



## Underutilized Lignocellulosic Waste as Sources of Feedstock for Biofuel Production in Developing Countries

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The need for a reliable and sustainable energy source, stability in energy price and solution to environmental challenges of fossil fuel has led to searching for an alternative energy source to fossil fuel. Several alternative sources have been developed over time, but they are limited in one form or another. However, biofuel such as bioethanol has been identified as a superb alternative with superior properties to fossil fuel. One major challenge with biofuel is the high production cost resulting from feedstock, which may also serve as a food source. In order to address this challenge, research is focused on searching for cheap and sustainable feedstock for biofuel production. Currently, attention is on lignocellulosic waste as feedstock with a keen interest in developing the most appropriate technique for processing it to bioethanol, especially in developing countries, which is the focus of this review. This review involves converting lignocellulosic waste to bioethanol and the pretreatment steps involved as well as its challenges, prospect and economic aspect. Among the pretreatment steps reported, biological treatment remains outstanding but with a few challenges which can be managed. Biofuel has come to stay in developing countries with lots of opportunities that favours its production cost. Although the high cost of enzyme production has been identified as a challenge to the economic viability of lignocellulosic bioethanol, there is hope that developing an efficient bio-system for simultaneous saccharification and fermentation (SSF) and consolidated biomass processing may help circumvent the challenge. In conclusion, the effective utilization of lignocellulosic waste in an efficient biocatalyst system can serve as an economically viable means to overcome the challenge posed by fossil fuel.

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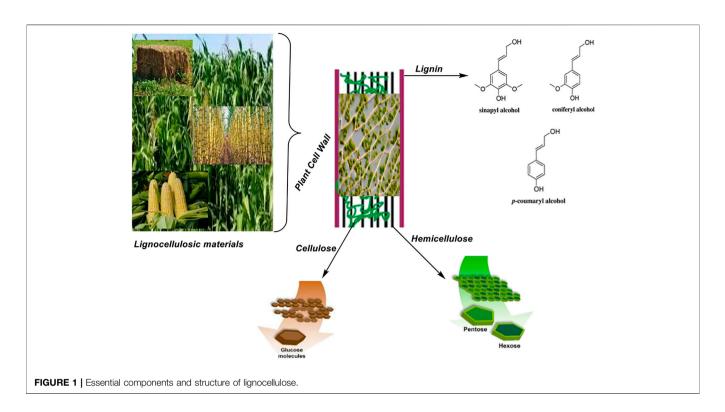
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### INTRODUCTION

Lignocellulosic materials are biomass capable of serving as a viable resource for renewable bioenergy. They are of plant biomass and in large quantity on earth. They include agricultural, forestry or wood wastes, which mainly contains cellulose, lignin, tannin, saponin and hemicellulose. Cellulose makes up the entire polysaccharide structure of plant cell wall corresponding to about 30–50% dry weight of lignocellulose (Harris and DeBolt, 2010), containing a linear chain of  $\beta(1\rightarrow 4)$  linked D-glucose units. The cellulose may be degraded by enzyme cellulase to glucose which is finally converted to biofuel. Unlike cellulose, hemicellulose contains many heteropolymers such as glucomannan, glucuronoxylan, xyloglucan, xylan and arabinoxylan (Scheller and Ulvskov, 2010), giving it a random amorphous structure. It may be hydrolyzed to simple sugar by acid or base hydrolysis



or by the action of enzyme hemicellulase. Simple sugar may serve as a source of biofuel. Authors (Scheller and Ulvskov, 2010) have reported lignin as the second most abundant carbon source after cellulose. It is different from cellulose and hemicellulose because it is a non-carbohydrate aromatic heteropolymer that may be obtained from three different building blocks, including coniferyl alcohol (G), p-coumaryl alcohol (H), and sinapyl alcohol (S) as shown in Figure 1. It is necessary to remove lignin during biofuel production since it does not have a carbohydrate base structure that can contribute simple sugar as a monomeric base for biofuel production. Plant-sourced biomass such as lignocellulosic materials is an important feedstock for many biotechnological processes and a sure sustainable and renewable source of energy for now and the future. A study (Kumar et al., 2008) revealed the capacity of a process plant to produce up to  $1.3 \times 10^{10}$  metric tons of biomass (on a dry weight basis) annually via photosynthetic fixation of CO<sub>2</sub> in the presence of sunlight. The study further revealed that such biomass possesses equivalent energy of about two-thirds of the world's energy requirement. This showcased lignocellulosic materials, an example of plant-sourced biomass as a potential sustainable energy source for the present and future needs.

The use of lignocellulosic materials as feedstock for bioethanol production as an alternative fuel to fossil fuel is an essential means of circumventing fossil fuel challenges. The conversion of lignocellulosic materials may be achieved in three steps: removal of lignin to have access to cellulose and hemicellulose, hydrolysis of the carbohydrate polymers to obtain free sugars and fermentation of the free sugar to bioethanol. A large quantity of lignocellulosic waste generated annually from agricultural and food processing industries can serve as feedstock for biofuel production. The use of these wastes from agricultural and food processing industries can help reduce the cost of producing biofuel as well as maintain a clean environment. Other materials like grasses, forest waste and municipal solid wastes are also cheap, abundant renewable resources that may serve as feedstock. The use of lignocellulosic materials for biofuel production can help reduce greenhouse gases emission, which may help mitigate global warming (Jeswani et al., 2020). Presently, research is focused on improving or developing adequate technology that can efficiently convert lignocellulosic waste to biofuel with very minimal production cost. The use of lignocellulosic waste is one way to avoid competition between the use of agricultural proceeds as food or biofuel. As previously reported (Elumalai et al., 2018), furfural, 5hydroxymethylfurfural, and levulinic acid are examples of potential chemicals obtained from lignocellulose conversion that may serve as a precursor for fuel or building blocks for other valuable products.

The use of lignocellulose as a source of biofuel may suffer from a few challenges; this may include the difficulty associated with finding and sustaining a cost-effective technology for cellulose conversion and sugar extraction processes. Achieving this requires specific pretreatment steps due to the bio-component of lignocellulose substrate, which needs to be taken care of. Such pretreatment may end up increasing the production cost. Apart from this, the use of various chemicals in lignocellulose pretreatment generates chemical wastes or inhibitory compounds, which may negatively affect the performance of the hydrolysis and fermentation processes. Moreover, distinct removal of the chemical wastes or inhibitory compounds from the pretreatment stage further increases production cost as this

TABLE 1	Composition of selected lignocellulosic materia	als.
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Biomass	Cellulose (%)	Hemicellulose (%)	Lignin (%)	References
Grasses	25–40	25–50	10–30	(Stewart et al., 1997), (Hon et al., 2000)
Cotton stalk	31	11	30	Rubio et al. (1998)
Bamboo	49–50	18–20	23	(Reguant and Rinaudo, 2000), (Alves et al., 2010)
Wheat straw	35–39	22–30	12–16	(Prasad et al., 2007), (Grohmann et al., 1986)
Sugarcane tops	35	32	14	Jeon et al. (2010)
Tall fescue straw	31	20	14	Kumar and Murthy, (2011)
Douglas fir	44	11	27	Isikgor and Becer, (2015)
Switchgrass	35-40	25–30	15–20	Howard et al. (2003)
Nut shell	25-30	22–28	30-40	Sinner et al. (1979)
Jute fibre	45-53	18–21	21–26	Mosihuzzaman et al. (1982)
Elephant grass	38–40	18–24	25	De Vrije et al. (2002), Sørensen et al. (2008)
Hardwood	40–55	20–40	18–25	Betts et al. (1991), McKendry, (2002)
Rice hulls	28-38	17	16–22	Guo and Rockstraw, (2007), Rabemanolontsoa and Saka, (2013

may even affect the quality of biofuel produced. In order to avoid the challenges that come with the use of chemical pretreatment, enzymatic pretreatment of lignocellulose is considered to be more efficient with less chemical waste. Although the use of enzymatic pretreatment is selective, precise, and environmentally friendly, it is expensive with a high cost to sustain, which poses a challenge in poor developing countries as these countries lack the financial capacity to sustain or maintain the technology. The enzymes are temperature sensitive, and the process conditions will have to be carefully maintained. Different strains of microorganisms have been studied for their capacity and novelty to function as viable sources of lignocellulolytic enzymes. A previous study (Druzhinina et al., 2012) reviewed the three species of Trichoderma, namelv Trichoderma atroviride. Trichodermareesei, and Trichoderma virens, as sources of carbohydrate-active enzymes. Large number of lignocellulosic wastes are generated yearly in developing countries. Moreover, these lignocellulosic wastes are inappropriately disposed creating a waste disposal challenge. Therefore, it is important to identify the possible use of these waste by converting them to useful products making it a win-win situation more like waste-towealth. However, some of these wastes have been identified but some of them are still underutilized. Therefore, this study aimed to consider the prospect of underutilized lignocellulosic wastes generated in developing countries as possible sources of feedstock for biofuel production, focusing on bioethanol with keen interest on evaluating the different pretreatment steps and process modifications necessary for obtaining better bioethanol production.

