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# Temperature-dependent carbon isotope fractionation in coccolithophores

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**Introduction:** The stable carbon isotope ratio of long-chain alkenones produced by marine haptophyte phytoplankton has often been used to estimate past variations in atmospheric CO<sub>2</sub> throughout the Cenozoic. However, previous experimental studies and surveys of alkenones from surface sediment and suspended particulate matter document additional environmental and physiological influences on carbon isotopic fractionation in alkenones.

**Methods:** To clarify the non-CO<sub>2</sub> effects on the alkenone carbon isotope fractionations, an important alkenone producer, *Gephyrocapsa oceanica*, was cultured in laboratory. To separate effects of different environment parameters, *G. oceanica* was grown in continuous cultures under a matrix of environmental conditions in order to explore the influence of temperature independently of CO<sub>2</sub>(aq). Through careful manipulation of the media carbon system, we can control the variation of the media CO<sub>2</sub>(aq) independently of temperature solubility. Carbon isotope fractionations from alkenones, coccolith, and particulate organic carbon were measured from this steady state system.

**Results and Discussion:** We find  $\epsilon_p$  in alkenones and particulate organic carbon inversely correlates with temperature, and temperature affects  $\epsilon_p$  more strongly than CO<sub>2</sub>(aq). The magnitude of the temperature effect can be explained by higher growth rates at warmer temperatures with a similar growth rate dependence as observed in previous cultures in which growth rate was regulated by other factors. Where the past temperature influence on growth rate could be constrained using the U<sup>K</sup><sub>37</sub> alkenone index in the same samples, our finding offers an approach to deconvolve an important physiological factor affecting ancient alkenones  $\epsilon_p$ , and may therefore improve past pCO<sub>2</sub> estimates.

## KEYWORDS

carbon isotopic fractionation ( $\epsilon_p$ ), Climate proxy, alkenone, coccolithophore, paleo-CO<sub>2</sub>, *Gephyrocapsa oceanica*, continuous cultivation

## 1 Introduction

Carbon isotope fractionation has been documented in phytoplankton for over half a century (Deuser et al., 1968). Early cellular models, inspired by similar models for land plants (Farquhar et al., 1982), considered a fractionation control of phytoplankton cells in which the diffusive carbon flux that entered the cell was partly fixed into organic carbon, while subject to fractionation through the Rubisco enzyme. In such a system it was shown that the overall photosynthetic fractionation ( $\epsilon_p$ ) was dependent on the CO<sub>2</sub> aqueous (CO<sub>2</sub>(aq)) supply, the phytoplankton growth rate, the cells surface area to volume ratio, and

the cell permeability to CO<sub>2</sub> (Rau et al., 1996). Since then, carbon isotope fractionation has been used to estimate growth rates of living photic zone phytoplankton (Laws et al., 1995; 1997), or to estimate trophic levels and animal migration (Magozzi et al., 2017), but the widest use has been to estimate variations in atmospheric CO<sub>2</sub> in the geological past (see review by Pagani et al., 2011; Rae et al., 2021).

Biomarkers such as alkenones have been an appealing class of stable lipids for carbon isotopic analysis of both water column and sediments because their production is restricted to certain haptophytes of the Isochrysidales order. In particular, marine haptophytes with Reticulofenestrinid coccoliths are the dominant alkenone producers in the modern ocean and their alkenone profile falls into the Group III haptophyte classification of Longo et al. (2016). This group includes species such as *Emiliania huxleyi*, *Gephyrocapsa oceanica*, and *Gephyrocapsa muelleriae*, which allows for better constraining cell geometry by their carbonate fossils, coccoliths. Therefore, the use of alkenones attributable to this select group can potentially limit the known influence of the surface area/volume ratio on fractionation.

Genomics studies have concluded that these alkenone producers—species of the *Gephyrocapsa* genus, *E. huxleyi* and some Reticulofenestrinids—are genetically similar and monophyletic (Bendif et al., 2019). In the modern ocean, *E. huxleyi* is dominant in subtropical to subpolar settings characterized by deep winter mixing and production in the spring, while *G. oceanica* occurs at temperatures between 12°C and 27°C but favors temperatures above 20°C such as in low latitude upwelling settings (Sett et al., 2014; von Dassow et al., 2021). *E. huxleyi* emerged 0.27 million years ago, and diverse *Gephyrocapsa* strains are recognized from the fossil record over more than 5 million years (Bown, 1998).

Yet, even when the alkenone-producing organism and its cell geometry are well constrained, previous studies from suspended particulate matter (SPM) in the ocean (Bidigare et al., 1997), surface sediments (Benthien et al., 2007; Hernández-Almeida et al., 2020; Phelps et al., 2021b), and laboratory experiments have demonstrated significant non-CO<sub>2</sub>-related effects on alkenone  $\epsilon_p$ . Indeed, alkenone samples obtained from SPM produced at a range of CO<sub>2</sub>(aq) in the modern surface ocean, demonstrated that the CO<sub>2</sub>(aq) concentration is not the dominant component of variation in alkenone  $\epsilon_p$ . Lower than expected  $\epsilon_p$  in SPM of the higher latitudes and upwelling regions were initially attributed to higher growth rates in these regions (Bidigare et al., 1997). However, subsequent models of coccolithophorid growth rate in the ocean (Krumhardt et al., 2017; Nissen et al., 2018) have simulated significantly lower growth rates in the high latitude regions compared to low latitude regions. This is because at the low temperatures of higher latitude regions, models simulate lower maximum growth rates for a given degree of nutrient or light limitation. This paradox in explanations has been reviewed in (Hernández-Almeida et al., 2020; Phelps et al., 2021b). Experimental cultures of alkenone producers have evaluated the effects of varying CO<sub>2</sub>(aq) on fractionation. In some studies changing nitrogen availability (Bidigare et al., 1997), varying irradiance, or day length have been employed to modulate growth rate variability (Rost et al., 2002; Tchernov et al., 2014). Given a strong influence of light, growth rate has been considered to have negligible effect on  $\epsilon_p$  in some formulations (Phelps et al., 2021b). To date, the effect of temperature-regulated growth rate

variations on  $\epsilon_p$  has not been rigorously evaluated. One published experiment varied temperature but maintained a constant pCO<sub>2</sub>, leading to an inverse variation of CO<sub>2</sub>(aq) and growth rate due to temperature-controlled CO<sub>2</sub> solubility (Tchernov et al., 2014).

To better characterize the influence of temperature on  $\epsilon_p$ , we grew continuous cultures of *G. oceanica*. We used a turbidostat system, in which the CO<sub>2</sub>(aq) can be manipulated independently of the temperature solubility effect through manipulating the partial pressure of CO<sub>2</sub> bubbled into the media as well as the media pH and DIC. Our continuous cultures will therefore permit for the first time, the separation of the influence of CO<sub>2</sub>(aq) and temperature on  $\epsilon_p$ . We find temperature to have a significant effect on  $\epsilon_p$  of a magnitude consistent with control by phytoplankton growth rate. This confirms that standard temperature-dependent growth rate formulations for phytoplankton (Fielding, 2013) apply to the fractionation process. Since multiple temperature proxies exist, some derived directly from alkenones, our results can thus contribute to improved estimations of past changes in phytoplankton growth rates and more robust pCO<sub>2</sub> reconstructions from alkenone  $\epsilon_p$ .

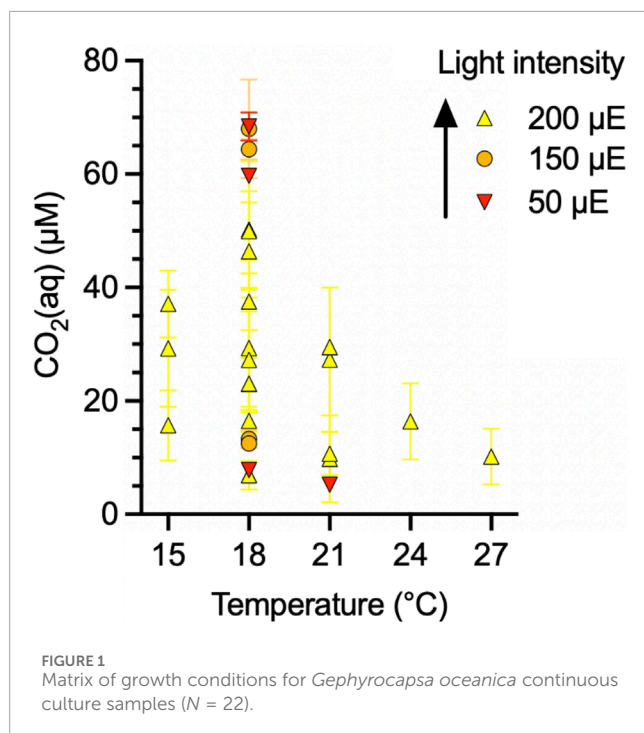
## 2 Materials and methods

### 2.1 Cell culturing

Cultures of the monoclonal coccolithophore strain *G. oceanica* RCC1303 (Roscoff Culture Collection) were grown in 1 or 3-L flat panel photobioreactors (PBR; FMT150, Photon Systems Instruments) with cool-white LED panels. The setup for continuous cultivation was the same as described in Zhang et al. (2022). The PBRs were equipped with a Peltier element at the bottom to maintain a constant temperature ( $\pm 0.1^\circ\text{C}$ ), with tested temperatures ranging from 15°C to 27°C. A sinusoidal diurnal cycle of 16:8 light/dark hours was set at different intensity maxima, with photosynthetically active radiations (PAR) between 50–200  $\mu\text{E}$  ( $\mu\text{mol photon m}^{-2} \text{s}^{-1}$ ). Light intensity was checked with an external quantum photometer (LI-COR) ( $\pm 1 \mu\text{E}$ ). Shading was estimated by measuring the light on the opposite wall of the LED panel past the vessel with the turbid culture and accounts for less than 30% of maximum light intensity for the employed low threshold of cell concentration. In total, the cultures were grown at 22 different combinations of temperature, CO<sub>2</sub>(aq), and light (Figure 1).

