



New Insights Into *Acidithiobacillus thiooxidans* Sulfur Metabolism Through Coupled Gene Expression, Solution Chemistry, Microscopy, and Spectroscopy Analyses

David Camacho^{1†}, Rodolfo Frazao^{2†}, Aurélien Fouillen^{2,3}, Antonio Nanci^{2,3}, B. Franz Lang², Simon C. Apte⁴, Christian Baron^{2†} and Lesley A. Warren^{1,5*†}

¹ School of Geography and Earth Science, Faculty of Science, McMaster University, Hamilton, ON, Canada, ² Department of Biochemistry and Molecular Medicine, Faculty of Medicine, Université de Montréal, Montreal, QC, Canada, ³ Laboratory for the Study of Calcified Tissues and Biomaterials, Faculty of Dentistry, Université de Montréal, Montreal, QC, Canada, ⁴ CSIRO, Land and Water, Lucas Heights, NSW, Australia, ⁵ Department of Civil and Mineral Engineering, Faculty of Applied Science and Engineering, University of Toronto, Toronto, ON, Canada

OPEN ACCESS

Edited by:

Gordon T. Taylor,
Stony Brook University, United States

Reviewed by:

Eric D. van Hullebusch,
Université de Paris, France
Delong Meng,
Central South University, China

*Correspondence:

Lesley A. Warren
lesley.warren@utoronto.ca

† These authors have contributed
equally to this work and share senior
authorship

Specialty section:

This article was submitted to
Microbiological Chemistry
and Geomicrobiology,
a section of the journal
Frontiers in Microbiology

Received: 26 February 2019

Accepted: 27 February 2020

Published: 13 March 2020

Citation:

Camacho D, Frazao R, Fouillen A,
Nanci A, Lang BF, Apte SC, Baron C
and Warren LA (2020) New Insights
Into *Acidithiobacillus thiooxidans*
Sulfur Metabolism Through Coupled
Gene Expression, Solution Chemistry,
Microscopy, and Spectroscopy
Analyses. *Front. Microbiol.* 11:411.
doi: 10.3389/fmicb.2020.00411

Here, we experimentally expand understanding of the reactions and enzymes involved in *Acidithiobacillus thiooxidans* ATCC 19377 S^0 and $S_2O_3^{2-}$ metabolism by developing models that integrate gene expression analyzed by RNA-Seq, solution sulfur speciation, electron microscopy and spectroscopy. The *A. thiooxidans* $S_2O_3^{2-}$ metabolism model involves the conversion of $S_2O_3^{2-}$ to SO_4^{2-} , S^0 and $S_4O_6^{2-}$, mediated by the sulfur oxidase complex (Sox), tetrathionate hydrolase (TetH), sulfide quinone reductase (Sqr), and heterodisulfate reductase (Hdr) proteins. These same proteins, with the addition of rhodanese (Rhd), were identified to convert S^0 to SO_3^{2-} , $S_2O_3^{2-}$ and polythionates in the *A. thiooxidans* S^0 metabolism model. Our combined results shed light onto the important role specifically of TetH in $S_2O_3^{2-}$ metabolism. Also, we show that activity of Hdr proteins rather than Sdo are likely associated with S^0 oxidation. Finally, our data suggest that formation of intracellular $S_2O_3^{2-}$ is a critical step in S^0 metabolism, and that recycling of internally generated SO_3^{2-} occurs, through comproportionating reactions that result in $S_2O_3^{2-}$. Electron microscopy and spectroscopy confirmed intracellular production and storage of S^0 during growth on both S^0 and $S_2O_3^{2-}$ substrates.

Keywords: sulfur metabolism, gene expression, geochemistry, *Acidithiobacillus thiooxidans*, sulfur oxidation, modeling

INTRODUCTION

The stepwise oxidation of reduced sulfur species from sulfide to sulfate can occur via several pathways involving a variety of sulfur oxidation intermediate (SOI) compounds that are dynamically influenced by environmental and geochemical characteristics as well as the microbes involved (Schippers et al., 1996; Schippers and Sand, 1999; Nordstrom, 2015). This range of

sulfur oxidation states contributes to a complex, and only partially constrained biogeochemical cycle in which sulfur compounds can be variably reduced, oxidized and disproportionated via abiotic and/or biotic processes depending on environmental conditions (Johnston and McAmish, 1973; Kelly and Baker, 1990; Pronk et al., 1990; Druschel, 2002; Zopfi et al., 2004; Bernier and Warren, 2007; Boyd and Druschel, 2013). The geochemical challenges to closing the sulfur biogeochemical cycle reflect the existence of multiple semi-stable SOI compounds, which are either not comprehensively constrained to date and/or lack readily available analytical methods for their characterization (Miranda-Trevino et al., 2013). For instance, the challenges in measuring polythionates and other higher oxidation state sulfur compounds have impeded the delineation of their roles in the chain of reactions culminating in sulfate (Johnson and Hallberg, 2003; Nordstrom et al., 2015). The complexities of sulfur chemistry underscore the need for mass balance of all sulfur within systems, in order to quantify how much sulfur may be tied up in a currently unidentified or, as referred to here, “other SOI” pool. However, sulfur mass balance is rarely employed in studies of sulfur cycling.

Further, microbial catalysis, dependent on the specific bacteria, growth stage and sulfur substrates involved, is important for initiating or accelerating rates for some of these sulfur oxidation reactions (Bacelar-Nicolau and Johnson, 1999; Druschel et al., 2004; Bernier and Warren, 2005, 2007; Beller et al., 2006; Warren et al., 2008; Bobadilla Fazzini et al., 2013). Several studies have demonstrated flexibility of the sulfur oxidation metabolism by assessing the solution chemical changes in some intermediate sulfur species, or inferred pathways from what is known about identified sulfur metabolism genes within an organism or community (Bobadilla Fazzini et al., 2013; Jones et al., 2014; Yin et al., 2014; Houghton et al., 2016). Intermediate species of sulfur, especially S^0 , $S_2O_3^{2-}$, and polythionates [$S_nO_6^{2-}$ ($n > 2$)], are important in microbial processing of sulfur, even though their concentrations in solution may be low. Indeed, these intermediate sulfur compounds are thought to be involved in the so-called “cryptic” sulfur cycle, an enigmatic process in which sulfur is recycled amongst lower state sulfur species that is not well-characterized to date (Thamdrup et al., 1994; Jørgensen and Nelson, 2004; Canfield et al., 2010; Houghton et al., 2016).

Further, gaps in understanding of which proteins catalyze specific sulfur pathways also exist (Friedrich et al., 2001; Sauvé et al., 2007; Valdes et al., 2011; Jones et al., 2014). The literature to date indicates that some sulfur metabolic enzymes catalyze a broad suite of sulfur oxidative reactions, e.g., the Sox (sulfur oxidizing) complex, while others seem to catalyze more specific sulfur reactions, e.g., Sdo (sulfur dioxygenase) (Kelly et al., 1997; Friedrich et al., 2001; Rohwerder and Sand, 2003; Hensen et al., 2006; Sauvé et al., 2007; Wang et al., 2019). Some microorganisms capable of sulfur oxidation can possess a suite of these genes, enabling them to carry out many different reactions, while others have a more limited set of sulfur genes, restricting them to select reactions only (Hallberg and Johnson, 2003; Ghosh and Dam, 2009; Zhu et al., 2012; Nuñez et al., 2017). Recent works reviewing *Acidithiobacillus*

spp. sulfur metabolism have identified diverse pathways for this genus dependent on the species, as well as the sulfur substrate(s) (S^0 , $S_2O_3^{2-}$, $S_4O_6^{2-}$) and the different sulfur metabolism genes available to them (Wang et al., 2019; Zhan et al., 2019). These studies have provided updated models for *A. caldus* and *A. ferrooxidans* based on the existing literature of studies using either genomics, proteomics or sulfur chemistry analyses. For both species, S^0 metabolism is proposed as oxidation to SO_3^{2-} via Sdo, followed by oxidation to SO_4^{2-} via the sulfate adenylyltransferase dissimilatory-type (SAT) gene (Wang et al., 2019). While the $S_2O_3^{2-}$ metabolism is proposed to differ between the two species, where in *A. caldus* it is through the S_4I pathway and Sox complex, and in *A. ferrooxidans* via the S_4I pathway and thiosulfate dehydrogenase (TSD) (Wang et al., 2019; Zhan et al., 2019). The S_4I pathway utilizing the *doxD* (thiosulfate:quinone oxidoreductase) and *tetH* (tetrathionate hydrolase) genes (Wang et al., 2019). While further notable genes present in the sulfur metabolism for *Acidithiobacillus* spp. include the *sqr* (sulfide quinone reductase), *sor* (sulfur oxygenase reductase), *rhd* (rhodanese) and the heterodisulfide reductase or Hdr-like complex (*hdrA*, *hdrB*, and *hdrC*) (Ghosh and Dam, 2009; Valdes et al., 2011; Jones et al., 2014; Yin et al., 2014; Wang et al., 2019).

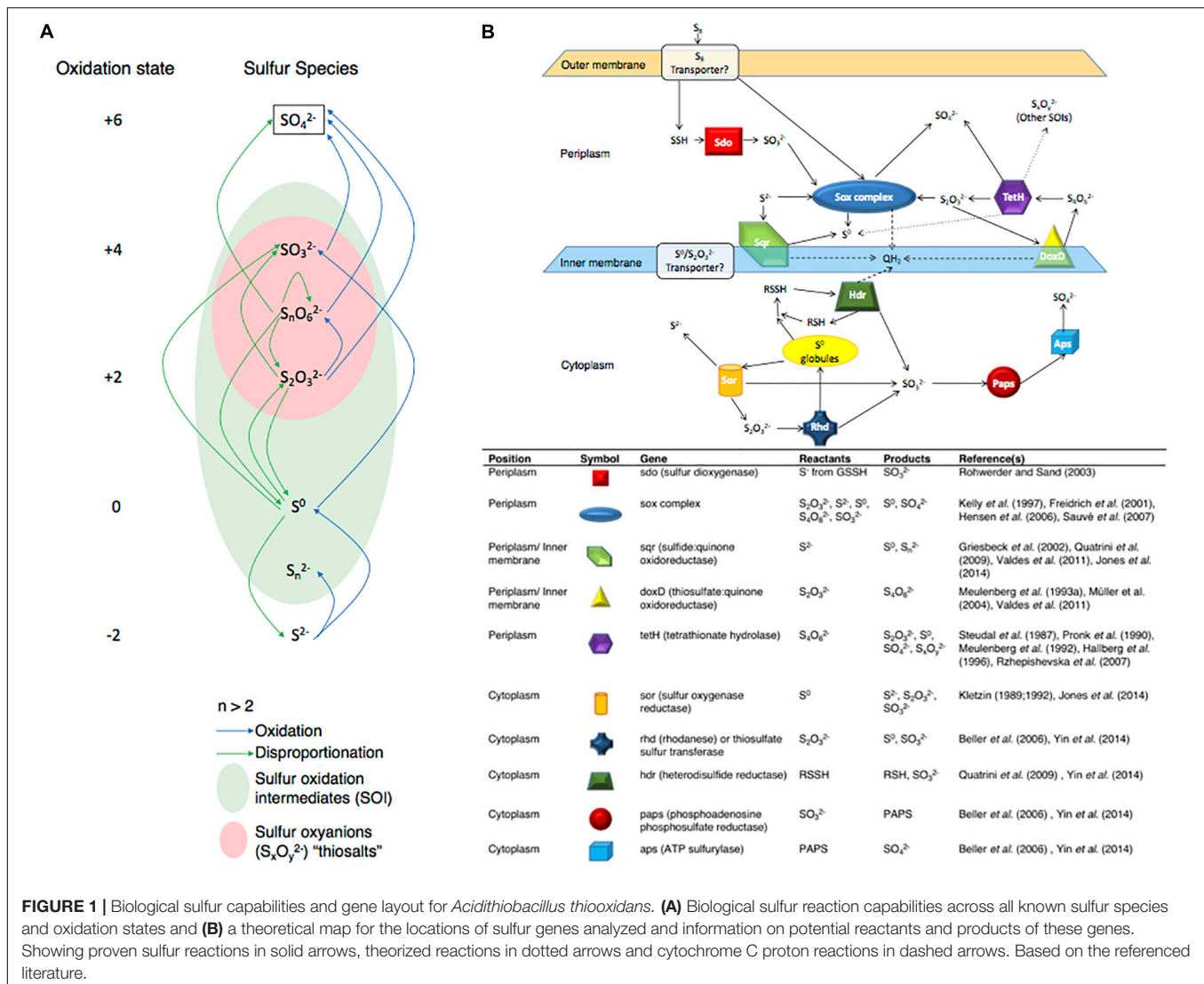
Here, the objectives were to characterize both the levels of gene expression at high resolution (RNA-Seq) for *Acidithiobacillus thiooxidans*, and the changes in sulfur speciation associated with its experimental growth on either S^0 or $S_2O_3^{2-}$ to generate models for *A. thiooxidans* sulfur metabolism. *A. thiooxidans* is a strict autotroph only able to carry out sulfur oxidation/disproportionation reactions (Figure 1A) and a well-studied sulfur oxidizing microorganism (Kelly et al., 1997; Suzuki et al., 1999; Masau et al., 2001; Rohwerder and Sand, 2003). The model organism *A. thiooxidans* ATCC 19377 used here, encodes at least 10 known proteins or protein complexes thought to be involved in sulfur metabolism, which includes the aforementioned S_4I pathway and Sox complex in the periplasm, and the Hdr-like complex in the cytoplasm (Valdes et al., 2011; Bobadilla Fazzini et al., 2013; Yin et al., 2014) (Figure 1B). Our integrated approach provides important novel insights since previous studies have designed models for this species based solely on solution chemistry (Bobadilla Fazzini et al., 2013) or gene expression (Yin et al., 2014).

MATERIALS AND METHODS

Experimental Design, Cell Growth, and Counting

Experimental Design

In order to jointly assess both gene expression and changes in sulfur speciation, the experimental design included collection of samples for cell counts, gene expression, microscopy, S speciation and pH for *A. thiooxidans* grown in both S^0 and $S_2O_3^{2-}$ treatments over 12 days to ensure both exponential and stationary phases were encompassed in the characterization.