#### Composition and Structure of Lignocellulose Biomass

The main compositions of lignocellulose are cellulose (20–50%), hemicellulose (15–35%) and lignin (5–30%) as previously reported (Lynd et al., 2002; Chen, 2014; Michelin et al., 2015); this comes along with other molecular compounds which are organized in complex structures as shown in **Figure 1**. The cellulose, hemicellulose and lignin composition of some selected lignocellulosic materials are summarized as presented in **Table 1**. Despite the potential exhibited by lignocellulosic materials, they express resistance to degradation due to the hydrophobicity of lignin, the lignin-hemicellulose matrix encapsulation and the crystallinity of cellulose (Isikgor and Becer, 2015). Cellulose is the main component of interest in lignocellulose for the production of biofuel. It is a polymer containing d-anhydro-glucopyranose units linked together by  $\beta$ -1,4 linkages; with number of glucose molecules in a chain varying from 100 to 1,000 depending on the source of the lignocellulose and this has a key role to play in the yield of biofuel because the higher the number of glucose molecule per chain length the higher the biofuel conversion yield would be expected to be. Cellulose contains hydroxyl groups that promote inter and intramolecular hydrogen bonding resulting in crystalline structure (with some level of amorphous structure) of cellulose, which reduces its solubility in water and resistance to hydrolysis that poses resistance to ease of conversion to glucose units and finally, bioethanol. Hemicellulose, on the other hand, is heterogeneous, containing sugars and uronic acids. It is amorphous, short-chained and capable of joining lignin and cellulose together (Chundawat et al., 2010). It contains several heteropolymers such as, glucomannan, galactomannan, xylan, glucuronoxylan, xyloglucan, and arabinoxylan (Isikgor and Becer, 2015); the composition of hemicelluloses varies depending on the source of the lignocellulose. Lignin remains the most complex component of lignocellulose, which is amorphous and capable of forming a three-dimensional and irregular network when complexed with cellulose and hemicellulose; this complex provides structural stability for the lignocellulose. As shown in Table 1, the distribution of cellulose, hemicellulose and lignin from a single source is not uniform.

## Lignocellulosic Waste as Sources of Feedstock

A large number of wastes are annually generated in developing countries. Most of the wastes are of an environmental challenge as they constitute a nuisance in the environment when disposed of inappropriately; waste disposal is a major problem in poor developing countries. It is important to find practical applications for these wastes. Lignocellulosic wastes are one of the main groups of wastes suitable for the production of biofuel. Sources of lignocellulosic waste may be classified as waste generated from forestry resources, agricultural resources, and by-products of agricultural-based industry. The waste from forestry resources is generated from forest cutting, reforestation, forest processing, forest protection, etc., while waste generated from agricultural resources include crop straws, agricultural crops, by-products from crop harvest, etc. An essential way of using these lignocellulosic wastes is in their conversion to biofuel, majorly bioethanol production, where they supplement and replace nonrenewable sources of energy. The lignocellulosic wastes are the most abundant waste in developing countries with significant cellulose and hemicellulose content, requiring pretreatment steps before conversion to biofuel. The use of lignocellulosic wastes as a renewable source of biofuel depends on developing an economically viable process technique (Kricka et al., 2015). The process requires the use of cellulase to hydrolyze the β-1,4-glucosidic linkages of cellulose to low molecular weight sugar that can be converted to bioethanol.

# Agro-Based Wastes as Lignocellulosic Materials for Biofuel Production

Several agricultural wastes have been screened for their potential use as sources of biofuel. Biofuel production using agricultural or industrial wastes containing lignocellulosic material is of great interest because it minimizes the human 'food versus fuel' dilemma encountered with the substrates of the firstgeneration biofuel (Clauser et al., 2021; Malode et al., 2021). Lignocellulosic wastes are considered to be second-generation feedstock. Examples include bagasse wastes, rice wastes, corn wastes, spent grains and miscellaneous wastes.

Bagasse waste is a by-product of the sugar industry; it is a dry pulpy fibrous material left after crushing sorghum or sugarcane stalks. Apart from its high moisture content, the composition is similar to that of wood. The dry weight contains cellulose, lignin, and hemicellulose, etc. It has found application in biofuel production due to the hydrolysis of its hemicellulose and cellulose (Bezerra and Ragauskas, 2016). Sugarcane bagasse has played an essential role in bioethanol production. Developing countries like Brazil, where the sugarcane industry plays a crucial role in power generation, receive incentives from the government to support energy production. Molasses containing sugar have been fermented for bioethanol production as previously reported (Bhatnagar et al., 2016), other studies (Merschmann et al., 2016; Barbosa et al., 2017; Carminati et al., 2019) have revealed the use of sugarcane bagasse in bioethanol production.

Cashew apple bagasse is the by-product obtained from the cashew apple juice industry. It is a lignocellulosic waste material that is underutilized globally without a specific use in most developing countries. It is considered as waste, and about 15% (w/w) of cashew apple bagasse produced during juice production are discarded as waste (de Souza Bezerra et al., 2015); finding application for this waste is essential as the lignocellulose content suggests the potential of being a good source for biofuel production. Brazil, a developing country, accounts for 11% of the world cashew nut production, which represents over five million tons of cashew production (dos Santos Lima et al., 2012);

with the large volume of waste produced, cashew agroindustry can serve as an outstanding source of raw material for biofuel production. To achieve the best use of cashew apple bagasse, pretreatment methods have been developed, such as the use of dilute sulphuric acid (Rocha et al., 2014), sodium hydroxide (Rodrigues et al., 2011) and alkaline hydrogen peroxide (da Costa et al., 2015).

Rice wastes are mainly rice straw and husk. They are currently receiving close attention as an alternative source of energy due to their cellulose and hemicellulose contents. They have been reported to contain a high amount of cellulose and hemicellulose (Ibitoyea et al., 2020; Tufail et al., 2021). Over 120 million tons of rice husks are produced annually as waste. One primary application of rice husk is biofuel production (Hossain et al., 2018); this is the situation in a country like India, where large rice production occurs annually. Studies have shown that subjecting rice husk to the appropriate method can generate sufficient power to help power communities in a country like India and China, where rice is produced in substantial quantities annually (Pradhan et al., 2013). However, rice husk has been reported to be poorly saccharified due to its high levels of lignin. It produces high fermentation chemical inhibitors, which indicates the need for a better treatment procedure (Wood et al., 2016; Khaleghian et al., 2017). Compared to other lignocellulosic cereals, rice husk and straw contain high levels of silica in their cell walls, with rice husk having the highest (Wu et al., 2018; Chun and Lee, 2020). An estimation of four million tons of rice straw is produced in Egypt annually and globally, producing about 205 billion liters of bioethanol annually (Belal, 2013). It contains lignin (5-24%), cellulose 32-47%, ashes (18.8%) and hemicelluloses (19-27%), as previously reported (Belal, 2013). Bioethanol production from rice straw by saccharification and fermentation with optimized cellulase cocktail and fermenting fungus has been reported (Takano and Hoshino, 2018). A study reported bioethanol production from a mixture of rice hull and orange peel wastes (Taghizadeh-Alisaraei et al., 2019); the study revealed a bioethanol yield of 0.295 L kg<sup>-1</sup> from dry rice hull and orange peel. Rice bran is also an essential source of feedstock for biofuel production; an estimated annual world production is about 76 million tons (Kahlon, 2009). It contains about 34-52% carbohydrate (Webb, 1991), which suggests it is a primary source of glucose for bioethanol production. Production of bioethanol from rice bran using Bacillus cereus strain McR-3 has been reported (Tiwari et al., 2015).