In order to maintain the cultures in log-phase, the PBRs were operated in turbidostat mode through continuous measurement of the optical density at 680 nm (OD680). Automatic dilution would be triggered when the OD680 exceeded a set threshold equivalent to  $1\text{--}2 \times 10^5$  cells mL<sup>-1</sup>. Above this threshold, a peristaltic pump would initiate dilution inflow pulses equivalent to <3% vessel volume into the PBR. Equivalent outflow volumes were discharged into a waste bottle through an outlet tubing placed at  $\approx 10\%$  volume air headspace. Gas overpressure within the air headspace would facilitate liquid outflow and a constant vessel liquid volume was maintained. A gently rotating magnet was placed in the middle of the PBR and, with the aid of bubbling and daily manual resuspension, cells were kept in suspension.





## 2.2 Maintaining the carbonate system

Cells were kept in artificial seawater (ASW) K/2 media (Kester et al., 1967; Keller et al., 1987). To better characterize the water chemistry, Tris buffer was not added. Media was adjusted to yield an initial dissolved inorganic carbon (DIC) concentration of 2.05 or 4.10  $\mu\text{mol kg}^{-1}$  through addition of 2.33  $\text{mmol kg}^{-1}$   $\text{NaHCO}_3$  and 1.74  $\text{mmol kg}^{-1}$   $\text{HCl}$  (double amounts for the higher DIC). The lower DIC concentration was to simulate modern ocean and the higher DIC concentration media was employed to prevent pH from falling below 7.8 during high  $\text{pCO}_2$  bubbling at 2000 ppm and still allow for coccolithophore growth. The CO2SYS excel macro (Pierrot et al., 2006) was used with inputs of pH and DIC, the NBS scale, 35‰ salinity, 7  $\mu\text{M}$  phosphate, and 0  $\mu\text{M}$  silicate to calculate the media's carbonate system.

Continuous pH monitoring was done with an internal pH-probe with an integrated temperature sensor (InPro 3253SG/120/PT1000, Mettler Toledo) inserted into the PBR. The pH of harvested samples was measured externally using a pH-meter (FiveEasy, Mettler Toledo) and calibrated with NIST buffer standards (Mettler Toledo). Variability of the internal and external pH probes was within error ( $\pm 0.1$ ).

After external pH measurement, 50 mL of the harvested sample was centrifuged at 4,000  $g$  for 5 minutes. Aliquots were taken of the supernatant and subsequently analyzed for DIC on an Apollo AS-D1 DIC- $\delta^{13}\text{C}$  Analyzer (Apollo Scitech, United States) coupled to a Picarro G-2131-I Cavity Ringdown Spectrometer (CRDS; Picarro Inc, United States). 3.5 mL of the aliquoted supernatant were reacted with 0.9 mL of a 3%  $\text{H}_3\text{PO}_4$  and 7%  $\text{NaCl}$  solution in the Apollo AS-D1. The DIC concentration was subsequently measured from the released  $\text{CO}_2$ . Each measurement was performed twice for each sample. Seawater reference material from the National Oceanic and Atmospheric Administration (NOAA) batch 186 and two in-house

$\text{NaHCO}_3$  of known concentrations were used for calibration. The coupled Apollo-Picarro instrument had an analytical error of 10  $\mu\text{M}$  for DIC. To confirm that stable DIC concentrations were maintained within the PBR, DIC was measured at cell harvest and at least once the day before.

Carbonate chemistry equilibrium was maintained through pre- and continuous bubbling of the PBR and fresh input media with different compressed air and  $\text{CO}_2$  mixtures. Firstly, an NDC-140  $\text{CO}_2$  scrubber (F-DGSi) was used to remove residual  $\text{CO}_2$  from compressed air. Subsequently, a Gas Mixing System (GMS 150, Photon Systems Instruments) was used to create a mixture with a set  $\text{pCO}_2$ , which flowed into a Mass Flow Controller (MFC; Vögtlin Instruments). Mixtures containing 200–2000 ppm  $\text{pCO}_2$  were used. The MFC regulated and maintained the bubbling rate at 50–100  $\text{mL min}^{-1}$ . Backflow and system overpressure were prevented through a metallic air tank equipped with a safety valve that would trigger above 0.2 bar between the MFC and the fresh media bottle.

After the safety valve, the mixture with set  $\text{pCO}_2$  went through a 0.2  $\mu\text{m}$  gas filter and into the fresh media bottle. Bubbling occurred from the bottom of the bottle to allow for complete humidification and equilibration of the fresh media. Gas from the bottle headspace subsequently went through two more gas filters and a bubble-interrupting valve, which interrupted flow during OD680 measurements, before flowing into a U-shaped sparger at the bottom of the PBR vessel. The gas in the vessel headspace exited the PBR through shared gas- and liquid outflow tubing into the outflow bottle. Gas exited the system through a one-way gas filter valve fitted into the outflow bottle cap.

The  $\text{CO}_2$  concentration of the gaseous mixtures inputted into the PBR were measured using the Picarro CRDS and corrected with certified gas mixtures of known ppm (PanGas and Air Liquide). An aliquot from the GMS150 was used to fill air-tight gas bags and subsequently inserted into the Picarro inlet tube. Stable  $\text{CO}_2$  measurements were considered to have been reached after the inserted gas maintained a  $\text{pCO}_2$  of  $\pm 2$  ppm.

## 2.3 Cell monitoring and growth rate

At least once a day, 1 mL of culture was harvested for cell growth monitoring. Care was taken to harvest at the same hour every day throughout each experiment to minimize cell size variability. Cell density per mL and mean cell diameter were measured with a Multisizer 4e particle counter and sizer (Beckman Coulter). Cell volume per mL was also measured, assuming a spherical shape. As coccolithophores tend to divide during the dark and grow in size during the light period, care was taken to include entire day/night cycles for growth rate ( $\mu$ ) calculations. As such, cell concentrations were measured at 24-h intervals. For an ideal turbidostat,  $\mu$  is the same as the dilution rate ( $D$ ). However, there are several factors in our system that cause differences from an ideal turbidostat: (i) non-homogenous growth due to day/night cycles, (ii) dilution occurred in pulses rather than constant dilution, and (iii) a fraction of cells flocculated and did not grow in suspension with equal exposure to light. Thus, to account for small daily fluctuations in cell

concentration, we adapted the growth rate calculation from Shuler and Kargi, (2008) in Eq. 1:

$$\mu = D + \frac{\ln X_1 - \ln X_0}{t_1 - t_0}, D = \frac{V_O + V_H}{V_{PBR}(t_1 - t_0)} \quad (1)$$

where  $X_0$  is the cell concentration at time 0 ( $t_0$ ),  $X_1$  is the cell concentration after 1 day ( $t_1$ ),  $V_0$  is the outflow volume,  $V_H$  is the volume harvested during sampling, and  $V_{PBR}$  is the PBR vessel liquid volume without in or outflow.

Cell growth in the turbidostat varied considerably between conditions. During our experiments, we experienced difficulties in establishing steady state, continuous cultures after initial inoculation. The method that led to a higher degree of success included halting the magnet stirring and gas bubbling after inoculation. The culture would then be kept in batch mode until the target concentration was reached and, when switching to turbidostat mode, stirring and bubbling were kept at a minimal intensity to homogenize the cultures but still reach the target pH or  $\text{CO}_2(\text{aq})$  levels. Nonetheless, a peak of loose coccoliths was visible in our Coulter counter measurements and microscope observations. Altogether, we believe that during growth inside our PBRs, coccolithophores must be affected by different degrees of shear stress potentially triggering partial cell damage or lysis (Thomas and Gibson, 1990). Given that cultures had  $1.2 \pm 0.9$  loose coccoliths per cell, the mortality rate should be below 10% assuming more than 10 coccoliths per cell.

## 2.4 Elemental analysis (EA-IRMS) of particulate organic carbon and nitrogen (POC and PON)

An aliquot of 50 mL of cultures was snap-frozen in liquid nitrogen and filtered on precombusted quartz fiber filters (QM-A, Whatman). In order to remove the inorganic carbon through conversion of  $\text{CaCO}_3$  into gaseous  $\text{CO}_2$ , the filters were placed inside a desiccator and exposed to acid fumes originating from 50 mL of a 6%  $\text{H}_2\text{SO}_3$  solution. The acid solution was evaporated through maintaining the vacuum below 20 mbar. The desiccator was closed for at least 24 h before reopening to allow for a complete reaction. This procedure was verified to fully remove  $\text{CaCO}_3$  carbon from filters. Subsequently, the filters were dried overnight at 60°C, compacted, and wrapped in tin cups using a manual press. The wrapped filter pellets were first introduced to a ThermoFisher Flash-EA 1112 with a ConFlo IV interface, combusted in an oxidation column at 1,020°C, and then passed through a reduction column at 650°C to produce  $\text{N}_2$  and  $\text{CO}_2$ . These gases were separated by chromatography and transferred to a ThermoFisher Delta V-IRMS through an open split for on-line isotope measurement.

