



Quasi-Unidirectional Transport Bilayer Two-Dimensional Nanopores for Highly-Efficient Molecular Sieving

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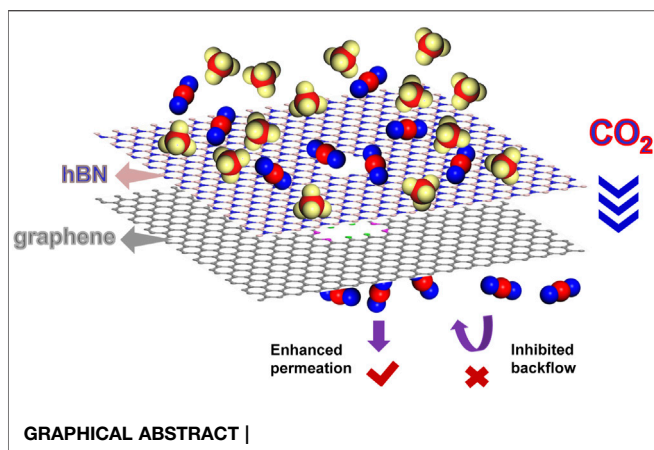
Two-dimensional nanopores are very promising for high-permeance molecular sieving, but the molecular backflow from permeate-side to feed-side is not beneficial for improving molecular permeance. We study the quasi-unidirectional molecular transport through a graphene-hexagonal boron nitride bilayer nanopore, aiming to realize a high-permeance molecular sieving. Molecular dynamics simulations of CO₂/CH₄ separations show that the bilayer pore presents 3.7 times higher selectivity comparing to the single-layer graphene nanopore with the same size. The quasi-unidirectional molecular transport is attributed to the distinctive adsorption abilities of gas molecules on the two sides of bilayer nanopores and the inhibited molecular backflow from permeate-side to feed-side. This work provides a promising way to realize the ultra-permeable porous membranes with molecular permeance even higher than the single-layer atomic-thickness membranes.

Keywords: nanopore, molecular sieve, molecular dynamics, gas separation membrane, nanoporous graphene

1 INTRODUCTION

Graphene (Geim and Novoselov, 2007; Geim, 2009) and its derivatives have been proposed as a perfect candidate for the molecular separation membranes owing to its atomic thickness, because the permeance of membrane is generally inversely proportional to the thickness of membrane materials. Thus, the one-atomic thickness graphene can be a high-efficiency separation membrane with a high molecular permeance (Wen et al., 2015; Sun et al., 2020a). To realize the graphene-based membranes, the selective nanopores should be introduced into the impermeable pristine graphene to generate the nanoporous graphene (NPG) sheets (Koenig et al., 2012). In 2009, Jiang et al. (Jiang et al., 2009) firstly demonstrated that the NPGs can realize the high-efficiency separation of H₂/CH₄ mixtures with a high H₂ permeance of 2.985×10⁹ GPU (1 GPU = 3.35 × 10⁻¹⁰ mol/s m² Pa). Currently, the industrial-scale NPG membranes have been successfully synthesized with a high molecular permeance as the theoretically expected (Boutillier et al., 2017; Wang et al., 2017; Zhao et al., 2019). Meanwhile, the molecular permeation mechanisms through NPG membranes were also well revealed (Drahushuk and Strano, 2012; Sun et al., 2014; Yuan et al., 2017; Sun et al., 2019a; Sun et al., 2021a; Sun et al., 2021b). In short, the NPG membranes for molecular separation has become a reality, especially for the application of gas separation.

For the NPG membranes, the structure of selective nanopores in graphene is strictly symmetrical, accordingly molecules can permeate through the graphene nanopores from both sides in two directions. Unlike other one-way permeable membranes, the molecular permeation through NPG membranes can appear the backflow from permeate-side to feed-side, which would greatly decrease the net permeation rate of NPG membranes. If the backflow can be effectively inhibited or totally eliminated, the permeance of NPG membranes would be further improved.



