



Contamination of Stream Sediment With Heavy Metals in the Awetu Watershed of Southwestern Ethiopia

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Surface sediment samples were collected from different streams of Awetu Watershed in southwestern Ethiopia. Sediment samples were analyzed for As, Cd, Cr, Pb, and Hg levels using inductively coupled plasma optical emission spectrometry. The heavy metal concentration ranged from 183.60 to 1,102.80 mg/kg for As (mean 623.32 ± 291.65 mg/kg), 4.40-303.20 mg/kg for Cd (151.09 ± 111.5 mg/kg), 149.20-807.20 mg/kg for Cr (375 ± 212.03 mg/kg), 485.60-3,748.80 mg/kg for Pb (2005.94 ± 954.99 mg/kg) and 3.6-5.6 mg/kg for Hg (4.64 ± 0.59 mg/kg). The mean heavy metal concentration in the streams followed the decreasing order of Pb > As > Cr > Cd > Hg. As, Cr and Pb are detected at high concentrations with values of 623.32, 375.00, and 2,005.94 mg/kg respectively. A low level of heavy concentration (3.6 mg/kg) was recorded for Hg. The contamination factor (CF) of all the studied heavy metals ranged from a low degree (CF < 1) to a very high degree (CF \leq 6). Mainly, Dololo and Kito streams show a very high degree of contamination (CF \leq 6) than Awetu and Boye streams. Specifically, As, Cd and Cr in the Dololo and Kito streams have significantly elevated concentrations than others. Geo-accumulation index (Igeo) shows low to moderate contamination level with As, Pb, and Hg; uncontaminated to heavily contaminated by Cr; and moderate to extreme contamination by Cd. Untreated solid waste, garages and farmlands were sources of contamination. Streams receiving wastewater effluents from teaching institutions had higher heavy metal concentrations. Dumping of electronic wastes and car washing discharges also identified as another source of pollution.

Keywords: Awetu watershed, contamination, Ethiopia, heavy metals, sediment

INTRODUCTION

Quality of water is affected by human-induced or natural activities in the upstream watershed (Sany et al., 2013). As a result of the natural flow of the water, most pollutants are drained into a one-point collection site, such as reservoirs that can serve as a sink for different pollutants (Nowrouzi and Pourkhabbaz, 2014a). Due to its potential and toxic environmental and public health effects and the ability to accumulate, heavy metal contamination of the aquatic ecosystems is becoming a potential global problem (Sharma et al., 2015).

Sediment contamination with heavy metals could be either from natural geogenic sources or sourced from anthropogenic activities (Giouri et al., 2010). The benthic environment of aquatic

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ecosystems receives and absorbs heavy metals from natural weathering, erosion, industrial wastes, and atmospheric deposition (Jaishankar et al., 2014). Anthropogenic activities, such as industrial and agricultural discharges, inappropriate disposal of industrial wastes, dumping of domestic and municipal wastes, faulty drainage systems are some of the causes for heavy metal contamination of aquatic ecosystems (Hahladakis and Smaragdaki, 2013; Islam et al., 2015).

Several studies indicated that heavy metal concentration in stream sediments is relatively high due to significant anthropogenic metal loadings carried by tributary rivers (Li et al., 2011). As a result, surficial sediments may serve as a metal puddle that can release metals to the overlying water that could potentially adversely affect the riverine ecosystems (Evans et al., 2003; Reda and Ayu, 2016). It is well-known that the mobility and availability of heavy metals in aquatic environments are primarily affected by physicochemical parameters of water, such as pH, dissolved oxygen, and organic matter content (Sim et al., 2016).

Due to the absence of waste treatment facilities in cities and the discharge of wastes into the nearby water bodies, rivers located near cities are often suffering from heavy metal contamination (Giridharan et al., 2008). A study conducted by Mekonnen and his colleagues (2014) on the Akaki river that crosses Addis Ababa city showed severe contamination of the stream water with heavy metals sourced from industrial, residential, and agricultural wastes. The quality of the water gets worsened as heavy metals bioaccumulate in algal blooms in the downstream section of the water bodies where wastes are discharged from the upstream (Melaku et al., 2007).

Other studies of sediment and water pollution assessment in Ethiopia and in the region showed that the concentration of heavy metals rapidly increases when it entered Addis Ababa city (Akele et al., 2016; Aschale et al., 2016; Woldetsadik et al., 2017), Kombolcha city (north-central part of Ethiopia) (Zinabu et al., 2019), but significantly decreased downstream of the cities. This decrement of heavy metal in the downstream of the town might be due to heavy metals adsorbed into the sediment portion of the river (Khan et al., 2017). Bottom stream sediments are sensitive indicators for monitoring contaminants as they can act as a sink and a carrier for pollutants in the aquatic environment (Benson and Etesin, 2008; Asefa and Beranu, 2016). The presence and behavior of metals in the natural water is a function of the substrate sediment composition, the suspended sediment composition, and the water chemistry (Suresh et al., 2012). Heavy metals are potentially accumulated in sediments and aquatic organisms in the fresh water environment and subsequently transferred to man through the food chain. As a result the concentration of heavy metals in aquatic ecosystems are usually scrutinized by measuring their concentrations in sediments (Ekeanyanwu et al., 2010; El Bouraie et al., 2010).

