



# Large Stimulation of Recalcitrant Dissolved Organic Carbon Degradation by Increasing Ocean Temperatures

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### Specialty section:

This article was submitted to  
Marine Biogeochemistry,  
a section of the journal  
Frontiers in Marine Science

**Received:** 30 October 2017

**Accepted:** 18 December 2017

**Published:** 12 January 2018

### Citation:

Lønborg C, Álvarez-Salgado XA,  
Letscher RT and Hansell DA (2018)  
Large Stimulation of Recalcitrant  
Dissolved Organic Carbon  
Degradation by Increasing Ocean  
Temperatures. *Front. Mar. Sci.* 4:436.  
doi: 10.3389/fmars.2017.00436

More than 96% of organic carbon in the ocean is in the dissolved form, most of it with lifetimes of decades to millennia. Yet, we know very little about the temperature sensitivity of dissolved organic carbon (DOC) degradation in a warming ocean. Combining independent estimates from laboratory experiments, oceanographic cruises and a global ocean DOC cycling model, we assess the relationship between DOC decay constants and seawater temperatures. Our results show that the apparent activation energy of DOC decay ( $E_a$ ) increases by three-fold from the labile (lifetime of days) and semi-labile (lifetime of months) to the semi-refractory (lifetime of decades) DOC pools, with only minor differences between the world's largest ocean basins. This translates into increasing temperature coefficients ( $Q_{10}$ ) from 1.7–1.8 to 4–8, showing that the generalized assumption of a constant  $Q_{10}$  of  $\sim 2$  for biological rates is not universally applicable for the microbial degradation of DOC in the ocean. Therefore, rising ocean temperatures will preferentially impact the microbial degradation of the more recalcitrant and larger of the three studied pools. Assuming a uniform 1°C warming scenario throughout the ocean, our model predicts a global decrease of the DOC reservoir by  $7 \pm 1$  Pg C. This represents a 15% reduction of the semi-labile + semi-refractory DOC pools.

**Keywords:** temperature, dissolved organic carbon, mineralization, Arrhenius law, global ocean biogeochemical model

## INTRODUCTION

Dissolved organic carbon (DOC) in the ocean represents one of the largest reservoirs of organic matter on the Earth's surface, containing a similar amount of carbon as atmospheric carbon dioxide ( $\text{CO}_2$ ) (Hedges, 2002). The main processes responsible for the removal of DOC from the ocean water column are: (1) photochemistry, where DOC is degraded directly to  $\text{CO}_2$ , carbon monoxide (CO) or low molecular weight organic compounds readily available for prokaryote uptake (Moran and Zepp, 1997; Mopper et al., 2015); (2) abiotic aggregation into microparticles (Kerner et al., 2003) or sorption to particles (Chin et al., 1998); (3) abiotic degradation via free radical reactions with oxygen (Rontani et al., 2014); and (4) degradation by marine heterotrophic prokaryotes (Lønborg and Álvarez-Salgado, 2012), with the latter representing the major DOC sink in the ocean (Carlson and Hansell, 2015).

Although the DOC pool comprises a myriad of compounds covering a widespread molecular size-reactivity continuum (Amon and Benner, 1996; Benner and Amon, 2015), it is operationally divided into a biologically labile fraction ( $\text{DOC}_L$ ) that is degraded over hours to days, a semi-labile fraction ( $\text{DOC}_{SL}$ ) with turnover times from weeks to months, a semi-refractory fraction ( $\text{DOC}_{SR}$ ) that can be stored for decades, a refractory fraction ( $\text{DOC}_R$ ) with lifetimes of thousands of years, and an ultra-refractory fraction ( $\text{DOC}_{UR}$ ) resistant to removal for tens of thousands of years (Hansell, 2013). Bioavailability has traditionally been linked to the chemical composition and structure of DOC, but more recently it has been shown to also vary depending on a range of other external factors (Raymond and Spencer, 2015). These factors include nutrient status, redox state, mineral-particle associations, terrestrial inputs, sunlight, biological production of recalcitrant compounds, changing microbial community composition, and the effect of priming or the extreme dilution of individual molecules (e.g., Amon and Benner, 1996; Thingstad et al., 1999; Del-Giorgio and Davies, 2003; Bianchi, 2011; Keil and Mayer, 2014; Arrieta et al., 2015; Letscher et al., 2015). About 4% of the ocean's net primary production escapes degradation in the surface layer to be exported as DOC to the ocean interior at a rate of about  $2 \text{ Pg C year}^{-1}$  (Hansell et al., 2012), roughly equalling the annual net ocean uptake of  $\text{CO}_2$  (Sarmiento and Gruber, 2006). Therefore, climate induced changes in DOC production, degradation, and/or export to deeper waters could potentially impact the atmospheric  $\text{CO}_2$  concentration and the Earth climate system. Previous studies have proposed that changes in the degradation of ocean DOC over global scales could have an important role in regulating the long-term climate, with the rate and magnitude of degradation impacting atmospheric  $\text{CO}_2$  levels, which in turn influences the cooling and warming of the Earth surface (Peltier et al., 2007; Swanson-Hysell et al., 2010; Sexton et al., 2011).

The temperature sensitivity of natural organic matter degradation has been studied extensively in soils (Davidson et al., 2000; Davidson and Janssens, 2006), demonstrating that increasing ambient temperature preferentially stimulates the microbial degradation of the more recalcitrant organic matter fractions. These studies suggest that the lability of a substrate can be defined by the change in the activation energy ( $E_a$ ) needed in order for the degradation to take place; the more recalcitrant the compound, the higher its  $E_a$  and, following the Arrhenius law, the higher its temperature sensitivity to microbial degradation (Davidson et al., 2000; Davidson and Janssens, 2006; Sierra, 2012). In the marine environment, studies of the microbial activity response to temperature in sediments have shown that the  $E_a$  varies as a function of compound class and the organisms degrading the organic matter (Westrich and Berner, 1988; Middelburg et al., 1996). More recently, the Arrhenius law has been used for calculating the  $E_a$  of plankton community metabolism, suggesting that marine autotrophs and heterotrophs might react differently to ocean warming (Yvon-Durocher et al., 2012; Chen and Laws, 2017). It has also been shown that the temperature dependence of heterotrophic prokaryote production is fundamentally different between the upper and

deeper layers of the ocean (Lønborg et al., 2016). Dissolved oxygen utilization rates in the ocean, computed from dissolved oxygen concentrations and apparent water age estimates, have also revealed a noteworthy temperature sensitivity of microbial respiration (Brewer and Peltzer, 2016, 2017).

Here, we quantify the  $E_a$  associated with the microbial degradation of  $\text{DOC}_L$ ,  $\text{DOC}_{SL}$ , and  $\text{DOC}_{SR}$ , which partly controls the ocean prokaryotic production and oxygen consumption. To achieve this goal, we combine data from laboratory microbial cultures and latitudinal DOC concentration and water mass age gradients collected during oceanographic cruises, with the outputs from a global ocean DOC cycling model. This analysis allows us to also assess the impact of future ocean warming on the microbial degradation and size of the DOC fractions and the potential implications for the ocean carbon cycle.

## METHODS

Sensitivity of the  $\text{DOC}_L$  and  $\text{DOC}_{SL}$  decay constants to temperature, on the order of days to months, can be tested with microbial incubation experiments. However, this approach is not feasible for  $\text{DOC}_{SR}$ , which turns over in years to decades. In this case, calculating rates using latitudinal gradients of DOC concentrations and water ventilation times obtained during oceanographic cruises are more suitable (Carlson et al., 2010). Furthermore, models that combine ocean circulation with DOC biogeochemical cycling provide insights into the global implications of temperature dependence of  $\text{DOC}_{SL}$  and  $\text{DOC}_{SR}$  degradation.

### Determination of $\text{DOC}_L$ and $\text{DOC}_{SL}$ Decay Constants from Incubation Studies

Studies of the degradation of  $\text{DOC}_L$  usually involve the dilution of a natural microbial community to follow the utilization of DOC over days in controlled laboratory conditions; when the incubation time is extended for months, the kinetics of the degradation of  $\text{DOC}_{SL}$  can also be followed (Lønborg and Álvarez-Salgado, 2012). While numerous studies have determined the  $\text{DOC}_L$  and  $\text{DOC}_{SL}$  decay constants ( $k_C$ ) measuring changes in DOC concentration over time using incubation experiments (see Lønborg and Álvarez-Salgado, 2012; for recent overview) very few were conducted at different temperatures (Table 1). In order to overcome methodological differences between the few available studies, we only included data where changes in DOC concentration over time had been measured at different temperatures. These include laboratory incubations using phytoplankton-derived DOC (Seiki et al., 1991), marine humic substances (Bussmann, 1999), and DOC collected in a coastal system (Lønborg et al., 2009). In one case (Bussmann, 1999), only initial DOC was measured but the time course of dissolved oxygen was followed over time. We included this study in the analysis because of the interest on the degradation of natural humic substances. Therefore, it should be kept in mind that in this case the temperature response of DOC respiration rather than utilization (= growth + respiration) has