Hexagonal boron nitride (hBN), another type of two-dimensional material, has become a hot topic of research thanks to its structural equivalency to graphene and its outstanding mechanical (Chen et al., 2015), thermodynamic (Liu and Turner, 2014), and electronic properties (Li et al., 2012). hBN is composed of the heterogeneous boron and nitrogen atoms and thus it would exhibit a stronger molecular adsorption ability comparing to the graphene. Therefore, we adopt hBN and graphene hybrid sheets to form a two-dimensional membrane with stronger adsorption surface in one side and weaker adsorption surface in another side. In this study, we systematically study the quasi-unidirectional molecular transport characteristics through a graphene-hBN bilayer nanopore, following the asymmetric two-layer porous membranes proposed by Liu et al. (Liu et al., 2020). We show that the quasi-unidirectional molecular transport is realized by inhibiting the molecular backflow from permeate-side to feed-side. Through molecular dynamics (MD) simulations of the CO₂/CH₄ separation, it is shown that the permeance of the graphene-hBN bilayer nanopore is higher than that of the monolayer graphene nanopore with the same size. Even more, the selectivity of molecular separation is improved by employing the bilayer two-dimensional nanopores. This enhancement of the molecular permeance is caused by the relatively stronger CO₂ adsorption abilities on the hBN surfaces compared with those on the graphene surface. It is expected that this type of graphene nanopores are promising for high-efficiency membranes for molecular separation as well as other processes involving molecular permeation (Sun et al., 2020b). This work presents a promising road to achieve ultra-permeable porous membranes even with a higher molecular permeability compared to the single-layer membranes. Furthermore, the stacking of multiple atomic layers makes such membranes more feasible for large-area industrial production.

2 SIMULATION MODEL

2.1 Pore Structure

During the molecular permeation, the hBN surface is towards the feed-side while the graphene surface is towards the permeate-side. Thus, the CO₂/CH₄ mixtures are initially arranged in the

hBN side. The graphene-hBN bilayer nanopore is constructed by stacking the graphene and hBN layers together and generating a nanopore with the same size (Figure 1A). To realize a selectivity of the separation of CO₂/CH₄ mixtures, the pores are functionalized by the N and H atoms (Figure 1B), as done in our early work (Sun and Bai, 2017). The N functionalization not only enlarges the size of the pore to achieve a high permeance of CO₂ molecules, but also enhances the adsorption intensities of CO₂ molecules on the surface which further improves the permeability and selectivity. The spacing distance of two layers is set as 3.4 Å in the simulation (Figure 1C).