Greater details on collection and analyses of samples for each of these variables are provided subsequently.

Culture Conditions

Acidithiobacillus thiooxidans ATCC 19377 cells were grown in liquid elemental sulfur or thiosulfate media (Staley et al., 1989). The media contained two components, the salt medium and the sulfur source. Elemental sulfur salt medium: (NH₄)₂SO₄, 0.2 g; MgSO₄ × 7 H₂O, 0.5 g; CaCl₂ × 2 H₂O, 0.331 g; KH₂PO₄, 3.0 g; FeSO₄ × 7 H₂O, 9.15 mg; distilled water, 1,000 ml. The salt medium was sterilized by passing through a 0.22 μm filter. Elemental sulfur powder was heated in an oven at 100°C for 30 min and the cycle was repeated three times. The salt medium was then added to the culture flasks and the final sulfur concentration was 1% (m/v). Thiosulfate medium: salt medium as above and Na₂S₂O₃ was added at 0.2% (m/v), followed by filter sterilization (0.22 μm filter). For both cultures, the total volume of medium corresponded to a fifth of the total volume of the Erlenmeyer flask. All cultures were initially inoculated at

5% v/v with cultures pre-grown in the corresponding media; the inoculant bacteria were washed with sterile 1% NaCl solution prior to inoculation. All cultures were grown under aerobic conditions at 30°C and flasks were shaken at 120 rpm.

Fluorescence-Activated Cell Sorting (FACS)

Cells were harvested at the desired time points (days 1, 2, 3, 4, 5, 8, 10, and 12) and washed with 1% NaCl. Optical density (O.D.) values were determined to generate cell counts; however errors introduced by S⁰ clumping precluded their use for these experiments. Thus, for the growth curves, 2 μl of the Live/Dead marker mixture of component A and component B at a ratio of 1:1 (L7012 LIVE/DEAD® BacLight, Bacterial Viability Kit, Thermo Fisher Scientific) were added to 1.5 ml of bacterial suspension. The rationale behind the Live/Dead stain is that all cells will be stained green, because SYTO 9 penetrates into live and dead cells and stains their DNA, whereas propidium iodide (red stain) penetrates only into dead or damaged cells with leaky membranes staining their DNA. For the negative control

(dead cells), the cells were first washed with 1% NaCl and then incubated in 70% ethanol for 1 h, followed by washing with 1% NaCl. Propidium iodide (Component B) was added (0.66 μ l for 1 ml of bacterial suspension). For the positive control, 0.66 μ l of SYTO 9 (Component A) was added to 1 ml of bacterial suspension. All samples were incubated in the dark at room temperature for 15 min, followed by counting in a FACS BD Canto II instrument. Experiments were conducted in triplicates.

Genetic Methods and Analyses

DNA Purification

Genomic DNA was purified from cells from 50 ml bacterial culture grown on elemental sulfur by manual cell disruption with a pestle in the presence of small glass beads (<106 μ m diameter; sufficient to form a thick paste). Genomic DNA was purified from combined washes with TE buffer (10 mM Tris, 1 mM EDTA, pH 8) following essentially the instructions of the Qiagen Genomic G20 kit, resulting in 10 μ g of purified total DNA.

Illumina DNA Sequencing

For paired-end Illumina sequencing (MISEQ-PE300, i.e., 300 nucleotides read length), a TruSeq library was constructed with sized DNA fragments (570 to 650 bp). The reads received from the sequencing service (McGill and Génome Québec Innovation Centre; Montreal, QC, Canada) were cleaned from adapters and quality-clipped with the Trimmomatic software (Bolger et al., 2014), resulting in a total of 2,254,174 read pairs. In addition, a Nextera mate-pair library (insert size 7–8 kbp) was sequenced on two flow cells of Illumina HiSeq (rapid mode; 150 nucleotides read length), and cleaned with Trimmomatic (8,224,769 read pairs).

Genome Assembly and Annotation

The genome was assembled with Spades v. 3.6.1 (Bankevich et al., 2012) using a coverage cutoff value of 3.0. The resulting set of contigs was annotated with Prokka v.1.13.3 (Seemann, 2014).

Total RNA Extraction

Cells were harvested on day 3 (exponential phase; pH 2.5) and 5 (stationary phase; pH 1.5) for S^0 media and day 5 (stationary phase; pH 2.5) for $S_2O_3^{2-}$ media, and washed with ice-cold NaCl 1%. They were then lysed and total RNA was extracted using the High Pure RNA Isolation Kit (Roche). Instead of 4 μ l of lysozyme as indicated in the kit, 20 μ l were added to efficiently break the cells. The lysozyme solution was prepared from egg white lysozyme (Bio Basic, Inc.; activity: 20,000 U/mg) at a final concentration of 50 mg/ml. The genomic DNA was removed using the TURBO DNA-free KitTM (Ambion). The concentration of total RNA was determined using a Nanodrop instrument and the quality of the preparation was assessed by agarose gel electrophoresis to monitor 16S and 23S ribosomal RNA. Samples were conserved at -80°C ; experiments were conducted in biological triplicates.

High-Throughput RNA Sequencing and Bioinformatics

Sequencing was done using Illumina Hi-seq technology (100 bases paired-end). Quality controls, DNA library construction

from isolated RNA and sequencing were performed at the Génome Québec Innovation Centre (Montreal, QC, Canada). Bioinformatics analysis was done using software available on the Galaxy server¹ (Giardine et al., 2005; Blankenberg et al., 2010; Goecks et al., 2010). Full-length reads (100 bases) were trimmed so that only portion 11 to 80 of each read was conserved. Quality control of the reads was done using FastQC (Galaxy Tool Version 0.63) before and after trimming to ensure quality of the reads. The quality format was changed to “Sanger & Illumina 1.8 +” using FASTQ Groomer (Galaxy Tool Version 1.0.4). Reads were mapped as paired-end using Tophat (Galaxy Tool Version 0.9). The mean inner distance between mate pairs was set to 150 bases and the standard deviation to 20. The reference genome of *A. thiooxidans* (Valdes et al., 2011) was used as guide to help align the reads and the defaults parameters of Tophat were selected. Finally, differential expression was analyzed using Cufflinks (Galaxy Tool Version 2.2.1.0). The “max intron length” was set to 300,000, the “min isoform fraction” was set to 0.1 and the “pre mRNA fraction” to 0.15. Cufflinks only counted fragments compatible with the reference annotation of the genome and it performed a biased correction using the genome assembly. Default Cufflinks parameters were selected.

Sulfur Chemistry Methods and Analyses

Biogeochemical Experiments

Nine sterile 1 L flasks were prepared for batch experimentation: six containing salt medium with 1% S^0 and three with 0.2% $S_2O_3^{2-}$ culture medium, followed by *A. thiooxidans* inoculation as described above. For each treatment, one flask was sacrificed for sulfur chemical analyses from the S^0 cultures on days 0, 1, 2, 3, 4, and 5 and from the $S_2O_3^{2-}$ cultures on days 0, 2, and 4. For each sampling time, the bulk solution pH was measured (Denver Instrument Model 225, Bohemia, NY, United States) prior to sampling for sulfur analyses. Triplicate samples were then collected for dissolved (<0.45 μ m), total sulfur (ΣS_{aq}) and sulfur speciation (SO_4^{2-} , S^{2-} , $S_2O_3^{2-}$, S^0 , and SO_3^{2-}) analyses as described subsequently.

ΣS_{aq} – Determination by ICP-AES

For total S (ΣS_{aq}), 40 ml of water samples were filtered by Pall Acrodisc[®] 25 mm 0.45 μ m Supor[®] membrane via polypropylene syringes into 50 ml FalconTM tubes, followed immediately by addition of 80 μ L of HNO_3 (Optima grade, Fisher Chemical) to each tube before storing at 4°C until analyses. To enable sulfur mass balance calculations, ΣS_{aq} analyses were performed by inductively coupled argon plasma emission spectrometry (ICPAES) (Varian730 ES, Mulgrave, VIC, Australia) using the operating conditions recommended by the manufacturer. Sulfur calibration standards were prepared from certified reference stock solutions (AccuStandard, New Haven, CT, United States) in 2% v/v HNO_3 . The limit of detection (LOD) for sulfur was 1 mg L^{-1} (calculated as three times the standard deviation of the mean blank). Subtracting the sum of all measured solution sulfur species concentrations, described subsequently (SO_4^{2-} , S^{2-} , $S_2O_3^{2-}$, S^0 , and SO_3^{2-}) from the total sulfur (ΣS_{aq}) concentration,

¹<https://usegalaxy.org/>

allowed us to quantify the concentration of S occurring within an unresolved or “Other” SOI pool.

SO_4^{2-} and S^{2-} – Determination by Spectrophotometry

At each sampling time point, samples were immediately fixed and analyzed using the HACH SulfaVer 4 Method and Methylene Blue Method for SO_4^{2-} and S^{2-} , respectively (Hach Company, Loveland, CO, United States) by spectrophotometry (Pharmacia Biotech Ultrospec 3000 UV/Visible Spectrophotometer).

$\text{S}_2\text{O}_3^{2-}$, S^0 , and SO_3^{2-} – Determination by HPLC

Sampling and analyses for individual SOI species $\text{S}_2\text{O}_3^{2-}$, S^0 , and SO_3^{2-} were concomitant with those for total S, $\Sigma\text{S}_{\text{aq}}$, and redox end members, SO_4^{2-} and S^{2-} . At each sampling time point, samples were taken and immediately preserved using a monobromobimane derivatization procedure for SOI analyses by HPLC (Rethmeier et al., 1997). The Alltima HP C18 (5 μm \times 150 mm \times 4.6 mm) reverse phase column and Shimadzu LC-20AD prominence HPLC instrument were used for all SOI analyses. Solvents used in protocols were: A = Water, B = Methanol, C = Acetonitrile, D = Acetic acid 0.25% v/v pH 3.5 adjusted with NaOH (1N). $\text{S}_2\text{O}_3^{2-}$ and SO_3^{2-} were assessed via fluorescence excitation at 380 nm and emission at 480 nm. Standards and calibrations for $\text{S}_2\text{O}_3^{2-}$ (0–10 mM) and SO_3^{2-} (0–1.7 mM) were made with $\text{Na}_2\text{S}_2\text{O}_3^{2-}$ and $\text{Na}_2\text{SO}_3^{2-}$, respectively. The thiosulfate and sulfite elution protocol was as follows: 0–1 min, 1 ml/min flow; 1–6 min, 1 to 0.85 ml/min flow linear gradient; 0–8 min B 35%, D 65% to B 40%, D 60% linear gradient, oven heated to 35°C. Sample size was 5 μl and elution times were 3 min for SO_3^{2-} and 6.5 min for $\text{S}_2\text{O}_3^{2-}$. S^0 was extracted with chloroform from both filtered (<0.45 μm , i.e., colloidal) and unfiltered samples (i.e., particulate and/or colloidal) and analyzed with reverse-phase HPLC and UV-absorption at 263 nm. Standards and calibrations (0–32 mM) were made from S^0 dissolved in chloroform. S^0 elution protocol: 1 ml/min flow, B 65%, C 35% isocratic; the sample size was 10 μl and the elution time was at 5 min.

Microscopy and Spectroscopy Analyses

Transmission Electron Microscopic (TEM) Analysis

25 ml cultures of bacteria were grown in 1% S^0 or 0.2% $\text{S}_2\text{O}_3^{2-}$ media, respectively. Cells were sedimented and rinsed three times with 0.1M phosphate buffer at pH 7.2 to eliminate the remaining medium. Cells were fixed with 4% paraformaldehyde (Acros Organics, Morris Plains, NJ, United States) and 0.1% glutaraldehyde (Electron Microscopy Sciences, Fort Washington, PA, United States) for 30 min at 4°C, followed by three wash with 0.1M phosphate buffer before osmification using 1% osmium tetroxide for 1 h at room temperature. The pellets were dehydrated using a graded ethyl-alcohol series and then processed for embedding in epon (Marivac, Halifax, NS, Canada). Ultrathin sections of 80–100 nm thickness were cut with a diamond knife, collected on Formvar-carbon (polyvinyl formate) coated 200-mesh nickel grids. Sections were then stained with 2% uranyl acetate and lead citrate and examined with a FEI Tecnai

12 (Eindhoven, Netherlands) transmission electron microscope operating at 80 kV.

Energy-Dispersive X-Ray Spectroscopy and Wavelength-Dispersive Spectroscopy Analysis

Bacterial sections were imaged using a transmission electron microscope (Jeol JEM-2100F, JEOL, Ltd., Tokyo, Japan) equipped for elemental analysis by energy-dispersive X-ray spectroscopy (EDS). In addition, a scanning electron microscope (Jeol JSM-7600F, JEOL, Ltd., Tokyo, Japan) was used for wavelength dispersive X-ray Spectroscopy (WDS) analysis to obtain a better isolation of the peaks of interest for quantitative analysis.

