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# Development of a highly sensitive lateral flow strip device for nucleic acid detection using molecular beacons

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Extensive research is focused on the development of highly sensitive, rapid on-site diagnostic devices. The lateral flow strip (LFS) is a paper-based point-of-care diagnostic device, which is highly promising because of its ease of use and low cost. Despite these advantages, LFS device is still less popular than other methods such as enzyme-linked immunosorbent assay (ELISA) or real-time polymerase chain reaction (qPCR) due to its low sensitivity. Here, we have developed a fluorescence-based lateral flow strip (f-LFS) device for DNA detection using a molecular beacon (MB), a short hairpin-forming DNA strand tagged with a fluorophore-quencher pair. Each paper and membrane component of f-LFS device was carefully selected based on their physicochemical properties including porosity, surface functionality, and autofluorescence. The limit of detection (LOD) of this device was substantially improved to 2.1 fg/mL by adding MgCl<sub>2</sub> to the reaction buffer and narrowing the test membrane dimension. Also, a portable fluorescence detection system for f-LFS was developed using a multi-pixel photon counter (MPPC), a sensitive detector detecting the signal on site. We anticipate that this highly sensitive paper-based diagnostic device can be utilized for on-site diagnosis of various diseases.

## KEYWORDS

paper-based diagnostic device, molecular beacon (MB), nucleic acid detection, point-of-care (POC), lateral flow strip (LFS)

## 1 Introduction

Various methods are available for biomolecular diagnoses, such as real-time polymerase chain reaction (qPCR), high-performance liquid chromatography (HPLC), enzyme-linked immunosorbent assay (ELISA), etc (Matuszewski et al., 2003; Postollec et al., 2011; Aydin, 2015; Choi et al., 2019). Although these techniques enable precise and accurate detection, they require expertise and involve complicated procedures. To overcome these drawbacks, researchers have developed various microfluidic devices,

e.g. digital microfluidics, modular microfluidics, and paper-based microdevices which utilize loop-mediated isothermal amplification (LAMP) of DNA or RNA, surface-enhanced Raman scattering (SERS) detection, or qPCR (Wang et al., 2009; Lo et al., 2013; Choi et al., 2015; Song et al., 2019). Amongst these recently developed methods, paper-based diagnostic devices (i.e., lateral flow strip, LFS) are some of the most promising technologies (Jafry et al., 2017; Cho et al., 2019; Jafry et al., 2019). They are capable of fast and on-site detection of various conditions or diseases; e.g., infectious organisms, drug addiction, and disease monitoring crucial to human physiology, along with animal disease and environmental testing (Anfossi et al., 2018; Land et al., 2019; Lim et al., 2019; Di Nardo et al., 2021).

The commercially available LFS device consists of cellulose-based sample pad and absorbent pad, a glass fiber-based conjugate pad, and a nitrocellulose-based test membrane (St John and Price, 2014; Sajid et al., 2015). The test line in the middle of the test membrane contains antibody-conjugated gold nanoparticles, which give rise to a color change when bound with antigen molecules. There are two distinct detection methods for LFS; sandwich and competitive immunoassays. The sandwich immunoassay is a method using two antibodies, which bind to different sites on the antigen (Cox et al., 2019). A labeled detector antibody moves through the test membrane with the antigens of interest. The antigen-antibody complex is captured by an immobilized, antigen-specific antibody, resulting in a colored line (Ching, 2015). The competitive immunoassay is based on antibody-antigen interactions in which the number of antigen-binding sites on the antibody is limited (Clarke, 2004). In this assay, the antigen of interest is immobilized on the test membrane. If antigen is present in the sample, it will bind with the detector antibody, and the complex will flow past the immobilized antigen, resulting in no colored line (Ching, 2015). This LFS device has been used for detecting not only antigen but also nucleic acids (Chen and Yang, 2015; Yao et al., 2019). In these applications, nucleic acid probes conjugated to the nanoparticles bind with their complementary target strands to cause nanoparticle aggregation and coloring of the test line. However, these conventional LFS devices suffer from low throughput (single target detection per loading) and low sensitivity.

To address these drawbacks, numerous devices have been developed since LFS device was invented and patented in the 1970s (Deutsch and Mead, 1978). To detect multiple targets for LFS, some research groups have reported the use of multiple test lines (Song et al., 2014; Peng et al., 2016; Wang et al., 2018). However, such LFS devices need a larger footprint and sample volume, and the nonlinear velocity reduction along a longer test line produces an inconsistent reaction time. Therefore, other types of LFS were developed to detect multiple antigens in one test line using nanoparticles of multiple shapes and colors. For example, several groups fabricated such LFS devices using red

nanoparticles and blue nanoparticles (Kim et al., 2019; Rodriguez-Quijada et al., 2020). They conjugated different antibodies to the red nanoparticles and blue nanoparticles to detect multiple targets on one test line. When one antigen is detected, the test line turns blue or red, and when two antigens are detected simultaneously, the test line turns purple.

