace PEMFC's numerical model in COMSOL. Sohani et al. (2016) defined different optimization scenarios for a PEMFC, considering efficiency, power density, the leveled cost, and size as key performance criteria, and compared the optimum results together. This study then employed a proposed approach to find the best optimization scenario, considering the preferences for both transportation and stationary applications. Wu and Luo, 2023 developed a gray correlation analysis and response surface approach for a multi-objective PEMFC performance optimization design. Optimization goals for this approach included current density, oxygen distribution homogeneity, and system efficiency. Compared to the basic model, the three-performance metrics improve significantly with optimal operating conditions. Xu et al. (2024) also introduced a comprehensive performance assessment of a high-power PEMFC system that considers the diffusion of substances in the catalyst layer under variable loading. They also proposed an NSGA-based strategy to optimize the PEMFC system's comprehensive performance.

Numerous researches also attempt to find an optimized channel structure to improve the geometric structure. Such as, Manso et al. (2011) investigated the performance of a PEMFC with a serpentine flow field design to determine how the channel cross-section aspect ratio, defined as the ratio of height to width, affects operation (performance). Furthermore, the results demonstrate that a higher

channel cross-section aspect ratio leads to improved performance capabilities. To improve the cell performance of a PEMFC, Perng et al. (2009) also look into installing a transverse rectangular cylinder along the gas diffusion layer in the flow channel. The PEMFC effectiveness was investigated concerning the width of the cylinder and varying gap sizes. In summary, all the above studies investigated different ways to improve PEMFC's performance in a deterministic manner without considering the uncertainty effect of the operating parameters. However, there were also a few studies conducted to examine the effect of operational parameter uncertainty on PEMFC performance.

Kannan et al. (2020) investigated the impact of a wide variety of input parameters, including electrochemical, physical, material, operating, and design parameters, on the performance of the PEMFC cell, as well as the management of water and thermal energy in all three regions of PEMFC operation: the activation loss region, the ohmic loss region, and the concentration loss region. Mawardi and Pitchumani (2006) also described a way to simulate how a fuel cell works when there is uncertainty, using a one-dimensional, non-isothermal description of the physical processes that control it. This study also created a sampling-based stochastic model and showed a parametric analysis to show how uncertainty in several operating parameters affects the fuel cell's power density variation. Zhu et al. (2023) proposed a model that integrates Bayesian theory and the Gated Recurrent Unit (GRU) to determine how the fuel cell voltage decay trend varies according to operating parameter uncertainty. The results show that the proposed uncertainty quantification prediction method can help hydrogen energy devices to make better decisions. These studies show that the operating parameter's uncertainties of PEMFC have a significant effect on the fuel cell's performance.

This study suggests a reliability-based design optimization for the PEMFC channel structure, considering the operating parameter's uncertainty. The cell potential is one of the most quantified effects of PEMFC due to the variability of the input parameters. In addition, this study considered investigating water management variability. As known, PEMFCs are categorized as high-temperature and low-temperature (LT) PEMFCs based on the membrane materials. In the LT-PEMFC, the oxygen reduction reaction has the slowest electrochemical kinetics. The LT-PEMFC's cell voltage losses are caused by the overpotential at the cathode as a result of the sluggish reaction kinetics. A dual-phase water system is also presented in the fuel cell while it is running at a lower temperature and atmospheric pressure. This dual-phase water system must be tightly controlled due to the membrane's stringent humidification needs, making water management complex. Flooding may also occur at this stage. Therefore, in addition to the cell potential, this study investigates the uncertain effect of operating parameters on the average water activity at the membrane during LT-PEMFC operation.

First, we developed the numerical model of the serpentine flow channel PEMFC in the COMSOL Multi-physics 6.1 version software and validated it using an experimental case, and the RBDO framework was developed. The primary contribution or novelty in this study takes the following form: (i) Gaussian processing regression (GPR) surrogate model is utilized to replace the computationally expensive COMSOL's numerical model of PEMFC. This model could be used for each response, such as cell potential, efficiency, and the average amount of water in the membrane. (ii) Sobol's global sensitivity analysis method is performed

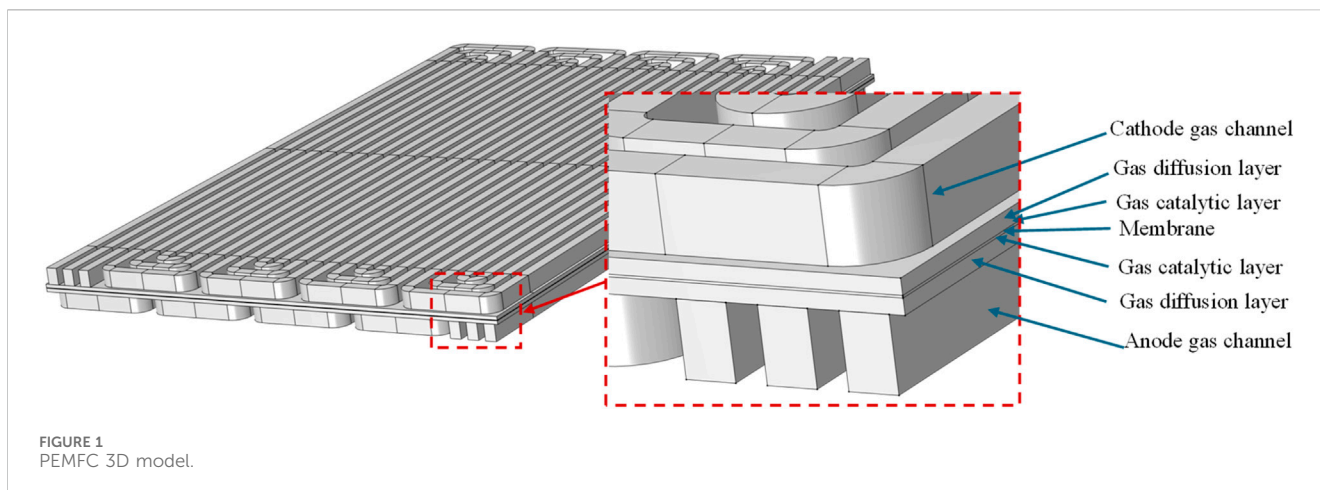


FIGURE 1  
PEMFC 3D model.

to identify and select the significant operating parameters. (iii) Using a genetic algorithm (GA), the optimal settings of the design variables are determined, and Monte Carlo simulations (MCS) are performed to determine how the input uncertain parameters affect PEMFC performance (cell potential and average water activity in the membrane). Furthermore, employing the GPR surrogate model will reduce the computational time while performing an MCS in the reliability analysis. When compared to the deterministic optimization, the proposed RBDO model shows more significant improvements in the performance reliability of the PEMFC due to the modification of the design variables.

The rest of the manuscript is also structured as follows: Section 2 discusses the numerical model in COMSOL, along with the governing equation. In Section 3, the proposed methods, including the surrogate model, sensitivity analysis, and RBDO formulation, have been explained. The results are discussed in Section 4, and finally, the conclusion is given in Section 5.

## 2 Materials and methods

### 2.1 Physical model

In this study, a three-dimensional geometry of a four-serpentine channel PEMFC is built in COMSOL (Figure 1). The geometric parameters are presented in Table 1. The model includes seven components, i.e., cathode gas channel, anode gas channel, gas catalytic layer, GDL, and proton exchange membrane. In the anode catalytic layer, hydrogen is catalytically decomposed into protons, which reach the cathode through the proton exchange membrane. The electrons generated by hydrogen decomposition go to the cathode through the load, which can generate electric power.

### 2.2 Governing equations

The models are assessed using the Batteries and Fuel Cells Module of COMSOL Multiphysics 6.1, and physical fields, such as Secondary Current Distribution, and Transportation of Concentrated Species, are used for the computational analysis.

First, the model introduces several simplification assumptions. 1) All processes operate under steady-state conditions; 2) The gas phase adheres to the ideal gas law; 3) Anisotropic electronic conductivities are used in gas diffusion layers, which have conductivities that are about an order of magnitude larger in the in-plane ( $x$  and  $y$ ) directions than in the through-plane ( $z$ ); 4) We assume a laminar and incompressible flows in the channels and porous layers; 5) The solid-phase electronic potential and membrane ionic potential, respectively, control the transportation of electrons and protons. The Governing equations of the model in Equations 1–16 are discussed as follows:

The convection and diffusion terms for the gaseous species in the channels, gas diffusion layers, and catalyst layers are given as:

$$\nabla \cdot (-D_i \nabla C_i + u_j C_i) = S_i, \quad (1)$$

where  $S$  is a specific surface area ( $m^{-1}$ ),  $C$  is the molar concentration ( $mol\ m^{-3}$ ),  $D$  is diffusivity ( $m^2\ s^{-1}$ ),  $u$  is velocity ( $ms^{-1}$ ), the subscript  $i$  is a gaseous species, and  $j$  can be gas or liquid.

The diffusion coefficient of gaseous species in the diffusion layer is expressed as:

$$D = D_{eff} \cdot (\epsilon)^{1.5}, \quad (2)$$

where the subscript  $eff$  represents effective value, and  $\epsilon$  represents a gas pore volume fraction.

Momentum equation for the gaseous species in channels, gas diffusion layers, and catalyst layers:

$$(u_j \cdot \nabla) u_j = F - \frac{1}{\rho_j} \nabla P_j + \frac{\mu_j}{\rho_j} \nabla^2 u_j, \quad (3)$$

$$\nabla \cdot u = 0, \quad (4)$$

where  $u$  is velocity ( $ms^{-1}$ ),  $P$  is a pressure (Pa),  $\rho$  denotes density ( $kg\ m^{-3}$ ),  $F$  is Faraday's constant ( $C\ mol^{-1}$ ),  $\mu$  is a dynamic viscosity ( $N\ s\ m^{-2}$ ).

