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Dispersion and fate of methane emissions from cold seeps on Hikurangi Margin, New Zealand

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The influence of cold seep methane on the surrounding benthos is welldocumented but the fate of dissolved methane and its impact on water column biogeochemistry remains less understood. To address this, the distribution of dissolved methane was determined around three seeps on the south-east Hikurangi Margin, south-east of New Zealand, by combining data from discrete water column sampling and a towed methane sensor. Integrating this with bottom water current flow data in a dynamic Gerris model determined an annual methane flux of 3 x 10⁵ kg at the main seep. This source was then applied in a Regional Ocean Modelling System (ROMS) simulation to visualize lateral transport of the dissolved methane plume, which dispersed over ~100 km in bottom water within 1 year. Extrapolation of this approach to four other regional seeps identified a combined plume volume of 3,500 km³ and annual methane emission of 0.4–3.2 x 10^6 kg CH₄ y⁻¹. This suggests a regional methane flux of $1.1-10.9 \times 10^7$ kg CH₄ y⁻¹ for the entire Hikurangi Margin, which is lower than previous hydroacoustic estimates. Carbon stable isotope values in dissolved methane indicated that lateral mixing was the primary determinant of methane in bottom water, with potential methane oxidation rates orders of magnitude lower than the dilution rate. Calculations indicate that oxidation of the annual total methane emitted from the five seeps would not significantly alter bottom water dissolved carbon dioxide, oxygen or pH; however, superimposition of methane plumes from different seeps, which was evident in the ROMS simulation, may have localized impacts. These findings highlight the value of characterizing methane release from multiple seeps within a hydrodynamic model framework to determine the biogeochemical impact, climate feedbacks and connectivity of cold seeps on continental shelf margins.

KEYWORDS

methane, hydrates, hydrodynamic model, methane oxidation, continental shelf, New Zealand, mixing, climate change

1 Introduction

Although the ocean only accounts for 1%-3% of global methane (CH₄) emissions (9–22 Tg CH₄ y⁻¹, Saunois et al., 2020) marine methane is attracting increasing attention, in part due to its sensitivity to climate change. Marine methane sources are dominated by cold seeps on continental shelf margins and coasts (Weber et al., 2019), which primarily originate from underlying gas hydrates. The latter form in the Gas Hydrate Stability Zone (GHSZ),

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at low temperature and high pressure, hence their subsurface presence on continental shelves and margins (Hester and Brewer, 2009). Estimates vary widely and indicate that 450-75,000 Pg CH₄ may be stored globally in these reservoirs (Klauda and Sandler, 2005; Wallmann et al., 2012), with annual global seafloor seepage potentially reaching 65 Tg (US EPA: Office of Atmospheric Programs, 2010). However, their contribution to atmospheric emissions is negligible (Saunois et al., 2020), due to a combination of physical, geological and biological barriers that minimize methane transfer to the ocean surface. Methane flux from subsurface hydrates is primarily suppressed by anaerobic methanotrophy in the overlying sediment, which significantly reduces methane release into bottom water to ~0.02 Gt C y⁻¹ (Boetius and Wenzhöfer, 2013). However, gaseous methane can evade this benthic filter (Sommer et al., 2006), with bubble ebullition and subsequent diffusion elevating dissolved methane in bottom waters (McGinnis et al., 2006). Dispersion and dilution by bottom currents (Kessler et al., 2011; Steinle et al., 2015), and oxidation via the pelagic filter of aerobic methanotrophs (Reeburgh, 2007) then further reduce dissolved methane concentration (de Angelis et al., 1993; Mau et al., 2013; Leonte et al., 2017). The dominance of lateral flow relative to vertical exchange in bottom and intermediate waters, combined with stratification in the main pycnocline, effectively caps the transfer of dissolved methane to surface waters and the atmosphere, with retention in deeper water ultimately resulting in aerobic oxidation to carbon dioxide (CO₂) (Valentine et al., 2001; Reeburgh, 2007).

Methane release at cold seeps has a significant influence on local benthic biology, as it supports microbial communities and associated biogeochemical pathways that sustain unique infauna and macrofauna assemblages via chemosynthesis (Boetius and Wenzhöfer, 2013; Bowden et al., 2013). There is also evidence around seep sites of trophic transfer of carbon derived from methane in food webs that include commercially important fisheries species (Niemann et al., 2013; Seabrook et al., 2019). The potential sensitivity of hydrates to tectonic activity (Crutchley et al., 2023; Rudebusch et al., 2023) and ocean warming (Ruppel and Kessler, 2017), and also interest in methane hydrates as a potential fuel source (Moridis et al., 2011), has focused attention on the fate of methane in bottom water. Warming of deep water may destabilize hydrates at the upper edge of the GHSZ, as indicated in the paleorecord by significant methane release from subsurface reservoirs coincident with periods of ocean warming (Maslin et al., 2010), and there is increasing contemporary evidence for this (Phrampus and Hornbach, 2012; Trivedi et al., 2022). The resulting increase in methane in deep waters could enhance CO₂ production and consumption of oxygen (O₂) via aerobic oxidation, so accelerating acidification and deoxygenation (Biastoch et al., 2011; Ruppel, 2011; Boudreau et al., 2015). However, the impact of seep methane on water column biogeochemistry has received less attention (Levin et al., 2016), in part due to the challenge of tracking dissolved methane in bottom water.

Fluid seepage and cold seeps are well-characterized along the Hikurangi Margin, east of New Zealand (Lewis and Marshall, 1996; Barnes et al., 2010; Watson et al., 2020), with subsurface hydrates evident as bottom-simulating reflectors in seismic data (Henrys et al., 2003; Klaucke et al., 2010; Crutchley et al., 2021; Crutchley et al., 2023). Regional analysis indicates seepage

occurring predominantly along the deforming backstop of the margin, with fluid release from both the subducting plate and the accretionary wedge (Watson et al., 2020). Fluid migration is facilitated by folding, fractures and faults arising from tectonic activity and strata of elevated permeability and porosity. Cold seep location is further evidenced by the presence of gas flares or bubble plumes in hydroacoustic backscatter data (Turco et al., 2022), that may extend up to 1 km above the seafloor (Faure et al., 2010; Law et al., 2010). To determine the fate of methane released from cold seeps, dissolved methane concentrations and related biogeochemical parameters including dissolved O2 and carbonate system parameters were measured in the water column at three seep sites on the south-east Hikurangi Margin. As tracking of dissolved methane in bottom waters is confounded by tidal currents and dispersal in three dimensions, the dissolved methane source and distribution at the main seep site was simulated at local scales in a hydrodynamic framework using a dynamic Gerris model, with dispersal along the margin then determined using the Regional Ocean Modelling System (ROMS). This was extended to four other regional seeps (Turco et al., 2022) to assess the combined dispersal of seep methane and influence on bottom water biogeochemistry. Methane loss was assessed by potential oxidation rate measurements, with oxidation also examined indirectly using carbon stable isotope values in dissolved methane (δ^{13} C-CH₄), to establish the relative influence of dispersion and oxidation on seep methane in bottom waters along the Hikurangi Margin.

