



Synthesis of Hollow Nanofibers and Application on Detecting SF₆ Decomposing Products

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Hollow structured nanofibers have attracted much attention in diverse domains owing to their unique physicochemical properties and characteristics. Gas sensing is one of the most promising applications. In electrical engineering, the detection of SF₆ decomposing gas products is significant to monitor the insulation status online of gas insulated switchgear (GIS). This mini-review presents the developments of hollow structured nanofibers in synthesis strategies and gas sensing application, especially in the detection of SF₆ decomposing gas products, including hydrogen disulfide (H₂S), sulfur dioxide (SO₂), thionyl fluoride (SOF₂), and sulfuryl fluoride (SO₂F₂). In addition, the gas sensing mechanism of metal oxide hollow nanofibers based gas sensor are discussed.

Keywords: hollow nanofibers, sensing application, sensing mechanism, SF₆ decomposing products, gas insulated switchgear

INTRODUCTION

Sulfur hexafluoride (SF₆) as an insulating gas is commonly utilized in gas insulated switchgear (GIS) (Zhang et al., 2018a). Though SF₆ has high stability at normal operating temperatures, parts of the SF₆ molecules would react with little water and solid insulated material under electrical stress conditions caused by arc, spark, and partial discharge, and then decompose to a variety of typical decomposition byproducts including SO₂, H₂S, SOF₂, SO₂F₂, SOF₄, S₂F₁₀, HF gases, and solid byproducts (Li et al., 2018; Zhou et al., 2018a; Chen et al., 2019c). This can lead to insulating property reduction of GIS. Furthermore, the research results showed that the type, concentration, and formation regularity of the SF₆ decomposition byproducts and the severity of the insulation fault have the close correlation (Tang et al., 2018; Zhang et al., 2018b). Therefore, the detection of the typical decomposition components of SF₆ gas is significant for identifying and diagnosing the insulation faults of the GIS. Over the past few years, photoacoustic (PA) spectroscopy (Luo et al., 2015), gas chromatography (Jong et al., 2015), and infrared absorption spectrometry (Dong et al., 2017) have been applied for the detection of the decomposition components of SF₆. Nevertheless, the preceding three methods only monitor the insulation status offline. With the development of gas sensitive materials, gas sensors for detecting SF₆ decomposing products have developed further in recent years. Gas sensors detection methods have grown to analyzing the insulation status of GIS online (Liu et al., 2017; Lu et al., 2018; Cui et al., 2019).

There are a variety of gas-sensitive materials that can be made into SF₆ decomposition component gas sensor, including semiconducting oxides, carbon nanotubes, molybdenum disulfide, etc. (Chen et al., 2018, 2019a,b). Nanostructure materials received extensive

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attention recent years due to the outstanding physical and chemical properties (Xu et al., 2017b,c; Zhou et al., 2018b). Hollow structured nanofibers as potential gas sensing nanomaterials have attracted increasing research due to their extensive specific surface areas, larger accessible active area, and higher aspect ratios (Zhang et al., 2009; Li et al., 2017; Diltemiz and Ecevit, 2019). Although the fabrication and application of hollow nanofibers have been well reviewed (Kenry and Lim, 2017; Thenmozhi et al., 2017). Few studies focused on the application of detecting SF₆ decomposing gas products in the field of electrical engineering, which developed in recent years. In this mini review, synthesis strategies of novel hollow nanofibers and their gas sensing application, especially detecting SF₆ decomposing gas products, are introduced.

