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Enhancing and upgrading biogas and biomethane production in anaerobic digestion: a comprehensive review

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Anaerobic digestion (AD) processes can face operational challenges or flaws such as substrate structure and characteristics complexity, process complexity, low productivity, inefficient biodegradability, and poor stability, which suppresses or reduces biogas and biomethane production. As a result of the need to overcome these challenges/shortcomings and improve or enhance biogas and biomethane yield, process intensification methods have gained attention. There is some literature review on pretreatment and co-digestion as a means of improving AD performance; however, there is no systematic information on the various strategies required for improving AD performance and, in turn, increasing biogas/biomethane yield. The AD process produces biogas, a valuable renewable biofuel. Biogas is composed primarily of biomethane and other undesirable components such as carbon dioxide, oxygen, hydrogen sulphide, water vapour, ammonia, siloxanes, nitrogen, hydrocarbons, and carbon monoxide, which act as impurities or contaminants and tend to reduce the biogas specific calorific value while also causing various problems with machine operation. As a result, various technologies are used to improve raw biogas quality by removing contaminants during biogas transformation to biomethane. As a result, this paper provides a comprehensive review of the various systematic process intensification strategies used to overcome AD process challenges/shortfalls, improve or enhance biogas and biomethane production, and conventional and emerging or advanced technologies for biogas purification, cleaning, and upgrading.

KEYWORDS

anaerobic digestion, biogas, biomethane, upgrading, calorific value, contaminants, pretreatment, co-digestion

1 Introduction

The global industrial revolution, together with simultaneous socio-economic and technological advancement, has been driven by the utilisation of fossil fuels (oil, gas, and coal), but its use has, over the years, negatively impacted the environment by causing climate change through the release of toxic pollutants and contributing to greenhouse

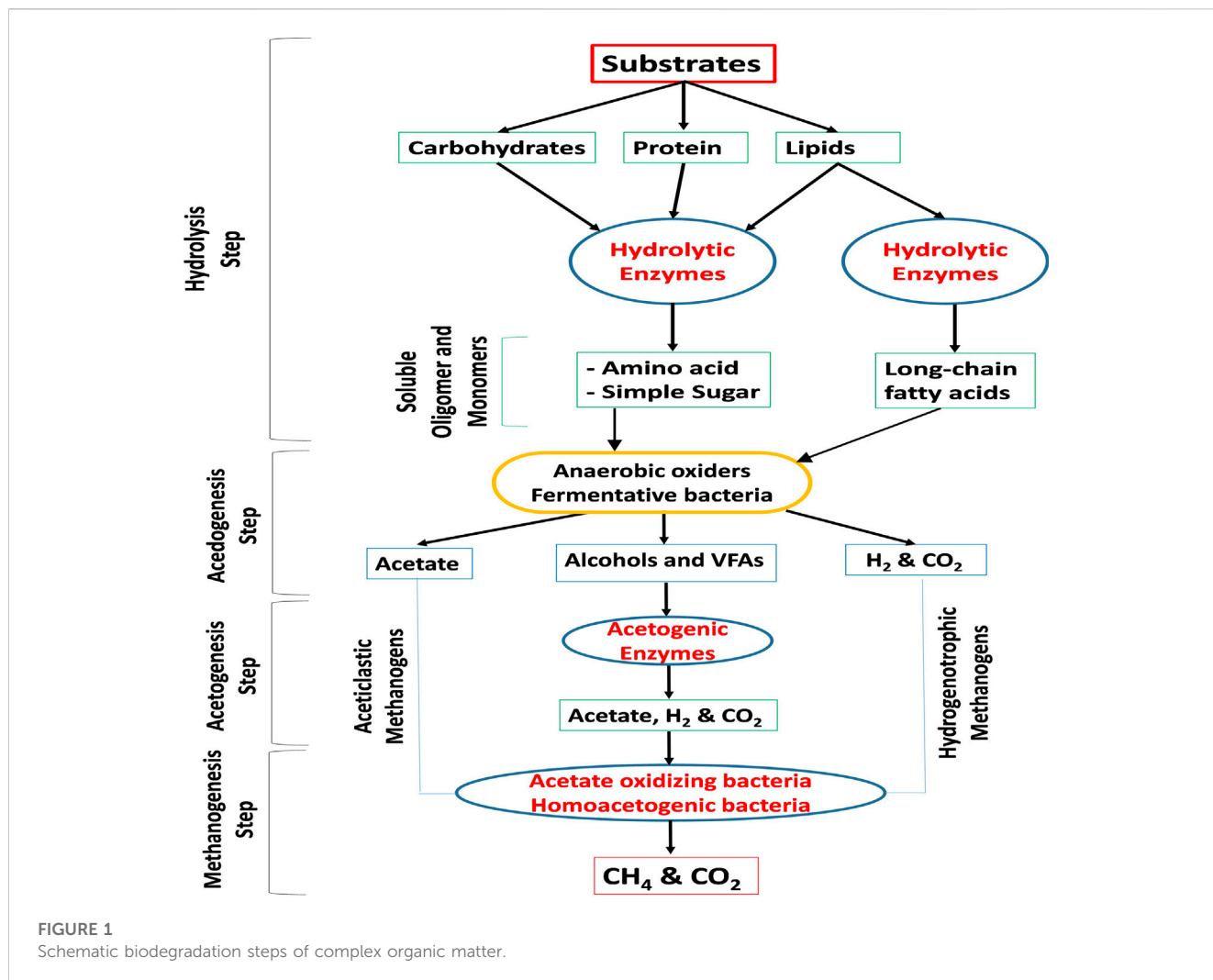


FIGURE 1 Schematic biodegradation steps of complex organic matter.

gas emission (Kasinath et al., 2021). Sustainable global energy strategies include the use of renewable energy resources such as geothermal, hydro, solar, biomass, and wind to reduce the use of fossil fuels (Chen et al., 2010). One of the renewable energy technologies that can be employed to generate bioenergy from biomass is the anaerobic digestion (AD) process technology.

The AD process technology is a complex multistep process that involves the degradation or breakdown of organic matter by a group of microorganisms in the absence of oxygen, resulting in the production of biogas/biomethane and digestate as a byproduct (Kasinath et al., 2021). Figure 1 illustrates the steps involved in the biodegradation of complex organic matter. As shown in Figure 1, during the hydrolysis step (the first step), hydrolytic enzymes released by bacteria break down substrates such as carbohydrates, proteins, lipids, and insoluble polymers into soluble oligomers and monomers. In the acidogenesis step (the second step), amino acids, simple sugars, and long-chain fatty acids are further broken down into volatile fatty acids (VFAs), acetate, alcohols, CO₂, and H₂. The third step, acetogenesis, involves the further metabolism of VFAs and alcohols into acetic acid and H₂. Finally, in the methanogenesis step (the fourth and final step), the mixture of CO₂, H₂, formate, methanol, and acetate is transformed into methane (or biomethane)

through acetoclastic, hydrogenotrophic, and methylotrophic pathways (Kasinath et al., 2021), which is the ultimate product of the process.

The digestate is a nutrient-rich biomass that can be further used as an organic fertilizer for organic farming (Khan et al., 2017; Cai et al., 2021). The biomass wastes generally utilised for biogas/biomethane production can be categorised into agricultural wastes, energy crops/residues, municipal biowaste, industrial wastes, and wastewater (Ljupka, 2010), as well as aquatic biomass (e.g., algae) (Ward et al., 2014) as schematically presented in Figure 2.

The AD is normally carried out in biodigesters or bioreactors. The performance of biogas digesters or reactors with respect to biogas and biomethane production or yield is linked to various physical, chemical, and biological factors such as the structure or nature and characteristics (properties) of the organic waste (feedstock), total solid content; hydraulic retention time, pH, stirring, carbon/nitrogen ratio, temperature, organic loading rate, volatile fatty acids (VFA) and seeding (Rocamora et al., 2020; Uddin et al., 2021). Any forms of modification that might be made to these factors can lead to changes in the surrounding environment of the microbial community and movement within the AD biodigester.

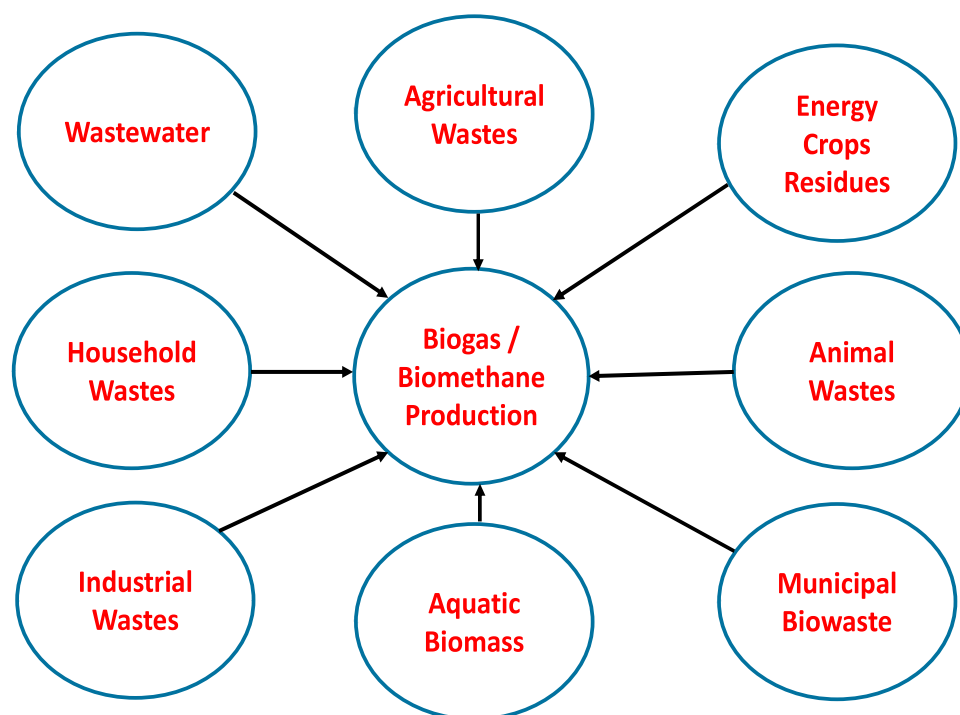


FIGURE 2
Substrates for biogas/biomethane production.

Hence, to optimize biogas and biomethane production, it is essential to control these factors. These factors can be manipulated within an appropriate and acceptable range so that the AD biodigester or reactor can run optimally and efficiently without failure or collapse. Furthermore, other operational shortcomings, such as substrate complexity, low productivity, process complexity, inefficient biodegradability, and poor stability, tend to suppress or reduce biogas and biomethane production. Therefore, the need to overcome these shortcomings and improve or enhance biogas and biomethane yield has drawn attention to process intensification methods. Different process intensification strategies have been employed to overcome these shortcomings and improve or enhance biogas and biomethane production (Yadvika et al., 2004).

The raw biogas produced from organic biomass wastes through the AD processes consists majorly of biomethane (CH_4) and carbon dioxide (CO_2). Aside from the CH_4 and CO_2 , the raw biogas also contains some small or minor amounts of hydrogen sulphide (H_2S), ammonia (NH_3), water vapour (H_2O), nitrogen (N_2), oxygen (O_2), carbon monoxide (CO), hydrogen (H_2), siloxanes and hydrocarbons which are regarded as impurities or pollutants (Sun et al., 2015; Angelidaki et al., 2018). Some of these impurities have significant negative effects on the biogas utilisation system, such as calorific value reduction, corrosion, etc. (Sun et al., 2015). In order to reduce or remove these unwanted components that are harmful to the biogas utilisation systems and increase the biogas calorific value, it is important to clean or purify the raw biogas and upgrade it to a higher biofuel standard. The process is referred to as biogas cleaning or purification and upgrading (Sun et al., 2015; Angelidaki et al.,

2018). Upgrading biogas to biomethane is one of the technologies that has attracted great interest in the bioenergy industry (Sun et al., 2015). When the biogas is purified or cleaned and upgraded to required specifications that are similar to natural gas, the final gas product is called biomethane (Kougias et al., 2017; Angelidaki et al., 2018).