Statistical Analyses

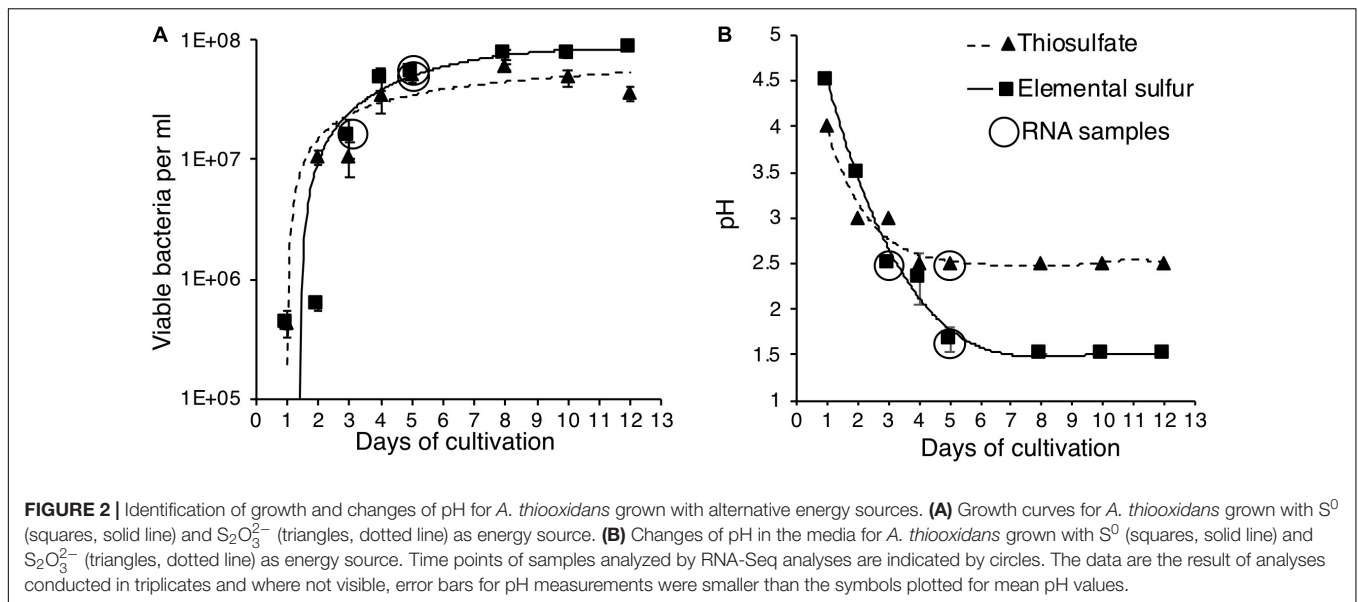
Growth curve and pH results for the two treatments were compared by *t*-test analyses: paired two samples for means via Microsoft Excel 2016, with each treatment having three replicates per data point. RNA-seq analysis is a whole genome approach allowing the detection of low and highly expressed genes using the parameter fragments per kilobase of transcript per million mapped reads (FPKM), and the standard deviations between each treatment's triplicates. Further analyses on FPKM values was carried out to make pairwise comparisons using independent *t*-test on the FPKM between RNA-seq experiments and for the relative levels of gene expression based on Log2 values between samples for the suite of known sulfur genes: (1) across growth curve stage within the S^0 media, (2) between S^0 and $\text{S}_2\text{O}_3^{2-}$ media at the same solution pH and (3) at the same growth curve stage via Microsoft Excel 2016. The chemical relationships between the different S species and $[\text{H}^+]$ (pH) were tested using ANOVA regression statistics via Microsoft Excel 2016 and significance of *p*-value < 0.05 are stated. Intracellular S^0 globules were analyzed after TEM to determine the quantity and size of globules found inside the cells using Image J software.² Manual modeling and stoichiometric balancing methodology is presented in **Supplementary Text**.

RESULTS

Growth, pH, and Sulfur Species Related to Gene Expression

We cultivated *A. thiooxidans* on minimal media with S^0 or $\text{S}_2\text{O}_3^{2-}$ as the source of energy. The results indicate that the organism can extract energy with equal efficiency from both compounds, as evidenced by statistically identical growth patterns for the two media (*p* < 0.05) (**Figure 2A**). However, the amount of acid generated was higher in the S^0 media (final pH of 1.5 compared to 2.5 in the $\text{S}_2\text{O}_3^{2-}$ media) with a corresponding higher slope of pH decrease (0.68 vs. 0.45) as compared to the results on $\text{S}_2\text{O}_3^{2-}$ media over the experimental time period (days 0–5) (**Figure 2B**). These results indicate *A. thiooxidans* catalyzes sulfur substrate-dependent metabolic

²<https://imagej.nih.gov/ij/>



reactions, which may correspondingly be reflected in differential gene induction profiles.

Genomic Analyses Sequencing, Assembly, and Annotation of the *A. thiooxidans* Genome

To correlate the results of the analysis of sulfur species in the medium with expression of the sulfur metabolism genes using RNA-seq we first needed to generate a more robust genome sequence than the available draft version (Valdes et al., 2011). The published draft genome sequence (GenBank: AFOH01000000) has 164 contigs at low coverage and a total genome size of 3,019,868 bp, which may lead to incomplete transcriptome analyses. For this reason, we re-sequenced the genome of *A. thiooxidans* ATCC 19377 and **Table 1** shows the characteristics of the assembly comprising 22 unique contigs and a total of 3,404,101 bp (almost 13% larger than previously published), with the largest contig (2,390,830 bp) spanning 70% of the total sequence. Two contigs have a highly elevated genome coverage, most likely representing circular plasmids. 27 small contigs (size range between 129 and 7,095 bp) carry polymorphic sites and are therefore not counted in the total genome size but included in the GenBank submission. This Whole Genome Shotgun project has been deposited at DDBJ/ENA/GenBank under the accession SZUV00000000. The version described in this paper is version SZUV01000000. A significantly larger fraction of RNA-seq reads (92% for all growth conditions) aligned to our new genome assembly as compared to the previous draft (29–60%) showing that the quality of assembly was greatly improved over the published GenBank record (**Table 2**). Gene annotation identified all known genes encoding enzymes of sulfur metabolism such as *sdo* (sulfur dioxygenase), the Sox (sulfur oxidation) complex (*soxA*, *soxB*, *soxX*, *soxY*, and *soxZ*), *sqr* (sulfide quinone reductase), *doxD* (thiosulfate:quinone oxidoreductase), *tetH* (tetrathionate hydrolase), *sor* (sulfur oxygenase reductase),

rhd (rhodanese), the heterodisulfide reductase (*hdrA*, *hdrB*, and *hdrC*), *paps* (phosphoadenosine phosphosulfate reductase) and *aps* (ATP sulfurylase) (**Figure 3** and **Supplementary Tables S1–S4**) (Kletzin, 1989, 1992; Griesbeck et al., 2002; Rzhapishevska et al., 2007; Valdes et al., 2008; Quatrini et al., 2009; Valdes et al., 2009; Mangold et al., 2011; You et al., 2011). The genome contains three copies of *sdo*, two operons encoding the Sox complex, two copies of *rhd* and three copies of *hdrA*. The plasmids apparently do not code for genes that are of interest in this context, with the potential exception of a gene for a “divalent metal cation transporter” (MntH), which may have been recruited via a plasmid to manage the high metal ion concentrations in its natural environment.

Expression Analysis of the Sulfur Metabolism Genes Using RNA-Seq

For transcriptome analysis, we collected total RNA from cultures of *A. thiooxidans* grown on elemental S^0 and on $S_2O_3^{2-}$ media (three biological replicates) to compare gene expression on two differing oxidation state sulfur substrates. RNA-Seq analysis is a

TABLE 1 | Assembly and annotation of the *Acidithiobacillus thiooxidans* genome ATCC 19377.

Characteristic	Value
Total genome size	3,404,101 bp
Total number of unique contigs (including two potential circular plasmids)	22
Largest contig	2,390,830 bp
Contigs carrying polymorphisms	27
Average% GC	52.6
Number of tRNA genes	64
Number of rRNA genes	4
Total number of coding sequences	3,505
Number of proteins with known function	2,242
Number of hypothetical proteins	1,263

TABLE 2 | Comparison of the percentage of concordant pair alignments between RNA-seq data and the new assembly of the *A. thiooxidans* ATCC 19377 genome and the published draft genome with 164 contigs.

Genome	Elemental sulfur pH 2.5	Elemental sulfur pH 1.5	Thiosulfate pH 2.5
44 contigs genome	92.1	92.3	92.4
164 contigs genome	59.5	29.5	56.8

whole genome approach allowing the detection of low and highly expressed genes using the parameter fragments per kilobase of transcript per million mapped reads (FPKM) [Sequence Read Archive (SRA) accession: PRJNA541131]. To assess the quality of mapping of the RNA-Seq sequences on the genome assembly, we compared the percentage of concordant pair alignments using the same raw RNA-Seq data and the two available genomes [our new assembly and the previously published draft genome (Valdes et al., 2011)]. We observed an increase of more than 30% of the total concordant pair alignments of the RNA-Seq data for the newly assembled genome for each individual sample as compared to the draft (Table 2). These data underline the quality of the new genome assembly that was used for all the following analyses. A direct representation of the FPKM values, i.e., relative expression levels for the three growth conditions (exponential and stationary growth phases on S^0 and stationary growth phase on $S_2O_3^{2-}$) is shown in Figure 3A. FPKM values under 200 are interpreted as low to no expression, as compared to low expression (200–1,000 FPKM), medium expression (1,000–4,000 FPKM), high (4,000–10,000 FPKM), and very highly expressed (more than 10,000 FPKM).

The genes encoding the Sox complex (*soxA*, *B*, *X*, *Y*, *Z*) are generally highly expressed, but interestingly the relative expression of the two *sox* operons changes during growth on elemental sulfur at pH 2.5 (day 3) and pH 1.5 (day 5); *sox-1* strongly decreases and *sox-2* increases to medium levels. In contrast, the *sox-1* operon is very highly expressed during growth on $S_2O_3^{2-}$ and we also observe medium to high expression of the *sox-2* operon showing the importance of the gene products under this condition.

The *sqr* gene is medium to highly expressed in all three conditions at comparable levels suggesting that the gene product sulfide quinone reductase also plays an important role in *A. thiooxidans* S metabolism. Other genes are relatively weakly expressed, and whereas there is some variation of gene expression, it is difficult to assess whether they provide major contributions to sulfur metabolism under these conditions (*aps*, *doxD*, *sor*, and *paps*). We observe low expression of the *rhd* gene and medium to very high expression of *hdrA*, *hdrB*, *hdrC* genes under all conditions. In the case of *sdo*, encoding sulfur dioxygenase required for the entry of elemental sulfur into the cell, the expression of one copy is low under all conditions, whereas two gene copies are below 200 FPKM values (Figure 3A).

Further Pairwise Expression Analysis of the Sulfur Metabolism Genes Using RNA-seq

Expression of most of the *A. thiooxidans* sulfur genes (with exceptions of the *sox-2* operon, *hdr*, all but *hdrA-2*, and

paps genes) was higher on day 3 during exponential growth on S^0 media, as compared to day 5 when cells were in the stationary phase (Figure 3B-i). It thus appears that *A. thiooxidans* exhibits greater metabolic variability in the genes involved, producing higher oxidation state sulfur species (e.g., polythionates) (Figure 1A), during exponential phase, which shifts during stationary phase to a greater processing of polythionates and decreasing pH values (Figure 2). In addition, *hdrA-1* and *hdrA-3* expression strongly increases at pH 1.5 as compared to pH 2.5 during growth on sulfur, suggesting an increased importance of heterodisulfide reductase in the late growth phase. In contrast, the tetrathionate hydrolase encoding gene (*tetH*) is highly expressed only during stationary growth on thiosulfate (day 5), suggesting that this protein plays a specific role in growth on this SOI compound.

Acidithiobacillus thiooxidans gene expression also differed between the two growth media, when an identical pH of 2.5 had been reached. Higher expression levels of the *sox* complex, *tetH*, *hdrA-3*, and *paps* genes were observed for growth on $S_2O_3^{2-}$ (day 5, stationary phase), whilst higher expression levels of all the *sdo* copies, *sqr*, *rhd-2*, *aps* and all the *hdr* genes except *hdrA-3* were observed during growth on S^0 (day 3, exponential phase) (Figure 3B-ii). Gene expression levels also differed for day 5 (stationary phase) for *A. thiooxidans* growth in the two sulfur media (Figure 3B-iii): all sulfur genes with the exceptions of *sdo-1*, *sdo-3*, *sqr* and all *hdr* genes were more highly expressed when grown on $S_2O_3^{2-}$ compared to growth on S^0 .

Genome Wide Analysis of Gene Expression

While the analysis of sulfur genes is vital to the comprehension of autotrophic metabolism, the analysis of the complete transcriptome may lead to the identification of genes that are correlated with this metabolic adaptation. To this effect, we conducted pairwise comparisons of relative gene expression levels (FPKM values) to identify additional up- and down-regulated genes. Analysis of gene expression after growth on elemental sulfur at pH 2.5 compared to pH 1.5 (Supplementary Figure S1a), showed that 20% of the genes (660) are upregulated and 12% (404) are downregulated. The top 50 upregulated genes with the highest degree of differential expression are presented in Supplementary Table S5; several of these genes encode chemotaxis and flagellar components. We also analyzed the top 50 downregulated genes and most encode hypothetical proteins (Supplementary Table S6). Analysis of gene expression after growth on elemental sulfur at pH 2.5 compared to thiosulfate pH 2.5 (Supplementary Figure S1b), shows that 18% (594) are upregulated and 8% (269) are downregulated. The top 50 upregulated genes comprise genes encoding chemotaxis components as well as ATP synthase subunits (Supplementary Table S7). We analyzed the top 50 downregulated genes finding hypothetical proteins as well as transcription factors involved in osmoregulation as well as proteins cytochrome C biogenesis among them (Supplementary Table S8). Finally, analysis of gene expression after growth on thiosulfate at pH 2.5 compared to elemental sulfur at pH 1.5 (Supplementary Figure S1c) shows that 8% of the genes are upregulated (271) and 11% are downregulated (347). The top 50 upregulated