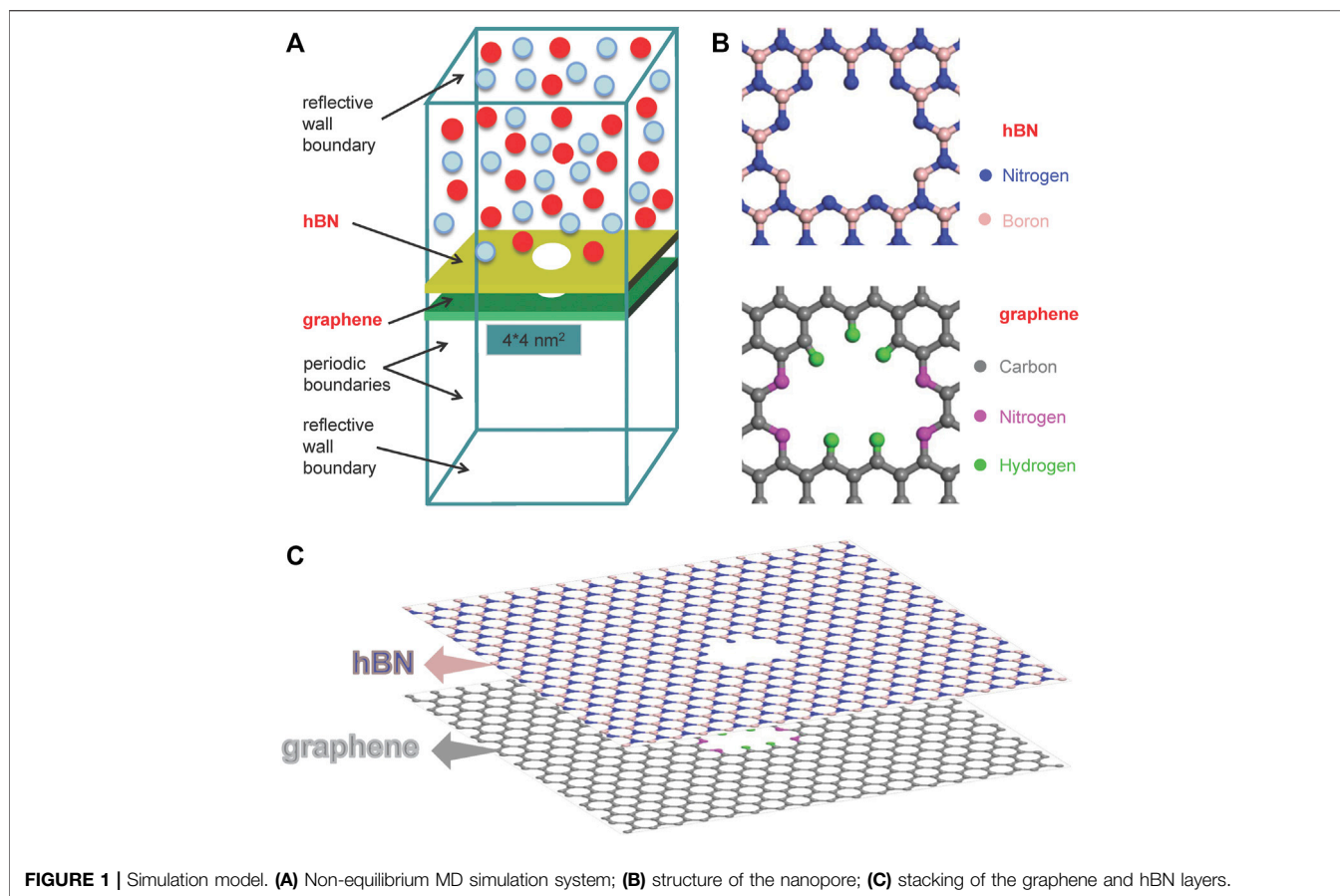
2.2 Simulation System and Method

To examine the sieving effect of nanopores, we perform MD simulations for the separation of CO₂/CH₄ mixtures in a non-equilibrium system based on the platform of LAMMPS. In the system, there are 1,000 mixture gas molecules, namely 500 CO₂ molecules and 500 CH₄ molecules. Initially, the molecules are arranged alternatively and uniformly in the feed side, while the permeate side is kept vacuum, as seen from Figure 1A. The nanopore is located at the center of the simulation box of height 140 nm. The heights of box in the permeate side and feed side are both 66.6 nm. Periodic boundary conditions are applied in the *x*- and *y*-directions (parallel to the surface), while reflective wall condition is applied in the *z*-direction (perpendicular to the surface). The spacing distance among the nanopores in the porous membranes is assigned as 4 nm. In this case, the number density of the pores is 6.25 × 10¹² cm⁻², which is comparable with the pore densities in the fabricated NPG membranes in laboratory (O'Hern et al., 2014). Owing to the application of periodic boundary conditions, the area of sheet is assigned as 4 × 4 nm² based on the spacing distance among nanopores. The simulation is run in a NVT ensemble with a temperature of 350 K, which is close to the running temperature in gas separation industry. A simulation period of 6.6 × 10⁷ timesteps (time step is 0.3 fs) is chosen such that the permeable gases transport through the nanopore with a relatively fast rate, ensuring a high accuracy in the calculation of permeance. Based on the ideal gas equation, the initial pressure of the feed side is estimated as 21.1 bar.