A continuous function that is based on Darcy's law is used to compute the average velocity of phase  $j$ , which is given as follows:

$$\nabla \cdot \left( \rho_j \left( -\frac{\kappa \kappa_{r,j}}{\mu_j} \nabla P_j \right) \right) = H_j, \quad (5)$$

$$u_j = -\frac{\kappa}{\mu} \nabla P_j, \quad (6)$$

TABLE 1 List of the investigated parameters.

No	Input parameters	Input parameters name	Mean value	Remark
1	$w_{plate}$	Minimum plate width[m]	0.054	
2	$w_{rib}$	Rib width [m]	9.48E-4	
3	$w_{ch}$	Channel width[m]	9.48E-4	
4	$H_{gdl}$	Gas diffusion layer (GDL) height [m]	2E-4	
5	$H_{mem}$	Membrane thickness [m]	10E-5	
6	$stoich_{H_2}$	Hydrogen flow stoichiometry	1.2	
7	$stoich_{O_2}$	Oxygen flow stoichiometry	2.5	
8	$RH_{an}$	Inlet relative humidity in anode side [%]	0.70	
9	$RH_{ca}$	Inlet relative humidity in cathode side [%]	0.70	
9	$O_2N_2$	Oxygen molar fraction in cathode compared with nitrogen	0.21	$N_2O_2 = 1 - O_2N_2$
10	$T$	Cell temperature [°c]	60	
11	$L_{CL}$	Catalytic layer thickness [m]	1E-5	
12	$i_0H_2ref$	Reference exchange current density at hydrogen oxidation [A/m <sup>2</sup> ]	100	
13	$i_0O_2ref$	Reference exchange current density at oxygen reduction [A/m <sup>2</sup> ]	1E-4	
14	$a_{CL}$	Specific area of catalytic layers[1/m]	5E7	
15	$sigmas_{GDL,IP}$	Electric in-plane conductivity, gas diffusion layer[S/m]	5,000	
16	$sigmas_{GDL,TP}$	Electric thru-plane conductivity, gas diffusion layer [S/m]	200	
17	$kappag_{GDL}$	Gas permeability, gas diffusion layer [m <sup>2</sup> ]	5E-12	
18	$alpha_{O_2}$	Anodic transfer coefficient, oxygen reduction	3	
19	$eps_{GDL}$	Solid phase volume fraction, gas diffusion layer	0.4	Gas phase volume fraction, gas diffusion layers (Macroscopic porosity between agglomerates) ( $eps_{g_{GDL}} = 1 - eps_{GDL}$ )

where  $H$  is a mass source term,  $\kappa$  represents gas permeability,  $k_r$  is a relative permeability.

Since there is no electrochemical reaction occurring in the gas diffusion layers, the source term of the electronic current has been set to zero, and it is expressed as:

$$-\nabla(-\sigma_s \nabla \varnothing_s) = 0, \tag{7}$$

where  $\varnothing$  is a cell potential (V), and  $\sigma$  is electronic conductivity ( $S m^{-1}$ ).

In the same way, the source term for the ionic current in PEMFC is also zero, which is also given as follow:

$$-\nabla(-\sigma_l \nabla \varnothing_l) = 0. \tag{8}$$

The catalyst layer is the region in which the reaction takes place, and the ionic balance and electron balance can be defined in the following way:

$$-\nabla(-\sigma_{l,eff} \nabla \varnothing_l) = -S \cdot i_{loc}, \tag{9}$$

$$-\nabla(-\sigma_{s,eff} \nabla \varnothing_s) = S \cdot i_{loc}, \tag{10}$$

where  $i_{loc}$  is the local current density. According to the Butler-Volmer equation, the local current density by the porous electrode reaction in the catalyst layer is defined as:

$$i_{loc} = i_0 \left( \exp\left(\alpha_a F \frac{\eta}{RT}\right) - \exp\left(-\alpha_c F \frac{\eta}{RT}\right) \right), \tag{11}$$

where  $i_0$  is exchange current density ( $A cm^{-2}$ ),  $\alpha_a$  and  $\alpha_c$  are anode and cathode transfer coefficient, respectively, and  $R$  is a gas constant ( $mol cm^3 s^{-1}$ ).

$$M_{H_2,in} = M_{H_2} + \frac{AI}{2F}, \tag{12}$$

$$M_{O_2,in} = M_{O_2} + \frac{AI}{4F}, \tag{13}$$

where  $A$  is the total active area of the MEA ( $cm^2$ ),  $M$  is the molar flow rate ( $mol s^{-1}$ ). Hydrogen and oxygen consumption are represented by the second term on the right side of Equations 12, 13, respectively. The

hydrogen and oxygen molar flow rates that enter the channels are defined by their respective stoichiometric ratios ( $\lambda$ ).

$$M_{H_2, \text{in}} = \lambda_{H_2} + \frac{AI}{2F}, \quad (14)$$

$$M_{O_2, \text{in}} = \lambda_{O_2} + \frac{AI}{4F}. \quad (15)$$

In polymer electrolytes, there is an interconnection between ion transport and water transport. Because of this intercoupling, water content affects polymer electrolyte conductivity. When modeling the water-ion-polymer system, the transport equations for both ions and water molecules must also consider the water-ion friction forces. COMSOL used [Weber-Newman's \(2004\)](#) concentrated solution theory to show how the three pairs of water molecules, the charge-carrying ion, and the immobilized polymer matrix talk to each other. The model uses the chemical potential of water in the polymer  $\mu_0$  ( $J/mol$ ) as dependent parameter, and then the polymer water flux,  $N_0$  ( $mol/(m^2 \cdot s)$ ) is defined as:

$$N_0 = -\frac{\sigma_l \xi}{F} \nabla \phi_l + \left( \alpha + \frac{\sigma_l \xi^2}{F^2} \right) \nabla \mu_0, \quad (16)$$

where  $\xi$  is the electroosmotic coefficient and  $\alpha$  ( $mol^2/(J \cdot m \cdot s)$ ) is the water transport coefficient, with the mass balance of  $\nabla N_0 = 0$  in the domain.

## 2.3 The study's objective functions

### 2.3.1 Cell potential

The cell potential (voltage) can be conveyed as the difference between the thermodynamically reversible cell voltage and the losses caused by overpotential,

$$V = V_{oc} + \frac{R(273 + T)}{2F} \ln \left( \frac{P_{H_2} P_{O_2}^{0.5}}{P_{H_2O}} \right) - \frac{R(273 + T)}{1/2F} \ln \left( \frac{I}{i_0 P_{O_2}} \right) - \frac{I t_m}{\sigma_m} - \beta I^k \ln \left( \frac{I}{I_L - I} \right), \quad (17)$$

where  $R$  is a gas constant ( $8.314 J mol^{-1} K^{-1}$  or  $82.057 cm^3 atm mol^{-1} K^{-1}$ ),  $T$  is temperature ( $^{\circ}C$ ),  $F$  is Faraday's constant ( $96,487 C equiv^{-1}$ ) ([Ang et al., 2010](#)),  $t_m$  is membrane thickness ( $cm$ ),  $\sigma_m$  is membrane conductivity ( $\Omega^{-1} cm^{-1}$ ).  $V_{oc}$  represents the open-circuit potential,  $\beta I^k$  the amplification term for total mass transit overpotential in potential units ([Squadrito, et al., 1999](#)), and  $I_L$  denotes the limiting current density. According to the Nernst equation ([Golbert and Lewin, 2004](#)), the first two terms on the right of [Equation 17](#) reflect the thermodynamic reversible voltage. The third term is the activation overpotential ([Nguyen and White, 1993](#)), which is voltage loss owing to electrode reactions. The activation overpotential is assumed to be mostly at the cathode. The fourth term is the ohmic overpotential ([Nguyen and White, 1993](#)), the voltage drop caused by electrolyte proton resistance. Finally, the concentration overpotential ([Squadrito, et al., 1999](#)) is the mass transport limitation-induced voltage loss.

### 2.3.2 Water activity

The water activity on either side of the membrane is computed by dividing the partial pressure by the corresponding saturation vapor pressure, which depends on the cell's temperature.

$$a_{H_2O} = \frac{x_{H_2O} \times P}{P_{vap}}, \quad (18)$$

where  $x_{H_2O}$  is the water molar fraction in the gas phase,  $P$  is gas pressure (Pa), and  $P_{vap}$  is the saturated vapor pressure (Pa). The saturated vapor pressure values were obtained from the cell temperature ( $T$ ) and it is empirical relation is defined as ([Buck, 1996](#)):

$$P_{vap} = 0.61121 \times \exp \left\{ \left( 18.678 - \frac{T}{234.5} \right) \times \left( \frac{T}{257.14 + T} \right) \right\} \quad (19)$$

### 2.3.3 System efficiency

Efficiency is one of the key characteristics of a fuel cell used to evaluate its performance. The system efficiency is defined as follows:

$$\eta_{sys} = \frac{W - W_{prs}}{W_{fuel}}, \quad (20)$$

where  $W$  represents the power output of the PEMFC,  $W_{prs}$  denotes the parasitic power, and  $W_{fuel}$  is represents the fuel inherent power. They are expressed as follows:

$$W = wA, \quad (21)$$

Where  $w$  is a power density, which can be expressed as:

$$w = iV, \quad (22)$$

where  $i$  is the current density and  $V$  is the corresponding cell potential (voltage) of the PEMFC.

$$W_{prs} = W_{comp} + W_{others}, \quad (23)$$

where  $W_{comp}$  is the power consumption of the compressor and are other power losses. They are also expressed as:

$$W_{comp} = \frac{c_p T_e}{\eta_c \eta_{mt}} \left[ \left( \frac{P}{P_{in}} \right)^{0.286} - 1 \right] m_{air}, \quad (24)$$

$$m_{air} = 3.57 \times 10^{-7} \lambda_c iA, \quad (25)$$

$$W_{others} = 0.05 \times W, \quad (26)$$

where  $m_{air}$  the mass flow rate of air ( $kg s^{-1}$ ),  $\eta_c$  compressor connecting efficiency,  $\eta_{mt}$  motor efficiency,  $c_p$  specific heat constant of air ( $J K^{-1} kg^{-1}$ ),  $T_e$  entry air temperature (K),  $\lambda_c$  the stoichiometric ratio of the cathode,

The fuel inherent power ( $W_{fuel}$ ) also express as:

$$W_{fuel} = \lambda_a \frac{iA}{2F} LHV, \quad (27)$$

where  $\lambda_a$  is a stoichiometric ratio of anode,  $LHV$  Lower heating value of hydrogen ( $2.4 \times 10^5 J mol^{-1}$ ),  $F$  Faraday's constant ( $96,487 C equiv^{-1}$ ).

