



# Pin-Hole-Free, Homogeneous, Pure CsPbBr<sub>3</sub> Films on Flat Substrates by Simple Spin-Coating Modification

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We show a simple modification of spin-coating to obtain *pinhole-free, well-covering* and *uniform* films of an all-inorganic Pb-halide perovskite (Pb-HaP), *on a flat substrate* (i.e., not mesoporous) *at standard temperature and pressure*. In Pb-HaP-based devices the active film is generally fabricated by spin-coating. This is a challenging task for purely inorganic Pb-HaPs, as, except if nano-crystallites can be used, mostly rough and inhomogeneous thin films with poor optoelectronic quality are obtained. We describe a *gas flow-assisted* method to fabricate thin *conformal* films of phase-pure CsPbBr<sub>3</sub>. First cells made with the resulting CsPbBr<sub>3</sub>-films, yield solar conversion efficiencies up to 2.5%. Electron beam-induced current measurements of device cross-sections show uniform charge generation profiles, implying  $\geq ~ 0.2 \,\mu$ m diffusion lengths. Non-encapsulated devices generate stable photocurrent for > 1 hr under continuous illumination at maximum power in ambient.

#### **OPEN ACCESS**

#### Edited by:

Chunling Wang, Shanghai Jiao Tong University, China

#### Reviewed by:

Xiaoming Li, Nanjing University of Science and Technology, China Ren Yuan Hang, Fudan University, China

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#### Specialty section:

This article was submitted to Solar Energy, a section of the journal Frontiers in Energy Research

Received: 13 October 2019 Accepted: 07 May 2020 Published: 09 June 2020

#### Citation:

Gupta S, Kulbak M and Cahen D (2020) Pin-Hole-Free, Homogeneous, Pure CsPbBr<sub>3</sub> Films on Flat Substrates by Simple Spin-Coating Modification. Front. Energy Res. 8:100. doi: 10.3389/fenrg.2020.00100 Keywords: thin films, solar cell, halide perovskite, stability, electron beam-induced current

# INTRODUCTION

Lead-based halide perovskites (HaPs) with stoichiometry AMX<sub>3</sub> (where A is a monovalent organic cation such as methylammonium-MA<sup>+</sup>; M is a divalent metal cation-Pb<sup>2+</sup> and X is a monovalent anion, a halide, iodide-I<sup>-</sup>, bromide-Br<sup>-</sup> or chloride-Cl<sup>-</sup>) have evolved as an important class of semiconducting materials for optoelectronics, due to their excellent optical and electronic properties (Hodes, 2013; Stranks et al., 2013; Yin et al., 2014; Saliba et al., 2016; Yang et al., 2017). With some specific AMX<sub>3</sub> compositions photovoltaic energy conversion efficiencies close to 25% were demonstrated recently<sup>1</sup>. We focus on bromide-based HaPs (with  $\sim$  2.3 eV band gap, E<sub>G</sub>), as they exhibit high open-circuit voltage (V<sub>OC</sub>) (Edri et al., 2013, 2014; Kedem et al., 2015; Kulbak et al., 2015, 2016), which makes them attractive starting materials for tandem cells as partner with c-Si ones. Bromide based halide perovskites have been used for high performance photodetector application (Zeng et al., 2018). CsPbBr<sub>3</sub> is a broad band-gap material ( $E_{BG} = 2.32 \text{ eV}$ ). Such materials are less studied than the smaller bandgap ones and methods to synthesize them are limited. Importantly, solar cells or light-emitting devices, made with such perovskites generate high open circuit voltage/ green-blue emission. High voltage solar cells are of interest for use in tandem or spectral splitting systems. While the bandgap of pure CsPbBr<sub>3</sub> is higher than optimal for use in tandem cells, understanding and optimizing fabrication of the more optimal mixed bromide/iodide analogs requires studying and understanding the parent compounds. Pure CsPbBr<sub>3</sub> is of direct interest for solar energy conversion, and storage as chemical energy (in chemical bonds) through water splitting and CO<sub>2</sub>

 $^{1} https://www.nrel.gov/pv/assets/pdfs/best-research-cell-efficiencies.20190802.pdf$ 

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spinning, gas is blown from the top surface, resulting in the uniform film formation.