Additionally, the general limit of detection (LOD) of LFS is as low as 0.1–10 ng/mL when it is detected by color appearance or change (Nguyen et al., 2020). Several studies have been conducted to modify the materials or the detection method to improve LOD. Parolo et al. improved LOD to ~ 200 pg/mL by functionalizing the surface of gold nanoparticles with an enzyme called horseradish peroxidase and adding a metal enhancer (Parolo et al., 2013). The enzyme produces insoluble and non-flowing chromogens, which are then captured by the enhancers on the gold nanoparticles, leading to a 5-fold increase in the color intensity of the test line. The color intensity of LFS was also enhanced by a double-targeted detection method, where antibodies of a target antigen and proteins are conjugated to small-sized gold nanoparticles, and antibodies of the protein to large-sized gold nanoparticles (Choi et al., 2010). The smaller gold nanoparticles were caught by antigens at the test line, and sequentially the larger nanoparticles are caught by the smaller gold nanoparticles. This double-targeted detection method increased the sensitivity ~ 10- to 100-fold (LOD ~ 10–100 pg/mL). Takalkar et al. enhanced the LOD of LFS ~15-fold up to ~ 67 pg/mL by combining large silica nanorods with a large number of gold nanoparticles (Takalkar et al., 2016).

Recently, researchers employed fluorescence to improve sensitivity using the portable fluorescence reader or fluorescence scanner, etc (Han et al., 2020; Lee et al., 2021). Berlina et al. and Qu et al. used quantum dots instead of gold nanoparticles for LFS (Berlina et al., 2013; Qu et al., 2016). They could quantitatively determine the antigen level from the signal-to-noise ratio (SNR) of the fluorescence signal. Lou et al. attached a fluorophore to a specific part of the antibody and use this to functionalize polystyrene microspheres (Lou et al., 2018). This functionalization greatly increased the fluorescence intensity of the polystyrene microspheres, resulting in ~31-fold increase in sensitivity (LOD ~ 32 pg/mL). Wang et al. fabricated a magnetic particle coated with three layers of quantum dots; the Fe<sub>3</sub>O<sub>4</sub> triple QD-shell nanocomposite (named as MagTQD) (Wang et al., 2021). This nanocomposite particle had numerous quantum dots on the surface, which could improve the detection limit by about a factor of 2000 (LOD ~ 0.5 pg/mL). Ao et al. fabricated the LFS device using the near infrared-II (NIR-II, 1,000–1,700 nm) emitting lead sulfide quantum dots (Ao et al., 2022). They concentrated the quantum dots in porous silica particles to amplify the fluorescence intensity and improved the sensitivity of the LFS by about a factor of 9 (LOD ~ 110 pg/mL).

In this study, we report a sensitive and rapid point-of-care fluorescence-based lateral flow strip device (f-LFS device). We used molecular beacons (MB), hairpin-shaped molecules

composed of a single-stranded oligonucleotide probe labeled with a fluorophore and a quencher, to detect human immunodeficiency virus type 1 (HIV-1) as the target. To improve the detection sensitivity, the materials for the LFS device including paper and membrane were selected based on measurements of porosity, pore size, total intrusion volume, and autofluorescence. And, a portable fluorescence detection system was developed using multi-pixel photon counter (MPPC). The reactivity between the designed MB and the target nucleic acid in solution was enhanced by adding  $MgCl_2$  which serves as a charge screening agent. Furthermore, we improved the sensitivity by reducing the width of the test membrane, thus increasing the target concentration during flow. We expect the f-LFS device to be widely adopted in the diagnostic field due to its rapid processing, high sensitivity, low cost, and ease of fabrication.

## 2 Materials and methods

### 2.1 Materials and reagents

Multiple Grade 1 chromatography papers, CF3, CF5, standard 17, glass fiber filter, and FF170HP were purchased from Whatman, United Kingdom, and AP1004700 and GFCP203000 were purchased from Millipore, United States. Deoxyribonucleic acid (DNA) oligos and Molecular Beacon (MB) conjugated with fluorophore (6-carboxyfluorescein; 6-FAM)-quencher (black hole quencher-1; BHQ-1) pair were purchased from Integrated DNA Technologies (IDT), United States. Sodium chloroacetate, N-(3-Dimethylaminopropyl)-N'-ethyl carbodiimide hydrochloride (EDC), N-hydroxysuccinimide (NHS), cystamine dihydrochloride, iodoacetic anhydride, 1,2-dichloroethane and tris(2-carboxyethyl)phosphine hydrochloride (TCEP) were purchased from Sigma-Aldrich, United States. Sodium hydroxide and acetone were purchased from Samchun Chemicals, Rep. of Korea. Polyethylene glycol spacer (maleimide-polyethylene glycol (PEG)- $NH_2$ , MW 2000) was purchased from Nanocs, United States. N,N-Dimethylformamide (99.8% purity) was purchased from Alfa Aesar, United States. The nucleic acid purifying kit (Oligo clean and concentrator) was purchased from Zymo Research, United States.