Corn is an integral part of food in most developing countries. With about a 44.74% increase in corn production in 2001–2011 (Zhang et al., 2012), the generation of corn waste such as corn husk and corn stover increased. The cobs, leaves and stalks left after the harvest of corn are referred to as stover. It is abundantly available, and it is considered as waste since there is no specific use for it presently. Bioethanol production was reported from corn cobs via a combined thermochemical and fermentative processes (Luque et al., 2016). Corn cobs have been reported to be a better source of glucose than stalks and leaves (Chun and Lee, 2020). A study (Akinola and Erkurt, 2014) using wastewater containing cornstalk for bioethanol production was reported to have desirable output on reducing sugar yield. The study further

<b>TABLE 2</b>   Selected biofuel producing companies, feedstock and location.	
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Company	Location	Feedstock
Global biofuels	Nigeria	Sweet sorghum
Caltech Ventures	Ghana	Cassava
China Resources Alcohol Corporation	China	Corn stover
ACABIO Coop. Lta	Argentina	Corn and grain sorghum
ANDRITZ Feed and Biofuel	Chile	Biomass
Miranda	Russia	Wheat
ETIP Bioenergy	Brazil	Sugarcane
Zagros Green Fuel development Company	Iran	Sugarcane
Mabele Fuels Pty (Ltd.)	South Africa	Sorghum
Bioenergy	South Africa	Distillers grain

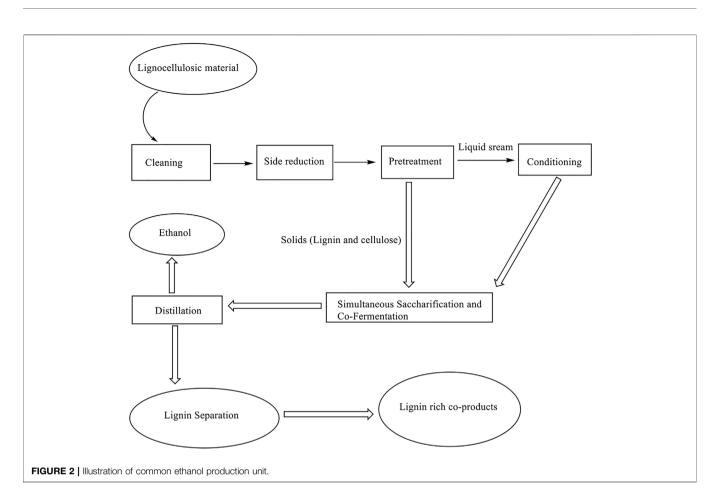
identified the pretreatment method of three-stage alkaline-acidenzymes as the rate-limiting step for bioethanol production. An optimum yield for reducing sugar was reported to be 604.96 mg  $g^{-1}$  cornstalk. Cornstalk juice has also been reported in bioethanol production using *Saccharomyces cerevisiae* (Bautista et al., 2019).

Spent grain from the brewery industry contains lignocellulose (Patel et al., 2018) that may be useful in biofuel production. It is the most abundant by-product emanating from the brewing industry (Kavalopoulos et al., 2021). The composition of brewery spent grain varies according to the process it is subjected to and the type of additive added to the production process. However, it contains mainly cellulose and hemicellulose. A study (Xiros and Christakopoulos, 2009) reported enhancing bioethanol production from brewer's spent grain using Fusarium oxysporum. The influence of temperature and phosphoric acid concentration on brewer's spent grain treatment in preparation for bioethanol production has been reported (Rojas-Chamorro et al., 2018). The study revealed that phosphoric acid as a pretreatment agent enhanced sugar production and subsequently good yield of ethanol. It has become evident that brewer's spent grain is a feedstock for producing an easily hydrolysable simple sugar for the production of bioethanol. The main parameters and mechanism of drying kinetics of brewer's spent grain with the capacity to produce biodiesel were reported (Mallen and Najdanovic-Visak, 2018); the authors used acid catalyst in the transesterification. The study revealed a new approach towards the volarisation of brewer's spent grain. A previous study reported bioethanol production from brewer's spent grains using consolidated bioprocessing (Wilkinson et al., 2017). The process recognized Aspergillus oryzae and Saccharomyces cerevisiae NCYC479 as outstanding fungi that gave the highest ethanol yield, indicating that 1 ton of brewer's spent grain yields 94 kg of ethanol.

Agro-industrial residues are examples of lignocellulosic materials; different types of miscellaneous agro-industrial residues have been tested for their potentials as feedstock for biofuel. They are majorly solid wastes referred to as secondgeneration feedstocks for biofuel and are primarily composed of polysaccharides (Nissilä et al., 2014; Gottumukkala et al., 2016; Sadh et al., 2018). Agriculturally based industries produce a large volume of wastes every day that can be of immense benefit. If these wastes are released directly into the environment without being treated, they may become pollutants. They are underutilized; therefore, it is vital to find a way of making the best use of them. They may be sugarcane bagasse, cassava peel, groundnut oil cake, coconut oil cake, stem-bark, fruit peels, sawdust and vegetable leftovers. Sludge generated from pulp and paper industries may also fall into this category. They are rich in cellulose fibers. Most indigenous local industries in developing countries produce large volumes of waste that may find useful applications in biofuel production. With the provision of appropriate technology, making use of these wastes will generate energy and provide employment that will help reduce the poverty rate. Some selected biofuel companies in developing countries are shown in Table 2. Lignocellulosic waste remains cheap source of feedstock for biofuel production, however, they are challenged with their recalcitrant content like the presence of lignin which needs to be removed. Therefore, necessitating the need for a pretreatment of the waste before conversion to biofuel. It is important to select the most appropriate pretreatment to avoid production of side products that may interfere or detrimental to the growth of the microorganism involved in the bioprocessing of lignocellulosic material to biofuel.

## Treatment of Lignocellulosic Materials for Biofuel Production

Treatment of lignocellulosic materials is vital in preparedness for biofuel production. This pretreatment stage is the most energyintensive and expensive stage in the convention of lignocellulosic materials to biofuel (Ab Rasid et al., 2021). The pretreatment of lignocellulosic materials may result in physical or chemical changes causing changes in particle size and porosity to enhance penetration of solvent and catalyst. Such treatment may affect cellulose crystallinity, linkage among lignin, cellulose and hemicellulose, and the degree of polymerization. Treatment cost is one of the significant factors that determine the type of treatment used; moreover, the nature of chemical reagent and lignocellulose material plays an important role as well in determining the treatment to use. In order to minimize the challenges from chemical side products, the current production of bioethanol considers the use of biological treatment, which is a three-step process involving pretreatment of lignocellulose feedstock, hydrolysis of sugar polymer such as cellulose and hemicellulose and fermentation of sugar; this is described in Figure 2. During pretreatment, the aim is to separate lignocellulose into its major components, cellulose, hemicellulose, and lignin to achieve conversion to biofuel easily. Single-step pretreatment such as pyrolysis may not be efficient due to the numerous molecules present in lignocellulosic materials. The pretreatment step will require more than one step in most cases. The pretreatment methods may be categorized as; physical, chemical, mechanical and biological treatments; however, the various methods may be combined (Barakat et al., 2013) to achieve high efficiency. Previously reported methods include hydrolysis, solubility, steam explosion, ammonia fiber explosion (AFEX), and fractionation in order to separate cellulose, hemicellulose and



lignin (Saha, 2005; Taherzadeh and Karimi, 2008); the essence of this is to make the conversion process as simple as possible; however, despite the success attained with the reported methods, the methods still suffer from certain limitations such as long residence time, generation of toxic residues, and high power consumption.

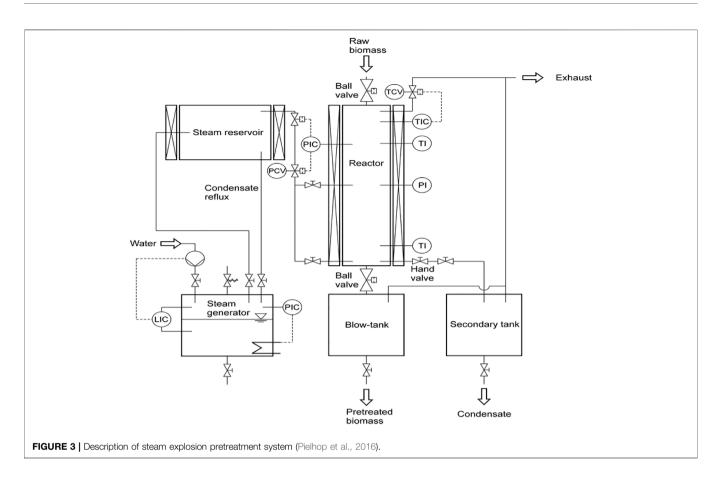
### **Physical Pretreatment**

This may also be referred to as mechanical treatment. It involves size reduction treatments that do not change the lignocellulosic waste composition but change size, shape, and bulk density. Methods employed include cutting, milling, grinding, extrusion and irradiation. In most cases, physical treatment is applied first in preparation for other treatment processes. The treatment helps increase the available surface area of lignocellulose for interaction. When the milling method is used in size reduction, the relationship between the final size obtained and milling energy is not linear; moreover, the type of lignocellulosic material, initial material size, and moisture content influence the milling energy required (Liu et al., 2020). The energy eventually relies on the final particle size and the magnitude of crystallinity reduction. Although the treatment enhances reactivity, the physical structure is altered, which is the case for cellulose where crystallinity is reduced, making it susceptible to chemical attack; when this happens, the cellulose becomes more amorphous, favouring hydrolysis at low

temperature (Zhao et al., 2006; Yu and Wu, 2010). Reduction in crystallinity and particle size is determined by the type of lignocellulosic material, milling method and process time (Kumar et al., 2017). A study on the pyrolysis of wheat straw using hammer-milling and rod-milling as pretreatment steps showed a significant reduction in particle size and crystallinity of the biomass, which led to high surface area and pore volume of the wheat straw (Bai et al., 2018). Recently, a combination of physical treatment and catalysis known as mechanocatalysis has gained wide attention. A good example of this is the milling of lignocellulose followed by the use of a catalyst to enhance conversion rate. A study on the effect of planetary ball milling on wood fibre gave rise to high xylose and glucose via enzymatic hydrolysis (Gu et al., 2018). Previous authors have shown that the inclusion of mineral acid to cellulose agitation in a mill aids hydrolysis to glucose (Hick et al., 2010; Meine et al., 2012; Shrotri et al., 2013) which can be fermented for biofuel production. However, this is not a direct route process; in most cases, mechanocatalysis leads to the production of a composite of depolymerized materials (Carrasquillo-Flores et al., 2013), which may promote the removal of hemicellulose and lignin.