## 2.4 Proposed methods

The proposed methods involved four main procedures, as shown in the flow chart of [Figure 2](#). Initially, we created a

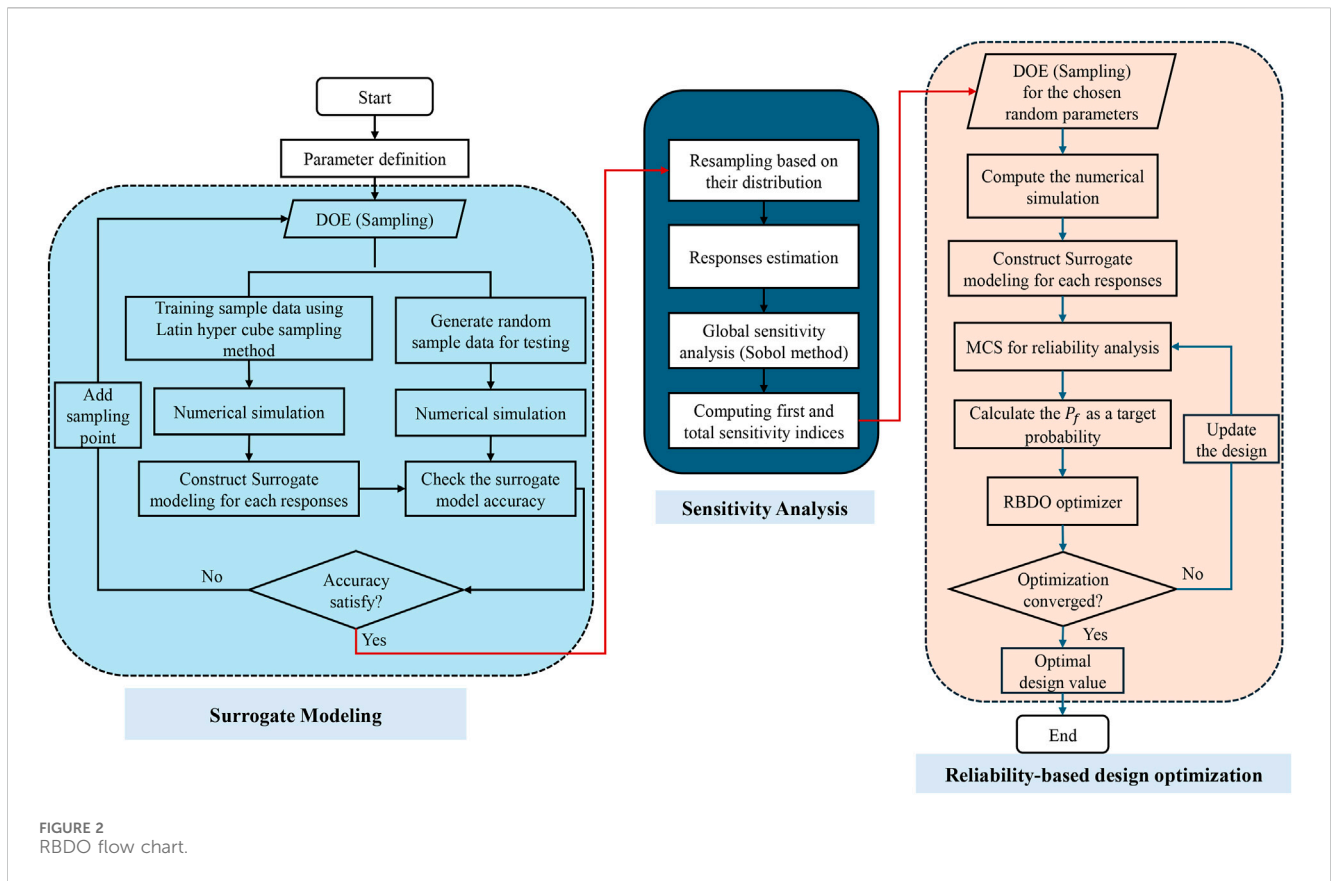


FIGURE 2 RBDO flow chart.

numerical model for the serpentine PEMFC using COMSOL multi-physics software. To confirm the accuracy of this model, we conducted an actual experiment using the same variables. The study's main goal is to develop a reliability-based design optimization model for the serpentine PEMFC flow channel, considering the fuel cell's efficiency as an objective function, cell potential (voltage), and average water activity in the membrane as a constraint function. We initially examined 17 uncertain variables to identify the most significant ones for cell potential and water activity and then employed the surrogate-based Sobol's sensitivity method. After selecting the significant uncertainty parameter for the respective constraint functions, the study conducted reliability-based design optimization using Monte Carlo simulation. The next section discusses the theory behind the proposed methods.

A surrogate model is an estimation of a high-fidelity model that is computationally accurate. Using a surrogate model, you can estimate the response for a given input without needing to carry out any additional experiments or run a high-fidelity simulation model. In other words, it can be used to estimate the output for a given set of inputs. It is also essential that the surrogate model be computationally effective. Gaussian processes (Rasmussen, 2003; Chen et al., 2023), neural network (Tripathy and Bilonis, 2018), and support vector machines (Xiang et al., 2017) are some of the most common approaches to developing a surrogate model. In this study, a Gaussian Process Regression (GPR) was employed as a surrogate model to replace the computationally expensive numerical model of PEMFC in COMSOL. We also used GPR to determine the feasibility of estimating the propagation of uncertain responses with different sets of uncertain operating parameters.

GPR is a type of supervised machine learning technique that requires only a few parameters to yield a prediction. A Gaussian process is defined as an endless extension of the multivariate normal distribution. The relationship between the input vector and the output parameter can be expressed as:

$$y = f(\mathbf{x}) + \varepsilon \tag{28}$$

where  $f(\mathbf{x})$  is the function representing the independent variable for observation,  $\varepsilon$  represents the noise added to the observed variables. The aim is to use the GPR framework to model an unknown function  $f(\mathbf{x})$ , by assuming that  $f(\mathbf{x})$  at any point  $\mathbf{x}$  is a Gaussian random variable  $N(\mu, \sigma^2)$ , where  $\mu$  and  $\sigma$  are two constants independent of  $\mathbf{x}$ . For any  $\mathbf{x}$ ,  $f(\mathbf{x})$  is a sample of  $\mu + \varepsilon(\mathbf{x})$ , where,  $\varepsilon \sim \mathcal{N}(0, \sigma_n^2)$ . For any  $\mathbf{x}, \mathbf{x}' \in \mathbb{R}^d$ ,  $\psi(\mathbf{x}, \mathbf{x}')$ , the correlation between  $\varepsilon(\mathbf{x})$  and  $\varepsilon(\mathbf{x}')$ , depends on  $\mathbf{x} - \mathbf{x}'$ . More precisely

$$\psi(\mathbf{x}, \mathbf{x}') = \exp\left(-\sum_{k=1}^d \theta_k |\mathbf{x}_k - \mathbf{x}'_k|^{p_k}\right), \tag{29}$$

where the hyper-parameter  $p_k$  is in between  $1 \leq p_k \leq 2$  which is related to the smoothness of  $f(\mathbf{x})$  to  $x_k$ , in this study, we took  $p = 2$  and the hyper-parameter  $\theta_k$  should be above zero (0), which indicates the importance of the input variable  $x_k$  on  $f(\mathbf{x})$ . To estimate the hyper-parameters  $\mu$ ,  $\sigma$  and  $\theta_1, \theta_2, \dots, \theta_k$ , let's consider a set of N-number of input points  $(x_1, x_2, \dots, x_N) \in \mathbb{R}^d$ , and their responses will be  $y = [y_1, y_2, \dots, y_N]^T$ . So, the hyper-parameters can be estimated by maximizing the likelihood function that  $f(\mathbf{x}) = y_k$  at  $\mathbf{x} = x_k$ , which is given as:

$$\mathcal{L}(y|\mu, \sigma^2, \theta) = \frac{1}{(2\pi\sigma^2)^{N/2} \|\Psi\|^{1/2}} \exp\left(-\frac{(y - \mu\mathbf{1})^T \Psi^{-1} (y - \mu\mathbf{1})}{2\sigma^2}\right), \tag{30}$$

where  $\Psi \in \mathbb{R}^{N \times N}$  is an  $N \times N$  correlation matrix whose elements depend on  $\psi(x_i, x_j)$ , and  $\mathbf{1}$  is an  $N$ -dimensional column vector of ones. To facilitate the hyper-parameter estimation by the method of maximum likelihood estimation, we can formulate a log-likelihood function as follows:

$$\ln \mathcal{L}(y|\mu, \sigma^2, \theta) = -\frac{N}{2} \ln(2\pi) - \frac{N}{2} \ln(\sigma^2) - \frac{1}{2} \ln \|\Psi\| - \frac{(y - \mu\mathbf{1})^T \Psi^{-1} (y - \mu\mathbf{1})}{2\sigma^2} \tag{31}$$

Out of the three hyper-parameters, we can use analysis to find the maximum likely values for  $\mu$  and  $\sigma^2$  by having the first derivatives of  $\ln \mathcal{L}$  with respect to  $\mu$  and  $\sigma^2$  as follow:

$$\frac{\partial \ln \mathcal{L}}{\partial \mu} = 0 \Rightarrow \hat{\mu} = \frac{\mathbf{1}^T \Psi^{-1} y^N}{\mathbf{1}^T \Psi^{-1} \mathbf{1}} \tag{32}$$

$$\frac{\partial \ln \mathcal{L}}{\partial \sigma^2} = 0 \Rightarrow \hat{\sigma}^2 = \frac{(y^N - \mathbf{1}\hat{\mu})^T \Psi^{-1} (y^N - \mathbf{1}\hat{\mu})}{N} \tag{33}$$

By substituting Equations 32, 33 into Equation 31, the likelihood function becomes dependent only on hyper-parameter  $\theta$ . Therefore, Equation 31 can be optimized to estimate  $\theta$ . In this study, MATLAB 2022b global optimization toolbox has been used to optimize the likelihood function.