### 2.2 Physical characterization of papers and membranes

The paper and membrane characteristics of porosity, pore size, and total intrusion volume were measured using the automatic porosimeter (AutoPore IV, Micromeritics, United States) at Gacheon University, Rep. of Korea. A paper was placed inside a glass penetrometer bulb, which was then

sealed and loaded into the automatic porosimeter. Afterwards, pressure was applied to permeate mercury through the paper, and the porosity, pore size, and total intrusion volume of the paper were calculated. The fiber width was calculated by averaging the fiber radii observed by the scanning electron microscope (SEM, JSM7000F, JEOL Ltd., Japan) at Sungkyunkwan University, Rep. of Korea.

For imaging, an inverted microscope (Ti-U, Nikon, Japan) with a 2X magnification objective lens was used. For fluorescence imaging of 6-carboxyfluorescein (6-FAM), we used a mercury lamp as the excitation source with an excitation bandpass filter (465–495 nm) and an emission filter (513–557 nm), the center wavelengths of which overlap with the excitation and emission wavelengths of 6-carboxyfluorescein (6-FAM), respectively. Images were taken using a CMOS camera (UCMOS03100KPA, Hangzhou ToupTek Photonics Co., Ltd., China).

### 2.3 Functionalization of molecular beacon for test membrane immobilization

For MB to be immobilized on glass fiber, N, N-Bis( $\alpha$ -iodoacetyl)-2, 2'-dithiobis (ethylamine) (BIDBE) was synthesized to create the thiol group (-SH) on MB, following the literature procedure (Ludueña et al., 1981). First, 18 mM of cystamine dihydrochloride solution (1 mL) in 0.1 M of NaOH was mixed with 220 mM of iodoacetic anhydride (0.25 mL) in 1, 2-dichloroethane solution, and the mixture was vortexed for 1 min. The white precipitate in this solution was collected *via* centrifugation (Centrifuge 5430R, Eppendorf, Germany) for 15 min (12,000 rcf and 20°C) and dried in a vacuum oven. After drying, the precipitate was dissolved in acetone and centrifuged (12,000 rcf and 15°C) for 15 min to remove any remaining iodoacetate. The supernatant was collected from the centrifuged solution and dried to obtain white BIDBE powder. For linking BIDBE to phosphorothioate on MB and reducing the disulfide bond in BIDBE, we followed a previously published protocol (Lee et al., 2007; Lee et al., 2010). To make BIDBE conjugated MB, 1 mM of MB containing phosphorothioate modification (10  $\mu$ L) was mixed with 100 mM of BIDBE solution (20  $\mu$ L, dissolved in DMF) and 10 mM of phosphate buffer (36  $\mu$ L, pH 7.0) followed by heating at 50°C for 5–6 h. After heating, BIDBE conjugated MB (BIDBE-MB) was washed and collected *via* centrifugation using the Oligo clean and concentrator. To cleave the disulfide bond from BIDBE, 12  $\mu$ M of BIDBE-MB (100  $\mu$ L), 50 mM of acetate buffer (1  $\mu$ L, pH 5.2) and 1 mM of TCEP (5  $\mu$ L) were mixed and incubated at room temperature for 1 h, followed by washing and collecting *via* centrifugation using the Oligo clean and concentrator kit again.

To attach thiol-modified MB to glass fibers in the test membrane, we terminated the glass surface with maleimide groups in a two-step process. The hydroxyl group (-OH) of

glass fiber was substituted with a carboxyl group (-COOH) in a reaction with a solution of 3 M of sodium chloroacetate and 10 wt% of sodium hydroxides at 90°C for 4 h, followed by washing with DI water for 10 min twice. The surface-modified glass fiber was immersed in 10 mM of EDC solution followed by 10 mM of NHS solution at room temperature for 3 h. EDC and NHS solutions were used at 1 mL per 1 cm<sup>2</sup> of glass fiber. Afterwards, the substrate was washed thoroughly with DI water and immersed in 10 nM of maleimide-PEG-NH<sub>2</sub> solution at room temperature for 12 h to functionalize the glass fiber surface. After maleimide incorporation, the surface was washed using DI water. Finally, MB-thiol was applied to the surface for conjugation to the PEG polymer on glass fiber.