For carbon and hydrogen atoms in graphene sheet and CH₄ molecules, the atomic interactions are modeled by the AIREBO potential model; while for CO₂ molecules and hBN sheets, they are modeled by the Lennard-Jones (LJ) potential model. The detailed introduction and parameters of AIREBO potential can be found in the literature (Stuart et al., 2000). A three-site model with three partial charges is adopted for CO₂ molecules. In the model, the LJ potentials and Coulombic potentials are coupled together, as follows:

$$\phi(r_{ij}) = \begin{cases} 4\epsilon \left[\left(\frac{\sigma}{r_{ij}} \right)^{12} - \left(\frac{\sigma}{r_{ij}} \right)^6 \right] + \frac{Cq_i q_j}{\chi r_{ij}} & (r_{ij} < r_{cut}) \\ 0 & (r_{ij} \geq r_{cut}) \end{cases} \quad (1)$$

where σ is length scale, ϵ is energy scale, q_i and q_j are the charges on atom *i* and atom *j*, C is an energy-conversion constant, and χ is



the dielectric constant. The parameters for cross atoms are estimated by the Lorentz-Berthelot mixing rules based on the following parameters for C-C interactions: $\sigma = 3.40 \text{ \AA}$, $\epsilon = 2.413 \times 10^{-3} \text{ eV}$ (Du et al., 2011), H-H interactions: $\sigma = 2.50 \text{ \AA}$, $\epsilon = 1.300 \times 10^{-3} \text{ eV}$ (Liu et al., 2015), and $N_{\text{graphene}}-N_{\text{graphene}}$ interactions: $\sigma = 3.297 \text{ \AA}$, $\epsilon = 3.126 \times 10^{-3} \text{ eV}$ (Chae and Violi, 2011), B-B interactions: $\sigma = 3.453 \text{ \AA}$, $\epsilon = 4.116 \times 10^{-3} \text{ eV}$ (Baowan and Hill, 2007), $N_{\text{hBN}}-N_{\text{hBN}}$ interactions: $\sigma = 3.365 \text{ \AA}$, $\epsilon = 6.281 \times 10^{-3} \text{ eV}$ (Baowan and Hill, 2007). The bond and angle potentials in the gas molecules (CO_2) and the functionalized groups are all modeled by harmonic potential. The parameters in these potential models can refer to our previous work (Sun and Bai, 2017). The charges on the atoms near the pore are obtained through a density functional theory calculation using the DMol₃ module in Material Studio software.

3 RESULTS AND DISCUSSION

3.1 Molecular Permeance

As seen from **Figure 2A**, the number of permeated molecules increases over time but the increasing rate slows down gradually owing to the non-equilibrium permeation process. The molecular number in the permeate side ($z < -0.6 \text{ nm}$) of permeating molecules sharply increases with the passage of time, while

that of non-permeating molecules increases very slowly and only a few CH_4 molecules appear in the permeate-side. Namely, the graphene-hBN bilayer nanopore exhibits a phenomenon of selective molecular sieving for the separation of CO_2/CH_4 mixtures. Comparing to the single-layer graphene nanopore, the performance of selective molecular sieving of graphene-hBN bilayer nanopore is more significant.

In order to quantitatively analyze the molecular sieving performance of nanopores, the molecular permeance is calculated based on the relationship between the number of permeated molecules N and time t , as follows (Sun and Bai, 2017):

$$N = (250 - N_{\text{al}}/2) \times (1 - e^{-8.24 \times 10^{10} Pt}) \quad (2)$$

where N_{al} is the average number of molecules adsorbed on both sides of graphene, P is molecular permeance, and the constant 250 is related to the molecular number 500 in the simulation box. On the basis of this function, the curves of molecular number N versus time t are fitted and the molecular permeance P is then obtained.