2 Methods

2.1 Regional setting

Sampling was carried out during July 2019 onboard *R/V Tangaroa*, as part of the New Zealand HYDEE (Gas Hydrates: Economic Opportunities and Environmental Implications) programme. The study area lies at the southern end of the Hikurangi subduction margin (Figure 1) where the Pacific Plate subducts obliquely beneath the Australian Plate. Measurements were focused on three seeps - Maungaroa, Glendhu Ridge and Uruti Ridge - with data from two additional seep sites–Pahaua Bank and Uruti Ridge West - also considered in the analysis (Turco et al., 2022; see Table 1). Dispersal modelling focused on the main seep site at Maungaroa, where a subsurface layer of gas of >200 m height below the GHSZ has generated doming, fracturing and formation of a conduit that facilitates gas transfer to the seafloor (Crutchley et al., 2021).

2.2 Bottom water hydrography and properties

A seafloor mooring was deployed in the vicinity of each of the three seeps consecutively to record bottom water physical and biogeochemical properties. The mooring frame was attached to a 460 kg ballast weight and supported a 600 kHz Acoustic Doppler Current Profiler (ADCP) for measuring current direction and velocity. A SBE 37 ODO MicroCAT with conductivity, temperature and optical dissolved O_2 sensors was also deployed on the mooring at ~15 m above the seafloor. The mooring was deployed for periods



FIGURE 1

Location of the five seep sites on the south-east Hikurangi Margin plotted against bathymetric depth (m), as shown in the colour bar. The three seep sites surveyed in this study (Maungaroa, Uruti Ridge and Glendhu Ridge) are indicated by yellow circles, and the two additional sites (Uruti Ridge West and Pahaua Bank, (Turco et al., 2022) by white circles. The *x*-axis shows longitude (°East) and *y*-axis latitude (°South).

TABLE 1 Seep locatio	ons and dimensions or	the south-east Hikur	angi Margin (all data fi	rom Turco et al., 2022	<pre>2, except for * Crutchle</pre>	y et al., 2021), with					
the sites sampled in this study indicated by †.											

Seep	Latitude (ºE)	Longitude (ºS)	Depth (m)	Seafloor acoustic anomaly area (km ²)	Bubble plume height (m)	Gas flux from bubble plume (10 ⁶ kg y ⁻¹)
Maungaroa†	175.310	41.940	2081	0.2 *	1,400	2.52-7.18
Pahaua Bank	175.705	41.716	1,500	0.21	810	1.47-5.58
Glendhu Ridge†	176.089	41.768	1950	0.17	1,200	1.57-4.34
Uruti Ridge West	176.294	41.426	1,110	0.45	470	1.82-6.54
Uruti Ridge†	176.347	41.426	1,240	0.45	500-600	1.28-3.55

of 3–5 days at each site, which either preceded or followed other sampling activities to ensure the mooring and sensors were not compromised. On recovery, the mooring frame was decoupled from the ballast by remote activation of an acoustic release device whilst the ballast remained on the seafloor. The ADCP data was processed and quality controlled to generate a filtered data set with records at 120 s intervals over the lifetime of each respective deployment.

2.3 Water column sampling and characterisation

Dissolved methane concentrations were determined by four different approaches: discrete water sample collection in Niskin bottles on a CTD (Conductivity, Temperature and Depth) rosette in either a) standard vertical profiling mode or b) lateral towed mode, with methane concentration also measured by a methane

sensor deployed on c) the CTD or d) a Deep Towed Imaging System (DTIS, Bowden et al., 2013). As the dissolved methane plume was largely retained in near-bottom waters, the CTD and DTIS lateral tows focused on the bottom 10 m of water overlying the seafloor, with each sensor tow orientated relative to the seep by reference to a predetermined location and information from preceding tows. Each sensor tow started ~1 km from the seep and crossed this at the transect mid-point before completion at a similar distance of 1 km. During the CTD sensor tow, Niskin bottle water samples were obtained with increased sampling frequency close to the seep, with the remaining bottles fired in the bottom 200-500 m during retrieval. At least one vertical CTD cast was carried out at a distance of >1 km from the seep at all three sites to characterize background conditions. Bottom waters at Maungaroa received the most intense sampling, with six DTIS tows and 12 CTDs. Stratification was determined using buoyancy frequency squared (N²), which was calculated as:

$$N^2 = g/\rho_0 \, \delta\rho/\rho z$$

where *g* is gravitational acceleration, ρ_0 is a reference density, and $\delta\rho/\rho z$ is the vertical density gradient (Stevens et al., 2012).

2.4 Dissolved methane measurements

2.4.1 Analysis of discrete water samples

Seawater from the 10-L Niskin bottles was sub-sampled into 240 mL glass serum bottles, which were overflowed by at least 100% to exclude bubbles and then sealed with butyl rubber septa. The samples warmed passively to ambient temperature, with analysis within 4–5 h of collection. A 40 mL headspace of O₂-free nitrogen (N₂) was injected into the bottle whilst 40 mL of seawater was expelled, with the two phases subsequently equilibrated at room temperature for 30 min. The equilibrated headspace was displaced through a drying tube containing magnesium perchlorate and then a 2.9 mL sample loop, the contents of which were injected onto a 1/8" molecular sieve chromatographic column with methane detection by a Flame Ionisation Detector (Law et al., 2010). Calibration was carried out using prepared gas standards certified by comparison in an international calibration exercise (Wilson et al., 2018), with peak area variation of <0.35% over a 12-h analysis period. Dissolved methane concentration (nmol L⁻¹) was calculated from the measured partial pressure of methane in the headspace, following correction for water vapour and using the equilibration temperature and salinity with the solubility coefficients of Weisenburg and Guinasso (1979). Precision of duplicate sample analysis was 2.5% (1 standard deviation), consistent with the reproducibility reported in an international intercomparison (Wilson et al., 2018). Methane saturation from atmospheric equilibrium was calculated using the mean atmospheric methane mixing ratio of 1740 ppb recorded at the nearby National Institute of Water and Atmospheric Research (NIWA) Baring Head Global Atmosphere Watch Programme station (World Data Centre for Greenhouse Gases; kishou.go.jp), with dissolved methane concentration expressed as a percentage of expected saturation. An example of the vertical profile data for dissolved methane and associated parameters is available in Supplementary Table S1 (Supplementary Material), with the data for all profiles available at the associated website.