Figure 1A shows the absorption spectrum of the  $\sim$ 200 nm thick CsPbBr<sub>3</sub> films, deposited over ~70 nm dense TiO<sub>2</sub> (d-TiO<sub>2</sub>), coated on fluorine-doped tin oxide (FTO). The onset of absorption corresponds to, E<sub>G</sub>, of 2.32 eV, consistent with earlier reports. The photoluminescence (PL) emission peak of CsPbBr<sub>3</sub> (deposited on a bare glass substrate) is at 530 nm (Figure 1A). The XRD pattern shows no evidence of different Cs-Pb-Br or Pb-Br or Cs-Br phases, i.e., within the accuracy of this method, phase-pure CsPbBr3 formed. XRD pattern of reference CsPbBr3 powder (# 97851 in the ICSD database) is shown in Figure 1B. All peaks in the diffraction patterns of the samples can be identified as belonging to, and indexed for CsPbBr<sub>3</sub>, plus those from the TiO<sub>2</sub> (substrate marked with \*), as shown in Figure 1B. Such a result is obtained within a precursor concentration range between 0.4 (M) and 0.45 (M), and after annealing at 250°C for 10 min. For concentrations > 0.45 (M), additional peaks were found (Figure 1B), corresponding to Cs<sub>2</sub>PbBr<sub>6</sub>. SEM images show smooth and pinhole-free films on the flat d-TiO<sub>2</sub>-coated FTO on glass substrate, as can be seen in the cross-section, shown in Figure 1C. The SEM image of the surface of the film is shown in Figure S1.

We compared the films of CsPbBr<sub>3</sub>, grown by gas flow-assisted with those grown by two-step deposition, both on FTO-coated glass slides. Cross-section SEM of the CsPbBr<sub>3</sub>, fabricated by the two-step deposition technique (Kulbak et al., 2015) over a dense TiO<sub>2</sub>, i.e., d-TiO<sub>2</sub>-coated FTO-on-glass slide is shown in **Figure 1D** (for planar SEM image see SI, **Figure S2**). The films grown by the two- step method are very inhomogeneous and show irregular crystallites that are oriented rather randomly on the d-TiO<sub>2</sub> on the FTO-coated glass substrate. Thus, a mesoporous, mp-TiO<sub>2</sub> layer had to be used with the two-step method, to obtain reasonable conversion efficiencies of CsPbBr<sub>3</sub> solar cells, made with the two-step process.

Having the CsPbBr<sub>3</sub> on a mesoporous substrate, while a practical solution, limits options for using other substrates and above all, the ability to study the HaP itself! In contrast, the gas-flow assisted method produces smooth and pin-hole free films on relatively smooth substrates, such as d-TiO<sub>2</sub>, and hence



reduction. Hence, studies of  $CsPbBr_3$  are of direct importance for energy conversion and storage.

In previous work we have shown that if caesium  $(Cs^+)$  is used as A cation instead of MA in the HaP, used to make the solar cell, not only are efficiencies similar to those of hybrid MAPbBr<sub>3</sub>based cells, but solar cell stability is better than for MAPbBr<sub>3</sub>based cells under prolonged AM1.5 illumination (Kulbak et al., 2016). However, preparing CsPbBr<sub>3</sub> films that are

- phase-pure (i.e., are composed of only the perovskite with ABX<sub>3</sub> stoichiometry),
- continuous, pinhole-free and, thus, cover, completely also flat, non-mesoporous substrates,
- smooth (absence of randomly oriented irregular shaped crystallites with RMS roughness of < ~10 nm)
- is a challenging task.

Indeed, most devices use a mesoporous oxide substrate to get acceptable films. For studying many of the material's fundamental properties for optoelectronic applications, a smooth, continuous well-covering thin film on a regular substrate is preferred or, at times, even needed. In this work we adapted a gas flow-assisted method, earlier used for organic-inorganic hybrid HaPs (Yu et al., 2018), to deposit such CsPbBr<sub>3</sub> films on a surface that is not mesoporous but flat. The flow of gas (through a nozzle) during spin-coating retards, due to rapid cooling by solvent evaporation, growth of nuclei, increasing the density of nucleation sites, thus helping to make the film that grows more uniform. Our focus in this preliminary study was to explore a new method to produce phase-pure and pinhole-free CsPbBr<sub>3</sub>, which was attained by the method, discussed in the work. Preliminary results of the performance of first solar cells, prepared with such CsPbBr<sub>3</sub> films, are promising. Further work is needed to enhance efficiency of the devices and measurement of their EQE and EIS, which will be our future studies.