A great number of biogas cleaning and upgrading technologies have been developed, and some of these technologies are commercially available (Sun et al., 2015). To this end, some recent researchers have presented a review of biogas cleaning/purification and upgrading technologies (Sun et al., 2015; Angelidaki et al., 2018; Ryckebosch et al., 2011; Bauer et al., 2013a; Muñoz et al., 2015; Kadam and Panwar, 2017; Struk et al., 2020; Domingues et al., 2021). In light of the constantly evolving biogas cleaning and upgrading technologies and market, it is important to conduct frequent assessments of the available technologies. While some literature reviews have been published exploring strategies to improve AD performance and increase biogas/biomethane yield, such as co-digestion, pretreatments, and adding additives, these publications are not comprehensive (Yadvika et al., 2004; Deepanraj et al., 2013; Romero-Güiza et al., 2016; Sarker et al., 2019; Kasinath et al., 2021). Therefore, a more thorough evaluation of the available strategies is necessary to ensure optimal performance of the AD process. Also, for different purposes of utilising biogas/biomethane, selecting the appropriate biomass pretreatment technique prior to biogas production and the appropriate biogas upgrading technology based on fundamental criteria is pertinent, rather than simply selecting the cheapest one without basic knowledge. If the cheapest technology is chosen and

does not satisfy the biogas utilization requirements can result in serious damaging consequences, which may even destroy the system or result in a much higher total cost (Sun et al., 2015). Therefore, comprehensive information on the parameters or criteria for consideration in selecting the appropriate pretreatment technique for biomass prior to biogas/biomethane generation as well as the appropriate biogas upgrading technology required to satisfy biomethane quality is still limited.

The purpose of this paper is to provide a comprehensive review of recent advances in enhancing the production of biogas and biomethane. Additionally, this review discusses various biogas cleaning, purification, and upgrading technologies that are used globally and commercially. The intention is for this review to serve as a comprehensive guide for improving AD processes, selecting appropriate biomass pretreatment techniques, and choosing biogas upgrading technologies. One of the main contributions of this study is to provide guidance on selecting pretreatment techniques based on factors such as biomass hydrolysis rate, generation of toxic inhibitory compounds, operational cost, energy requirement, and effectiveness. Additionally, this review offers information on selecting biogas upgrading technologies based on investment cost, efficiency, maintenance cost, and other relevant criteria. Finally, this study includes information on the challenges faced by biogas upgrading technologies and recommended mitigation strategies.

1.1 Biogas and biomethane

Biogas is a non-polluting, clear combustible gas that is considered an alternative energy source (Umar et al., 2013). The physical properties of biogas are presented in Table 1.

It is an odourless and colourless gas that is flammable and smokeless (i.e., burns with blue flame). It has a specific gravity of 0.847–1.004, an ignition temperature between the range of 650°C–750°C, a calorific value of 4,740–7,500 kcal/Nm³, and an energy content of 37.3 MJ/m³ (Korbag et al., 2020). Biogas consists mainly of methane (CH₄) and carbon dioxide (CO₂), with traces of hydrogen sulphide (H₂S), water vapour (H₂O), ammonia (NH₃), oxygen (O₂) and nitrogen (N₂) (Sun et al., 2015; Agbede et al., 2019; Hosseinipour and Mehrpooya, 2019; Struk et al., 2020; Domingues et al., 2021; Moya et al., 2022). It may sometimes contain very small quantities of siloxanes, aromatic and halogenated compounds (Ward et al., 2014). Table 2 presents the findings of various investigations conducted on the composition of typical biogas, which indicate that methane (CH₄) constitutes between 45% and 75% of the biogas, and carbon dioxide (CO₂) is present in a range of 25%–50% (Zhou et al., 2016; Angelidaki et al., 2018; Hosseinipour and Mehrpooya, 2019; Struk et al., 2020; Kasinath et al., 2021; Moya et al., 2022). Insignificant amounts of hydrogen sulphide (H₂S) (<2% or 0–10000 ppm), ammonia (NH₃) (<1% or 0–100 ppm), water vapour (H₂O) (1%–10%), nitrogen (N₂) (0%–15%), oxygen (O₂) (0%–3%), and siloxanes (<0.02%) were also reported (Khan et al., 2017; Struk et al., 2020; Moya et al., 2022).

The biogas composition is intrinsically dependent on the carbon redox state of the waste's organic matter, the origin and type of feedstock, and the type of AD process technology (Jönsson et al., 2003). The composition of the biogas revealed that CH₄ is the most

TABLE 1 Physical properties of biogas.

Property	Value
Colour	Colourless
Odour	Odourless
Specific gravity	0.847–1.004
Ignition temperature	650–750 °C
Calorific value	4,740–7,500 kcal/Nm ³
Energy	37.3 MJ/m ³

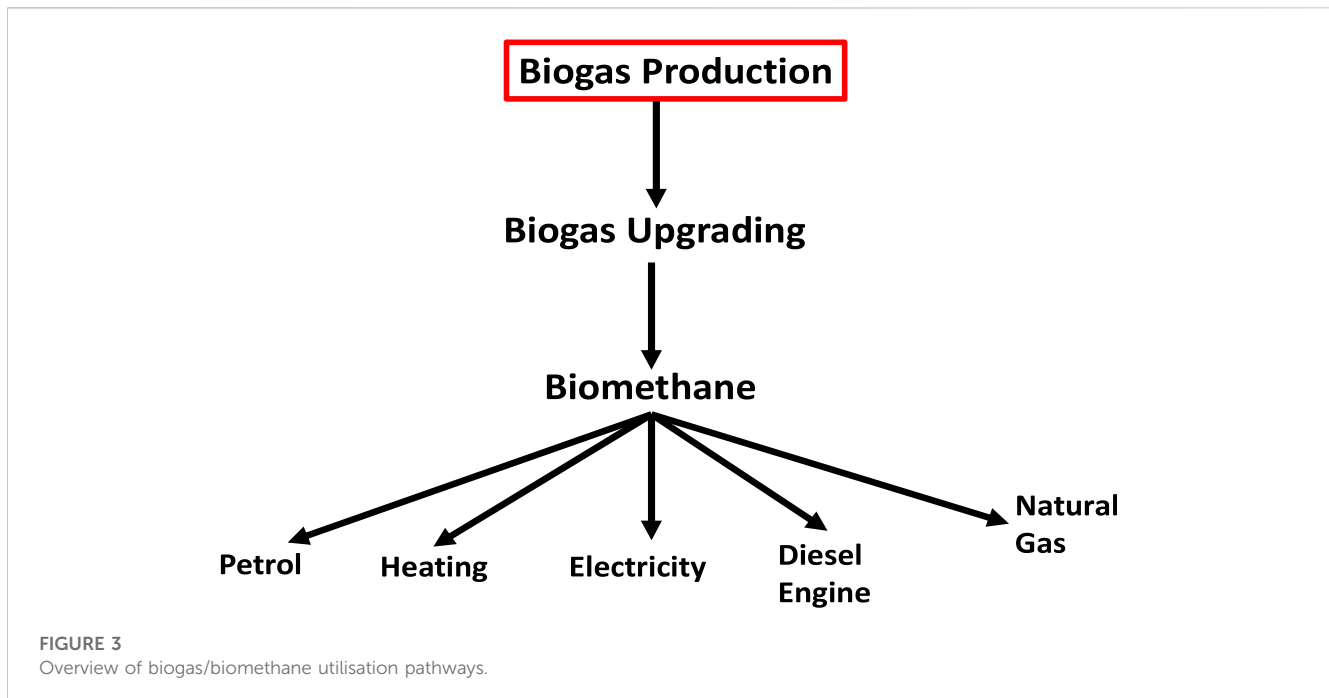
TABLE 2 Typical composition of biogas from biowaste

Component	Symbols	Concentration (Vol %)
Methane	CH ₄	45–75
Carbon dioxide	CO ₂	25–50
Hydrogen sulphide	H ₂ S	<2 (0–10,000 ppm)
Ammonia	NH ₃	<1 (0–100 ppm)
Water	H ₂ O	1–10
Nitrogen	N ₂	0–15
Oxygen	O ₂	0–3
Siloxanes	H(OSiH ₂) _n OH and (OSiH ₂) _n	0.02

significant component, and thus the high content of CH₄ in the biogas brought about the idea of biomethanation (anaerobic digestion of biodegradable organic wastes to produce biomethane), which is often used to describe the process of biogas production (Prakash and Singh, 2013).

Biomethane, which can be obtained from biogas by purification (i.e., cleaning) and upgrading (Sun et al., 2015), is a colourless and odourless gas that is not very soluble in water. It has a high energy content that ranges from 50 to 55 MJ/kg (Angelidaki et al., 2018; Tabatabaei et al., 2020; Mahmudul et al., 2021). In the last 5 years, the market for biomethane has grown exponentially (Moya et al., 2022). According to Moya et al. (Moya et al., 2022), the European Biogas Association (EBA) reported that in Europe, there were 483 existing active biomethane production plants in 2018 with a total production capacity of 22,787 GWh/year and that this number increased to 729 active plants at the end of the year 2020 (i.e., about 51% increase) with France and Germany being the major producers of biomethane. Figure 3 shows the overview of biogas/biomethane utilisation pathways.

The utilisation of biogas as fuel for various purposes is depicted in Figure 3. It can be used as fuel for both petrol (Supporting Material) and diesel (CI) engines and utilised as fuel for boilers, turbines, and internal combustion engines to generate electricity, heat, and steam. Furthermore, biogas is employed as a substitute for natural gas in domestic and industrial settings and as a feedstock in fuel cells (Andriani



et al., 2014; Kadam and Panwar, 2017). In developing countries, biogas is mostly used for cooking and lighting purposes in stoves and lamps, respectively (Khan et al., 2017; Domingues et al., 2021). The storage of biogas in compressed cylinders and its transportation through pipelines can make it accessible in remote areas (Hosseini pour and Mehrpooya, 2019).

1.2 Strategies for AD process enhancement

Different strategies have been employed to enhance biogas and biomethane production during the AD processing of raw materials (Romero-Güiza et al., 2016), as schematically illustrated in Figure 4.

Figure 4 shows that the strategies include (i) use of pretreatments, (ii) use of co-digestion, (iii) bioaugmentation (use of inoculum or seeding bacteria), (iv) biostimulation (use of macro and micro nutrients), (v) use of carbon-based conductive materials and (vi) use of improved reactor configuration and operation conditions.

1.3 Use of pretreatment

Pretreatment, also referred to as the conditioning process, is any action performed on feedstock to weaken or overcome the recalcitrance attributed to its cell wall and structural characteristic (Andriani et al., 2014; Seidl and Goulart, 2016), thereby facilitating its solubilization and hydrolysis (Morales-Polo, 2018). Feedstock pretreatment is adapted based on the feedstock structure and characteristics (Kasinath et al., 2021). That is, feedstock such as lignocellulosic biomass and municipal solid waste that is composed of complex organic matter (cellulose, hemicellulose, and lignin) or refractory compounds that are not readily biodegradable or highly recalcitrant are considered for

pretreatment before its utilisation in the AD process. The most recalcitrant component of the feedstock is lignin, which protects the cellulose and hemicelluloses by forming tight bonds, thereby reducing the available surface area for enzyme attack and consequently hindering the digestion or total breakdown (degradation) of the entire substrate structure (Paudel et al., 2017). Examples of feedstocks resistant or recalcitrant to microbial digestion are lignocellulosic feedstocks or biomass, such as energy crops and plant residues (Carlsson et al., 2012). Pretreatment is employed to facilitate or enhance microbial degradation or digestion (i.e., solubilisation and hydrolysis) of complex organic matter into simple components by removing the obstacles and making the organic content of the feedstock easily available and useable by the microbial populations (Patinvoh et al., 2017; Kasinath et al., 2021), as well as to enhance biogas and biomethane production and to improve dewatering and the digestate quality (Zhen et al., 2017; Kasinath et al., 2021).