2.4.2 Continuous dissolved methane measurement

The dissolved methane dataset was extended to greater temporal and spatial resolution by deployment of a METS Sensor (Franatech, Oslo, Norway). This sensor was used in two modes: i) attached to the CTD, which enabled comparison with discrete sample data so enabling sensor calibration at in situ temperature and pressure; and ii) mounted on the DTIS (Bowden et al., 2013), which generated detailed transects across the seep region in bottom waters between 2 and 8 m above the seafloor. Each CTD lateral tow was at slightly shallower depths (5-15 m above the seabed), with the CTD initially held for 5-10 min to allow equilibration of the METS sensor before towing commenced. The METS relies upon methane diffusion across a permeable membrane, with diffusion assisted by a pump head, and detection by semi-conductor technology. Methane adsorption results in a change in conductivity which is converted to voltage as an output signal, with temperature sensitivity accounted for by a separate voltage (Krabbenhoefft et al., 2010). As diffusion across the membrane is rate-limiting this results in a lag in METS response, which was corrected for by comparison with in situ discrete sample data. The METS working temperature minimum of +2°C and methane concentration sensitivity of ~1 nmol L^{-1} were well suited to measurements in bottom waters along the Hikurangi Margin.

2.5 Dissolved inorganic carbon and total alkalinity

Bottom water collected in the Niskin bottles was subsampled into 1-L and 2-L glass bottles on selected CTD tows and profiles at seep and background stations for analysis of dissolved inorganic carbon (DIC) and total alkalinity (TA), respectively. The bottles were overflowed by 100%, with samples preserved by saturated mercuric chloride addition, sealed with parafilm and stored in the dark, with subsequent analysis in the laboratory. DIC concentration was determined coulometrically (Dickson et al., 2007) with an estimated accuracy and precision of ±1 µmol kg⁻¹ based on repeat analysis of Dickson Certified Reference Materials. TA was determined by closed-cell potentiometric titration (Dickson et al., 2007), using a curve-fitting optimization with least squares analysis of the titration curve, with an estimated accuracy and precision $\pm 2 \mu mol \text{ kg}^{-1}$ also based on repeat analysis of Dickson Certified Reference Materials measurements. The DIC and TA data were used to calculate in situ pH on the total scale using in situ temperature, salinity and pressure and Mehrbach dissociation constants (Mehrbach et al., 1973), as refitted by Dickson and Millero (1987).

2.6 Dissolved methane carbon isotope analysis

Stable C isotope ratios of CO₂ were measured using a GasBench II and GC-PAL autosampler coupled to a DELTA V Plus isotope ratio mass spectrometer (IRMS) (all Thermo Fisher Scientific, Bremen, Germany). Daily checks confirmed that IRMS linearity was <0.06‰ volt⁻¹ for a known m/z 44 voltage span, and the standard deviation of δ^{13} C values from 10 peaks of working standard CO₂ was

always <0.06‰. Values of sample δ^{13} C were derived relative to the CO₂ working standard gas introduced directly into the IRMS and corrected for ¹⁷O (Santrock et al., 1985) by Isodat software (version 3.0.94.17; Thermo Fisher Scientific, Bremen, Germany). Isotopic values were reported in δ -notation relative to the Vienna Pee Dee Belemnite scale. This approach was used for both dissolved methane (Section 2.6.1), and also for DIC in the potential oxidation rate measurement (Section 2.7).

2.6.1 Sample preparation for methane carbon isotope analysis

Seawater samples for dissolved δ^{13} C-CH₄ analysis were subsampled from the Niskin bottles into 285 mL glass serum bottles, poisoned with mercuric chloride and sealed with butyl rubber septa. Sample temperature was slowly raised to ambient temperature with a hypodermic needle inserted through the septa to allow water to expel during volume change. In the laboratory, a 15 mL headspace of ultra-high purity helium was introduced while withdrawing an equivalent volume of seawater to obtain a seawater:headspace volume ratio of 18, with the two phases equilibrated on a shaker for 30 min. Using a GasBench II and GC PAL autosampler (Thermo Fisher Scientific, Bremen, Germany) the headspace gas was purged at 20 mL min⁻¹ in a helium carrier flow and directed through a series of in-line traps to remove water, carbon monoxide and CO₂. A PreCon cold trap immersed in liquid N₂ removed nitrous oxide before sample methane was trapped on a second PreCon automated cold trap comprised of 100 cm of coiled GS-Q capillary column (divinylbenzene (DVB), wide-bore 0.32 mm D; Agilent Technologies, United States) also held in liquid N₂ (Yarnes, 2013). On removal of the second trap from the liquid N₂ the methane was transferred into the helium carrier flow and combusted by passing through the PreCon combustion reactor (99.8% alumina, 1.6 mm outer diameter, 0.8 mm bore, 327 mm length; McDanel Ceramics, United States) at 1,000°C. After drying, the CO₂ produced by methane combustion was cryo-focused in the GasBench II cold trap held in liquid N2, and then separated from residual gases by a chromatographic column (CP-PoraPLOT Q) at 70°C. C stable isotope analysis was as described above, with δ^{13} C-CH₄ from clean air from the Baring Head station used to correct sample δ^{13} C-CH₄ values (Paul et al., 2007).

2.7 Methane oxidation rate

Potential methane oxidation rate was measured in selected bottom water samples using ¹³C-CH₄ addition, following the method of Leonte et al. (2017). Seawater was transferred from the Niskin bottle into 6 x 240 mL serum bottles for each depth sample, sealed with butyl rubber septa and aluminium seals, and placed in the dark at ambient bottom water temperature (2–2.5°C). Within three-five hours, four of the six serum bottles were injected with 50 μ L ¹³C-CH₄ mixed with 50 μ L ambient air from a gastight syringe, during which the serum bottle was inverted and the rubber seal perforated by a separate needle to maintain atmospheric pressure in the bottle. The remaining two serum bottles were not amended and provided a time zero for the rate measurement. All bottles were gently agitated for 20 min on a wrist-action shaker, after which the four amended samples were incubated in the dark at ambient bottom water temperature for three-five days. The incubation was terminated at time zero, Day 2 and Day 4 by removal of 2 mL of water; this was injected into a 12 mL gas-tight glass vial sealed with a rubber septum previously flushed with high purity helium and containing 100 µL of phosphoric acid. In the laboratory, acidified samples were equilibrated overnight at 24°C, during which the CO₂ released by acidification of DIC had equilibrated with the headspace. C stable isotope analysis was as described above in Section 2.6. Reference materials IAEA NBS-18 and IAEA-603 were used to normalize all $\delta^{13}C_{gas}$ values (Paul et al., 2007). Values of $\delta^{13}\text{C-DIC}$ were calculated from $\delta^{13}\text{C}_{\text{gas}}$ values by accounting for the C stable isotope fractionation associated with the CO₂ gas-aqueous partition (Assayag et al., 2006). Potential methane oxidation rate was determined by the transfer of ¹³C from the methane gas to DIC. Background in situ DIC concentrations were determined using the individual sample CO2 peak values generated by the GasBench II, calibrated against a subset of samples analysed by coulometry, as detailed in Section 2.5. Rates (nmol l⁻¹ d⁻¹, hereafter identified as nM d⁻¹) were derived using the approach of Leonte et al. (2017), and are regarded as "potential" oxidation, reflecting the large ¹³C-CH₄ addition relative to the background dissolved methane concentration.