## 2.4 f-LFS device fabrication and fluorescence detection

A CO<sub>2</sub> laser cutter (VLS 6.60, Universal Laser Systems, United States) and a blade cutter (silhouette cameo 2, Silhouette America, United States) were used to cut the selected papers and membranes after characterization. To construct the f-LFS device, the sample pad, test membrane and absorbent pad were fixed to a lateral flow backing card (Adhesive polyester film, Fancylobby, Rep. of Korea) with an overlap of 1 mm between them. The test line was formed by drawing MB solution onto the membrane using an in-house linear motor control system connected with the reagent dispenser (Claremont Biosolutions LLC, United States) and a syringe pump (PHD Ultra Advanced Syringe Pump, Harvard Apparatus, United States). After several trials, we found the optimal flow rate and concentration of MB solution (20 µL/min at 0.5 µM), linear motor speed (150 mm/min), and number of drawings (one time) needed to form the test line. The inverted microscope and a multi-pixel photon counter (MPPC) (C11208 MPPC module, Hamamatsu Photonics, Japan) were used to detect the fluorescence intensity from the f-LFS device and MB before and after binding with target DNA. To check the effect of the MgCl<sub>2</sub> on the interaction between the MB and target, a multimode microplate reader (Varioskan LUX, ThermoFisher, United States) was used to measure the fluorescence intensity in DI water. To evaluate the selectivity and sensitivity of f-LFS, 150 µL of the target and non-target DNA molecules were placed on the sample pad at concentrations ranging from 0.21 to 210 fg/mL. The fluorescence intensities with and without DNA at the test line were measured using the fluorescence measurement system consists of MPPC, and the signal-to-noise ratio (SNR) was calculated.

## 3 Results and discussion

For the conventional LFS device, the conjugate pad holds nanoparticles functionalized with antibodies or nucleic acid probes. In comparison, our f-LFS device does not have a

conjugate pad, and MBs were directly attached to the fibers of the test membrane. Our f-LFS device (0.5 cm × 9.8 cm; 1-mm overlap between paper and membrane) consists of a sample pad (0.5 cm × 2 cm), MB-functionalized test membrane (0.5 cm × 4 cm), and an absorbent pad (0.5 cm × 4 cm), which are all placed on an adhesive backing card as illustrated in Figure 1A. Activating the paper surface with carboxyl groups and subsequent reactions with EDC/NHS and maleimide-PEG-NH<sub>2</sub> expose maleimide groups on the surface. These maleimides react with the thiol groups on MB, leading to MB-anchored paper shown in Figure 1B.

The schematic shown in Figure 1C represents the mechanism for detecting the target nucleic acid using MB. HIV-1, one of the viruses that require continuous monitoring of infection and on-site diagnosis has various mutations, such as U3, R, US and the U3 region has low variability among the gene regions of HIV-1 (Tatt et al., 2001; Mbondji-Wonje et al., 2018). Therefore, the U3 region of HIV-1 was selected as the target for detecting HIV-1 and our MB is based on the U3 region of HIV-1 (5'-CTGGCCCTGGTGTGTAGTTCTGCC-3') with the addition of 5'-AG-3' to the 3'-end for detecting short nucleic acid target molecules, such as microRNA, which are biomarkers for disease detection (Lee et al., 2022). The complementary sequences 5'-CTGGC-3' and 5'-GCCAG-3' at the ends form the stem, and the rest of the sequence forms the loop. The fluorophore, 6-FAM, and the quencher, BHQ-1, were incorporated at 5'- and 3'-end respectively. Using the OligoAnalyzer™ Tool (IDT, United States), the Gibbs free energy for binding of the MB to its complementary target (5'-GGCAGAACTACACACCAGGGCCAGGGATCA-3') is calculated to be -47.64 kcal/mol. We also designed a non-target DNA with the sequence 5'-GCTTTCCTTACGTGACGTGATCGGAGTCT-3', whose Gibbs free energy for binding to MB is -4.64 kcal/mol, much weaker than the target as expected. The free energy between MBs of the same sequence is also much lower at -9.28 kcal/mol. Hence, fluorescence emission from MB predominantly arises from binding with target DNA.

The portable fluorescence detection system was developed, as shown in Figure 1D, to measure the selectivity and sensitivity of fluorescence, quantitatively. The light source of a LED light module (29 mm-3030-7 LED-CL, WELED, Rep. of Korea) was used as the excitation source with the excitation band-pass filter (465–495 nm) and the emission band-pass filter (513–557 nm) for detecting 6-FAM functionalized in MB. Also, the system had a convex plano lens to focus and collect the emission from the test line only. And, a very sensitive MPPC was used for fluorescence measurement unlike most strip tests which are detected with naked eyes and have sensitivity issues, so that the control line was not introduced in the developed LFS (Kang and Kim, 2020; Jung et al., 2022).