Nowadays, various indices are developed to assess the level of sediment contamination and ecological risk. Geo-accumulation index (I_{geo}), enrichment factor (EF), contamination factor (CF), and pollution load index (PLI) methods have been commonly used for geochemical standardization approaches (Müller, 1979). The presence of heavy metals in stream sediments has created an

alarming situation that requires immediate attention. Thus, sediment analysis plays a vital role in evaluating the aquatic environment (S. Li and Zhang, 2010). However, information on metal concentrations in stream sediments at a spatial scale along a course of streams in the urban and semi-urban environment is insignificant. Hence, this study addresses existing research gaps and provides valuable information regarding the spatial distributions of selected heavy metals in the urban rivers around the Awetu watershed catchment. This paper discusses the concentrations of heavy metals (As, Cd, Cr, Pb, and Hg) in the surface sediments of four streams around the Awetu watershed catchment, comparing with the Sediment Quality Guidelines (SQGs), and assess the heavy metal contamination using the I_{reco} , CF, and PLI methods.

METHODS AND MATERIALS

Study Area and Site Description

Awetu watershed encompasses Jimma City and the surrounding urban and semi-urban regions of southwestern Ethiopia. The streams under the Awetu watershed are receiving pollution from different sources, such as different laboratories of Jimma University, Jimma Medical Center, garages and carwashing points, untreated Jimma town municipal effluent, agricultural non-point source discharges, and animal feedlots. The watershed mainly contains four different streams, of which Awetu is the largest, which divides Jimma city into two while Dololo, Kito, and Boye are tributaries of Awetu stream. The major land-use activities in the Awetu watershed along with the site code and the coordinate points are listed in **Table 1**. Sampling sites were mapped using the ArcGIS 10.3 for Desktop (ESRI, Redlands, California, United States). Twenty surface integrated sediment samples were collected from each sampling point (**Figure 1**).

Sediment Samples Collection and Preparation

Submerged surface integrated sediment samples collected from 20 sampling stations in the urban and semi-urban streams of the Awetu watershed are used. As described by Decena et al., 2018, each sample constituted a 0-15 cm depth of sediment collected using a stainless steel bottom sampling dredge (grab sampler). Different grain size of the sediments were considered and homogenized to keep the uniformity of samples collected from each sampling stations (Figure 2). The grab sampler was washed with detergent and rinsed with distilled water before each use to reduce possible contamination. Sediment samples were placed in a new polyethylene zip-lock bag and placed in a cooler with ice, transported to the Environmental Health Laboratory of Jimma University within a few hours sampling, and kept at 4°C until analysis. During sample collection, a hand-held GPS was used to identify the sites (Garmin eTrex®). The sediment samples were air-dried and ground using a pestle and mortar after homogenized. Cobles, pebbles, and other coarse debris were manually removed. Finally, the grounded samples were sieved to pass through 63 µm mesh size to obtain fine-powdered

Stream name	Site name	Site code	GPS coordinate	Major anthropogenic activities of the site
Awetu	Awetu 1	A1	36° 49′ 51.15″ E 7° 42′ 28.14″N	Agricultural activities and grazing
	Awetu 2	A2	36° 49′ 52.39″E 7 °42′ 4.03″N	Agricultural activities, grazing, washing clothes, and bathing
	Awetu 3	A3	36° 49′ 52.89″E 7° 41′ 43.38″N	Horticulture, recreational, residential and commercial, vehicle traffic, and agricultural runoff
	Awetu 4	A4	36° 49′ 57.27″E 7 °41′ 13.20″N	Washing, swimming, and fetching water for household consumption
	Awetu 5	A5	36° 50′ 5.95″E 7 °40′ 46.11″N	Vehicle traffic, washing, car washing, and seedling plantation
	Awetu 6	A6	36° 50′ 9.79″E 7 °40′ 15.47″N	High vehicle traffic, residential area, and small scale industries
Dololo	Dololo 1	D1	36° 50′ 50.67″E 7° 40′ 45.52″N	Public institutions, domestic activities, vehicle traffic, hospital, chemical and biological laboratories, and construction sites
	Dololo 2	D2	36° 50′ 41.99″E 7° 40′ 0.28″N	Car washing, small scale enterprises like garages, woodwork, and vehicle traffic
	Dololo 3	D3	36° 50′ 41.65″E 7° 39′ 44.82″N	Commercial area, high vehicle traffic, garages, gas/petro station, and seepage
	Dololo 4	D4	36° 51′ 4.81″E 7° 39′ 22.63″N	Car washing, gas/fuel station, garages, residential and commercial and seepage
	Dololo 5	D5	36° 51′ 43.74″E 7° 40′ 2.46″N	Commercial, recreational, vehicle traffic, bus park, gas/petrol station, cement stores, metal works and fabrications, and seepage
Kito	Kito 1	K1	36° 48′ 34.68″E 7° 40′ 23.97″N	Grazing, institutional wastes, waste stabilization pond, wood and metalwork enterprise, garage, car washing, agricultural activities, and bridge
	Kito 2	K2	36° 49′ 15.04″E 7° 40′ 19.80″N	Residential, commercial, garage, seepage, and agricultural activities
	Kito 3	K3	36° 49′ 32.86″E 7° 39′ 54 87″N	Residential, commercial, seepage, agricultural activities, and airport
	Kito 4	K4	36° 49′ 53.70″E 7° 39′ 31 91″N	Grazing, agricultural activities, and small scale enterprises like garages
	Kito 5	K5	36° 50′ 18.86″E 7° 39′ 1.56″N	Solid waste dump sites, horticulture, residential and vehicle traffic
Boye	Boye 1	B1	36° 50′ 46.97″E 7 °38′ 57 16″N	Car washing, vehicle traffic, residential and commercial area
	Boye 2	B2	36° 51′ 23.16″E 7° 39′ 19.66″N	Agriculture runoff, irrigation, and residential area
	Boye 3	B3	36° 51′ 44.99″E 7° 39′ 38.48″N	Irrigation, agricultural runoff, slaughterhouse, and residential area
	Boye 4	B4	36° 52′ 12.42″E 7° 39′ 42.30″N	Wetland, grazing, agricultural activities, fishing and recreational

TABLE 1 | Sampling stations and their code, GPS coordinates, and major anthropogenic activities in the Awetu watershed streams.

particles and placed in a clean polyethylene zip-lock bag. The processed samples were refrigerated at -20° C until further analysis.