## 2.5 Particulate inorganic carbon (PIC) determination

One to five million cells were filtered in triplicate on 0.8  $\mu\text{m}$  polycarbonate membranes (Whatman Cyclopore), with their ASW flow-through discarded. The residue on the filters were washed with

milli-Q water and stored at  $-20^\circ\text{C}$ . The PIC of the filter residue was dissolved through the immersion and vortexing in 1 mL 2%  $\text{HNO}_3$  solution. The solution was then centrifuged at 11,000 g for 2 min, with cell debris collecting at the bottom while the dissolved PIC went into the supernatant. An aliquot of 400  $\mu\text{L}$  of the supernatant was transferred into Teflon tubes for Inductively Coupled Plasma-Mass Spectrometry analysis (Agilent 8800 ICP-QQQ-MS). The concentration of PIC was calculated through the measured molar concentration of Ca and harvested cell count, and is expressed as  $\mu\text{g CaCO}_3$  per cell.

## 2.6 Chlorophyll analysis

About  $10^6$ – $10^7$  cells were harvested from the PBRs in duplicate. Cells were pelleted at 4,000 g for 5 min, the supernatant discarded, and the pellet snap-frozen in liquid nitrogen and stored at  $-20^\circ\text{C}$ . Pellets were thoroughly resuspended in 1 mL pure methanol and transferred to 1.5 mL tubes. Pigment extraction was then allowed for at least 1 hour at  $-20^\circ\text{C}$ . Tubes were centrifuged for 5 min at 11,000 g, with supernatants transferred into polystyrene cuvettes for spectrophotometer readings at 470, 652.4, 665.2, and 750 nm. The empirical formula in Eq. 2 was employed to quantify chlorophyll (Lichtenthaler, 1987):

$$\text{Chlorophyll} \left( \frac{\mu\text{g}}{\text{mL}} \right) = 19.71 (\text{Abs}_{652.4} - \text{Abs}_{750}) + 4.44 (\text{Abs}_{665.2} - \text{Abs}_{750}) \quad (2)$$

## 2.7 Coccolith carbon and oxygen isotope analysis

After the methanol extraction of the previous section, cell pellets were dried at 60°C overnight. Aliquots of 300  $\mu\text{g}$  of the remaining coccolith powder were transferred into glass vials, then closed airtightly and flushed with He. Calcite was dissolved with five drops of orthophosphoric acid at 70°C in the same Gas Bench system. The reacted PIC  $\delta^{13}\text{C}$  and  $\delta^{18}\text{O}$  were measured and calibrated with in-house standards that were calibrated with the above-mentioned international standards. The carbon isotope fractionation between coccoliths (or PIC) and DIC was calculated as in Eq. 3:

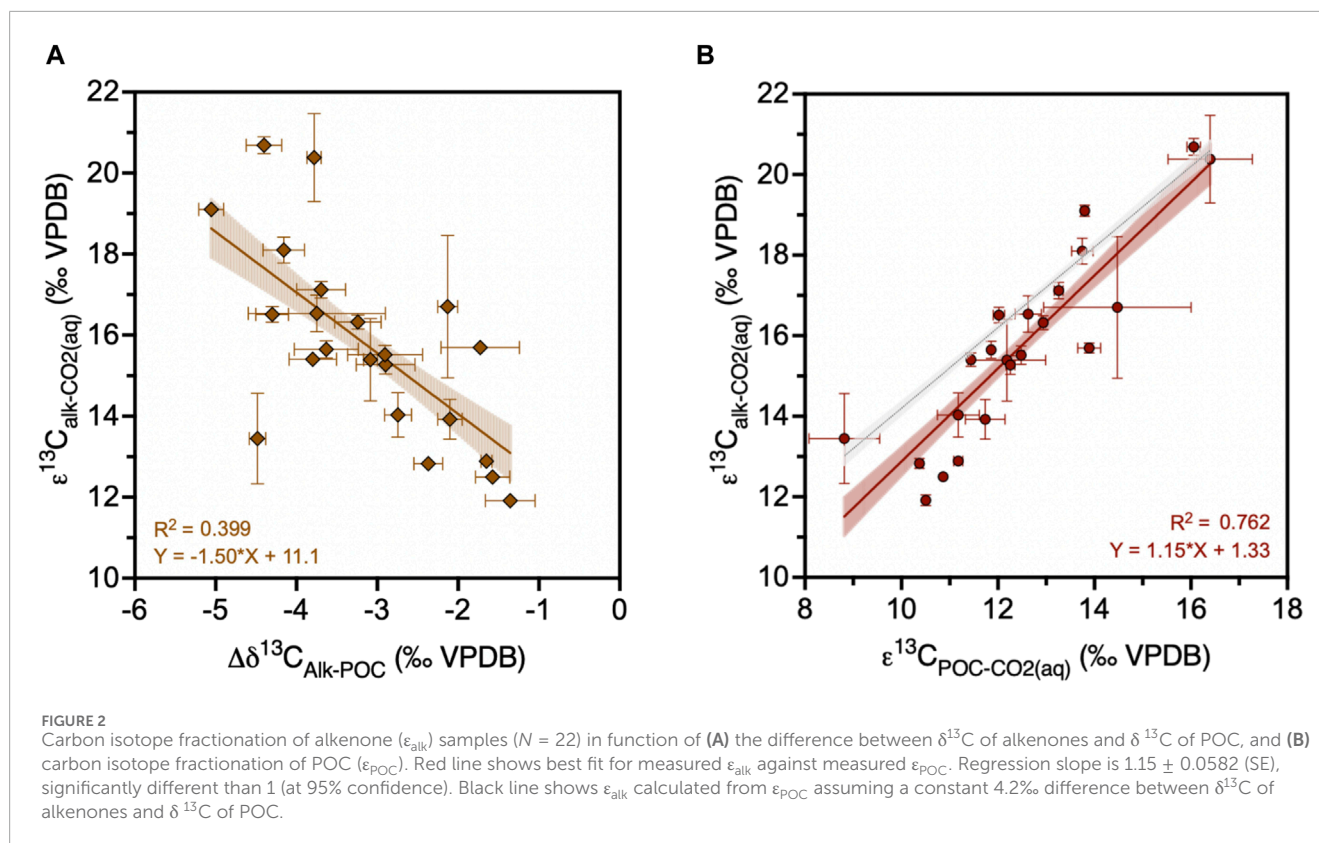
$$\epsilon_{\text{cocco}} = 1000 \left( \frac{\delta^{13}\text{C}_{\text{DIC}} + 1000}{\delta^{13}\text{C}_{\text{coccolith}} + 1000} - 1 \right) \quad (3)$$

Water  $\delta^{18}\text{O}$  was analyzed on a Picarro L2120-*i* cavity ringdown spectrometer using reference water materials: SLAP2, GRESP and VSMOW2. Each water sample without cells was injected ten times and first two to four injections were discarded to avoid effects of instrumental memory. Accuracy ( $2\sigma$ ) for  $\delta^{18}\text{O}$  was 0.2‰, based on reproducibility of six measurements of USGS W-67400-S international standard.

The oxygen isotope fractionation between coccoliths and water was calculated as follows in Eq. 4:

$$\epsilon^{18}\text{O}_{\text{cocco-water}} = 1000 \left( \frac{\delta^{18}\text{O}_{\text{water}} + 1000}{\delta^{18}\text{O}_{\text{coccolith}} + 1000} - 1 \right) \quad (4)$$

where  $\delta^{18}\text{O}$  (‰ VSMOW) =  $1.03092 * \delta^{18}\text{O}$  (‰ VPDB) + 30.92.



## 2.8 Lipid extraction and quantification

Duplicate or triplicate samples of 100–200 mL of *G. oceanica* were harvested on precombusted glass fiber filters, snap-frozen in liquid nitrogen, and stored at  $-20^{\circ}C$ . Extraction and quantification of the total lipid fraction was performed using an isopropanol and methyl tert-butyl ether (MTBE) protocol (Légeret et al., 2016). One mL of preheated isopropanol containing 0.01% butylated hydroxytoluene was added to a vial containing the filter in order to unfreeze the pellet. Vortexing and sonication then homogenized and suspended the pellet into the solution. Immediate heating at  $85^{\circ}C$  for 10 min caused cells lysis and inactivation of lipases. The solution was then allowed to cool down to room temperature. To account for extraction yield and quantification of lipids, 10  $\mu g$  of standard C27 alkenone (14-heptacosanone) (Sigma) were added. Then to allow for phase separation, 3 mL of MTBE were added, the solution was vortexed, 1 mL of DI water was added, and centrifuged for 5 min at 3,200 g. The upper organic phase, which is free of cell debris and the filter, was transferred to a clean glass tube. Re-extraction with an additional 1 mL MTBE and pooling of the organic phases gave higher extraction yields. The organic solvents were evaporated in a bath of  $\leq 30^{\circ}C$  and under a gentle  $N_2$  (g) stream. Lipids were re-dissolved in hexane or toluene for gas chromatographic (GC) analysis of alkenones.