Generally, pretreatment enhances the production of biomethane from the pretreated lignocellulosic feedstock or biomass in the 25%–120% range compared with the untreated feedstock (Mustafa et al., 2017; Rani et al., 2021). Biomass or feedstock pretreatment can shorten the duration of the hydrolysis stage (the rate-limiting step) of AD while at the same time accelerating soluble substances' bioavailability for methanogenic bacterial enzymes (Kasinath et al., 2021). Pretreatment improves the feedstock accessibility to the microbial community by increasing biomass porosity, decrystallization, surface area and solubilisation (Carlsson et al., 2012). The efficiency of the feedstock pretreatment can directly be expressed based on the increase in the yield of biogas or biomethane or indirectly expressed based on the increase in the soluble components obtained (Kasinath et al., 2021). Pretreatment techniques or strategies as schematically represented in Figure 5 can broadly be classified as physical (mechanical, thermal (temperature induced and steam explosion), sonication,

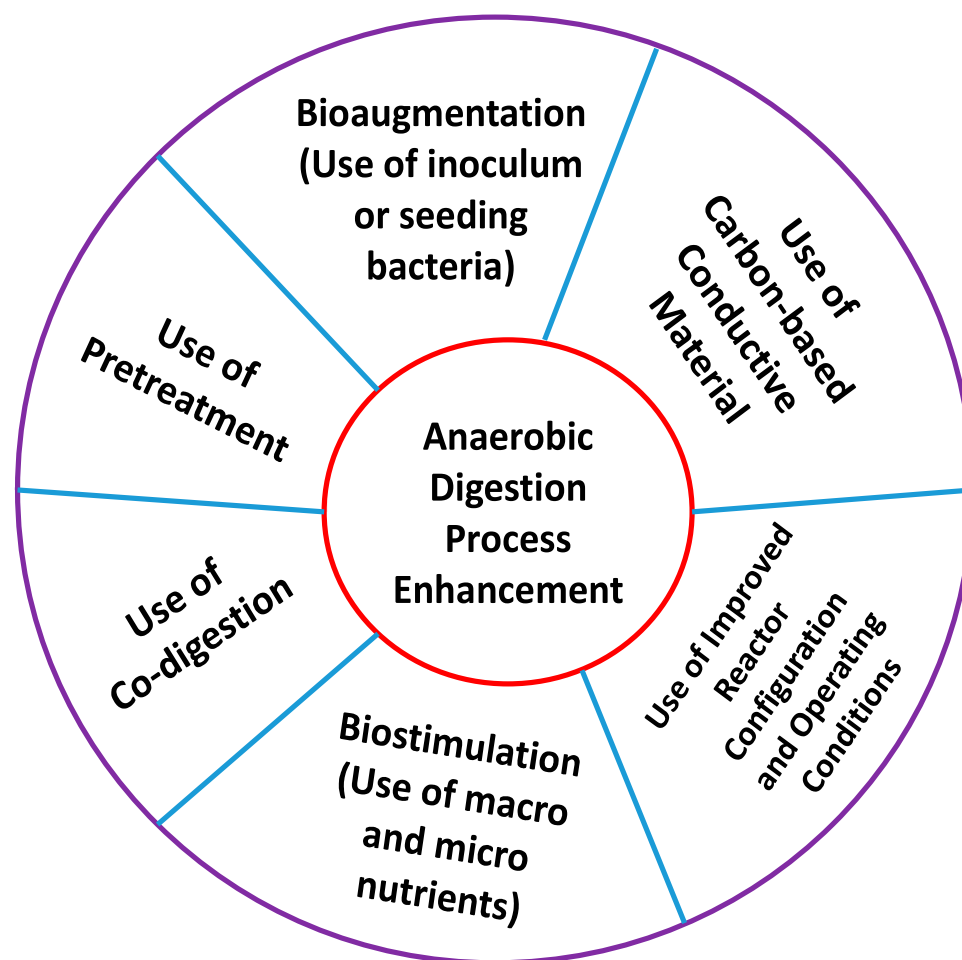


FIGURE 4
Strategies for anaerobic digestion process enhancement.

microwave irradiation), chemical (acidic, alkali, ozonation, Fenton and Fe(II)-activated persulfate oxidation), and biological (bacterial enzyme and fungi) (Montingelli et al., 2017; Kasinath et al., 2021).

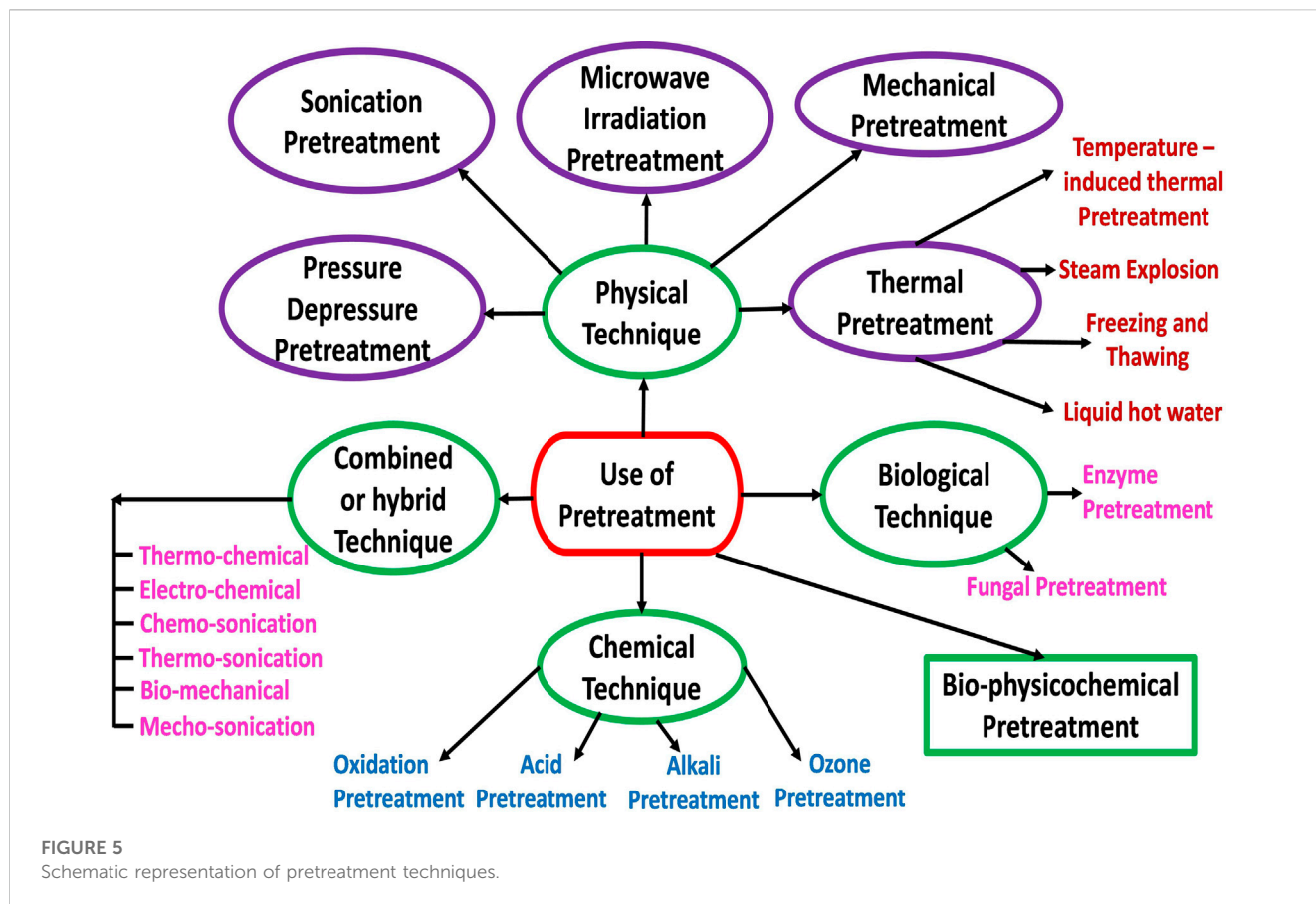
1.3.1 Physical technique

The physical technique includes operations such as mechanical operations (milling, grinding, lysing, centrifuge, high-pressure homogeniser, and electroporation), thermal operations, sonication (ultrasound), and microwave irradiation (Deepanraj et al., 2013; Morales-Polo, 2018). The physical pretreatment technique of feedstock affects the cellulose crystallinity, degree of polymerization, particle size, surface area and pore size of the feedstock (Tsapekos et al., 2015; Khan et al., 2017).

1.3.1.1 Mechanical pretreatment

Mechanical pretreatment (e.g., milling/grinding) is based on increasing the feedstock contact surface area by disintegrating and reducing its particle size. The increased surface area will allow for better contact and interaction between the feedstock and the microbial community (Zhang et al., 2014; Morales-Polo, 2018). When there is sufficient particle size reduction, up to a 40% increase or improvement in biomethane yield can be achieved

(Morales-Polo, 2018). The most common procedures in reducing particle size include screw presses, milling/grinding, and disc screening (Morales-Polo, 2018). Mechanical milling can be achieved using different mills such as attrition, ball, centrifugal, colloid, hammer, extruders, knife, pin, and vibratory mills (Cheng and Timilsina, 2011). With the use of mechanical milling, the degree of polymerization, particle size, and cellulose crystallinity is reduced, and thus the surface area and feedstock digestibility are enhanced. Mechanical pretreatment improves flow properties, bulk density, bioconversion effectiveness, particle densification and distribution, porosity and the overall lignocellulosic biomass conversion without any toxic by-products (Barakat et al., 2014). Mechanical pretreatment using screw press extrusion has been employed on dip litter manure (solid fraction and straw mixture) to achieve up to a 30% increment in methane yield (Hjorth et al., 2011), and Pilarski et al. (Pilarski et al., 2016) reported a 16.5% increase in the yield of methane after maize straw silage pretreatment using single screw extrusion as compared to the untreated control. Dell'Omo and Froschia (Dell'Omo and Froschia, 2018), using knife milling pretreatment on wheat straw, reported an increase in methane yield by 49.3% when compared with the untreated, while an increase in the yield of methane by 44.7% after ball milling



pretreatment of grass as compared to the untreated was reported by Pengyu et al. (Pengyu et al., 2017). Also, Mönch-Tegeder et al. (Mönch-Tegeder et al., 2014) reported a 26.5% enhancement in methane production when horse manure was pretreated using the mechanical pretreatment method compared to the untreated. In addition, Agyeman and Tao (Agyeman and Tao, 2014) gave a report of a 9%–34% of improvement in the yield of biogas when food waste was mechanically pretreated using the grinding method prior to its AD processing. It is to be noted that an excessive particle size reduction can result in hydrolysis overloading and subsequent accumulation of volatile fatty acids (Izumi et al., 2010).

1.3.1.2 Thermal pretreatment

Thermal pretreatment has to do with exposing the feedstock to temperature change over a duration of time. This can be subdivided into the following.

(a) Temperature-induced thermal pretreatment: This takes place at a vast temperature range of 70 °C–275 °C (Raju et al., 2013). However, during temperature-induced thermal pretreatment, temperature above 250 °C should be avoided and this is because, beyond 250 °C, pyrolysis reactions that are not wanted will begin to occur (Deepanraj et al., 2013). The thermal treatment enhances the hydrolysis of the feedstock leading to an increased rate and extent of AD. Temperature induced thermal pretreatment conducted on feedstocks such as food waste (Marañón et al., 2012), waste activated sludge (Bougrier

et al., 2006), municipal solid waste (Ariunbaatar et al., 2014), algae (Schwede et al., 2013), agricultural byproducts (Menardo et al., 2012), and grass (Menardo et al., 2012) resulted in an increased yield of biomethane. Also, Carrère et al. (Carrère et al., 2009) performed a temperature induced thermal pretreatment on pig manure (total liquid and solid fractions) at a temperature of 190°C and reported an improved or enhanced yield of biogas, while Rafique et al. (Rafique et al., 2010) reported that there was no improvement in the yield of biomethane from a similar feedstock (solid fraction of pig manure) subjected to a pretreatment temperature of over 100 °C. According to Rafique et al. (Rafique et al., 2010), the yield of biogas and biomethane from temperature-induced thermal pretreated feedstocks is not always higher than those not subjected to pretreatment.

(b) Steam explosion: This is temperature and pressure induced, which is most commonly applied for the pretreatment of lignocellulosic biomass. The applied pressure can range from 5 to 50 bar, while the temperature can be in a range of 160°C–250°C (Zheng et al., 2014; Khan et al., 2017). In this method, the feedstock is put into a closed container or vessel and high temperature-pressure steam is directly passed into it for a few minutes or a short duration (Deepanraj et al., 2013). In the steam explosion, the pretreatment time is dependent on the feedstock moisture content (MC). If the MC is high, the optimum steam explosion pretreatment time will be long (Deepanraj et al., 2013). This pretreatment method is often

defined using a severity factor. This factor is calculated from the time duration and temperature of the process, which is expressed in Eq. 1 (Amin et al., 2017):

$$\log R_O = \log\left(t \cdot e^{\left(\frac{T-100}{14.75}\right)}\right) \quad (1)$$

where $\log R_O$ = severity factor; T = temperature ($^{\circ}\text{C}$); t = residence time (min); and 14.75 = activation energy (follows first-order kinetics and is Arrhenius temperature dependent).