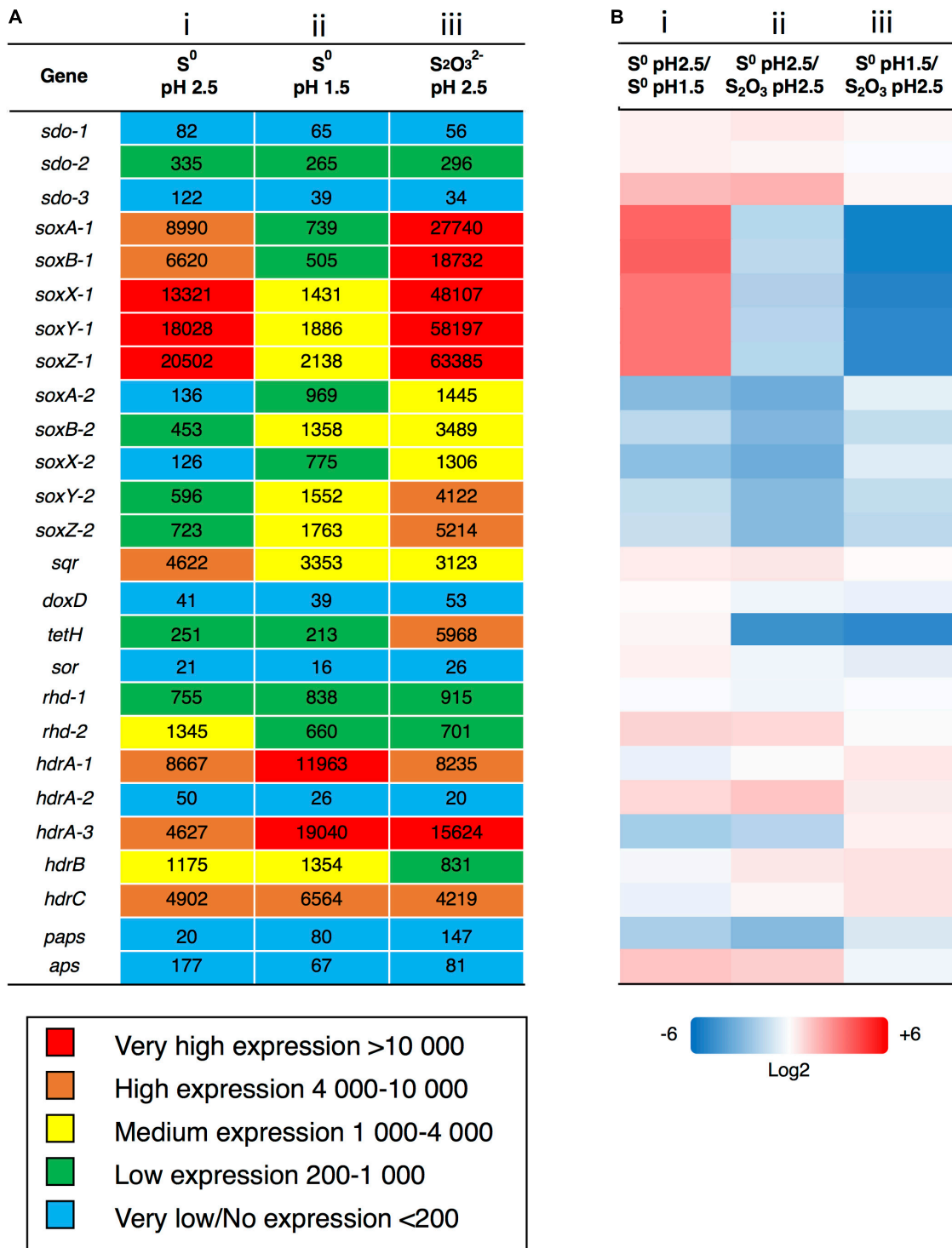


FIGURE 3 | Analysis of gene expression after growth with S⁰ or S₂O₃²⁻ as energy source. **(A)** Gene expression based on FPKM values was analyzed after growth on S⁰ to **(i)** pH-value 2.5 and at **(ii)** 1.5 and on **(iii)** S₂O₃²⁻ to pH-value 2.5. Color scale against indicates relative expression values with blue being the very low, green is low, yellow is intermediate, orange is high and red represents very highly expressed genes. **(B)** Comparative gene expression for FPKM values based on Log₂ ratio. **(i)** Growth on same substrate (S⁰) at different points on pH and growth curve (pH 2.5 = day 3/pH 1.5 = day 5), **(ii)** growth to same pH (2.5) on different substrates and points on growth curve (S⁰ = day 3/S₂O₃²⁻ = day 5), **(iii)** growth until day 5 on different substrates and to different pH values (S⁰ = pH 1.5/S₂O₃²⁻ = pH 2.5). Color scale against each comparison test based on Log₂ values; blue = -6 (i.e., numerator expressed less than denominator), white = 0 (i.e., expression equal), red = +6 (i.e., numerator expressed more than denominator).

genes comprise genes encoding components of cytochrome C biogenesis and of proteins involved in protein folding and outer membrane stability (Supplementary Table S9). Analysis of the top 50 downregulated showed that most encode hypothetical proteins (Supplementary Table S10). Further discussion on these broader metabolic characteristics can be found in Supplementary Text.

Insights Into Sulfur Pathways Catalyzed by *A. thiooxidans* Grown on S⁰ and S₂O₃²⁻

Consistent with the notion that *A. thiooxidans* catalyzes sulfur substrate-dependent metabolic reactions suggested by differential acid production (Figure 2), solution sulfur speciation also differed in the two growth media (Figure 4). *A. thiooxidans* growth on S⁰ resulted in relatively higher concentrations of produced *Other SOI* (i.e., unresolved S species; 25.3 mM versus 6.9 mM on S₂O₃²⁻) and SO₄²⁻ (13.7 mM versus 7.8 mM on S₂O₃²⁻; Figures 4A,B), while growth on S₂O₃²⁻ resulted in near equal generation of *Other SOI*, sulfate and S⁰ (Figures 4A,B). Further, S²⁻ and SO₃²⁻ were largely non-detectable in solution, with the exception of a very low amount of SO₃²⁻ on day 5 in the S⁰ growth experiment (Supplementary Table S11), while both sulfur species were detected at low concentrations (<0.5 mM) throughout growth on S₂O₃²⁻ (Supplementary Table S11).

Sulfur mass balance identified that concentrations of unresolved sulfur species, *Other SOI*, occurred at appreciable levels under both growth conditions (Figures 4A,B). This *Other SOI* pool may variably comprise a number of possible sulfur intermediate oxidation compounds, such as species associated with oxidation pathways, i.e., polythionates, as well as products of disproportionation reactions, i.e., polysulfides. While our results do not identify the specific species sulfur species occurring within this pool, insights provided through analysis of the relationships between changes in (1) [*Other SOI*] and (2) [SO₄²⁻] to [H⁺] (Figures 4C–H), suggest that the unresolved sulfur species differ in their composition between the two growth treatments. The high correlations and statistical significance (*p*-value < 0.05) for Figures 4C–H assist in providing a strong rationale for the basis of stoichiometric reactions occurring in the respective sulfur substrates individual metabolism. The higher slopes observed during growth on S⁰ (Figures 4C,E) alongside the greater overall H⁺ generation (10-fold higher total H⁺ increase) imply greater overall oxidation compared to growth on S₂O₃²⁻ (Figures 4D,F and Supplementary Figure S2a). During growth on S⁰, a decrease in ΔS⁰, and increases in both Δ*Other SOI* and ΔSO₄²⁻ imply that S⁰ is first converted to higher oxidation state SOI, e.g., polythionate species, and ultimately to SO₄²⁻ (Supplementary Figure S2a); consistent with predominantly oxidative (i.e., acid generating) pathways (i.e., Eqs 2–6; Table 3). During growth on S₂O₃²⁻, Δ*Other SOI* and ΔSO₄²⁻ increase from days 0 to 2, while, Δ*Other SOI* subsequently decreases and ΔSO₄²⁻ does not change from days 2 to 4 (Supplementary Figure S2b). These results are consistent with oxidative pathways occurring initially (i.e., Eqs 5, 6, 9,

and 10; Table 3), followed by disproportionating pathways (e.g., Eq. 11; Table 3; as shown further and in Supplementary Text), reflected in an increase in ΔS⁰. Consistent with a potential shift from oxidative (i.e., greater acid generating) to disproportionating reactions dominating, Δ[H⁺] increased between days 0 and 2, and subsequently decreased from days 2 to 4 (Supplementary Figure S2b).

Electron Microscopic and Spectroscopic Analyses of Intracellular S⁰ Storage

Transmission electron microscopy in tandem with energy-dispersive X-ray spectroscopy (EDS) and wavelength dispersive spectroscopy (WDS) revealed sulfur globule formation in the cells (Figures 5A–H). The globules did not differ in size (Supplementary Figure S3), but quantification indicated that a higher number (45.6 per 100 bacteria) were observed for *A. thiooxidans* grown on S⁰, while a lower number of internal S⁰ globules (13.5 per 100 bacteria) occurred for *A. thiooxidans* grown on S₂O₃²⁻ (Figures 5A,B vs. C,D), consistent with sulfur speciation and mass balance results (Supplementary Table S12).

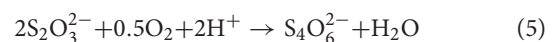
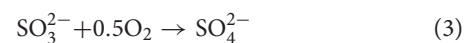
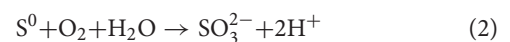
Sulfur Metabolism Models

Stoichiometric Sulfur Metabolism Arrays

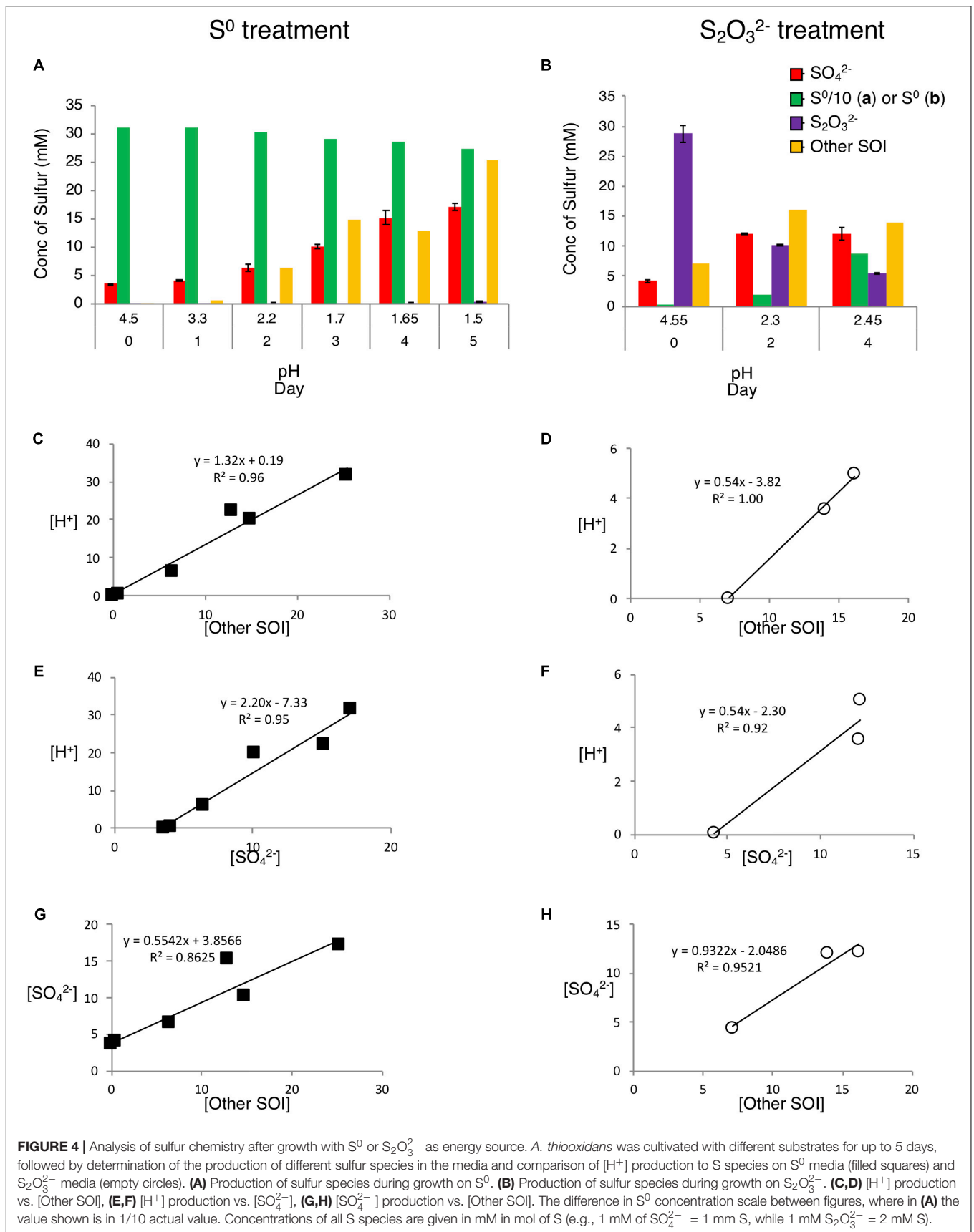
We developed *A. thiooxidans* metabolism models by combining observed solution S speciation and [H⁺] changes with FPKM gene expression levels to elucidate the most likely pathways being catalyzed. The generated *A. thiooxidans* S⁰ metabolism model identifies conversion of S⁰ into 1/3 SO₄²⁻ and 2/3 S^{OtherSOI} [Eq. 1; assumption of initial *Other SOI* generated to be S₄O₆²⁻; the initial metabolism reaction from S₂O₃²⁻ (Eq. 5, Table 3)]. While there are uncertainties as to whether the *Other SOI* pool is solely polythionate species and/or comprises the same polythionates at any given sampling point in either treatment, the highly significant correlations between acid generation and this specific sulfur pool (Figures 4G,H) are consistent with this assumption (Eqs 1, 7, and 8, Figures 6A–C and Table 3, respectively).



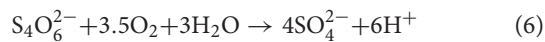
Thus our *A. thiooxidans* S⁰ metabolism model identifies the following suite of reactions occur throughout the time course of the experiment (Figure 6A).



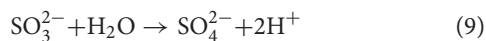
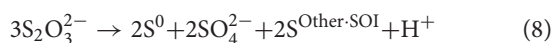
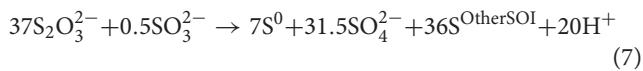
The model stoichiometrically balances the observed changes in elemental sulfur concentration. However, the model predicts a greater acid generation than observed. Specifically, the model predicts production of 6H⁺ for every 6S⁰ converted to 2SO₄²⁻ and 4S (as *Other SOI*); whereas we observe 5H⁺. The same observed lower H⁺ generation relative to expected, also occurs for a model incorporating successive oxidative processing of sulfur by an



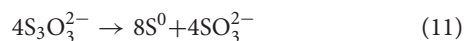
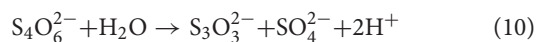
alternative set of pathways that would exclude SO_3^{2-} oxidation to SO_4^{2-} (Eq. 3), and proceed via oxidation of $\text{S}_2\text{O}_3^{2-}$ and $\text{S}_4\text{O}_6^{2-}$ and other polythionates to SO_4^{2-} (Table 3, Eqs. 5 and 6).