**TABLE 1** | Summary of values obtained applying the Arrhenius law to experimental DOC degradation.

	DOC pool	$Q_{10}$	$E_a$ (kJ mol <sup>-1</sup> )	Temp. range (°C)	Ref. temp (°C)	$k_C$ (10 <sup>-3</sup> day <sup>-1</sup> )	Turnover time (day)	Ref.
Incubation of phytoplankton-derived DOC	DOC <sub>L</sub>	1.7 ± 0.1	37 ± 3	[5, 30]	15	110 ± 10	9.2 ± 0.8	29
Incubation of natural marine DOC (spring)	DOC <sub>L</sub>	1.7 ± 0.3	35 ± 10	[8, 18]	11	61 ± 6	16 ± 2	31
Incubation of natural marine DOC (summer)	DOC <sub>L</sub>	1.8 ± 0.3	41 ± 11	[8, 18]	14	101 ± 4	9.9 ± 0.3	31
Incubation of natural marine DOC (winter)	DOC <sub>SL</sub>	2.1 ± 0.4	50 ± 11	[8, 18]	8	13 ± 1	76 ± 8	31
Incubation of isolated humic substances	DOC <sub>SL</sub>	2.7 ± 0.4	67 ± 11	[-1, 10]	0	5 ± 1	220 ± 26	30
Bathypelagic N Atlantic DOC gradients/ventilation times	DOC <sub>SR</sub>	3.8 ± 0.5	87 ± 9	[2.5, 7.5]	5	0.20 ± 0.04	5 ± 1 × 10 <sup>3</sup>	32
Mesopelagic N Atlantic DOC gradients/ventilation times	DOC <sub>SR</sub>	8 ± 3	146 ± 21	[7.5, 14.5]	11	0.20 ± 0.07	5 ± 2 × 10 <sup>3</sup>	32

The temperature coefficient,  $Q_{10}$ , apparent activation energy,  $E_a$ , temperature ranges used for calculation, and the average DOC decay constants standardized to a reference temperature,  $k_C$ . Average values ± standard errors are shown. DOC pool assignments were based on the DOC turnover times; DOC<sub>L</sub>, days; DOC<sub>SL</sub>, months; DOC<sub>SR</sub>, years.

been followed. DOC samples collected during these experiments were either left unfiltered (Bussmann, 1999) or passed through a 0.45 μm membrane (Seiki et al., 1991) or GF/F filter (Lønborg et al., 2009). All studies but one (Seiki et al., 1991), which used wet oxidation, employed high-temperature combustion to determine the DOC concentration. Note that studies inferring the temperature sensitivity of marine organic matter degradation in long-term microbial incubation experiments using oxygen consumption estimates were not included in our analysis as they either did not distinguish between the particulate and dissolved fraction and/or did not report initial DOC values (e.g., Robinson and Williams, 2005; Hansen and Bendtsen, 2014; Bendtsen et al., 2015).

## Estimation of DOC<sub>SR</sub> Decay Constants from Latitudinal Gradients

Field data used in this work were collected in June–July 2003 along the CLIVAR section A16N in the North Atlantic Ocean (latitudinal range: 6°S to 63°N; longitude range: 20°–29°W) as part of the US CLIVAR Repeat Hydrography program (See Figure 6 in Hansell, 2013). Hydrographic data were obtained at 73 stations spaced by about 50 km using a conductivity, temperature, and depth (CTD) profiler equipped with 36 twelve-liters Niskin bottles. More than two thousand DOC samples were collected during the cruise. Samples taken in the depth range 0–250 m were passed through an in-line polycarbonate filter cartridge holding a combusted GF/F filter attached directly to the Niskin bottle, while samples below this depth were left unfiltered.

The ventilation time of each sample, an estimate of water apparent age, was estimated from measurements of chlorofluorocarbon (CFC) concentrations as described elsewhere (Carlson et al., 2010). In brief, the water mass ventilation year was estimated by comparing the partial pressure of CFC-12 in each water sample with the annual average of CFC-12 in the atmosphere as a function of time (Bullister and Weiss, 1988; Karstensen and Tomczak, 1998). The accuracy of the CFC-12 method is not only impacted by the analytical error but also by under saturation in surface waters at the time of water

masses formation and isopycnal/diapycnal mixing of more recent high-CFC with older zero-CFC water, which adds uncertainty to the ventilation times estimates. The CFC-12 method has an approximate accuracy of 1 year for the water masses with an age range between 3 to 5 years, 0.5 year in the age range 5 to 45 years and 2 years for the waters with an age between 45 and 55 years (Carlson et al., 2010).

Fitting the decline of DOC with water apparent age along isothermal surfaces provides DOC decay constants at different temperatures for the North Atlantic Ocean. Given the apparent age range of the water masses in this hydrographic section, from 3 to 55 years, the DOC<sub>SR</sub> pool is studied in this case. For the Pacific Ocean two DOC decay estimates for the recalcitrant pools are available but they were measured at approximately equal temperatures (Hansell et al., 2012). As multiple DOC decay estimates over a range of temperatures are needed, we did not include them in our analysis.

## DOC Decay Function

Organic matter consists of a wide spectrum of reactivity, each component with its own rate of degradation; therefore, continuum models, rather than those based on the separation of organic matter into distinct reactivity pools, are the most suitable to study the kinetics of organic matter degradation (e.g., Middelburg et al., 1993). However, most DOC degradation studies have been based on the first-order model developed by (Berner (1964). Following this simple model, we used the following 2-pool first-order exponential decay function to fit the DOC decay in the incubation experiments and latitudinal gradients processed in this study:

$$DOC(t) = b_1 \cdot e^{-k_C \cdot t} + b_2 \quad (1)$$

Where  $DOC(t)$  is the concentration of DOC at time  $t$ ,  $b_1$  is the degraded pool (μmol kg<sup>-1</sup>),  $k_C$  the decay constant (in year<sup>-1</sup>),  $t$  the time (in years) and  $b_2$  the resistant pool (μmol kg<sup>-1</sup>). For the latitudinal gradients, we fitted DOC and water mass apparent ages ( $t$ ) to the exponential decay function for different temperature bins (see Table S1).

## Arrhenius Law

The temperature sensitivity of  $k_C$  is defined by changes in activation energy ( $E_a$ ):

$$K_C = A \cdot e^{-E_a/R \cdot T} \quad (2)$$

Where  $A$  is the theoretical  $k_C$  in the absence of  $E_a$  (in year<sup>-1</sup>);  $E_a$  is the energy barrier to be surpassed in order for the reaction to take place (in J mol<sup>-1</sup>);  $R$  is the universal gas constant (8.314 J mol<sup>-1</sup> K<sup>-1</sup>); and  $T$  is the temperature in Kelvin (K). The factor  $e^{-E_a/R \cdot T}$  is proportional to the fraction of DOC molecules with kinetic energies in excess of  $E_a$  (Arrhenius, 1889). The Arrhenius equation suggests that more recalcitrant DOC requires a higher amount of energy to be degraded than labile compounds and therefore its degradation is slower. An estimate of the  $E_a$  for DOC degradation was derived from the slope of an Arrhenius plot where the ln of DOC decay constants (ln  $k_C$ ) is plotted against the inverse absolute temperature (1/T) (Figure 1). The Arrhenius law assumes that DOC degradation is a well-defined enzymatic reaction with a constant  $E_a$  and that temperature is the only factor affecting the rate, while factors such as nutrient limitation are ignored. As all the experimental and *in situ* studies included here had nutrients at non-limiting levels this is a valid assumption (data not shown). Furthermore, as DOC consists of myriad compounds degraded by a variety of microbial populations with different physiologies (Repeta, 2015), the temperature response measured is really the sum of all processes involved and the calculated  $E_a$  should be seen as an apparent  $E_a$  (Westrich and Berner, 1988; Middelburg et al., 1996). Also as both the estimates from the laboratory experiments and latitudinal gradients were obtained using apparent  $E_a$  and apparent Arrhenius pre-exponential factors, the result of this analysis is a likely upper limit to the rate of enzymatic degradation.

$T$ -tests were performed to test the significance of differences and Linear regression analyses were used for the Arrhenius plots applying model II regression (Sokal and Rohlf, 1995). Prior to regressions, normality was checked, and the confidence level was set at 95%.

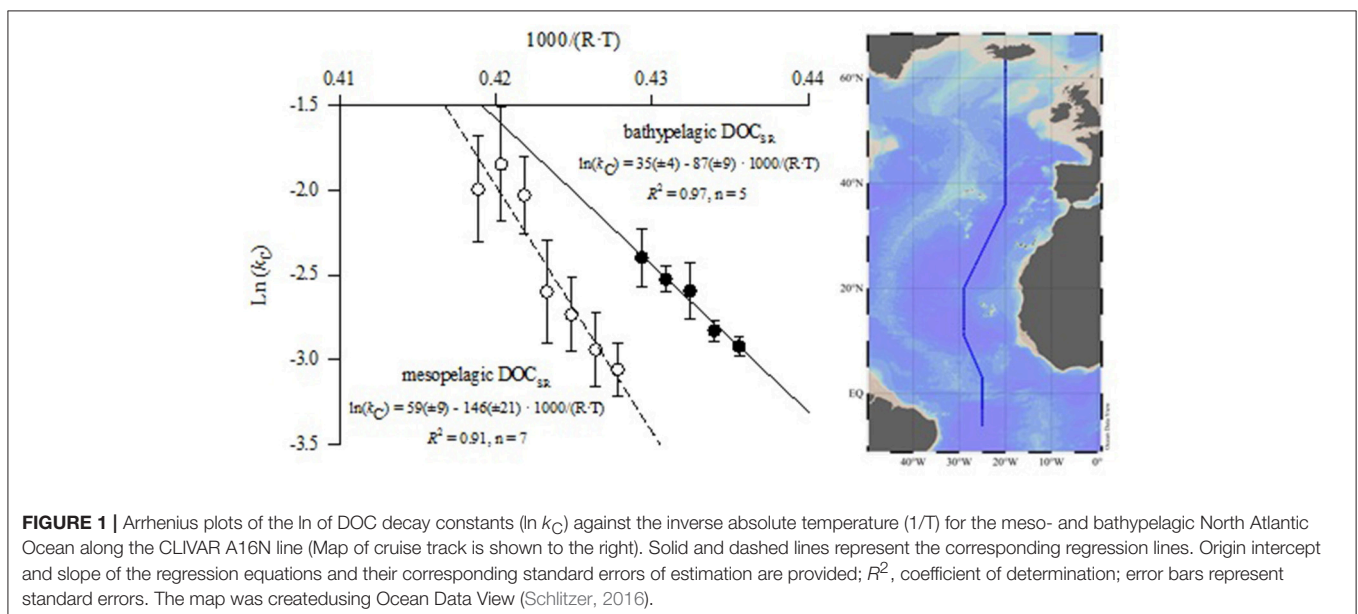
The relationship between temperature and biological rates has been modeled in various ways (Ratkowsky et al., 1983; Ahlgren, 1987), while the factor by which a decay constant increases with a 10°C increase ( $Q_{10}$ ) is mostly used.  $Q_{10}$  values have previously been calculated using multiple functions (Sierra, 2012). In this study we assumed that the relationship between rates and temperature is exponential and we used the following function (Li and Dickie, 1987):