### **Steam Explosion**

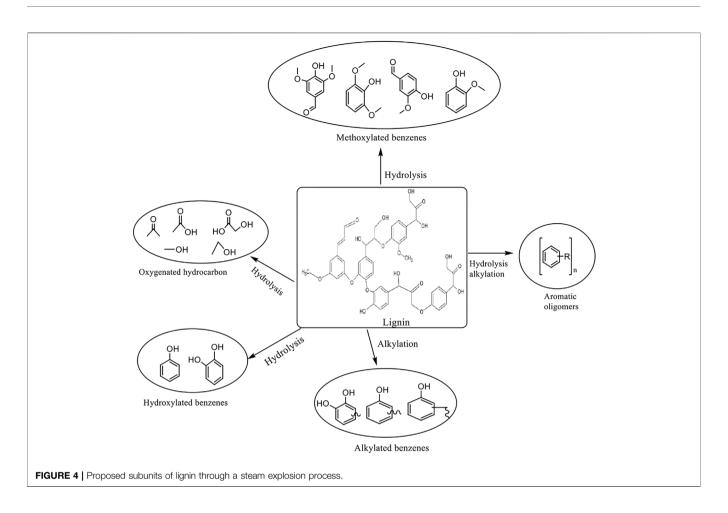
The lignocellulosic material is treated with high-pressure saturated steam to disrupt bonding between polymeric components in order to break the lignocellulose structure, as



shown in Figure 3. It is a physicochemical process that is initiated at a temperature of about 433-533 K and pressure of 0.69-4.83 MPa; the process is brought to an end by an explosive decomposition of the material. The observed 'explosion' is due to the depressurization and cooling of the biomass, which causes the water in the biomass to explode. The process enhances cellulose hydrolysis by degrading hemicellulose and lignin (Varga et al., 2003; Ruiz et al., 2006). The process is catalyzed by hot water or organic acids released via hemicellulose degradation; in some cases, an acid catalyst such as sulphuric acid may be added to boost the hydrolysis and help reduce sugar degradation (Weil et al., 1997; Stenberg et al., 1998). The inclusion of acid, referred to as acid-catalyzed steam explosion, improves enzymatic hydrolysis of cellulose to promote the removal of hemicellulose. The process is relatively green as it avoids the use of organic solvent and corrosive chemicals, making it of industrial value. Although the method is effective, it cannot wholly fractionate lignocellulose as lignin particles can backagglomerate; some of the possible fractions of chemicals that may arise from lignin as a result of the treatment are presented in Figure 4. Furthermore, sugar obtained can also degrade during the process (Donaldson et al., 1988), leading to the formation of inhibitory by-products that may retard the process. When compared with a mechanical or physical process, steam explosion requires less energy for actualization, and it is most effective for agro-lignocellulosic materials. A study (Li et al., 2011) has shown that pit membrane and the parenchyma of wood is damaged by steam explosion improving permeability. A study on Abies lasiocarpa showed that five kinds of changes could take place in the aspirated pit; these changes may include distortion and depression of the pit membrane, cracking of pit border and cell wall, and breaking of pit torus. A report on the steam explosion treatment of softwood showed that the process could enhance the enzymatic digestibility of lignocellulosic materials (Pielhop et al., 2016).

#### **Ammonia Recycle Percolation**

Ammonia fiber treatment is a physicochemical process that is similar to steam explosion treatment. It uses ammonia at high pressure (2-3 MPa) and moderate temperature (333-373 K). The ammonia treatment is AFEX and ammonia recycle percolation (ARP). During the process, lignocellulose is exposed to ammonia over a period of time. The dosage is about 1-2 kg of liquid ammonia per kg of lignocellulosic material; AFEX significantly improves fermentation rate (Zheng et al., 2009) but with low solubility of hemicellulose. However, improved hydrolysis of hemicellulose and cellulose is possible. Care must be taken to recover ammonia vapour from the environment (Sun and Cheng, 2002). The approach is green and improves the surface area, digestibility and enzyme accessibility (Kamiya et al., 2008). The ammonia can be recovered and reused. Unfortunately, the process does not remove hemicellulose significantly and may reduce final sugar yield. The process is not efficient for lignocellulosic material



with high lignin content (Taherzadeh and Karimi, 2008). During the ARP process, aqueous ammonia is passed through the lignocellulosic material at a high temperature of about 423–443 K (Galbe and Zacchi, 2007). The ammonia reacts with lignin causing depolymerization and cleavage of lignin-carbohydrate linkages. A study on silver grass revealed that moisture content of lignocellulosic material plays an essential role in AFEX (Lee and Kuan, 2015). Mainly ARP makes use of aqueous ammonia solution while AFEX relies on liquid ammonia. Both processes do not produce process inhibitors and reduce lignin and hemicellulose with improved cellulose hydrolysis (Sun and Cheng, 2002). The energy required for ARP is higher than that of AFEX due to the high temperature and longer contact time of the process.

#### **Chemical Pretreatments**

Chemical pretreatment aims at removing hemicellulose and lignin in order to improve the biodegradability of cellulose. Presently, supercritical  $CO_2$  fluid is commonly used in the treatment of wood.

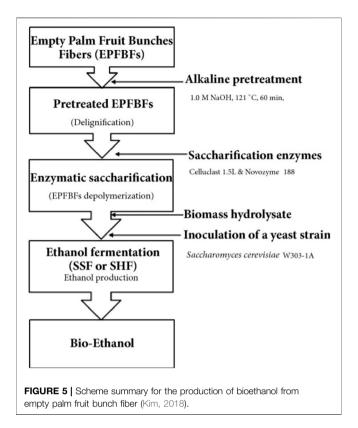
#### **Acid Pretreatment**

During acid pretreatment, the rigid structure of lignocellulosic materials is broken into simple sugar, which can be fermented into biofuel. The process leads to the disruption of intra and intermolecular forces holding the components of the lignocellulosic materials together, which results in the reduction of cellulose and solubilization of hemicellulose (Li et al., 2010). Acid pretreatment can be achieved using either concentrated acids at low temperatures or dilute acids at high temperatures. Using concentrated acids at low temperatures increases the sugar conversion rate; however, the concentrated acids come with disadvantages of being corrosive, toxic and increase process cost due to additional maintenance cost; apart from this, it might lead to undesired products from cellulose degradation. Dilute acid is sometimes used because it permits high reaction rates with improved cellulose hydrolysis; however, lignin hardly solubilizes in either dilute or strong acid concentration, but it gets disrupted to a great extent to allow for easy access to cellulose hydrolysis. Therefore, many industrial processes would prefer the use of dilute acids. Several methods have been developed using inorganic and organic acids. Authors have reported the use of phosphoric acid on wheat bran in preparation for enzymatic hydrolysis and, subsequently ethanol production (Nair et al., 2015). Corn stover has been pretreated with nitric acid for the production of bioethanol (Kim et al., 2015). Formic acid has also been reportedly used in the hydrolysis pretreatment of functional cellulose nanofibrils (Du et al., 2016); oxalic acid (Jeong and Lee, 2016) and maleic acid (Jung et al., 2015) have also been reported as chemical

pretreatment agents for lignocellulosic materials in biofuel production. It is very important to understand reaction kinetics and the nature of the lignocellulosic material in order to design, configure, and operate a suitable reactor for acid pretreatment. Previous studies showed that hemicellulose is converted to xylose by a first-order reaction; complete conversion of lignocellulose materials with low and high furfural is a two-stage process. The first stage involves slow hydrolysis of hemicellulose with concentrated acid producing hemicellulosic sugars while the second stage is the fast hydrolysis of hemicellulose and hydrolysis of cellulose to glucose (Amin et al., 2017). A study on wild rice grass compared the use of dilute acid (H<sub>2</sub>SO<sub>4</sub>) and alkali (NaOH) pretreatment on enzymatic hydrolysis of biomass (Sahoo et al., 2018). The results revealed and supported the feasibility of dilute acid treatment over alkali treatment. Several other studies have also supported the use of H<sub>2</sub>SO<sub>4</sub> as a suitable treatment agent in acid pretreatment (Sindhu et al., 2014; Kärcher et al., 2015).