#### **RESULTS AND DISCUSSIONS**

The process of the gas flow-assisted method is shown in **Scheme 1**, in which the perovskite precursor solution is dropped over the substrate and then the spinning is started. During





mesoporous layer is not necessary for the solar cells. Moreover, XRD revealed that films grown by two-step method are not phase-pure and consist of a mixture of CsPbBr<sub>3</sub>, CsPb<sub>2</sub>Br<sub>5</sub>, and Cs<sub>4</sub>PbBr<sub>6</sub> (**Figure S3**).

To fabricate cells, CsPbBr<sub>3</sub> films (on d-TiO<sub>2</sub>-coated FTO) were used in an *n-i-p* configuration, where TiO<sub>2</sub> serves as the n-type electron transport layer and poly-triarylamine (PTAA) as the p-type hole transport layer. The device structure was completed by thermal deposition of 100 nm of Au as contact on the PTAA. An SEM image of the cell cross-section is shown in **Figure 2B**. The cell parameters (open-circuit potential, V<sub>OC</sub>, short-circuit current density, J<sub>SC</sub>, fill factor, FF and light-to-electrical power conversion efficiency) were measured using simulated 1 sun illumination (AM1.5), with a solar simulator. The CsPbBr<sub>3</sub> device exhibited a PCE of 2.5%. The current density-voltage (J-V) curves of the device (light and dark) are shown in **Figure 2A** and the device parameters are given in **Table S1**.

We did EBIC analysis on the cross section of the best device to understand the charge generation profile (shown in **Figure 3**) from various interfaces. It can be observed from **Figure 3** that charge generation occurs throughout the CsPbBr<sub>3</sub> film. Monte Carlo simulations of electron trajectory estimates that at 3 keV, the vast majority of electrons will reach  $\sim 60$  nm away from the area of interaction. Thus, charge carrier diffusion lengths must exceed  $\sim 80$  nm (film thickness minus generation volume diameter), which implies that the EBIC signal is governed by the minority carrier diffusions length, and not by the electron beammatter volume of interaction. This further testifies to the good quality of the CsPbBr<sub>3</sub> films produced by this method.

The importance of this method can be realized from the extremely simple and non-demanding fabrication route, that lends itself to optimization to produce solar cell quality materials. Furthermore, the stability of the non-encapsulated devices was tracked by keeping these devices at maximum power point under AM1.5 simulated illuminations for 1.5 h (Figure 2C). It can be observed that the non-encapsulated device generated stable photocurrent during the period of illumination. At present we show fair unencapsulated stability, at least comparable to that of CsPbBr<sub>3</sub> devices made by other methods. In future study focus will also be on stability of the solar cells as a function of temperature and light illumination. The straightforward way in which a widely used method can be adapted to yield



what we describe and exemplify here, makes the information in this short communication is very valuable for researchers that require such thin films for their work, be it academic or applied.

# **EXPERIMENTAL SECTION**

#### **Device Fabrication**

Fluorine (F)-doped tin oxide (FTO)-based transparent conducting substrates (Xinyan Technology TCO-XY15) were cut into specific size and thoroughly cleaned by sonication in warm aqueous alconox solution, deionized water, acetone and ethanol, followed by drying under a N<sub>2</sub> stream. After an oxygen plasma treatment, a compact ~60 nm thin TiO<sub>2</sub> layer was deposited to the clean substrate by spray pyrolysis of a 30 mM titanium diisopropoxide bis(acetylacetonate) (Sigma Aldrich) solution in isopropanol using air as the carrier gas on a hot plate set to 450°C, followed by a two-step annealing procedure at 160°C and 500°C, each for 1 h in air (Kulbak et al., 2015).

The  $CsPbBr_3$  films were prepared by adding a flow nitrogen  $(N_2)$  stream at a specific stage of the process, as follows:

First the substrates (FTO coated with thin layer of TiO<sub>2</sub>; substrate size:  $2 \text{ cm} \times 2 \text{ cm}$ ) were etched by oxygen plasma asher for 10 minutes to clean the surface and enhance wettability. Immediately after this treatment, the substrates were placed on the spin-coater. After that, 250 µL of the precursor solution (containing various concentrations of both CsBr and PbBr<sub>2</sub>, viz. (0.4, 0.45, and 0.5 M) were added to the substrate (FTO coated with thin layer of TiO<sub>2</sub>) and spin coating was started immediately. The spin -coating was carried out in two consecutive stages. For the first 10s the spinning speed was 500 RPM, followed by 20s at 2000 RPM. After 10 sec of spinning (in the second stage), nitrogen gas was blown (at 0.5 LPM-liter per minute) from the top of the spin coater (vertically onto the substrate) for 10s. The method of film fabrication by gas-flow assisted method was repeated 25-30 times and the results were observed to be consistent from SEM and UV -Visible analysis. Smooth film was observed to