SYNTHESIS STRATEGIES OF HOLLOW NANOFIBERS

Hollow Nanofibers Based on Template Synthesis

Up to now, template synthesis, self-assembly, and electrospinning technique have been employed for producing the hollow structured nanofibers (Khajavi and Abbasipour, 2012; Wang and Lu, 2012; Kim et al., 2019). Among them, the template synthesis method is more adaptable, because it is easy to control the width of the middle hollow (Homaeigohar et al., 2017). As shown in **Figure 1A**, the template is mostly prepared by electrospinning. Then the target precursor is coated or deposited on the prepared template. The employed deposition technique includes atomic layer, electrochemistry, chemical vapor deposition, and so on. The last process is calcination in order to remove the core template fibers. Donmez et al. fabricated hafnia hollow nanofibers by atomic layer deposition using electrospun nylon 6,6 nanofibers as templates (Donmez et al., 2013). Hollow TiO₂ nanofibers were fabricated by using oxidized electrospun polyacrylonitrile fibers as templates (Ouyang et al., 2018). Han et al. used controllable sacrifice template route to synthesize polyaniline hollow nanofibers with adjustable hollow structure (Han et al., 2019b). Kim et al. successfully synthesized ultra-thin SiO₂ hollow nanofibers with surface-replicated morphologies using herringbone-, platelet-, and tubular-type carbon nanofibers templates (Kim et al., 2019). The template synthesis methods can create hollow nanofibers of different diameters and hollow structure but cannot prepare them continuously.

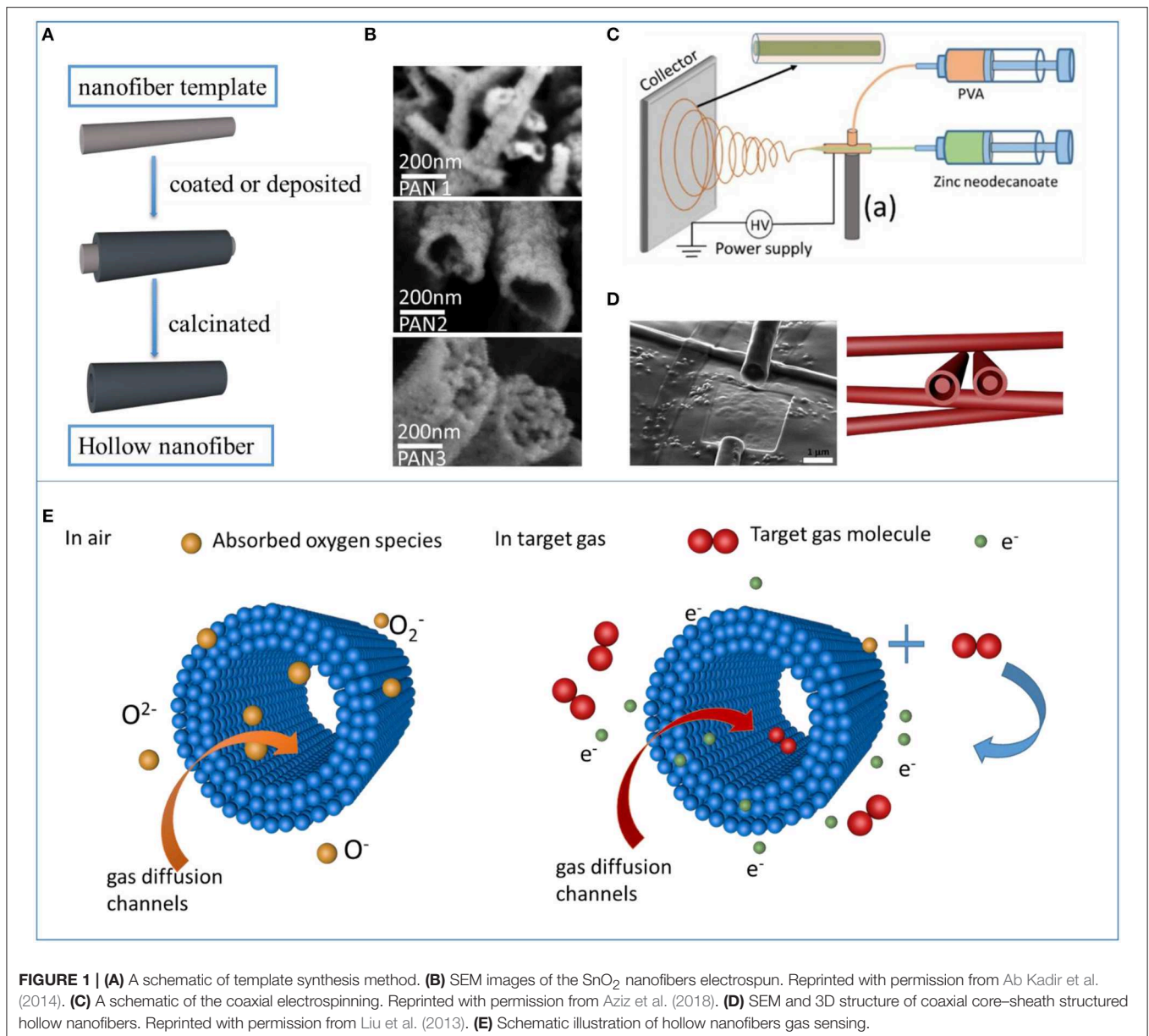
Hollow Nanofibers Based on Self-Assembly Methods

