



Corrigendum: Living at the Extremes: Extremophiles and the Limits of Life in a Planetary Context

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A Corrigendum on

Living at the Extremes: Extremophiles and the Limits of Life in a Planetary Context

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In the original article, there was a mistake in the legend for **Table 4** as published. The legend in **Table 4** is missing two parentheses around "Poly." The correct legend appears below.

"Table 4. Examples of notable (Poly)extremophiles and their physiological requirements."

Additionally, there was a mistake in **Table 3** and **Table 5** as published. In **Table 3**, the lowest temperature listed for *Planococcus halocryophilus* Or1 is "−18°C." It should be "−15°C" instead. In addition, the pH range is "nr" but should be "6–11" instead. In the temperature column, 37 is bold type, but this should be regular type.

In **Table 5**, the atmosphere entry for Earth > Atmosphere > Geochemistry is listed as "8.1% N₂," but the actual composition of Earth's atmosphere is "78% N₂."

The corrected **Table 3** and **Table 5** appear below.

Lastly, there is a grammatical error in the original article.

A correction has therefore been made to the section **Can Life Originate, Evolve, or Survive on Other Planetary Bodies?**, paragraph five:

"Solar and galactic cosmic rays (high-energy particles with energies from 10 MeV to >10 GeV) present challenges to life on the surface and near-surface of Mars and other planetary bodies. However, any subsurface aquifer deeper than a few meters would be protected from damaging radiation. Dartnell et al. (2007) calculated the galactic cosmic ray dosage rates and the corresponding survival times (which they defined as a million-fold decrease in cell number) of characteristic microbes at different depths in Mars's subsurface. At the surface, *E. coli* has a survival time of 1,200 years, while at 20-m depth, that survival time jumps to 1.5×10^8 years. Compared to *E. coli*, *D. radiodurans* has survival times an order of magnitude longer. These survival times are, in fact, lower limits in light of recent measurements by the Radiation Assessment Detector onboard the Mars Science Laboratory (Hassler et al., 2014), which found that the actual dose rate at Gale Crater (76 mGy year^{-1}) is a factor of 2 lower than that modeled by Dartnell et al. (2007)."

The authors apologize for these errors and state that they do not change the scientific conclusions of the article in any way. The original article has been updated.

TABLE 3 | Limits of life as identified by (poly)extremophilic organisms in pure cultures.

Strain	Domain	Extremophile Type	Isolation ecosystem	Temperature (°C)	pH	Pressure (Mpa)	Salinity (%)	Water activity (a _w)	References
<i>Picrophilus oshimae</i> KAW 2/2	Archaea	Hyperacidophile	Hot springs, Solfataras	47–65 (60) ^a	−0.06 –1.8 (0.7)	nr	0–20	nr	Schleper et al., 1995, 1996
<i>Serpentinomonas</i> sp. B1	Bacteria	Alkaliophile	Serpentinizing system (water)	18–37 (30)	9– 12.5 (11)	nr	0–0.5 (0)	nr	Suzuki et al., 2014
<i>Methanopyrus kandleri</i> 116	Archaea	Hyperthermophile	Deep-sea hydrothermal vent	90– 122 (105)	(6.3–6.6)	0.4–40	0.5–4.5 (3.0)	nr	Takai et al., 2008
<i>Planococcus halocryophilus</i> Or1	Bacteria	Halopsychrophile	Sea ice core	−15 –37 (25)	6–11 (7–8)	nr	0–19 (2)	nr	Mykytczuk et al., 2012, 2013
<i>Halarsenatibacter silvermanii</i> SLAS-1	Bacteria	Haloalkaliphile	Soda lake	28–55 (44)	8.7–9.8 (9.4)	nr	20–35 (35)	nr	Oremland et al., 2005
<i>Thermococcus piezophilus</i> CDGS	Archaea	Piezothermophile	Deep-sea hydrothermal vent	60–95 (75)	5.5–9 (6)	0.1– 125 (50)	2–6 (3)	nr	Dalmasso et al., 2016
Haloarchaeal strains GN-2 and GN-5	Archaea	Xerophile	Solar salterns (brine)	nr	nr	nr	nr	0.635	Javor, 1984

^aData presented as range (optimum) for each parameter. nr, not reported in the original publication. Current limits are highlighted in bold.

TABLE 5 | Boundary conditions for different planetary bodies of astrobiological interest (compared to Earth), split into atmosphere, surface, and subsurface layers.