To fabricate the f-LFS device, we first selected suitable paper and membrane materials. The selected materials and their

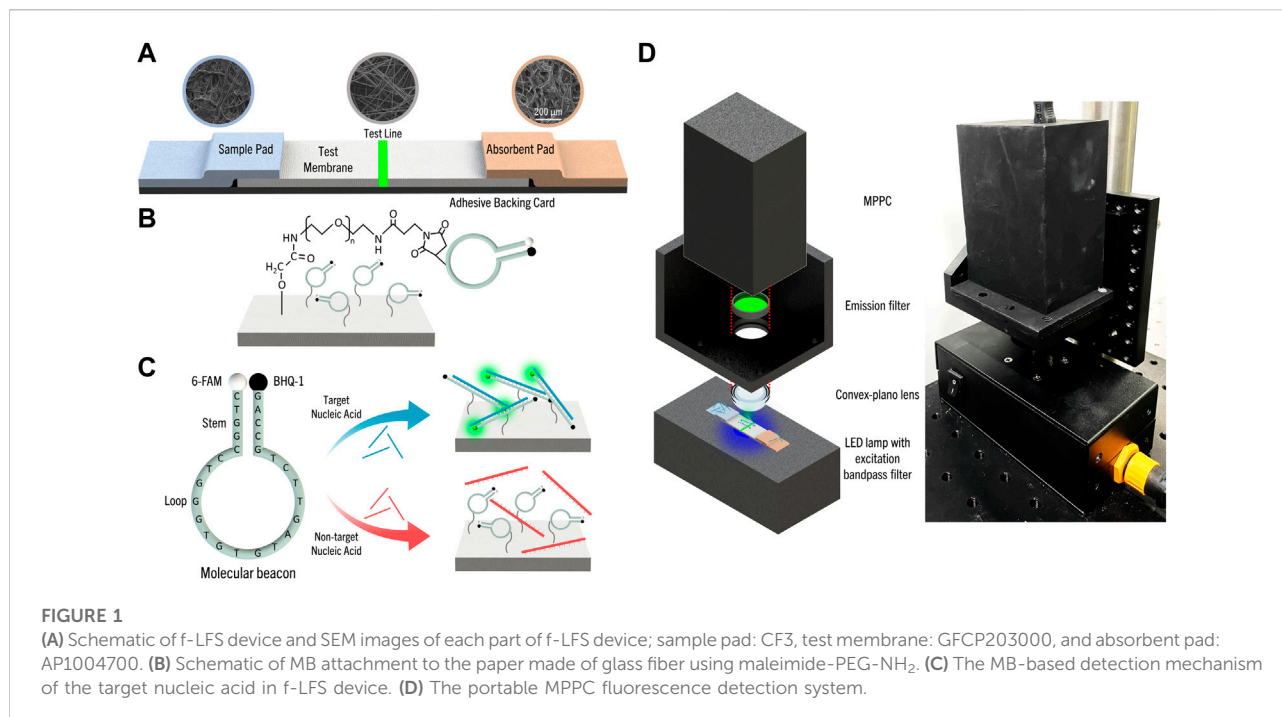


TABLE 1 Physical characteristics of papers for choosing sample pad, absorbent pad, and test membrane.