The molecular permeance (P) and selectivity for the separation of CO_2/CH_4 mixtures through the bilayer and single-layer nanopores are displayed in **Figure 2B**. For CO_2/CH_4 , the CO_2 permeance of bilayer nanopore is higher than that of single-layer nanopore while the CH_4 permeance is lower, accordingly a higher selectivity appears for the bilayer nanopore. The bilayer pore

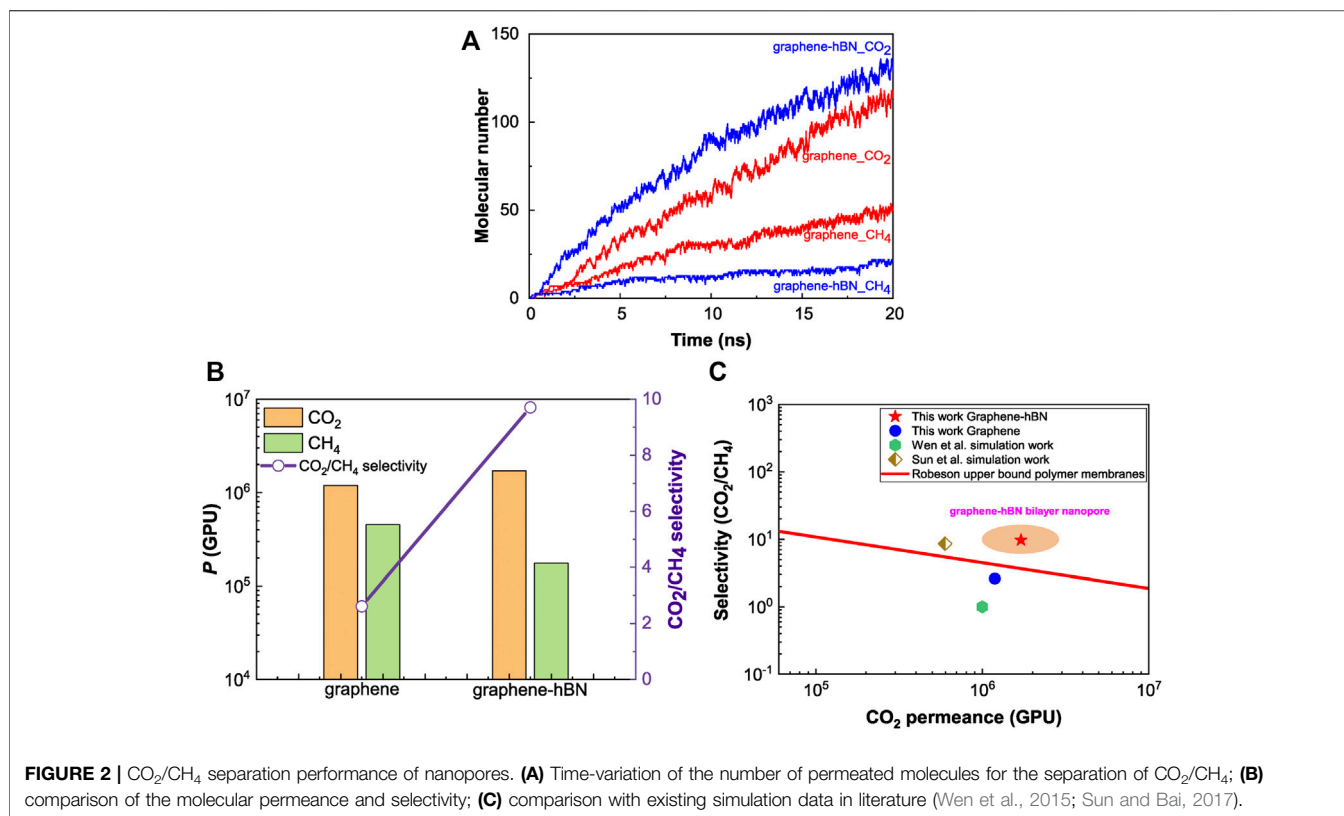


FIGURE 2 | CO₂/CH₄ separation performance of nanopores. **(A)** Time-variation of the number of permeated molecules for the separation of CO₂/CH₄; **(B)** comparison of the molecular permeance and selectivity; **(C)** comparison with existing simulation data in literature (Wen et al., 2015; Sun and Bai, 2017).