The self-assembly technique has attracted wide interest and employed to form polymers owing to the omitting preparation of template and post-treatment of removing template. There have been many studies on preparing nanotubes. However, there is very little research on hollow nanofiber structured polymers fabricated by self-assembly because of the special continuous hollow structure. Wang et al. successfully prepared polyaniline

nanomaterials with different morphologies including 1D-nanotubes, 3D-nanofiber by simply changing the concentration of aniline and the molar ratio of dichloroacetic acid to aniline. They found that strong hydrogen bonding effect between dichloroacetic and aniline may result in the formation of self-assembly nanofibers networks, and thus producing phenazine-like units to further self-assembly (Wang and Lu, 2012).

Hollow Nanofibers Based on Electrospinning Methods

Compared with template synthesis and self-assembly methods which need complex manipulations of molecules to form the nanofibers structure, electrospinning is the most straightforward, and efficient method to prepare continuous hollow nanofibers (Xia et al., 2012). Firstly, the high voltage is applied between a positively-charged syringe needle and a grounded collector. Then, the charged solution is stretch under the electronic forces to produce nanofibers. Usually, in order to obtain hollow nanofibers by electrospinning, the nanofibers with core-shell structure should be prepared firstly. Then, the inner layer material is removed by heat treatment or solvent extraction to obtain the hollow structured nanofibers (Di et al., 2008; Zhang and Hsieh, 2009). Many kinds of electrospinning apparatus and methods including single spinneret, coaxial, triaxial, microfluidic, and emulsion electrospinning have been applied to prepare hollow nanofibers (Wei et al., 2011; Zhang et al., 2019). The mechanism of single spinneret electrospinning to produce hollow nanofibers is the phase separation (Wu et al., 2017). Tang et al. fabricated multichannel hollow TiO₂ nanofibers by single-nozzle electrospinning. The hollow structure was formed because the polyvinyl pyrrolidone (PVP) and polyacrylonitrile (PAN) are immiscible caused phase separation (Tang et al., 2013). Ab Kadir et al. prepared hollow SnO₂ nanofibers and demonstrated that the diameter of nanofibers depends on the polymer concentration in the electrospinning solution (Ab Kadir et al., 2014). The SEM is shown in **Figure 1B**. The PAN concentration of three kinds of samples is different from PAN 1 to PAN 3 (the ratio of them is 1:2:3). The results showed that with the increase of solution concentration from PAN 1 to PAN 2 and PAN 3, thicker nanofibers will be formed. As shown in SEM of PAN 3, the nanostructure is filled nanofibers. That is because the stirring speed is different from PAN 1 and PAN2 to PAN 3 which is faster than the others. As the stirring speed increases, the droplet diameter of PAN decreases, and the PAN which are supposed to form the core of the nanofibers will disperse in the PVP solution more randomly. Coaxial electrospinning technology is more flexible to form hollow nanofiber. The experimental setup has been shown in **Figure 1C** (Aziz et al., 2018). The coaxial electrospinning technology follows the traditional electrospinning principal, just replaces the traditional single spinneret with a spinneret composed of two coaxial capillaries. Compared with single spinneret electrospinning which can only produce hollow nanofibers of a limited kind of material, the advantages of coaxial electrospinning is that a diverse range of materials can be made into hollow nanofibers (Yoon et al., 2018). Hyun et al.