Planetary body	Type	Layer	Temperature (°C)	pH	Pressure (MPa)	Salinity (% NaCl)	Geochemistry	References
Earth	Planet	Atmosphere	-100 – 40	Neutral, local acidic conditions possible due to volcanism and human activities	0.0001 – 0.1	0	78% N ₂ , 21% O ₂ , 9340 ppm Ar, 400 ppm CO ₂ , 18.2 ppm Ne, 5.2 ppm He, 1.7 ppm CH ₄ , 1.1 ppm Kr, 0.6 ppm H ₂ , variable H ₂	Hans Wedepohl, 1995; McDonough and Sun, 1995; Wayne, 2000
		Surface	-98.6 – 464	-3.6 – 13.3	0.003 – 112	0 – saturation	Soils and sediments of varying lithologies, siliceous crust, ranging from mafic to felsic composition. Extensive ocean (70% planet surface), with 4,000 m average depth, 4°C and 3.5% average temperature and salinity respectively	
		Subsurface	3.25 – <400	~1 – 12.8	<800	0.05 – saturation	Soils and sediments of varying lithologies, siliceous crust, ranging from mafic to felsic composition, ultramafic mantle	
Venus	Planet	Atmosphere	-40 – 482 ^a	0 ^b	0.1 – 9.3 ^c	nr	96.5% CO ₂ , 3.5% N ₂ ; small quantities of CO, SO ₂ , HCl, HF, HDO, and H ₂ O; H ₂ SO ₄ condensates	Cockell, 1999; Basilevsky and Head, 2003; Schulze-Makuch et al., 2004; Lang and Hansen, 2006; Bertaux et al., 2007; Airey et al., 2017
		Surface	377 – 482	nr	4.5 – 9.3 ^c	nr	Rocks are similar to tholeiitic and alkaline basalts; no liquid water	
		Subsurface	nr	nr	nr	nr	Fluid channels; volcanism	
Mars	Planet	Atmosphere	-138 – 35 ^d	nr	0.0001–0.0009	nr	95.3% CO ₂ , 2.7% N ₂ , 1.6% Ar, 0.13% O ₂ , 0.08% CO; trace amounts of H ₂ O, NO, Ne, Kr, Xe	Varnes et al., 2003; Fairén et al., 2004; Nicholson and Schuerger, 2005; Hecht et al., 2009; Smith et al., 2009; Johnson et al., 2011; Jones et al., 2011; Michalski et al., 2013; Longstaff, 2014; Wordsworth, 2016; Sinha et al., 2017; NASA, 2018
		Surface	-138 – 30	7.7 ^e	0.0004–0.0009	5.2–5.8	Basaltic, Fe-/Mg-rich phyllosilicates, perchlorate salts, Al-rich clays, sulfates, chlorides, calcite, and silicas; potential cryosphere	
		Subsurface	55 ^g	4.96–9.13 ^h	10–303 ^g	Cl-rich brines	Potential groundwater; basalt crust; possible serpentinization	

(Continued)

TABLE 5 | Continued

Planetary body	Type	Layer	Temperature (°C)	pH	Pressure (MPa)	Salinity (% NaCl)	Geochemistry	References
Enceladus	Icy moon	Plume jets	0	~8.5 – 9	High velocity jets	> 0.5	90–99% H ₂ O, ≤0.61–4.27% N ₂ , 0.3–5.3% CO ₂ , 0.1–1.68% CH ₄ , 0.4–0.9% NH ₃ , 0.4–39% H ₂ , trace amounts of hydrocarbons; high mass organic cations, silicates, sodium, potassium, carbonates	Gioia et al., 2007; Postberg et al., 2009, 2018; Waite et al., 2009; Zolotov et al., 2011; Glein et al., 2015; Holm et al., 2015; Hsu et al., 2015; Taubner et al., 2018
		Icy shell (~10 km thick)	–233 – –23	nr	nr	May have ammonia brine pockets	May have tectonics	
		Subsurface global ocean (~0–170 km depth)	<90	8.5 – 12.2 ^k	1 – 8	0.45 – < 4	Possible serpentinization	
Titan	Icy moon	Atmosphere	–183 – –73 ^j	nr	> 0.01 – 0.15	nr	98.4% N ₂ , 1.4% CH ₄ , 0.2% H ₂ , trace hydrocarbons and organics; 95% N ₂ , 5% CH ₄ , 0.1% H ₂ ; ~50 ppmv CO and ~15 ppbv CO ₂ ; C ₂ H ₃ CN; clouds	Fulchignoni et al., 2005; de Kok et al., 2007; Norman, 2011; Baland et al., 2014; Mastrogiuseppe et al., 2014; Mitri et al., 2014; Sohl et al., 2014; Jennings et al., 2016; McKay, 2016; Mitchell and Lora, 2016; Brassé et al., 2017; Cordier et al., 2017
		Surface	–183 – –179	nr	0.15–0.35 ⁱ	nr	Lakes and sea have CH ₄ , C ₂ H ₄ , and dissolved nitrogen; dunes of solid organic material; low-latitude deserts and high-latitude moist climates	
		Subsurface	–18	11.8 ^l	50–300 ^m	Likely dense subsurface ocean (≤1,350 kg m ^{–3}) suggesting high salinity	CH ₄ and C ₂ H ₆	
Ceres	Dwarf planet	Atmosphere	nr	nr	nr	nr	Transient atmosphere with possible water vapo	Fanale and Salvai, 1989; Zolotov, 2009, 2017; Küppers et al., 2014; Hayne and Aharonson, 2015; Neveu and Desch, 2015; Hendrix et al., 2016; Villarreal et al., 2017; Vu et al., 2017; Castillo-Rogez et al., 2018; McCord and Castillo-Rogez, 2018; McCord and Zambon, 2019

(Continued)