Analysis of Sediment Samples

A 0.25 g sediment sample was accurately weighed and placed in a dry and clean Teflon microwave digestion vessel and digested with 20 ml aqua regia (3 HNO₃:1 HCl v/v) until the solution turned colorless in an open thermostatically controlled hot plate. The digest samples were heated near to dryness and cooled to ambient temperature. The beaker walls were rinsed with 10 ml de-ionized water, and 5 ml HCl were added, mixed, and heated. After that, filtration of the sample into a 50 ml volumetric flask using Whatman No. 42 filter paper was made. Then the digest was allowed to cool and transferred into a 100 ml standard flask and filled to the mark with de-ionized water. The digested samples were subjected to metal analysis using inductively coupled plasma optical emission spectrometry (ICP-OES)

(SPECTRO ARCOS Model: ARCOS FHS12, Germany). The calibration curves obtained from concentration and absorbance were used to determine the level of heavy metals of each sample. Data were statistically analyzed using the fitting of a straight line (r > 0.999). A blank reading was used to make a necessary correction during the calculation of concentrations.

Data Analyses

Different contamination indices were used to evaluate the findings with different standards. The contamination status of sediment was assessed based on the geo-accumulation index (I_{geo}), contamination factor (CF), and pollution load index (PLI) (Graça et al., 2002).

Contamination Factor

CF was used to determine the simple and effective tool in monitoring the level of heavy metal contamination at each site using the following formula (Graça et al., 2002).



$$CF = \frac{C \, sample}{C \, background} \tag{1}$$

Where C_{sample} is the mean metal content in sample sediment, $C_{\text{background}}$ is the mean natural background value of the metal. The natural background sample was collected from A1, which is about 10 km from the main areas where maximum anthropogenic activities are performed. This area is assumed to be free from the known anthropogenic source of selected heavy metals. The ratio of the measured concentration to the natural abundance of a given metal had been proposed as the index CF being classified into four grades for monitoring the pollution of a single metal over a period of time (Ali et al., 2016): low degree (CF < 1), moderate degree (1 ≤ CF < 3), considerable degree (3 ≤ CF < 6), and very high degree (CF ≤ 6). Thus, the CF values can monitor the enrichment of a given metal in sediments over a while.

Pollution Load Index

PLI is used to evaluate sediment quality. PLI of the combined approaches of the five heavy metals were calculated according to (M. A. Islam et al., 2017). The PLI is defined as the nth root of the multiplications of the contamination factor of the target heavy metals (CF).

$$PLI = (CF1 \times CF2 \times CF3 \times \ldots \times CFn)1/n$$
(2)

Where CF_1 is the concentration of the first metal, CF_2 is the concentration of the second metal, CF_3 is the concentration of the third metal and CF_n is the concentration of metal nth, and n = the total number of studied heavy metals in the sample. PLI = 0 indicates excellence; PLI = 1 suggests the presence of only a baseline level of pollutants and PLI > 1 indicates progressive deterioration of the site and estuarine quality (Tomlinson et al., 1980). The PLI evaluated the overall toxicity status of the sample and its contribution to the contribution of the five metals.

Geo-Accumulation Index (Igeo)

The degree of contamination from the heavy metals could be assessed by measuring the geo-accumulation index (I_{geo}). The index of geo-accumulation has been widely used for the assessment of sediment contamination (Islam et al., 2014). To characterize the level of heavy metal contamination in sediment samples, geo-accumulation index (I_{geo}) was calculated using the equation:

$$Igeo = Log2\left[\frac{Cn}{1.5 Bn}\right]$$
(3)

Where; Cn is the content of measured metal "n" in the samples, Bn is the crustal shale background content of the metal "n", the constant of 1.5 is introduced to minimize the



variation of background values due to lithogenic origins, and I_{geo} is a quantitative index of metal enrichment or contamination levels.

Geo-accumulation index (I_{geo}) values were interpreted as $I_{geo} \le 0$ uncontaminated; $0 \le I_{geo} \le 1$ —uncontaminated to moderately contaminated; $1 \le I_{geo} \le 2$ —moderately contaminated; $2 < I_{geo} \le 3$ —moderately to heavily contaminated; $3 \le I_{geo} \le 4$ — heavily contaminated; $4 \le I_{geo} \le 5$ —heavily to extremely contaminated; and $5 < I_{geo}$ —extremely contaminated (Müller, 1979).

Statistical Analyses

Heavy metal concentration data were log-transformed to reduce the variability and minimize outliers. R statistical package (R Core Team, 2019) was used to undertake a paired permutation test evaluating the presence of a statistically significant difference between the streams. To determine heavy metal concentrations variability among sediment samples between the urban and semiurban sites, a one-way ANOVA was applied. Pearson's correlation was made to examine the correlation between heavy metals and selected physicochemical properties. A probability of 0.05 was considered as a level of significance.