$$Q_{10} = e^{\frac{E_a}{R} \cdot \frac{10}{T_1 \cdot T_2}} \quad (3)$$

where  $T_1$  and  $T_2$  are the temperatures at which the DOC decay constants were measured.

## Describing DOC<sub>SL</sub> and DOC<sub>SR</sub> Decay Using a Global Ocean DOC Cycling Model

Following the first-order decay model (Equation 1), a linear DOC cycling model that considers two DOC pools, semi-labile (DOC<sub>SL</sub>) and semi-refractory (DOC<sub>SR</sub>), was coupled with an ocean circulation inverse model (Primeau et al., 2013) constrained by tracer observations of temperature, salinity, sea surface height, heat, and freshwater fluxes,  $\Delta^{14}\text{C}$ -DIC, CFC-11, and phosphate that accurately represents the transport and ideal age of ocean water masses (Devries and Primeau, 2011). The horizontal resolution of the model is 2° × 2° with 24 vertical layers ranging in thickness from ~36 m near the surface to ~663 m at depth. The temperature sensitive DOC decay parameters  $A$  and  $E_a$  (Equation 2) were estimated using a Bayesian inversion procedure combined with a numerical optimization technique that finds the most probable value



for each parameter conditioned on the global DOC tracer observations from Letscher and Moore (2015). The tracer conservation equations of the two DOC pools are given by:

$$\frac{\partial \text{DOC}_{\text{SL}}}{\partial t} + \mathbf{T} \cdot \text{DOC}_{\text{SL}} = f_1 \cdot \mathbf{J}_p \cdot r_{\text{CP}} - A_1 e^{-E_{a1}/RT} \cdot \text{DOC}_{\text{SL}} \quad (4)$$

$$\frac{\partial \text{DOC}_{\text{SR}}}{\partial t} + \mathbf{T} \cdot \text{DOC}_{\text{SR}} = f_2 \cdot \mathbf{J}_p \cdot r_{\text{CP}} - A_2 e^{-E_{a2}/RT} \cdot \text{DOC}_{\text{SR}} \quad (5)$$

where  $\mathbf{T}$  is the 3D advection-diffusion operator of the annually resolved steady-state ocean circulation,  $\mathbf{J}_p$  is the biological new production rate of organic matter in the euphotic zone (upper 73 m) which has been optimized using the phosphate tracer distribution and is detailed elsewhere (Primeau et al., 2013),  $r_{\text{CP}}$  is the ratio of carbon to phosphorus in newly produced organic matter which is assigned a modified Redfield ratio of 117:1,  $R$  is the universal gas constant and  $T$ , the temperature in degrees Kelvin, is obtained from the ocean circulation inverse model output.

The 6-tunable parameter set  $[f_1 f_2 A_1 A_2 E_{a1} E_{a2}]$  contains the fraction of biological organic matter production that accumulates as  $\text{DOC}_{\text{SL}}$  ( $f_1$ ; unit less) and  $\text{DOC}_{\text{SR}}$  ( $f_2$ ), the theoretical decay constants for  $\text{DOC}_{\text{SL}}$  ( $A_1$ ;  $\text{s}^{-1}$ ) and  $\text{DOC}_{\text{SR}}$  ( $A_2$ ;  $\text{s}^{-1}$ ) in the absence of an energy barrier, and the apparent activation energies ( $E_{a1}$ ;  $\text{kJ mol}^{-1}$ ) and ( $E_{a2}$ ;  $\text{kJ mol}^{-1}$ ) for  $\text{DOC}_{\text{SL}}$  and  $\text{DOC}_{\text{SR}}$ , respectively. The most probable values of the tuneable parameter set are determined by minimizing an objective function using the Nelder-Mead method (Lagarias et al., 1998), which considers the squared deviations between model-predicted and observed DOC pairs from the global dataset, with additional priors related to the acceptable ranges of each parameter, e.g.,  $0 \leq f \leq 0.5$ ;  $0 \leq E_a \leq 150 \text{ kJ mol}^{-1}$ , etc. An additional constraint, that the global inventory of  $\text{DOC}_{\text{SR}} > \text{DOC}_{\text{SL}}$  (Hansell, 2013), was needed to accurately simulate the known distributions of each DOC pool.

The linear Equations (4, 5) are solved by direct matrix inversion with the resulting 3D distributions of  $\text{DOC}_{\text{SL}} + \text{DOC}_{\text{SR}}$  added to the concentration of recalcitrant DOC ( $\text{DOC}_{\text{R}}$ ), which was not simulated, and varied with ocean basin. Concentrations of  $\text{DOC}_{\text{R}}$  were estimated from the mean concentrations observed below 1500 m as  $43.3 \mu\text{mol kg}^{-1}$  (North Atlantic;  $65^\circ\text{N}-0^\circ$ ),  $40.5 \mu\text{mol kg}^{-1}$  (South Atlantic;  $40^\circ\text{S}-0^\circ$ ),  $38.6 \mu\text{mol kg}^{-1}$  (Pacific),  $41.4 \mu\text{mol kg}^{-1}$  (Indian),  $41.7 \mu\text{mol kg}^{-1}$  (Arctic;  $> 65^\circ\text{N}$ ), and  $40.9 \mu\text{mol kg}^{-1}$  (Southern;  $> 40^\circ\text{S}$ ).

We tested the sensitivity of the inferred DOC decay temperature parameters to alternate assumptions with respect to marine DOC production and consumption processes (Table 2, Figures 2, 3). Model I considers both  $\text{DOC}_{\text{SL}}$  and  $\text{DOC}_{\text{SR}}$  are produced biologically in the euphotic zone as separate fractions with a modified Redfield  $r_{\text{CP}}$  117:1, and is represented by Equations (4, 5). Model II is based on the size-reactivity continuum concept of DOC cycling with only  $\text{DOC}_{\text{SL}}$  being produced biologically in the euphotic zone (with  $r_{\text{CP}} = 117:1$ ) and  $\text{DOC}_{\text{SR}}$  being produced as a fraction of  $\text{DOC}_{\text{SL}}$  decay by heterotrophic consumption. The Model II tracer conservation equation for  $\text{DOC}_{\text{SR}}$  is:

$$\frac{\partial \text{DOC}_{\text{SR}}}{\partial t} + \mathbf{T} \cdot \text{DOC}_{\text{SR}} = f_2 \cdot (A_1 e^{-E_{a1}/RT} \cdot \text{DOC}_{\text{SL}} - A_2 e^{-E_{a2}/RT} \cdot \text{DOC}_{\text{SR}}) \quad (6)$$

With  $f_2$  being the fraction of  $\text{DOC}_{\text{SL}}$  decay that produces  $\text{DOC}_{\text{SR}}$ . The Model II tracer conservation equation for  $\text{DOC}_{\text{SL}}$  is equal to Equation (4). Sensitivity of the model results to variability in the  $r_{\text{CP}}$  of DOM production was also tested, using the spatially varying  $r_{\text{CP}}$  of upper ocean organic matter export inferred for twelve ocean biomes in Teng et al. (2014). The twelve ocean biomes are nominally separated by the  $0.3 \mu\text{M} [\text{PO}_4]$  iso-concentration at the surface from World Ocean Atlas 2009. The