TABLE 5 | Continued

Planetary body	Type	Layer	Temperature (°C)	pH	Pressure (MPa)	Salinity (% NaCl)	Geochemistry	References
		Surface	(−157– −30) ⁿ	9.7–11.3 ⁿ	nr	<10 ⁿ	Surface clays; (Mg, Ca)-carbonates; (Mg, NH ₄)-phyllosilicates; Fe-rich clays; salt deposits; chloride salts; water-rock interactions; brucite and magnetite; sulfur species and graphitized carbon; localized Na-carbonates (e.g., Na ₂ CO ₃), NH ₄ Cl, NH ₄ HCO ₃	
		Subsurface	−143 – −93 ^o	Likely alkaline	<140 – 200 ^p	Potentially has briny or NH ₃ -rich subsurface liquid	Active water/ice-driven subsurface processes	
Europa	Icy moon	Atmosphere (tenuous)	nr	nr	0.1–12 – 1–12	nr	Ion sputtering of the surface; potential water plumes; O ₂ ; trace amounts of sodium and potassium	Spencer et al., 1999; Chyba and Phillips, 2001; Marion et al., 2005; McGrath et al., 2009; Zolotov and Kargel, 2009; Travis et al., 2012; Cassidy et al., 2013; Muñoz-Iglesias et al., 2013; Kattenhorn and Prockter, 2014; Soderlund et al., 2014; Hand and Carlson, 2015; Kimura and Kitadai, 2015; Noell et al., 2015; Vance et al., 2016; Teolis et al., 2017; Zhu et al., 2017; Jones et al., 2018; Martin and McMinn, 2018; Pavlov et al., 2018
		Surface (icy shell)	−187 – −141	nr	0.1–12	May be saline, as delivered to the surface from a salty ocean, may have brine or salt inclusions	H ₂ O ₂ , H ₂ SO ₄ , CO ₂ ; salts concentrated in cracks; oxidants and simple organics; potentially MgSO ₄ , Na ₂ SO ₄ , Na ₂ CO ₃ , may have gas inclusions; may have tectonics	
		Subsurface ocean	Daily inundation of seawater at T= −4 – 0	Potential for wide range ^q	0.1 – 30 ^r	<3.5	Likely contains Mg ²⁺ , SO ₄ ²⁻ , Na ⁺ , Cl ⁻ ; oxidants and simple organics	

The observed or putative geochemistry as well as other potential influences are also listed. ^aThermosphere can be as cold as −173°C (Bertaux et al., 2007); the upper-to-middle cloud layers are between −40 and 60°C (Cockell, 1999). ^bAcid concentration in upper cloud layer is 81%, in lower layers up to 98% (Cockell, 1999). ^cUp to 11 MPa in a deep depression (Basilevsky and Head, 2003). ^dSummer air temperatures on Mars near the equator can reach a maximum of 35°C (Longstaff, 2014). ^eMeasured by the Phoenix Mars Lander Wet Chemistry Laboratory at the northern plains of the Vastitas Borealis (Hecht et al., 2009). ^fLiquid water may have had water activity > 0.95 (Fairén et al., 2009). ^gCalculated temperature at a depth of 1–30 km (Jones et al., 2011; Sinha et al., 2017); at a depth ~310 km, the calculated temperature is <427°C (Jones et al., 2011); the Martian core has temperature 1527°C (Longstaff, 2014). ^hCalculated groundwater pH (Varnes et al., 2003). ⁱCalculated pressure at Titan's large sea, Ligeia Mare, is 0.20–0.35 MPa (Cordier et al., 2017). ^jTropospheric temperature can be −193°C; 80% of incident sunlight is absorbed by Titan's atmosphere, suggesting that there are greenhouse and anti-greenhouse effects (Mitchell and Lora, 2016). ^kThe subsurface ocean on Enceladus could also have pH range 10.8–13.5 (Glein et al., 2015). ^lCalculated ocean pH with 5 wt% NH₃ (Brassé et al., 2017). ^mCalculated pressure for the subsurface ocean with thickness 100 km and outer shell thickness 40–170 km (Baland et al., 2014); 800 MPa at the mantle ice shell-core boundary (Sohl et al., 2014). ⁿCalculated surface temperatures, illuminated surfaces can have temperature <−173°C (Hayne and Aharonson, 2015); calculated pH and salinity for bright deposits in Occator crater (Zolotov, 2017); temperature for bright deposits in Occator crater might reach <−0.2°C (Zolotov, 2017). ^oInternal temperature might reach 77°C (McCord and Sotin, 2005). ^pCeres' center pressure (Zolotov, 2009). ^qAcid brine may result from hydrothermal systems and be enriched with sulfuric acid (Kargel et al., 2000); neutral brine may occur as leachate from chondritic material and be enriched with magnesium sulfate (Kargel et al., 2000; Pasek and Greenberg, 2012); alkaline brine may occur in areas with natron (Na₂CO₃·10H₂O), produced from the venting of CO₂ from aqueous reservoirs (Langmuir, 1971; Millero and Rabintra, 1997). ^rAt the base of a 100 km European ocean, the pressure is calculated to be 146 MPa (Marion et al., 2005).

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