The steam explosion causes the lignocellulosic biomass fibres to open up, making them more accessible for degradation. According to Horn et al. (Horn et al., 2011) and Zhou et al. (Zhou et al., 2016), the yield of biogas from *Salix* woodchips and rice straw (lignocellulosic biomass) was maximised using a steam explosion method at 210 $^{\circ}\text{C}$ for 10 min and at 200 $^{\circ}\text{C}$ for 2 min duration, respectively. Also, Vivekanand et al. (Vivekanand et al., 2013) reported a two-fold increase in the yield of biomethane from birch wood chips when a steam explosion method at a temperature of 220 $^{\circ}\text{C}$ with a severity of 4.5 was employed. The steam explosion's severity factor for most feedstocks usually falls within the range of 3.14–3.56 (Amin et al., 2017). Furthermore, Li et al. (Li et al., 2016), Lizasoain et al. (2017), Mulat et al. (Mulat et al., 2018), Steinbach et al. (Steinbach et al., 2019) and Weber et al. (2020) respectively reported that *Miscanthus lutarioriparius*, corn stover, Birchwood chips, rice straw and Agave bagasse feedstocks that were subjected to initial steam explosion pretreatment at 153–198 $^{\circ}\text{C}$, 140–220 $^{\circ}\text{C}$, 162–240 $^{\circ}\text{C}$ and 142–179 $^{\circ}\text{C}$ resulted in an increased yield of methane by 49%, 22%, 140% and 11% as compared to the untreated after undergoing anaerobic digestion process.

(c) Liquid hot water or hydrothermal pretreatment: This is also known as wet torrefaction (Alvarez-Chavez et al., 2020). In this case, liquid hot water is utilised in place of steam. The liquid hot water is obtained using heat combined with high pressure (Rodriguez et al., 2017; Rocamora et al., 2020). Generally, it is carried out at a temperature range of 100–140 $^{\circ}\text{C}$, together with a pressure range between 1 and 2 bar (Passos and Ferrer, 2015). However, in some cases, the pretreatment can be performed at a temperature range of 150 $^{\circ}\text{C}$ –260 $^{\circ}\text{C}$ (Chang et al., 2013). The hydrothermal pretreatment can be carried out in an autoclave (Passos and Ferrer, 2015) as well as in stainless kettles heated by an oil batch (Qiao et al., 2013). Its purpose is to selectively break down or degrade the hemicellulose of the substrate. That is, the hot water easily and readily solubilises the feedstock and prevents the formation of inhibitory compounds. In using this method, the pH level should be maintained at the range of 4–7 as this acts as a catalyst for the cellulosic raw material and helps minimise monosaccharide production (Hendrickson et al., 2011). Qiao et al. (Qiao et al., 2013) reported that after the hydrothermal pretreatment (170 $^{\circ}\text{C}$ for 1 h) of pig manure, municipal solid waste, cow manure and fruit/vegetable waste, the production of biogas respectively increased by 7.8%, 67.8%, 13.3% and 18.5%, while the methane yield increased by 14.6%, 65.8% and 16.1% for pig manure, municipal solid waste and fruit/vegetable waste, respectively and decreased by 6.9% for cow manure. Also, Passos and Ferrer (Passos and Ferrer, 2015) reported an

increase in methane yield by 17%–39% after the hydrothermal pretreatment of microalgal biomass. Jiang et al. (Jiang et al., 2012), Panigrahi et al. (Panigrahi et al., 2019) and Shang et al. (Shang et al., 2019) respectively reported 31%, 10.8% and 62.9% increment in methane yield obtained from anaerobic digestion of giant reed, yard waste and wheat straw after their prior liquid hot water pretreatment.

(d) Freezing and thawing: This case entails subjecting the feedstock to sub-zero temperature (freezing) and, after that, recovering back its temperature (thawing). Elmashad (Elmashad, 2004) reported a 30% increase in the yield of biogas when cattle manure was pretreated using the freezing and thawing pretreatment method. Although this pretreatment method is adequate and feasible for food waste, it has not been widely used, mainly due to its high operational cost (Morales-Polo, 2018).

1.3.1.3 Microwave irradiation pretreatment

This method directly involves the application of the components of electric and magnetic field to the molecular structure of the feedstock components, eliciting biological, chemical and physical reactions as a result of the heat and extensive collisions by the ion movements and the polar molecules rotations or vibrations (Deepanraj et al., 2013; Peng et al., 2014). Microwave irradiation benefits include lignin depolymerisation, decreased cellulose crystallinity and polymerisation, increased feedstock surface area, and increased enzyme attack accessibility (Diaz et al., 2015; Deepanraj et al., 2017). Microwave irradiation can be used alone or in a combined form with various solvents such as acid, salt, water, alkali, ionic liquid and organic solvents. Carrerre et al. (Carrerre et al., 2016) reported an enhanced yield of biomethane (i.e., about 60% yield increase) from microalgae during its anaerobic digestion in a continuous reactor when the feedstock was subjected alone to microwave irradiation pretreatment. However, there are technical difficulties in applying this method on an industrial scale. Thus to date, its application is limited to the lab scale.

1.3.1.4 Sonication pretreatment

This entails the application of sound waves at high frequencies. In this method, high-intensity ultrasonic waves with a frequency equal to or greater than 20 kHz are applied, which by cavitation of bubbles generates huge hydro-mechanical shear force (Deepanraj et al., 2013). Ultrasound exposure can result in the thinning of the microbial cell walls, leading to the release of cytoplasm and consequently facilitating the solubilization and availability of both extracellular and intracellular matter for further degradation during the AD process (Sarker et al., 2019). In addition, sonication of high solid feedstocks causes disturbance to the acoustic waves' homogeneity, high cavitation and increased dewaterability (Carrerre et al., 2016). The sonication method has been widely utilised for sewage or waste-activated sludge (Martin et al., 2013; Ormaechea et al., 2017), as this feedstock, when compared to other feedstocks, requires lower sonication energy and time. This is because past studies have shown that longer or higher sonication energy and time may lead to a lower efficiency of solubilised matter conversion into biomethane (Carrerre et al., 2016). The specific sonication energy threshold ranges from 1,000 to 16,000 kJ/kg TS (total solids). This is highly correlated to feedstock's solid content

(Carrere et al., 2016). Carrère et al. (Carrère et al., 2010) reported that the biomethane potential (BMP) of waste activated sludge and cattle manure (5.8% total solids) was respectively improved by 140% and 19% when the individual feedstocks were pretreated using the sonication method. Also, it has been reported by Martín et al. (Martín et al., 2013) that the use of sonication (ultrasound) in the pretreatment of sewage sludge increased the yield of biomethane by 95%. This method could be suitable for a wide range of feedstock as it was found to be energetically feasible as the accrued benefit due to the earned revenue from higher production of biogas (35%–55%) overcompensated the investment cost on sonication (Pérez-Elvira et al., 2009).

1.3.1.5 Pressure-depressure (high-pressure homogenizer) pretreatment

Pressure-depressure or high-pressure homogenizer pretreatment involves subjection of the feedstock to high pressure of about 10 bar and thereafter depressurized (known as blasting) to approximately 1 bar. This sudden pressure drop causes strong turbulent eddies and shearing forces to occur, which break up or rupture the cell walls and structure (Morales-Polo, 2018; Mahmudul et al., 2021). Ma et al. (Ma et al., 2011) reported that food waste pretreated with the pressure-depressure treatment method resulted in up to a 35% increase in the yield of biogas.

1.3.2 Chemical technique

The chemical technique comprises the use of alkalis, acids, ammonia, peroxides, ozone and organic solvents (e.g., ethanol, tetrahydrofuran (THF) and N-Methyl morpholine-N-oxide (NMMO)) (Seidl and Goulart, 2016; Morales-Polo, 2018).

1.3.2.1 Acid pretreatment

This method is performed with the use of dilute or strong acids such as hydrochloric acid (HCl), nitric acid (HNO₃), sulphuric acid (H₂SO₄), acetic acid (CH₃COOH), maleic acid and phosphoric acid (H₃PO₄) (Khan et al., 2017). These acids accelerate the solubilisation of hemicellulose into oligomers and lignin removal (Sarker et al., 2019). Pretreatment with strong acids (30%–70%) is usually performed at a temperature that is less than 100°C, whereas pretreatment using dilute acids is done at a temperature range of 100–250°C (Khan et al., 2017). Dilute acids with concentrations less than 4% w/w are commonly used for acid pretreatment, because strong acids cause excessive substrate degradation leading to fermentable sugar loss as well as results in the production of unwanted byproducts like furfural and its derivatives that can greatly inhibit the AD process (Paudel et al., 2017). Biomass pretreated with dilute acids at high temperatures for a very short duration (e.g., 10–30 min) may have no positive effect on biogas or biomethane yield. However, there is a significant positive effect on the yield when the dilute acid pretreatment takes place for a higher duration, such as 1–2 h (Taherdanak et al., 2016). Zhang et al. (Zhang et al., 2011) reported that lignocellulosic waste subjected to acidic pretreatment using H₂SO₄ elicited a 57% increase in the yield of biomethane. Venturin et al. (Venturin et al., 2018) also reported that the use of acid pretreatment on corn stalks resulted in a 31.9% increment in the yield of biogas. Acidic pretreatment is usually not suitable for food waste as it leads to a decrease in the yield of biogas (Ma et al., 2011) due to inhibitor (carboxylic acids and phenolic compounds) accumulation at low pH (Yang et al., 2014). It has been

suggested that inorganic acids have a more positive effect on biogas/ biomethane yield (i.e., more effective) as compared to the use of organic acids (Mankar et al., 2021). For instance, Song et al. (Song et al., 2014) reported that sulfuric acid and hydrochloric acid exhibited a significant improvement or increase in methane yield compared with the use of acetic acid pretreatment, which resulted in a lower increase in the yield of methane.

1.3.2.2 Alkali pretreatment

This method also referred to as saponification, involves the use of sodium hydroxide (NaOH), calcium hydroxide (Ca(OH)₂), ammonium hydroxide (NH₄OH), and potassium hydroxide (KOH). These alkalis assist in the breakdown of lignin alongside hemicellulose solubilization (Carrere et al., 2016; Norraahim et al., 2021). It also helps to accelerate lipid-like feedstocks (Battimelli et al., 2010). The temperature of pretreatment, residence or reaction time and alkali concentration are the basic variables that affect or influence alkali pretreatment (Khan et al., 2017). Alkali pretreatment has demonstrated more effectiveness in enhancing the production of biogas and biomethane when compared with acid pretreatment (Carrere et al., 2016; Ugwu and Enweremadu, 2019). The reason for this is that alkali provides a better condition for the AD process by preventing a drop in the pH (Li et al., 2012). Antonopoulou et al. (Antonopoulou et al., 2020) and Dasgupta and Chandel (Dasgupta and Chandel, 2020) respectively pretreated the grass lawn waste and the organic fraction of municipal solid waste using NaOH as an alkali pretreatment agent, and the results revealed that methane yield was correspondingly 25.7% and 34.8% higher when compared with the untreated during the anaerobic digestion processing. Rani et al. (Rani et al., 2021) also reported an enhanced biomethane production in the AD processing of wheat straw and animal manure when the wheat straw was pretreated with 10% Ca(OH)₂ in comparison to the untreated wheat straw. The workers observed a lower biomethane yield when 15% Ca(OH)₂ was used. The Alkali pretreatment method is still the most attractive and cost-effective method of removing lignin from lignocellulosic feedstocks (Seidl and Goulart, 2016).