2.8 Modelling

2.8.1 Near-field dispersion of dissolved methane in bottom water

A dynamic model of nearfield dispersion of dissolved methane around the Maungaroa seep was developed using Gerris, a semistructured adaptive grid model for the solution of the nonhydrostatic Navier-Stokes equations (S. Popinet, http://gfs.sf.net; Rickard, 2020). The primary aim of the Gerris modelling was to estimate the source input of methane into the water column by assessing advection and diffusion of the methane tracer via fitting to the spatial and temporal transect observations. Dissolved methane was treated as a passive tracer and so was advected and diffused with the imposed flow. As Gerris is applicable to small scales it was appropriate for modelling dispersion of dissolved methane within a 1 km radius of the seep where most of the water sampling took place. The three-dimensional model domain spanned 2.25 km in each horizontal direction and 250 m in the vertical plane, with grid spacing over most of the domain of 7.8 m. The model incorporated advection associated with M2 tidal currents, as measured by the ADCP, with the seep methane assumed to be entrained within 50 m above the seafloor (see Results). The model was spun up for a number of M2 tidal cycles, thereby allowing the methane tracer to advect and diffuse across the domain, with two to four M2 tidal cycles producing tracer patterns that best fitted the observations. Using these best fits an "optimal" ratio of the observed maximum methane concentration to the Gerris model maximum was determined for each transect and profile, with the average ratio subsequently used to calculate methane emission.

2.8.2 Far-field dispersion of dissolved methane

The ROMS (Haidvogel et al., 2008) was used to simulate the advection of passive 'dye' tracers released at each of the five seep sites (Table 1) along the southern Hikurangi Margin over a period of

1 year. The model domain spans the continental shelf of the eastern North Island of New Zealand, from Cook Strait in the south to East Cape in the North, with a horizontal resolution of 2 km and 30 terrain-following vertical coordinates. The model bathymetry was constructed from various sources including the NIWA gridded bathymetry database, land elevation data, regional coastline data, and the General Bathymetric Chart of the Oceans (GEBCO) gridded ocean bathymetry. The model was initialized on 1 January 2013 with initial and boundary conditions derived from a global ocean reanalysis based on the Hybrid Coordinate Ocean Model (HYCOM; Chassignet et al., 2009). The global HYCOM product used provides daily snapshots of the 3-dimensional state of the global ocean on a 1/12° grid. Tidal currents were added to the HYCOM boundary conditions using amplitude and phase data for 13 tidal constituents derived from the New Zealand tidal model described by Walters et al. (2001). The ROMS simulation was forced at the surface with wind stress calculated from 3-hourly winds obtained from the 12 km New Zealand Limited Area Model (NZLAM; Lane et al., 2009). Heat and freshwater fluxes were obtained from six-hourly NCEP/NCAR reanalysis data with a horizonal resolution of 2.5° (Kalnay et al., 1996).

The passive tracer computational capabilities of ROMS were used to determine the dispersal of methane via a conservative dye tracer released in the bottom three grid cells (150–200 m above the bottom) at each of the five seep sites. The passive tracers were advected from the release locations following the same governing equations as temperature and salinity. The passive tracer has a unit concentration and was initialized from each location with a value of 100. To simulate a continuous release, the passive tracers were nudged back to a value of 100 at the release locations on a daily basis using a nudging coefficient of 0.05 days⁻¹.

The calculated methane source function generated by the Gerris model was applied to the ROMS dye tracer to determine dispersion from the Maungaroa seep in bottom waters along the Hikurangi Margin. The ratio of the Gerris source function to an independent estimate of methane emission from hydroacoustic backscatter (Turco et al., 2022) was generated for Maungaroa, and then applied to the estimated methane emission from hydroacoustic backscatter for the other four sites (Turco et al., 2022) to re-evaluate the total emission from the five seeps.

3 Results

3.1 Dissolved methane distribution

Analysis of the integrated discrete water sample and highresolution sensor data determined that most of the dissolved methane was retained in bottom waters up to 200 m above the seafloor, although this was biased by the large number of sensor measurements obtained within 5 m of the seafloor. Dissolved methane concentration varied over four orders of magnitude, with maximum concentrations exceeding 2 μ mol L⁻¹ (supersaturation of 60,000 relative to atmospheric methane) at the Maungaroa and Uruti Ridge sites (Figure 2, Supplementary Table S1). The lower dissolved methane maximum at Glendhu Ridge may reflect that this site is characterized by a number of smaller dispersed seeps (Turco et al., 2022). All three sites showed methane undersaturation, indicative of methane oxidation, in the mid-water column at depths of 750-1,000 m for the shallower Uruti Ridge site and 1,000-1,500 m for the two deeper sites (Figure 2; Supplementary Table S1). Mid-water column methane concentrations were similar at the background and seep stations at Maungaroa, although elevated methane concentration of ~20 nmol L⁻¹ was evident at 1,600-1800 m at the background sites. All three sites exhibited a mean methane concentration of ~4.5 nmol L⁻¹ between 500 m and the surface, with a maximum of ~12.5 nmol L^{-1} at a background site near Maungaroa. The near-surface methane supersaturation of <150% was consistent with previous regional surveys (Law et al., 2010), confirming low regional methane flux to the atmosphere. Although there was no evidence of seep methane reaching the surface, emissions from shallower seeps along the Hikurangi Margin may occur (Higgs et al., 2019). Alternatively, the low methane supersaturation in surface waters may be produced by methane phosphonate or methylotrophic pathways (Karl et al., 2008; Weller et al., 2013).

