



Peroxide scavenging potential of ultraviolet-B-absorbing mycosporine-like amino acids isolated from a marine red alga *Bryocladia* sp.

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Ultraviolet-B (UV-B; 280–315 nm)-absorbing mycosporine-like amino acids (MAAs) were extracted and purified from a marine red alga *Bryocladia* sp. by using high performance liquid chromatography. We have detected four MAAs having retention times 3.23, 2.94, 3.56, and 2.67 min with absorbance maxima (λ_{max}) at 323, 328, 335, and 340 nm, respectively. The effect of UV-B on the induction of these MAAs was studied. In comparison to control, there was 3–22% induction of MAAs after 12 and 24 h of UV-B exposure. Apart from MAAs, other pigments such as chl *a*, carotenoids and total proteins were inversely affected by UV-B irradiation. In addition, peroxide scavenging potential of these MAAs were also investigated. With 2 mM hydrogen peroxide (H_2O_2) concentration, only <5% of MAAs were found to be affected. However, with the increased H_2O_2 , 40–60% decline in the MAAs concentration with a corresponding peak shifting toward the blue wavelength was recorded. In addition, most of the MAAs were found to be reacting slowly with increasing H_2O_2 (upto 10 mM) concentration after an incubation period of 5 and 30 min, which indicates the remarkable scavenging potential and stability of MAAs against oxidative stress. Thus, the isolated MAAs from marine red alga *Bryocladia* sp. may act as an efficient peroxide scavenger.

Keywords: *Bryocladia* sp., mycosporine-like amino acids, hydrogen peroxide, UV-B irradiation, pigmentation

INTRODUCTION

Earth's atmosphere is surrounded by ozone layer in the stratosphere, which absorbs harmful ultraviolet radiation to protect the living beings (Lee and Shiu, 2009). Anthropogenically and naturally released atmospheric pollutant such as chlorofluorocarbons (CFCs), chlorocarbons (CCs), organobromides (OBs), and nitrogen containing reactive species (NO, ONOO, N_2O) are potentially responsible for continuous declining of ozone layer since 1970 (Crutzen, 1992; Russell et al., 1996; Ravishankara et al., 2009). The UV-B radiation (280–315 nm) on the Earth's surface has increased due to depletion of the stratospheric ozone layer (Hoffman and Deshler, 1991; Madronich et al., 1998; Sahoo et al., 2005). Ultraviolet radiation is highly energetic short wavelength radiation that can penetrate up to 70 meter in sea water column (Smith et al., 1992). Ambient UV-B radiation is more harmful at low altitude sea level due to high solar angle and reduction in the thickness of ozone layer which penetrates deeper water column of sea (Lee and Shiu, 2009). These assumptions suggest that tropical regions of the world face higher impact of UV-B radiation as compared to the polar and temperate region (Häder, 1993). UV-B radiation has been reported to have the negative impact on the life forms of terrestrial (Ballaré et al., 2011) as well as aquatic ecosystems including photosynthetic cyanobacteria, algae and phytoplankton (Häder et al., 2011).

UV-B radiation may severely affect the physiological and biochemical processes such as specific growth rate, photosynthesis,

CO_2 uptake, inorganic, and organic nutrient uptake, DNA damage and destruction of proteins of several organisms thereby reducing the productivity of ecosystems (Rastogi et al., 2011; Pessoa, 2012; Richa et al., 2013). In addition, extensive exposure may have drastic effect on the photosynthetic electron transport system, resulting in excessive ROS production leading to oxidative stress and consistent damage of cells (Langebartels et al., 2000). UV-B-induced oxidative stress has been observed in micro as well as macroalgae (He and Häder, 2002; Rijstenbil, 2002; Shiu and Lee, 2005). However, certain organisms have developed defense mechanisms such as induction of photoprotective compounds, scavenging of free radicals, photorepair, and programmed cell death that counteract the damaging effects of UV radiation (Janknegt et al., 2008; Lee and Shiu, 2009; Rastogi and Incharoensakdi, 2013). Interestingly, most of the algae have evolved to synthesize mycosporine-like amino acids (MAAs) that efficiently absorb UV-B radiation and scavenges the reactive oxygen species (ROS) to protect the cells (Riegger and Robinson, 1997; Dunlap and Shick, 1998; Conde et al., 2000).

MAAs are colorless, water soluble, and small (<400 Da) nitrogenous substances structurally characterized by an aminocyclohexenone or aminocyclohexenimine ring chromophore conjugated with the nitrogen substituent of an amino acid or imino alcohol group, having absorption maxima between 310 and 362 nm (Nakamura et al., 1982; Sinha et al., 2007). MAAs have very high molar extinction coefficient ($\epsilon = 28$,

100–50,000 $M^{-1}cm^{-1}$) with photosensitized chromophore that indicates additional photo stability in both terrestrial as well as aquatic habitats (Shick and Dunlap, 2002).