A stoichiometrically balanced sulfur and H^+ model of *A. thiooxidans* $\text{S}_2\text{O}_3^{2-}$ metabolism developed for days 0–2 or for the entire time course of days 0–4 (Table 3; Eqs. 7 and 8, respectively) identifies the most likely occurring reactions would include conversion of $\text{S}_2\text{O}_3^{2-}$ to SO_3^{2-} , S^0 and polythionates (*Other SOI*) and ultimately to SO_4^{2-} , with the reverse of Eq. 4 followed by Eqs. 3, 5, 6, and 9 as the dominant reactions (Figure 6B) (Table 3).



However, the $\text{S}_2\text{O}_3^{2-}$ metabolism model of *A. thiooxidans* for days 2–4 indicates disproportionation of $\text{S}_4\text{O}_6^{2-}$ and $\text{S}_3\text{O}_3^{2-}$ to S^0 and SO_3^{2-} are occurring (Table 3, Eqs. 10 and 11).



These disproportionation reactions would recycle sulfur back to $\text{S}_2\text{O}_3^{2-}$, continuing to consume H^+ via regenerated reduced SOI species such as $\text{S}_2\text{O}_3^{2-}$ and S^0 over the time period of days 2–4. $\text{S}_2\text{O}_3^{2-}$ model reaction arrays (Figures 6B,C) can also be stoichiometrically balanced via other pathways involving oxidation of polythionates and thiosulfate to sulfate. However, informed by gene expression, results, S metabolism for days 0–2 and days 2–4 is more consistent with the reactions identified above. The most robust model for days 0–4 based on currently theorized/known sulfur reactions follows the series of reactions shown in Figure 6C, identifying the important formation and accumulation of S^0 . Stepwise reactions for Figure 6 are identified in Supplementary Text.

Models of *A. thiooxidans* S^0 and $\text{S}_2\text{O}_3^{2-}$ Metabolism

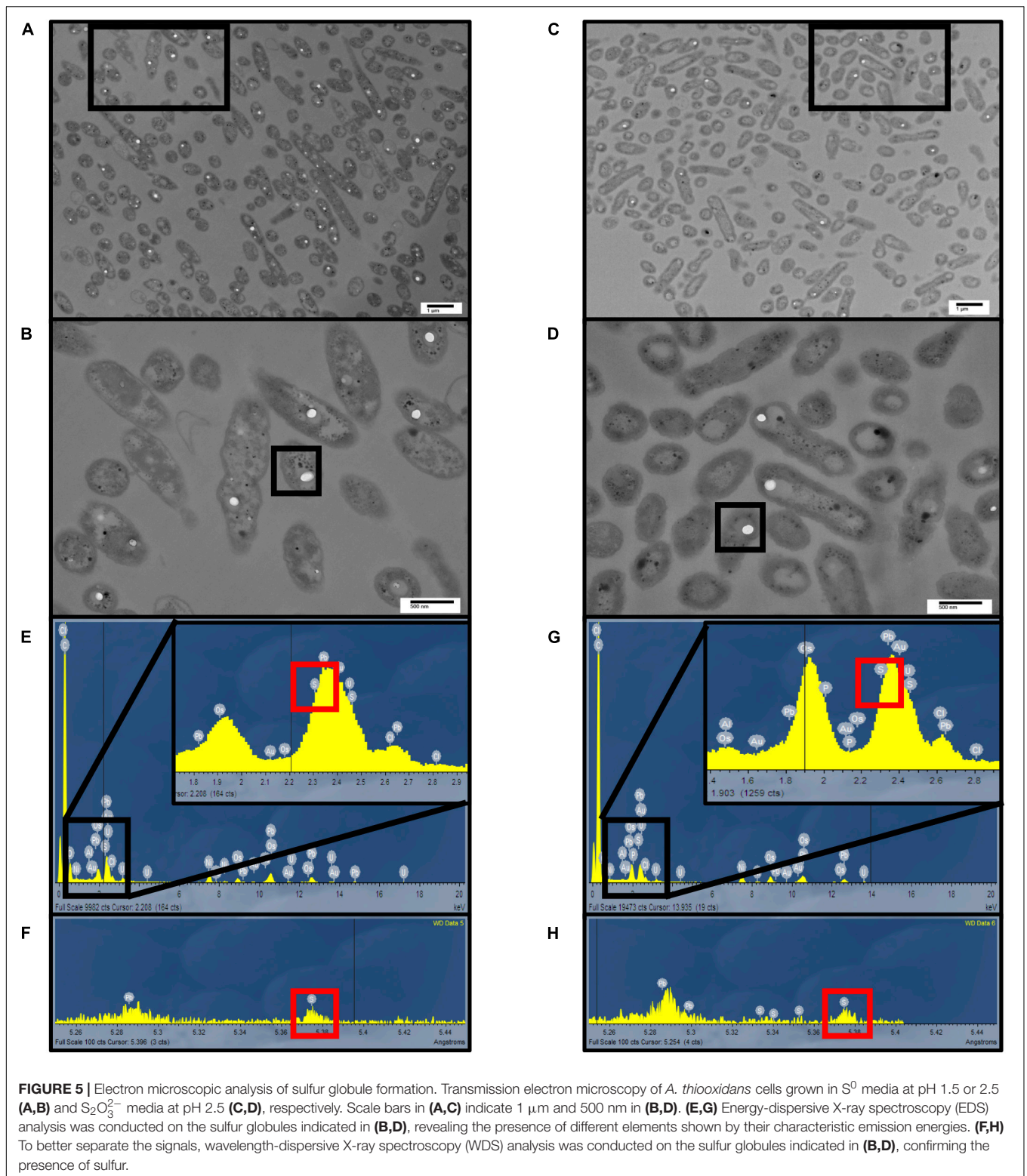
By combining the analysis of gene expression, solution sulfur speciation and electron microscopy, our results provide new insights into *A. thiooxidans* sulfur metabolism revealing the importance of intracellular pathways. Based on these data we propose models for the metabolism of *A. thiooxidans* grown on S^0 (Figures 7A,B) suggesting that the Sox complex plays a major role initiating metabolism after entry of S^0 into the cell via unknown transporters. There is little published information on the transport of S^0 into cells to date, however, it has been postulated by other studies to occur via outer membrane proteins (Sugio et al., 1991; Buonfiglio et al., 1999; Rohwerder and Sand, 2003). Sdo is not highly expressed, but it may also contribute to S^0 metabolism. The intracellular S^0 is metabolized

TABLE 3 | Mass balance S reactions for the two treatments and potential S abiotic and biotic reactions important for stoichiometric balancing.

Formula	Eq. #	References
$6\text{S}^0 \rightarrow 2\text{SO}_4^{2-} + 4\text{S}^{\text{OtherSOI}} + 5\text{H}^+$	1	This paper, S^0 treatment days 0–5
$\text{S}^0 + \text{O}_2 + \text{H}_2\text{O} \rightarrow \text{SO}_3^{2-} + 2\text{H}^+$	2	Based on Suzuki (1999)
$\text{SO}_3^{2-} + 0.5\text{O}_2 \rightarrow \text{SO}_4^{2-}$	3	Based on Suzuki (1999)
$\text{S}^0 + \text{SO}_3^{2-} \leftrightarrow \text{S}_2\text{O}_3^{2-}$	4	Based on Johnston and McAmish (1973) and Suzuki (1999)
$2\text{S}_2\text{O}_3^{2-} + 0.5\text{O}_2 + 2\text{H}^+ \rightarrow \text{S}_4\text{O}_6^{2-} + \text{H}_2\text{O}$	5	Based on Suzuki (1999)
$\text{S}_4\text{O}_6^{2-} + 3.5\text{O}_2 + 3\text{H}_2\text{O} \rightarrow 4\text{SO}_4^{2-} + 6\text{H}^+$	6	Based on Suzuki (1999)
$37\text{S}_2\text{O}_3^{2-} + 0.5\text{SO}_3^{2-} \rightarrow 7\text{S}^0 + 31.5\text{SO}_4^{2-} + 36\text{S}^{\text{OtherSOI}} + 20\text{H}^+$	7	This paper, $\text{S}_2\text{O}_3^{2-}$ treatment days 0–2
$3\text{S}_2\text{O}_3^{2-} \rightarrow 2\text{S}^0 + 2\text{SO}_4^{2-} + 2\text{S}^{\text{OtherSOI}} + \text{H}^+$	8	This paper, $\text{S}_2\text{O}_3^{2-}$ treatment days 0–4
$\text{SO}_3^{2-} + \text{H}_2\text{O} \rightarrow \text{SO}_4^{2-} + 2\text{H}^+$	9	Based on Suzuki (1999)
$\text{S}_4\text{O}_6^{2-} + \text{H}_2\text{O} \rightarrow \text{S}_3\text{O}_3^{2-} + \text{SO}_4^{2-} + 2\text{H}^+$	10	Based on Pronk et al. (1990) and Suzuki (1999)
$4\text{S}_3\text{O}_3^{2-} \rightarrow 8\text{S}^0 + 4\text{SO}_3^{2-}$	11	Based on Steudel et al. (1987) and Pronk et al. (1990)

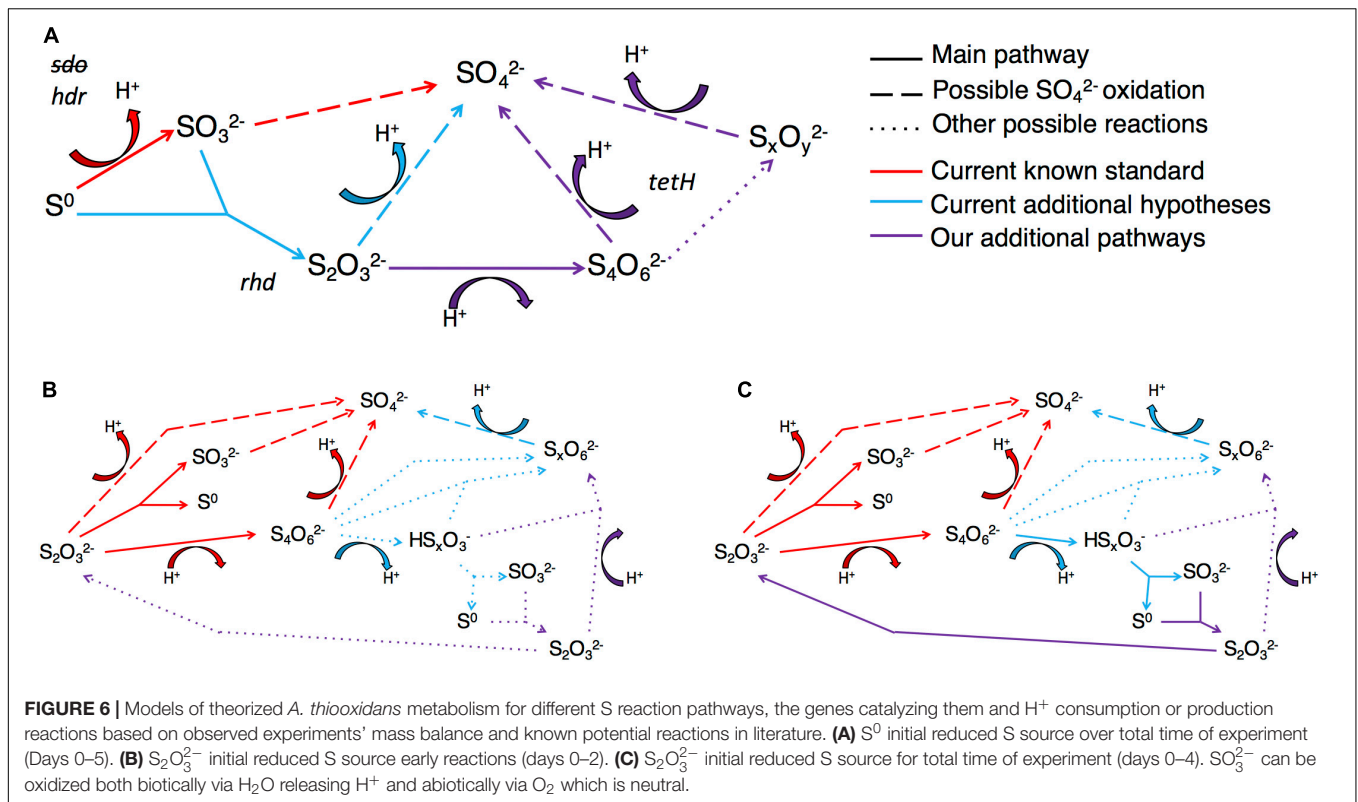
subsequently through both oxidative and comproportionating pathways. Cytoplasmic Hdr catalyzes S^0 oxidation generating intracellular SO_3^{2-} . While it is not certain which gene(s) are involved in intracellular S^0 comproportionation and buildup of sulfur granules, the high expression of genes responsible for SO_3^{2-} production (*hdr*) yet low concentrations in bulk solution, suggest that this pathway generates $\text{S}_2\text{O}_3^{2-}$. We believe that this pathway is active, because we observe medium-level expression of Rhd known to catalyze $\text{S}_2\text{O}_3^{2-}$ disproportionation (Figure 7A), possibly acting in a reverse reaction utilizing the high intracellular S^0 and SO_3^{2-} to produce $\text{S}_2\text{O}_3^{2-}$, which is then oxidized to higher order S species (e.g., tetra- and other polythionates). This possibility is consistent with the observed increased concentration of the *Other SOI* pool and ultimately SO_4^{2-} (Figure 4A). These higher oxidation state S species (i.e., $\text{S}_4\text{O}_6^{2-}$ and/or other higher chain polythionates represented, we believe, in the *Other SOI* fraction based on S speciation, $[\text{H}^+]$ changes and gene expression results presented above) generated through S^0 comproportionation are oxidized through TetH catalysis resulting in SO_4^{2-} . The observed increase in Hdr expression from exponential growth (Figure 7A) to stationary growth (Figure 7B) supports the notion that this pathway would catalyze growth through intracellular recycling of sulfur, and implies the synthesis of sulfur storage granules.