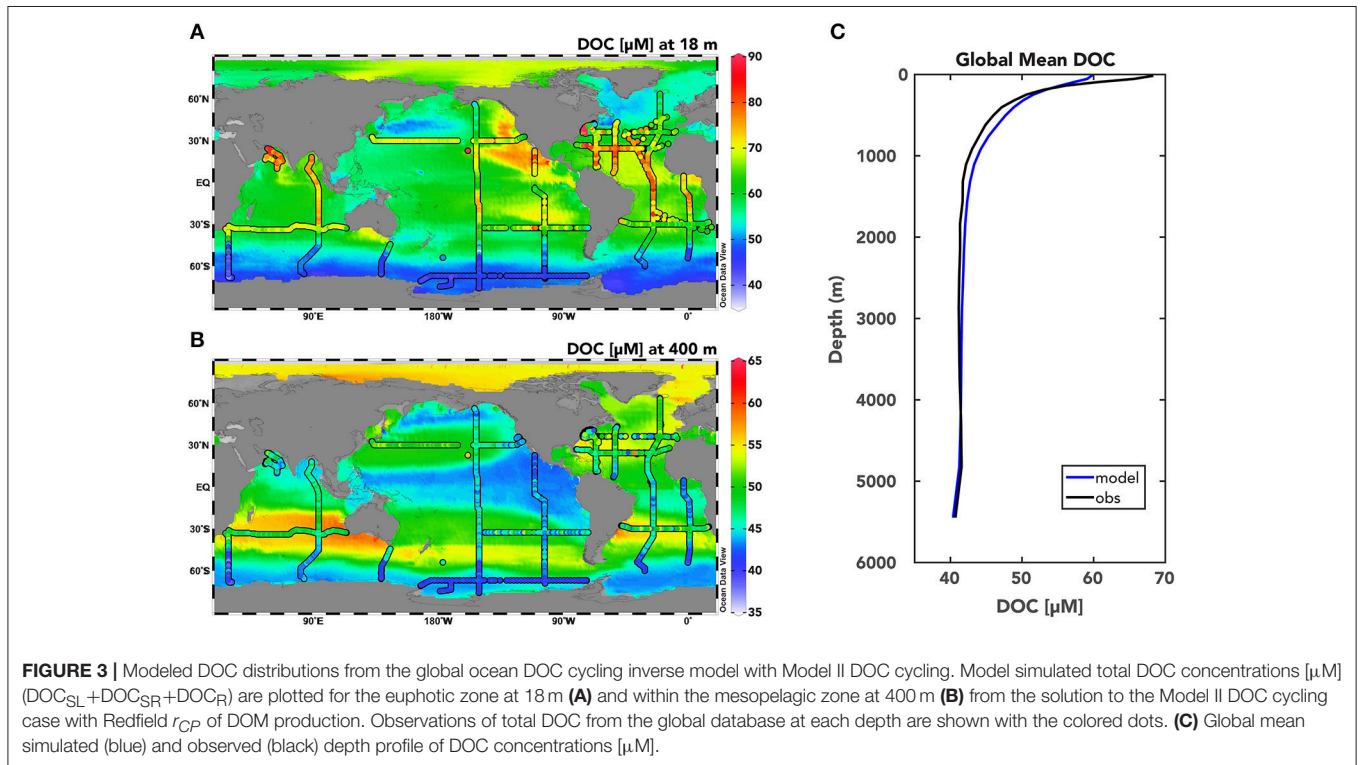
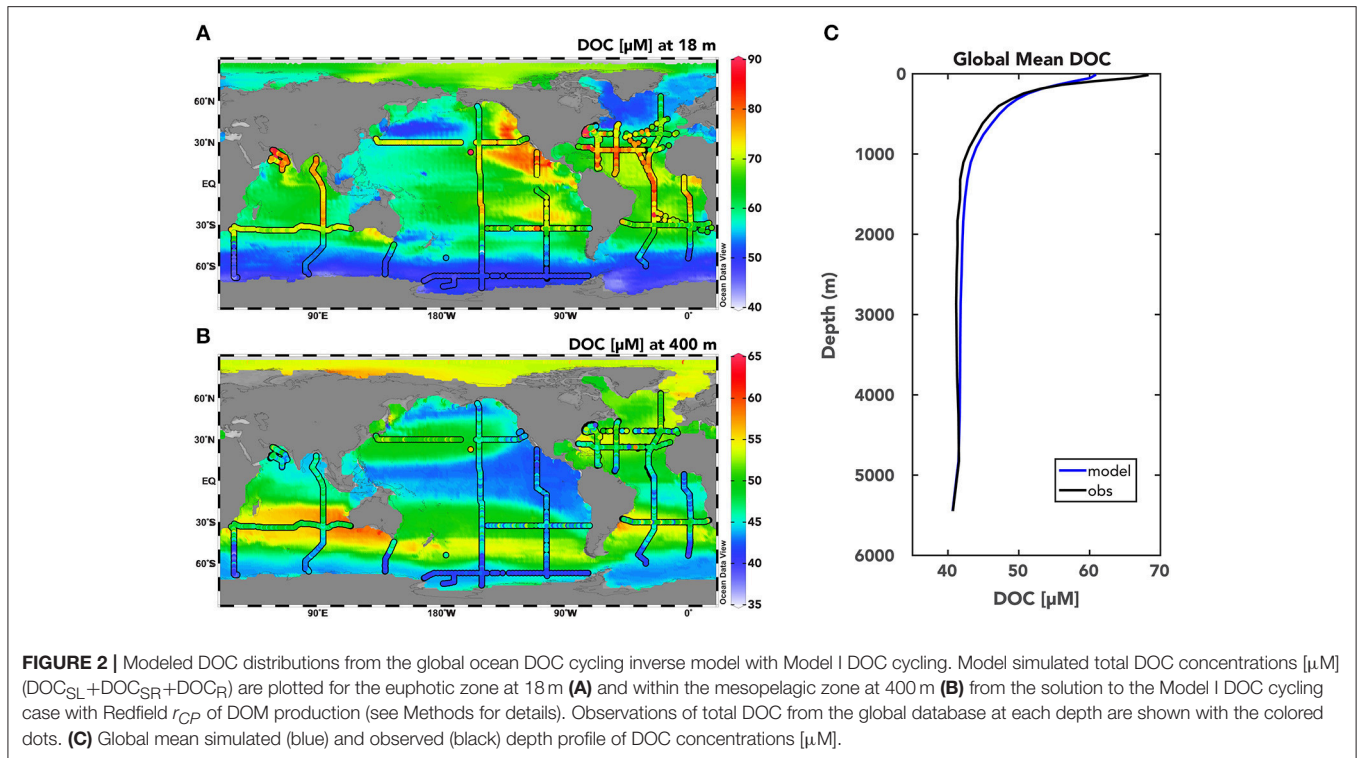
**TABLE 2** | Model-obtained values for the temperature sensitivity of DOC degradation.

	DOC pool	$Q_{10}$	$E_a$ ( $\text{kJ mol}^{-1}$ )	Ref. temp ( $^\circ\text{C}$ )	$k_C$ ( $10^{-3} \text{ day}^{-1}$ )	Present reservoir (Pg)	Reservoir + $1^\circ\text{C}$ (Pg)	Diff. (%)
<b>REDFIELD C:P</b>								
Model I*	$\text{DOC}_{\text{SL}}$	2.2	$53.1 \pm 1.3$	15	$10.7 \pm 0.5$	19.0	17.4	8
	$\text{DOC}_{\text{SR}}$	4.4	$101.3 \pm 31.5$	15	$0.06 \pm 3\text{e-}6$	38.0	31.1	18
Model II**	$\text{DOC}_{\text{SL}}$	2.3	$55.8 \pm 2.3$	15	$3.2 \pm 0.07$	18.7	17.1	9
	$\text{DOC}_{\text{SR}}$	4.4	$100.3 \pm 25.2$	15	$0.05 \pm 1\text{e-}6$	37.5	30.9	18
<b>VARIABLE C:P</b>								
Model I*	$\text{DOC}_{\text{SL}}$	2.3	$55.0 \pm 2.0$	15	$3.9 \pm 0.1$	14.1	12.9	9
	$\text{DOC}_{\text{SR}}$	4.5	$101.6 \pm 33.6$	15	$0.03 \pm 5\text{e-}7$	28.3	23.3	17
Model II**	$\text{DOC}_{\text{SL}}$	2.3	$57.3 \pm 3.2$	15	$0.6 \pm 0.004$	8.7	8.0	8
	$\text{DOC}_{\text{SR}}$	3.4	$82.8 \pm 15.6$	15	$0.07 \pm 8\text{e-}6$	34.8	29.6	15

The temperature coefficient,  $Q_{10}$ , apparent activation energy,  $E_a$ , reference temperature used for calculation, and the average DOC decay constants, standardized to a reference temperature,  $k_C$  are shown for all tested cases of DOC cycling pathways and production stoichiometry. Present DOC reservoir estimates are shown together with the difference in pool size from an equilibrium  $1^\circ\text{C}$  increase in temperature. Average values with  $\pm$  standard errors are shown.

\*Model I: In this case both  $\text{DOC}_{\text{SL}}$  and  $\text{DOC}_{\text{SR}}$  are produced as constant fractions of the new production field. Thus, all  $\text{DOC}_{\text{SR}}$  is produced in the euphotic zone and exported to the deep ocean with the overturning circulation.

\*\*Model II: In this case, the model assumes that  $\text{DOC}_{\text{SR}}$  is produced as some fraction of  $\text{DOC}_{\text{SL}}$  degradation. Thus,  $\text{DOC}_{\text{SR}}$  is produced throughout all depths of the ocean as  $\text{DOC}_{\text{SL}}$  is microbially degraded.



spatially varying  $r_{\text{CP}}$  values are reported in Table S2. All cases fit the data equally well as, and neither significantly altering the conclusions drawn from, the standard case (Model I, Redfield

$r_{\text{CP}}$ ) presented above, i.e.,  $A_2 \gg A_1$ ;  $E_{a2} > E_{a1}$ . The reported model parameter standard error bars were assigned by expansion of the log posterior probability function in a Taylor series about

the most probable values and truncating at second order, i.e., Laplace's approximation.

For all approaches used it should be remembered that we are assuming that geochemical DOC scavenging processes, such as interaction with suspended particles (Druffel et al., 1992), are negligible. As these processes are purely physical/chemical they should have low temperature sensitivity ( $Q_{10}$  around 1; Oelkers, 1991) and the importance of these processes will therefore not be as severely impacted by changing ocean temperature as the biological processes. Furthermore, the major pool of  $\text{DOC}_R$ , which amounts to  $630 \pm 32$  Pg C and has a half-life of thousands of years (Hansell, 2013), has been excluded from our analysis because the nature of its removal, biotic or geochemical, is still unclear.