UV-B radiation has been shown to have remarkable inhibitory effects on photosynthetic electron transport and photochemical reaction thereby (Foyer et al., 1994; Shimizu et al., 2010) inducing the ROS formation (Langebartels et al., 2000). Enhanced ROS production causes rapid disruption of macromolecules (Rijstenbil, 2002). The cyanobacteria have developed enzymatic as well as non-enzymatic defense mechanisms in order to avoid excessive ROS production (Wada et al., 2013). Hydrogen peroxide is most efficient ROS inevitably produced as by-products in photochemical reactions. The detoxification of hydrogen peroxide has been attributed by catalase, peroxidase and ascorbic peroxidase (Karyotou and Donaldson, 2005). In the present investigation an attempt has been made to elucidate the stability and scavenging potential of MAAs against peroxide.

MATERIALS AND METHODS

GEOGRAPHICAL REGION AND COLLECTION OF ALGAL MATERIAL

The test organism *Bryocladia* sp. (Sahoo et al., 2001) belonging to the family rhodophyceae was collected from the rock surface near Bheemli beach (17°46'49.13"N, 83°23'10.98"E), Vishakhapatnam, Andhra Pradesh, India (Figure 1A). The morphology of alga was observed by light microscopy (Dewinter Light Microscope, New Delhi, India). The alga was growing in close association with the green alga *Ulva* sp. (Figure 1B). *Bryocladia* sp. is dark red-brown in color, prostrate, polysiphonous

branchlets spirally arranged with 2–5 cm long plant height (Figures 1C,D).

SOURCE AND MODE OF UV-B IRRADIATION

Algal sample was thoroughly washed with Milli Q water, transferred into sterile transparent Petri dishes (75 × 75 mm) filled with artificial sea water and placed on a rotary shaker for uniform exposure of $1 Wm^{-2}$ UV-B radiation reaching to the culture (Philips Ultraviolet-B TL 40 W:12, Holland). Temperature was maintained at $20 \pm 5^\circ C$ to avoid heat shock effects. A 295 nm cut-off filter foil (Ultraplan, UV Opak Digepra, Munich, Germany) was used to get the required regime of UV-B radiation. Aliquots (1.0 g) were withdrawn after 12, 24, and 48 h of UV-B irradiation. All the experiments were done twice (in triplicates) with consistently the same results.

EXTRACTION AND PURIFICATION OF MYCOSPORINE-LIKE AMINO ACIDS (MAAs)

Extraction and purification of MAAs were done following the method of Sinha et al. (1999). Briefly, the samples were homogenized in 100% methanol (HPLC grade, Spectrochem, Mumbai) with the help of Mortar and Pestle and incubated for 24 h at $4^\circ C$. Thereafter, aliquots were centrifuged at 5000 g for 5 min. The supernatant was evaporated to dryness, redissolved in 1 ml Milli Q water and centrifuged (10000 g for 10 min) to remove the pigment contamination. The supernatant was filtered through $0.2 \mu m$ (Axiva Biotech., New Delhi) size membrane filter. Further purification and separation of MAAs was done by HPLC.