The model for growth of *A. thiooxidans* on $\text{S}_2\text{O}_3^{2-}$ implies that the Sox complex catalyzes $\text{S}_2\text{O}_3^{2-}$ disproportionation to S^0 and SO_4^{2-} (Figure 7C), while TetH catalyzes oxidation and conversion of $\text{S}_4\text{O}_6^{2-}$ to other higher chain polythionates [consistent with detection of *Other SOI*, which would include these unresolved S compounds (Figure 4B)]. These higher oxidation S compounds are then disproportionated via the



Sox complex and/or TetH catalysis, resulting in intracellular S^0 , and the subsequent intracellular generation of SO_3^{2-} indicated to occur by the high level of Hdr expression (Figure 7C). Comproportionation reforming $S_2O_3^{2-}$ from the

high intracellular S^0 and SO_3^{2-} catalyzed by Rhd may also be possible, thereby recycling S within the cell. Alternatively, the low levels of expression of DoxD (Figure 3A), suggest that either TetH may be catalyzing a reverse reaction from $S_2O_3^{2-}$ to $S_4O_6^{2-}$



(or to *Other SOI*), or there may be other proteins responsible for S₂O₃²⁻ oxidation to higher chain polythionates.

DISCUSSION

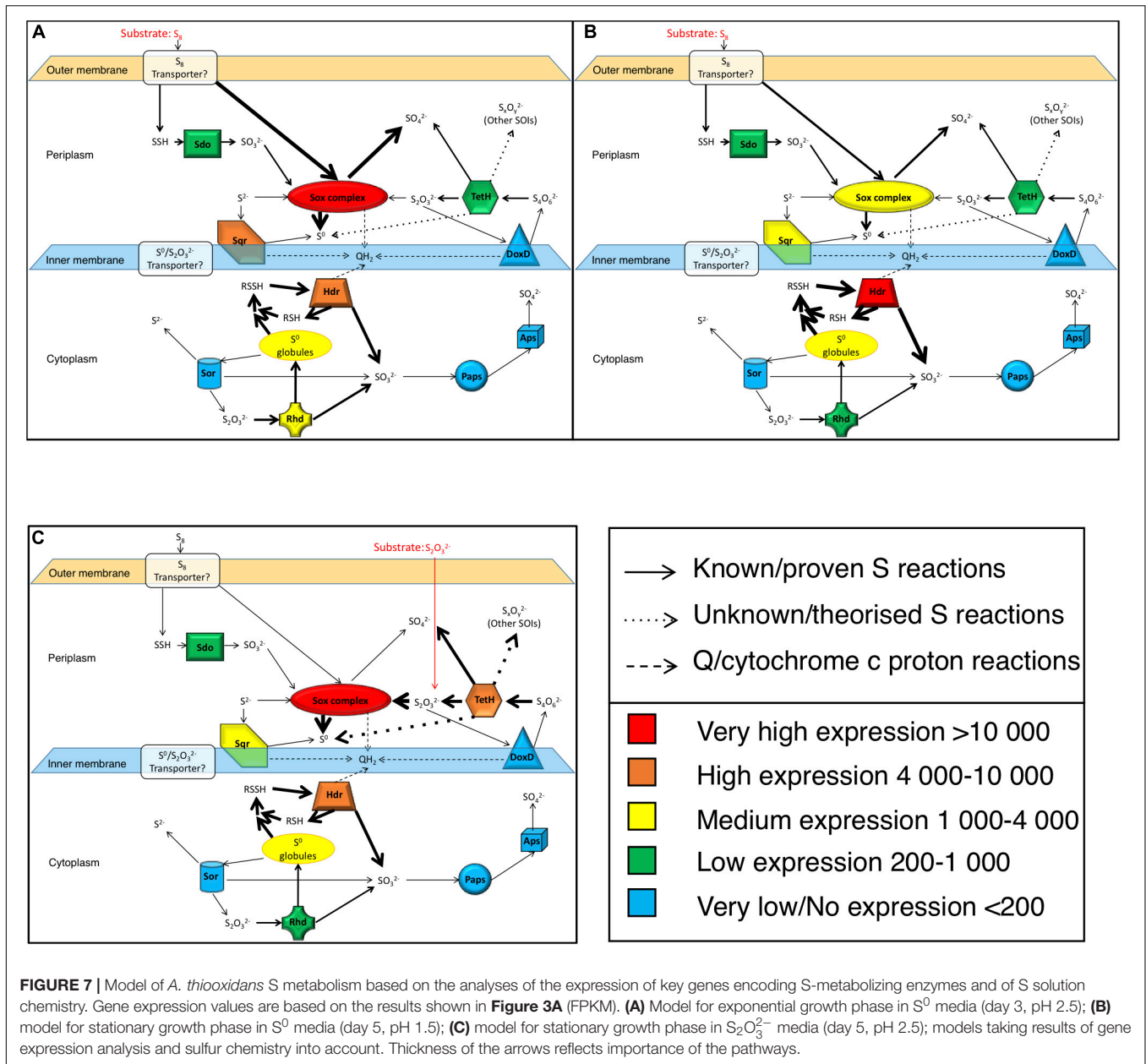
Novel Insights Into S-Metabolism: Importance of S⁰, S₂O₃²⁻, SO₃²⁻ and Intracellular Reactions

Comparisons to Previous Literature

The models of *A. thiooxidans* sulfur metabolism that were developed through integrated analysis of gene expression, sulfur chemistry, sulfur mass balance and electron microscopy reveal new insights into the importance of intracellular reactions involving TetH- and Hdr-catalyzed transformation of S⁰ into SO₃²⁻ species, compared to previous models. Bobadilla Fazzini et al. (2013) analyzed the solution concentrations of two sulfur species, S⁰ and S₂O₃²⁻, for *A. thiooxidans* DSM 17318 at stationary phase when grown in S⁰ and S₄O₆²⁻ media at low pH (1.8 and 2.5, respectively). Their chemically based model identified the same comproportionation reaction involving S⁰ and SO₃²⁻ to form S₂O₃²⁻ (Eq. 4; Table 3) as identified here for *A. thiooxidans* S⁰ growth. However, they speculated that Sdo was the most important protein for SO₃²⁻ production, while our results are more consistent with the Hdr protein catalyzing this reaction. Further, their S⁰ metabolism model does not account for activity of the TetH enzyme, resulting in less S⁰ storage and *Other SOI* (e.g., polythionates) production. Bobadilla Fazzini et al. (2013)

also modeled *A. thiooxidans* growth on S₄O₆²⁻ and suggest, based only on their chemical analyses of S⁰ and S₂O₃²⁻ that S⁰ production from polythionates occurs with no involvement of TetH. In contrast, our combined chemical and gene expression results assessing *A. thiooxidans* growth on S₂O₃²⁻, show that TetH is highly expressed (Figure 7C) and associated with the evident production of intracellular S⁰ determined by microscopy and elemental analyses (Figure 5C). This intracellular S⁰ plays a central role in S₂O₃²⁻ metabolism (Figure 6C). The Bobadilla Fazzini et al. (2013) model did not predict any S⁰ storage for *A. thiooxidans* grown on S⁰, or storage in tandem with TetH activity for *A. thiooxidans* grown on S₄O₆²⁻ and did not include the *hdr*, *rhd*, *paps*, and *aps* genes.

Yin et al. (2014) examined *A. thiooxidans* A01 via gene expression proposing a similar sulfur gene model to our *A. thiooxidans* ATCC 19377 model. Due to our improved draft genome of *A. thiooxidans* ATCC 19377, we were able to find and confirm the previously elusive *sor* gene (Valdes et al., 2011) identifying that the same sulfur genes are present in the two strains (Figure 1B). However, differences in the number of gene copies identified for *hdrA*, *rhd*, *paps*, and *aps* exist between the strains, where for *A. thiooxidans* ATCC 19377, we found three copies of *hdrA*, two copies of *rhd* and one copy each for *paps* and *aps* (Figure 3). In contrast, for *A. thiooxidans* A01 one copy of *hdrA*, five copies of *rhd*, three copies of *paps*, and two copies of *aps* were identified (Yin et al., 2014). That study found that most sulfur metabolic genes were more strongly expressed in *A. thiooxidans* A01 when grown on S⁰ compared to S₂O₃²⁻ during exponential growth phase (Yin et al., 2014),



showing opposing results to our relative expression levels for *A. thiooxidans* ATCC 19377 stationary phase growth on S⁰ compared to S₂O₃²⁻ (Figure 3B-iii). These results suggest that relative gene expression switches from lower to higher in S₂O₃²⁻, and higher to lower in S⁰, as *A. thiooxidans* goes from exponential to stationary growth phase. However, the Yin et al., 2014 study only examined gene expression during exponential growth phase, and their hypothetical models included pathways identified from previous studies depicting models for other *Acidithiobacillus* species (Yin et al., 2014). Thus, their model was not able to identify the importance of intracellular S⁰, and SO₃²⁻ and the *hdr* gene as observed here.

In comparison to other *Acidithiobacillus* species models of S metabolism, our gene model for *A. thiooxidans* shows closest

similarity to *A. caldus*. Differing only that in *A. caldus*, Sdo has been determined to be located in the cytoplasm instead of the periplasm (Wu et al., 2017) and it has the addition of SAT responsible for oxidation of sulfite to sulfate (Wang et al., 2019). While the *A. ferrooxidans* gene model shows greater differences to our *A. thiooxidans* model, most notably in its absence of the Sox complex and *sor* gene, and its inclusion of SAT and TSD (Wang et al., 2019; Zhan et al., 2019). For *A. caldus* and *A. ferrooxidans*, the current S⁰ metabolism is proposed to be oxidation to SO₃²⁻ via Sdo, followed by oxidation to SO₄²⁻ via SAT, with the bacteria acquiring S⁰ from extracellular sources (Mangold et al., 2011; Zhan et al., 2019). This differs to our proposed S⁰ metabolic pathway (Figure 6A), which is elaborated upon further below. The current proposed model of S₂O₃²⁻ oxidation metabolism,

shows that both *A. caldus* and *A. ferrooxidans* utilize the S₄I pathway, however, *A. caldus* also uses the Sox system while *A. ferrooxidans* also uses TSD (Ghosh and Dam, 2009; Wang et al., 2019). Our *A. thiooxidans* S metabolism model follows the same S₂O₃²⁻ oxidation metabolism as *A. caldus* employing both the S₄I pathway and the Sox system.

Integrated Gene Expression and Sulfur Chemistry

A. thiooxidans Metabolism Models

The models generated here provide new insights into the likely pathways involved in *A. thiooxidans* sulfur metabolism, closing some of the gaps in the current understanding. Specifically, our results identify internal cell S⁰ generation, storage and use, as well as the importance and rapid conversion of SO₃²⁻ in these models, both confirming the speculated importance of these two S compounds (Suzuki et al., 1992) and explaining why they have not previously been definitively confirmed by solution chemical characterization alone.

Based on the published studies to date, the first step in microbial S⁰ metabolism is thought to be a relatively linear pathway beginning with oxidation to SO₃²⁻, followed by further oxidation to SO₄²⁻ (Suzuki et al., 1992; Rohwerder and Sand, 2003). However, here the model developed through combined sulfur chemical and gene expression analyses indicates that S₂O₃²⁻ oxidation/disproportionation reactions are occurring as formation of significant amounts of *Other SOI* (i.e., indicating the presence of polythionates) and small amounts of S₂O₃²⁻ are observed (Figure 4A). Consistent with these pathways, expression specifically of *tetH* and *rhd*, genes known to encode enzymes for S₂O₃²⁻ and polythionate oxidation/disproportionation reactions were being expressed (Figures 7A,B) (Meulenberg et al., 1992; Hallberg et al., 1996; Beller et al., 2006; Rzhepishevska et al., 2007).

Further lending support to these alternative pathways, higher relative expression of the *hdr* genes was observed (Figure 3A), which should result in high levels of SO₃²⁻, and thus subsequent high SO₄²⁻ values. However, our results here indicate lower values of SO₄²⁻ than expected, consistent with recycling of this SO₃²⁻ through comproportionating reactions that would generate S₂O₃²⁻ instead. The specific presence of S₂O₃²⁻, despite likely abiotic disproportionation at this low pH < 2, SO₃²⁻ (Supplementary Table S11) and activity of *hdr* genes associated with sulfur back reactions, underscore the formation of S₂O₃²⁻ as a critical step in S⁰ metabolism (Figure 6A) generating the precursor to most reactions involving the increased pool of *Other SOI*, e.g., polythionates (Meulenberg et al., 1993; Müller et al., 2004).

The formation of S₂O₃²⁻ from SO₃²⁻ comproportionation during *A. thiooxidans* S⁰ metabolism is supported by three lines of evidence. First, the intracellular neutral pH of *A. thiooxidans* (Suzuki et al., 1999) makes the neutrophilic reaction combining S⁰ with SO₃²⁻ to form S₂O₃²⁻ favorable (Eq. 4, Table 1). Second, *A. thiooxidans* possesses rhodanese/sulfur transferase (Yin et al., 2014), which may include a rhodanese capable of binding a sulfane group sulfur (e.g., S⁰) to SO₃²⁻ to form S₂O₃²⁻ (Hildebrandt and Grieshaber, 2008; Zhang et al., 2013). Third, the metabolic bonding of S⁰ with SO₃²⁻ is mediated via the Sox

complex, which is highly expressed by *A. thiooxidans* grown on S⁰ (Figures 7A,B). The versatility of the Sox complex would support this pathway (Sauvé et al., 2007; Wang et al., 2019). The gene expression results are consistent with comproportionation, as results here identify that within the Sox complex, SoxYZ (carriers) and SoxAX (binders) are more highly expressed than the oxidizing enzyme (SoxB) in all analyses (Figure 3A).

Metabolic modeling results from growth on S₂O₃²⁻ indicate *A. thiooxidans* S₂O₃²⁻ oxidation closely follows the S₄I pathway proposed in the literature, further suggesting higher oxidation chain polythionate formation (Figures 6B,C and Supplementary Text) (Hallberg et al., 1996; Masau et al., 2001; Müller et al., 2004; Ghosh and Dam, 2009). However, the ability to effectively measure all the possible sulfur species remains an analytical challenge (Houghton et al., 2016), which precludes 100% certainty in our model fitting.