utilized the coaxial electrospinning technology to prepare solid, hollow, multi-channel, and hollow multi-channel nanofibers by adjusting the shell-core spinning solution and combining the post-heat treatment process (Hyun et al., 2016). To format nanofibers with higher surface areas, the triaxial electrospinning was developed in recent years. The triaxial electrospinning has been widely used in the field of biomedical applications. Liu et al. designed a triaxial electrospinning system and successfully developed a biodegradable multilayer nanofiber (Liu et al., 2013). The structure of triaxial electrospinning can produce coaxial core-sheath structured hollow nanofibers as shown in **Figure 1D**. The spinneret of microfluidic electrospinning apparatus consists of several inner capillaries. This method is mainly used in biomedical and textile field (Chae et al., 2013). Emulsion electrospinning spinning can be used to produce

hollow nanofibers without changing the spinneret, but the continuity of the prepared nanofibers is low (Wei et al., 2015).

Up to now, diverse electrospun hollow nanofibers which are made of a range of materials such as polymers, ceramics, semiconductors, and composites have been made for different application including batteries, hydrogen storage and generation, biological sensing, and gas sensing. In the next part of this mini review, we focus on the gas sensing application, especially the SF_6 decomposing gas products.

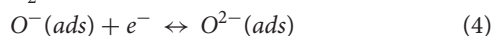
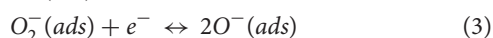
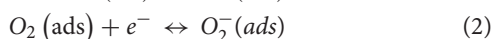
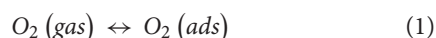
APPLICATION OF HOLLOW NANOFIBERS IN GAS SENSING

Gas sensing is one of important applications of hollow nanofibers. Among gas sensing materials, semiconducting metal

oxides such as SnO₂, ZnO, TiO₂, and CuO are applied the most widely in gas sensors due to the excellent gas sensing properties in terms of response, response and recovery times, and stability (Xu et al., 2017a; Zhou et al., 2018c; Wang et al., 2019a). The higher specific surface area of sensing materials can provide the higher sensitivity. The surface area of hollow structured nanofibers is around two times than that of solid nanofibers (Wali et al., 2014). Therefore, the gas sensing applications such as the detection of volatile organic compounds, inflammable gases, and toxic gases of hollow nanofibers, have been activity studied in recent years. Katoch et al. reported that the ZnO hollow fibers with smaller hole diameters show higher gas sensitivity to both oxidizing and reducing gases than those with larger hole diameters. The improved gas sensing properties is attributed to the increased surface area of the nanomaterials (Katoch et al., 2016). Besides increasing the active surface area, by doping metallic nanoparticles or constructing nano-heterojunction can also improve the sensing capability especially enhance the selectivity of gas sensor. Gas sensor produced by electrospinning of pure ZnO hollow nanofibers and Ce doped ZnO hollow nanofibers have been successfully synthesized and their gas sensing properties were measured. The result showed that Ce doping can improve the response performance toward acetone, ethanol, acetic acid, DMF and ammonia (Li et al., 2015). In addition, 0.6 wt% Pr-doped SnO₂ hollow nanofibers exhibit the higher response of ethanol (Li et al., 2014). ZnO/CoNiO₂ composite hollow nanofibers offered a good choice of ammonia gas sensing materials (Alali et al., 2017). Besides, the metallic nanoparticles doping concentration obviously affects specific gas sensitivity. Choi et al. reported 0.08 and 0.4 wt% Pd doped SnO₂ hollow nanofibers increase and decrease the sensitive of C₂H₅OH, respectively (Choi et al., 2010).