The alkenone compositions of the continuous cultures were quantified through gas chromatography-flame ion detection (GC-FID). Alkenones were separated with a Thermo GC-FID (Trace 1,310) on a VF-200 ms of 60 m x 0.25 mm, 0.25  $\mu m$  film column. Samples were injected with a TriPlus RSH Autosampler to the

column using the PTV inlet in pulsed splitless mode with  $H_2$  as carrier gas. The detector temperature was  $320^{\circ}C$  with the following oven settings:  $120^{\circ}C$  for 1 min, ramp up to  $200^{\circ}C$  at  $40^{\circ}C/min$ , and up to  $300^{\circ}C$  at  $5^{\circ}C/min$ , then hold for 15 min, heat up to  $320^{\circ}C$  at  $10^{\circ}C/min$  and hold for 1 min. Alkenones were quantified with the internal standard C27 alkenone accounting for extraction yields.

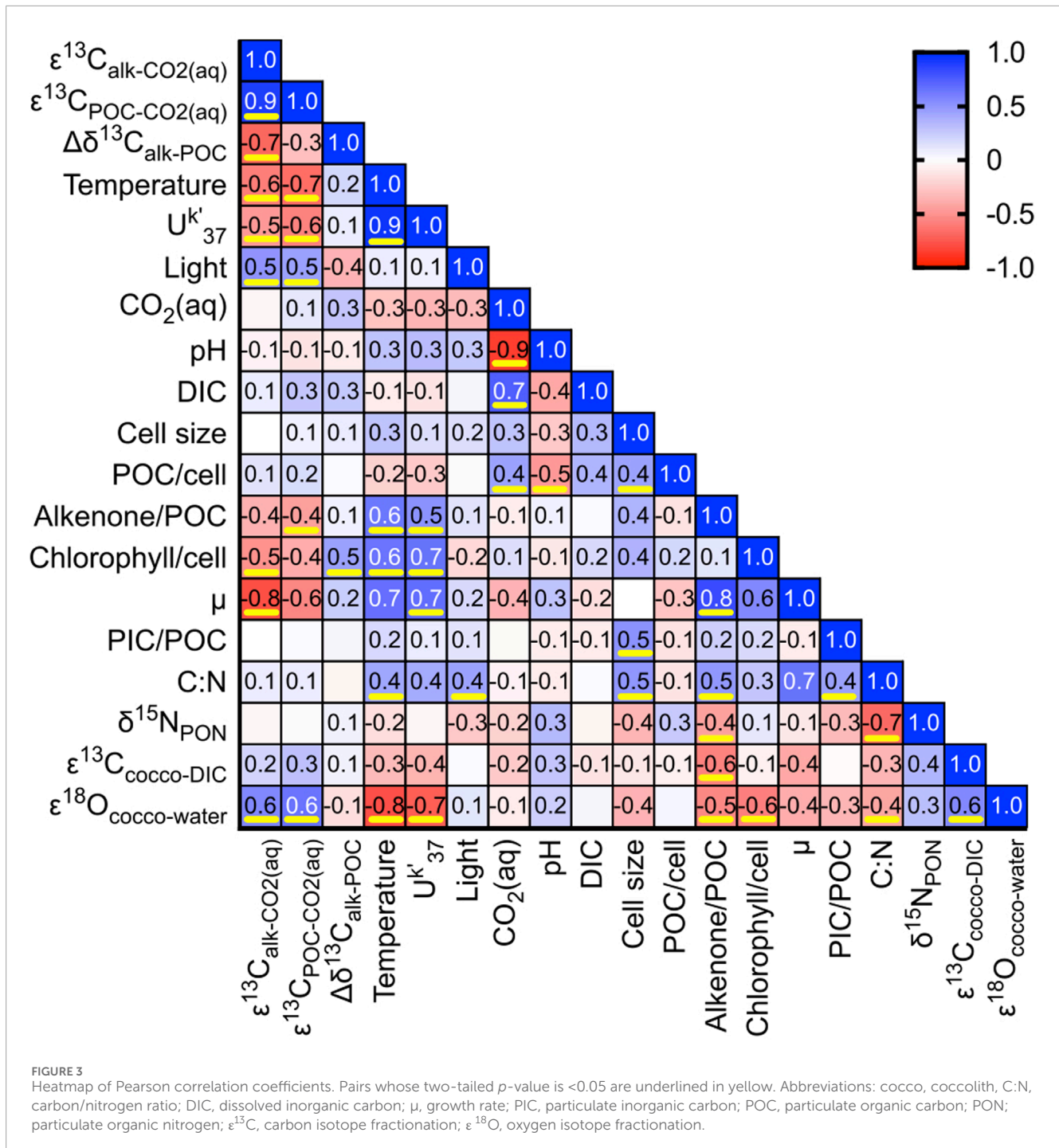
## 2.9 Alkenone isotope analysis

The stable carbon isotopic compositions were determined using GC-isotope ratio mass spectrometry (GC-IRMS) consisting of a Thermo Trace 1310 GC detector and a combustion reactor oxidating alkenones to  $CO_2$  (GC Isolink II). Alkenones were separated in a Rtx 200, 0.32 mm ID, 0.5  $\mu m$ , 60 m column (Restek). Injection was splitless with helium as a carrier gas at a constant flow of  $2 mL min^{-1}$  and detector temperature of  $320^{\circ}C$ . Oven temperature cycle was set at  $90^{\circ}C$  for injection to be ramped after 1 min to  $250^{\circ}C$  at  $25^{\circ}C min^{-1}$ ,  $1^{\circ}C min^{-1}$  to  $305^{\circ}C$ , and finally to  $320^{\circ}C$  at  $10^{\circ}C min^{-1}$ . The temperature was kept isothermal for 10 min. Reported C isotope values were corrected with an *n*-alkane mixture (A7, Dr. Arndt Schimmelmann, Indiana University). Each biological replicate was injected three times and mean values were reported in ‰ relative to VPDB for  $\delta^{13}C$ .

The carbon isotope fractionation between alkenones and  $CO_2(aq)$  ( $\epsilon_{alk}$ ) was calculated using Eq. 5:

$$\epsilon_{alk} = 1000 \left( \frac{\delta^{13}C_{CO_2(aq)} + 1000}{\delta^{13}C_{alkenones} + 1000} - 1 \right) \quad (5)$$





Similarly, the carbon isotope fractionation between POC and  $\text{CO}_2(\text{aq})$  ( $\epsilon_{\text{POC}}$ ) was calculated using Eq. 6:

$$\epsilon_{\text{POC}} = 1000 \left( \frac{\delta^{13}\text{C}_{\text{CO}_2(\text{aq})} + 1000}{\delta^{13}\text{C}_{\text{POC}} + 1000} - 1 \right) \quad (6)$$

The  $\delta^{13}\text{C}_{\text{CO}_2(\text{aq})}$  in Eqs 5 and 6 was calculated by measured DIC carbon isotope ratios ( $\delta^{13}\text{C}_{\text{DIC}}$ ) and temperature (Rau et al., 1996) with Eq. 7:

$$\delta^{13}\text{C}_{\text{CO}_2(\text{aq})} = \delta^{13}\text{C}_{\text{DIC}} + 23.644 - \frac{9701.5}{T} \quad (7)$$

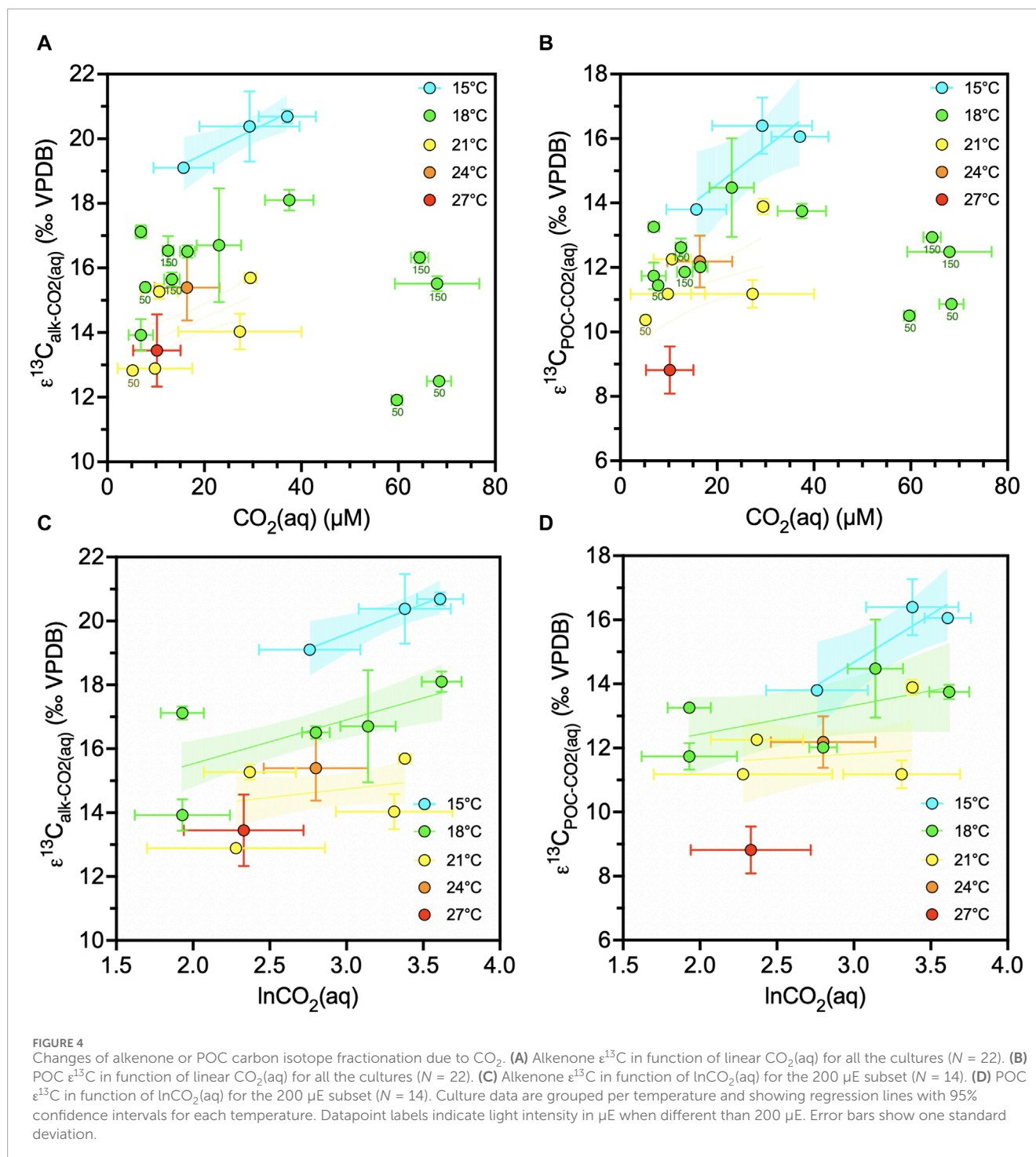
where  $T$  is the temperature in K.