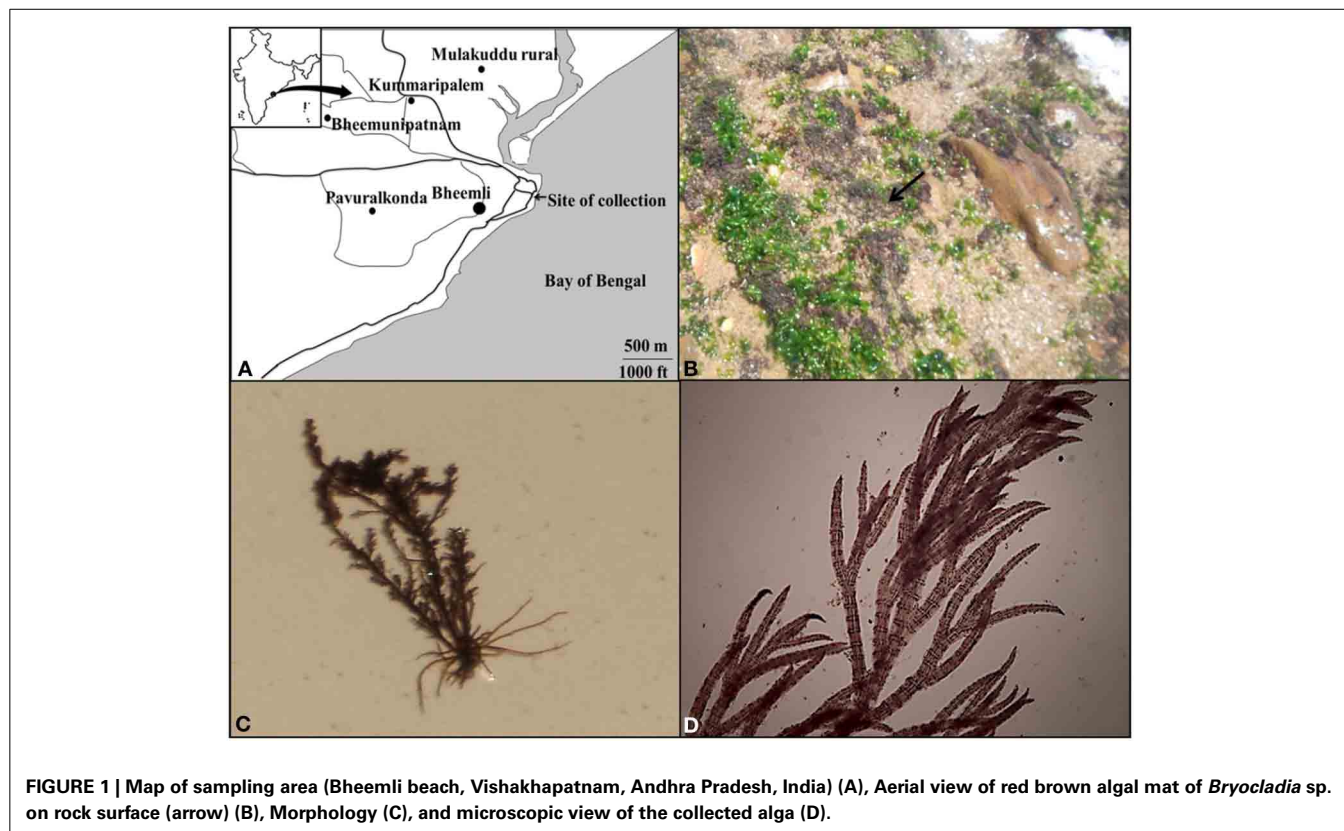


FIGURE 1 | Map of sampling area (Bheemli beach, Vishakhapatnam, Andhra Pradesh, India) (A), Aerial view of red brown algal mat of *Bryocladia* sp. on rock surface (arrow) (B), Morphology (C), and microscopic view of the collected alga (D).

HIGH PERFORMANCE LIQUID CHROMATOGRAPHY AND QUANTIFICATION OF MAAs

MAAs were analyzed with HPLC (Water 2998 with PDA detector, 515 PUMP, auto injector 717 plus, Milford, USA) equipped with Empower-2 software. The system has an ODS-2 (RP18) column (Water, Spherisorb analytical column, 5 μm , 4.6 \times 250 mm diameter, Ireland) along with guard (4.6 \times 10 mm). A 50 μl sample volume was used for injection and 0.2% acetic acid was used as mobile phase. The wavelength of detection was 330 nm with a flow rate of 1 ml min^{-1} . The compounds showing variable peaks were eluted and lyophilized. The quantification of MAAs was carried out by using molar extinction coefficient of standard MAAs (P-334).

ABSORPTION SPECTROSCOPY

The spectroscopic analysis of the water soluble aliquots was performed by using a UV-Vis double beam spectrophotometer (U-2910, 2J1-0012, Hitachi, Tokyo, Japan) in the range of 200–700 nm. The raw spectra were transferred to computer and peaks were analyzed by UV-solution software (Version-2.2, Hitachi, Tokyo, Japan).

DETERMINATION OF PHOTOSYNTHETIC PIGMENTS

Extraction of Chl *a* from *Bryocladia* sp. was carried out by the method as described by Tripathi (1983). Samples were sonicated (130 W, 20 kHz; Sonic and Materials, USA) with sodium carbonate in 80% acetone and kept for 15 min at 80°C in a water bath. The supernatant obtained after centrifugation was stored and pellet was repeatedly washed with acetone till the pellet became colorless. The pool supernatants were mixed with equal volume of water and petroleum ether followed by centrifugation at 5000 \times g for 15 min. The green top layer of supernatant was collected and lyophilized to evaporate the petroleum ether. Dried pellet was further dissolved in 80% acetone (3 ml) and spectra were taken in a UV-Vis spectrophotometer. Quantification of chlorophyll *a*

and carotenoids were carried out by the formula as described by Mackinney (1941) and Myers and Kratz (1955), respectively.

CELLULAR PROTEIN CONTENTS

Cellular protein was extracted by using the method of Wagner et al. (2002). Quantification of protein was done by measuring the absorbance of samples at both 280 and 260 nm and applying a correction formula of Warburg and Christian (1942).