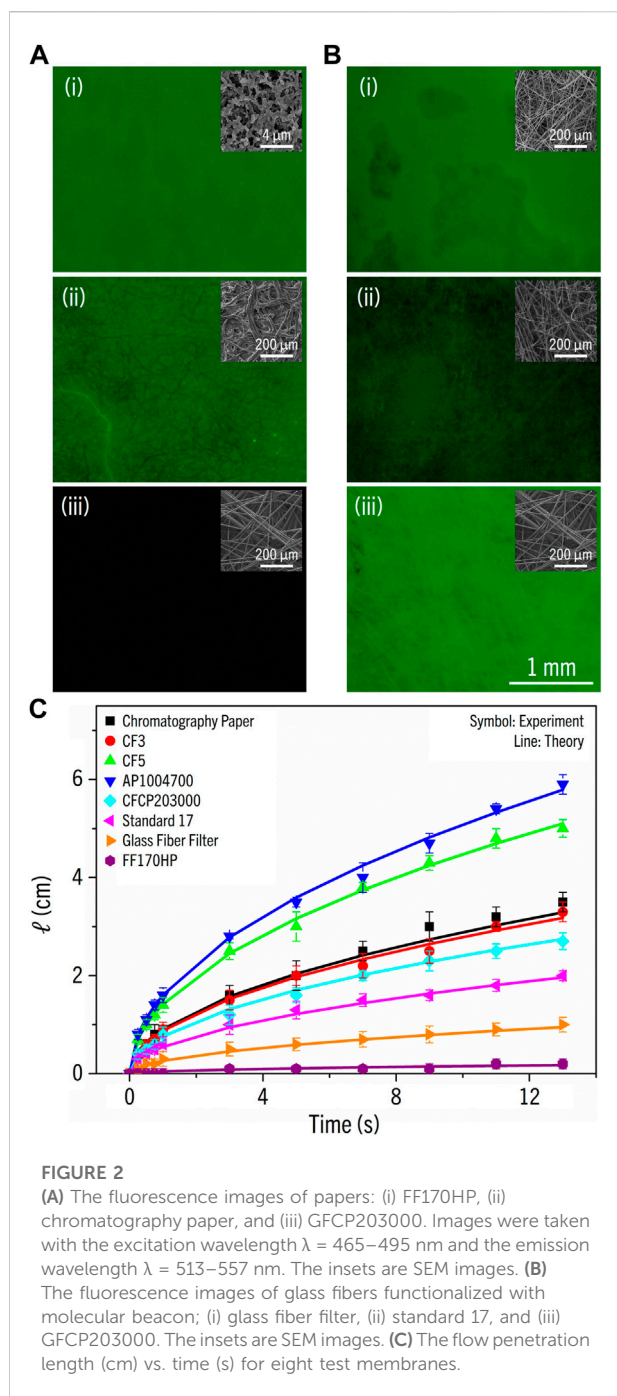
	Product	Porosity	Pore diameter (μm)	Total intrusion volume (mL/g)	Fiber width (μm)	Material
Sample pad and Absorbent pad	Chromatography paper (Whatman)	0.585	11	1.1814	10.0	Cellulose
	CF3 (Whatman)	0.589	3.2	1.0502	5.15	
	CF5 (Whatman)	0.716	4.9	1.7929	6.17	
	AP1004700 (Millipore)	0.767	7.7	2.2811	6.97	
Test membrane	Chromatography paper (Whatman)	0.585	11	1.1814	10.0	Cellulose
	CF3 (Whatman)	0.589	3.2	1.0502	5.15	
	CF5 (Whatman)	0.716	4.9	1.7929	6.17	
	AP1004700 (Millipore)	0.767	7.7	2.2811	6.97	
	GFCP203000 (Millipore)	0.909	68	6.7564	3.95	Glass fiber
	Standard 17 (Whatman)	0.875	35	7.4013	2.69	
	Glass fiber filter (Whatman)	0.864	3.9	6.4702	0.47	
	FF170HP (Whatman)	0.454	0.19	0.5975	0.12	Nitrocellulose

characteristics such as the porosity, pore diameter, and total intrusion volume are listed in Table 1. The list includes cellulose (chromatography paper, CF3, CF5, and AP1004700), glass fiber (GFCP203000, standard 17, and glass fiber filter), and nitrocellulose (FF170HP). Here, the porosity is the ratio of the void to the total volume of paper, and the total intrusion volume is the absorbed volume of liquid for unit mass of paper.

The total intrusion volume is the primary factor for selecting the sample pad and absorbent pad. Papers with a large total

intrusion volume retain fluid better than those with a small total intrusion volume. Therefore, small intrusion volume is suitable for the sample pad, but large intrusion volume for the absorbent pad. Among the products, CF3 was selected as the sample pad because of its smallest total intrusion volume, and AP1004700 from Millipore was selected as the absorbent pad for having the largest total intrusion volume.

The absence of autofluorescence of the material is essential for selecting the test membrane. We measured the fluorescence



intensities of the test membrane candidates: (i) FF170HP for nitrocellulose, (ii) chromatography paper for cellulose, and (iii) GFPCP203000 for glass fiber (Figure 2A). Because GFPCP203000 showed the lowest level of autofluorescence, we concluded that glass fiber was the most suitable material for the test membrane. Next, we took images of different glass fibers functionalized with MB (Figure 2B) and calculated their standard deviations of pixel intensity to quantify the uniformity of MB functionalization; (i) glass fiber filter, (ii)

standard 17, and (iii) GFPCP203000. The standard deviation was calculated as  $\sum_{i=1}^{n_p} (I_i - \bar{I})^2 / n_p$ , where  $I_i$  is the measured intensity of each pixel,  $\bar{I}$  is the average intensity of the measured area, and  $n_p$  is the number of pixels. The calculated standard deviations for (i), (ii), and (iii) of Figure 2B were 5.76, 6.07, and 4.84, respectively. Amongst the glass fibers, GFPCP203000 of Figure 2B-(iii) had the most uniform intensity distribution and therefore was selected as the test membrane. Furthermore, as shown in the SEM images (insets, Figure 2B), the fabric structure of GFPCP203000 is the coarsest. We argue that the coarse network structure is beneficial for the fluorescence emission of MB as the distance between neighboring MBs would become longer on average, reducing the chance for intermolecular quenching between a pair of dye and quencher belonging to different MBs.