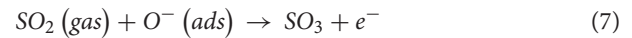
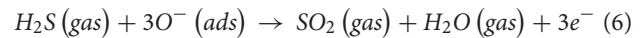
Gas Sensing Mechanism

The gas sensing mechanism of the hollow nanofibers materials is relevant to the reactions between the target gas molecule and the oxygen species pre-adsorbed (O₂⁻, O⁻, O²⁻) on the surface as shown in **Figure 1E**. When the sensor is exposed in air, oxygen molecules on the surface of nanomaterials convert into surface adsorbed oxygen species including O₂⁻, O⁻, and O²⁻ by capturing electrons from conduction band of the nanomaterials. Therefore, the electron depletion region extends and it can lead to the increase of the gas sensor resistance. The reaction can be described as follow (Li et al., 2014):



When the sensor is exposed to the target gases, these gas molecules react with adsorbed oxygen and then release the electrons back to the conduction band of nanomaterials. Thus, the resistance of the sensor decreases at target gas atmosphere.

The following equations take H₂S and SO₂ gas as examples (Han et al., 2019a; Queraltó et al., 2019):



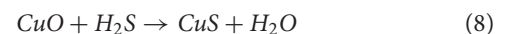
The high specific surface area offered by inner and outer walls of hollow nanofibers provides more active centers of gas molecules adsorption. In addition, the sufficient diffusion channels makes it easier for gas molecules to diffuse into the inner surface.

Hollow Nanofibers Sensor for the SF₆ Decomposing Products

For the SF₆ decomposition component, current research efforts are focused on the detection of SO₂ and H₂S gases. Combined with the research, some of the typical gas sensing performance of SO₂ and H₂S sensors are listed in **Table 1**.

S, the gas sensing sensitivity of the gas sensor. The S was determined by the formula in the **Table 1**. Ra is the resistance of sensor in air. Rg is the resistance of sensor in SO₂ or H₂S gas.

As shown in the table above, compared with filled nanofibers or other nanostructure, hollow structured nanofibers sensors show excellent gas sensitivity, and lower working temperature to H₂S and SO₂. Albert Queraltó et al. prepared single phase LaFeO₃ (LFO) perovskite nanofibers for high detection of sulfur-containing gases. The detection limitation of H₂S and SO₂ was as low as 0.5 ppm at 250°C (Queraltó et al., 2019). Nanomaterials decorated CuO such as CuO-ZnO and CuO-SnO are potential H₂S gas sensing materials. The 0.04 CuO-ZnO composite hollow spheres prepared by Kim et al. exhibited high response to 5 ppm H₂S (Ra/Rg = 32.4) at 336°C (Kim et al., 2012). 0.3 CuO-ZnO composite hollow nanofibers showed high performance to H₂S gas detection at 250°C (Han et al., 2019a). The response value of CuO modified hollow SnO₂ nanofibers gas sensor to 10 ppm H₂S gas reached 410 at 125°C. And this sensor is almost no response to other gases when the temperature is lower than 200°C indicating the excellent selectivity toward H₂S gas (Yang et al., 2017). Liang et al. synthesized CuO-loaded In₂O₃ hollow nanofibers for H₂S gas sensing application. It is observed that the sensor shows high gas response (9.17 × 10³ toward 5 ppm H₂S) at room temperature (Liang et al., 2015). The high response of hollow nanofibers decorated CuO can be attributed to the specific interaction between H₂S molecular and CuO nanoparticles. The reaction between CuO and H₂S is as follow:



At present, there are limited researches on the detection of other SF₆ decomposing products such as SOF₂ and SO₂F₂. Present research mainly focus on the sensor with nanotube structure. Zhang et al. were devoted to the application of TiO₂ nanotubes and carbon nanotubes sensors to monitor SF₆ decomposition byproducts online of GIS equipment. The team researched the adsorption process between gas sensing material (including functional group-modified or metal/non-metal-doped carbon nanotubes) and SF₆ decomposing products based on density functional theory. For example, they investigated the adsorption

TABLE 1 | Comparison of the representative hollow nanofibers based sensors for H₂S and SO₂.