3.1.1 High resolution dissolved methane in bottom waters

Comparison of discrete sample and sensor data on three CTD tows at the Maungaroa seep (Figure 3) indicated that the sensor generally overestimated background dissolved methane concentration and underestimated high values. This was not surprising as diffusion limits the sensor response with equilibrium taking longer than exposure to the methane maximum during towing. This also resulted in a "memory" effect, with temporal smearing generating lower broad dissolved methane concentrations relative to the sharper and higher concentrations in the discrete bottle data (see Figure 3). Corrected in situ dissolved methane concentrations (CH4 insitu) were generated from the sensor data by applying the equation $CH_{4 insitu} = (CH4_{sens} -2)^{1.15}$, which was derived from in situ calibration of the sensor relative to discrete bottom water samples. The limiting step of diffusion also caused a sensor response lag of ~5-6 min, as confirmed by comparison of sensor and discrete sample data, and so the dissolved CH4 insitu was also corrected for this delay (Figure 3). However, the in situ temporal lag was shorter than in prevoyage laboratory tests, potentially reflecting increased diffusion at the elevated in situ pressure in bottom water. Although the sensor "memory" could not be explicitly determined, the combined concentration and temporal corrections showed good agreement between the calculated CH4 insitu and the bottle data (Figures 3A, B), with some indication of smearing following the methane maxima.

The lateral distribution of dissolved methane at Maungaroa was determined by a series of transects carried out with reference to the seep location and also elevated backscatter, which indicated the presence of authigenic carbonates formed by precipitation that is characteristic of seeps along the Hikurangi Margin (Figure 4C; Bowden et al., 2013; Crutchley et al., 2021). In two of the CTD transects dissolved methane associated with the bubble plume reached a water column depth of ~1,600 m with lateral transport to the north-east (towards 4 in Figure 4B). Retention in bottom waters occurred, with over 50% of total dissolved methane within 50 m of the seafloor and 75% in the bottom 200 m. This largely reflects rapid diffusion of methane out of the bubbles (McGinnis et al., 2006),



FIGURE 2

Vertical profiles of dissolved methane concentration (nmol/l) at the three seep sites: (A) Maungaroa, (B) Glendhu Ridge and (C) Uruti Ridge, with consistent depth and concentration scales used for comparison. The dashed line indicates the dissolved methane concentration expected from equilibrium with atmospheric methane. In (A) samples within 1 km of the Maungaroa seep are shown by blue circles, and those from background sites >1 km from the seep by open circles.



FIGURE 3

(A,B) CTD-sensor transects in bottom waters 5 m above the seafloor at Maungaroa, showing dissolved methane concentration (nmol/L) as determined by the sensor (CH_{4 sens}, dashed line), and discrete Niskin samples (open circles), overlain by the corrected sensor data (CH_{4 insitu}, black lines). The *y*-axis indicates the duration of the sensor tow in hours.



which is then transferred laterally in dissolved form by bottom currents. This may be supplemented by dissolved methane in fluid release at the seafloor (Krabbenhoeff et al., 2010), although these two sources are indistinguishable with the techniques employed in the current study.

3.2 Bottom water currents and properties

Regional circulation in the water column overlying the Maungaroa Seep is influenced by eddies and mixing between the northward-flowing Southland Current and the D'Urville Current which flows through the Cook Strait. This generates the Wairarapa Coastal Current (WCC), a relatively cool northward flow along the outer shelf and upper slope (Figure 5D), inshore of the southward-flowing East Cape Current (ECC, Chiswell, 2003). The observed salinity variation between 200 m and 400 m (Figure 5C) likely reflects variability in the relative contributions of the Southland Current, ECC and D'Urville Current between sampling profiles. Below this the Antarctic Intermediate Water (AAIW) is evident as a salinity minimum around 1,000 m depth, with salinity increasing to the seafloor in Pacific Deep Water (PDW).

The mean Buoyancy Frequency squared (N^2) profile showed strongest vertical stratification between 250 m and 400 m depth at the base of the WCC (Figure 5B), and then decreased to the core of the AAIW at 1,000 m depth. N^2 subsequently increased to ~1,250 m depth, at the interface between the AAIW and PDW, before decreasing again to the seafloor. Comparison with

the dissolved methane profile at Maungaroa (Figure 5A; also Figures 4A, B) suggests that water column structure and stability have a limited influence on vertical dissolved methane distribution. The bubble plume extended to a water column depth of ~1,000 m, shallower than the mid-water column stratification maximum at 1,250 m (compare Figures 5B, E), whereas dissolved methane was largely retained below 1,500 m where there was no corresponding N² maximum (compare Figures 5A, B). The absence of nearseafloor stratification (Figure 5B) is in contrast to Opouawe Bank nearby, where dissolved methane from a shallower seep (depth of ~1,080 m) was constrained by elevated N² just above the seafloor (Law et al., 2010).

The mooring at Maungaroa was located less than 100 m from the seep centre, with sensor measurements at 15 m above the seafloor. The ADCP recorded relatively low currents ($<0.15 \text{ m s}^{-1}$), with north-easterly flow dominating and tidal currents running along the shelf parallel to the coast (Supplementary Figure S1). Fitting of M2 tides confirmed that tidal forcing was the dominant factor, despite an apparent anomaly in flow direction on Day 2. The tidal influence was also apparent in bottom water temperature and salinity, which showed an inverse relationship and variability of <0.2°C and 0.02 ppt, respectively, over the 5-day period. Dissolved O_2 concentration showed minor variation of <2 µmol L⁻¹ relative to a bottom water mean of 152.5–155 $\mu mol \ L^{-1}$ (44.8%–45.5% saturation with respect to the atmosphere). This variability primarily reflected temperature-related variation in solubility and there was no temperature or dissolved O2 signal associated with the seep plume (Supplementary Figure S1), as reported in a previous regional study (Krabbenhoefft et al., 2010).



3.3 Modelled near-field methane distribution

The modelled methane was scaled by reference to the observed dissolved methane concentration in bottom waters within a 1 km radius around the seep using the Gerris model to estimate the methane source function, which was then converted to a flux by applying this to a source volume. Figures 6A, B show two sensor tows that crossed the Maungaroa seep, with the modelled distribution at the corresponding tidal state shown in Figures 6C, D. Overlaying the transects in Figures 6C, D highlights the overriding influence of the tidal cycle on the spatial distribution of seep-derived methane in bottom waters; despite both transects passing within 50 m of the seep, the modelled methane concentrations and distribution differ markedly due to the tidal state, as shown in the corresponding model simulations.

The footprint of the seep was assessed using different approaches. Previous measurement of water column backscatter indicated a seep footprint of $\sim 200 \text{ m}^2$ at Maungaroa (Crutchley et al., 2021), although this is likely an overestimate as the bubble plume will be dispersed over a broader area by bottom currents and advection relative to the actual seafloor release area. An alternative approach to estimating seep spatial extent utilized the decline in meiobenthos abundance from the seep centre to background (D. Leduc, pers. comm.) which provided a horizontal length scale of $\sim 50 \text{ m}$. As the latter

estimate is consistent with the length scale reported for active fluid release at seeps along the Hikurangi Margin (Klaucke et al., 2010), a footprint of 50 m x 50 m was applied with the Gerris simulation, to generate a source function of 4.7×10^{-6} mol m⁻³ s⁻¹ methane, and so an annual methane emission of 3 x 10⁵ kg at the Maungaroa seep.