1.3.2.3 Oxidation pretreatment

This involves the addition of oxidizing agents like H₂O₂ and peracetic acid (Deepanraj et al., 2013). However, this pretreatment could be either in the form of wet oxidation pretreatment or advanced wet explosion pretreatment (Khan et al., 2017). Wet oxidation pretreatment involves the addition of water to the biomass and then followed by the addition of an oxidizing agent. The pretreatment is exothermic and is often conducted at a higher temperature range of 125–300°C and a higher pressure range of 0.5–20 MPa. Nevertheless, it can also be carried out at a lower temperature. The advanced wet explosion pretreatment is performed at a temperature range of 140–220°C and a pressure range of 0–3.5 MPa. Thus, temperature, pressure, and residence time are the major parameters that influence the wet oxidation and advanced wet explosion pretreatment modes (Khan et al., 2017). The pertinent difference between wet oxidation and an advanced wet explosion is the use of a decompression unit in an advanced wet explosion where the pressure is reduced and results in biomass physical disruption, which has not been utilised in wet oxidation (Biswas et al., 2014). Oxidation pretreatment removes or solubilizes the hemicelluloses and decomposes the lignin and thus increasing

the cellulose accessibility (Deepanraj et al., 2013; Ahring et al., 2015). Caution should be taken in the process of oxidation agents' addition as this can form aromatic compounds that may serve as inhibitors for the AD process. Appels et al. (Appels et al., 2011) reported a 21% increase in biogas yield when waste-activated sludge was pretreated using peracetic acid as an oxidation agent prior to AD processing. Lee et al. (Lee et al., 2020) and Lee et al. (Lee et al., 2021) have both used the wet oxidation method to pretreat oil palm empty fruit bunches prior to anaerobic digestion, and they reported that the wet oxidation pretreatment resulted in 43% and 30% increment in methane yield during mesophilic and thermophilic anaerobic digestion, respectively. Biswas et al. (Biswas et al., 2014) and Ahring et al. (Ahring et al., 2015), using advanced wet explosion pretreatment for digested manure fibres and feedlot manure, showed that the pretreatment significantly and correspondingly enhanced the methane yield by 129% and 357%.

1.3.2.4 Ozone (ozonolysis) pretreatment

Ozonolysis or ozonation pretreatment involves the application of ozone, which is normally conducted at ambient or room temperature and pressure (Deepanraj et al., 2013; Khan et al., 2017). In this method, ozone reacts with the substrates and transforms them into smaller molecular-weight compounds leading to the rupturing of the cellular membrane such that the cell cytoplasm is spilled out (Cesaro and Belgiorno, 2014). If the dose of ozone is sufficiently high, it can lead to the mineralisation of the cellular compounds released (Cesaro and Belgiorno, 2014). The main parameters usually considered in this method, which affects its effectiveness, are ozone concentration in the gas flow, feedstock moisture content, and particle size (Khan et al., 2017). This pretreatment method can partly remove hemicellulose and effectively remove lignin (Taherzadeh and Karimi, 2008). This method does not leave toxic, basic and acidic residues (Deepanraj et al., 2013). Cesaro and Belgiorno (Cesaro and Belgiorno, 2014), using ozonolysis or ozonation pretreatment for organic solid waste before AD, reported that an ozone dose of 0.16 g_{O₃}/g_{TS} improved the production of biogas by 37%.

1.3.3 Biological technique

The biological technique involves the use of biological agents such as enzymes and fungi. The biological technique shows several advantages in lignin removal from lignocellulosic raw materials; however, its low lignin removal efficiency places limitations on its practical application (Seidl and Goulart, 2016).

1.3.3.1 Bacterial enzyme pretreatment

Bacterial enzyme pretreatment involves the application of oxidative and hydrolytic enzymes, which are often produced by bacteria and fungi (Koupaie et al., 2019). For enzyme pretreatment, over forty enzymes have been tested (Koupaie et al., 2019). Among these enzymes, α -amylases, endoglucanase, β -glucosidase cellulases, xylanases, cellobiases, peroxidases, pectinases, proteases, and laccases are some of the enzymes that can be employed for pretreatment (Carrerre et al., 2016). The impact of the enzyme on the feedstock is dependent on the enzyme type and the composition of the feedstock being treated. This is due to the enzyme's specificity in terms of the type of reaction it catalyses (Koupaie et al., 2019). The effectiveness and activities of the

enzyme reaction are influenced by factors such as the substrate or feedstock composition, pH, incubation time, reactor configuration, and temperature (Parawira, 2012; Koupaie et al., 2019). Enzyme degradation of cellulose and lignin during various hydrolysis stages leads to the release or production of oligomers and several monomer sugars (Tanjore and Richard, 2015). According to Quiñones et al. (Quiñones et al., 2009), the pretreatment of solid cattle manure with enzymes prior to anaerobic digestion (AD) resulted in a significant improvement of approximately 105% in methane production. Lin et al. (Lin et al., 2017) found that pretreating pulp and paper sludge with endoglucanase and laccase enzymes before subjecting it to mesophilic anaerobic digestion (AD) led to a significant improvement of 34% in methane production. Similarly, Frigon et al. (Frigon et al., 2012) observed a 29% and 42% increase in methane production when using lignin peroxidase and manganese peroxidase enzymes, respectively, for the enzyme pretreatment of switchgrass.

A variety of enzyme pretreatment approaches have been explored in the literature. Koupaie et al. (Koupaie et al., 2019) have conducted a comprehensive review of this method, demonstrating the impact of over forty enzymes. Their review highlights the potential of this pretreatment method for enhancing biomethane production through mesophilic and thermophilic anaerobic digestion (AD) of lignocellulosic biomass. However, the authors note that further research is needed to better understand the relationship between lignocellulosic substrate composition, enzyme or bacterial type, and optimal pretreatment conditions. The benefits of this pretreatment method include its relatively short reaction time, low nutritional requirements for enzymatic reactions, and the fact that most enzymes are unaffected by other microbial metabolisms and inhibitors (Wei, 2016), as well as not requiring expensive processing equipment (Ometto et al., 2014). However, the major limitations of enzyme pretreatment, which restrict its application, are the high cost of enzymes and their selectivity (Sarker et al., 2019). This highlights the need for further research in the context of AD process optimisation.

1.3.3.2 Fungal pretreatment

The enzymes used for AD processes are usually produced from different forms of fungi like *Aspergillus* and *Trichoderma* genus (Schimpf et al., 2013). Therefore, instead of producing the enzymes from these fungi, the fungi can be directly used as an alternative so that the cost of enzyme production can be totally avoided (Sarker et al., 2019). In addition, these fungi, combined with an auxiliary enzyme such as Lytic polysaccharide mono-oxygenases (LPMO), can be employed to enhance cellulose degradation (Dollhofer et al., 2015). Various fungi, such as soft-rot fungi, white-rot fungi, and brown-rot fungi, are commonly used for lignin and hemicellulose degradations in the feedstock (Deepanraj et al., 2013; Khan et al., 2017; Wagner et al., 2018). Soft and white rot fungi are mainly used to attack lignin and cellulose, while brown-rot fungi are mainly used for cellulose degradation (Deepanraj et al., 2013; Wagner et al., 2018). Muller and Trosch (Muller and Trosch, 1986) biologically pretreated wheat straw using white-rot fungi and reported a double-fold increase in the yield of biogas to the untreated wheat straw.

In addition, Srilatha et al. (Srilatha et al., 1995) reported an increase in the production of biogas when an orange processing waste was biologically pretreated using *Aspergillus*, *Sporotrichum*,

Penicillium and *Fusarium*. Ghosh and Bhattacharyya (Ghosh and Bhattacharyya, 1999) reported that using brown rot and white rot fungi to respectively pre-treat bamboo prior to AD operation resulted in the yield of biomethane respectively, increasing by 32% and 46%. Mackulak et al. (Mackulak et al., 2012) reported a 15% enhancement in methane yield during anaerobic digestion of hay and leaves that were subjected to prior fungal pretreatment using *Auricularia auricula-judan*. Ali et al. (Ali et al., 2014) reported that the fungal pretreatment of water hyacinth and maize cob heart using *Volveria diplasia* and *Phanerochaete chrysosporium* prior to AD processing resulted in an increased yield of biogas at short hydraulic retention time in comparison with the untreated feedstocks. Furthermore, it was reported by Rouches et al. (Rouches et al., 2016) that fungal pretreatment of wheat straw using *Polyporus brumalis* resulted in an increase of biomethane yield by 45%. On the other hand, it was observed and reported by Paul et al. (Paul et al., 2018) that fungal pretreatment of agricultural biomass did not improve biomethane yield. This observed variation is expected as the biological pretreatment technique is still being developed.

1.3.4 Combined or hybrid technique

Any two or three of the physical, chemical and biological techniques of pretreatment can be combined to increase the biogas and biomethane yield as well as lower energy consumption and minimise cost (Shirkavand et al., 2016; Khan et al., 2017; Kumar et al., 2017). For example, physical pretreatment techniques can be combined with chemical pretreatment techniques such as thermo-chemical (i.e., combination of thermal and chemical pretreatments) and Chemo-sonication (combined ultrasonication and chemical). Also, physical pretreatment techniques can be combined with biological pretreatment techniques such as bio-mechanical (i.e., combination of biological and mechanical pretreatments), while physical, chemical and biological pretreatment techniques can be combined together. To this end, different researchers have utilised these combined or hybrid pretreatment methods and have reported positive results (Shirkavand et al., 2016; Khan et al., 2017; Passos et al., 2017).

1.3.4.1 Thermo-chemical pretreatment

The combination of heat (thermal) and chemicals can also be applied as a pretreatment strategy. This method helps to contribute to the enhancement of AD productivity by reducing the particle size of the feedstock (Izumi et al., 2010), increasing the volatile solid reduction (Mladenovska et al., 2006) and increasing the solubility of the chemical oxygen demand (COD) (Valo et al., 2004). The chemicals that are predominantly utilised for thermo-chemical pretreatment are alkali (Lin et al., 2009), acids (Devlin et al., 2011), and ozone (Bougrier et al., 2006). Shehu et al. (Shehu et al., 2012) reported a 36% increase in biogas production when sewage sludge was previously subjected to thermo-alkaline pretreatment using 2.29 M NaOH and 88.50 °C. Also, Passos and Ferrer (Passos and Ferrer, 2015), using the thermal-alkali pretreatment method (10% NaOH at 100 °C for 5 min) on dairy cow manure feedstock, reported a 23.6% enhancement in methane production when compared with the untreated. The use of microwave irradiation in combination with solvent can result in the rapid hydrolysis of more sugars in

the lignocellulosic biomass (Saha et al., 2008) and remove or degrade more hemicellulose and lignin compositions of the lignocellulosic feedstock (Zhu et al., 2006) when compared to the conventional thermal treatment. Kaur and Phutela (Kaur and Phutela, 2016) reported that paddy straw subjected to thermochemical pretreatment using the combination of microwave and sodium hydroxide resulted in 55% increment in methane yield as compared to the untreated control.

1.3.4.2 Electro-chemical pretreatment

The electrochemical technique includes electro-oxidation, electro-deposition, electro-coagulation, and electro-flotation. Yu et al. (Yu et al., 2014) reported a 63.4% increase in biogas yield after the electrochemical pretreatment of waste activated sludge using a pair of Ti/RuO₂ mesh plates as electrodes. Kumar et al. (Kumar et al., 2017) reported an improved methane yield during anaerobic digestion when mixed microalgae biomass was subjected to initial pretreatment using the combination of electrolysis and ultrasonic pretreatment.

1.3.4.3 Chemo-sonication pretreatment

This is the combination of chemical and sonication (ultrasound), which can be employed as a pretreatment method. Wang et al. (Wang et al., 2012) reported an increase in the production of biogas by 67%–76% after the chemo-sonication pretreatment of rice stalks using 2% NaOH and ultrasonication frequency of 30 kHz for 1 h.

1.3.4.4 Thermo-sonication pretreatment

This involves the combination of thermal and sonication. Dhar et al. (Dhar et al., 2012) reported that the thermo-sonication pretreatment of waste-activated sludge using a temperature range of 50–90 °C and sonication energy range of 1,000–10,000 kJ/kg TSS resulted in 30% increment in the production of methane and 29%–38% reduction in volatile solids.

1.3.4.5 Bio-mechanical pretreatment

In this case, it is the combination of biological and mechanical techniques utilised as pretreatment method. The combination of mechanical milling with fungal pretreatment can also be utilised to achieve enhanced biomethane yield (Mustafa et al., 2017). Pérez-Rodríguez et al. (Pérez-Rodríguez et al., 2017), using the combination of extrusion and enzymatic pretreatment for pretreating corn cob, reported a higher production of methane when compared with the untreated corn cob.

1.3.4.6 Mechanical-sonication pretreatment

This is the combination of mechanical and sonication pretreatment methods. Elbeshbisy and Nakhla (Elbeshbisy and Nakhla, 2011) and Cesaro et al. (Cesaro et al., 2012) obtained about 94% increase in the yield of both biogas and biomethane when FW was previously subjected to the combination of grinding (i.e., mechanical) and sonication pretreatment methods.

1.3.4.7 Bio-physico-chemical pretreatment

This involves the combination of biological, physical and chemical pretreatment techniques. Peng et al. (Peng et al., 2014) reported a 280% increase in the production of biogas from oily

wastewater when it was previously pretreated using a bio-physico-chemical pretreatment method (*Bacillus*, sonication and acid).