In lateral flow strip, the reaction time must be optimized for the signal to reflect the true reactivity between MB and the target nucleic acid; if the reaction time is too long, nonspecific reactions may produce a false negative. The fluid flow through a porous medium such as paper where flow penetration is induced by surface tension can be described by the Lucas-Washburn equation. The penetration length is expressed as (Mendez et al., 2010; Jafry et al., 2016):

$$\ell(t) = 2 \sqrt{\left( \frac{k\gamma \cos \theta}{\phi\mu r} \right) t} \quad (1)$$

where  $\ell(t)$  is the penetration length of the fluid,  $k$  is the permeability of the porous medium,  $\gamma$  is the surface tension of the fluid,  $\theta$  is the contact angle,  $\phi$  is the porosity, and  $r$  is the pore radius. The permeability of porous media is represented in terms of pore radius and porosity as (Nabovati et al., 2009):

$$k = a^2 C_1 \left( \sqrt{\frac{1 - \phi_c}{1 - \phi}} - 1 \right)^{C_2} \quad (2)$$

where  $a$  is the fiber radius,  $C_1$  and  $C_2$  are constants related to the geometry of the network, and  $\phi_c$  is the critical porosity (permeation threshold). We set the contact angle of water on paper as  $0^\circ$  (Liukkonen, 1997). The values of  $\phi_c$ ,  $C_1$  and  $C_2$  are  $1 - \pi/2\sqrt{3}$ ,  $16/9\sqrt{6}\pi$ , and  $5/2$ , respectively (Gebart, 1992). The theoretical penetration lengths were evaluated using the values in Table 1 and compared with the experimental results in Figure 2C. The agreement is excellent with a maximum discrepancy of only 9.63%.

The reactivity between MB and the target DNA in a buffer without salt was extremely low, as indicated by the measured fluorescence close to the base line level (blue line, Figure 3A). This low reactivity is expected without salt because of the repulsion between the negatively charged phosphate backbones of MB and target. Hence, we added  $\text{MgCl}_2$  to the buffer, which is known to increase the hybridization rate of DNA due to effective charge screening of DNA by the divalent ions (Misra and Draper, 1999). Consistent with this prediction, the

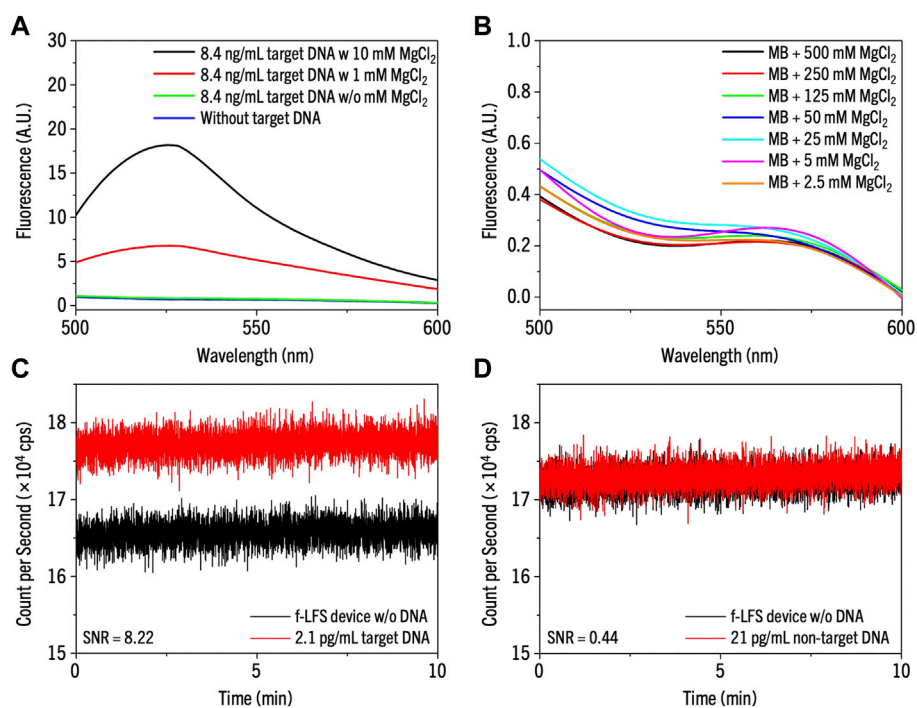


FIGURE 3

(A,B) The reactivity test in the solution of (A) only MB without target DNA (blue line) and MB with 8.4 ng/mL target DNA with  $\text{MgCl}_2$  of 10 mM (black), 1 mM (red), and without  $\text{MgCl}_2$  (green), and (B) MB without target DNA varying the concentration of  $\text{MgCl}_2$  from 2.5 mM to 500 mM. (C,D) The selectivity test of the MB on f-LFS device with 500 mM of  $\text{MgCl}_2$  for (C) 2.1 pg/mL of target DNA and (D) 21 pg/mL of non-target DNA.

fluorescence intensity of MB increased as  $\text{Mg}^{2+}$  concentration increased (Figure 3A). In a control experiment, we also showed that  $\text{MgCl}_2$  alone is not responsible for this fluorescence increase (Figure 3B).