Quality Control and Quality Assurance

The quality of the analytical results is assured by laboratory quality assurance and quality control methods. These were

implemented by pre-cleaning of laboratory materials with 10% HNO₃, use of standard operating procedures, analysis of blanks, calibration with the standard, and recovery of known additions. Each heavy metal was analyzed in three replicates, and the results were presented as mean. The recovery percentages attained for the reference materials of sediment is between 97 and 120%, meeting the acceptable recovery of 80–120% recommended by the United States-EPA. In this study, the metal contents were measured according to dry weight to ensure consistency.

RESULTS AND DISCUSSION

Description of Anthropogenic Activities

Heavy metals concentration in stream sediment indicates the magnitude of pollution in water ecosystems and the capacity to accumulate contaminants (Nowrouzi and Pourkhabbaz, 2014b). Speciation and bio-availability of heavy metals in aquatic ecosystems are strongly dependent on pH, and values <4 increase the toxicity of heavy metals (Edokpayi et al., 2016). The average pH values determined varied between 5.82 and 7.60, reflecting the availability of carbonate hosts in the area (Moore et al., 2011). Mobility of metal is also affected by pH, adsorption/ desorption processes, salinity, sulfur, and carbonates (Giouri et al., 2010). Bottom sediments with a higher concentration of

Metals A1 A5 A6 D1 D2 D3 D4 D5 K1 K2 K3 K4 K5 B1 B2 B3 B4 A1 A2 A3 A4 A5 A6 D1 D2 D3 D4 D5 K1 K2 K3 K4 K5 B1 B2 B3 B4 A5 369.2 57.2 52.8 47.0 444 183.2 1026 101.16 873.2 720.8 1102.8 103.7 2899.6 664 591.6 351.2 249.2 362.8 823.32 284.26 20 17 C4 232 57.2 657.6 430.2 267.2 540.8 307.2 247.6 198 192.2 199.2 195.7 105.6 166 10 17 172 192 307.2 267.6 607.2 540.8 377.6 198.7 150.6 198.6 106.6 166.4 191.6 191.8	(APS)									Sampling (stations										Mean	StDev	aEC	PEL	°WRSE
As 369.2 503.2 47.0 444 183.5 102.6 101.6 87.3.2 720.8 1102.8 1037.2 899.6 664 591.6 551.2 243.2 224.45 20 17 Cd 23.2 57.2 52.8 430.4 813.2 102.6 101.6 87.32 720.8 1102.8 1037.2 899.6 664 591.6 551.2 244.26 20 17 Cd 23.2 57.2 567.6 303.2 247.2 247.6 171.2 192 36.8 452.2 224 151.09 106.67 3 3.53 Cr 174 173.6 186.8 163.6 567.6 549.8 507.2 540.8 57.6 419.6 419.8 132.8 149.2 207.6 438 3.55 Pb 952 44.4 4.4 5.00.84 370.8 500.4 210.04 118.16 900.890.8 100.6 74 4.3 05.9 49.4	etals A1	A2	A3	A4	A5	A6	5	D2	D3	D4	D5	Ł	K2	K3	K4	K5	B	B2	ß	2					
Cd 232 572 528 472 44 40.12 2364 2956 302 2896 2012 3032 247.2 247.6 171.2 192 336 28 452 224 151.09 108.67 3 353 Cr 174 173.6 186.8 135.6 186.8 557.2 647.2 647.2 567.6 498.8 807.2 540.8 57.6 418.6 418.8 122.8 492 207.6 436 375 206.6 ND 90 Cr 174 173.6 186.8 125.8 195.8 2381 537.2 647.2 647.2 547.8 300.4 236.8 307.2 540.8 57.6 418.6 418.9 122.8 149.2 207.6 436 375 206.6 ND 90 Cr 174 173.6 186.8 125.8 45.6 135.8 195.8 2381 537.2 847.2 647.2 547.8 300.4 257.6 219.6 111.6 900 980.9 105.6 774 2005.4 901.8 91.3 41 4.8 4.8 4.8 4.8 4.8 4.8 4.4 4.8 5.2 4.8 4.4 4.8 5.0 4.8 4.4 5.2 4.8 4.4 4.6 5.6 0.46 90.9 13.8 138.1 15.0 10.8 174 5.0 4.8 4.4 4.8 5.0 4.8 4.8 4.7 5.7 4.8 4.8 4.8 4.8 5.7 4.8 4.8 4.8 4.8 4.8 5.7 4.8 4.8 4.8 5.7 4.8 4.8 5.7 4.8 4.8 5.7 4.8 4.8 5.7 4.8 4.8 5.7 4.8 5.7 4.8 4.8 5.7 4.8 4.8 5.7 4.8 5.7 4.8 5.7 4.8 5.7 4.8 5.7 4.8 5.7 4.8 5.7 4.8 5.7 4.8 5.7 4.7 5.7 5.7 5.7 5.7 5.7 5.7 5.7 5.7 5.7 5	369.2	503.2	470	444	183.6	430.4	813.2	1026	1011.6	873.2	720.8	1102.8	1037.2	899.6	664	591.6	351.2	249.2	362.8	362.8	523.32	284.26	20	17	-
Cr 174 1736 186.8 169.6 133.6 158.8 537.2 607.2 643.2 567.6 498.8 807.2 540.8 537.6 419.6 418.8 132.8 149.2 207.6 436 375 206.66 ND 90 Pb 952 1585.2 147.2 1235.6 485.6 1382 2381.6 2940.8 3330.8 2308.4 235.4.8 3748.8 3080.4 2767.6 2100.4 1811.6 900 880.8 1056 2744 2005.94 930.81 300 91.3 Hg 4.4 4.8 4.4 5.2 4.8 4.4 4.4 6 3.6 5.2 4.4 4 4.8 4.8 4.4 4 5.2 4 4.4 5.6 4.8 4. 6.4 4. 6.7 20 0.486	23.2	57.2	52.8	47.2	4.4	40.12	236.4	295.6	302	269.6	201.2	303.2	247.2	247.6	171.2	192	33.6	28	45.2	224	151.09	108.67	с С	3.53	0.25
Pb 952 1585.2 1472.4 1235.6 485.6 1382 2381.6 2940.8 3330.8 2908.4 235.4.8 3748.8 3080.4 2767.6 2100.4 1811.6 900 880.8 1056 2744 2005.94 930.81 300 91.3 Hg 44 4.8 4.4 5.2 4.8 4.4 4.4 6 3.6 5.2 4.4 4 4.8 4.8 4.4 4 5.2 4 4 4.6 5.6 4.8 4.64 0.57 20 0.486	174	173.6	186.8	169.6	133.6	158.8	537.2	607.2	643.2	567.6	498.8	807.2	540.8	537.6	419.6	418.8	132.8	149.2	207.6	436	375	206.66	Q	06	55.5
Hg 44 48 44 52 48 44 44 6 3.6 5.2 4,4 4 4,8 4 4,8 4 4.8 20 0.486	952	1585.2	1472.4	1235.6	485.6	1382	2381.6	2940.8	3330.8	2908.4	2354.8	3748.8	3080.4	2767.6	2100.4	1811.6	006	880.8	1056	2744 2	005.94	930.81	300	91.3	48
	4.4	4.8	4.4	5.2	4.8	4.4	4.4	9	3.6	5.2	4.4	4	4.8	4.4	4	5.2	4	4.4	5.6	4.8	4.64	0.57	20 C	.486	0.4