presents a 3.7 times higher selectivity comparing to the single-layer graphene nanopore with the same size. Meanwhile, we compare the separation performance of the bilayer nanopore and graphene nanopore with available simulation data of NPG membranes in literature (Wen et al., 2015; Sun and Bai, 2017), as shown in **Figure 2C**. It can be found that both the single-layer graphene and bilayer nanopores show a high molecular permeance. The graphene-hBN bilayer nanopore exhibits an ultra-high permeance and high selectivity, such that its separation performance exceeds the upper bound of polymer membranes.

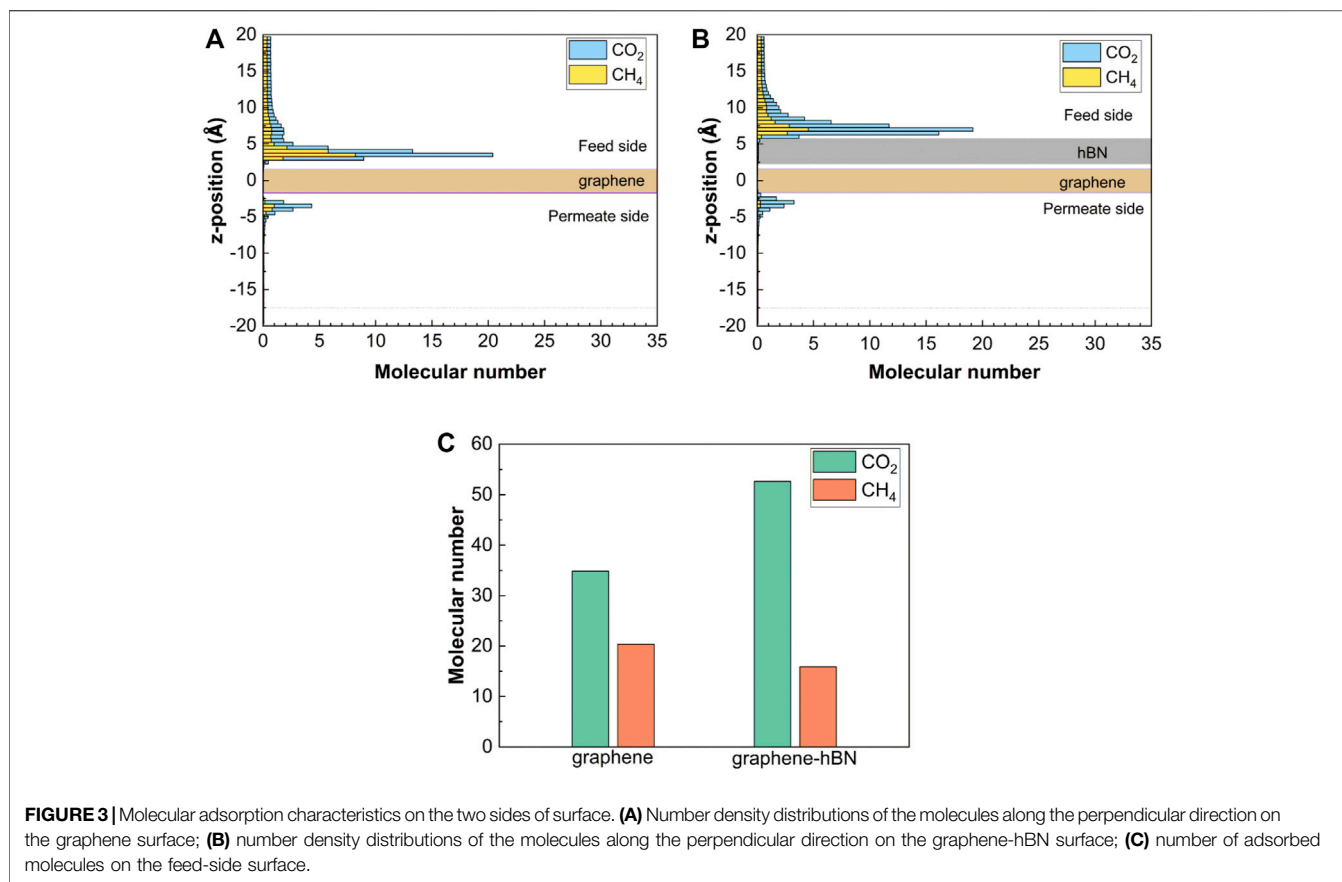
3.2 Molecular Adsorption

To understand why the graphene-hBN bilayer nanopore can present the quasi-unidirectional transport characteristic and finally exhibit a better molecular separation performance comparing to the single-layer nanopore, we firstly analyze the molecular adsorption characteristics on the two sides of the graphene-hBN bilayer surface and single-layer graphene surface. From the number density distributions of gas molecules along the perpendicular direction (**Figure 3A,B**), it can be found that the density of adsorbed molecules for CO₂ is higher than that of CH₄. This means that the interactions between CO₂ molecules and hBN or graphene surfaces are stronger than those of CH₄ molecules. Owing to the non-equilibrium permeation process, the density of adsorbed molecules in permeate side is weaker than that in feed side. Comparisons