We can now generate the prediction of  $y(\mathbf{x})$  at a new point  $\mathbf{x}$  given the maximal likelihood estimates of the hyper-parameters  $(\mu, \sigma^2, \theta)$ . Let  $\mathbf{r}$  represent the correlation vector between the new point and the  $N$  data points, which is the  $i^{th}$  element of  $\mathbf{r} = (\psi(x, x^1), \dots, \psi(x, x^N))^T$ . The optimal linear unbiased estimator for  $y(\mathbf{x})$  can be expressed as:

$$\hat{y}(\mathbf{x}) = \hat{\mu} + \mathbf{r}^T \Psi^{-1} (y - \mathbf{1}\hat{\mu}), \tag{34}$$

and its mean square error can be obtained using the following equation:

$$s^2 = \hat{\sigma}^2 \left[ 1 - \mathbf{r}^T \Psi^{-1} \mathbf{r} + \frac{(\mathbf{1} - \mathbf{1}^T \Psi^{-1} \mathbf{r})^2}{\mathbf{1}^T \Psi^{-1} \mathbf{r}} \right]. \tag{35}$$

Based on this, the GPR models can predict  $y(\mathbf{x})$  at new points using the observed sample data.

To check that the approximation model is accurate/precise enough, we use numerical measures of the root mean square error (RMSE) and coefficient of determination ( $R^2$ ). Both the RMSE and  $R^2$  values quantify the degree to which a regression model accurately fits a given dataset. The RMSE quantifies a regression model's accuracy in predicting the value of a response variable, measured in absolute terms. However,  $R^2$  measures the extent to which the predictor variables can account for the variability observed in the response variable.

The RMSE formulation is given as:

$$RMSE = \sqrt{\frac{1}{n} \sum_{i=1}^n (\hat{y}_i - y_i)^2}, \tag{36}$$

where  $n$  is the number of sample points,  $y$  is the actual value, and  $\hat{y}$  is the predicted one.

The formulation for the  $R^2$  is also expressed as follows:

$$R^2 = 1 - \frac{\sum (\hat{y} - y_i)^2}{\sum (\bar{y} - y_i)^2}, \tag{37}$$

where  $\bar{y}$  is the mean value of the observed data.

The impact of uncertain input parameters on the cell potential and water content on the membrane is measured by Sobol' indices. Sobol' global sensitivity analysis method is a variance-based sensitivity metrics (Sobol, 1993; Bergamini et al., 2019), which can be easily derived from the surrogate model that casts the input-output relationship. The Sobol' indices simply represents the ratio of partial variances to the total variance of  $y$ . For instance, the Sobol' index due to the  $S^{th}$ -order interactive effect of the input variables  $\{x_{i_1}, \dots, x_{i_s}\}$  is estimated as  $S_{i_1, \dots, i_s} = Var_{i_1, \dots, i_s} / Var$ , the value of which is always between 0 and 1. With a higher Sobol' index, the contribution of the associated parameters to the variance of  $y$  becomes more important. In particular, the first order, and total Sobol' indices can be derived from Equations 38, 39 respectively. The first-order effect can be defined as

$$S_i = \frac{Var_{x_i}(E_{x_{-i}}(y|x_i))}{Var(y)}, \tag{38}$$

where  $x_i$  indicates the  $i^{th}$  input parameter and  $x_{-i}$  represents all parameters except  $x_i$ . While maintaining  $x_i$  constant, the inner expectation value indicates that we can compute the mean of  $y$  over all potential values of  $x_{-i}$ . Then, the outer variance is calculated over all possible  $x_i$  values. The first-order Sobol' index  $S_i$  measures the influence due to the parameter  $x_i$  alone, which reflects the marginal effect of  $x_i$ .

The total effect is also derived as:

$$S_{T,i} = 1 - \frac{Var_{x_{-i}}(E_{x_i}(y|x_{-i}))}{Var(y)}, \tag{39}$$

The total Sobol's index,  $S_{T,i}$ , summarizes the overall contribution of the input parameter,  $x_i$ , by taking into account its marginal effect as well as its interactive effects. It should be noted that the more  $S_{T,i}$  deviates from  $S_i$ , the more preponderant the interaction effects among parameters. The second component in the equation can be seen as the first-order influence of all parameters, except the  $i^{th}$  parameter. Thus, subtracting the second term from one yields the contribution of all terms related to  $x_i$ .

## 2.5 Formulation of reliability-based design optimization

When there are uncertainties, the design optimization problem may involve both probabilistic and deterministic design criteria. Probabilistic criteria are frequently incorporated as constraints to limit the likelihood of failure. However, one can also employ them to establish objectives, like minimizing the cost's function or maximizing the probability of achieving a specific value.

Given a set of optimization design variables  $d$  and a set of random (uncertain) parameters  $X$ , the fundamental RBDO formulation can be expressed as follows:

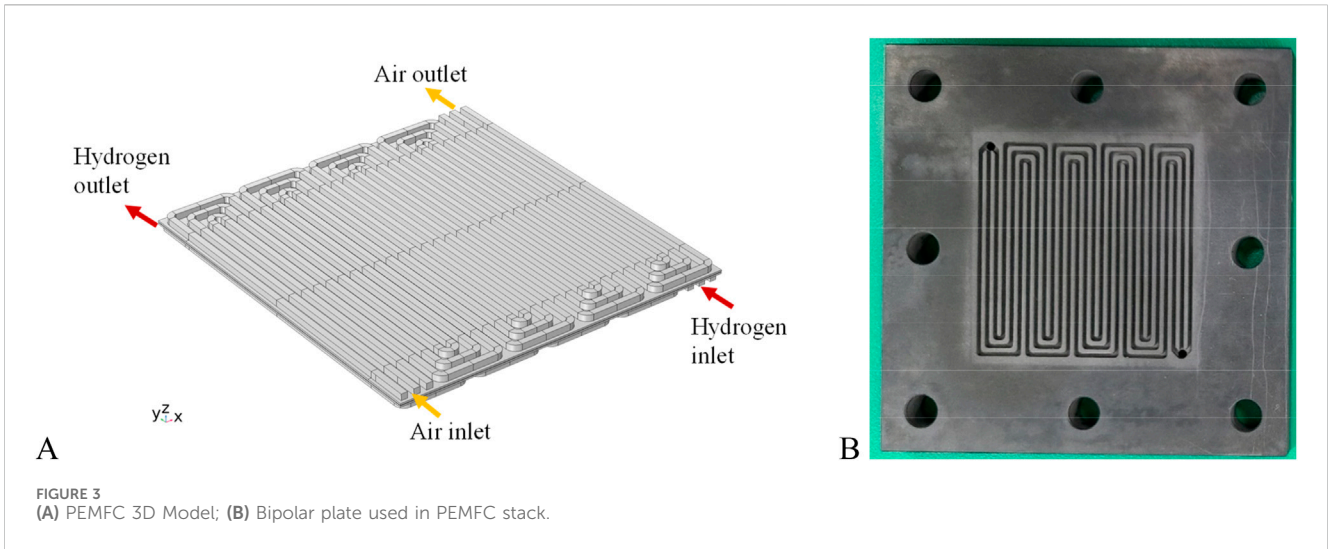


FIGURE 3 (A) PEMFC 3D Model; (B) Bipolar plate used in PEMFC stack.

$$\begin{aligned}
 & \text{minimize } f(d, \mu_x) \\
 & \text{s.t.} \\
 & P\{g_i(d, X) \leq 0\} \leq P_f^i \\
 & h_j(d) = 0 \\
 & d_L \leq d \leq d_U
 \end{aligned} \tag{40}$$

where,  $\mu_x$  represent the mean value of the uncertain (random) parameters,  $h_j$  is the  $j^{th}$  equality deterministic constraints,  $g_i(d, X)$  represents the  $i^{th}$  non-equality deterministic constraint functions. If the constrain function  $g_i(d, X_k)$  is less than 0 at the design point  $X_k$ , the design point  $X_k$  would be a performance failure point in terms of  $i^{th}$  constraint of  $g_i$ . Furthermore, P denotes the likelihood of failure, thus  $P_f^i$  means the target likelihood of failure in the  $i^{th}$  constraint.

The probability of failure,  $P_f$ , can be expressed as follows:

$$\begin{aligned}
 P_f &= P[g(X_1, X_1, \dots, X_n) \leq 0] \\
 &= \int \int \dots \int f_{X_1, X_1, \dots, X_n}(x_1, x_2, \dots, x_n) dx_1 dx_2 \dots dx_n, \\
 & \quad g(X_1, X_1, \dots, X_n) \leq 0
 \end{aligned} \tag{41}$$

where  $(x_1, x_2, \dots, x_n)$  are the uncertain (random) parameters and  $f_{X_1, X_1, \dots, X_n}(x_1, x_2, \dots, x_n)$  is the joint probability density function. Typically, it is not possible to obtain the analytical solution of Equation 41. Various numerical methods, including Monte Carlo simulation (MCS), importance sampling, and subset simulations, have been developed to calculate the probability of failure. However, this study has adapted the most well-known method, MCS, to determine the failure probability.