## RESULTS

Apparent  $E_a$  values, temperature coefficients ( $Q_{10}$ ), and average decay constants ( $k_C$ ) of DOC degradation estimated from microbial incubation experiments conducted with phytoplankton-derived DOC, marine humic substances, and natural DOC from a coastal embayment are summarized in **Table 1**. Applying the Arrhenius law, we estimated apparent  $E_a$  values varying from (average  $\pm$  standard error)  $35 \pm 10$  kJ mol<sup>-1</sup> for the  $\text{DOC}_L$  to  $67 \pm 11$  kJ mol<sup>-1</sup> for the  $\text{DOC}_{SL}$  pool, while the  $Q_{10}$  increased from 1.7 to 2.7 (**Table 1**). The corresponding average DOC decay constants ranged from  $110 \pm 0.1 \times 10^{-3}$  day<sup>-1</sup> (half-life,  $\ln 2/k_C$ ,  $6.3 \pm 0.6$  days) for the  $\text{DOC}_L$  to  $5 \pm 1 \times 10^{-3}$  day<sup>-1</sup> (half-life,  $152 \pm 18$  days) for the  $\text{DOC}_{SL}$  (**Table 1**).

The Arrhenius plot of the  $\text{DOC}_{SR}$  decay constants, obtained from the DOC concentrations and CFC apparent water ages collected during the CLIVAR A16N cruise, clearly show two distinct data clusters for the bathy- and mesopelagic North Atlantic Ocean (**Figure 1**) with significantly different ( $t$ -test,  $p < 0.05$ ) apparent  $E_a$  (average  $\pm$  standard error) of  $87 \pm 9$  and  $146 \pm 21$  kJ mol<sup>-1</sup>, which translate into  $Q_{10}$  values of  $3.8 \pm 0.5$  and  $8 \pm 3$ , respectively (**Table 1**). The most likely reason for this difference is that the bathypelagic North Atlantic is filled with fresher DOC during the massive water mass formation that occurs in the Arctic seas, which constitutes the engine of the meridional overturning circulation. Uncertainties in this analysis of the temperature response of  $\text{DOC}_{SR}$  primarily stem from the fact that the surface ocean waters ultimately ventilating the meso- and bathypelagic zones do not have identical initial DOC content; while both waters contain  $\text{DOC}_{SR}$  and  $\text{DOC}_R$ , they are at different concentrations and likely partially different chemical compositions. Note that for the bathypelagic ocean,  $b_1$  (degraded pool) and  $b_2$  (resistant pool) in Equation (1) remained constant at  $5.6 \pm 0.4$  and  $40.6 \pm 0.7$   $\mu\text{mol kg}^{-1}$  for all temperatures, but for the mesopelagic ocean  $b_1$  decreased and  $b_2$  increased with increasing temperature (Table S1). These changes are most likely explained by an increasing proportion of recalcitrant DOC in subtropical compared with subpolar mode waters, although the influence of decreasing age range with increasing temperature cannot be disregarded (Table S1). In any case, to avoid this effect,

we restricted our analysis of the temperature response of the mesopelagic layer to the 8 to 11°C isotherms, where  $b_1$  and  $b_2$  remained constant at  $11 \pm 2$  and  $42 \pm 2$   $\mu\text{mol kg}^{-1}$  for all temperatures. When extrapolating this analysis, it should be remembered that our dataset is restricted to one CLIVAR section (A16N). Other cruises have not been included as they contained too few data points (2–3) and it was therefore not possible to obtain  $k_C$  over a wide temperature range, which is a prerequisite for calculating reliable  $E_a$  and  $Q_{10}$  values.

The  $E_a$  values of  $\text{DOC}_{SR}$  obtained for the North Atlantic could be affected by water mass mixing because the latitudinal gradients of DOC and CFC apparent age were followed along isothermal rather than isopycnal surfaces. Note that although isopycnal surfaces are preferred to minimize the impact of mixing, it is compulsory to use isothermals when studying temperature sensitivities. Therefore, we used a global ocean DOC cycling model that accounts for ocean mixing of DOC to both constrain the validity of our approach (i.e., following gradients along isothermals) and provide estimates of the global impact. The most probable values for the apparent  $E_a$  conditioned on the global database of DOC observations were found to be  $53.1 \pm 1.3$  kJ mol<sup>-1</sup> (average  $\pm$  standard error) for  $\text{DOC}_{SL}$  and  $101.3 \pm 31.5$  kJ mol<sup>-1</sup> for  $\text{DOC}_{SR}$  (**Table 2**). Corresponding  $Q_{10}$  values using a reference temperature of 15°C yield a value of  $2.2 \pm 0.1$  for the  $\text{DOC}_{SL}$  pool and  $4.4 \pm 2.1$  for  $\text{DOC}_{SR}$ . These estimates are all within the ranges obtained from the incubation studies and the latitudinal gradients of DOC in the north Atlantic (**Tables 1, 2**). This good agreement between the model and experimental data, shows that the water mass mixing on the latitudinal DOC gradients did not significantly affect the estimate of  $\text{DOC}_{SR}$  decay constants, and secondly that our results are generally valid.

The most probable basin-specific apparent  $E_a$  values were also tested for the Atlantic and Pacific basin, yielding Atlantic values of  $55.2 \pm 2.1$  kJ mol<sup>-1</sup> ( $Q_{10} = 2.3 \pm 0.1$ ) and  $86.2 \pm 13.4$  kJ mol<sup>-1</sup> ( $Q_{10} = 3.1 \pm 0.7$ ) and Pacific values of  $54.6 \pm 1.8$  kJ mol<sup>-1</sup> ( $Q_{10} = 2.2 \pm 0.1$ ) and  $92.7 \pm 52.0$  kJ mol<sup>-1</sup> ( $Q_{10} = 3.9 \pm 3.3$ ) for the  $\text{DOC}_{SL}$  and  $\text{DOC}_{SR}$  pools, respectively (data not shown). Thus, apparent  $E_a$  ( $Q_{10}$ ) values for the  $\text{DOC}_{SL}$  pool are statistically the same between the Atlantic and Pacific basins. More data are required to reduce the uncertainties in the inferred basin specific  $E_a$  for the  $\text{DOC}_{SR}$  pool in order to address the possibility of differing temperature sensitivities between ocean basins.

Using our estimates, we calculated the sensitivity of the size of the global ocean DOC reservoir per 1°C of equilibrium ocean warming, by comparing the modern ocean reservoir, computed with the most probable parameter values to the DOC cycling model, with that assessed by uniformly increasing the ocean temperature by 1°C. The modern ocean is found to hold 57.0 Pg C of  $\text{DOC}_{SL} + \text{DOC}_{SR}$  and 630.0 Pg C of  $\text{DOC}_R$  decreasing to 48.5 Pg C of  $\text{DOC}_{SL} + \text{DOC}_{SR}$  for a 1°C warming (**Table 2**; Model I, Redfield  $r_{CP}$  case). Therefore, in terms of size changes of the DOC reservoir for a 1°C warming scenario, our DOC cycling model predicts a decrease by  $7 \pm 1$  Pg C (mean and  $SD$  of the four modeled DOC cycling sensitivity tests), i.e.,  $\sim 15\%$  of the  $\text{DOC}_{SL} + \text{DOC}_{SR}$  reservoirs (**Table 2**). In these calculations, we assume for simplicity a uniform warming throughout the

ocean, which approximates the sensitivity of the ocean DOC reservoir to an equilibrium warming of the global ocean by 1°C. We ignore regional differences in warming and that most heat is absorbed by the surface and then slowly redistributed throughout the ocean interior that defines the transient response of the ocean to Earth surface warming (Drijfhout et al., 2014). The impact of changing ocean temperatures on DOC degradation will therefore vary, and would for the deep ocean be delayed relative to both the initial warming and the subsequent uptake of CO<sub>2</sub> and heat. Given the current observed rate of deep ocean (>1,000 m) warming of ~0.01°C per decade (Purkey and Johnson, 2010), the deep ocean would warm by 1°C from present in ~1 millennium, i.e., the 7 ± 1 Pg C reduction in the global ocean DOC reservoir would occur on a similar timescale as deep ocean ventilation. The temperature coefficients, Q<sub>10</sub>, at a reference temperature of 15°C (see Equation 3 in methods) are 1.7–1.8 for DOC<sub>L</sub>, 2–3 for DOC<sub>SL</sub>, and 4–8 for DOC<sub>SR</sub>, indicating that DOC compounds with higher activation energies have higher temperature dependence.

## DISCUSSION

Considering that most of the 662 ± 32 Pg C of DOC in the oceans has half-life times of decades to thousands of years (Hansell, 2013), knowledge on the impact of rising ocean temperatures on the decay constants of the less bioavailable DOC pools is relevant for predicting how climate change will affect the global carbon cycle.