SF ₆ decomposing products	Sensing material	Synthesis method	Working temperature (°C)	S/ppm	Formula of S	Detection limitation	References
H ₂ S	LaFeO ₃ hollow nanofibers	Electrospinning	200	160/4	(Rg-Ra)/Ra	0.5 ppm	Queralto et al., 2019
	p-CuO/n-ZnO hollow nanofibers (Ratio Zn/Cu = 15.6)	Combining electrospinning, ALD, and calcination methods	250	60.5/100	Ra/Rg	/	Han et al., 2019a
	CuO modified hollow SnO ₂ nanofibers	Electrospinning	125	410/10	Rg/Ra	Below 2 ppm	Yang et al., 2017
	CuO on In ₂ O ₃ nanofiber	Electrospinning	25	9,170/5	Ra/Rg	<400 ppb	Liang et al., 2015
SO ₂	LaFeO ₃ hollow nanofibers	Electrospinning	250	100/1	(Rg-Ra)/Ra	0.5 ppm	Queralto et al., 2019
	Ni-doped ZnO nanorods	Hydrothermal method	220	40/50	Ra/Rg	/	Wang et al., 2019b
	TiO ₂ nanotube array	Assisted-template method	200	14.35/10	Ra/Rg	/	Zhang et al., 2016

process between four types (H₂S, SO₂, SOF₂, and SO₂F₂) of SF₆ decomposing products and Co-doped single-walled carbon nanotubes (SWCNT) using density functional theory calculations. The result showed that the adsorption leads to the obvious bond breakings of SOF₂ and SO₂F₂. The conductance of Co-SWCNT changed significantly when the SOF₂ or SO₂F₂ gases absorbed on the sensing materials, which indicated that the Co-SWCNT is a promising SOF₂ or SO₂F₂ gas sensor (Zhang et al., 2017). Gui et al. studied the adsorption mechanism between N-doped SWCNT and SOF₂, SO₂F₂ gases. The N doping reduces the adsorption capacity of N-doped SWCNT to SOF₂, SO₂F₂ gases. Thus, the problem of poisoning sensor can be fixed (Gui et al., 2018). The team also discussed TiO₂ nanotube arrays and the adsorption of SO₂, SOF₂, and SO₂F₂ on different surface of TiO₂. The gas sensing responses of Pt, Au- modified TiO₂ are better than pure TiO₂ (Zhang et al., 2016). Although the most of present studies focus on the TiO₂ and carbon nanotubes, the hollow structure and density functional theory calculations can inspire further research on the detection of SF₆ decomposing products. Meanwhile, the adsorption process between traditional metal oxide semiconductor and SF₆ decomposing products has been researched based on density functional theory. Wang et al. reported that Ni dopant can improve the adsorption capacity and selectivity of ZnO materials to SO₂, SOF₂, SO₂F₂ gases (Wang et al., 2019b). The authors believe that the hollow nanofibers for SF₆ decomposing products detection application is feasible.

CONCLUSION

In this mini review, the synthesis strategies and gas sensing application of hollow structured nanofibers have been summarized, especially the detection of SF₆ decomposing

products. Various gas sensors based on hollow nanofibers materials with ultrahigh response value and low detection limit have been applied to gas detection. The hollow structure provides more diffusion channels and active surface area between target gas and gas sensing nanomaterials. It can be foreseen that more hollow nanofibers gas sensing materials will be synthesized and applied in the near future. Although achieving good performance, there is still a lot of room for development. Further enlarging the surface area of nanofibers is an effective way to enhance gas sensing properties. For example, the multichannel hollow nanofibers with more diffusion channels for gas spreading can be applied in gas sensing tests. In addition, the further study on high performance gas sensor and its application in the detection of SF₆ decomposing products should be investigated, aiming to the actual industrial application in SF₆ insulated equipment.

AUTHOR CONTRIBUTIONS

All authors listed have made a substantial, direct and intellectual contribution to the work, and approved it for publication.

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Conflict of Interest Statement: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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