1.3.5 Evaluation of pretreatment efficiency in biogas production: Energy cost as a criteria

The main purpose or goal of AD is biogas/biomethane or energy production. Evaluation of the efficiency of different biomass or lignocellulosic biomass pretreatments methods can be based on the operational energy costs that are associated with the different pretreatment methods (Ruggeri et al., 2015). These costs can be determined based on the direct energy (heat and electricity) consumed during each process and the indirect energy involved during the pretreatment. To this end, two energy parameters can be utilised to grade the pretreatment methods (Ruggeri et al., 2013a). The first parameter is the energy conversion efficiency (ξ). This parameter evaluates the ratio of energy (biogas/biomethane) produced as biogas/biomethane during pretreatment/bioreaction to the initial amount of available energy contained in the feedstock or biomass. The second parameter is the energy sustainability index (ESI).

1.3.5.1 Energy conversion efficiency

The energy conversion efficiency (ξ) is given as:

$$\xi = \frac{E_{TP}}{E_0} \times 100 \quad (2)$$

Where E_p stands for the total energy produced as biogas or biomethane (kJ/L) and E_0 depicts the initial available energy contained in the feedstock or biomass (kJ/L). The total energy produced (E_p) in terms of biomethane can be obtained using Eq. 3:

$$E_{TP} = \frac{G_{\max}}{22.4 \times LHV_{CH_4}} \quad (3)$$

Where G_{\max} represents the total biomethane volume (NL/L) produced at 1 atm and absolute temperature (0 °C), which is based on the mean biomethane concentration in biogas; 22.4 is the equivalent molar volume (L/mol); and LHV_{CH_4} represents the biomethane's Low Heat Value ($LHV = 800.32$ kJ/mol). In the calculation of E_0 , it is only the energy within the feedstock or biomass waste that the microorganisms can utilize for cellular activity that should be considered (Nelson et al., 2008; Ruggeri et al., 2015), however, this task is difficult. Nevertheless, for the purpose of scoring or ranking the different pretreatment outcomes performed on the feedstock, a good approximation of E_0 could be achieved utilizing the feedstock's LHV. The E_0 can be estimated using Eq. 4:

$$E_0 = LHV_{FS} \times TS_0 \times 10^{-3} \quad (4)$$

Where LHV_{FS} (kJ/kg) stands for the low heat value of feedstock and TS_0 for the initial total solid concentration (g/L) present in the medium. It is essential to note that the inherent energy within the feedstock is not all available for biomethane production. It is only a portion of this inherent energy that can be harvested or obtained as biomethane using AD technology, while a fraction of this energy is released as heat, a fraction is utilized during biological metabolism, a fraction is embedded within the metabolites produced during the AD reaction, and a fraction is stored within the newly divided

microbial cells. It is to be noted that not all carbon-carbon (C-C) or carbon-hydrogen (C-H) bonds present in feedstock or biomass wastes are of the same type and only the bonds of simple molecules such as glucose that can easily be degraded by microorganisms. The second parameter evaluated was the efficacy (η), which relates the efficiency of AD performed after pretreatment to the efficiency of AD performed without pretreatment. η can be calculated as:

$$\eta = \frac{\xi_{PT}}{\xi_{NP}} \quad (5)$$

Where ξ_{PT} represents the efficiency attained with pretreatment and ξ_{NP} represents the efficiency achieved without pretreatment. Thus, η is a ratio of obtained energy after pretreatment process to the energy obtained without the use of pretreatment. $\eta < 1$ signify that pretreatment had a negative influence on efficiency; $\eta = 1$ signify that pretreatment had no effect on the AD performance; and $\eta > 1$ show that pretreatment increased the energy produced as biomethane. In this way, η allows for the comparative ranking or scoring of different pretreatment methods.

1.3.5.2 Energy sustainability index

It is pertinent to verify the energy sustainability of AD technology by comparison of the energy produced in the form of biogas or biomethane with the energy spent or consumed during the AD process. This can be carried out by employing ESI. The ESI can be utilized to evaluate which feedstock pretreatments or biogas upgrade technologies could be most effectively coupled with AD in order to select the most energetically effective techniques of processing the feedstock. The ESI is presented as given in Eq. 6:

$$ESI = \frac{E_{AD}}{E_{TC}} \quad (6)$$

Where E_{AD} represents the energy obtained as biogas or biomethane during AD E_{TC} , represents the total energy consumed or spent on the pretreatment or biogas upgrade. When the value of ESI is greater than one (1), it signifies an increased energy production and the possibility of an energetically sustainable process (Ruggeri et al., 2013b) and when ESI value is less than one (1), it indicates that the process is not energetically sustainable, even when there is an increased energy production (i.e., $\eta > 1$). For each pretreatment or biogas upgrade as the case may be, the direct energy (e.g., heat and electricity) consumed or spent to conduct the pretreatment or biogas upgrade has to be considered plus the energy spent or consumed to produce the chemicals used as the case may be:

$$E_{TC} = E_H + E_{EE} + E_{IE} \quad (7)$$

Where E_H , E_{EE} , and E_{IE} represents heat energy, electrical energy and indirect energy, respectively.

The heat energy can be calculated using Eq. 8:

$$E_H = \rho C_p (T_1 - T_2) \quad (8)$$

Where ρ depicts the mass of feedstock present per unit reactor or digester working volume; C_p stands for the specific heat value; T_1 represents the ambient temperature and T_2 represents the working temperature. The E_{EE} value can be obtained by multiplication of the

equipment power with the duration time utilized for the pretreatment or biogas upgrade as the case may be. The EIE being the energy spent or used to produce materials like chemicals (NaOH, H₂SO₄, HCl and salts) can be evaluated as Global Energy Requirements (GER; kJ/kg) by employing data from the Ecoinvent (2010) database (Frischknecht and Jungbluth, 2010). Hence, E_{IE} values can be gotten by multiplication of the specific GER with the quantity of materials utilized for the pretreatment or biogas upgrade. Ruggeri et al. (2015) employing the energy conversion efficiency and energy sustainability index criteria, evaluated the effects of different pretreatments on energy produced as methane during the AD of olive mill wastewaters (OMWW) and olive pomace (OP). The different pretreatment methods that were evaluated included, basic pretreatment (BP); physical pretreatment using ultrasound (USP); salt pretreatment using FeSO₄ (SP_{FeII}), FeCl₃ (SP_{FeIII}), MnSO₄ (SP_{Mn}) and CaCO₃ (SP_{Ca}); and double combinations of the pretreatments, such as ultrasound plus FeCl₃ (CP_{FeIII}) or ultrasound plus CaCO₃ (CP_{Ca}). Ruggeri et al. (2015) observed and reported that for the goal of methane production, the pretreatments that were most effective included salt pretreatment using CaCO₃ or FeCl₃ addition and ultrasonic pretreatment. The ultrasonic pretreatment had the highest efficacy of (Cesaro and Belgiorno, 2013). However, from the energetic point of view, the ultrasonic pretreatment was found to have an ESI value of 0.09, suggesting that the energy gotten as methane during the AD after the pretreatment is not enough to cover the energy utilized or spent on ultrasonic pretreatment despite its high efficacy. Amongst these three most efficient pretreatment methods, the salt pretreatment method using CaCO₃ was found to be the only method with the highest ESI value of 14, suggesting its capability to result in a positive net energy balance between energy recovered and energy utilized or spent.

1.3.6 Comparison of pretreatment techniques

Comparisons of the pretreatment techniques based on five essential criteria (rate of hydrolysis, toxic inhibitory compounds generation, operational cost, energy requirement, and effectiveness) are presented in Table 3.

1.3.6.1 Hydrolysis rate

From Table 3, among the pretreatment techniques, it is seen that physical techniques suggest to be the fastest. This is more so for microwave (MW) pretreatment (Wu et al., 2015b; Kumar and Sharma, 2017). This is in agreement with Yuan et al. (2014), who posited that short duration time is an advantage of mechanical pretreatment. Conversely, Gumisiriza et al. (2017) reported slow irradiation processes. There are contrasting reports on the effect of chemical pretreatment techniques on the rate of feedstock hydrolysis during AD process. Yuan et al. (2014) and Li et al. (2015) have reported that the chemical pretreatment technique is a fast process, while Kumar and Sharma (2017), Gumisiriza et al. (2017) and Amin et al. (2017) reported that the technique is a slow process. This difference could be due to variations in their chemical pretreatment conditions. For instance, chemical pretreatment's retention time is affected by the temperature of the reaction (Theuretzbacher et al., 2015; Kamusoko et al., 2019). Data on combined methods is rather limited. As presented in Table 3, the major challenge of biological pretreatment is that it is a slow process (Kumar and Sharma, 2017; Gumisiriza et al., 2017; Amin et al., 2017;

Wagner et al., 2018; Den et al., 2018; Kamusoko et al., 2019). Combined or hybrid pretreatment technique has been reported to be a fast process (Kim et al., 2018; Kamusoko et al., 2019).

1.3.6.2 Energy requirement

Table 3 revealed that biological pretreatment techniques demonstrated low energy requirements or cost among the pretreatment techniques. Several studies have indicated that the major merit of biological pretreatment is the possession of low energy cost or requirement (Singh et al., 2014; Zieminski and Kowalska-Wentel, 2017; Kumar and Sharma, 2017; Gumisiriza et al., 2017; Wagner et al., 2018; Den et al., 2018). Biological pretreatment enables savings on chemicals and energy. The physical pretreatment technique has been reported to be a high-energy requirement process (Table 3). That is, the technique is highly energy intensive given the involvement of high temperatures and pressures as in the case of thermal treatment (Mulakhudair et al., 2016; Speda et al., 2017; Gumisiriza et al., 2017; Wagner et al., 2018; Kamusoko et al., 2019). For example, mechanical pretreatment, such as milling, has a high energy requirement (Yuan et al., 2014; Luo et al., 2015; Chandra et al., 2015; Kumar and Sharma, 2017; Baruah et al., 2018). Also, microwave irradiation pretreatment has the challenge of high energy costs (Kostas et al., 2017). In contrast, Wu et al. (2015b), Kumar and Sharma (2017), Kostas et al. (2017), and Kamusoko et al. (2019) reported that the physical pretreatment technique with respect to microwave pretreatment is energy efficient. Nevertheless, due to the reason of high energy requirement, the physical pretreatment technique is considered to be economically not viable for large industrial-scale applications (Zheng et al., 2014; Kamusoko et al., 2019). Furthermore, the chemical pretreatment technique is considered a high-energy cost process (Table 3). This has been reported in several studies (Wikandari et al., 2015; Speda et al., 2017; Rouches et al., 2017; Gumisiriza et al., 2017; Wagner et al., 2018). Combined or hybrid pretreatment technique as seen in Table 3 has been reported to be a high energy-demanding process (Speda et al., 2017; Gumisiriza et al., 2017; Perendeci et al., 2018). On the other hand, Kumar and Sharma (2017), Zhang et al. (2018), and Kim et al. (2018) reported combined or hybrid pretreatment techniques to be a low energy-demanding process. This observed variation may be due to combined or hybrid pretreatment technique complexity; thus, further research will be required.

1.3.6.3 Effectiveness in biogas/biomethane enhancement

Table 3 depicts that among the pretreatment techniques, the chemical pretreatment technique suggest to be the most effective technique. Chemical pretreatment technique in terms of acid, alkali, oxidation and ozonolysis pretreatments has been reported to be a very effective technique for biogas/biomethane yield enhancement (Appels et al., 2011; Zhang et al., 2011; Biswas et al., 2014; Cesaro and Belgiorno, 2014; Ahring et al., 2015; Venturin et al., 2018; Antonopoulou et al., 2020; Dasgupta and Chandel, 2020; Lee et al., 2020). Physical pretreatment technique with respect to mechanical (milling), thermal (temperature-induced, steam explosion, liquid hot water), microwave irradiation, sonication and high-pressure homogenizer pretreatments have also been observed and reported to be moderately effective in the increase of biogas/biomethane yield (Hjorth et al., 2011; Pilarski et al., 2016; Dell'Omo and Froschia, 2018; Pengyu et al., 2017; M'onch-Tegeder et al., 2014; Marañón et al., 2012; Bougrier et al., 2006; Ariunbaatar et al., 2014; Schwede et al.,

TABLE 3 Comparison of pretreatment techniques (Sibiya et al., 2018).