Probable Effects Level (PEL) (Naclu et al., 2014) and Canadian Freshwater Sediment Guidelines (CFSG) (Shafie et al., 2015).

World River Sediment Background Concentration (WRSB)

organic matter also influence the solubility and mobility of the heavy metals in the aquatic ecosystems (Journal et al., 2010). The formation of potential mobile metal dissolved organic carbon complexes under oxidizing conditions prevents metals from coprecipitation with or adsorbing to oxides of metallic ions (Du Laing et al., 2009).

Heavy Metal Distribution

Heavy metal concentrations in sediments of the Awetu watershed catchment channelized streams with different world standards are summarised in Table 2. The present study exposed all the heavy metals in stream sediment were found above the permissible limit set by WHO (1993) and EU (1998) (Bhuyan et al., 2017). A significant difference in the concentration of As, Cr, Cd, and Pb between streams was identified (p < 0.01).

The heavy metal concentration ranged from 183.60 to 1,102.80 mg/kg for As (mean $623.32 \pm 291.65 \text{ mg/kg}$), 4.40-303.20 mg/kg for Cd (151.09 ± 111.5 mg/kg), 149.20-807.20 mg/kg for Cr (375 ± 212.03 mg/kg), 485.60-3,748.80 mg/kg for Pb (2005.94 ± 954.99 mg/kg) and 3.6-5.6 mg/kg for Hg ($4.64 \pm 0.59 \text{ mg/kg}$). The mean heavy metal concentration in the streams followed the decreasing order of Pb > As > Cr > Cd > Hg. As, Cr and Pb are detected at high concentrations with values of 623.32, 375.00 and 2,005.94 mg/kg respectively and the minimum concentration found is Hg with a value of 3.6 mg/kg.

The heavy metals concentrations of sediment samples from individual streams showed high variability where the Dololo stream was the highest for all studied heavy metals except Hg. The concentration of heavy metals in all stream sediments are by far higher than the Sediment Quality Guidelines (SQG) values set by World Rivers Khan et al., 2005), EC and PEL (Nadu et al., 2014), USEPA water quality criteria (Hahladakis and Smaragdaki, 2013), Canadian Freshwater Sediment Guidelines (Gergen et al., 2015), and World River Sediment Background Concentration (WRSB) concentration and background concentration of this study.

The highest concentrations of As were observed in D2, D3, K1, and K2 sampling sites where it is the center of the city at which maximum anthropogenic activities are experienced. Some of the heavy metal sources of pollution are the burning of fossil fuels, the use of arsenical fungicides, herbicides and insecticides in agriculture, and wood preservatives in woodwork enterprises (Shankar et al., 2014). The highest value of As is perceived at lower pH (5.82). The study supports this; pH is the most critical factor controlling arsenic speciation under oxidizing conditions and H₂AsO₄⁻ is dominant at a lower value (pH < 6.9) (Smedley and Kinniburgh, 2001). The interchange in oxidation and solubility of arsenic species affects the environmental behavior and subsequent transportation of arsenic species in the aquatic ecosystems (Chatterjee et al., 2017).

The highest values of Cd are found at K1, D2, and D3 sampling sites. The value at K1 could be attributed to the direct influence of effluents coming from institutional wastes, garage, car-washing, and agricultural activities where phosphate fertilizer has been in

l _{geo}				F	PLI			CF
Class	Value	Classification	Level	Value	Contamination	Level	Value	Level
0	<0	Uncontaminated	1	CF < 1	Low degree	1	PLI = 0	Excellent
1	0–1	Uncontaminatedto moderately contaminated	2	CF = 1–3	Moderate degree	2	PLI = 1	Baseline
2	1–2	Moderately contaminated	3	CF = 3–6	Considerable degree	3	PLI >1	Progressive deterioration
3	2–3	Moderately and heavily contaminated	4	CF > 6	Very high degree			-
4	3–4	Heavily contaminated						
5	4–5	Heavily to extremely contaminated						
6	>5	Extremely contaminated						

TABLE 3 | Contamination categories based on a geo-accumulation index (I_{aeo}), pollution load index (PLI), and contamination factor (CF).

use. The values at D2 and D3 sites could be attributed to commercial waste, high vehicle traffic, garages, gas/petrol stations, and seepage from chemical laboratories and public institutions. The high level of Cd in the sediment compared to its levels in the water is to be expected for sediments that have been described as a sink or reservoir for pollutants in water (Topi et al., 2012). The concentration of Cd from all sites is significantly different from the background concentrations (23.2 mg/kg). It is also significantly differ World River Sediment background concentrations by EC (3 mg/kg), PEL and CFSG (3.53 mg/kg), and WRSB (0.25 mg/kg) (**Table 2**).