between the bilayer surface and single-layer surface in **Figure 3C** show that the bilayer nanopore has a stronger adsorption ability of CO₂ gas molecules because the interaction of the hBN surface is greater than that of graphene surface. Meanwhile, the competitive adsorption of CO₂ and CH₄ results in a reduction in the adsorption density of CH₄ on the feed-side surface of bilayer nanopore.

The distinctive molecular adsorption on the surface can partially explain why the graphene-hBN bilayer nanopore can improve selectivity while maintain ultra-high permeability. The adsorption of gas molecules on the surfaces is beneficial for the molecular permeation through the nanopore (Drahushuk and Strano, 2012; Sun et al., 2014), therefore the stronger molecular adsorption in the feed side can promote the molecular permeation from feed-side to permeate-side. Thus, the distinctive adsorption abilities of CO₂ and CH₄ molecules on the feed side of the graphene-hBN bilayer surface is the main reason for the high-efficiency molecular sieving.

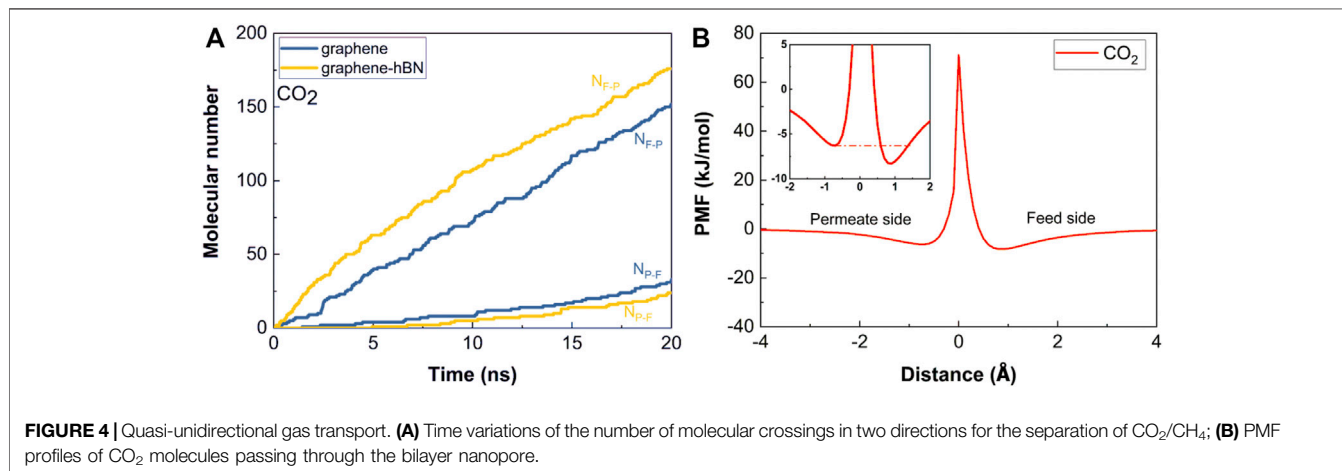
The molecular adsorption on the graphene surface may have another positive contribution on the molecular permeance by inhibiting molecular backflow from permeate-side to feed-side. This inhibition effect can be well analyzed from the aspect of molecular trajectories during permeation. We obtain the molecular crossing number N_{F-P} and N_{P-F} , respectively; N_{F-P} is the molecular crossing number from feed-side to permeate-side, while N_{P-F} is the molecular crossing number from permeate-side to feed-side. **Figure 4A**