### 3 Results

In our dataset of 22 cultures at different  $\text{CO}_2(\text{aq})$ , temperature and light conditions (Figure 1), the  $\delta^{13}\text{C}$  of alkenones, POC and coccoliths span from  $-34.5$  to  $-24.4\text{‰}$ ,  $-30.8$  to  $-19.9\text{‰}$ , and  $-6.0$  to  $-1.4\text{‰}$ , respectively (Dataset 1). The difference between  $\delta^{13}\text{C}$  of POC and  $\delta^{13}\text{C}$  of alkenone is not constant as it varies from  $-1.4$  to  $-5.1\text{‰}$  (Figure 2). Their relationship can be estimated with the regression in Eq. 8:

$$\epsilon_{\text{alk}} = 1.15 \epsilon_{\text{POC}} + 1.33 \quad (8)$$

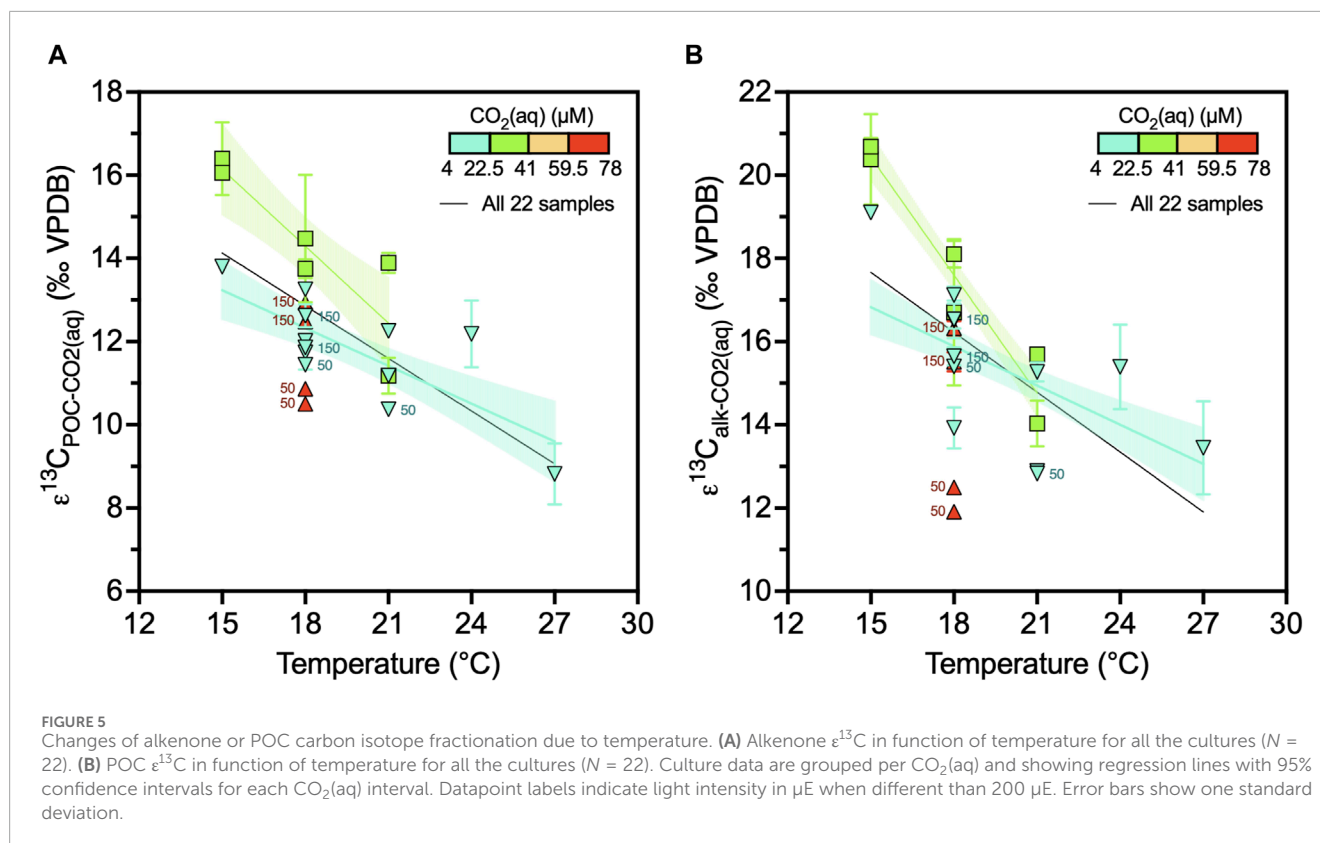




The isotope measurements are analyzed together with environmental and physiological parameters and plotted in a correlation matrix (Figure 3). The  $\delta^{13}\text{C}$  difference between alkenones and POC ( $\Delta\delta^{13}\text{C}_{\text{alk-POC}}$ ) is not correlated with the amount of POC per cell nor its allocation into alkenones (alkenones/POC). However, the  $\Delta\delta^{13}\text{C}_{\text{alk-POC}}$  is reduced when the carbon isotopic fractionation into alkenones ( $\epsilon_{\text{alk}}$ ) and POC ( $\epsilon_{\text{POC}}$ ) is lower, indicating a greater divergence of isotopic composition in conditions that cause greater carbon isotopic fractionation. Both  $\epsilon_{\text{alk}}$  and  $\epsilon_{\text{POC}}$

are similarly controlled: both have a positive correlation with light and a negative correlation with temperature,  $U_{37}^K$ , alkenones/POC, chlorophyll/cell, and  $\mu$ . Because higher light levels correlate with an increased fractionation, the  $\Delta\delta^{13}\text{C}_{\text{alk-POC}}$  is also moderately correlated with light. Surprisingly, our correlation matrix did not identify a positive linear correlation between  $\epsilon_{\text{alk}}$  or  $\epsilon_{\text{POC}}$  with CO<sub>2</sub>(aq).

When we group cultures by temperature, at any given temperature there is a positive correlation between  $\epsilon_{\text{alk}}$  (and  $\epsilon_{\text{POC}}$ )



and  $\text{CO}_2(\text{aq})$  within the  $\text{CO}_2(\text{aq})$  range of 5.2–37.5  $\mu\text{M}$ , but not for the four cultures grown at the lowest pH conditions and higher  $\text{CO}_2(\text{aq})$  range of 59.7–68.4  $\mu\text{M}$  (Figure 4AB). When examining the subset of the 14 cultures at 200  $\mu\text{E}$  light, the positive correlation of  $\epsilon_{\text{alk}}$  (and  $\epsilon_{\text{POC}}$ ) with  $\text{CO}_2$  is also evident for a given temperature (Figure 4CD). Since  $\text{CO}_2(\text{aq})$  solubility is affected by temperature, and both parameters show opposite relationships with  $\epsilon_{\text{alk}}$  (and  $\epsilon_{\text{POC}}$ ), we isolate the temperature effect in Figure 5. Indeed, at similar  $\text{CO}_2(\text{aq})$ , higher temperatures lead to lower  $\epsilon_{\text{alk}}$  (and  $\epsilon_{\text{POC}}$ ). In our culture subset with available growth rate data ( $N = 9$ ),  $\mu$  has a significantly negative correlation with  $\epsilon_{\text{alk}}$  (Pearson  $r = -0.79$ ,  $p = 0.012$ ) (Figure 6).

