



# Corrigendum: Evaluation of General Synthesis Procedures for Bioflavonoid–Metal Complexes in Air-Saturated Alkaline Solutions

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## A Corrigendum on

### Evaluation of General Synthesis Procedures for Bioflavonoid–Metal Complexes in Air-Saturated Alkaline Solutions

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In the original article, there was a mistake in the legends for **Figures 3, 4, 7, and 8** as published. In these Figures, marks such as a, b, c, d, e, f, and g have not been explained. The updated legends appear below.

**Figure 3.** (A) UV-vis spectra and (B) main peak positions of DHM refluxed for different times. (a) DHM; (b–g) DHM refluxing for 30, 60, 90, 120, 150, and 180 min.

**Figure 4.** (A) UV-vis spectra and (B) main peak positions of DHM stirred for different times at pH = 8.2. (a) DHM; (b–g) DHM stirring for 1, 3, 5, 10, 20, and 30 min in pH 8.2.

**Figure 7.** EPR spectra of DHM in air-, nitrogen-, and oxygen-saturated alkaline solutions. (a) DHM only; (b) DHM + DMPO; (c–e) DHM + DMPO, in nitrogen-, air-, oxygen-saturated alkaline solution.

**Figure 8.** EPR spectra of superoxide-anion radical generation from DMSO reacted with a base in the presence of oxygen. (a) DMSO + DMPO + O<sub>2</sub>; (b) Na<sup>+</sup>PhO<sup>-</sup> + DMPO + O<sub>2</sub>; (c) Na<sup>+</sup>PhO<sup>-</sup> + O<sub>2</sub>; (d) DMSO + Na<sup>+</sup>PhO<sup>-</sup> + O<sub>2</sub>; (e) DMSO + Na<sup>+</sup>PhO<sup>-</sup> + DMPO + O<sub>2</sub>.

The authors apologize for this error and state that this does not change the scientific conclusions of the article in any way. The original article has been updated.

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