After we confirmed the specificity of hybridization between MB and target DNA in solution, we tested the same reaction on the f-LFS platform. We spotted 2.1 pg/mL of the target DNA and 21 pg/mL of the non-target DNA on the sample pad of the f-LFS device and waited until the DNA sample reached the MB-coated test line on the test membrane. The fluorescence signal was then measured using MPPC. We chose SNR larger than 3 as the positive signal for target DNA detection (Sharma et al., 2014). Figures 3C,D show the results with the target and the non-target DNA solutions, respectively. Although we used a 10-fold lower target DNA concentration (2.1 pg/mL) compared to the non-target DNA (21 pg/mL), the SNR value from target DNA (8.22) was much higher than that from non-target DNA (0.44), suggesting high specificity of our MB-based f-LFS device.

We also tried different methods to increase the sensitivity of the device. We reasoned that increasing the effective concentrations of MB and target DNA at the point of detection would increase the sensitivity, and confining the reactants to a narrower area could bring about such an effect (Eriksson et al., 2019). Hence, we reduced the width of the test

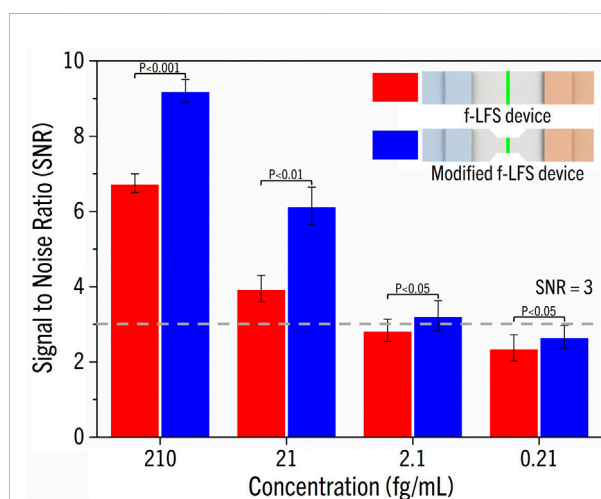


FIGURE 4

SNR showing for both f-LFS and modified f-LFS devices. The LOD of f-LFS is 21 fg/mL, and 2.1 fg/mL for modified f-LFS.

membrane to 0.3 cm and measured the limit of detection (LOD) by applying 150  $\mu\text{L}$  of target DNA solution to the MB-coated test line while varying DNA concentrations from 210 to 0.21 fg/mL.

As shown in [Figure 4](#), SNR of the original f-LFS device dropped below 3 at 2.1 fg/mL of target DNA, and therefore, the LOD was determined to be 21 fg/mL. In comparison, the modified f-LFS device with the narrower test membrane yielded SNR = 3.23 at 2.1 fg/mL of target DNA, resulting in a higher LOD at 2.1 fg/mL. Our f-LFS device thus represents a significant improvement over LFS device of previous studies ([Mao et al., 2009](#); [Liu et al., 2019](#)).

## 4 Conclusion

A novel f-LFS device was fabricated using a sample pad, an absorbent pad, and a test membrane. Each component was selected based on its measured mechanical properties (pore size, porosity and total intrusion volume) and autofluorescence. We selected CF3 with a small total intrusion volume and AP1004700 with a large intrusion volume as the sample pad and the absorbent pad, respectively. To maximize the SNR and avoid false positives, we selected GFPCP203000 with the minimum autofluorescence and large void size as the test membrane. To test f-LFS performance, we selected the U3 region with low variability among the genetic regions from HIV-1 and designed a MB labeled with a dye-quencher pair. The MB was loaded on the test membrane using an in-house test line drawer. We showed that  $MgCl_2$  is critical for enhancing the binding reaction between MB and target DNA, probably because of the screening capability of  $Mg^{2+}$ . The portable and sensitive detection system was developed using MPPC. We tested f-LFS devices with different widths and showed that the narrower one performed better than the wider one, with a 10-fold improvement in LOD from 21 fg/mL to 2.1 fg/mL. In conclusion, we demonstrated the excellent performance of a novel MB-based f-LFS device for DNA detection. We believe that this portable f-LFS device can find many applications for point-of-care disease diagnosis.

## Data availability statement

The datasets presented in this study can be found in online repositories. The names of the repository/repositories and

accession number(s) can be found in the article/supplementary material.

## Author contributions

All authors listed have made a substantial, direct, and intellectual contribution to the work and approved it for publication.

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## Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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