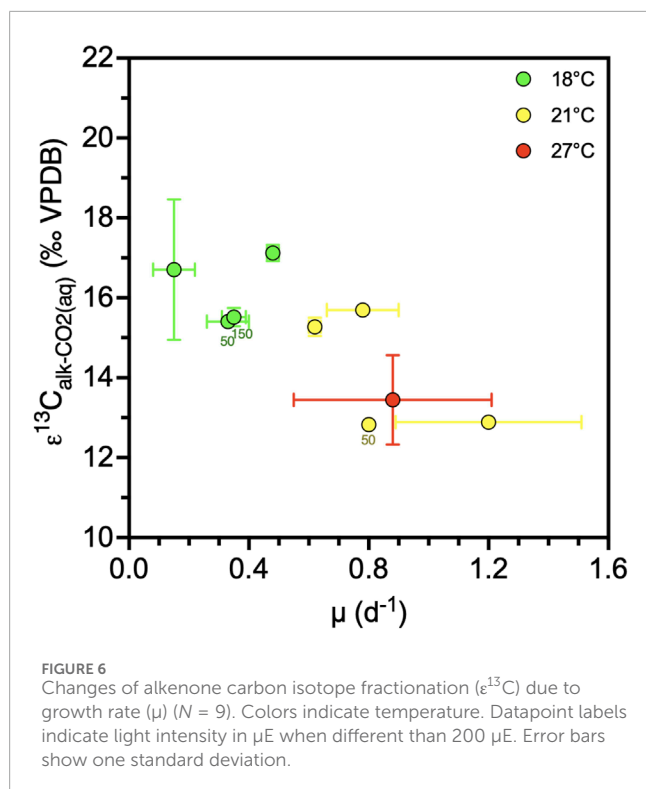
After identifying the complex non-linearity and multivariate effects of temperature, light, and  $\text{CO}_2(\text{aq})$  on  $\epsilon_{\text{alk}}$  or  $\epsilon_{\text{POC}}$ , we run several combinations of multiple linear regressions to obtain the highest  $R^2$  that gives the best prediction of carbon isotope fractionations from environmental variables (shown in bold in Table 1). The highest  $R^2$  regression using the cultures from the 5.2–37.5  $\mu\text{M}$  range of  $\text{CO}_2(\text{aq})$  (i.e.,  $N = 18$ ) can predict up to 80% of variance. In this model, temperature is the most significant influence on  $\epsilon_{\text{alk}}$  (and  $\epsilon_{\text{POC}}$ ), with the lowest  $p$ -value, followed by  $\text{CO}_2(\text{aq})$  or  $\ln(\text{CO}_2(\text{aq}))$ . In this data subset, the influence of light could not be rigorously evaluated due to the dominance of experiments at 150 and 200  $\mu\text{E}$  and only two at a significantly lower light intensity (50  $\mu\text{E}$ ), and therefore there is a low statistical significance of light on  $\epsilon_{\text{alk}}$  and  $\epsilon_{\text{POC}}$ .  $\epsilon_{\text{cocco}}$  varies from  $-1.2\%$  to  $1.1\%$ , and is not well predicted by temperature, light and  $\text{CO}_2(\text{aq})$  ( $R^2 = 0.29$ ). A linear combination of temperature, light, and  $\text{CO}_2(\text{aq})$  also provides only a weak prediction of  $\mu$  ( $R^2 = 0.42$ ) (Table 1), potentially because the

studied variable range flanks the point of maximal growth rate, and because the range of pH and DIC also influence the energy required for calcification and affect growth rate.

## 4 Discussion

### 4.1 Fractionation among alkenones, POC, and coccoliths

Our results suggest a more variable  $\Delta\delta^{13}\text{C}_{\text{alk-POC}}$  compared to previous studies that suggested a constant  $-4.2\%$  offset (Popp et al., 1998). Our maximum difference ( $-5\%$ ) was similar to previous reports (Figure 2), but in our experiments, the  $\Delta\delta^{13}\text{C}_{\text{alk-POC}}$  was significantly smaller when the carbon isotopic fractionation between alkenones and  $\text{CO}_2(\text{aq})$  was lower (Pearson  $r = -0.69$ ,  $p = 0.0004$ ). It has been previously described that alkenones or lipids in general are lower in  $\delta^{13}\text{C}$ , i.e., more  $^{13}\text{C}$ -depleted, than other cellular components (Popp et al., 1998). Although we have not quantified the relative abundance of lipids, previous studies suggest that alkenones reflect a high proportion of storage lipids (Bakku et al., 2018). However, we find no significant correlation between the  $\Delta\delta^{13}\text{C}_{\text{alk-POC}}$  and the relative abundance of alkenones per POC. Potentially, other biomass composition changes such as the amount of carbohydrates, which we have not measured, or other parameters affecting alkenone biosynthesis and function can lead to variability in the fractionation. While both POC and alkenone  $\delta^{13}\text{C}$  show similar relationships with  $\text{CO}_2(\text{aq})$  and temperature in our study (Figures 4, 5; Table 1), future studies with larger datasets should test



if the quantitative sensitivities of  $\epsilon_{\text{alk}}$  on  $\text{CO}_2$  differ from those of  $\epsilon_{\text{POC}}$  on  $\text{CO}_2$ , which should be accounted for when applying  $\epsilon_{\text{POC}}$  culture calibrations to sedimentary  $\epsilon_{\text{alk}}$  determinations.

Our results suggest a limited range of variation in  $\epsilon_{\text{cocco}}$ ; the  $\delta^{13}\text{C}$  of coccoliths ( $\delta^{13}\text{C}_{\text{cocco}}$ ) is relatively close to the  $\delta^{13}\text{C}$  of DIC ( $\delta^{13}\text{C}_{\text{DIC}}$ ) and shows no systematic change with  $\text{CO}_2(\text{aq})$ . For studies of the past ocean, to calculate  $\epsilon_p$ , the  $\delta^{13}\text{C}_{\text{DIC}}$  has been estimated by measuring the  $\delta^{13}\text{C}$  of planktic foraminifera or by measuring the  $\delta^{13}\text{C}_{\text{cocco}}$ . We find a very strong correlation ( $R^2 = 0.894$ ) between  $\delta^{13}\text{C}_{\text{alk}} - \delta^{13}\text{C}_{\text{cocco}}$  and the  $\epsilon_{\text{alk}}$  calculated from the measured  $\delta^{13}\text{C}_{\text{DIC}}$  and the slope of this relationship is statistically indistinguishable from  $-1$  (Figure 7). This relationship could be advantageous because in paleo applications, the estimation of  $\delta^{13}\text{C}$  of DIC can be challenging due to limited planktic foraminifera abundance, uncertain vital effects in extinct planktic forams, and different depth or seasonal habitat of alkenone producers and planktic foraminifera. Several studies have applied  $\delta^{13}\text{C}_{\text{cocco}}$  as a DIC indicator (Pagani et al., 2005; Badger et al., 2013) and our results support this practice for sediment fractions with coccoliths dominated by alkenone producer species.

## 4.2 Temperature and growth rate effects on fractionation

We find that the carbon isotope fractionation due to photosynthesis,  $\epsilon_p$ , decreases significantly with increasing temperature (Pearson  $r = -0.6$ ,  $N = 22$ ,  $p = 0.0034$ ) (Figure 5). One explanation could be that higher temperature increases enzymatic rates and metabolic rates. We do not have accurate growth rate measurements for all experiments; thus, we cannot confirm whether

higher metabolic rates directly translate to higher net cell division rates. In a subset of our data with more robust cell division rates, we do find higher division rates at higher temperatures, as predicted by numerous previous studies of phytoplankton growth within their temperature range. *G. oceanica* RCC1303 was isolated at the Arcachon Bay on the coast of France, where monthly average sea surface temperatures range between  $12^\circ\text{C}$  and  $20^\circ\text{C}$ , and the cultured range therefore falls well within the temperature variation to which the strain was exposed. Attempts of growing this strain at  $12^\circ\text{C}$  or  $30^\circ\text{C}$  in our continuous cultivation setup were unsuccessful.

On the basis of theoretical grounds (Rau et al., 1996) as well as a previous culture meta-analysis (Stoll et al., 2019), it has been shown that higher coccolithophorid growth rate,  $\mu$ , decreases  $\epsilon_p$ . Most culture experiments measuring  $\epsilon_p$  have regulated growth rate using light intensity and/or day length (Rost et al., 2002; 2006), and a few have regulated nutrients (Bidigare et al., 1997). Our results suggest that the temperature influence on growth rate may strongly affect  $\epsilon_p$ . As shown in our subset of cultures with more robust cell division rates (Figure 5), we estimate a slope of  $\epsilon_p$  dependence on instantaneous growth rate of  $-3.85\text{‰} \pm 0.583\text{‰} \text{ day div}^{-1}$ , which is a similar order of magnitude as that found in previous culture meta-analysis: slope of  $-6.98\text{‰} \text{ day div}^{-1}$  (95% CI: 8.16 to  $-5.81$ ) (Stoll et al., 2019). Given the uncertainties in our growth rate quantification, to better evaluate whether a temperature dependence on growth rate might drive the temperature dependence of  $\epsilon_p$  in our dataset, we estimate the expected growth rate for each temperature using the temperature sensitivity of Fielding (2013). A maximum growth rate of  $1.3\text{--}2.1 \text{ d}^{-1}$  over the temperature range  $15^\circ\text{C}\text{--}27^\circ\text{C}$  for nutrient replete conditions such as our cultures is found.

We then estimate the consequences of this growth rate variation on  $\epsilon_p$ , using the  $\epsilon_p$  sensitivity to growth rate from Stoll et al. (2019): 95% CI of  $-8.16$  to  $-5.81\text{‰} \text{ day div}^{-1}$ . This calculation predicts a 5‰ range in  $\epsilon_p$  across our temperature range (Figure 8). The resulting calculated temperature sensitivity of  $\epsilon_p$  is statistically identical to the sensitivity determined from the MLR between temperature and  $\epsilon_p$  in our cultures (Table 1; Figure 8). This strongly suggests that the magnitude of the temperature-sensitive  $\epsilon_p$  variation we observe could be fully explained by the temperature sensitivity of  $\mu$ . Because of the dominance of light-regulated growth rate culture studies, there have been suggestions that  $\mu$  is not a significant term for  $\epsilon_p$  prediction and that only the light intensity was sufficient (Phelps et al., 2021a; 2021b). Our results suggest that, independent of light, growth rate itself is a necessary term for  $\epsilon_p$  prediction.