#### Alkali Treatment

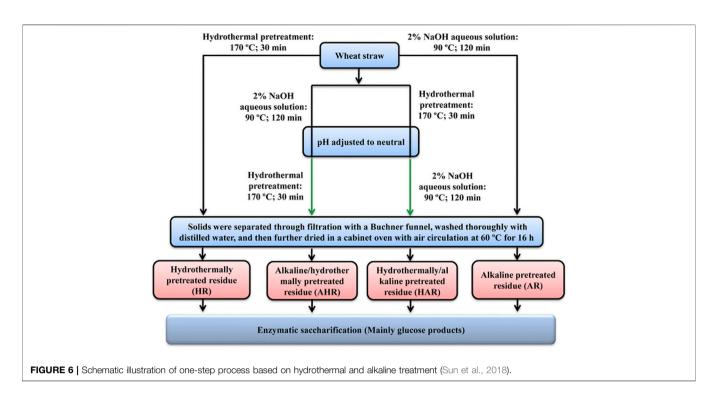
This treatment is commonly used in the solubilization of lignin. It causes the breakdown of ester and glycosidic side chains in the structure of lignin; the cellulose becomes swollen with a reduction in the degree of crystallinity while hemicellulose is solubilized (Sills and Gossett, 2011; Badiei et al., 2014; Behera et al., 2014; Maurya et al., 2015). Alkaline solutions used include calcium, ammonium, sodium and potassium hydroxide. However, sodium hydroxide is commonly used, and it is the most frequently reported and the most effective (Zhu et al., 2010; Kim et al., 2016); however, its usage leads to the production of methane gas which is unwanted and, as a result limiting the use of sodium hydroxide on an industrial scale (Zheng et al., 2009). This concern has shifted attention towards the use of potassium hydroxide (Jaffar et al., 2016) as well as calcium hydroxide (Singh et al., 2015). Studies on corn, switchgrass, bagasse, wheat, and rice straw have been documented (Zhu et al., 2010; Satlewal et al., 2018; Zheng et al., 2018; Tan et al., 2021). The use of alkaline leads to saponification reaction, which helps promotes the accessibility of enzymatic interaction (Sun et al., 2016). The process is very effective when the lignocellulosic material contains a low amount of lignin. An author (Kim, 2018) revealed the use of empty palm fruit bunch fibre using alkali pretreatment (NaOH, 0.5-2.0 M) at 121°C and 60 min for bioethanol production. The result revealed a delignification yield in the range of 55.4-56.9%; the fermentation process yielded 21 g  $L^{-1}$  ethanol within 28 h; the overall scheme for the reported process is as shown in Figure 5. Similarly, the effect of alkaline pretreatment has been reported on a cotton stalk (Dimos et al., 2019). Other authors have reported a combined process where alkaline pretreatment and other pretreatment steps are combined for effective and efficient treatment. An example of this is a two-step process based on alkaline and hydrothermal treatment of wheat straw for improved enzymatic saccharification (Sun et al., 2018). The treatment helped improve the production of bioethanol; a schematic illustration of the process is shown in Figure 6. A few studies, like the alkaline pretreatment of rice straw using 1% NaOH (Shetty et al., 2017), have revealed the advantages of the alkali



treatment step by reducing lignin and hemicellulose content. Another study (Shen et al., 2017) optimized the conditions applied for the pretreatment step in converting vinegar residue to biofuel.

#### **Biological Treatment**

Biological treatment of lignocellulosic materials is conducted to degrade the materials; the process may involve fungal pretreatment, enzymatic hydrolysis and aeration (Rafieenia et al., 2018). The fungi secrete the enzymes which are involved in the hydrolysis of cellulosic compounds to simple fermentable sugars. A few enzymes such as xylanase, hemicellulase, α-amylase, and arabinose have been identified that can be used directly to achieve hydrolysis (Cui et al., 2010; Zhao et al., 2012). The bioprocess is safe and environmentally friendly (Okano et al., 2005; Pan et al., 2006). The enzymes required for the hydrolysis are mostly fungi produced; they may be classed as soft, brown and white-rot fungi. However, the most studies and promising among the white fungi class is the Basidiomycetes. It has been shown that few of the white-rot fungi can degrade polysaccharides and lignin while others will selectively degrade lignin; in whichever process that a particular white-rot fungi displays, two extracellular enzymatic systems may be involved, which are hydrolytic and ligninolytic systems. The hydrolytic system describes the degradation of cellulose and hemicellulose, while the ligninolytic system describes the depolymerization of lignin (Vats et al., 2013; Wagner et al., 2018). During pretreatment, the white-rot is used as a whole-cell organism. Common whiterot fungi reported for the ligninolytic pretreatment are Pleurotus



streatus, Ceriporia lacerate, Ceriporiopsis subvermispora, Pleurotus ostreaus, Gloeophyllum trabeum, Cyathus stercolerus, Postia placenta, Pycnoporus cinnarbarinus, Phanerochaete chrysosporium, and Phlebia subserialis (Kumar et al., 2017; Baruah et al., 2018). Despite the performance reported for white-rot fungi, it still faces the challenge of not being able to adequately degrade lignin because of the carbon-carbon bonds in the lignin polymeric structure (Martinez et al., 2004). One disadvantage of the biological process is the slow rate of the process; apart from this, the microorganisms also consume carbohydrates for metabolic activity. However, most times, the enzymatic biological treatment is carried out in combination with other pretreatment methods in order to achieve high efficiency (Zhao et al., 2012). Other processes have made use of accessory enzymes laccase and xylanases in order to control the amount of enzyme needed to complete the conversion as well as to achieve cost-effectiveness. In fact, a previous study has revealed that the degradability of lignocellulosic material by enzymes depend on the degree of lignification, crystallization, and acetylation of the material (Chang and Holtzapple, 2000). The strength and extent of bonding among lignin, cellulose, and hemicellulose content further affect the hydrolysis conversion process (Tarasov et al., 2018). Deacetylation and delignification are two important processes that require attention in order to achieve complete conversion as these processes may produce substances that may inhibit the conversion process of lignocellulosic material if care is not taken (Chang and Holtzapple, 2000). Therefore, the processes must be monitored for the removal of chemical inhibitors of hydrolysis. In a previous study, the use of Saccharomyces cerevisiae resulted in an alcohol yield of 90%, but it is unable to ferment xylose and arabinose sugars and limited in the operational temperature range (McMillan, 1994; Jørgensen,

2009). Although the study on *Zymomonas mobilis* (bacteria) gave a high yield of alcohol, it has a low tolerance to chemical inhibitors with a neutral pH range (McMillan, 1994). Several authors have reported the use of some microorganisms for biological pretreatment at specified conditions, as shown in **Table 3**. When an adequate pretreatment is given to a lignocellulosic material, the amount of fermentable sugar after saccharification increases, increasing the total process efficiency.

The hydrolysis by enzymatic activity is affected by the level of accessibility of cellulose by cellulase. There is a relationship between enzyme accessed site and rate of hydrolysis. The performance of cellulase is not only affected by lignin but by xylan as well. Despite the fact that the presence of lignin and xylan affects the performance of cellulase, the influence of xylan on the accessibility of cellulase is more than that of lignin (Sindhu et al., 2016). This has indicated the need for the supplementation of accessory enzymes in order to improve efficiency. There has been a study on the isolation and purification of accessory enzymes in order to develop an eco-viable process for efficient hydrolysis to bioethanol. The accessory enzymes act on linkages in plant cell walls; they include galactanase, arabinases, pectinases, and lyases.