Our combined modeling and experimental approaches indicate that DOC<sub>SR</sub> has an apparent  $E_a$  of ~83–146 kJ mol<sup>-1</sup>, which is comparable to values found for soil (67–120 kJ mol<sup>-1</sup>; Leifeld and Von Lützw, 2013) and mesopelagic ocean dissolved oxygen consumption (86.5 mol<sup>-1</sup>; Brewer and Peltzer, 2016). Comparatively, DOC<sub>L</sub> and DOC<sub>SL</sub> have lower apparent  $E_a$  of ~35–40 and ~50–70 kJ mol<sup>-1</sup>, respectively, which are comparable to values estimated for the degradation of glucose (30 kJ mol<sup>-1</sup>), tannin (70 kJ mol<sup>-1</sup>), labile soil organic matter (~44 kJ mol<sup>-1</sup>), and values recently predicted for marine respiration (~56 kJ mol<sup>-1</sup>) and mesopelagic prokaryote production (~72 kJ mol<sup>-1</sup>) (Knorr et al., 2005; Davidson and Janssens, 2006; Yvon-Durocher et al., 2012; Lønborg et al., 2016). Overall this suggests that the DOC<sub>SR</sub> pool has higher activation energies and temperature dependence than the DOC<sub>L</sub> and DOC<sub>SL</sub> pools. A recent study demonstrated that in soils the Q<sub>10</sub> values scales inversely with temperature, with low temperature environments having a higher Q<sub>10</sub> than warmer ones (Koven et al., 2017). In a similar way we find that DOC<sub>SR</sub>, mostly present in the deep ocean with lower temperatures, has a higher Q<sub>10</sub> of ~3–4, while DOC<sub>SL</sub>, with higher concentrations in warmer mesopelagic waters, has a lower Q<sub>10</sub> of ~1.8–2. DOC decay in the future ocean will also vary depending on how factors other than temperature will be impacted by global change. These factors, amongst others, include changes in ocean circulation and the supply of inorganic nutrients, UV light exposure, dissolved oxygen levels, bacterial community composition, DOC chemical structure, and/or an extreme dilution of individual organic substrates (Lønborg et al.,

2009; Kattner et al., 2011; Mopper et al., 2015; Repeta, 2015). Ocean models have predicted that warming may strengthen vertical stratification, thereby suppressing upward mixing of nutrients from the deep ocean and causing longer exposure to UV-light, with possible consequences for the DOC decay. Currently we do not have sufficient data to determine how this would impact our findings of the temperature sensitivity of DOC cycling. Over the last few years it has been demonstrated that global increases in ocean temperatures will reduce the overall O<sub>2</sub> content (i.e., “ocean deoxygenation”; Keeling et al., 2010) both directly (e.g., decreased solubility) and indirectly (e.g., increased respiration and decreased ocean ventilation). Studies in marine sediments have found that the degradation of organic matter under hypoxic and anoxic conditions is slower and less efficient than under oxic conditions (Zehnder and Stumm, 1988; Fenchel and Finlay, 1995), but due to the lack of data on DOC cycling under different O<sub>2</sub> levels we are not able to conclude how this might influence the relationships found in this study. The deep ocean bacterial community composition has been shown to vary over spatial and temporal scales, which could potentially affect DOC decay rates in the deep Ocean (Delong et al., 2006). However, a major influence seems unlikely as previous studies have suggested that DOC decay is generally independent of the bacterial community composition (Sjöstedt et al., 2013). The chemical composition of DOC in the deep ocean varies spatially (e.g., Martínez-Pérez et al., 2017 and our higher diagnosed activation energy for the DOC<sub>SR</sub> than the DOC<sub>SL</sub> pool point to a compositional control on the temperature sensitivity, i.e., the molecule types that make up the DOC<sub>SR</sub> pool require more energy to be consumed than DOC<sub>SL</sub> molecules. A similar conclusion was reached by Brewer and Peltzer (2017), who showed that the activation energy of ocean oxygen consumption varies widely (60.8 to 1662.9 kJ mol<sup>-1</sup>) which they suggested was partly explained by some compounds being harder to oxidize than others.

Recent studies have also suggested that the dilute nature of individual compounds that constitute the DOC pool control its persistence in the oceans (Kattner et al., 2011; Arrieta et al., 2015). At very low concentrations, molecular diffusion could limit the utilization of DOC. Since molecular diffusion of organic matter is temperature sensitive, this could influence our findings. However, the  $E_a$  for molecular diffusion (~15 kJ mol<sup>-1</sup>; Oelkers, 1991) is low compared with the  $E_a$  obtained in this study. The approaches combined here are also biased by different limitations; as an example, the incubation studies do not include the effects of some environmental factors (e.g., sunlight, pressure) and they could also be impacted by so called “bottle effects,” which could lead to different conditions compared with those found in the natural environment. On the other hand, both the DOC latitudinal gradients and modeling approach do not explicitly measure the microbial degradation but infer it indirectly from subsurface spatial gradients in DOC and temperature. However, it can be assumed that these approaches assess the temperature effects on microbial consumption of DOC, as long as abiotic particle-DOC scavenging or solubilization processes are minimal (Druffel et al., 1992, 2016).



Ocean warming varies spatially and over time, partly linked with the steady heat accumulation in the deep ocean, resulting in ~30% of the ocean warming occurring below 1,000 m (Whitney et al., 2007; Meehl et al., 2011; Balmaseda et al., 2013). As the ocean will most likely experience temperature increases of a few (1–4°C) degrees Celsius over the next decades (Meehl et al., 2011), we have estimated that the decay rates of the DOC<sub>L</sub>, DOC<sub>SL</sub>, and DOC<sub>SR</sub> pools will increase by 4–6, 7–12, 15–18% for 1°C of warming. Note that for an increase of 2°C, which is possible for the surface ocean layer by the end of this century, the degradation rate of the DOC<sub>SR</sub> pool will increase by 32–40% compared with the current situation. According to the microbial carbon pump (MCP) hypothesis, stimulation of microbial respiration will lead to an increase in the production of DOC<sub>R</sub> relative to the production of the other DOC pools, with the magnitude and impact of this stimulation depend on the sensitivity of DOC<sub>R</sub> degradation to changing temperature (Jiao et al., 2010; Legendre et al., 2015). In terms of size changes of the DOC reservoir for a 1°C warming scenario, our DOC cycling model predicts a decrease by  $7 \pm 1$  Pg C (~15%) of the DOC<sub>SL</sub> + DOC<sub>SR</sub> reservoirs. Assuming a warming rate equal to the current sea surface increase of  $0.01^\circ\text{C year}^{-1}$  (Rhein et al., 2013), the annual loss of DOC<sub>SL</sub> + DOC<sub>SR</sub> would be  $0.07$  Pg C  $\text{year}^{-1}$ , which represents about 3.5% of the annual net ocean uptake of CO<sub>2</sub> (Sarmiento and Gruber, 2006). Therefore, the overall loss of bioavailable DOC would significantly, but not massively, counteract the ocean CO<sub>2</sub> sink.

## CONCLUSION

Our study found that, (1) the assumption of a constant  $Q_{10}$  of ~2 is not universally applicable for the microbial degradation of DOC in the open ocean, (2) the empirically estimated  $Q_{10}$  values indicate that increasing ocean temperature will enhance the microbial degradation of the more biologically resistant DOC with a high apparent  $E_a$  compared to the more labile DOC fractions with a low apparent  $E_a$ , and (3) increasing the

equilibrium ocean temperature by just 1°C will result in a net loss of ~7 Pg C of bioavailable DOC. In this analysis we have not tested the temperature sensitivity for the DOC<sub>R</sub> pool, since knowledge on the controls for removal of this pool is lacking, however abiotic processes are thought to contribute significantly (Hansell and Carlson, 2013). Further studies are therefore needed to determine the dominant loss processes for DOC<sub>R</sub> before the possible temperature sensitivity can be assessed on this largest contributor to the global ocean carbon reservoir.

## AUTHOR CONTRIBUTIONS

CL and XÁ-S conceived the study. CL, XÁ-S, and DH authors contributed equally to the data survey, analysis, and interpretation. RL performed the model simulation, developed new tracer transport diagnostics, and interpreted the model outputs. All authors wrote the paper.

## ACKNOWLEDGMENTS

This study was funded by the Australian Institute of Marine Science and a grant from the Carlsberg Foundation to CL. XÁ-S has been partially funded by the Spanish Ministry of Economy and Competitiveness grant numbers CSD2008–00077 (MALASPINA 2010), CTM2011–30010–C02–02 (HOT-MIX), and CTM2015–69392–C3–2–R (FLUXES), the latter two co-financed with FEDER funds. RL acknowledges support from the U.S. DOE Office of Science (DE-SC0016329). DH was supported by a grant from the US National Science Foundation (OCE-1436748). This work is a contribution of the SCOR Working Group 134 on the Microbial Carbon Pump in the Ocean.