PEROXIDE STABILITY AND SCAVENGING POTENTIAL OF MAAs

Peroxide stability and scavenging potential of MAAs was analyzed in the presence of hydrogen peroxide. For the analysis of stability of MAAs, 30% H_2O_2 (0–10 mM) was mixed with 0.074 mmol. g dry wt^{-1} of different MAAs and absorbance was taken at 323, 328, 335, and 340 nm by UV-Vis spectrophotometer after 5 and 30 min of incubation at 20 \pm 5°C. The stability was analyzed by the percentage inhibition. The quantification of scavenging potential of MAAs against peroxide was determined by the relationship between absorbance of different MAAs and H_2O_2 (240 nm). The molar extinction coefficient of H_2O_2 (43.6 $\text{M}^{-1} \text{cm}^{-1}$) was used for the calculation of relative inhibition.

STATISTICAL ANALYSIS

All the experiments were conducted in triplicate. Results were expressed as mean \pm SD ($n = 3$). A One Way ANOVA was applied for the statistical ($P \leq 0.05$) analysis. Sigma plot 11 and SPSS-16 software were used for the purpose of statistical analyses.

RESULTS

HIGH-PERFORMANCE LIQUID CHROMATOGRAPHY ANALYSIS OF MAAs

Mycosporine-like amino acids were extracted in Mili Q water and purified by high performance liquid chromatography (HPLC). Four peaks at retention times 2.65, 2.94, 3.23, and 3.56 min (Figure 2A) having absorption maxima at 340, 328, 335, and 323 nm, respectively (B).

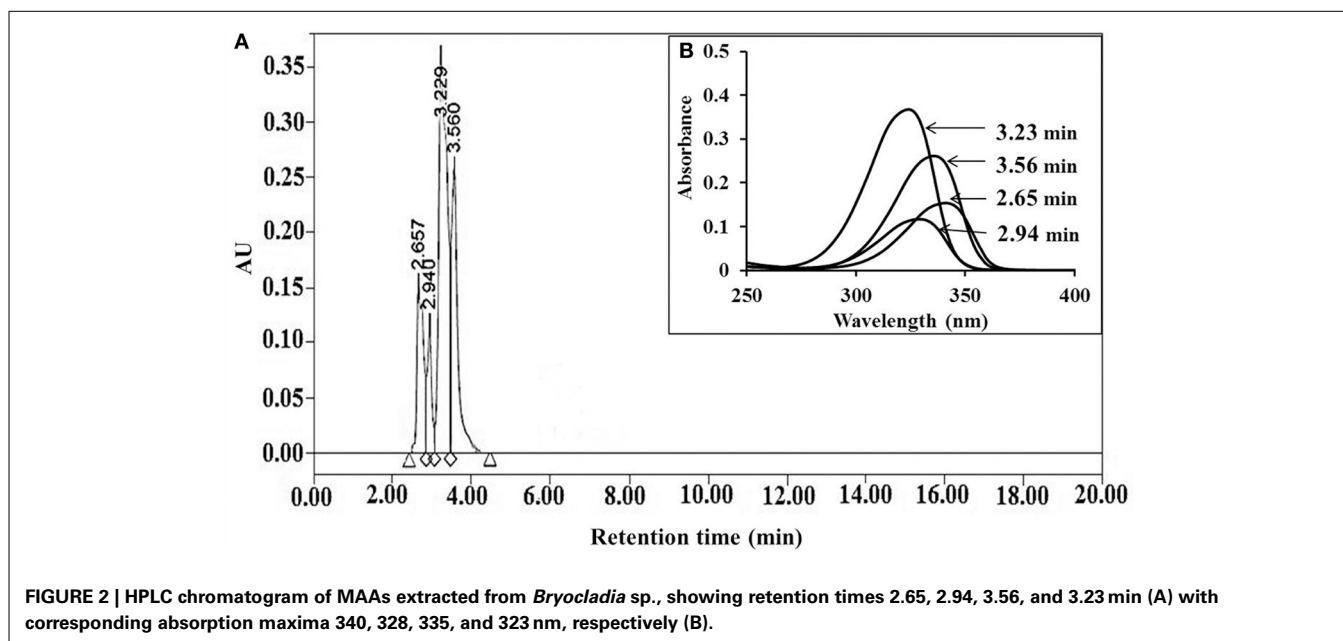


FIGURE 2 | HPLC chromatogram of MAAs extracted from *Bryocladia* sp., showing retention times 2.65, 2.94, 3.56, and 3.23 min (A) with corresponding absorption maxima 340, 328, 335, and 323 nm, respectively (B).

335 nm, respectively, were eluted (Figure 2B). Eluted MAAs were collected and lyophilized for oxidative analyses.