The MCS approach provides for the estimation of the probability of failure, which is provided by:

$$\bar{P}_f = \frac{1}{N} \sum_{i=1}^N I(X_1, X_1, \dots, X_n), \tag{42}$$

where  $I(X_1, X_1, \dots, X_n)$  is a function defined by

$$I(X_1, X_1, \dots, X_n) = \begin{cases} 1 & \text{if } g(X_1, X_1, \dots, X_n) \leq 0 \\ 0 & \text{if } g(X_1, X_1, \dots, X_n) > 0 \end{cases} \tag{43}$$

According to Equation 41,  $N$  independent sets of values  $X_1, X_1, \dots, X_n$  are obtained by the probability distribution for

each random variable and the failure function is calculated for each sample. Using MCS, an estimated probability of the failure is obtained by  $N$ -independent sets of values,  $X_1, X_1, \dots, X_n$  are derived from the probability distribution of each random variable in accordance with Equation 41, and the failure function is calculated for each sample. An estimation of the failure probability is produced by using MCS.

$$\bar{P}_f = \frac{N_f}{N}, \tag{44}$$

where  $N_f$  is the total number of cases where failure has occurred.

### 3 Results and discussion

#### 3.1 Numerical model

The numerical model of the PEMFC was developed in COMSOL 6.1 Multiphysics software, Figure 3 shows a  $54 \times 54 \text{mm}^2$  three-channel with four repeating units serpentine flow channels type. It is demonstrated that the fuel cell MEA is positioned in a sandwich configuration between two gas diffusion layers, as well as the serpentine flow channels for hydrogen and oxygen. Both the air side and the hydrogen side are placed lower than the MEA. The air side is located above the MEA. In this study, the current distribution in an LT-PEMFC is investigated for serpentine flow field patterns. This is accomplished by operating the cell in counter-flow mode, which ensures that the oxygen and hydrogen inlet flow streams are situated on opposite sides of the cell relative to the membrane's in-plane direction, as depicted in Figures 3A, B also shows the bipolar palate used for the simulation and testing.

This model solves for the electrode and electrolyte phase potentials in the gas diffusion layers and the membrane, furthermore, it solves for the molar fractions, the gas pressure and the flow velocity on each side of the membrane. In addition, the model incorporates the permeability and electroosmotic drag mechanisms that are responsible for the transport of water through the membrane. We ran a single simulation to demonstrate the numerical model result based on the initial operating parameter value shown in Table 1.



TABLE 2 Summary of the boundary and operating conditions.

Description	Expression
Hydrogen molar flow rate	$M_{H_2,in} = \lambda_{H_2} + \frac{AI}{2F}$
Oxygen molar flow rate	$M_{O_2,in} = \lambda_{O_2} + \frac{AI}{4F}$
Vapor pressure	$P_{vap} = 0.61121 \times \exp \left\{ \left( 18.678 - \frac{T}{234.5} \right) \times \left( \frac{T}{257.14+T} \right) \right\}$
Vapor molar fraction in anode flow stream	$x_{H_2O,an} = (RH_{an} \times P_{vap})/1 \text{ [atm]}$
Vapor molar fraction in cathode flow stream	$x_{H_2O,ca} = (RH_{ca} \times P_{vap})/1 \text{ [atm]}$
Hydrogen molar fraction in anode flow stream	$x_{H_2,an} = 1 - x_{H_2O,an}$
Oxygen molar fraction in cathode flow stream	$x_{O_2,ca} = O_2N_2 \times (1 - x_{H_2O,ca})$
Nitrogen molar fraction in cathode flow stream	$x_{N_2,ca} = 1 - x_{O_2,ca} - x_{H_2O,ca}$
Vapor molar flow rate, anode side	$M_{H_2O,an} = M_{H_2,in} \times x_{H_2O,an}/x_{H_2}$
Vapor molar flow rate, cathode side	$M_{H_2O,ca} = M_{O_2,in} \times x_{H_2O,ca} \times x_{O_2,ca}$
Nitrogen molar flow rate	$M_{N_2} = M_{O_2,in} \times x_{N_2,ca} \times x_{O_2,ca}$
Hydrogen mass flow rate	$m_{H_2} = M_{H_2} \times 2 \text{ [g/mol]}$
Vapor mass flow rate, anode side	$m_{H_2O,an} = M_{H_2O,an} \times 18 \text{ [g/mol]}$
Vapor mass flow rate, cathode side	$m_{H_2O,ca} = M_{H_2O,ca} \times 18 \text{ [g/mol]}$
Nitrogen mass flow rate	$m_{N_2} = M_{N_2} \times 28 \text{ [g/mol]}$
Oxygen mass flow rate	$m_{O_2} = M_{O_2} \times 32 \text{ [g/mol]}$
Anode total mass flow rate	$m_{total,an} = m_{H_2O,an} + m_{H_2}$
Cathode total mass flow rate	$m_{total,ca} = m_{O_2} + m_{N_2} + m_{H_2O,ca}$
Water mass fraction in anode flow stream	$\omega_{H_2O,an} = m_{H_2O,an}/m_{total,an}$
Water mass fraction in cathode flow stream	$\omega_{H_2O,ca} = m_{H_2O,ca}/m_{total,ca}$
Nitrogen mass fraction in cathode flow stream	$\omega_{N_2} = m_{N_2}/m_{total,ca}$

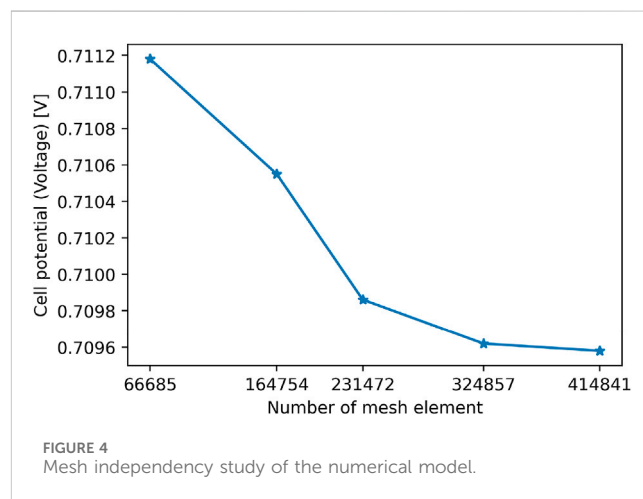
### 3.1.1 Boundary conditions

The gas flow channel inlet boundary conditions are set to mass-flow rate at the anode and cathode inlets, as indicated in Figure 3, and the inlet mass flow rate is determined using Equations 14, 15. The pressures at the anode and cathode outlets have been set to the operating pressure. In the case of the other variables, the gradients are required to be zero. It is also assumed that all of the outer walls are non-slip walls with zero fluxes.

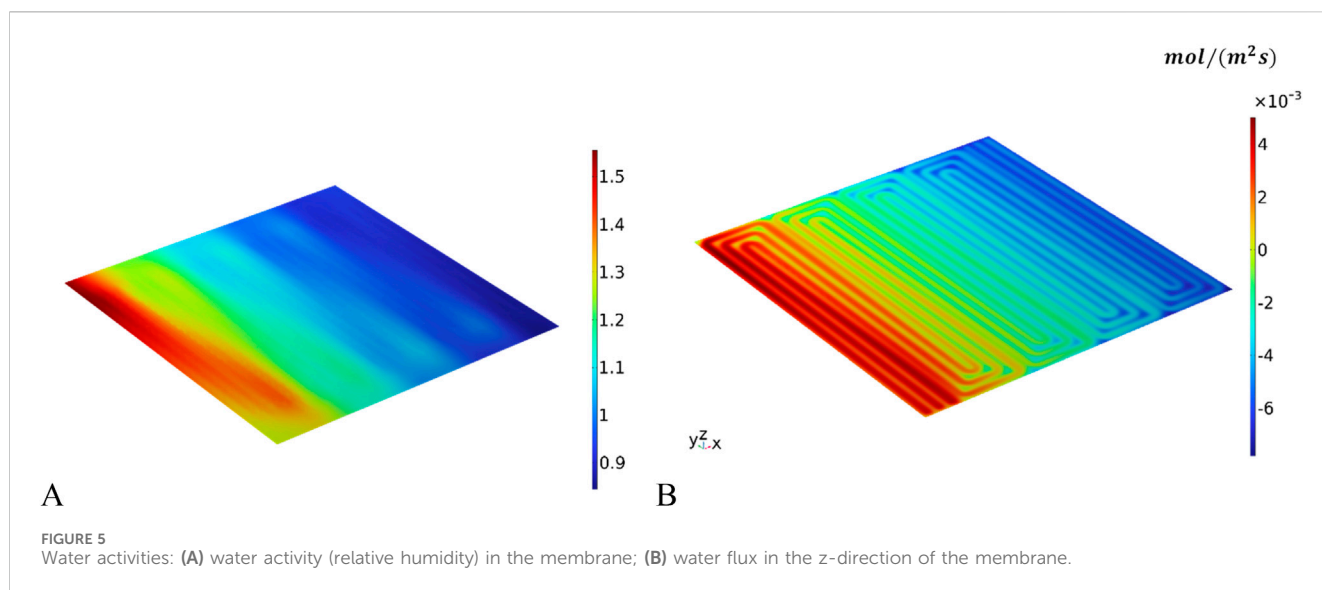
In addition, as shown in Table 1, we set the cell temperature as 60 °C, with the relative humidities of the hydrogen and air inlet streams humidified at 40% and 60%, respectively. We also set the molar flow rates of hydrogen and oxygen proportionate to the total current, with a 20% excess of hydrogen and a 150% excess of oxygen. This resulted in a stoichiometry for the flow of hydrogen that was 1.2 and a stoichiometry for the flow of oxygen that was 2.5. In addition to the operational parameters that are shown in Table 1, the boundary conditions and operating conditions of the model are summarized in Table 2.

### 3.1.2 Mesh independency study

Firstly, we conducted a convergence study to assess the mesh independence of the developed numerical model. We examined five different mesh element size models with an operating current



density of 0.6 A/cm<sup>2</sup>. The number of mesh elements employed was 66,685, 184,754, 231,472, 324,857, and 414,841, as depicted in Figure 4. As the result shows, the cell potential (voltage) values do not show a significant change between 324,857 and 414,841 mesh elements, so this study used 324,857 mesh elements for the simulation. The computational time for a single simulation



requires about 4 min on an AMD Ryzen Threadripper 1950X 16-Core Processor with 3.40 GHz and 64.0 GB of RAM.