3.3.1 Modelled far-field dispersion of dissolved methane

ROMS-simulated bottom water currents were comparable with the ADCP currents in strength ($<0.15 \text{ m s}^{-1}$) and direction, with north-easterly and south-easterly flow combining with tides to generate a net north-easterly advection (Figure 7A). Dye tracer released at the Maungaroa seep site in the ROMS simulation advected along the 2000 m isobath a maximum distance of $\sim 87 \text{ km}$ north-east of the seep location over 1 year. Scaling the ROMS tracer concentration to the Gerris-derived source function showed that modelled methane concentration was highest in a plume extending north-east of the seep centre (Figure 7B), but this declined sharply, falling below 500 and 200 nmol L⁻¹ within 10 and 20 km of the seep location, respectively, with a total plume surface area of 4,000 km² after 1 year (see Supplementary Table S2).

The ROMS model was subsequently applied across the southeast Hikurangi Margin to assess plume dispersion for the four other cold seeps, characterized by Turco et al. (2022); see Table 1). Although the tracer was advected primarily to the north-east



transects by the continuous black line, and the modelled dissolved methane concentration by colour corresponding to the underlying colour bar. The white arrows show the tidal flow vectors at the time of the concentration contour snapshot in each frame, to indicate how flow influences dispersal. The total domain is 2.25 km x 2.25 km, with the numbers on the x- and y-axis corresponding to increments of 250 m.

at Maungaroa (Figures 7, 8), individual snapshots of tracer concentration over the simulation period showed that some seep plumes advected south-west on occasions (Figure 9). Lateral plume dispersion varied between seeps, with the tracer released at Pahaua Bank advecting a maximum distance of 104 km between the 1,000 m and 2000 m isobaths, in contrast to the shallower Uruti Ridge site, where dispersal was only 50 km within a year. At all sites elevated dye concentration (>0.25%) was generally constrained within 10 km of the release location, although elevated tracer plumes occasionally extended 50 km from the release site (Figure 9).

The extent of the dye tracer footprint for each seep plume was estimated based on a polygon enclosed by the contour associated with a dye concentration of 0.005. After 1 year the surface area of the individual dye tracer plumes ranged from 2,370 to 4,000 km² (Supplementary Table S2). The modelled vertical plume dispersal extended 1,000 m above the seafloor, resulting in a combined plume volume for the five seeps of 3,500 km³ (Supplementary Table S3). The combination of reversing tidal currents, the relative proximity

of seeps and their co-location primarily at similar depths resulted in overlap of the tracer plumes from individual seeps. This is evident as extended plumes of elevated tracer concentration (>0.25%) beyond 20 km from the seep location on occasions, such as for Maungaroa and Glendhu Ridge in September 2013 (Figure 9C). The ratio of the Gerris model flux estimate to the hydroacoustic flux estimate for Maungaroa (Turco et al., 2022) was then applied to the other seeps in Table 1 to generate a total methane flux for the five seeps of 0.4–3.2 x 10⁶ kg CH₄ y⁻¹. Extrapolating this estimate using the approach of Turco et al. (2022) to all seeps across the Hikurangi Margin (Watson et al., 2020) results in a total methane emission of 1.1–10.9 x 10⁷ kg CH₄ y⁻¹.

3.4 Potential methane oxidation rate

Measurements in 15 bottom water samples obtained within 1 km of the three seep sites (Maungaroa, Uruti Ridge and Glendhu Ridge)



(mmol m⁻³) for Maungaroa for the simulation period 1 January 2013–16 December 2013, with the turquoise, blue and green concentric circles indicating radii of 5 km, 10 km and 20 km from the release point, respectively. The modelled dissolved methane concentration is shown on the colour bar, with isobaths indicated by the grey contours.

generated potential methane oxidation rates of 1.79 ± 0.64 nM d⁻¹ (Supplementary Figure S2). Rates were relatively insensitive to bottom water DIC concentration (mean 2058.1 μ mol kg⁻¹), and also showed no trend with distance from seep location or between the three seeps. In addition, there was no relationship between potential oxidation rate and *in situ* methane concentration, despite three orders of magnitude variation in the latter (Supplementary Figure S2).

3.5 Dissolved δ^{13} C-CH₄

Hydrates release methane with a characteristic low stable isotope $(\delta^{13}C-CH_4)$ value that results from fractionation during bacterial methanogenesis (Reeburgh, 2007), and consequently elevated methane concentrations in bottom water overlying the Hikurangi Margin seeps were associated with low δ^{13} C-CH₄. The three seep sites showed a similar relationship between dissolved methane concentration and $\delta^{13}\text{C-CH}_4$ in waters between 1,000 m depth and the seafloor (Figure 10A), suggesting that there was no significant differences in either the source or processing of methane in bottom waters at the three sites. However, the more extensive dataset obtained at Maungaroa showed greater variability than the other two sites. Analysis of data from this site for the bottom 200 m of water, where most seep methane was retained (Figures 4, 5), showed a linear relationship between methane concentration and δ^{13} C-CH₄ (Figure 10B), as would be expected for lateral mixing of seep-derived methane with background water. However, variability precluded identification of end-member values, and indicated the influence of other factors.

4 Discussion

4.1 Methane emissions

Integration of in situ measurements in the Gerris model and ROMS simulation provided a new approach to estimating methane seep emissions on the Hikurangi Margin. The resulting methane flux for the Maungaroa seep of 3 x 10^5 kg CH₄ y⁻¹ is 4%–12% of a previous estimate of $2.5-7.2 \times 10^6$ kg CH₄ y⁻¹ derived from hydroacoustic data (Turco et al., 2022). These different estimates reflect methodological assumptions and uncertainties, as described in Turco et al. (2022) and other studies (Mitchell, et al., 2022). The hydroacoustic calculation likely overestimates the flux, as diffusive loss of methane during bubble rise (McGinnis et al., 2006) was not accounted for, whereas the Gerris model included uncertainty regarding the spatial dimensions of the methane source (see Section 3.3). The dissolved methane measurements in this study may also have included fluid input, although previous regional studies indicate that fluid release is relatively minor in relation to free gas (Luo et al., 2016). Although emissions from seeps are generally consistent in terms of location and magnitude on the Hikurangi Margin (Law et al., 2010), temporal variation in emissions has been reported (Naudts et al., 2010). Consequently, the emission estimates in this study and that of Turco et al. (2022) provide boundaries on annual methane emissions at Maungaroa of 0.3-7.2 x 10⁶ kg CH₄ y⁻¹. This estimate is high relative to other seeps on Hikurangi Margin, for which annual emissions range from 10^2 to 10^4 kg CH₄ y⁻¹ (Naudts et al., 2010; Luo et al., 2016), but is consistent with a first order estimate derived from hydroacoustic data and residence time for the Opouawe Bank seep located at a shallower depth