shows the time variations of N_{F-P} , N_{P-F} for the CO_2 molecules. Over time, the two numbers all necessarily increase gradually with the permeation of molecules, but N_{F-P} increases more sharply and N_{P-F} increases very slowly. Interestingly, for the bilayer nanopore N_{F-P} increases while N_{P-F} decreases comparing to the single-layer nanopore, demonstrating that the bilayer nanopore can promote the molecular permeation from feed-side to permeate-side while inhibit the molecular

backflow. This phenomenon exactly proves the quasi-unidirectional transport characteristics of the graphene-hBN bilayer nanopore.

Then, we further obtain the interacting energy distributions between gas molecules and graphene-hBN bilayer nanopore by calculating the potential of mean force (PMF). The PMF profiles of gas molecules passing through the nanopore are calculated based on the sampling method of Adaptive Biasing Force (Sun



et al., 2019b). It can be clearly seen from **Figure 4B** that the PMF distribution is asymmetric, namely, CO₂ molecules have a relatively deeper attractive potential well in the feed side comparing to the permeate side. This further proves that there is a stronger interaction between gas molecules and graphene-hBN bilayer surface in feed side. PMF calculations give a strong evidence for the distinctive molecular adsorption abilities on the two sides and the resulted quasi-unidirectional molecular transport characteristics from the insight of thermodynamic physics.

4 CONCLUSION

We study the quasi-unidirectional transport characteristics through a graphene-hBN bilayer nanopore to realize high-efficient molecular sieving. Compared to the single-layer graphene nanopores, the bilayer nanopores exhibit both higher molecular permeance and selectivity for the separation of CO₂/CH₄ mixtures. For the bilayer nanopore, the molecular permeation from feed-side to permeate-side is enhanced while the molecular crossings from permeate-side to feed-side (molecular backflow) are weakened. The quasi-unidirectional transport characteristics are caused by the distinctive molecular adsorption abilities on two sides of the graphene-hBN bilayer surface, which are confirmed from the molecular number density distributions perpendicular to the surface and PMF profiles of gas molecules passing through the nanopore. This work provides a promising way to realize the ultra-

permeable porous membranes with molecular permeance even higher than the single-layer graphene membranes. Meanwhile, the fabrication of such two-layer membranes is instead more feasible comparing to the single-layer graphene membranes because of the elimination of intrinsic defects via the overlap of atomic-layer sheets.

DATA AVAILABILITY STATEMENT

The original contributions presented in the study are included in the article, further inquiries can be directed to the corresponding author.

AUTHOR CONTRIBUTIONS

CS performed the simulations, analyzed the data, wrote the first draft of manuscript and revised the manuscript. CL and KL analyzed the data and discussed the simulation results. BB conceived and designed the simulations. All authors contributed to manuscript revision, read, and approved the submitted version.

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