### 3.1.3 Numerical model result discussion

Figure 5A shows the contour plot of water activity across the membrane while giving a  $0.6 \text{ A/cm}^2$  current density. As demonstrated, the water activity rises as it approaches the outflow. On the oxygen side, the increased water activity is a direct result of the water that is being created within the cell. Hydrogen depletion is the primary factor contributing to the increase in water activity on the hydrogen side. The contour plot of Figure 5A also shows that certain regions exhibit a water activity (relative humidity) value greater than one, which indicates the existence of condensation on the membrane. The membrane water flux in the z-direction is also shown in Figure 5B. As was seen, the flux is positive near the oxygen intake and the hydrogen outlet. This is because the hydrogen side is more humidified than the oxygen side, which indicates that water is carried from the hydrogen side to the oxygen side. When the oxygen input and hydrogen outlet are located near one another, the water is carried across the membrane in the negative z-direction.

### 3.1.4 Model validation

We performed three operating conditions under different temperatures and compared the numerical simulation results with the experimental data under the same working conditions. We evaluated the model for an initial case corresponding to a stack with a MEA total active area of  $54 \times 54 \text{ mm}^2$  and an operating temperature of  $50^\circ\text{C}$ ,  $60^\circ\text{C}$ , and  $70^\circ\text{C}$ . As the initial geometry and operating parameters in Table 1 show, we employed both hydrogen and air at a relative humidity of 60%. The hydrogen stoichiometric ratio is 1.25 (80% hydrogen utilization rate), whereas the air stoichiometric ratio is 2.

Figure 6 displays the effects of current density for both experimental and numerical results on the polarization and power density curves. The polarization curve, a widely used performance indicator for fuel cell systems, has a direct relationship between system efficiency and current density. As

the current density goes up, the voltage and efficiency go down because of the irreversible effects of activation, ohmic, and concentration overpotentials. On the other hand, the power density reaches a maximum at a specific current density value and rises with increasing current density. Overall, we observe that the simulation results under three different working conditions are all in good agreement with the experimental data. Given that the error falls within an acceptable range, the model appears feasible in this case, leading to the use of a numerical model for this study.

## 3.2 Sensitivity analysis

To investigate the uncertain effect and to choose the most significant noise (random) parameters for cell potential and average water activity at the membrane, first we developed a GPR model for each response. To train the model 100 simulations were conducted by applying a Latin hypercubic sampling method (Johnson et al., 1990). For testing 30 simulations were also conducted by generating a random sample dataset. The GPR model was developed for both cell potential and water activity. We calculated the  $R^2$  and RMSE values to assess the accuracy of the model. The results showed that the cell potential model had an  $R^2$  of 0.9909 and an RMSE of 0.0045, while the average water activity at the membrane model had an  $R^2$  of 0.9896 and an RMSE of 0.0134. Figure 7 displays the Actual vs. Prediction plot, which demonstrates a perfect match between the predicted and actual values of both cell potential and water activity. In addition, when we use the numerical simulation, the computing time for a single simulation takes approximately 4 min to yield the cell potential and water activity results. However, the GBR model will return the result in 0.001 s for a single run.

As discussed in Section 3, for sensitivity analysis, the study applied the Sobol method, which is one of the global variance-based sensitivity analysis methods. Figure 8 also displays the computed Sobol indices, demonstrating that parameters such as cell temperature (T), exchange current density at reference oxygen

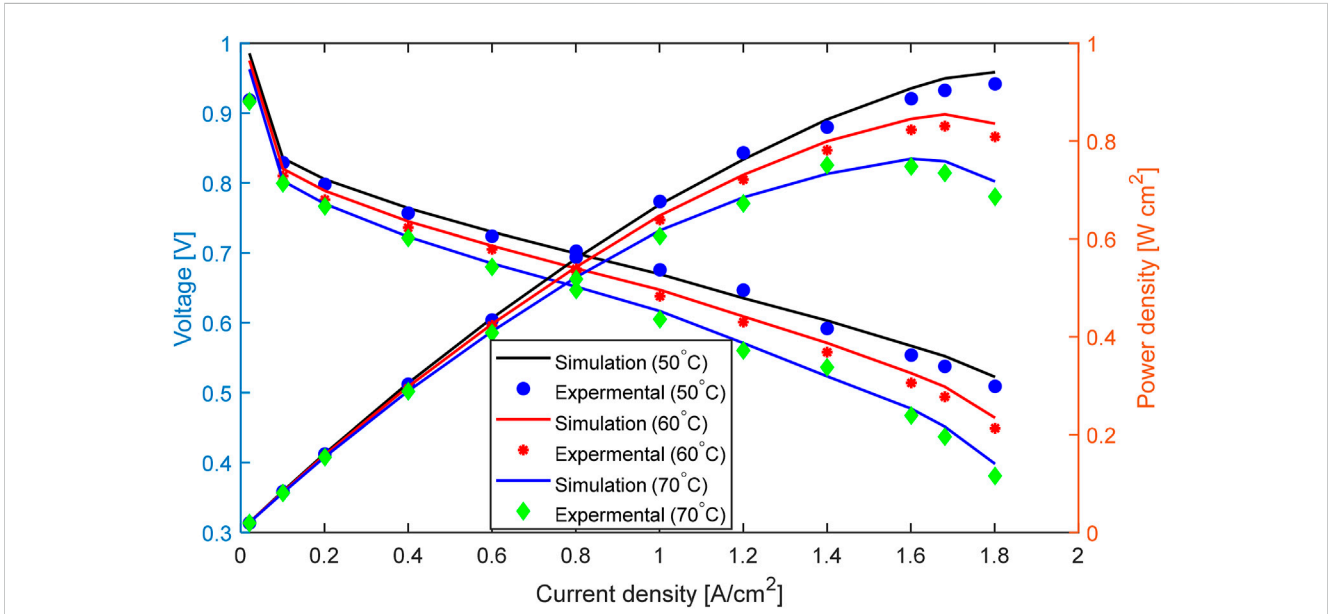


FIGURE 6 Comparison of simulation and experimental results for polarization and power density curves at three operating temperatures.

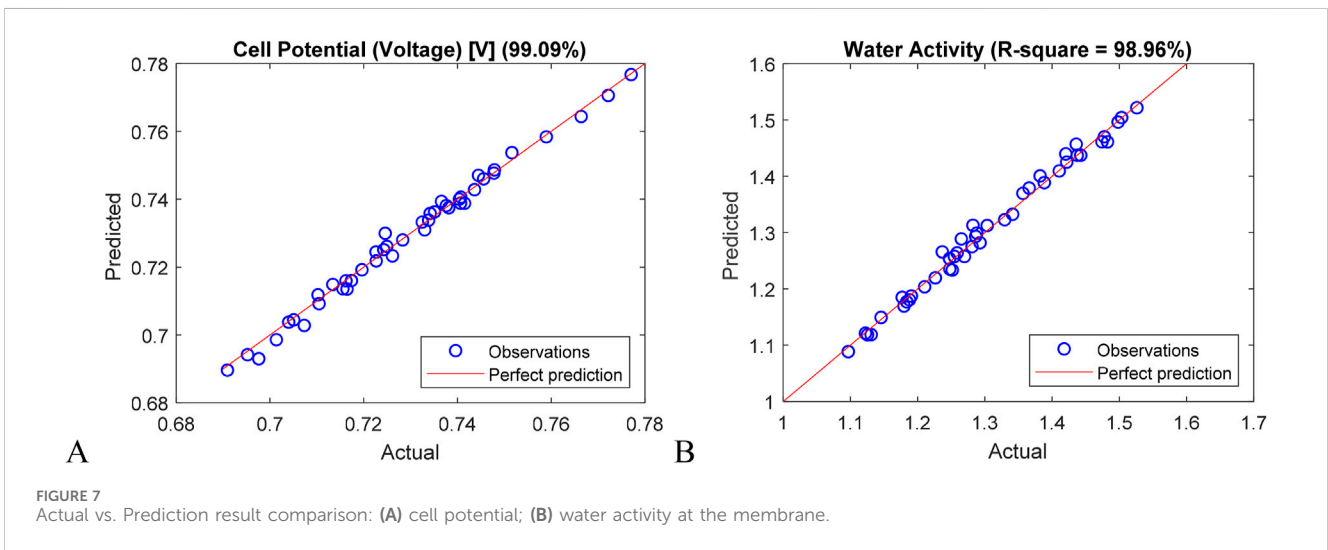


FIGURE 7 Actual vs. Prediction result comparison: (A) cell potential; (B) water activity at the membrane.

reduction ( $i_{o_2ref}$ ), specific area, catalytic layers ( $a_{CL}$ ), and the oxygen molar fraction in the cathode relative to nitrogen ( $O_2N_2$ ), significantly influence cell potential. Cell temperature ( $T$ ), hydrogen flow stoichiometry ( $stoch_{H_2}$ ), oxygen flow stoichiometry ( $stoch_{O_2}$ ), inlet relative humidity at the anode side ( $RH_{an}$ ), and inlet relative humidity at the cathode side ( $RH_{ca}$ ) also significantly influence the water content at the membrane.

### 3.3 Deterministic optimization

Deterministic optimization has been carried out after the selection of the most significant random (uncertain) variables for the respective target responses. The mean operating parameters are applied in addition to the design variables in deterministic optimization. The PEMFC's channel structure is then optimized

using a multi-objective optimization technique. Genetic algorithms and numerical simulation are combined in the multi-objective optimization method. The height of the channel  $h_{ch}$ , the width of the channel  $w_{ch}$ , and the width of the ribs  $w_{rib}$  are chosen to be variables of the optimization process. Additionally, the objective is to maximize cell potential output and efficiency while minimizing the average water activity at the membrane. The multi-objective optimization is described by the following mathematical model.

$$\begin{aligned} \min_{st} & [-E_{cell}, -\eta, Wa] \\ & d_L \leq d \leq d_U \end{aligned} \quad (45)$$

where  $E_{cell}$  is cell potential,  $\eta$  is efficiency,  $Wa$  is the average water activity at the membrane, and  $d$  is the design variables. The range of the design variables is also given in Table 3.