## SUPPLEMENTARY MATERIAL

The Supplementary Material for this article can be found online at: <https://www.frontiersin.org/articles/10.3389/fmars.2017.00436/full#supplementary-material>

## REFERENCES

- Ahlgren, G. (1987). Temperature functions in biology and their application to algal growth constants. *Oikos* 49, 177–190. doi: 10.2307/3566025
- Amon, R. M. W., and Benner, R. (1996). Bacterial utilization of different size classes of dissolved organic matter. *Limnol. Oceanogr.* 41, 41–51. doi: 10.4319/lo.1996.41.1.0041
- Arrhenius, S. Z. (1889). Über die reaktionsgeschwindigkeit bei der inversion von rohrzucker durch säuren. *Phys. Chem.* 4, 226–248.
- Arrieta, J. M., Mayol, E., Hansman, R. L., Herndl, G. J., Dittmar, T., and Duarte, C. M. (2015). Dilution limits dissolved organic carbon utilization in the deep ocean. *Science* 348, 331–333. doi: 10.1126/science.1258955
- Balmaseda, M. A., Trenberth, K. E., and Källén, E. (2013). Distinctive climate signals in reanalysis of global ocean heat content. *Geophys. Res. Lett.* 40, 1754–1759. doi: 10.1002/grl.50382
- Bendtsen, J., Hilligsøe, K. M., Hansen, J. L. S., and Richardson, K. (2015). Analysis of remineralisation, lability, temperature sensitivity and structural composition of organic matter from the upper ocean. *Prog. Oceanogr.* 130, 125–145. doi: 10.1016/j.pocean.2014.10.009
- Benner, R., and Amon, R. M. (2015). The size-reactivity continuum of major bioelements in the ocean. *Ann. Rev. Mar. Sci.* 7, 185–205. doi: 10.1146/annurev-marine-010213-135126
- Berner, R. A. (1964). An idealized model of dissolved sulfate distribution in recent sediments *Geochim. Cosmochim. Acta* 28, 1497–1503. doi: 10.1016/0016-7037(64)90164-4
- Bianchi, T. S. (2011). The role of terrestrially derived organic carbon in the coastal ocean: a changing paradigm and the priming effect. *Proc. Natl. Acad. Sci. U.S.A.* 108, 19473–19481. doi: 10.1073/pnas.1017982108
- Brewer, P. G., and Peltzer, E. T. (2016). Ocean chemistry, ocean warming, and emerging hypoxia: Commentary. *J. Geophys. Res. Oceans* 121, 3659–3667. doi: 10.1002/2016JC011651
- Brewer, P. G., and Peltzer, E. T. (2017). Depth perception: the need to report ocean biogeochemical rates as functions of temperature, not depth. *Philos. Trans. A Math. Phys. Eng. Sci.* 375: 20160319. doi: 10.1098/rsta.2016.0319
- Bullister, J. L., and Weiss, R. F. (1988). Determination of CCl<sub>3</sub> and CCl<sub>2</sub> F<sub>2</sub> in seawater and air. *Deep Sea Res.* 35, 839–853. doi: 10.1016/0198-0149(88)90033-7
- Bussmann, I. (1999). Bacterial utilization of humic substances from the Arctic Ocean. *Aquat. Microb. Ecol.* 19, 37–45. doi: 10.3354/ame019037

- Carlson, C. A., and Hansell, D. A. (2015). "DOM sources, sinks, reactivity, and budgets," in *Biogeochemistry of Marine Dissolved Organic Matter*, eds D. A. Hansell and C. A. Carlson (San Diego, CA: Academic Press), 65–126.
- Carlson, C. A., Hansell, D. A., Nelson, N. B. A., Siegel, D., Smethie, W. M., Khattiwala, S., et al. (2010). Dissolved organic carbon export and subsequent remineralization in the mesopelagic and bathypelagic realms of the North Atlantic basin. *Deep Sea Res. II* 57, 1433–1445. doi: 10.1016/j.dsr2.2010.02.013
- Chen, B., and Laws, E. A. (2017). Is there a difference of temperature sensitivity between marine phytoplankton and heterotrophs? *Limnol. Oceanogr.* 62, 806–817. doi: 10.1002/lno.10462
- Chin, W. C., Orellana, M. V., and Verdugo, P. (1998). Spontaneous assembly of marine dissolved organic matter into polymer gels. *Nature* 391, 568–572. doi: 10.1038/35345
- Davidson, E. A., and Janssens, I. A. (2006). Temperature sensitivity of soil carbon decomposition and feedbacks to climate change. *Nature* 440, 165–173. doi: 10.1038/nature04514
- Davidson, E. A., Trumbore, S. E., and Amundson, R. (2000). Soil warming and organic carbon content. *Nature* 408, 789–790. doi: 10.1038/35048672
- Del-Giorgio, P., and Davies, J. (2003). "Patterns of dissolved organic matter lability and consumption across aquatic ecosystems," in *Aquatic Ecosystems: Interactivity of Dissolved Organic Matter*, eds S. E. G. Findlay and R. L. Sinsabaugh (San Diego, CA: Academic Press), 399–424.
- Delong, E. F., Preston, C. M., Mincer, T., Rich, V., Hallam, S. J., Frigaard, N. U., et al. (2006). Community genomics among stratified microbial assemblages in the ocean's interior. *Science* 311, 496–503. doi: 10.1126/science.1120250
- Devries, T., and Primeau, F. (2011). Dynamically and observationally constrained estimates of water-mass distributions and ages in the global ocean. *J. Phys. Oceanogr.* 41, 2381–2401. doi: 10.1175/JPO-D-10-05011.1
- Drijfhout, S. S., Blaker, A. T., Josey, S. A., Nurser, A. J. G., Sinha, B., and Balmaseda, M. A. (2014). Surface warming hiatus caused by increased heat uptake across multiple ocean basins. *Geophys. Res. Lett.* 41, 7868–7874. doi: 10.1002/2014GL061456
- Druffel, E. R. M., Griffin, S., Coppola, A. I., and Walker, B. D. (2016). Radiocarbon in dissolved organic carbon of the Atlantic Ocean. *Geophys. Res. Lett.* 43, 5279–5286 doi: 10.1002/2016GL068746
- Druffel, E. R. M., Williams, P. M., Bauer, J. E., and Ertel, J. (1992). Cycling of dissolved and particulate organic matter in the open ocean. *J. Geophys. Res.* 97, 15539–15569. doi: 10.1029/92JC01511
- Fenchel, T., and Finlay, B. J. (1995). *Ecology and Evolution in Anoxic Worlds*. Oxford: Oxford University Press.
- Hansell, D. A. (2013). Recalcitrant dissolved organic carbon fractions. *Ann. Rev. Mar. Sci.* 5, 421–445. doi: 10.1146/annurev-marine-120710-100757
- Hansell, D. A., and Carlson, C. A. (2013). Localized refractory dissolved organic carbon sinks in the deep ocean. *Global Biogeochem. Cycles* 27, 705–710. doi: 10.1002/gbc.20067
- Hansell, D. A., Carlson, C. A., and Schlitzer, R. (2012). Net removal of major marine dissolved organic carbon fractions in the subsurface ocean. *Global Biogeochem. Cycles* 26:GB1016. doi: 10.1029/2011GB004069
- Hansen, J. L. S., and Bendtsen, J. (2014). Seasonal bottom water respiration in the North Sea–Baltic Sea transition zone: rates, temperature sensitivity and sources of organic material. *Marine Ecol. Prog. Ser.* 499, 19–34. doi: 10.3354/meps10633
- Hedges, J. I. (2002). "Why Dissolved organics matter" in *Biogeochemistry of Marine Dissolved Organic Matter*, eds D. A. Hansell and C. A. Carlson (New York, NY: Academic Press), 1–33.
- Jiao, N., Herndl, G. J., Hansell, D. A., Benner, R., Kattner, G., Wilhelm, S. W., et al. (2010). Microbial production of recalcitrant dissolved organic matter: long-term carbon storage in the global ocean. *Nat. Rev. Microbiol.* 8, 593–599. doi: 10.1038/nrmicro2386
- Karstensen, J., and Tomczak, M. (1998). Age determination of mixed water masses using CFC and oxygen data. *J. Geophys. Res. Oceans* 103, 18599–18609. doi: 10.1029/98JC00889
- Kattner, G., Simon, M., and Koch, B. P. (2011). "Molecular characterization of dissolved organic matter and constraints for prokaryotic utilization," in *Microbial Carbon Pump in the Ocean*, eds N. Jiao, F. Azam, and S. Sansers (Washington, DC: Science/AAAS), 60–61.
- Keeling, R. E., Kortzinger, A., and Gruber, N. (2010). Ocean deoxygenation in a warming world. *Ann. Rev. Mar. Sci.* 2, 199–229. doi: 10.1146/annurev.marine.010908.163855
- Keil, R. G., and Mayer, L. M. (2014). "Mineral Matrices and Organic Matter," in *Treatise on Geochemistry, 2nd Edn.*, eds H. Holland and K. Turekian (Amsterdam: Elsevier Science), 337–359.
- Kerner, M., Hohenberg, H., Ertl, S., Reckermann, M., and Spitz, A. (2003). Self-organization of dissolved organic matter micelle-like microparticles in river water. *Nature* 422, 150–154. doi: 10.1038/nature01469
- Knorr, W., Prentice, I. C., House, J. I., and Holland, E. A. (2005). Long-term sensitivity of soil carbon turnover to warming. *Nature* 433, 298–301. doi: 10.1038/nature03226
- Koven, C. D., Hugelius, G., Lawrence, D. M., and Wieder, W. R. (2017). Higher climatological temperature sensitivity of soil carbon in cold than warm climates. *Nat. Clim. Change* 7, 817–822. doi: 10.1038/nclimate3421
- Lagarias, J. C., Reeds, J. A., Wright, M. H., and Wright, P. E. (1998). Convergence properties of the nelder-mead simplex method in low dimensions. *SIAM J. Optimiz.* 9, 112–147. doi: 10.1137/S1052623496303470
- Legendre, L., Rivkin, R. B., Weinbauer, M. G., Guidi, L., and Uitz, J. (2015). The microbial carbon pump concept: potential biogeochemical significance in the globally changing ocean. *Prog. Oceanogr.* 134, 432–450. doi: 10.1016/j.pocan.2015.01.008
- Leifeld, J., and Von Lützw, M. (2013). Chemical and microbial activation energies of soil organic matter decomposition. *Biol. Fertil. Soils* 50, 147–153. doi: 10.1007/s00374-013-0822-6
- Letscher, R. T., Knapp, A. N., James, A. K., Carlson, C. A., Santoro, A. E., and Hansell, D. A. (2015). Microbial community composition and nitrogen availability influence DOC remineralization in the South Pacific Gyre. *Mar. Chem.* 177, 325–334. doi: 10.1016/j.marchem.2015.06.024
- Letscher, R. T., and Moore, J. K. (2015). Preferential remineralization of dissolved organic phosphorus and non-Redfield DOM dynamics in the global ocean: impacts on marine productivity, nitrogen fixation, and carbon export. *Global Biogeochem. Cycles* 29, 325–340. doi: 10.1002/2014GB004904
- Li, W. K. W., and Dickie, P. M. (1987). Temperature characteristics of photosynthetic and heterotrophic activities: seasonal variations in temperate microbial plankton. *Appl. Environ. Microbiol.* 53, 2282–2295.
- Lønborg, C., Cuevas, L. A., Reinthaler, T., Herndl, G. J., Gasol, J. M., Morán, X., et al. (2016). Depth dependent relationships between temperature and ocean heterotrophic prokaryotic production. *Front. Mar. Sci.* 3:90. doi: 10.3389/fmars.2016.00090
- Lønborg, C., Davidson, K., Álvarez-Salgado, X. A., and Miller, A. E. J. (2009). Bioavailability and bacterial degradation rates of dissolved organic matter in a temperate coastal area during an annual cycle. *Mar. Chem.* 113, 219–226. doi: 10.1016/j.marchem.2009.02.003
- Lønborg, C., and Álvarez-Salgado, X. A. (2012). Recycling versus export of bioavailable dissolved organic matter in the coastal ocean and efficiency of the continental shelf pump. *Global Biogeochem. Cycles* 26:GB3018. doi: 10.1029/2012GB004353
- Martínez-Pérez, A. M., Osterholz, H., Nieto-Cid, M., Álvarez, M., Dittmar, T., and Álvarez-Salgado, X. A. (2017). Molecular composition of dissolved organic matter in the Mediterranean Sea. *Limnol. Oceanogr.* 62, 2699–2712. doi: 10.1002/lno.10600
- Meehl, G. A., Arblaster, J. M., Fasullo, J. T., Hu, A., and Trenberth, K. E. (2011). Model-based evidence of deep-ocean heat uptake during surface-temperature hiatus periods. *Nat. Clim. Change* 1, 360–364. doi: 10.1038/nclimate1229
- Middelburg, J. J., Klaver, G., Nieuwenhuize, J., Wielemaker, A., De Haas, W., Van Der Nat, J. F. W. A. et al. (1996). Organic matter mineralization in intertidal sediments along an estuarine gradient. *Mar. Ecol. Prog. Ser.* 132, 157–168. doi: 10.3354/meps132157
- Middelburg, J. J., Vlug, T., and Nat, J. F. W. (1993). Organic matter mineralization in marine systems. *Global Planet. Change* 8, 47–58. doi: 10.1016/0921-8181(93)90062-S
- Mopper, K., Kieber, D. J., and Stubbins, A. (2015). "Marine photochemistry of organic matter," in *Biogeochemistry of Marine Dissolved Organic Matter*, eds D. A. Hansell and C. A. Carlson (San Diego, CA: Academic Press), 389–450.
- Moran, M. A., and Zepp, R. G. (1997). Role of photoreactions in the formation of biologically labile compounds from dissolved organic matter. *Limnol. Oceanogr.* 42, 1307–1316. doi: 10.4319/lno.1997.42.6.1307
- Oelkers, E. H. (1991). Calculation of diffusion coefficients for aqueous organic species at temperatures from 0 to 350°C. *Geochim. Cosmochim. Acta* 55, 3515–3529. doi: 10.1016/0016-7037(91)90052-7