Certain factors are known to affect the biological pretreatment process. The nature and type of lignocellulosic waste used plays an important role; this includes the distribution of cellulose, hemicellulose, lignin and other components in the lignocellulosic waste. Constituents like lipid, protein, and salts affect the treatment process. Since the lignocellulosics are biological and mostly plant-sourced, the growth condition and maturation before harvest goes a long way in determining the constituents and distribution of the compositions. The incubation time needed for the treatment is a factor to be considered. Suppose the incubation time is too long, the process cost

Lignocellulosic material	Microorganism	Process effect	References
Wheat straw	Ceriporiopsis subvermispora	Minimal cellulose loss	Cianchetta et al. (2014)
Eucalyptus grandis saw dust	P. ostreatus/P. pulmonarius	Twenty fold increase in hydrolysis	Castoldi et al. (2014)
Straw	Fungal consortium	Seven fold increase in hydrolysis	Taha et al. (2015)
Sorghum husk	Phanerochaete chrysosporium	Increase in reducing sugar production	Waghmare et al. (2018)
Japanese cedar wood	Ceriporiopsis subvermispora	28% lignin removal in initial substrate	Amirta et al. (2006)
Corn stover silage	Phanerochaete chrysosporium	39% lignin removal of initial substrate, improved degradationof substrate cell wall components	Liu et al. (2014)
Yard trimmings	Ceriporiopsis subvermispora	20.9% degradation of initial lignin content	Zhao et al. (2014)
Rice straw	Pleurotus ostreatus and Trichoderma reesei	33% lignin removal of initial substrate with wrf and 23.6% with brf lignin-to-cellulose ratio after treatment: wrf 4.2, brf 2.88	Mustafa et al. (2016)
Rubber wood	Ceriporiopsis subvermispora	Increased fermentation yield	Forough et al. (2013)
Agropyron elongatum	Flammulina velutipes	Improve biodegradation	Lalak et al. (2016)

<b>TABLE 3</b> Different microorganisms reported for biological treatment of lignocellulo	c wastes

increases which, in the long run, discourages the use of biological pretreatment. The incubation time must be short. The optimum process temperature is also a key factor that determines the success of biological pretreatment. The temperature depends on the type of microorganism and lignocellulosic waste used. The higher the temperature, the higher the energy consumption and the overall cost go up. If heat is unnecessarily accumulated, it can have a negative impact on the success of the process. The type of microorganism used affects the process. Several microorganisms have been identified over time. Fungi and bacteria have been used. Many studies have revealed that the use of a combination of fungi can help give an effective result. The particle size of the lignocellulosic material and process pH are essential. The smaller the particle size, the larger the available surface for interaction for the pretreatment. The penetration of the fungi is limited when the particle size is large, which further prevents air and water diffusion as well as other metabolite products into the lignocellulosic particles. The pH is an important factor in cultivating the fungi; report has shown that white-rot performs well in pH range of 4-5 (Sindhu et al., 2016).

#### Saccharification System for Converting Cellulose and Hemicellulose to Bioethanol

Before the conversion of cellulose and hemicellulose to bioethanol, pretreatment of lignocellulosic waste is an essential step in the production of bioethanol from lignocellulosic waste to improve the accessibility of degrading enzymes to cellulose and hemicellulose. The effect is to improve the yield from saccharification. The pretreatment step also further helped reduce the cost of production, although a few authors reported the pretreatment step to be one of the most expensive steps in converting cellulose to bioethanol (Alvira et al., 2010; Kim et al., 2011). Some of these pretreatment steps have been mentioned above, however, more sophisticated and advance processes might involve using an advanced fractionation method to obtain a highly purified bioethanol as a high-grade biofuel. The use of acidified zinc chloride in effective fractionation has been reported (Yoo et al., 2012; Kim et al., 2013), which is effectively designed to separate hemicellulose units containing xylan. The use of zinc

chloride showed many advantages, such as being very effective in enhancing enzymatic digestion of cellulosic waste, which promotes the ease of fermenting the wastes to ethanol (Yoo et al., 2012). A study (Kim et al., 2014) reported hemicellulose hydrolysis from barley straw and enzymatic saccharification. The study evaluated the effect of process parameters such as the effect of reaction time, temperature and concentration of zinc chloride on the conversion process. The study further revealed that the use of zinc chloride improves the solubility of hemicellulose and enhances the enzymatic digestion of treated cellulose.

The saccharification involves the use of cellulases (for cellulose) and hemicellulases (for hemicellulose) in enzymatic hydrolysis to degrade polysaccharides (hemicellulose and cellulose) to fermentable sugars. It is obvious that cellulase plays an essential role in catalyzing the process. The catalase may be sourced from plants, fungi, and bacteria; based on structural properties, three key types of cellulase activities are involved in the hydrolysis of cellulose, including endoglucanases and exoglucanases and  $\beta$ -glucosidases (Menon and Rao, 2012). Cellulose hydrolysis may involve individual enzymes or a multi enzymatic process with modular structure in both cases. The enzymes have a catalytic site where the hydrolytic reaction occurs and a domain where it binds to the cellulose; both the hydrolytic site and the cellulose-binding domain of the enzyme are linked by a peptide (Carrard et al., 2000). The best operating condition reported for most cellulase is pH value within the range 4.0-5.0 and temperature of 50°C (Kristensen, 2008; Świątek et al., 2014; Walker et al., 2018). Depending on the residue sugar, hemicellulose can be classified into two groups, with the common groups being xylan and mannan. Since the presence of hemicellulose shields access to cellulose, its removal reduced the amount of cellulase required for the hydrolysis of cellulose. Xylan's main polysaccharide chain is degraded by endoxylanases including into monomers, smaller β-xylosidase and that oligosaccharides finally separates into xylose. Furthermore, mannan can be degraded by  $\beta$ -mannosidase and endomannanase. Understanding the structural mode by which the enzyme binds to the polysaccharides can really help in having insight into the mechanism of saccharification. Knowledge of the mechanism guides in designing the most appropriate approach or condition for the process. Although it is not as simple as it may

seem due to the inherent homogeneity and complexity of cellulose; however, understanding the molecular mechanism behind cellulose and hemicellulose degradation is an important factors in designing the most appropriate approach for saccharification process (Louime et al., 2011).

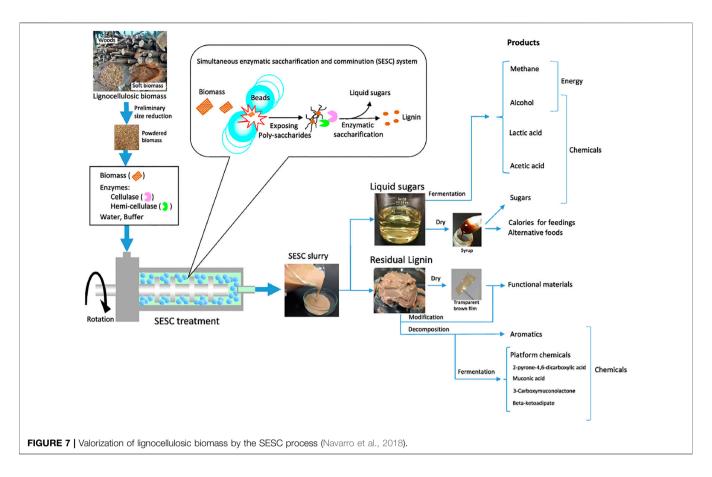
Trichoderma reesei is the most common source of cellulases; however, enzymes from Trichoderma reesei has been reported to exhibit low-glucosidase activity leading to incomplete hydrolysis of cellobiose (Holtzapple et al., 1992). It is important that the required cellulase should be highly active on the lignocellulosic waste for hydrolysis (Knauf and Moniruzzaman, 2004). Other studies have also been reported that involved the use of acid (Hamelinck et al., 2005); a major disadvantage of this is the fact that the process requires a high amount of acid, which makes it less attractive. It mostly leads to chemical dehydration of monosaccharides, causing the formation of other undesired products in biofuel produced. Hemicellulases contain catalytic and non-catalytic modules. The catalytic modules are either carbohydrate esterases or glycoside hydrolases. The noncatalytic modules contain the carbohydrate-binding domains, which promotes the binding of the catalytic domain to either the enzymatic complexes or microbial cell surface (Shallom and Shoham, 2003; Menon and Rao, 2012). Several microorganisms have been studied as alternative sources of cellulase for hydrolysis of cellulose; some of the reported organisms for this purpose include Candida tropicalis, Trichoderma harzianum, Aspergillus niger, Fusarium oxysporum, and Trichoderma sp. (Mattam et al., 2016; Mezule et al., 2016; Mkhize et al., 2016; Mishra and Jana, 2017). Recently, some authors (Hwangbo et al., 2019) reported a one-step saccharification of pretreated lignocellulosic biomass using immobilized magnetite biocatalyst system. The authors reported five saccharifying enzymes namely endo-1,4-βxylanase (an endoxylanase, xynC), endo-1,4-β-D-glucanase (an endoglucanase, eglS), β-xylosidase (bxlB) and cellobiohydrolase (an exoglucanase, cbhA), and β-glucosidase (bglH). These enzymes performed better than commonly used enzymes with better reusability. It has also become important to optimize saccharification conditions of lignocellulosic biomass for biofuel production. A recent study (Kucharska et al., 2020) revealed the use of Box-Behnken design as a promising approach for achieving this. Furthermore, other authors (Navarro et al., 2018) applied simultaneous enzymatic saccharification and comminution on lignocellulosic biomass (Figure 7) as a promising approach towards biofuel conversion, the approach achieved a saccharification efficiency of 80% at pH six using an energy consumption of 50 M.J. kg<sup>-</sup> biomass.