Pretreatment techniques	Criteria for comparison				
	Hydrolysis rate	Inhibitory compounds generation	Operational cost	Energy cost	Effectiveness in biogas/Biomethane enhancement
Physical:					
(a) Mechanical (milling)	Very fast	No	Very expensive	Very high	Moderate
(b) Thermal					
(i) Temperature-induced	Very fast	Yes	Very expensive		
(ii) Steam explosion	Very fast	Yes	Very expensive	Low	
(iii) Liquid hot water	Very fast	Yes	Very expensive		
(iv) Freezing and thawing	Very fast	Yes	Very expensive		
(c) Microwave irradiation	Very fast	Yes	Very expensive		
(d) Sonication	Very fast	Yes	Very expensive		
(e) Pressure-depressure			Very expensive		
Chemical:					
(a) Acid	Fast	Yes	Very expensive	High	Very effective
(b) Alkali	Fast	Yes	Cost-effective	High	Very effective
(c) Oxidation	Fast	No	Very expensive	High	Very effective
(d) Ozone or Ozonolysis	Fast	No & Yes	Very expensive	High	Very effective
Biological:					
(a) Enzyme	Fast	No	Very expensive	Very low	Effective
(b) Fungal	Slow	No	Cost-effective	Very low	Less effective
Combined or Hybrid	Fast	Yes	Cost-effective	Moderate	Effective

2013; Menardo et al., 2012; Horn et al., 2011; Li et al., 2016; Lizasoain et al., 2017; Mulat et al., 2018; Steinbach et al., 2019; Weber et al., 2020; Passos and Ferrer, 2015; Qiao et al., 2013; Jiang et al., 2012; Panigrahi et al., 2019; Shang et al., 2019; Carrerre et al., 2016; Martin et al., 2013; Carrère et al., 2010; Ma et al., 2011). Table 3 information show that the biological pretreatment technique in terms of enzyme and fungal pretreatments has been reported to be effective in enhancing biogas/biomethane yield as the other techniques (Muller and Trosch, 1986; Srilatha et al., 1995; Ghosh and Bhattacharyya, 1999; Quiñones et al., 2009; Mackulak et al., 2012; Ali et al., 2014; Tanjore and Richard, 2015; Rouches et al., 2016; Lin et al., 2017). However, fungal pretreatment has been reported to be less effective in comparison with other pretreatment techniques (Kamusoko et al., 2019). Combined or hybrid pretreatment technique with respect to thermo-chemical, electro-chemical, chemo-sonication, thermo-sonication, bio-mechanical, mechanical-sonication and bio-physico-chemical pretreatments has been found and reported to be effective in the enhancement of biogas/biomethane production (Elbeshbishy and Nakhla, 2011; Cesaro et al., 2012; Dhar et al., 2012; Shehu et al., 2012; Wang et al., 2012; Peng et al., 2014; Yu et al., 2014; Passos and Ferrer, 2015; Kaur and Phutela, 2016; Kumar et al., 2017; Pérez-Rodríguez et al., 2017).

1.3.6.4 Operational cost

Table 3 data compares the different pretreatment techniques in terms of operational cost. Despite the report that the biological pretreatment technique in terms of fungal pretreatment is less effective, fungal pretreatment is considered to be cost-effective in terms of operational cost (Rouches et al., 2017). Biological pretreatment techniques have been reported not to be expensive (Mulakhudair et al., 2016; Gumisiriza et al., 2017; Wagner et al., 2018; Den et al., 2018). This is due to the fact that fungi helps to lessen the pretreatment steps and costs by avoidance of the enzyme recovery step (Carrerre et al., 2016). It has been reported that enzyme pretreatment possesses low enzyme utility costs due to the utilisation of mild conditions (Kudanga and Le Roes-Hill, 2014). On the contrary, the biological pretreatment technique in terms of enzyme pretreatment using commercial enzymes has been reported to be costly with respect to production cost (Mulakhudair et al., 2016; Kamusoko et al., 2019). It is noted, according to the information in Table 3, that physical and chemical pretreatment techniques are predominantly very expensive. Selected works or studies from the literature revealed that physical or chemical pretreatment techniques are expensive and thus not economically applicable for biogas/biomethane generation from agricultural

biomass (Kamusoko et al., 2019). Physical pretreatment techniques are more expensive because of higher capital and energy costs (Sadhukhan et al., 2018). Mechanical pretreatment (such as milling and grinding), microwave irradiation pretreatment, and sonication (ultrasonic) pretreatment have high equipment maintenance and energy costs (Kumar and Sharma, 2017; Amin et al., 2017; Ismail et al., 2017; Gumisiriza et al., 2017; Kamusoko et al., 2019). Also, the reasons for chemical pretreatment techniques' expensive status were observed to be due to the followings; (i) use of expensive auxiliary equipment (Speda et al., 2017; Kamusoko et al., 2019), (ii) utilization of chemicals and downstream processing that incurs high cost (Sträuber et al., 2015; Kumar and Sharma, 2017; Amin et al., 2017; Sadhukhan et al., 2018; Den et al., 2018), (iii) high operational and maintenance costs (Rouches et al., 2017), and (iv) high costs of digestion residues' disposal (Wagner et al., 2018). Nevertheless, some studies have reported that chemical pretreatment technique with regard to alkali pretreatment is not expensive (Song et al., 2014; Kumar and Sharma, 2017; Amin et al., 2017; Ismail et al., 2017; Gumisiriza et al., 2017). Thus, alkali pretreatment will be favoured over the other chemical pretreatment techniques as a result of low operational costs (Ismail et al., 2017). Combined or hybrid pretreatment techniques would be influenced or affected by the associated costs of the other pretreatment techniques.

1.3.6.5 Generation of inhibitory compounds

Table 3 provides information on the comparison of the pretreatment techniques with regard to the generation of inhibitory compounds. Table 3 information reveals that physical pretreatment techniques generally result in the generation of inhibitory compounds. For example, thermal pretreatment at a temperature that is above 160 °C could result in partial degradation of lignin and polysaccharides to generate heterocyclic and phenolic compounds (Zieminski and Kowalska-Wentel, 2017). However, some other studies show that mechanical pretreatment, such as milling, do not produce inhibitory compounds (Baeta et al., 2016; Kumar and Sharma, 2017). It is presented in Table 3 that chemical pretreatment techniques with respect to acid and alkali pretreatments generate inhibitory compounds, while chemical pretreatment techniques with regard to oxidation and ozonolysis pretreatments do not generate inhibitory compounds. There are about eleven research studies in the literature that reported that chemical pretreatment techniques lead to the generation of toxic inhibitory compounds (Pei et al., 2014; Kudanga and Le Roes-Hill, 2014; Amin et al., 2017; Speda et al., 2017; Eskicioglu et al., 2017; Rouches et al., 2017; Kumar and Sharma, 2017; Gumisiriza et al., 2017; Nair et al., 2018; Den et al., 2018; Paul et al., 2018). Thus, inhibitory compounds generation is one of the challenges or demerits of chemical pretreatment techniques. Meanwhile, there are some chemical pretreatment techniques that do not generate inhibitory compounds (Sahito and Mahar, 2014; Gumisiriza et al., 2017; Paul et al., 2018; Den et al., 2018). Furthermore, biological pretreatment techniques concerning enzyme and fungal pretreatments do not result in the formation of inhibitory compounds (Singh et al., 2014; Amin et al., 2017; Wagner et al., 2018), as presented in Table 3. This is one of the positive attributes or advantages of biological pretreatment techniques. Table 3 also shows that combined or hybrid pretreatment techniques lead to inhibitory toxic compound generation. There are seven research works in the literature that have shown that combined or hybrid pretreatment techniques lead to the formation of inhibitory compounds (Speda et al., 2017; Amin et al., 2017;

Gumisiriza et al., 2017; Eskicioglu et al., 2017; Kumar and Sharma, 2017; Zhang et al., 2018).

1.4 Summary of comparison

Therefore, in summary, biological pretreatment techniques possessed more techno-economic merits across the five criteria compared to other pretreatment techniques. The merits of biological pretreatment techniques are associated with the capacity to prevent the generation of by-products that are inhibitory and toxic to methanogens, low operation cost, and low energy requirement. Nevertheless, there is a need to improve the biological pretreatment techniques' efficacy, and the main focus should be the rate of hydrolysis enhancement. In order to enhance or improve on the rate of hydrolysis, it is therefore recommended that parameters such as type of microorganisms and enzymes involved, pH, incubation temperature, incubation time, moisture content, inoculum concentration, aeration rate, and nature and composition of biomass should be optimised. Despite the high effectiveness of both physical and chemical pretreatment techniques, their major limitations or challenges are high energy requirements and cost.

1.5 Use of additives

Different additives have been employed to enhance biogas and biomethane production during the AD processing of raw materials (Romero-Güiza et al., 2016) as schematically represented in Figure 6. These additives include (i) biological additives, (ii) inorganic additives, (iii) bioaugmentation (use of inoculum or seeding bacteria), (iv) biostimulation (use of macro and micro nutrients), (v) use of adsorbent and (vi) use of co-digestion.

1.5.1 Biological additives

1.5.1.1 Bioaugmentation: Use of inoculum or use of seeding bacteria or starter bacteria

The inoculum-to-substrate (feedstock) ratio (I:S or I:F) or substrate (feedstock) to inoculum ratio (S:I or F:I) is one of the key parameters of AD. AD is a process that is performed by microorganisms. Hence there is a basic correlation that exists between the composition of the microbial community and the AD systems' behaviour (Kim et al., 2014; Mata-Alvarez et al., 2014). That is AD or biomethanation results from the interaction of various groups of microorganisms and substrates, which result in methane production. These microbes exist in nature and enter the reactor with the feeding of raw materials (Christy et al., 2013). If the fresh raw material enters the digester with only a few starting-bacteria, the fermentation period will be long, and the AD process will be slow (Palatsi et al., 2011). This is so because higher biogas and biomethane production requires a large amount of starting bacteria (Comparetti et al., 2013). Thus, the startup of the AD process has clearly been identified as the most crucial AD phase (Romero-Güiza et al., 2014) because, during this period, a proper microbial community is yet to be established. For this purpose, bioaugmentation is applied to manipulate and enhance the microbial community during the initial stages of AD such that the AD process performance can be improved or enhanced (Dhadse et al., 2012; St-Pierre and Wright, 2013). That is, the loading of inoculum or seeding bacteria (bioaugmentation) into the biodigester or bioreactor at

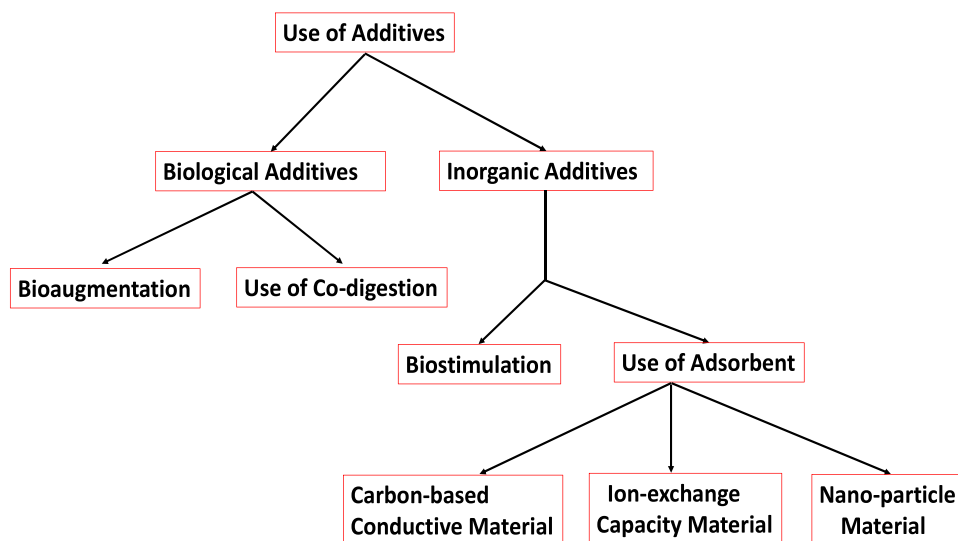


FIGURE 6

Various additives used in biogas production.

the start of the AD process is a means of accelerating or enhancing the start-up period as well as an effective way of providing the significant microbial population to the feedstock (substrate) (Di Maria et al., 2013).