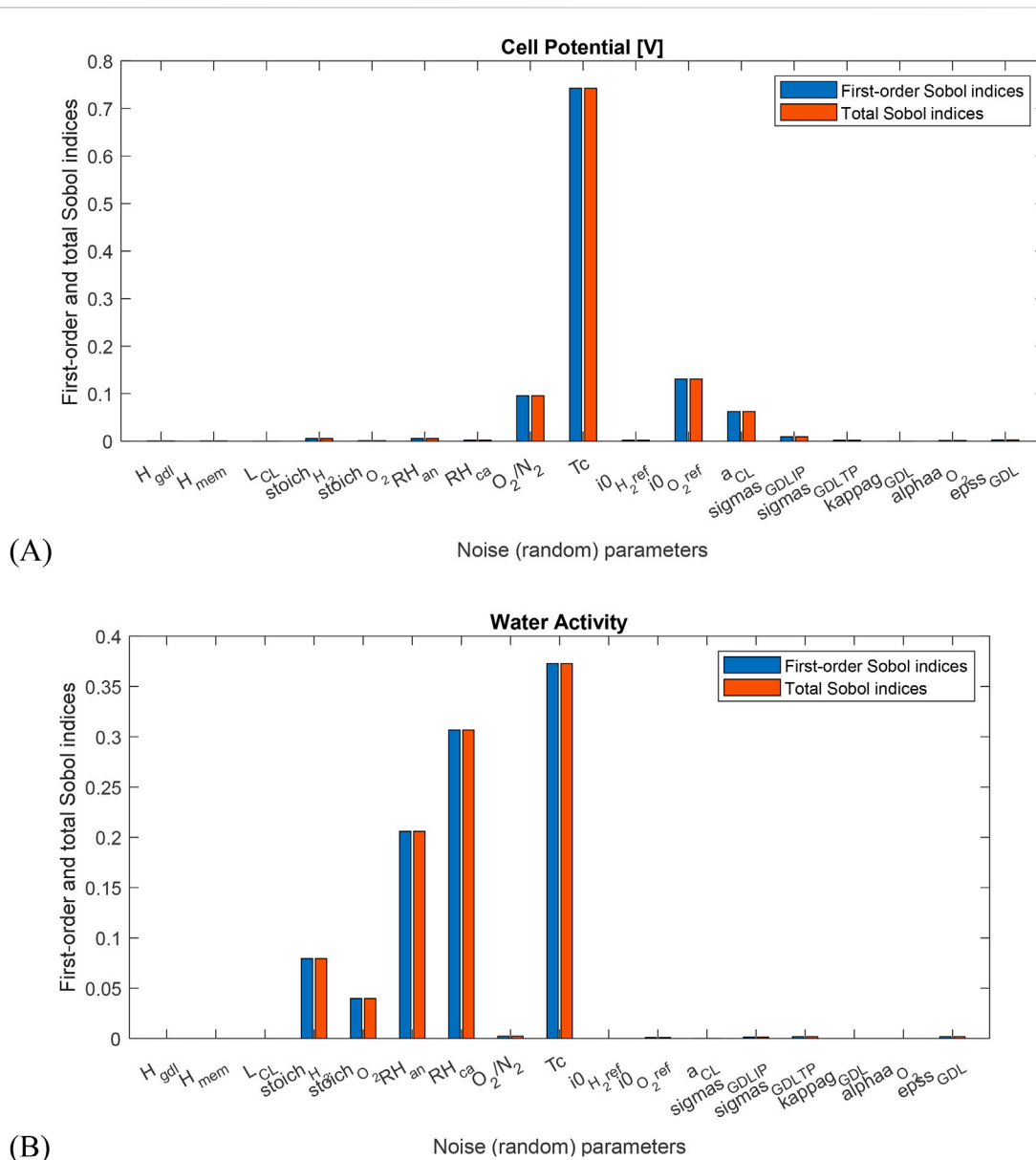


FIGURE 8 Sensitivity analysis with Sobol indices: (A) cell potential; (B) water activity.

Based on the selected random and design variables the surrogate model has been developed for each objective function which is cell potential, average water activity at the membrane and efficiency. Firstly, we developed the GPR surrogate model for each objective function by generating 100 training sample datasets using the hyper-Latin cube sampling method. Additionally, we generated 30 random datasets for model testing. Figure 9 shows the actual and prediction correlation plots for the testing dataset, along with the R2 value, demonstrating the accurate fitting of all three surrogate models.

As shown in Equation 45, the goal is to maximize cell potential output and efficiency while minimizing the average water activity at the membrane. In this investigation, we employed the multi-objective genetic algorithm to determine the Pareto front among the three objective functions. In this study, we utilize the Pareto fraction and the distance function to regulate the elitism of the

TABLE 3 Design variables information.

No	Input parameters	Description	Initial	lb	ub
1	$w_{\text{rib}}$	Rib width [mm]	0.948	0.4	1.5
2	$w_{\text{ch}}$	Channel width [mm]	0.948	0.4	1.5
3	$H_{\text{ch}}$	Channel height [mm]	0.948	0.4	1.5

genetic algorithm. By giving preference to candidates which are placed relatively far away from the front, the Pareto fraction option and the distance function both help to preserve variety on the front. This is accomplished by giving preference to candidates which are located at a distance. The Pareto fraction option places a limit on the

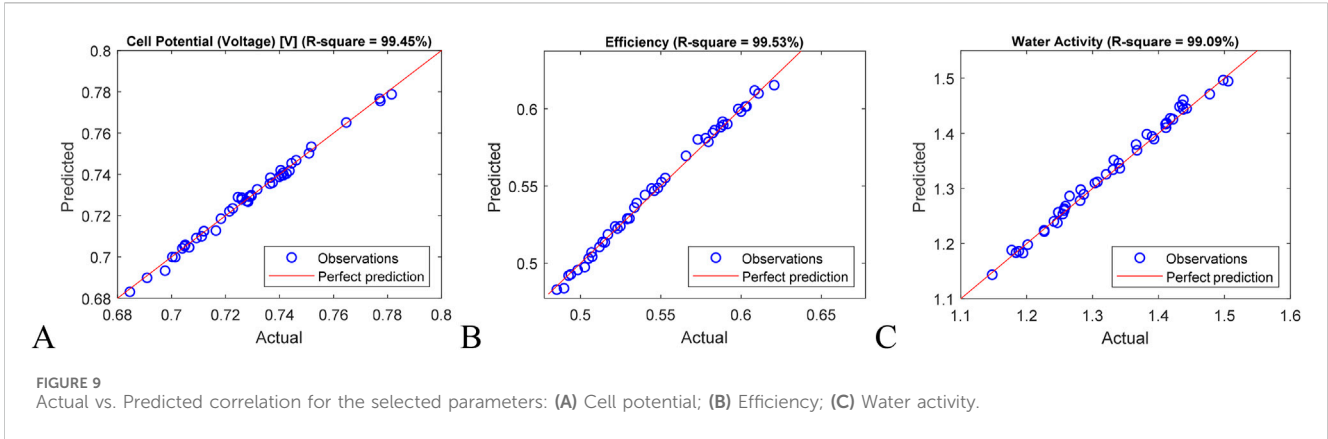


TABLE 4 TOPSIS results at different weights.

Scenario	Weights			TOPSIS results		
	System efficiency	Cell potential	Average water activity	System efficiency	Cell potential	Average water activity
1	0.333	0.333	0.333	-0.5796	-0.7572	1.2872
2	0.5	0.25	0.25	-0.5891	-0.7597	1.3429
3	0.25	0.5	0.25	-0.5803	-0.7594	1.3224
4	0.25	0.25	0.5	-0.5767	-0.7574	1.2868

number of candidates which are located on the Pareto front. In this study, we used a population size of 500 individuals, a Pareto percentage of 0.3 (which is thirty percent of the population size) and applied to 100 generations.

Several multiple-criteria decision-making methods have been suggested for different fields to find the best combination to choose from the Parato front result. Such as the analytical hierarchy process (AHP), the technique for order of preference by similarity to the ideal solution (TOPSIS), and the multi-level linguistic decision-making methodology (multi-level LDM). Various studies have also compared these methods from different perspectives (Kolios et al., 2016). This specific study employs the TOPSIS method, which is a widely used methodology in numerous fields that offers faster computational time compared to other methods (Tzeng and Huang, 2011). For the same weight the result shows that, compared to the initial (mean value), the efficiency and cell potential have increased from 0.5510 to 0.5796 and from 0.7273 to 0.7572 V, respectively. Additionally, the average water content at the membrane has decreased from 1.3416 to 1.2872. Table 4 also summarizes the TOPSIS result for different weight of weights of efficiency, cell potential, and average water activity at the membrane.