#### **Fermentation Strategies**

Saccharification is followed by fermentation which involves degrading the sugars in the presence of microorganisms into alcohols and other end products. *Saccharomyces cerevisiae* is the commonly used yeast that secrets enzyme that converts the sugars to ethanol at a temperature of about 30°C under anaerobic conditions with the generation of  $CO_2$  and N-based compounds as by-products. Despite the achievement with the use of *Saccharomyces cerevisiae*, it is unable to ferment pentose

sugar. Presently, research is focused on developing a microorganism system that ferments pentose and hexose sugars. Over time, two major approaches are known for the conversion of sugar to ethanol, namely, simultaneous saccharification and fermentation (SSF) and separate hydrolysis and fermentation (SHF). During SHF, the microorganism is added at the end of the hydrolysis in separate units. The process has shown some inherent limitations over time, including the formation of process inhibitors, contamination, need for co-equipment, and time consumption. In SSF, the hydrolysis and the enzymatic fermentation takes place in the same unit and simultaneously, which is more effective than SHF. However, in SHF, the inhibitory effect is neutralized; enzymes such as Kluveryomyces, Saccharomyces, Zymomanas, Pichia and Debaryomyces as well as their recombinants, have been reported (Philippidis, 2018). SSF has proved to be a better approach for lignocellulosic biomass (Kim and Holtzapple, 2006). The success of SSF depends on compatibility with process parameters such as temperature, pH, substrate concentration, kinetic, etc. (Doran-Peterson et al., 2010). One major advantage with the use of SHF is that each step involved in the process can be achieved at its optimal conditions to attain high product recovery (Dey et al., 2020). The process may require dilute media, which makes the product to be of low concentration and may require high enzyme dosage as well. Some level of freedom exists in SHF because the processes are carried out in separate vessels; however, this incurs additional production cost, which is not the case in SSF; although this additional production cost may be substantiated. Due to the level of purity that may be attained in SHF as a result of the removal of solid fractions after the enzymatic hydrolysis, the yeast may be recycled and reused, which is not possible in SSF. Therefore, in SSF, the yeast has to be prepared for each batch process, which suggests that sufficient Saccharomyces cerevisiae will have to be cultivated for the fedbatch mode to prevent low yield due to glucose repression; apart from this, care must be taken to avoid inhibition by the medium. Conclusively, the choice between SSF and SHF will depend on factors like product size, desired product spectrum in the biorefinery, and feedstock quality.

Apart from SHF and SSF, the use of simultaneous saccharification and co-fermentation (SSCF) and consolidated biomass processing (CBP) have been developed. There has been tremedous progress in the use of SSCF by developing an appropriate microorganism system for xylose fermentation. Genetically modified microorganisms with different approaches have been developed (Manzanares, 2010). SSCF is recognized as a possible means for ethanol production from xylose-rich lignocellulosic wastes with a genetically engineered strain of Saccharomyces cerevisiae (Olofsson et al., 2008; Bertilsson et al., 2009). A study (Olofsson et al., 2010) using SSCF increased ethanol yield from 40 to 50%. SSCF of steamexploded corn stover has been compared with other conversion processes (Liu and Chen, 2016). The study revealed the yield, concentration, and productivity of ethanol to be 90.0%, 34.3 g  $L^{-1}$ , and 2.61 g  $L^{-1}$  h<sup>-1</sup>, respectively, for the co-fermentation of 60 g L<sup>-1</sup> glucose and 10 g L<sup>-1</sup> xylose by Saccharomyces cerevisiae

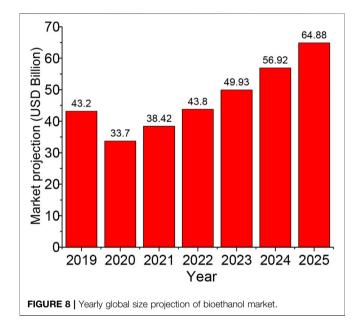


IPE003. It was concluded that SSCF increased ethanol production and can serve as an efficient lignocellulosic waste conversion process to ethanol. Another report (Sekhon et al., 2018) further revealed that the production of ethanol from insoluble fiber doubled when SSCF with S. cerevisiae and E. coli KO11 was used as a conversion process. The three major steps involved in the lignocellulosic biomass conversion (enzymes production, biological hydrolysis of biomass to sugars and oligomers, and fermentative metabolites production) can be combined into a single process known as CBP. The process makes use of a single type of microorganism in a single reactor via a biomass-tobioethanol conversion technology. The focus of CBP research has been on developing a more efficient microorganism for a well-designed process condition; this has led to the development of several mesophilic/thermophilic and cellulolytic/noncellulolytic microorganisms. These microorganisms include the use of Thermoanaerobacterium saccharolyticum, Clostridium thermocellum, Caldicellulosiruptor bescii, Clostridium phytofermentans, S. cerevisiae, K. marxianus, etc. (Shaw et al., 2008; Shao et al., 2011; Jin et al., 2012; Yamada et al., 2013; Chung et al., 2014). Genetically modified strains of microorganisms have also been developed for this purpose. Thermophilic microorganisms have the advantage of higher hydrolysis rate and less susceptibility to contaminations at high temperatures than the mesophilic group that are limited with high operation temperature (Olson et al., 2012). Recently, a study (Cunha et al., 2020) reported the use of engineered industrial Saccharomyces cerevisiae strains (whole-cell biocatalysts) as a promising commercial enzymatic cocktail for the production of ethanol from corn cob biomass. Other authors have also documented a one-pot CBP as a potential means for techno-economic feasibility of lignocellulosic-bioethanol-fuel production (Vaid et al., 2018). The authors optimized bioethanol production from pine needle biomass via single pot in-situ ionic liquid-based pretreatment and enzymatic saccharification. The adopted process produced a maximum ethanol yield of  $0.148 \text{ g s}^{-1}$  pine needle biomass in 72 h resulting in an efficiency of 41.39%. Table 4 shows bioethanol yield from some selected lignocellulosic materials. Sugarcane bagasse and switch grass are good sources of bioethanol which should be better made use of. Large amount of these are continuously generated from both local and international industries which can be used as sources of cheap raw materials for bioethanol production.

Optimization of fermentation condition is very crucial in order to obtain the best of yield and product. Obviously, the type of lignocellulose source, fermentation time, yeast amount are key conditions that should not be underplayed. Apart from ensuring an efficient pretreatment step, it is important to also use an efficient recovery process. Most times, there are challenges making use of appropriate recovery process. Recently, there has been attention on bioethanol recovery study in order to improve on the utilization of lignocelluloses. Gas stripping from fermentation broth has been studied to improve production yield. The use of gas

TABLE 4 | Bioethanol yield from selected underutilized lignocellulose materials.

Lignocellulose material	Biethanol yield (L t <sup>-1</sup> )	References
Sugar cane	70–75	Organization, (2008)
Cassava	137–180	Organization, (2008)
Sugarcane bagasse	318–500	(Nwakaire et al., 2013; Corbin et al.,
		2015)
Hardwood sawdust	381	Corbin et al. (2015)
Corn grain	470	Corbin et al. (2015)
Switch grass	392-457	Corbin et al. (2015)
Brassica carinata	128	Chin and H'ng, (2013)
Wheat straw	140	Chin and H'ng, (2013)



stripping has been shown to improve production efficiency (Bušić et al., 2018). Adsorption has also been introduced as a surface phenomenon for effective separation and recovery of ethanol in order to improve yield; which has led to the production of several adsorbents. The process takes place in two stages, the loading stage (adsorption) and the discharge stage (desorption). Application of adsorption as an efficient bioethanol recovery process has focused on the synthesis of cheap and affordable materials as adsorbents with excellent adsorption and desorption capacities. It is important that such adsorbent is easily regenerated for continuous usage. Silicalite-1, activated carbon, zeolite and sepabeads207 have been well used in the recovery of ethanol from broth (Beden et al., 2017; Zentou et al., 2019; Sahlan et al., 2021). Present studies are investigating the use of low-cost biomass adsorbents to achieve this purpose. In fact, other waste generated during the pretreatment and recovery stages can be converted to other chemical products which makes the use of lignocellulose a winwin process.