- Peltier, W. R., Liu, Y., and Crowley, J. W. (2007). Snowball Earth prevention by dissolved organic carbon remineralization. *Nature* 450, 813–818. doi: 10.1038/nature06354
- Primeau, F. W., Holzer, M., and Devries, T. (2013). Southern Ocean nutrient trapping and the efficiency of the biological pump. *J. Geophys. Res. Oceans* 118, 2547–2564. doi: 10.1002/jgrc.20181
- Purkey, S. G., and Johnson, G. C. (2010). Warming of global abyssal and deep Southern Ocean waters between the 1990s and 2000s: contributions to global heat and sea level rise budgets. *J. Clim.* 23, 6336–6351. doi: 10.1175/2010JCLI3682.1
- Ratkowsky, D. A., Lowry, R. K., Mcmeekin, T. A., Stokes, A. N., and Chandler, R. E. (1983). Model for bacterial culture growth rate throughout the entire biokinetic temperature range. *J. Bacteriol.* 154, 1222–1226.
- Raymond, P. A., and Spencer, R. G. M. (2015). “Riverine DOM,” in *Biogeochemistry of Marine Dissolved Organic Matter*, eds D. A. Hansell and C. A. Carlson (Amsterdam: Elsevier), 509–533.
- Repeta, D. J. (2015). “Chemical characterization and cycling of dissolved organic matter,” in *Biogeochemistry of Marine Dissolved Organic Matter*, eds D. A. Hansell and C. A. Carlson (San Diego, CA: Academic Press), 21–63.
- Rhein, M., Rintoul, S. R., Aoki, S., Campos, E., Chambers, D., Feely, R. A., et al. (2013). “Observations: Ocean,” in *Climate Change 2013: The Physical Science Basis. Contribution of Working Group I to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change*, eds T. F. Stocker, D. Qin, G.-K. Plattner, M. Tignor, S. K. Allen, J. Boschung, et al. (Cambridge; New York, NY: Cambridge University Press), 255–315.
- Robinson, C., and Williams, P. J. L. (2005). “Respiration and its measurement in surface marine waters,” in *Respiration in Aquatic Ecosystems*, eds P. A. Del Giorgio and P. J. L. Williams (Oxford: Oxford University Press), 147–180.
- Rontani, J.-F., Charrière, B., Sempéré, R., Doxaran, D., Vaultier, F., Vonk, J. E., et al. (2014). Degradation of sterols and terrigenous organic matter in waters of the Mackenzie Shelf, Canadian Arctic. *Organ. Geochem.* 75, 61–73. doi: 10.1016/j.orggeochem.2014.06.002
- Sarmiento, J. L., and Gruber, N. (2006). *Ocean Biogeochemical Dynamics*. Princeton, NJ: Princeton University Press.
- Schlitzer, R. (2016). *Ocean Data View*. Available online at: <https://odv.awi.de>
- Seiki, T., Date, E., and Izawa, H. (1991). Decomposition characteristics of particulate organic matter in Hiroshima Bay. *J. Oceanogr.* 47, 207–220. doi: 10.1007/BF02310036
- Sexton, P. F., Norris, R. D., Wilson, P. A., Pälike, H., Westerhold, T., Röhl, U. (2011). Eocene global warming events driven by ventilation of oceanic dissolved organic carbon. *Nature* 471, 349–353 doi: 10.1038/nature09826
- Sierra, C. A. (2012). Temperature sensitivity of organic matter decomposition in the Arrhenius equation: some theoretical considerations. *Biogeochemistry* 108, 1–15. doi: 10.1007/s10533-011-9596-9
- Sjöstedt, J., Pontarp, M., Tinta, T., Alfredsson, H., Turk, V., Lundberg, P., et al. (2013). Reduced diversity and changed bacterioplankton community composition do not affect utilization of dissolved organic matter in the Adriatic Sea. *Aquat. Microb. Ecol.* 71, 15–24. doi: 10.3354/ame01660
- Sokal, F. F., and Rohlf, F. J. (1995). *Biometry*. New York, NY: Freeman.
- Swanson-Hysell, N. L., Rose, C. V., Calmet, C. C., Halverson, G. P., Hurtgen, M. T., Maloof, A., et al. (2010). Cryogenian glaciation and the onset of carbon-isotope decoupling. *Science* 328, 608–611. doi: 10.1126/science.1184508
- Teng, Y.-C., Primeau, F. W., Moore, J. K., Lomas, M. W., and Martiny, A. C. (2014). Global-scale variations of the ratios of carbon to phosphorus in exported marine organic matter. *Nat. Geosci.* 7, 895–898. doi: 10.1038/ngeo2303
- Thingstad, T. F., Havskum, H., Kaas, H., Nielsen, T. G., Riemann, B., Lefevre, D., et al. (1999). Bacteria-protist interactions and organic matter degradation under P-limited conditions: analysis of an enclosure experiment using a simple model *Limnol. Oceanogr.* 44, 62–79. doi: 10.4319/lo.1999.44.1.0062
- Westrich, J. T., and Berner, R. A. (1988). The effect of temperature on rates of sulfate reduction in marine sediments. *Geomicrobiol. J.* 6, 99–117. doi: 10.1080/01490458809377828
- Whitney, F. A., Freeland, H. J., and Robert, M. (2007). Persistently declining oxygen levels in the interior waters of the eastern subarctic Pacific. *Prog. Oceanogr.* 75, 179–199. doi: 10.1016/j.pocean.2007.08.007
- Yvon-Durocher, G., Caffrey, J. M., Cescatti, A., Dossena, M., Del Giorgio, P., Gasol, J. M., et al. (2012). Reconciling the temperature dependence of respiration across timescales and ecosystem types. *Nature* 487, 472–476. doi: 10.1038/nature11205
- Zehnder, A. J. B., and Stumm, W. (1988). “Geochemistry and biogeochemistry of anaerobic habitat,” in *Biology of Anaerobic Microorganisms*, ed A. J. B. Zehnder (New York, NY: John Wiley and Sons), 1–38.

**Conflict of Interest Statement:** The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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