INDUCTION OF MAAs

MAAs were induced by UV-B irradiation (~1 Wm⁻²) (Figure 3A). Absorption spectrum obtained from HPLC showed significant induction of MAAs at wavelength 323 nm (RT 3.23 min) after 24 h of exposure in comparison to control. However, MAAs at 328 (RT 2.94 min), 335 (RT 3.56 min), and 340 nm (RT 2.65 min) were significantly induced up to 12 h of irradiation (Figure 3B). There was 22% induction in MAAs concentration as compared to control at 3.23 min with significant *p*-value (<0.05) after 24 h of exposure, whereas MAAs at 2.65 (5.3%), 2.94 (3.7%), and 3.56 (11.6%) min retention

times were comparatively less induced after 12 h of exposure. Further exposure resulted in the decline of MAAs concentration (Figure 3C). There was 42% reduction in MAAs concentration at 3.23 min after 48 h of exposure. Similarly, peak maxima with retention time of 2.65, 2.94, and 3.56 min showed significant (25–33%) decline (*p* < 0.05) after 48 h of exposure.

IMPACT OF UV-B ON PIGMENTS AND CELLULAR PROTEINS

The amount of chlorophyll (chl *a*) had declined up to 34, 66, and 98% after 12, 24, and 48 h of UV-B radiation, respectively (Figure 4A). However, the concentration of carotenoids was declined up to 33, 50, and 72% after 12, 24, and 48 h exposure of UV-B radiation, respectively (Figure 4B). The level of protein was significantly declined 23% after 12 h of exposure; however, it reached up to 40% after 48 h of UV-B irradiation (Figure 4C).

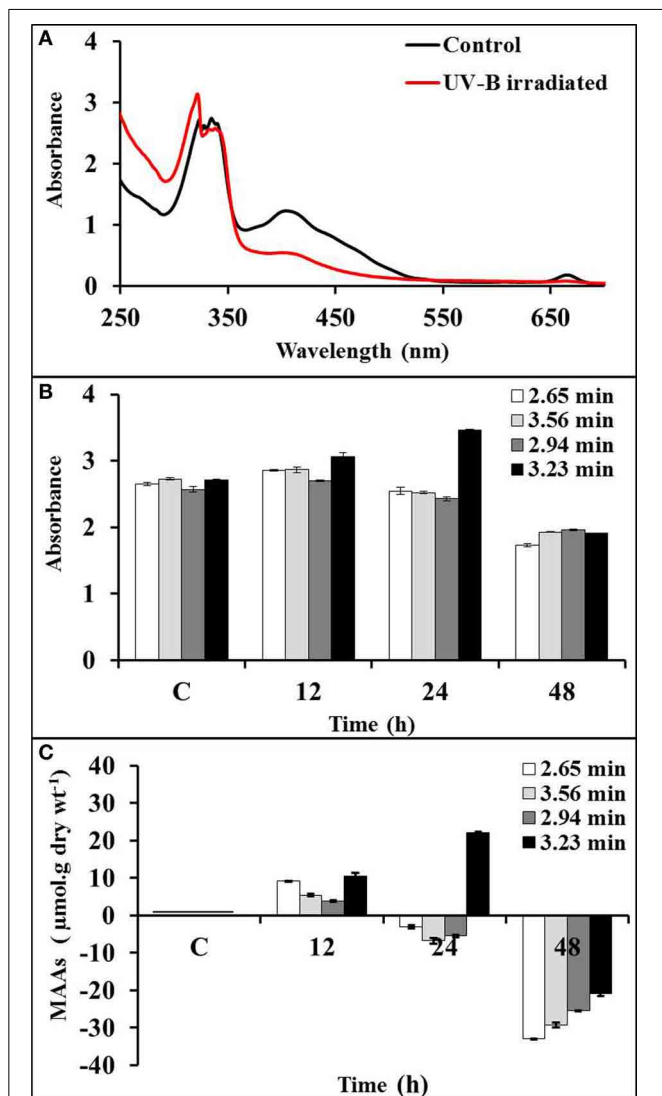


FIGURE 3 | Absorption spectra of control and UV-B irradiated *Bryocladia* sp. extracts (A) and showing relative induction of MAAs at retention times 2.65 (λ_{max} 340 nm), 2.94 (λ_{max} 328 nm), 3.56 (λ_{max} 335 nm), and 3.23 min (λ_{max} 323 nm) after 12, 24, and 48 h of UV-B irradiation (B). Actual percentage induction/reduction of MAAs (C).