The highest values of Cr are found at the K1 sampling site, which could be attributed to the direct influence of effluents coming from institutional wastes like seepage of laboratory and electroplating (Song et al., 2000; Baig et al., 2013). The high level of Cr (VI) in sediment is expected more than water, mainly in the mobile environment but has low mobility under moderately oxidizing and reducing conditions and nearly neutral pH (Decena et al., 2018) and discharged into the nearby natural waters bodies without treatment (**Figure 3**). Such contaminated environmental resources could be hazardous to humans, wildlife, and highly toxic to aquatic life (Olafisoye et al., 2013).

As explained on **Figure 4**, the highest values of Pb was recorded in Dololo (D1-D5), and Kito (K1-K3) streams with areas near intensive anthropogenic activities like garages, gasoline stations, chemical laboratories, and construction industries (Wuana and Okieimen, 2011). The other factors contributing to higher Pb concentrations might be due to the waste generated from public and private institutions released



FIGURE 3 | Mean metals contents in mg/kg (mean ± standard deviations) in sediments samples collected from stations in Awetu (A), Dololo (B), Kitto (C) and Boye (D) streams in Awetu watershed.



to the adjacent water bodies without treatment and old water pipelines (Adela et al., 2012). These values exceed the background concentrations of this study (952 mg/kg), EC (300 mg/kg), PEL and CFSG (91.3 mg/kg) and WRSB (48 mg/kg) guideline values (Nadu et al., 2014). Exposure to elevated concentration of Pb through drinking water or food may damage the kidney, increase blood pressure, and anemia (Basim and Khoshnood, 2016). Pb is a threat to public health even at very low concentrations because it usually bioaccumulates in the body. Specifically, it is essentially





harmful to children under six and causes mental and physical retardation (Kathuria, 2017).

The highest values of Hg is recorded D2 (6 mg/kg) followed by B3 (5.6 mg/kg) and D4 (5.2 mg/kg). This might be due to the anthropogenic emission sources of mercury, mostly from solid wastes (municipal and medical) incineration (Wang et al., 2004). As shown in **Table 2**, these values exceeded the background concentrations of this study (4.4 mg/kg), PEL (0.486 mg/kg) and WRSB (0.4 mg/kg).

The concentration of heavy metals alone does not provide sufficient information on the mobility and potential toxicity of contaminants or their potentially harmful effects on the environment, because different chemicals can inactivate and promote synergistic effects. Igeo, CF, and PLI determine the potentially harmful effect of heavy metals in the environment (**Table 3**). These indices provide a basis for assessing the effects of sediment-associated contaminants in sediment compared to the values concerning each index (Moore et al., 2011).

Contamination Level of the Tributaries Against the Different Indices

Based on the I_{geo} values, As and Pb made moderate pollution (0), Cr moderately to heavily polluted, Cd extreme pollution, Hg unpolluted to moderately polluted in the Dololo stream (**Table 4**).

The CF and PLI values are indicated in **Figure 5**. These values for heavy metals in sediments are essential to monitoring the enrichment of a given metal over time (Ali et al., 2016). The results indicate that the CF of As in all streams except Boye (0.9) shows a moderate degree of pollution ($1 \le CF < 3$) while Awetu, Dololo, and kito streams with the values of 1.08, 1.09, and 2.33 respectively showing a moderate degree of pollution ($1 \le CF < 3$). The CF of Cd in Awetu and Dololo streams are 1.62 and 1.1, respectively, which shows the sediment is in the status of a moderate degree of pollution ($1 \le CF < 3$). CF of Cd in the Boye riverine wetland is 3.56, which shows a considerable degree $(3 \le CF < 6)$, and Kito stream is 10.01, which shows a very high degree of pollution (CF \leq 6). The CF of Cr in Awetu stream is 0.95, which indicates the sediment is in a low degree of pollution (CF < 1) while the CF of kito stream is 3.13, which indicates the sediment is in a considerable degree of pollution ($3 \le CF < 6$). Dololo and Boye streams with the CF values of 1.06 and 1.33 respectively, show a moderate degree ($1 \le CF < 3$). The CF values of Pb in Awetu, Dololo, Kito, and Boye streams are 1.25, 1.17, 2.84, and 1.47 respectively, which shows these streams are in a moderate degree of pollution $(1 \le CF < 3)$ by Pb. The CF values of Hg in Awetu, Dololo, Kito, and Boye streams are 1.06, 1.07, 1.02, and 1.07 respectively, indicating streams are in a moderate degree of pollution $(1 \le CF < 3)$.

The PLI values calculated for each stream revealed the decreasing order of contamination: Kito > Boye > Awetu > Dololo. All the studied streams were found to be polluted (PLI > 1), suggesting inputs from anthropogenic sources. PLI was found to increase in sediments downstream along with Boye and

TABLE 4 | I_{geo} values for heavy metals in surface sediments of Awetu watershed streams.