In a previous turbidostat culture growing *E. huxleyi* with 24 h illumination (Tchernov et al., 2014), constant  $\text{pCO}_2$  was maintained while manipulating temperature, leading to covariation of temperature and  $\text{CO}_2(\text{aq})$ . As temperature decreased from  $18^\circ\text{C}$  to  $7^\circ\text{C}$ ,  $\text{CO}_2(\text{aq})$  increased from 12 to  $18 \mu\text{M}$  and  $\epsilon_p$  increased from 10‰ to 18‰. From the slope between temperature and POC in our *G. oceanica* cultures, the  $11^\circ$  cooling would be expected to contribute about 4‰ increase in  $\epsilon_p$ , suggesting that the remaining 4‰ increase in  $\epsilon_p$  was due either to the increased  $\text{CO}_2(\text{aq})$  or to the amplified change in growth rate (decreased from 1.2 to 2  $\text{days}^{-1}$ ) compared to that expected from the growth rate sensitivity to temperature of cultures reported in Fielding (2013) (Supplementary Figure S1).

**TABLE 1** Multiple linear regression (MLR) analysis using least squares as regression type. For each response variable (carbon isotope fractionation  $\epsilon$ , or growth rate  $\mu$ ), the most significant iterations (equations E1, E2 ... ) using different explanatory variables (parameters) are listed.

MLR	Parameters	Estimate	Standard error	p-value	R <sup>2</sup>
$\epsilon_{\text{alk}}$ E1	Intercept	20.47	2.767	<0.0001	<b>0.7549</b> (N=18)
	Temperature	-0.4834	0.1056	0.0004	
	Light	$5.061 \cdot 10^{-3}$	$7.802 \cdot 10^{-3}$	0.5109	
	Ln(CO <sub>2</sub> (aq))	1.480	0.6281	0.0335	
$\epsilon_{\text{alk}}$ E2	Intercept	22.58	2.382	<0.0001	0.7496 (N=18)
	Temperature	-0.4793	0.1077	0.0006	
	Light	$7.257 \cdot 10^{-3}$	$7.157 \cdot 10^{-3}$	0.3278	
	CO <sub>2</sub> (aq)	$8.026 \cdot 10^{-2}$	$3.539 \cdot 10^{-2}$	0.0397	
$\epsilon_{\text{alk}}$ E3	Intercept	18.89	2.432	<0.0001	0.6161 (N=18)
	U <sup>K</sup> <sub>37</sub>	-8.159	2.878	0.0132	
	Light	$3.875 \cdot 10^{-3}$	$8.763 \cdot 10^{-3}$	0.6651	
	CO <sub>2</sub> (aq)	$9.783 \cdot 10^{-2}$	$4.315 \cdot 10^{-2}$	0.0397	
$\epsilon_{\text{POC}}$ E1	Intercept	17.32	1.726	<0.0001	<b>0.8001</b> (N=18)
	Temperature	-0.3761	0.07809	0.0003	
	Light	$6.837 \cdot 10^{-3}$	$5.187 \cdot 10^{-3}$	0.2086	
	CO <sub>2</sub> (aq)	$7.418 \cdot 10^{-2}$	$2.565 \cdot 10^{-2}$	0.0118	
$\epsilon_{\text{POC}}$ E2	Intercept	15.61	2.075	<0.0001	0.7905 (N=18)
	Temperature	-0.3851	0.07921	0.0003	
	Light	$5.407 \cdot 10^{-3}$	$5.626 \cdot 10^{-3}$	0.3528	
	Ln(CO <sub>2</sub> (aq))	1.276	0.4710	0.0170	
$\epsilon_{\text{cocco}}$ E1	Intercept	2.164	0.8935	0.0296	0.2904 (N=18)
	Temperature	$-7.927 \cdot 10^{-2}$	$4.042 \cdot 10^{-2}$	0.0701	
	Light	$-2.606 \cdot 10^{-3}$	$2.685 \cdot 10^{-3}$	0.3482	
	CO <sub>2</sub> (aq)	$-6.515 \cdot 10^{-3}$	$1.328 \cdot 10^{-2}$	0.6313	
$\mu$ E1	Intercept	-0.6999	0.9208	0.4896	0.4247 (N=8)
	Temperature	$6.589 \cdot 10^{-2}$	$4.469 \cdot 10^{-2}$	0.2143	
	Light	$5.619 \cdot 10^{-3}$	$2.145 \cdot 10^{-3}$	0.8063	
	CO <sub>2</sub> (aq)	$-7.412 \cdot 10^{-3}$	$1.684 \cdot 10^{-2}$	0.6826	

Bold values indicate the best iteration for each response variable.

### 4.3 Implications for pCO<sub>2</sub> reconstruction

Our finding of a temperature effect on  $\epsilon_p$  could be relevant for pCO<sub>2</sub> proxy reconstructions. Diverse studies confirm temperature-regulated growth rates in natural ocean settings. *In situ* estimations

of phytoplankton community  $\mu$  in the upper photic zone (<30m depth), determined by radiolabel incubations with correction for grazing losses, reveal a strong temperature-growth rate relationship which is also reflected in lower growth rates in higher latitudes (Sherman et al., 2016). A field compilation (835 determinations)



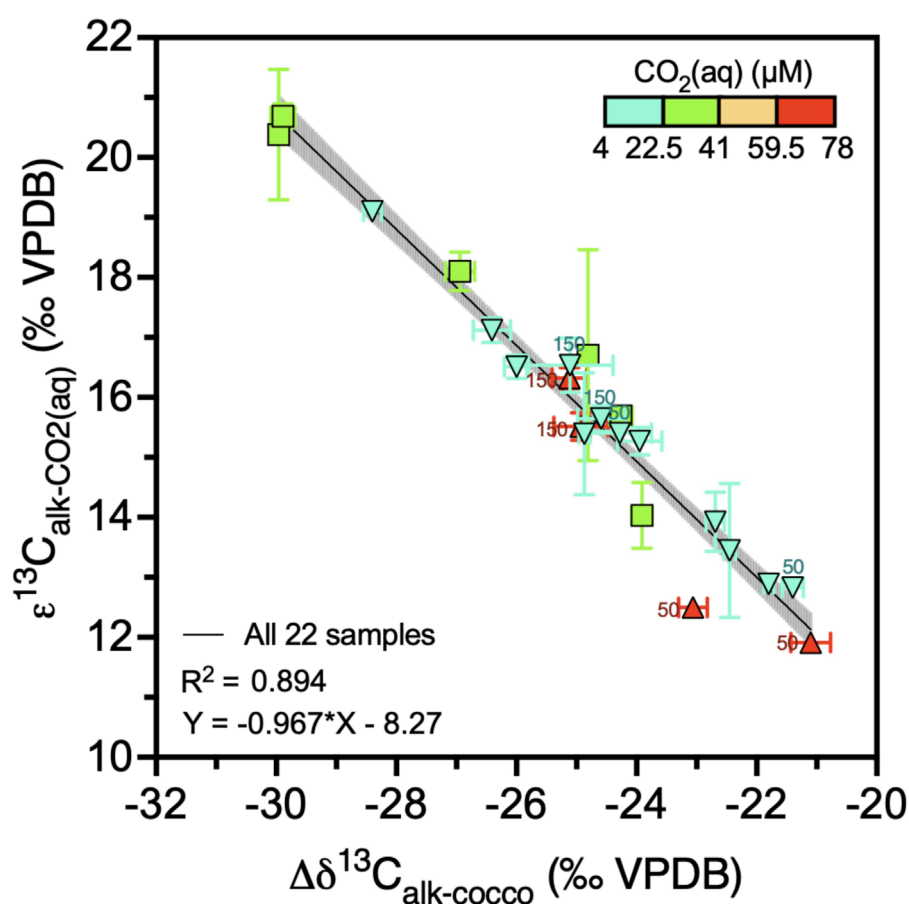


FIGURE 7

Alkenone carbon isotope fractionation in function of the difference between alkenone  $\delta^{13}\text{C}$  and coccolith  $\delta^{13}\text{C}$ . Colors indicate the  $\text{CO}_2(\text{aq})$  range and black line shows regression with 95% confidence intervals. Datapoint labels indicate light intensity in  $\mu\text{E}$  when different than 200  $\mu\text{E}$ . Error bars show one standard deviation.