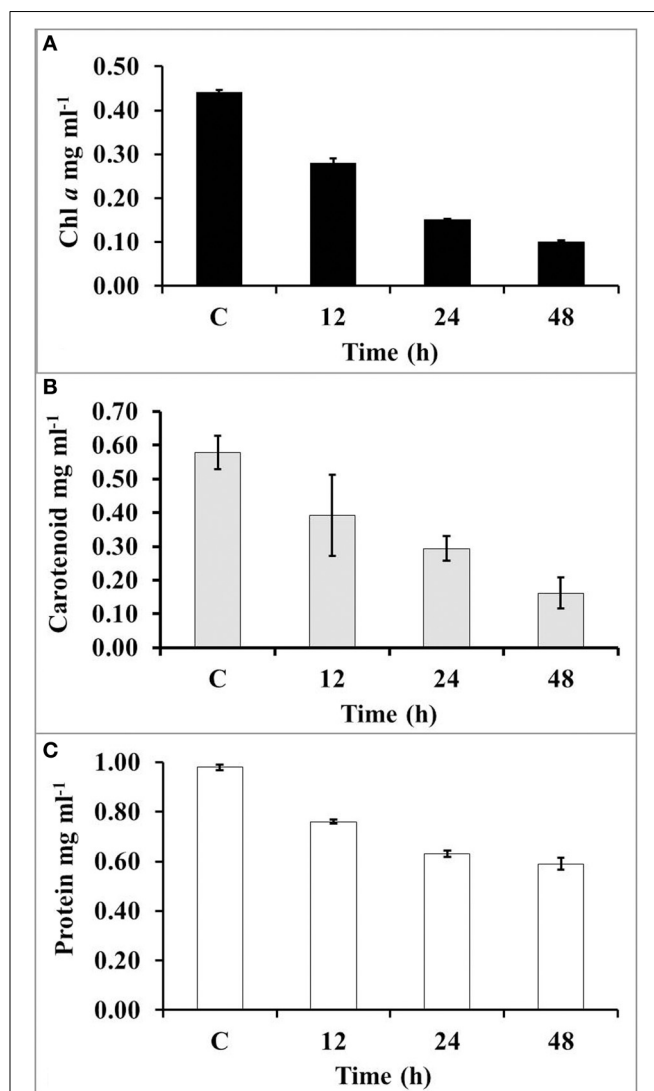


FIGURE 4 | The concentration of chl *a* (A), carotenoids (B), and total protein (C) after increasing time period of UV-B irradiation.

STABILITY AND SCAVENGING POTENTIAL OF MAAs

The overall stability of MAAs and their antioxidative behavior were analyzed by the treatment with different concentration (0–10 mM) of H_2O_2 (Figure 5). MAAs were significantly stable at 2 mM of H_2O_2 concentration. The increase (4–10 mM) in the concentration of H_2O_2 resulted in continuous decline in the absorbance maxima with a slight peak shift toward the shorter wavelength indicating the decline in absorbance of MAAs along with increasing absorbance of H_2O_2 at 240 nm.

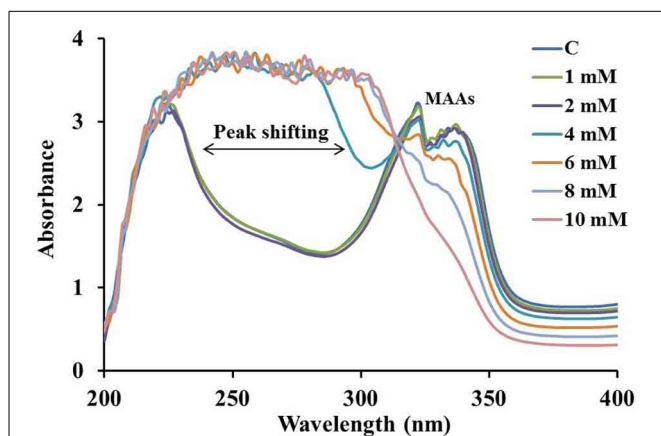


FIGURE 5 | Absorption spectrum of MAAs ($0.075 \text{ mmol.g dry wt}^{-1}$) in 0–10 mM of H_2O_2 (30%) showing peak shifting toward shorter wavelengths with increasing peroxide concentration.

STABILITY OF MAAs IN PEROXIDE TOXICITY

The stability of MAAs was analyzed in the presence of 0–10 mM H_2O_2 concentration with an incubation period of 5 and 30 min. Spectrophotometric analysis showed that MAAs were stable (>95%) up to 2 mM concentration of H_2O_2 . Thereafter, there was a gradual decline in the MAAs concentration with increasing (4–10 mM) concentration of H_2O_2 . Figure 6A indicates only 4 and 5% decline in MAAs concentration (323 nm) with 2 mM H_2O_2 after 5 and 30 min of incubation, respectively. However, further increase in H_2O_2 concentration (10 mM) had resulted 45 and 50% decline in MAAs concentration after 5 and 30 min of incubation, respectively. Similarly, the concentration of MAAs at wavelengths of 328 and 335 nm had declined up to 35–44% and 49–58% after 5 and 30 min, respectively, after incubation at 10 mM H_2O_2 concentration (Figures 6B,C). Interestingly, MAAs at 340 nm had invariably reduced (48–68%) after 5 and 30 min of incubation at 10 mM H_2O_2 , suggesting that this MAA was less stable after 30 min of incubation in H_2O_2 (Figure 6D).