**FIGURE 3 | (A)** EBIC response from a cross-section of a CsPbBr<sub>3</sub> PV device showing a signal through the entire film thickness and **(B)** the corresponding secondary electron image of the device (cf. to **Figure 2B**). Because the EBIC signal does not decay within the active material, its thickness sets a lower limit for the carrier diffusion lengths in the film.

appear after the spinning was stopped. The overall procedure was carried out in an ambient atmosphere. For complete device fabrication, poly[bis(4-phenyl) (2,4,6-trimethylphenyl)amine] (PTAA—Lumtec) was deposited by spin-coating 5 s at 500 rpm followed by 40 s at 2000 rpm. For CsPbBr<sub>3</sub>, the PTAA solution contained 12 mg in 1 mL of chlorobenzene, mixed with 7.5  $\mu$ L of tert-butylpyridine (TBP) and 7.5  $\mu$ L of 170 mg/mL LiTFSI [bis(trifluoromethane)sulfonamide (in acetonitrile)]. The samples were left overnight in the dark (under dry air atmosphere) before ~100 nm gold contacts were thermally evaporated through a shadow mask with 0.24 cm<sup>2</sup> rectangular holes. For two-step deposition of CsPbBr<sub>3</sub> films we used the method, described in earlier studies (Kulbak et al., 2015).

XRD measurements were carried-out on a Rigaku ULTIMA III operated with a Cu anode at 40 kV and 40 mA. The measurements were taken using a Bragg-Brentano configuration through a 10 mm slit, a convergence Soller  $5^{\circ}$  slit and a "Ni" filter.

Optical transmission and reflection spectra of films were measured using a Jasco V-570 spectrophotometer equipped with an integrating sphere. Transmission was corrected for reflection by using  $T_{corr} = T/(1-R)$ . A Zeiss SUPRA scanning electron microscope was used for SEM and EBIC imaging of the device cross-sections.

The J-V characteristics of the solar-cells were measured with a Keithley 2400-LV SourceMeter and controlled with a Labview-based, in-house written program. A solar simulator (ScienceTech SF-150; with a 300 W Xenon short arc lamp from USHIO Inc., Japan) equipped with an AM1.5 filter and calibrated with a Si solar cell IXOLARTM High Efficiency SolarBIT (IXYS XOB17-12x1) was used for illumination. The devices were characterized through a 0.16 cm<sup>2</sup> mask. The J-V characteristics were taken after light soaking for 10 s at open circuit and at a scan rate of 0.06 V/s (unless otherwise stated).

EBIC analysis was done in a Zeiss-Supra SEM using beam current of 5 pA and beam energy of 3 keV (Kulbak et al., 2016). Current was collected and amplified using Stanford Research Systems SR570 pre-amplifier. The device cross section was exposed by mechanical cleaving immediately (up to 2 min) before transferring the sample into the SEM vacuum chamber  $(10^{-5} \text{ mbar})$ .

### CONCLUSIONS

In this study, we showed how adding a "gas flow" allows fabricating smooth, pin-hole free, phase-pure thin films of CsPbBr<sub>3</sub> on a flat (i.e., not mesoporous) substrate, as shown with the aid of SEM and XRD analysis. Such CsPbBr<sub>3</sub> films can be used to make planar architecture solar cells with n-i-p configuration. Non-encapsulated solar cells give, in first experiments, conversion efficiencies up to 2.5% under AM1.5 illuminations and were relatively stable under constant illumination for >1 h.

### DATA AVAILABILITY STATEMENT

All datasets generated for this study are included in the article/**Supplementary Material**.

### **AUTHOR CONTRIBUTIONS**

SG prepared the films and characterized. MK have performed the EBIC and analyzed the results. DC helped in the preparation of manuscript and evaluation of results.

### ACKNOWLEDGMENTS

We thank Gary Hodes for discussions and Dr. Pallavi Singh for help with the X-ray diffraction data analysis. Partial support for this research work was provided by the Yotam project, via the Weizmann Institute of Science's Sustainability and Energy Research Initiative, SAERI, and by the Israel Ministry of Energy and Infrastructure.

### SUPPLEMENTARY MATERIAL

The Supplementary Material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/fenrg. 2020.00100/full#supplementary-material

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**Conflict of Interest:** The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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