FIGURE 8

Time-averaged concentration of dye tracer at the five seep locations on the south-east Hikurangi Margin for the period 1 January 2013–16 December 2013. The blue and green circles show radii of 10 and 20 km, respectively, from the release point at each seep. Dye tracer is shown as a percentage relative to 1.0 at the source location, as indicated in the colour bar, with isobaths indicated by the grey contours.



close to Maungaroa (4.9–9.8 x 10^5 kg CH₄ y⁻¹, Law et al., 2010). The revised methane emission for the entire Hikurangi Margin, of 1.1–10.9 x 10^7 kg CH₄ y⁻¹, is equivalent to 0.8%–7.9% of New Zealand terrestrial emissions (Ministry for Environment, 2022),

and of similar magnitude to the calculated air-sea methane flux for the entire New Zealand Exclusive Economic Zone ($4.1 \times 10^7 \text{ kg y}^{-1}$, based upon 130% surface supersaturation; Law et al., 2010). This highlights the critical role of horizontal dilution and the pelagic



methane filter in minimizing transfer of seep-derived methane to the atmosphere.

4.2 The relative roles of methane oxidation and dispersion

A primary aim of this study was to distinguish the relative contributions of dilution and methanotrophy in determining the distribution and concentration of methane in bottom waters along the Hikurangi Margin. The observed decrease in methane concentration in the vicinity of the seep, and corresponding diffusion and advection in the Gerris model, indicate dilution of two orders of magnitude. The equivalent dilution rate of $\sim 1 \text{ mol m}^{-3} \text{ d}^{-1}$ is considerably larger than the measured potential oxidation rate of $\sim 2 \times 10^{-6} \text{ mol m}^{-3} \text{ d}^{-1}$ and, consequently, advection and dilution are the dominant determinants of methane concentration for the volume of interest and timescales of days in this study.

However, methanotrophy is an important sink for dissolved methane that not only reduces methane concentrations in the deep ocean but also flux to the atmosphere (Mao et al., 2022). The potential contribution of methanotrophy in this study was examined by two independent techniques: measurement of potential oxidation rate in isotope tracer incubations and interpretation of the dissolved δ^{13} C-CH₄ values in bottom waters. The potential oxidation rates were relatively invariant, despite being sampled from within the seep plume through to background locations at all three seep sites (Supplementary Figure S2). This uniformity could reflect tidal retention in bottom waters along the shelf (Figure 9), resulting in a homogenised methanotroph community with uniform oxidation capacity. Although the potential oxidation rate technique relies upon addition of isotopic tracer at concentrations in excess of *in situ* methane concentrations (Leonte et al., 2018), the observed range and low variability are similar to rates reported at other seeps (5.6 \pm 2.3 nM d⁻¹; Leonte et al., 2017; $0-3 \text{ nM d}^{-1}$; Graves et al., 2015; $0-2.5 \text{ nM d}^{-1}$; Weinstein et al., 2016), and marine systems $(0-2.5 \text{ nM d}^{-1}, \text{ Steinle et al., 2015};$ Osudar et al., 2015; 1–10 nM d⁻¹; Mau et al., 2013; 0.01–7.5 nM d⁻¹; Mao et al., 2022). This consistency is surprising, considering the range of environments and ambient methane concentrations, and also the different techniques used to measure methane oxidation. However, it should be noted that rate measurements in this and other studies were predominately carried out at atmospheric pressure, which may alter microbial metabolism and so methane oxidation, relative to that at *in situ* pressure in the deep ocean (Fortunato et al., 2021).

The absence of a relationship between potential oxidation rate and in situ methane concentration, despite three orders of magnitude variation in the latter (Supplementary Figure S2), has been reported in other studies. However, a first-order relationship has also been reported for some hydrate seeps (Valentine et al., 2001), and hydrothermal plumes (de Angelis et al., 1993), with strong evidence from the Deep Water Horizon wellhead blowout in the Gulf of Mexico after which rates rose by four orders of magnitude (Crespo-Medina et al., 2014). Methanotrophic response is slow (Valentine et al., 2001), with rates increasing ~1-2 months after methane addition and reaching a maximum after 3-4 months, as confirmed by modelling, oxidation rate and in situ O2 measurements in the Gulf of Mexico (Kessler et al., 2011; Du and Kessler, 2012; Crespo-Medina et al., 2014). This may reflect slow adjustment of methanotroph metabolism or alternatively adaptation at the community level to elevated methane, which may account for the absence of a relationship between oxidation rate and in situ dissolved methane concentration (Supplementary Figure S2). Alternatively, this lag may reflect other controls of methane oxidation (Steinle et al., 2016), as suggested by shorter response times to methane addition at near-seep sites relative to sites at distance (Chan et al., 2019). As dilution was excluded in the Chan et al. study, the results suggest that near-seep methanotrophs were primed by prior exposure to elevated methane, or alternatively availability of a limiting factor such as trace metals or macronutrients (De Angelis et al., 1993; Beal et al., 2009).

The relatively constant potential oxidation rate across three orders of magnitude in dissolved methane concentration in this

study resulted in a range in turnover time of 6-1,000 days, consistent with other systems (7-560 days, De Angelis et al., 1993; 3-1,000 days; Osudar et al., 2015; 500 days to decades; Valentine et al., 2001). However, shorter turnover times have been recorded at seeps where mixing and circulation is constrained (0.3-3.7 days, Leonte et al., 2017; 1.3-24.0 days; Weinstein et al., 2016). Indeed, residence time is an important determinant of methane oxidation rate, with regional geomorphology, seasonal and water mass variation influencing methanotroph distribution, abundance and activity via dilution and current flow (Steinle et al., 2016; Gründger et al., 2021). The Maungaroa seep is located on a relatively flat ridge on the Hikurangi Margin (Turco et al., 2022), without canyons or seafloor morphology that might enhance regional retention and entrainment of bottom water, and so accumulation of dissolved methane, and thus methane oxidation, might be expected to be low. Although this is supported by the linear relationship between methane concentration and δ^{13} C-CH₄ in bottom waters, other influences are also apparent (Figure 10B). This may reflect tidal circulation and entrainment of methane in the vicinity of the seep, as visualized in the Gerris and ROMS model simulations (Figures 6 and 9), with fresh seep methane of lower δ^{13} C-CH₄ mixing with recirculated "older" methane with higher δ^{13} C-CH₄ values due to oxidation. Despite potential methane oxidation rates being low (Supplementary Figure S2) relative to mixing, local retention could enhance methanotroph adaptation to elevated methane, resulting in higher δ^{13} C-CH₄ values. Conversely, the variability at Maungaroa in Figure 10B may arise due to methane input from other regional seeps, as illustrated in the ROMS simulation in Figure 9. Methane release from hydrates will not have a uniform regional isotopic value but will vary between sites. In addition, thermogenic methane input has been reported for the Hikurangi Margin (Henrys et al., 2009), which could be a source of methane with higher δ^{13} C-CH₄. Although tidal circulation of bottom waters may increase methane oxidation and mixing with methane from other sources this also confounds determination of their relative contribution to local methane dynamics.