PEROXIDE SCAVENGING POTENTIAL OF MAAs

All the purified MAAs effectively scavenges H_2O_2 upto a concentration of ≤ 2 mM. There was negligible decline in MAAs (323 and 328 nm) concentration as compared to control ($0.075 \text{ mmol.g dry wt}^{-1}$) up to 2 mM H_2O_2 (Figures 7A,B). However, other MAAs (335 and 340 nm) were declined up to 0.008 and 0.012 $\text{mmol.g dry wt}^{-1}$ after 2 mM H_2O_2 treatment (Figures 7C,D). Further increase in H_2O_2 concentration (10 mM), the MAAs at wavelengths 323, 328, 335, and 340 nm had declined up to 0.026, 0.036, 0.029, and 0.021 $\text{mmol.g dry wt}^{-1}$, respectively.

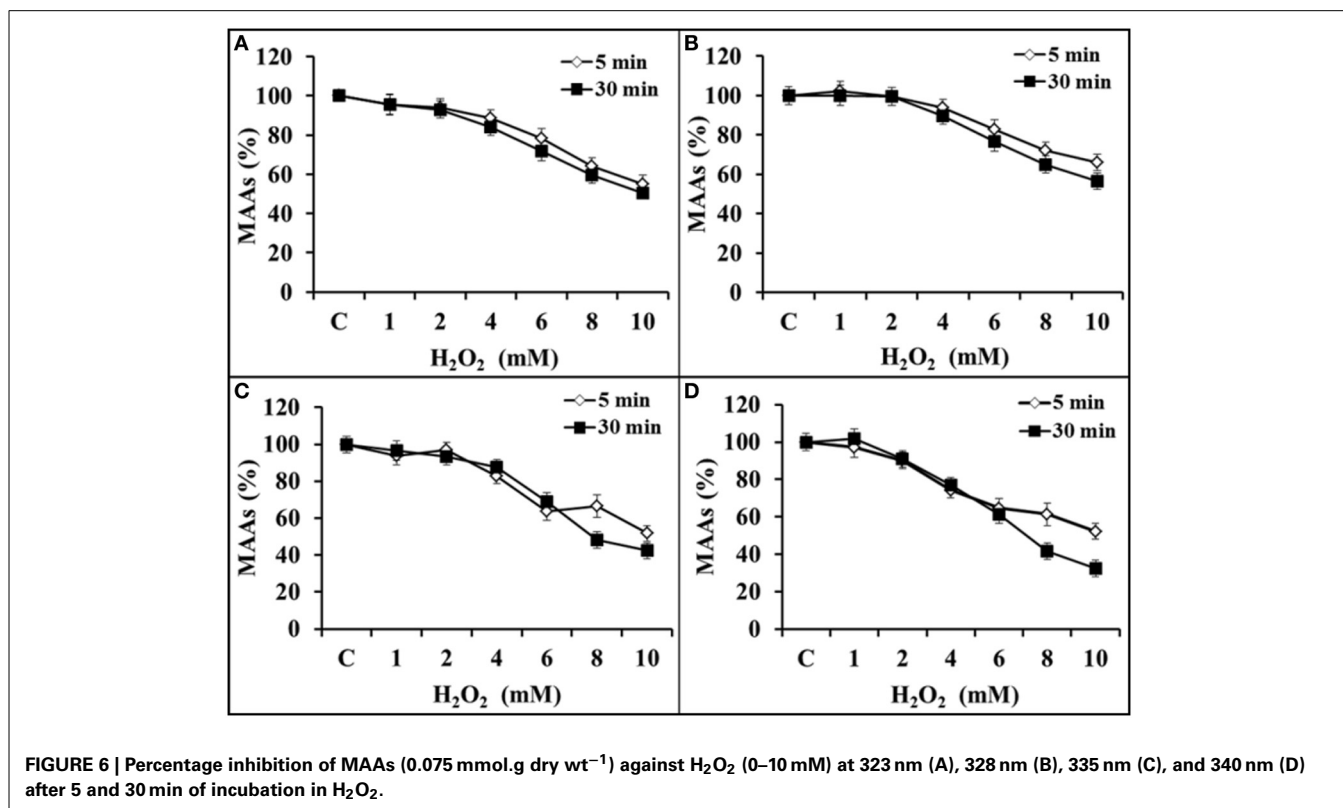


FIGURE 6 | Percentage inhibition of MAAs ($0.075 \text{ mmol.g dry wt}^{-1}$) against H_2O_2 (0–10 mM) at 323 nm (A), 328 nm (B), 335 nm (C), and 340 nm (D) after 5 and 30 min of incubation in H_2O_2 .