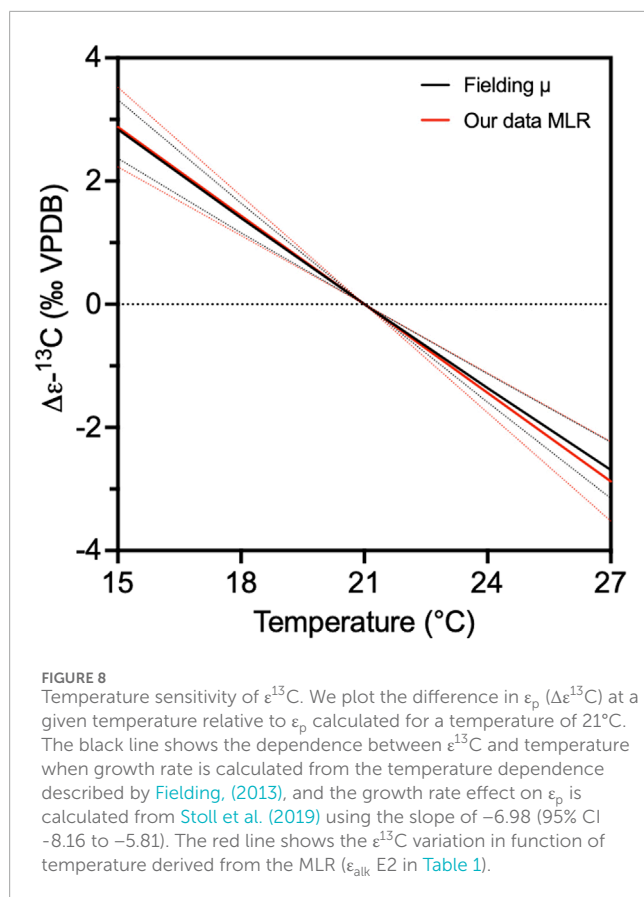
found a similar temperature- $\mu$  relationship to culture experiments (Fielding, 2013) and to  $\mu$  estimates from satellite and various earth system models (Behrenfeld et al., 2005; Sherman et al., 2016). In marine particulates of the southern Yellow Sea,  $\epsilon_p$  correlated strongly with temperature but not to *in situ*  $\text{CO}_2(\text{aq})$  (Liu et al., 2022). The residual non- $\text{CO}_2$  variation in  $\epsilon_p$  in alkenone photic zone samples is significantly correlated with temperature (Hernández-Almeida et al., 2020).

#### 4.3.1 Strategies for accounting for temperature-driven growth rate variations

While some studies have proposed that cell size may be an indicator that allows growth rate effects on  $\epsilon_p$  to be accounted for in  $p\text{CO}_2$  studies (Zhang et al., 2020), this approach has been controversial (Phelps et al., 2021a). We find no significant correlation between  $\epsilon_{\text{alk}}$  (and  $\epsilon_{\text{POC}}$ ) and cell size, or between temperature and cell size (Figure 3). This suggests that the large temperature influence on metabolic rates and  $\epsilon_p$  fractionation is not manifest in cell or coccolith size. We conclude that other approaches must be used to account for this temperature driven growth rate effect in  $p\text{CO}_2$  estimations.

The significance of temperature-modulated growth rate effects on  $\epsilon_p$  suggests that direct coccolithophore-derived temperature proxies may be more useful to constrain growth rate effects on  $\epsilon_p$  for estimation of past  $\text{CO}_2$ . One approach would be a direct regression of  $\epsilon_p$  from  $\text{CO}_2(\text{aq})$  and temperature sensitive proxy parameters such as the  $U^{K}_{37}$  ratio from the same alkenones. This multivariate regression has a  $r^2$  of 0.62 (Table 1,  $\epsilon_{\text{alk}}$  E3). We additionally note a strong correlation of  $\epsilon_p$  to  $\epsilon^{18}\text{O}_{\text{cocco-water}}$  as temperature is the dominant control on oxygen isotope fractionation in our study (Figure 3). However, for estimation of coccolith calcification temperatures in past oceans,  $\delta^{18}\text{O}_{\text{cocco}}$  is likely less useful than alkenone temperature, as the seawater  $\delta^{18}\text{O}$  is variable and cannot be independently constrained. A disadvantage of the linear regression of  $U^{K}_{37}$  to estimate  $\epsilon_p$  is that it does not account for the nonlinear relationship between temperature and growth rate.

We therefore suggest that a more robust approach may be the use of alkenone-derived temperature proxies to estimate the temperature influence on growth rate for past alkenone production using standard formulations of coccolithophorid growth rate in the ocean (Krumhardt et al., 2017). The results from our new *G. oceanica* cultures suggest a net  $\epsilon_p$  dependence on temperature-regulated growth statistically indistinguishable from



that derived previously from meta-analysis of culture  $\epsilon_p$  and temperature sensitivity of growth rate (section 4.2). Consequently, the new culture results support the approach used in previous studies accounting for the temperature sensitive growth rate effects on  $\epsilon_p$  such as applied to glacial-interglacial sequences (Stoll et al., 2019) or Late Miocene  $\text{pCO}_2$  estimations from  $\epsilon_p$  (Tanner et al., 2020).

#### 4.3.2 Potential impact of temperature on existing Cenozoic $\epsilon_p$ time series

Because declining atmospheric  $\text{pCO}_2$  leads to a decline in earth's mean surface temperature with a magnitude defined by earth system sensitivity, coupled declines in  $\text{pCO}_2$  and temperature (and growth rates) may lead to opposing effects on  $\epsilon_p$ . Several published paired  $\epsilon_p$  and alkenone SST records for the Pliocene feature  $3^\circ\text{C}$ – $7^\circ\text{C}$  cooling and  $0.4\text{‰}$ – $2\text{‰}$  declines in  $\epsilon_p$ . If the measured  $\epsilon_p$  were adjusted to remove the effect of changing temperature and growth rate as determined in our *G. oceanica* cultures, the adjusted  $\epsilon_p$  would exhibit a steeper decline in all sites, ranging from  $2.6\text{‰}$  to  $5.6\text{‰}$  (Supplementary Figure S2). Of two examined Paleogene paired  $\epsilon_p$  and alkenone SST records, that of ODP 511 exhibits a  $9^\circ\text{C}$  cooling over the latest Eocene, which could contribute to a nearly  $5\text{‰}$  increase in  $\epsilon_p$  due to temperature-stimulated growth rate. Accounting for this potential effect would impose a  $5\text{‰}$  decrease in adjusted  $\epsilon_p$ . Because temperatures in low latitude sites exceed the alkenone proxy sensitivity, the magnitude of cooling over the Cenozoic is not well constrained for sites such as western tropical Atlantic ODP Site 925, which

provides the longest continuous phytoplankton  $\text{pCO}_2$  estimates (Zhang et al., 2013). If GDGT temperature estimates at this site are accurate, then the temperature-growth rate effect has little influence on the  $\epsilon_p$  over this time interval. Clumped isotopic temperatures on coccoliths (Mejía et al., 2023) might be used to test the stability of coccolithophorid growth temperatures over this time interval at this site.

While our new results, coupled with estimates of coccolithophorid growth temperature, provide a mechanism for accounting for the effect of temperature-driven variation in phytoplankton growth rates on  $\epsilon_p$ , temperature is not the only influence on phytoplankton growth in the ocean. Nutrient and light limitation also have the potential to reduce growth rates, and independent proxies for these factors have been elusive. Relieving light limitation may increase growth rate which would tend to decrease  $\epsilon_p$ . Yet, in culture experiments, light intensity and photoperiod have been shown to have a large positive direct effect on  $\epsilon_p$  in coccolithophorids as well as other phytoplankton (Rost et al., 2002; Cassar et al., 2006; Phelps et al., 2021a). A comprehensive assessment of light and growth rate effects in  $\epsilon_p$  should consider these parameters as well.

## 5 Conclusion

Culture experiments of a strain of *G. oceanica* document that increasing temperature decreases the carbon isotopic fractionation during photosynthesis ( $\epsilon_p$ ) when measured either as  $\epsilon_{\text{POC}}$  or  $\epsilon_{\text{alk}}$ . This result is consistent in trend and magnitude with the expected increase in growth rate with increasing temperature, and decreased  $\epsilon_p$  with increasing growth rate. Our experiments characterized the effect in a single alkenone-producing species, which could be grown in continuous culture only over the range from  $15^\circ\text{C}$  to  $27^\circ\text{C}$ , so it could be advantageous for future experiments to confirm the observed effect in strains which can be cultured in the lower temperature ranges typical of higher latitude oceans. This experiment evaluated the effect of temperature in nutrient-replete cultures, and future experiments would be required to evaluate the interplay of temperature and nutrients on growth limitation and isotopic fractionation.

The finding that temperature influences  $\epsilon_p$  via growth rate has significant implications for the application of  $\epsilon_p$  from marine sediments as an indicator of past  $\text{pCO}_2$ . Where the surface ocean has experienced significant cooling accompanying a decline in  $\text{pCO}_2$ , the decreased growth rate may attenuate the declining trend in  $\epsilon_p$  resulting from declining  $\text{CO}_2(\text{aq})$ . Employing estimates of SST decline from alkenone unsaturation to correct for this temperature-induced growth rate effect may be one approach to improve the detection of past trends in  $\text{pCO}_2$  from alkenone-producing phytoplankton.

## Data availability statement

The original contributions presented in the study are included in the article/Supplementary Material, further inquiries can be directed to the corresponding authors.

## Author contributions

IT-R: Conceptualization, Data curation, Formal Analysis, Investigation, Methodology, Visualization, Writing–original draft, Writing–review and editing. AC: Formal Analysis, Investigation, Methodology, Writing–review and editing, Writing–original draft. RW: Investigation, Methodology, Validation, Writing–review and editing. MJ: Investigation, Methodology, Validation, Writing–review and editing. HZ: Investigation, Writing–review and editing. HS: Conceptualization, Funding acquisition, Investigation, Supervision, Writing–original draft, Writing–review and editing.

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## Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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## Supplementary material

The Supplementary Material for this article can be found online at: <https://www.frontiersin.org/articles/10.3389/feart.2024.1331179/full#supplementary-material>

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