The occurrence of S⁰ within the cells when grown on S⁰, can be attributed to the intake of the sulfur globules from the media via transport enzymes and outer membrane proteins (Rohwerder and Sand, 2003), and/or from SOI cycling through mechanisms such as S₂O₃²⁻ oxidation via the Sox complex when missing SoxCD, a characteristic for sulfur globule formation in bacteria species (Figure 5A) (Steudel et al., 1987; Pronk et al., 1990). However, formation of S⁰ within the cells was also observed when *A. thiooxidans* was grown on S₂O₃²⁻ associated with SOI cycling (Figure 5C), though at lower levels than that observed for *A. thiooxidans* grown on S⁰ (Figures 5A vs. C).

Relevance of Gene Expression Analysis

Relative Expression Levels Between Variable Conditions in S Metabolism

Results assessing relative changes in gene expression identify that the Sox complex, Sqr, Hdr, TetH, and Rhd are important in both S⁰ and S₂O₃²⁻ metabolism by *A. thiooxidans*. While measurements of gene expression does not allow firm conclusions on absolute protein levels or enzyme activities, they do identify specific genes and encoded enzymes likely to be important in a metabolic pathway. Relative levels of expression of these genes however differ between the two sulfur media and between growth stages for S⁰ (Figure 3A). The results illustrate the importance of the Sox complex and of TetH for S₂O₃²⁻ metabolism (Figure 7C). The Sox complex is a very important metabolic enzyme complex during growth on both substrates since it is highly expressed under all conditions (Figure 3A). Gene expression results indicate the *sox-1* operon is active in cells at less acidic pH values and underscore the geochemical flexibility and viability of the Sox complex as expression of the *sox-2* operon is used under more acidic conditions and with higher thiosulfate concentrations (Figure 3A). These observations are similar to those reported by others (Zhu et al., 2012; Jones et al., 2014; Yin et al., 2014; Li et al., 2017; Wang et al., 2019), whereby different gene copies of the same enzyme express at differing levels due to environmental parameters. The differential expression of these gene copies (Figure 3A) indicates that gene expression can provide insights into the geochemical conditions associated with sulfur metabolism.

Insight Into the Importance of Hdr Toward S Metabolism

Our results identify a key role of Hdr in *A. thiooxidans* ATCC 19377 S⁰ metabolism expanding the understanding of important genes and their roles in *A. thiooxidans* sulfur metabolism. Relatively high *hdr* expression levels were observed under all conditions in comparison to the low levels of *sdo* (Figure 3A). The inclusion of solution chemical data and electron microscopy suggest that Hdr is likely the primary S⁰ oxidizing enzyme rather than Sdo, which was previously identified as important for internal generation of SO₃²⁻ and S₂O₃²⁻ during growth on S⁰ (Rohwerder and Sand, 2003; Bobadilla Fazzini et al., 2013; Yin et al., 2014; Koch and Dahl, 2018). Catalysis by Hdr rather than Sdo is energetically more favorable since conversion of S⁰ to SO₃²⁻ is a non-quinone/cytochrome metabolic step for Sdo. Thus it would result in a loss of approximately 50% of the available potential energy considering the ΔG of -500 to 550 kJ per mol S in oxidation of S⁰ to SO₄²⁻ (Kelly, 1999). In contrast, catalysis of the *hdr* gene also found in *A. caldus* (Mangold et al., 2011), *A. ferrooxidans* (Quatrini et al., 2009) and *A. thiooxidans* A01 (Yin et al., 2014) enables *A. thiooxidans* to metabolize and access this energy. The identification of its role in sulfur metabolism here, may assist taxonomic classification and facilitate better understanding of the potential for sulfur metabolism across all *Acidithiobacilli* (Nuñez et al., 2017; Cao et al., 2018; Koch and Dahl, 2018; Wang et al., 2019) and other sulfur oxidizing microbes.

CONCLUSION

Here we are able to provide greater insight into the specific reactions being catalyzed by known sulfur genes and newly highlight the role of Hdr in *A. thiooxidans* sulfur metabolism by integrating gene expression levels with bulk solution S speciation. Our results further confirm the importance specifically of S₂O₃²⁻ and SO₃²⁻ in *A. thiooxidans* sulfur metabolism, which have been widely accepted in the literature to be important, though not definitively shown to date prior to this study (Suzuki et al., 1992; Suzuki, 1999; Bobadilla Fazzini et al., 2013). Further, our results generate new insights into the central role of intracellular S⁰ generation, transformation and pathways in both S₂O₃²⁻ and S⁰ metabolism and that SO₃²⁻ comproportionation to S₂O₃²⁻ is a critical step in S⁰ metabolism. Collectively these results highlight how the integration of molecular biology and chemistry approaches can better inform our understanding of biogeochemical cycling of sulfur by microbes.

REFERENCES

- Bacelar-Nicolau, P., and Johnson, D. B. (1999). Leaching of pyrite by acidophilic heterotrophic iron-oxidizing bacteria in pure and mixed cultures. *Appl. Environ. Microbiol.* 65, 585–590. doi: 10.1128/aem.65.2.585-590.1999
- Bankevich, A., Nurk, S., Antipov, D., Gurevich, A. A., Dvorkin, M., Kulikov, A. S., et al. (2012). SPAdes: a new genome assembly algorithm and its applications to single-cell sequencing. *J. Comp. Biol.* 19, 455–477. doi: 10.1089/cmb.2012.0021

DATA AVAILABILITY STATEMENT

The datasets generated for this study can be found in NCBI GenBank and NCBI Short Read Archive (SRA) repository, submitted and accession number for DNA is SZUV00000000, the accession number for RNA is PRJNA541131.

AUTHOR CONTRIBUTIONS

DC and RF did all experimentation, analyses, and wrote manuscript. AF did TEM, EDS, and WDS work. SA provided analyses on ΣS via ICP-AES. AN, BL, CB, and LW provided funding and laboratory expertise. CB and LW also provided manuscript edits and were main supervisors to this work.

FUNDING

Work in the laboratory of LW was financed by Ontario Genomics Seed Grant, NSERC Discovery Grant and Glencore Sudbury INO and NRCCan MEND Secretariat. Research was funded by Genome Canada GC LSARP 2015 Grant Number OGI-124, and ORF-RE Round 8 Grant Number RE08-007. Work in the laboratories of CB and BL was financed by NSERC Discovery grants.

ACKNOWLEDGMENTS

We would like to thank Serge Sénéchal in the Université de Montréal Department of Microbiology, Infectiology and Immunology for assistance with FACS analysis, Matt Sarrasin (Department of Biochemistry and Molecular Medicine) for assistance with bioinformatics work and Benoit Besette (Department of Biochemistry and Molecular Medicine) for technical assistance. EDS and WDS analysis were conducted in the CM² facility at Polytechnique Montréal (<https://www.polymtl.ca/expertises/centre-de-caracterisation-microscopique-des-materiaux-cm2>). AN is recipient of a Tier 1 Canada Research Chair in Calcified tissues, Biomaterials and Structural Imaging.

SUPPLEMENTARY MATERIAL

The Supplementary Material for this article can be found online at: <https://www.frontiersin.org/articles/10.3389/fmicb.2020.00411/full#supplementary-material>

- Beller, H. R., Chain, P. S., Letain, T. E., Chakicherla, A., Larimer, F. W., Richardson, P. M., et al. (2006). The genome sequence of the obligately chemolithoautotrophic, facultatively anaerobic bacterium *Thiobacillus denitrificans*. *J. Bacteriol.* 188, 1473–1488. doi: 10.1128/JB.188.4.1473-1488.2006
- Bernier, L., and Warren, L. (2005). Microbially driven acidity generation in a tailings lake. *Geobiology* 3, 115–133. doi: 10.1111/j.1472-4669.2005.00047.x

- Bernier, L., and Warren, L. A. (2007). Geochemical diversity in S processes mediated by culture-adapted and environmental-enrichments of *Acidithiobacillus* spp. *Geochim. Cosmochim. Acta* 71, 5684–5697. doi: 10.1016/j.gca.2007.08.010
- Blankenberg, D., Von Kuster, G., Coraor, N., Ananda, G., Lazarus, R., Mangan, M., et al. (2010). Galaxy: a web-based genome analysis tool for experimentalists. *Curr. Protoc. Mol. Biol.* Chapter 1, Unit19.10.1–21. doi: 10.1002/0471142727.mb1910s89
- Bobadilla Fazzini, R. A., Cortés, M. P., Padilla, L., Maturana, D., Budinich, M., Maass, A., et al. (2013). Stoichiometric modeling of oxidation of reduced inorganic sulfur compounds (Riscs) in *Acidithiobacillus thiooxidans*. *Biotechnol. Bioeng.* 110, 2242–2251. doi: 10.1002/bit.24875
- Bolger, A. M., Lohse, M., and Usadel, B. (2014). Trimmomatic: a flexible trimmer for Illumina sequence data. *Bioinformatics* 30, 2114–2120. doi: 10.1093/bioinformatics/btu170
- Boyd, E. S., and Druschel, G. K. (2013). Involvement of intermediate sulfur species in biological reduction of elemental sulfur under acidic, hydrothermal conditions. *Appl. Environ. Microbiol.* 79, 2061–2068. doi: 10.1128/AEM.03160-12
- Buonfiglio, V., Polidoro, M., Soyer, F., Valenti, P., and Shively, J. (1999). A novel gene encoding a sulfur-regulated outer membrane protein in *Thiobacillus ferrooxidans*. *J. Biotechnol.* 72, 85–93. doi: 10.1016/s0168-1656(99)00097-8
- Canfield, D. E., Stewart, F. J., Thamdrup, B., De Brabandere, L., Dalsgaard, T., Delong, E. F., et al. (2010). A cryptic sulfur cycle in oxygen-minimum-zone waters off the Chilean coast. *Science* 330, 1375–1378. doi: 10.1126/science.1196889
- Cao, X., Koch, T., Steffens, L., Finkensieper, J., Zigann, R., Cronan, J. E., et al. (2018). Lipoate-binding proteins and specific lipoate-protein ligases in microbial sulfur oxidation reveal an atypical role for an old cofactor. *Elife* 7:37439. doi: 10.7554/eLife.37439
- Druschel, G. K. (2002). *Sulfur Biochemistry: Kinetics of Intermediate Sulfur Species Reactions in the Environment*. Ph.D. dissertation, University of Wisconsin – Madison, Madison, WI.
- Druschel, G. K., Baker, B. J., Gihring, T. M., and Banfield, J. F. (2004). Acid mine drainage biogeochemistry at Iron Mountain, California. *Geochem. Trans.* 5:13. doi: 10.1186/1467-4866-5-13
- Friedrich, C. G., Rother, D., Bardischewsky, F., Quentmeier, A., and Fischer, J. (2001). Oxidation of reduced inorganic sulfur compounds by bacteria: emergence of a common mechanism? *Appl. Environ. Microbiol.* 67, 2873–2882. doi: 10.1128/AEM.67.7.2873-2882.2001
- Ghosh, W., and Dam, B. (2009). Biochemistry and molecular biology of lithotrophic sulfur oxidation by taxonomically and ecologically diverse bacteria and archaea. *FEMS Microbiol. Rev.* 33, 999–1043. doi: 10.1111/j.1574-6976.2009.00187.x
- Giardine, B., Riemer, C., Hardison, R. C., Burhans, R., Elntski, L., Shah, P., et al. (2005). Galaxy: a platform for interactive large-scale genome analysis. *Genome Res.* 15, 1451–1455. doi: 10.1101/gr.4086505
- Goecks, J., Nekrutenko, A., Taylor, J., and Galaxy, T. (2010). Galaxy: a comprehensive approach for supporting accessible, reproducible, and transparent computational research in the life sciences. *Genome Biol.* 11:R86. doi: 10.1186/gb-2010-11-8-r86
- Griesbeck, C., Schütz, M., Schödl, T., Bathe, S., Nausch, L., Mederer, N., et al. (2002). Mechanism of sulfide-quinone reductase investigated using site-directed mutagenesis and sulfur analysis. *Biochemistry* 41, 11552–11565. doi: 10.1021/bi026032b
- Hallberg, K. B., Dopson, M., and Lindstrom, E. B. (1996). Reduced sulfur compound oxidation by *Thiobacillus caldus*. *J. Bacteriol.* 178, 6–11. doi: 10.1128/jb.178.1.6-11.1996
- Hallberg, K. B., and Johnson, D. B. (2003). Novel acidophiles isolated from moderately acidic mine drainage waters. *Hydrometallurgy* 71, 139–148. doi: 10.1016/s0304-386x(03)00150-6
- Hensen, D., Sperling, D., Trüper, H. G., Brune, D. C., and Dahl, C. (2006). Thiosulphate oxidation in the phototrophic sulphur bacterium *Allochroamatium vinosum*. *Mol. Microbiol.* 62, 794–810. doi: 10.1111/j.1365-2958.2006.05408.x
- Hildebrandt, T. M., and Grieshaber, M. K. (2008). Three enzymatic activities catalyze the oxidation of sulfide to thiosulfate in mammalian and invertebrate mitochondria. *FEBS J.* 275, 3352–3361. doi: 10.1111/j.1742-4658.2008.06482.x
- Houghton, J. L., Foustoukos, D., Flynn, T. M., Vetriani, C., Bradley, A. S., and Fike, D. A. (2016). Thiosulfate oxidation by *Thiomicrospira thermophila*: metabolic flexibility in response to ambient geochemistry. *Environ. Microbiol.* 18, 3057–3072. doi: 10.1111/1462-2920.13232
- Johnson, D. B., and Hallberg, K. B. (2003). The microbiology of acidic mine waters. *Res. Microbiol.* 154, 466–473. doi: 10.1016/s0923-2508(03)00114-1
- Johnston, F., and McAmish, L. (1973). A study of the rates of sulfur production in acid thiosulfate solutions using S-35. *J. Colloid Interface Sci.* 42, 112–119. doi: 10.1016/0021-9797(73)90013-1
- Jones, D. S., Schaperdoth, I., and Macalady, J. L. (2014). Metagenomic evidence for sulfide oxidation in extremely acidic cave biofilms. *Geomicrobiol. J.* 31, 194–204. doi: 10.1080/01490451.2013.834008
- Jørgensen, B. B., and Nelson, D. C. (2004). Sulfide oxidation in marine sediments: geochemistry meets microbiology. *Geol. Soc. Am.* 379, 63–81.
- Kelly, D. P. (1999). Thermodynamic aspects of energy conservation by chemolithotrophic sulfur bacteria in relation to the sulfur oxidation pathways. *Arch. Microbiol.* 171, 219–229. doi: 10.1007/s002030050703
- Kelly, D. P., and Baker, S. C. (1990). The organosulphur cycle: aerobic and anaerobic processes leading to turnover of C1 -sulphur compounds. *FEMS Microbiol. Lett.* 87, 241–246. doi: 10.1111/j.1574-6968.1990.tb04919.x
- Kelly, D. P., Shergill, J. K., Lu, W., and Wood, A. P. (1997). Oxidative metabolism of inorganic sulfur compounds by bacteria. *Antonie Van Leeuwenhoek.* 71, 95–107.
- Kletzin, A. (1989). Coupled enzymatic production of sulfite, thiosulfate, and hydrogen sulfide from sulfur: purification and properties of a sulfur oxygenase reductase from the facultatively anaerobic archaeobacterium *Desulfurolobus ambivalens*. *J. Bacteriol.* 171, 1638–1643. doi: 10.1128/jb.171.3.1638-1643.1989
- Kletzin, A. (1992). Molecular characterization of the sor gene, which encodes the sulfur oxygenase/reductase of the thermoacidophilic Archaeum *Desulfurolobus ambivalens*. *J. Bacteriol.* 174, 5854–5859. doi: 10.1128/jb.174.18.5854-5859.1992
- Koch, T., and Dahl, C. (2018). A novel bacterial sulfur oxidation pathway provides a new link between the cycles of organic and inorganic sulfur compounds. *ISME J.* 12, 2479–2491. doi: 10.1038/s41396-018-0209-7
- Li, L. F., Fu, L. J., Lin, J. Q., Pang, X., Liu, X. M., Wang, R., et al. (2017). The sigma(54)-dependent two-component system regulating sulfur oxidation (Sox) system in *Acidithiobacillus caldus* and some chemolithotrophic bacteria. *Appl. Microbiol. Biotechnol.* 101, 2079–2092. doi: 10.1007/s00253-016-8026-2
- Mangold, S., Valdes, J., Holmes, D. S., and Dopson, M. (2011). Sulfur metabolism in the extreme acidophile *Acidithiobacillus caldus*. *Front. Microbiol.* 2:17. doi: 10.3389/fmicb.2011.00017
- Masau, R. J. Y., Oh, J. K., and Suzuki, I. (2001). Mechanism of oxidation of inorganic sulfur compounds by thiosulfate-grown *Thiobacillus thiooxidans*. *Can. J. Microbiol.* 47, 348–358. doi: 10.1139/w01-015
- Meulenberg, R., Pronk, J. T., Hazeu, W., Bos, P., and Kuenen, J. G. (1992). Oxidation of reduced sulphur compounds by intact cells of *Thiobacillus acidophilus*. *Arch. Microbiol.* 157, 161–168. doi: 10.1007/bf00245285
- Meulenberg, R., Pronk, J. T., Hazeu, W., van Dijken, J. P., Frank, J., Bos, P., et al. (1993). Purification and partial characterization of thiosulphate dehydrogenase from *Thiobacillus acidophilus*. *J. Gen. Microbiol.* 139, 2033–2039. doi: 10.1099/00221287-139-9-2033
- Miranda-Trevino, J. C., Pappoe, M., Hawboldt, K., and Bottaro, C. (2013). The importance of thiosalts speciation: review of analytical methods, kinetics, and treatment. *Crit. Rev. Environ. Sci. Technol.* 43, 2013–2070. doi: 10.1080/10643389.2012.672047
- Müller, F. H., Bandejas, T. M., Urich, T., Teixeira, M., Gomes, C. M., and Kletzin, A. (2004). Coupling of the pathway of sulphur oxidation to dioxygen reduction: characterization of a novel membrane-bound thiosulphate: quinone oxidoreductase. *Mol. Microbiol.* 53, 1147–1160. doi: 10.1111/j.1365-2958.2004.04193.x
- Nordstrom, D. K. (2015). Baseline and premining geochemical characterization of mined sites. *Appl. Geochem.* 57, 17–34. doi: 10.1016/j.apgeochem.2014.12.010
- Nordstrom, D. K., Blowes, D. W., and Ptacek, C. J. (2015). Hydrogeochemistry and microbiology of mine drainage: an update. *Appl. Geochem.* 57, 3–16. doi: 10.1016/j.apgeochem.2015.02.008
- Núñez, H., Moya-Beltrán, A., Covarrubias, P. C., Issotta, F., Cardenas, J. P., Gonzalez, M., et al. (2017). Molecular systematics of the genus *Acidithiobacillus*:

- insights into the phylogenetic structure and diversification of the taxon. *Front. Microbiol.* 8:30. doi: 10.3389/fmicb.2017.00030
- Pronk, J., Meulenber, R., Hazeu, W., Bos, P., and Kuenen, J. (1990). Oxidation of reduced inorganic sulphur compounds by acidophilic *Thiobacilli*. *FEMS Microbiol. Lett.* 75, 293–306. doi: 10.1111/j.1574-6968.1990.tb04103.x
- Quatrini, R., Appia-Ayme, C., Denis, Y., Jedlicki, E., Holmes, D. S., and Bonnefoy, V. (2009). Extending the models for iron and sulfur oxidation in the extreme acidophile *Acidithiobacillus ferrooxidans*. *BMC Genomics* 10:394. doi: 10.1186/1471-2164-10-394
- Rethmeier, J., Rabenstein, A., Langer, M., and Fischer, U. (1997). Detection of traces of oxidized and reduced sulfur compounds in small samples by combination of different high-performance liquid chromatography methods. *J. Chromatogr. A* 760, 295–302. doi: 10.1016/s0021-9673(96)00809-6
- Rohwerder, T., and Sand, W. (2003). The sulfane sulfur of persulfides is the actual substrate of the sulfur-oxidizing enzymes from *Acidithiobacillus* and *Acidiphilium* spp. *Microbiology* 149, 1699–1710. doi: 10.1099/mic.0.26212-0
- Rzhepishvska, O. I., Valdes, J., Marcinkeviciene, L., Gallardo, C. A., Meskys, R., Bonnefoy, V., et al. (2007). Regulation of a novel *Acidithiobacillus caldus* gene cluster involved in metabolism of reduced inorganic sulfur compounds. *Appl. Environ. Microbiol.* 73, 7367–7372. doi: 10.1128/AEM.01497-7
- Sauvé, V., Bruno, S., Berks, B. C., and Hemmings, A. M. (2007). The SoxYZ complex carries sulfur cycle intermediates on a peptide swinging arm. *J. Biol. Chem.* 282, 23194–23204. doi: 10.1074/jbc.m701602200
- Schippers, A., Jozsa, P., and Sand, W. (1996). Sulfur chemistry in bacterial leaching of pyrite. *Appl. Environ. Microbiol.* 62, 3424–3431. doi: 10.1128/aem.62.9.3424-3431.1996
- Schippers, A., and Sand, W. (1999). Bacterial leaching of metal sulfides proceeds by two indirect mechanisms via thiosulfate or via polysulfides and sulfur. *Appl. Environ. Microbiol.* 65, 319–321. doi: 10.1128/aem.65.1.319-321.1999
- Seemann, T. (2014). Prokka: rapid prokaryotic genome annotation. *Bioinformatics* 30, 2068–2069. doi: 10.1093/bioinformatics/btu153
- Staley, J. T., Bryant, M. P., Pfennig, N., and Holt, J. G. (1989). “Acidithiobacillus,” in *Bergey’s Manual of Systematic Bacteriology*, 1st Edn, Vol. 3, ed. W. A. Wilkins (Baltimore, MD: Springer), 1842–1858.
- Studel, R., Holdt, G., Göbel, T., and Hazeu, W. (1987). Chromatographic separation of higher polythionates SnO 62⊖(n= 3..22) and their detection in cultures of *Thiobacillus ferrooxidans*; Molecular Composition of Bacterial Sulfur Secretions. *Angewandte Chemie Int. Ed. Engl.* 26, 151–153. doi: 10.1002/anie.198701511
- Sugio, T., Suzuki, H., Oto, A., Inagaki, K., Tanaka, H., and Tano, T. (1991). Purification and some properties of a hydrogen sulfide-binding protein that is involved in sulfur oxidation of *Thiobacillus ferrooxidans*. *Agric. Biol. Chem.* 55, 2091–2097. doi: 10.1271/abb1961.55.2091
- Suzuki, I. (1999). Oxidation of inorganic sulfur compounds: chemical and enzymatic reactions. *Can. J. Microbiol.* 45, 97–105. doi: 10.1139/w98-223
- Suzuki, I., Chan, C. W., and Takeuchi, T. L. (1992). Oxidation of elemental sulfur to sulfite by *Thiobacillus thiooxidans* cells. *Appl. Environ. Microbiol.* 58, 3767–3769. doi: 10.1128/aem.58.11.3767-3769.1992
- Suzuki, I., Lee, D., Mackay, B., Harahuc, L., and Oh, J. K. (1999). Effect of various ions, pH, and osmotic pressure on oxidation of elemental sulfur by *Thiobacillus thiooxidans*. *Appl. Environ. Microbiol.* 65, 5163–5168. doi: 10.1128/aem.65.11.5163-5168.1999
- Thamdrup, B., Fossing, H., and Jørgensen, B. B. (1994). Manganese, iron and sulfur cycling in a coastal marine sediment, Aarhus Bay, Denmark. *Geochim. Cosmochim. Acta* 58, 5115–5129. doi: 10.1016/0016-7037(94)90298-4
- Valdes, J., Ossandon, F., Quatrini, R., Dopson, M., and Holmes, D. S. (2011). Draft genome sequence of the extremely acidophilic biomining bacterium *Acidithiobacillus thiooxidans* ATCC 19377 provides insights into the evolution of the *Acidithiobacillus* genus. *J. Bacteriol.* 193, 7003–7004. doi: 10.1128/JB.06281-11
- Valdes, J., Pedroso, I., Quatrini, R., Dodson, R. J., Tettelin, H., and Blake, R. II (2008). *Acidithiobacillus ferrooxidans* metabolism: from genome sequence to industrial applications. *BMC Genomics* 9:597. doi: 10.1186/1471-2164-9-597
- Valdes, J., Quatrini, R., Hallberg, K., Dopson, M., Valenzuela, P. D., and Holmes, D. S. (2009). Draft genome sequence of the extremely acidophilic bacterium *Acidithiobacillus caldus* ATCC 51756 reveals metabolic versatility in the genus *Acidithiobacillus*. *J. Bacteriol.* 191, 5877–5878. doi: 10.1128/JB.00843-09
- Wang, R., Lin, J. Q., Liu, X. M., Pang, X., Zhang, C. J., Gao, X. Y., et al. (2019). Sulfur oxidation in the acidophilic autotrophic *Acidithiobacillus* spp. *Front. Microbiol.* 9:3290. doi: 10.3389/fmicb.2018.03290
- Warren, L., Norlund, K. I., and Bernier, L. (2008). Microbial thiosulfate reaction arrays: the interactive roles of Fe (III), O₂ and microbial strain on disproportionation and oxidation pathways. *Geobiology* 6, 461–470. doi: 10.1111/j.1472-4669.2008.00173.x
- Wu, W., Pang, X., Lin, J., Liu, X., Wang, R., Lin, J., et al. (2017). Discovery of a new subgroup of sulfur dioxygenases and characterization of sulfur dioxygenases in the sulfur metabolic network of *Acidithiobacillus caldus*. *PLoS One* 12:e0183668. doi: 10.1371/journal.pone.0183668
- Yin, H., Zhang, X., Li, X., He, Z., Liang, Y., Guo, X., et al. (2014). Whole-genome sequencing reveals novel insights into sulfur oxidation in the extremophile *Acidithiobacillus thiooxidans*. *BMC Microbiol.* 14:179. doi: 10.1186/1471-2180-14-179
- You, X. Y., Guo, X., Zheng, H. J., Zhang, M. J., Liu, L. J., Zhu, Y. Q., et al. (2011). Unraveling the *Acidithiobacillus caldus* complete genome and its central metabolisms for carbon assimilation. *J. Genet. Genomics* 38, 243–252. doi: 10.1016/j.jgg.2011.04.006
- Zhan, Y., Yang, M., Zhang, S., Zhao, D., Duan, J., Wang, W., et al. (2019). Iron and sulfur oxidation pathways of *Acidithiobacillus ferrooxidans*. *World J. Microbiol. Biotechnol.* 35:60. doi: 10.1007/s11274-019-2632-y
- Zhang, L., Liu, X., Liu, J., and Zhang, Z. (2013). Characteristics and function of sulfur dioxygenase in echiuran worm *Urechis unicinctus*. *PLoS One* 8:e81885. doi: 10.1371/journal.pone.0081885
- Zhu, J., Jiao, W., Li, Q., Liu, X., Qin, W., Qiu, G., et al. (2012). Investigation of energy gene expressions and community structures of free and attached acidophilic bacteria in chalcopyrite bioleaching. *J. Ind. Microbiol. Biotechnol.* 39, 1833–1840. doi: 10.1007/s10295-012-1190-1
- Zopf, J., Ferdelman, T. G., and Fossing, H. (2004). “Distribution and fate of sulfur intermediates—sulfite, tetrathionate, thiosulfate, and elemental sulfur—in marine sediments,” in *Sulfur Biogeochemistry — Past and Present Special Paper 379*, eds J. P. Amend, K. Edwards, and T. W. Lyons (Boulder, CO: Geological Society of America), 97–116.

Conflict of Interest: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

Copyright © 2020 Camacho, Frazao, Fouillen, Nanci, Lang, Apte, Baron and Warren. This is an open-access article distributed under the terms of the Creative Commons Attribution License (CC BY). The use, distribution or reproduction in other forums is permitted, provided the original author(s) and the copyright owner(s) are credited and that the original publication in this journal is cited, in accordance with accepted academic practice. No use, distribution or reproduction is permitted which does not comply with these terms.