4.3 Potential influence of methane emissions on water column biogeochemistry

Methane oxidation provides a source of energy and C for methanotroph growth (Hanson and Hanson, 1996), and ultimately produces CO₂ whilst consuming O₂. This is reflected in the elevated O₂ consumption in seafloor seep communities relative to non-seep benthos (Boetius and Wenzhöfer, 2013), that arises directly from methanotrophy and indirectly from chemosynthetically-supported benthic communities. However, there was no evidence of the Hikurangi Margin seeps influencing bottom water O2 concentrations in either the mooring data (Supplementary Figure S1) or CTD profiles (data not shown), and the measured potential methane oxidation rates would have negligible effect on bottom water O2. If the total methane release by the five seeps was completely oxidized within the modeled plume volume (90% of dye tracer, ~1,000 km³) within 1 year, the decrease in dissolved O_2 concentration would be <0.3%. This assumes utilization of 2 mol of O2 during methanotrophy and so conversion of all methane-fueled biomass to CO₂ (Chan et al., 2019), with resulting CO₂ production of <0.2 µmol L⁻¹ and a corresponding decrease in pH of <0.002 relative to a background of 7.87. This minor influence is consistent with observations at the Hudson Canyon seeps, where aerobic methane oxidation accounted for only 0.3% of observed changes in DIC (Garcia-Tigreros and Kessler, 2018). A previous estimate based upon all Hikurangi Margin seeps suggested a greater impact of methane oxidation on dissolved O_2 and the carbonate system (Turco et al., 2022), but this was derived using a higher seep methane flux and also did not consider dilution and dispersion. Although the current assessment indicates negligible impact of seep methane on bottom water biogeochemistry it does not account for superimposition of plumes from different seeps along the Hikurangi Margin, as shown in Figure 9, which may result in localized impacts. For example, oxidation of seep methane in overlapping plumes could generate localized decreases of 0.53% in dissolved O2 concentration and 0.005 in pH. These localized methanotrophy-driven decreases would compound the current low dissolved O2 saturation in deep water along the Hikurangi Margin, and contribute to ongoing deoxygenation and carbonate saturation horizon shoaling (Feely et al., 2012; Oschlies et al., 2018).

4.4 Characterizing seep dispersal using hydrodynamic models

This study highlights the value of assessing the influence of methane seeps on bottom waters by integrating observations within a hydrodynamic framework. Although bottom water currents confound characterization of dissolved methane distribution around a seep, as illustrated in Figure 6, tidal dispersion and dilution was accounted for in a Gerris model. The ROMS model subsequently simulated methane concentration and distribution using the source function and location generated by the Gerris model but did not include loss to methane oxidation. Although rate measurements indicate that this process is insignificant relative to dispersion, further model development should incorporate methane oxidation, informed by measurements along the Hikurangi Margin to investigate the apparent uniformity in rate and also establish the relative contribution of oxidation to bottom water methane dynamics. The superimposition of multiple seeps could be further examined by applying the ROMS framework to the full suite of seeps along the Hikurangi Margin (Watson et al., 2020). As with many global studies, methane release from seeps is not apparent in surface waters along the Hikurangi Margin, except for some shallow seeps (Higgs et al., 2019); however, there is the potential for transfer via vertical mixing associated with geomorphological features and eddies. Application of the ROMS model framework to the broader shelf region could determine whether this represents a "shortcut" for seep methane transfer to surface waters and flux to the atmosphere along the Hikurangi Margin. The ROMS simulation also suggests that tidal retention and superimposition of plumes may facilitate connectivity between seeps, which should be examined with respect to larval dispersion and connectivity between seep populations (Levin et al., 2016) along the broader Hikurangi Margin. In addition, integration of regional hydrodynamic models with Earth System Models (Rickard et al., 2023) could examine future climate-driven changes in bottom water and the associated biogeochemical impacts along the Hikurangi Margin.

5 Conclusion

Integration of methane data in a ROMS simulation established the methane source strength of five cold seeps on the south-eastern slope of the Hikurangi Margin. Although the combined methane plumes dispersed over a volume of 3,500 km³ and depth of 1,000 m over a year, elevated concentrations were largely restricted to bottom waters 20 km downstream of the seep location. Potential methane oxidation rates were low and invariant in bottom waters, regardless of dissolved methane concentration, indicating slow turnover of dissolved methane. Despite dilution being orders of magnitude greater than aerobic oxidation, δ^{13} C-CH₄ values showed some variability in bottom waters, reflecting a potential contribution of methane oxidation or other sources. The total methane release was revised to $0.4-3.2 \ge 10^6 \text{ kg CH}_4 \text{ y}^{-1}$ for the five seeps, and 1.1-10.9 $x 10^7$ kg CH₄ y⁻¹ for the entire Hikurangi Margin, equivalent to the estimated air-sea flux for the entire Exclusive Economic Zone but less than 8% of New Zealand terrestrial emissions. Oxidation of total annual emissions from the five seeps would not significantly affect bottom water O₂, CO₂ and pH; however, superimposition of plumes from different seeps could support localised maxima in methane and methanotrophy in bottom waters. Integration of methane concentration within a hydrodynamic framework provides insight into the relative roles of dilution and oxidation in determining dissolved methane concentration in bottom waters, and also an approach for assessing the contribution to and sensitivity of cold seeps to climate change.

Data availability statement

The datasets presented in this study can be found in online repositories. The names of the repository/repositories and accession number(s) can be found below: Data and animations of the seep dispersal are available at https://zenodo.org/records/10373095.

Author contributions

CL: Conceptualization, Data curation, Formal Analysis, Funding acquisition, Investigation, Methodology, Project administration, Resources, Writing-original draft, Writing-review and editing. CC: Data curation, Investigation, Methodology, Software, Visualization, Writing-review and editing. AM: Data curation, Investigation, Methodology, Resources, Writing-review and editing. SB: Methodology, Resources, Writing-review and editing. JB: Formal Analysis, Investigation, Methodology, Resources,

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Writing-review and editing. GR: Data curation, Formal Analysis, Investigation, Methodology, Software, Validation, Visualization, Writing-review and editing.

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Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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Supplementary material

The Supplementary Material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/feart.2024. 1354388/full#supplementary-material

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