### 3.4 Reliability-based design optimization

Following the completion of the sensitivity analysis and deterministic optimization, we performed the RBDO by utilizing the significant random variables for each constraint function. Here we also employed the same surrogate model from the deterministic

optimization. The study’s RBDO formulation is also given as follows:

$$\text{minimize} - f(d, \mu_X)$$

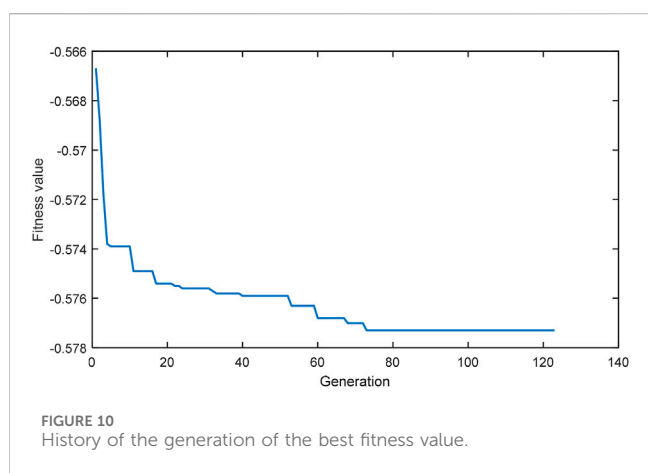
s.t.

$$\begin{aligned}
 P\{G_{W_a}(d, X) \leq G_{W_a}(d, \mu_X)\} &\leq P_f^{W_a}, \\
 X &= [T, i_o, O_2ref, a_{CL}, O_2/N_2] P\{G_{E_{cell}}(d, X) \\
 &\geq G_{E_{cell}}(d, \mu_X)\} \leq P_f^{E_{cell}}, \\
 X &= [T, stoich_{H_2}, stoich_{O_2}, RH_{an}, RH_{ca}] \\
 d_L &\leq d \leq d_U \\
 X &\sim \mathcal{N}(\mu_X, COV^2)
 \end{aligned}
 \tag{46}$$

where  $d$  denotes the design variable,  $X$  is the respective chosen random parameter,  $\mu_X$  is the mean value of the random parameters,  $f$  is the objective function (system efficiency),  $G_{wc}$  the average water activity function,  $G_{cp}$  cell potential function,  $P_f^{W_a}$  and  $P_f^{E_{cell}}$  target probability for average water activity and cell potential, respectively. Each constraint function’s feasibility region, or limit state, is also determined at each iteration. For instance, the cell potential’s limit state ( $G_{E_{cell}}(d, \mu_X)$ ) is obtained from the design variables  $d$  and the mean value of the random parameters  $\mu_X$ ; the failure state occurs when it falls less than the limit, while the average water activity’s failure value occurs when it falls above the determined limit value. The study also sets the  $P_f^{W_a}$  and  $P_f^{E_{cell}}$  to 0.05 to target a probability above 95% reliability for both average water activity at the membrane and cell potential. Table 5 also shows the random (uncertain) parameter values, assuming a normal distribution with a covariance of 0.05 for all random operating parameters.

TABLE 5 Information on the significant uncertain variables.

No	Input parameters	Description	Mean value	COV	Distribution
1	$stoich_{H_2}$	Hydrogen flow stoichiometry	1.2	0.05	Normal
2	$stoich_{O_2}$	Oxygen flow stoichiometry	2.5	0.05	Normal
3	$RH_{an}$	Inlet relative humidity at anode side [%]	0.60	0.05	Normal
4	$RH_{ca}$	Inlet relative humidity at cathode side [%]	0.60	0.05	Normal
5	$O_2/N_2$	Oxygen molar fraction in cathode compared with nitrogen	0.21	0.05	Normal
6	$T$	Cell temperature [°C]	60	0.05	Normal
7	$i_oO_2ref$	Reference exchange current density at oxygen reduction [A/m <sup>2</sup> ]	1e-4	0.05	Normal
8	$a_{CL}$	Specific area at catalytic layers[1/m]	5e7	0.05	Normal



To determine the reliability index for the average water activity in the membrane and the cell potential constraint, we used an MCS method, which involves generating 10,000 samples based on their distribution. For the objective function optimization, the study applied a genetic algorithm. In this investigation, we employed a maximum of 200 generations, a population size of 100, and a function tolerance of  $1e^{-6}$ . Figure 10 depicts the evolution of the fitness value for the objective function (system efficiency) across

generations until the genetic algorithm converges to the optimal value. As the result shows, the optimization ends before reaching the specified maximum number of generations. This demonstrates that the specified maximum number of generations is adequate.

Table 6 also summarizes the results for the deterministic and reliable optimal design variable, while Table 7 compares the optimal results of the deterministic optimization and RBDO with the initial values. The result demonstrates that the deterministic optimization yields higher efficiency, however, the reliability for cell potential is 60.92% and 79.31% for water activity at the membrane. The optimal system efficiency values obtained by RBDO are less than those obtained by DO; However, the reliability for cell potential is 95.01% and for water activity at the membrane is 96.85%, which is higher than the deterministic or the initial values. This suggests that the proposed RBDO provides a more reliable and robust design of the serpentine flow channel cross-section.

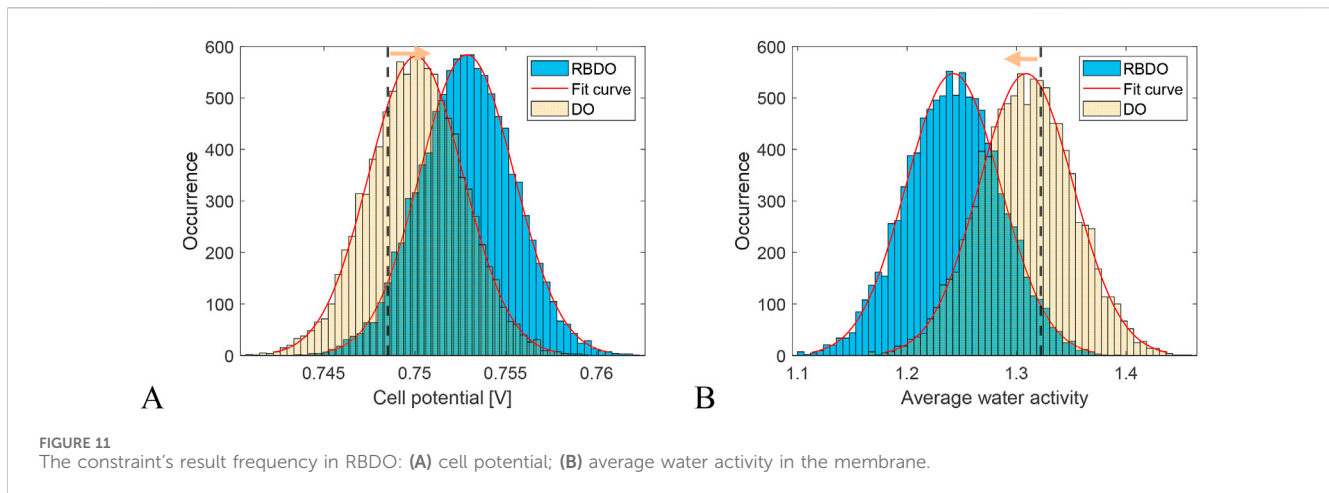
Figure 11 also shows the histograms of constraints (cell potentials and water activities) to show that the RBDO procedure determines a feasible, reliable design. As shown in the figure, the safe regions of the constraints are denoted by dashed black lines and an arrow, which represent the areas where the design is feasible due to the constraints being met. The histograms also demonstrate that the RBDO approach successfully preserves the system's level of reliability despite randomness (uncertainty), which may occur at any point in the system's operation.

TABLE 6 Optimal design variable results.

Method	Ribs width [mm]	Channel width [mm]	Channel height [mm]
DO	0.715	0.707	0.978
RBDO	0.733	0.721	0.923

TABLE 7 Result comparison.

	Efficiency [%]	Cell potential [V]	Water activity	Reliability	
				Cell potential (%)	Water activity
Mean (Initial)	0.5510	0.7273	1.3416	57.90	69.10%
DO	0.5796	0.7572	1.2872	79.31	60.92%
RBDO	0.5773	0.7485	1.3222	95.10	96.85%



## 4 Conclusion

RBDO of the PEMFC was performed by applying an uncertainty model to enhance the performance efficiency. It was assumed that uncertainty models of the operating parameters for the cell potential and average water activity in the membrane calculation are statistical models with normal distributions. First, we developed a numerical PEMFC model in COMOSL, followed by a surrogate model to replace the computationally expensive model. The GBR model demonstrates an efficient technique to solve the problems of performance nonlinearity and relatively high computing costs using numerical simulations, which demonstrate that with the GBR model, the computation time has been reduced from 4 min to 0.001 s. Then, a sensitivity analysis was conducted to identify and select the most significant operating parameters for the cell potential and average water activity at the membrane responses. MCS was executed to calculate the probability of failure of cell potential and average water activity in the membrane. In addition, the GPR model was adopted to determine the efficient performance levels of MCS and RBDO. We also utilize the genetic algorithm to exploit its advantages in the global optimization search process, making it particularly well-suited for the optimization problem of PEMFC.

The result of RBDO for the PEMFC system efficiency was compared with the result of deterministic optimization and the initial design variables. Based on the result, RBDO serves as a good alternative approach for nonlinear and uncertain design problems like PEMFC design optimization. Moreover, RBDO can improve the results of the initial channel cross-section and make up for the weak points of deterministic optimization by considering the uncertainty property of the serpentine PEMCF channel cross-section design. The results show that the deterministic optimization yields higher efficiency; however, the reliability for cell potential is 60.92% and 79.31% for water activity at the membrane. The RBDO gives us a cell potential reliability of 95.01% and an average water activity at the membrane of 96.85%, which is better than the initial or deterministic values. As a result, RBDO with an uncertainty model can be a useful and practical technique for obtaining a reliable and robust PEMFC design. This method can also be used to investigate and design reliability for any other fuel cell type by choosing the appropriate

objective functions (parameters). Further, in this study, water management and cell potential were considered as objective functions (parameters), but other parameters also need to be considered when designing PEMFCs, so a future study can be conducted to implement this study's proposed method considering other parameters, such as oxygen concentration in the catalytic layer and pressure drop in the channel.

## Data availability statement

The raw data supporting the conclusions of this article will be made available by the authors, without undue reservation.

## Author contributions

MA: Formal Analysis, Investigation, Software, Visualization, Writing—original draft. BK: Conceptualization, Funding acquisition, Resources, Supervision, Writing—review and editing. M-GK: Validation, Writing—review and editing. H-SK: Project administration, Resources, Writing—review and editing.

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## Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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