Streams	As	Cd	Cr	Pb	Hg
Awetu	-0.47 (0)	0.11 (1)	-0.64 (0)	-0.27 (0)	-0.49 (0)
Dololo	1.61 (1)	7.5 (6)	2.19 (3)	1.95 (2)	0.72 (1)
Kito	0.63 (1)	2.74 (3)	1.09 (2)	0.92 (1)	-0.56 (0)
Boye	-0.74 (0)	1.25 (2)	-0.17 (0)	-0.03 (0)	-0.49 (0)

			e, _e, p, ee.						
	DO	EC	рН	Turbidity	As	Cd	Cr	Pb	Hg
DO	1								
EC	-0.227	1							
рН	0.224	0.087	1						
Turbidity	0.444*	0.046	-0.227	1					
As	-0.040	0.467*	-0.054	0.574**	1				
Cd	-0.218	0.484*	-0.170	0.441	0.901**	1			
Cr	-0.166	0.448*	-0.268	0.513*	0.916**	0.978**	1		
Pb	-0.094	0.411	-0.172	0.555*	0.912**	0.963**	0.958**	1	
Hg	-0.241	-0.121	-0.121	-0.151	-0.209	-0.220	-0.186	-0.235	1

TABLE 5 | Correlation between heavy metals, DO, EC, pH, and Turbidity in sediment (n = 20).

*, ** Correlation is significant at 0.05 and 0.01 level respectively.

TABLE 6 | Correlation between heavy metals in sediment (n = 20).

	As	Cd	Cr	Pb	Hg
As	1				
Cd	0.901 ^a	1			
Cr	0.916 ^a	0.978 ^a	1		
Pb	0.912 ^a	0.963 ^a	0.958 ^a	1	
Hg	-0.065	-0.018	-0.084	-0.103	1

^aCorrelation is significant at the 0.01 level.

Awetu channelized streams. The I_{geo} of As shows practically unpolluted in Awetu and Boye, which shows a very low value (<0), unpolluted to moderately polluted in Kito (0–1), and moderately polluted in Dololo (1–2).

The Igeo values calculated for each stream of the studied metals is also indicated in table 4. Accordingly, the concentration level of As indicates unpolluted to moderately polluted status in Awetu and Boye (0-1) and moderately polluted in Dololo and kitto streams (1-2). Cd indicates unpolluted to moderately polluted status in Awetu (0-1), moderately polluted in Boye (1-2), moderately to heavily polluted in Kito (2-3), and extremely polluted in Dololo streams (>5). Cr shows practically unpolluted in Awetu and Boye (<0), moderately polluted in Kito (1–2), and moderately to heavily polluted in Dololo streams (3-4). Pb shows practically unpolluted in Awetu and Boye (<0), unpolluted to moderately polluted in Kito (1-2), and moderately polluted in Dololo (1-2) streams. Hg shows practically unpolluted in Awetu, Kito, and Boye streams (<0) and unpolluted to moderately polluted in the Dololo stream (0-1). The anthropogenic pressures largely contribute to the observed heavy metal concentrations rather than the mineralogical composition of the crust along the watershed. The higher values of Igeo were observed in samples from the Dololo stream as compared to other streams, which might be due to the kind and magnitude of anthropogenic pressures on the areas and differences in industrial and household waste management. However, both factors are likely to be the most significant cause of the observed differences. When compared with results from other developing nations, the concentrations measured in this study are much higher than rural and urban streams (Bai et al., 2011).

The box and whisker plot (Figure 6) indicates that relatively higher heavy metal concentrations were identified in the Dololo and Kito streams, except for Hg. Pairwise permutation test using R statistical package (R Core Team, 2019) showed no significant concentration difference between Awetu and Boye, and Dololo and Kito streams for As, Cd, and Cr concentrations between. These streams are known to receive wastewater discharges from Jimma University compasses. For example, the Dololo stream receives untreated wastewater discharge from the main campus and the Jimma Medical Center. Similarly, Kito stream receives the wastewater effluent from the waste stabilization pond of Jimma Institute of Technology. These campuses have different laboratories which could be a source of these contaminants. The highest level of electrical conductivity at Dololo, followed by Kito streams, could be a clear indication that the sources of contamination could be from those campuses. Though the further investigation of tracing which exact laboratory source from these campuses is crucial, this might be the possible reason why these two differently located streams showed a similar pattern of heavy metal contamination.

The interrelationship among metals in sediment of the aquatic environment provided important information on sources and pathways of variables (heavy metals). The result of bivariate correlations between parameters is shown in Table 5, 6. The elemental pairs, As and Cd (r = 0.901, p < 0.01); As and Cr (r =0.916, *p* < 0.01); As and Pb (0.912, *p* < 0.01); Cd and Cr (0.978, *p* < 0.01); Cd and Pb (0.963, *p* < 0.01); Cr and Pb (r = 0.958, *p* < 0.01) showed significant positive correlation. This result might indicate that similar pollution sources of two or more heavy metals. The elemental association may signify that each paired elements have a common sink in the stream sediments discharged from municipal wastes and agricultural inputs (Bhuyan et al., 2017). Metal and physicochemical associations show pairs EC/As (r = 0.467, *p* < 0.05), EC/Cd (r = 0.484, *p* < 0.05) and EC/Cr (0.448, *p* < 0.05) are correlated with each other, whereas the rest are not significantly correlated. The positive correlations of heavy metals concentration of the sediment with EC might be attributed to anthropogenic impacts (Alghobar and Suresha, 2017). Turbidity has correlated with As (r = 0.574, *p* < 0.01), Cr (r = 0.513, *p* < 0.05) and Pb (r = 555, p < 0.05). DO and pH are negatively correlated with all studied heavy metals in the stream sediment, which significantly affects heavy metals (Sekabira et al., 2010). The concentrations of heavy metals were significantly correlated

TABLE 7 | Heavy metal concentrations in sediment samples from ou study and from other studies from river sediment.