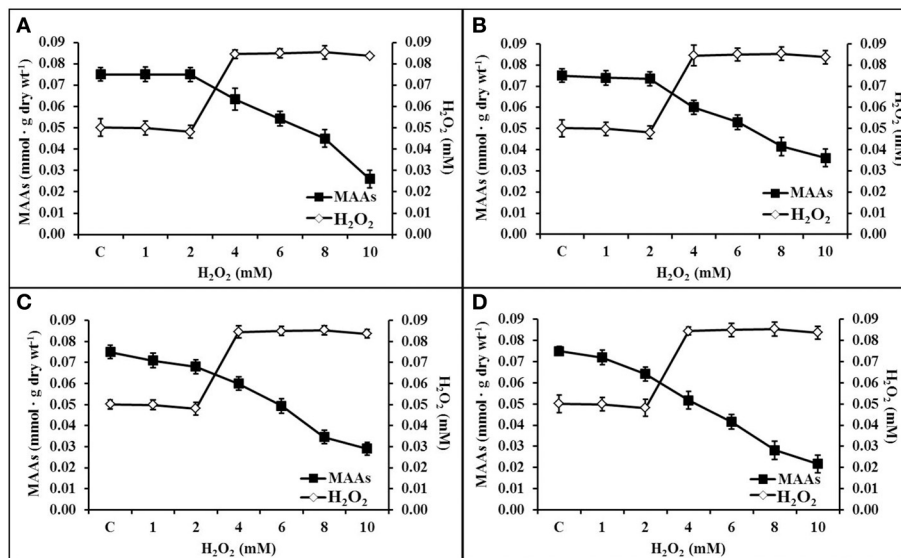


FIGURE 7 | Diagonal relationship of scavenging potential of MAAs in different concentrations of H₂O₂ (0–10 mM). Inactivation of H₂O₂ by MAAs at 323 nm (A), 328 nm (B), 335 nm (C), and 340 nm (D).

DISCUSSION

It is well established that highly energetic UV-B radiation is deleterious to all living organisms on the Earth's surface (Sinha et al., 2000). It has been reported that UV-B radiation promotes oxidative damage due to the overproduction of ROS inside the cells of several algae, cyanobacteria and plants (Rijstenbil, 2002; Shiu and Lee, 2005). H₂O₂ is the principle component of ROS that is initiated by UV-B irradiation in red algae (Lee and Shiu, 2009).

To cope up with such harsh oxidative conditions organisms have developed stress inhibition compounds that minimize the deterioration of biosynthetic, metabolic and native protein components with stable amino acids orientation. The database of one such UV-absorbing compounds MAAs has been reported in fungi, cyanobacteria, macroalgae, phytoplankton and animals (Sinha et al., 2007). MAAs are protective agent in living organisms that prevent 3 out of 10 photons of UV-B radiation thereby providing protection from irradiation to cyanobacteria (Garcia-Pichel et al., 1993). The algae of median tidal area are more efficient in producing UV-B screening compounds to cope up with its deleterious effect (van de Poll et al., 2003). Our results indicate that MAAs are inducible with UV-B radiation and can maintain good stability in peroxide stress conditions. Induction of MAA having RT 3.23 min (323 nm) after UV-B irradiation is unique in red algae because this MAA is more static in the range of 330–340 nm. Rastogi and Incharoensakdi (2013) have recently reported the presence of MAAs at 322 and 324 nm wavelengths in fresh water alga *Tetraspora* sp. CU2551 species that have potent photoprotective and antioxidative properties.

After UV-B irradiation, the chl *a* and carotenoids have been reported to be bleached due to the photooxidation of endogenous chromophores (Marwood and Greenberg, 1996; Sinha et al., 2000). Sinha et al. (2000) reported similar result of continuous decline in absorption peak at 665 and 475 nm in the red alga

Gracilaria cornea after UV-B radiation. The stability of growth and photosynthetic pigments in algae depends on the adaptation of species in harsh environmental conditions (Karsten et al., 2007). Production of antioxidative enzymes and induction of MAAs may help the organisms to reduce the impact of UV-B irradiation (Sinha et al., 2000; Rastogi et al., 2011). As stated earlier, UV-B irradiation induces the production of more ROS including peroxide and to cope up with this harsh condition of oxidative damage organisms have developed the ability to synthesize photoprotective MAAs. An oxo-carbonyl derivative linked MAAs have oxidative inhibition properties while imino-MAAs (P-334 and shinorine) present in red alga *Gracilaria cornea* are oxidatively inert (Sinha et al., 2000). We have first analyzed the *in vitro* effects of hydrogen peroxide on algal MAAs with a view to illustrate their level of stability and scavenging properties against ROS. It has been reported that mycosporine-glycine have oxidative quenching ability to protect the biological system against light oriented damage (Suh et al., 2003).

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