## Challenges, Prospects and Cost Evaluation of Biofuel Production

The focus of most studies on lignocellulosic waste to bioethanol has been on how to commercialize the product, meaning from cellulose to ethanol technology. As at present, commercialization is still expensive. Most economies in poor developing countries may not be able to sustain the current cost of setting up and running a biofuel plant; however, the governments in some of these countries are trying to help private investors by providing subsidies, tax waivers and agricultural funds to encourage young farmers. The effort is still ongoing to prepare the most efficient and cost-effective technique for biofuel production. The previous global report revealed bioethanol production (7.25 billion litres in 2000 to over 46 billion litres in 2007) to be on the increase (Balat, 2007). However, with market size of USD 43.2 billion in 2019, the global market size projection was estimated at a growth of USD 33.7 billion in 2020 to USD 64.8 billion by 2025 with an expectation of a compound annual growth rate of 14.0% from 2020 to 2025 (Pattnaik et al., 2021) as shown in Figure 8. The present growth was achieved because of the mandatory usage compelled in many developed countries. On the other hand, the situation is different in many developing countries, most especially those countries in Africa, due to poor policy, lack of support and inconsistent expression of willpower by the government. Although countries like South Africa, Malawi, Zimbabwe, Nigeria, and Kenya have made impacts but the numerous challenges such as land availability, food security, high poverty rate and inadequate government policies still remain major hurdles. China and Brazil are unique exceptions; bioethanol has gained profound usage as blends in the two countries as the government encouraged bioethanol producers with tax reduction and waivers. The feedstock varies in developing countries; despite that, about 25 million hectares of land are available for maize cultivation in Sub-Saharan Africa; the use of its waste in bioethanol production is limited due to poor technology development and low investment. Although the biggest producers of maize and maize waste are South Africa, Nigeria, Tanzania and Kenya, these countries do not have operational maize-fed ethanol plants, and this is the current situation in Sub-Saharan Africa except in South Africa, where this is ongoing. The use of sorghum stalks is increasingly encouraged as a feedstock for bioethanol, most especially in the Philippines, USA, and India but not in Africa, although Africa has large spans of land for sweet sorghum production. Nigeria remains the biggest sorghum producer in Sub-Sahara Africa, estimating for about 39% of the total production, followed by Sudan, which accounts for about 21%, not excluding Burkina Faso and Ethiopia. Nigeria has considered a large-scale production of sorghum for biofuel production through its Nigerian National Petroleum Corporation with a dedication of about 10,000 ha of land with an estimated production of about 1.5 million litres per day. Although cassava peel is considered for biofuel production, its production rate is low in most developing countries. With the construction of the cassava

bioethanol plant in Brazil, Nigeria being the largest producer of cassava in Africa, is known for the production of about 50-2000 L of ethanol per day from cassava, although this is in a small processing unit (Ohimain, 2013; Adewuyi, 2020). Despite the expansion expressed in sugarcane in Sub-Saharan countries, there are challenges with the wetlands, leading to less space for natural habitat, livestock, fisheries and other agricultural practices. This is a pointer for the need for check and balancing. It has become apparent that there could be a decline in water quality due to the water required to grow the crops; this was one of the previous challenges encountered in Brazil by the sugarcane growers. It is expected that irrigation may meet up the shortfall, but with climate change, rainfall is either not sufficient or sufficiently continuous, creating a control challenge in planning and designing efficient irrigation. The decline in water quality may also be in the form of sedimentation and eutrophication, nutrient leaching, and handling of waste generated from ethanol production may also be of concern.

International energy agency has indicated that biofuel can provide 27% of the total transport fuel by 2050 with the capacity to replace kerosene, jet fuel, and diesel. Due to the high cost of enzymes process conditions, research is focused on developing a cost-effective bio-technique for ethanol production, which is keen on using lignocellulosic material as feedstock. Unfortunately, most reported research works were based on a laboratory scale, indicating the need for large or industrial scale studies to understand better the performance of the newly researched techniques for commercialization. Apart from the benefits that come with the use of biofuel from lignocellulosic materials, studies have highlighted a few negative environmental and socioeconomic challenges that may associate with its use. Some of these may include exacerbation of soil, destabilization of agricultural and forest system, degradation of vegetation and overexploitation of forest (Koh and Ghazoul, 2008; Koul et al., 2011); this has been the case in most developing countries where there are no maintenance and sustenance plans. One major concern is the fear of private investors converting agricultural land meant for food production to biofuel production and may negatively affect the price of food commodities hampering food security (Headey and Fan, 2008). Although the positive aspect of this may include increased soil carbon content, reduced water erosion and wind, and reduced shallow landslides (Börjesson and Berndes, 2006), care still needs to be taken to control biofuel activities. Several other challenges have been identified in developing countries which include poor policy initiatives, market challenges, socioeconomic issues and sustainability. On a global view, developed countries like the United States and European Union have developed advanced techniques for massive food production, which leads to overproduction. Excess of this production is exported to food-importing countries in developing countries at rock bottom prices. Such a move keeps farmers in developing countries out of business as they find it difficult to compete. This negatively impacts agricultural practice, leading to a negative ripple effect on the availability of lignocellulose waste as the farmers quit farming due to the inability to compete in the market. In developing countries

with high population and sufficient land for agriculture, the population is a driving force to drive the biofuel market. This is the case in countries like China, Nigeria, India, and South Africa. However, in countries with a small population, like Tanzania, Uzbekistan, and Azerbaijan, a minor investment in land for farming will only be required for biofuel production to meet the national need. Poor infrastructure development is also a challenge; most developing countries have poor road and railway networks, which pose difficulties on logistics and most times contribute to production costs.

Understanding the production cost of biofuel is an integral part of commercialization feasibility. Previous studies have shown that the permissible cost of enzymes may go for about 15-30 cents/gallon of ethanol; however, when lignocellulosic biomass is used as feedstock, the estimated production cost, which includes the pretreatment stage, maybe about 50-80 cents/gallon of ethanol (Chapple et al., 2007). Other authors at the onset of using lignocellulosic materials, preestimated a market cost of about U.S. \$4 per gallon of ethanol (Louime and Uckelmann, 2008). An effort is required to gain a reduction in the estimated cost but the cost of hydrolysis and fermentation is sometimes complicated depending on the source of lignocellulosic feedstock. The cost was previously predicted to be about U.S.  $0.97 L^{-1}$  using sugarcane as feedstock (Demirbas, 2010). Economic factors to be considered in biofuel production include the cost of raw materials, transportation, labour, plant capacity, maintenance, process design and capital (Karmakar et al., 2010; Cansino et al., 2013). It has been shown that the cost of raw material covers about 75-80% of the entire production cost (Bracco, 2015; Gasparatos et al., 2015; Kousoulidou and Lonza, 2016). Authors have shown that the cost of producing bioethanol on a large-scale is about three times higher than that of fossil fuel in developed countries, on the contrary, the cost is lower in developing countries like those in Africa due to readily available agricultural wastes, large span of land for cultivation, and cheap labour (Abila, 2014; Ben-Iwo et al., 2016; Giovannetti and Ticci, 2016). Biofuel production can improve the trade balance of developing countries by substituting imported refined oils with locally made biofuel making such countries depend less on foreign markets to build a more stable macroeconomic business environment. Encouraging the use of lignocellulosic waste creates an avenue for wealth creation from waste (concept of waste to wealth) by providing job opportunities.

### CONCLUSION

With the current awareness of the environmental challenges of fossil fuel and the demand for neat energy, biofuel is becoming popular, with attention shifting towards lignocellulosic materials as feedstock for biofuel production. From cost consideration and environmental safety, lignocellulosic bioethanol represents a potential pathway to the global provision of renewable and sustainable energy that can suffice the current challenges to biofuel provision in developing countries. Although the high cost of enzyme production has been identified as a challenge to the economic viability of lignocellulosic bioethanol, but there is hope that developing an efficient biosystem for SSF and CBP may help circumvent the challenge. The future of biofuel in developing countries is very bright as there are numerous bioresources that can help make this a reality.

Despite the abundant of natural resources in developing countries, it is obvious that most developing countries are not yet fully strategized to become major producer or exporter of biofuel or its feedstock. Therefore, it is important that government in developing countries should promote policies that can help local producers of biofuel and its feedstock, and not just foreign investors. For better biofuel technology development, it is high time that both private and public sectors invest in science and technology programmes or initiatives that will drive the design and development of techniques to ease biofuel production, this might be inform of

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providing research grants for research-based institutions. Production of lignocellulosic biomass should be encouraged either through contract farming with smallholders or cooperative institutions. In order to make it a win-win process for socio-economic growth, there is need for massive investments in chemical process research, agricultural research and social science research that will develop evidence base policy to help policy makers in developing appropriate guidelines for biofuel production in developing countries.

### **AUTHOR CONTRIBUTIONS**

AA conceived the idea, executed the project and wrote the manuscript.

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