Study area	Mean	concent the sed	rations of iment (m	f heavy m g/Kg dw)	etals in	Source of contamination identified	References
	As	Cd	Cr	Pb	Hg		
Awetu watershed, Ethiopia	623.3	151.1	375.0	2005.9	4.6	Untreated municipal wastewater discharge and solidewaste damping sites	The present study
Tisza River and its tributaries, Serbia	а	а	11.37	19.13	а	Industrial activities, municipal sewage discharges, and agricultural inputs	Manojlovic (2009)
Yangtze River intertidal zone, China	а	0.261	78.9	27.3	а	Not mentioned	Zhang et al. (2009)
Yilong Lake wetland, China	15.46	0.76	86.73	53.19	а	Industrial discharges, coal and lead mining areas, and domestic effluents	Bai et al. (2011)
The northern part of Lake Babrukas, Lithuania	24.59	31.79	76.97	7.25	10.25	Municipal wastewater discharge	Raulinaitis (2012)
Butrinti Lagoon, Albania	а	0.125	56.5	31.2	0.16	Pedo-geological processes and human activity	Topi et al. (2012)
Upper Tigris River, Turkey	5.90	3.02	135.81	380.4	а	Copper mine plant	Varol and Şen, (2012)
Jinjiang river, China	9.6	1.6	32.2	95.1	0.08	Residential, industrial, and agricultural pollutant loads	Chen et al. (2013)
Turag, Buriganga and Shitalakha rivers, Bangladesh	35	17	695	356	а	From residents and household waste and sewage treatment facility	M. S. Islam et al. (2014)
Korotoa River of Bogra district urban area, Bangladesh	25	1.2	109	58	а	Domestic raw sewage, household waste, and industrial wastes	S. Islam et al., (2015)
Karnaphuli River, Bangladesh	81.1	2.01	20.3	43.7	а	Industrial effluent and domestic sewage discharge	Ali et al. (2016)
Lake Edku, Egypt	а	а	113.1	44.6	а	Run-off from agricultural fields	Waheshi (2017)
Jiaozhou Bay rivers, China	7.7	0.159	69.3	20.2	а	Industrial and domestic waste discharge	Xu et al. (2017)
Zhelin Bay, China	а	0.063	23.07	35.7	а	Not mentioned	Gu (2018)
Mustafakemalpas, Turkey	154	8.78	516	65.6	а	Mining and industrial activities	Omwene et al. (2018)
Miyun District, China	15.1	0.18	31.5	33.01	а	Not mentioned	Pan et al. (2018)

^aNo data.

with pH (negatively), turbidity (positively) except for Hg (negatively).

The absence of a significant correlation between the heavy metals and pH might be due to variation in sediment composition that implies, minerals are the only factors controlling the fixation of heavy metals (Ali et al., 2016). Other studies also showed that there is a significant variation in the concentrations of heavy metals based on the type of waste discharge at different sites (Nagajyoti et al., 2010). Heavy metals originate by their natural occurrence in soil and their concentrations depend on the rock type and environmental conditions, activating the weathering processes. Our findings showed elevated heavy metal concentration than many other published studies (**Table 7**). This indicates how much the aquatic environment and probably the biota is at risk of elevated heavy metal contamination.

Besides the teaching institutions' wastewater discharge, runoff and waste discharges from different sources, such as car washing facilities, hotels, garages are directly entering the stream system. This could be the main reason for the extremely exceeded Pb concentration than the other studies done in China, Turkey, and Bangladesh. Similarly, Liao et al. (2017) found that the concentrations of heavy metals were upto 120 fold than the background concentration, which was mainly sourced from mining activities. However, for our case, multiple sources of aquatic pollution might have contributed to these very high concentrations of heavy metals.

CONCLUSION

An investigation was made on sediments of streams from Awatu watershed that showed heavy metal contamination (As. Cd, Cr, Pb, and Hg) at various degrees in southwestern Ethiopia. The concentration of the metals in sediments decreased in the order of Pb > As > Cr > Cr > Hg. The concentrations of heavy metals in the streams of Awetu watershed sediments were remarkably high and varied among sampling points. The data analyses by Igeo, CF, and PLI values showed contamination of the sediments by all the studied heavy metals which exceeded the limits of average world concentration. Correlation analysis shows that heavy metals have common anthropogenic pollution sources that resulted in rapid urbanization and inappropriate waste discharge. As a result, the higher concentration level of heavy metals in sediment poses a risk of water pollution during sediment disturbance or changes in sediment chemistry, which eventually intermix and transferred easily with the water in the hydraulic movement. The results suggested that special attention must be given to the issue of heavy metal pollution since a considerable portion of elements in sediments is likely to release back into the water column. Therefore, public institution and commencial centers shold properly manage their waste discharge. Besides, authorities of in low income countries should inforce waste generating institutions to have waste treatment facilities before releasing to the environment. Studying seasonal influence of heavy meatal concentration in sedimets is recommended.

DATA AVAILABILITY STATEMENT

The raw data supporting the conclusions of this article will be made available by the authors, without undue reservation.

AUTHOR CONTRIBUTIONS

HA Conceived the idea, draft the methodology, collect and analyze samples, statistical analysis of the data AA Review the

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Conflict of Interest: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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