Although the introduction or addition of inoculum reduces the space available for AD, it helps control the retention time by shortening the retention and total reaction time and improving or increasing biogas and biomethane yield (Rocamora et al., 2020) and reducing and reducing the problem of ammonia inhibition (Yang et al., 2022). On the other hand, reducing or decreasing the quantity of start-up inoculum increases the AD process capacity, but it will lead to longer retention and total reaction times as well as a reduction in the yield of biogas and biomethane (Rocamora et al., 2020). The materials that can be utilised as inoculum include pre-rotted raw material, digestate obtained from a previous AD, wastewater from sewage sludge (Yadvika et al., 2004; Karthikeyan and Visvanathan, 2013), chicken-gizzard rumen (Aworanti et al., 2017a) and well-constructed bacteria consortia (Yang et al., 2022). Bioaugmentation is also applied as a response to stress conditions or when microbial community shift occurs during transitional phases (Ács et al., 2013) to increase AD performance or recover the system's efficiency (Kim et al., 2013). Bioaugmentation should be undertaken by selecting the appropriate inoculum and dosage based on the specific microbial community diversities present within the bioreactor/biodigester (Romero-Güiza et al., 2016). Microbial cultures immobilised on a support matrix (i.e., immobilised biofilm or microbial culture) can also be utilised for bioaugmentation. That is, bioaugmentation success can further be improved by the use of immobilized microbial cultures (Youngsukkasem et al., 2011).

1.5.1.2 Use of co-digestion/co-fermentation of different substrates

The anaerobic mono-digestion process is faced with the challenges of a slow start-up, relatively long or high retention time (about 20–50 days) and overall low degradation efficiency (about 20%–50%) (Shehu et al., 2012; Chen X. et al., 2014).

Therefore, there is a need to enhance or improve the overall AD process efficiency in the biogas and biomethane digesters. This enhancement or improvement can be achieved through the application of anaerobic co-digestion (AnCo-D). AnCo-D can be defined as the biotreatment of a mixture of at least two different feedstock or substrates anaerobically (Neczaj et al., 2012). AnCo-D of different feedstock or biomass wastes has been reported to give better performance or improve or enhance the cumulative biogas and biomethane yield than mono-digestion of separate feedstock (Umar et al., 2013; Hassan, 2014; Ogunleye et al., 2016; Haosagul et al., 2019; Liew et al., 2021; Rani et al., 2022). Co-digestion also improves the chances of the process of handling substrates containing toxic (poisonous) components. Thus, AnCo-D offers many benefits over mono-digestion of separate or individual feedstock, which are, increased substrate or feedstock biodegradation, increased biogas and biomethane production and increased cost-efficiency (Neczaj et al., 2012). In addition, co-digestion can bring about C: N ratio adjustment or enhancement together with improved pH buffering (Zhang et al., 2016; Sarker et al., 2019) as well as the improvement of the overall macronutrients balance by proportionally adjusting the carbon, sulphur, phosphorous and nitrogen content (Sarker et al., 2019). Therefore, a variety of substrates is desirable, as it increases the likelihood of a stable and robust process. There are a lot of reports in the literature dealing with the use of co-digestion or co-substrate fermentation making use of different feedstock or substrates as co-substrates. For example, the use of sewage sludge co-digested with other substrates like agriculture wastes (Rughoonundun et al., 2012), municipal solid waste (Lebiocka and Piotrowicz, 2012), cattle manure (Hassan, 2014) and chicken manure (Mahmoud et al., 2022) the use of animal wastes (cattle manure/dung, cow dung, pig/swine manure, chicken dung) co-digested with other substrates such as food waste, fruit waste, vegetable waste, grass clippings, municipal sewage sludge, wheat straw, oil palm fronds, sorghum stem and banana peels (Ossai, 2013; Prakash and Singh, 2013;

Latinwo and Agarry, 2015; Matheri, 2016; Zhang et al., 2016; Aworanti et al., 2017a; Achinas et al., 2019; Ahlberg-Eliasson and Westerholm, 2021; Rani et al., 2021; Rani et al., 2022). Co-digestion of many other substrates has been documented in a comprehensive review by Poulsen et al. (Poulsen and Adelard, 2016).

1.5.2 Inorganic additives

1.5.2.1 Biostimulation: Addition of macro and micro-nutrients (trace metals)

The addition of certain metals like Ca^{2+} , Fe^{2+} , Ni^{2+} , Zn^{2+} , Mg^{2+} , Cu^{2+} , Cd^{2+} , Co^{2+} , and W^{6+} (Facchin et al., 2013; Jagadabhi et al., 2019) as well as zero-valent iron (Dykstra and Pavlostathis, 2017) at specific concentrations, enhances the biogas yield from AD. This is because enzymes responsible for the biochemical activities of bacteria are chemically associated or linked with metals (metallo-enzymes). By augmenting with appropriate amounts of metals or trace elements, the appropriate amounts of nutrients in the biodigester will be maintained, thereby leading to an enhanced feedstock degradation and, consequently, increased yield of biogas and biomethane (Jiang et al., 2012). For instance, the enzyme called acetyl-CoA decarboxylase/synthase plays an essential role in the conversion of acetate to methane, and this can further be accelerated by the addition of nickel metal (Ni^{2+}) (Funk et al., 2004). Kumar et al. (Kumar et al., 2006) reported that in the mesophilic AD processing of potato waste and cattle manure (50:50), by supplementing with heavy metals such as Ni^{2+} , Zn^{2+} , and Cd^{2+} at a low concentration of 2.5 mg/L, the production of biogas was enhanced. Climenhaga and Banks (Climenhaga and Banks, 2008) also reported an increase in the yield of biomethane after the addition of Fe, Zn, Mn, Cu, and Mo into the biodigester in the AD processing of catering waste. In addition, Dykstra and Pavlostathis (Dykstra and Pavlostathis, 2017) reported a 123%–231% increase in total biomethane generation as a result of the biodigester amendment with zero-valent iron (ZVI). However, to prevent AD process collapse or failure, the addition of the excess amount of heavy metals should be avoided as it can reduce the biochemical activity of the methanogens (Kumar et al., 2006). Hence, the rightful amount of metal augmentation is essential.

1.5.2.2 Use of adsorbents

1.5.2.2.1 Carbon-based conductive material. The accumulation of toxic and inhibitory compounds during AD processing of organic wastes may cause a serious decrease in the yield of biogas and biomethane, thereby compromising the feasibility of the biogas plant. The application of carbon-based conductive materials or adsorbents can help mitigate or overcome the accumulation of inhibitory compounds such that there can be an improvement or enhancement in the production of biogas and biomethane (Dang et al., 2016). Using carbon-based conductive materials or adsorbents in AD to avoid the inhibition associated with high ammonia and volatile fatty acids levels during the process to enhance the AD performance or biogas/biomethane production has been widely studied (Chen S. et al., 2014; Xu et al., 2015; Lee et al., 2016; Zhang et al., 2020). These studies all reported that conductive carbon-based materials could enhance methane production from diverse, complex organic substrates, permit higher OLR and promote rapid recovery of soured reactors. Some of the conductive carbon-based materials that can be utilised include carbon cloth (Chen S. et al., 2014), activated carbon (powdered or granular) (Lee et al., 2016; Dastyar et al., 2021), and biochar (Zhang et al., 2020). The use of these carbon-based conductive materials facilitates direct

interspecies electron transfer (DIET) (Lee et al., 2016), methanogens enrichment and accelerates the start-up of methanogenesis (Xu et al., 2015). Furthermore, these conductive carbon-based materials can be doped with magnetite to achieve enhanced biogas/biomethane production. Recently, Barua et al. (Barua et al., 2019) reported a high-performance AD with the use of magnetite doped granular activated carbon.

1.5.2.2.2 Ion-exchange capacity material. Materials with ion exchange capability, like zeolites and bentonites as well as inorganic materials such as clay and manganese oxides, can also be utilised as support matrix to retain the biomass (i.e., immobilization) in the bioreactor or biodigester (Montalvo et al., 2012; Wang et al., 2012). This biomass immobilization allows the bioreactor or biodigester to operate at a lower HRT and higher OLR (Montalvo et al., 2012), thus improving or enhancing the AD process (Wang et al., 2012). In addition, zeolites can be modified to increase their ionic exchange capacity and to supply micronutrients (e.g., Ni, Co, Mg) (Milán et al., 2003). However, high amounts of zeolite should be avoided as it can lead to toxicity due to heavy metal accumulation (Montalvo et al., 2005). Yin et al. (Yin et al., 2019) reported a 79.4% increment in biomethane production from the AD of wasted activated sludge when the AD process was amended with ash. The ash amendment facilitated the sequestration of CO_2 as calcite leading to a high percentage of hydrogen in the biogas.

1.5.2.2.3 Nanoparticle materials. In recent times, the use of nanoparticle materials has demonstrated its benefits in promoting substrate degradation and enhancing AD performance as well as reducing the bioconversion period (Ellacuriaga et al., 2021). This is because of their unique characteristics of high surface area, a high number of active sites, specificity and high reactivity (Baniamerian et al., 2019). The presence of the nanoparticle materials in the AD digester or bioreactor, through its metal components' bioavailability that is essential for enzymatic reactions, stimulates microbial activity and thus enhances or accelerates cellular growth (Abdelsalam and Samer, 2019). The review of the nanoparticles mechanism and its effect on AD process has been presented by both Abdelsalam and Samer (Abdelsalam and Samer, 2019) and Faisal et al. (Faisal et al., 2019). These Nanoparticle materials enhance direct interspecies electron transfer and interspecies hydrogen transfer (Li et al., 2019). Some of the nanoparticle materials utilised include metal nanoparticles (iron, copper, cobalt, silver and nickel) (Casals et al., 2014; Zaidi et al., 2018; Abdallah et al., 2019; Grosse et al., 2021; Hassaan et al., 2021). Casals et al. (Casals et al., 2014) and Abdelwahab et al. (Abdelwahab et al., 2020) have both reported increased biomethane/biogas production with low H_2S when the AD process was supplemented with iron nanoparticles (NPs), while Farghali et al. (Farghali et al., 2019) also reported a twofold biogas increase and a decreased H_2S production when the AD process was amended with the addition of titanium dioxide (TiO_2) and iron oxide (Fe_2O_3) nanoparticles.

1.6 Use of improved reactor configuration and operation conditions

The manner of feedstock conversion during anaerobic digestion is also dependent on the reactor configuration and its operational

parameters (Sarker et al., 2019). A wide range of reactor configurations and designs include anaerobic baffled stacking reactor, anaerobic contact process, anaerobic filter, anaerobic fluidized bed reactor, etc. To obtain a higher biogas and biomethane yield, several modifications of reactor systems have been suggested, including the use of two-stage or multi-stage reactor systems (Voelklein et al., 2017; Zhang et al., 2017) and temperature-phased anaerobic digestion reactor (TPAD), where the thermophilic stage is followed by the mesophilic stage (Wu et al., 2015a). Also, the performance of reactors with respect to biogas and biomethane production or yield is linked to operational factors such as the pH, stirring or agitation, temperature, hydraulic retention time (HRT), organic loading rate (OLR), total solid content (TSC), carbon/nitrogen ratio, etc., (Rocamora et al., 2020; Uddin et al., 2021). Any form of modification or drastic change that might be made to these factors can lead to changes in the surrounding environment of the microbial community and movement within the bioreactor. Hence, to optimize biogas and biomethane production, it is essential to control these factors. These factors can be manipulated within an appropriate and acceptable range so that the AD bioreactor can run optimally and efficiently.

1.7 Biogas purification and upgrading technologies

Aside from CH₄, the rest of the gases (CO₂, H₂S, NH₃, N₂ and the water vapour (H₂O) present in the biogas are considered biogas impurities or pollutants (Tippayawong et al., 2010; Angelidaki et al., 2018), which decrease or reduce the heat or specific calorific value of biogas (Tippayawong et al., 2010). That is, the concentration or amount of CO₂ and H₂S as well as other impurities that are mixed with the biogas, determines the heating effectiveness or specific calorific value of the biogas or biomethane. The higher the amount or concentration of these impurities present in the biogas or mixed with the biomethane, the lower the heat effectiveness and specific calorific value and *vice versa*. In addition, these impurities or pollutants like water vapour can lead to corrosion of the energy conversion equipment, and similarly, NH₃ and H₂S are both dangerous as they are toxic and corrosive. NH₃, during combustion, generates products that are corrosive, which can cause damage to engines and biogas pipelines (Muñoz et al., 2015). Also, H₂S can result in the emission of sulphur dioxide from combustion and can as well reacts with condensate water to form sulphuric acid, which can cause corrosion of gas storage tanks, gas pipelines, engine parts, compressors and damage the combined heat and power unit (Angelidaki et al., 2018; Muñoz et al., 2015; Domingues et al., 2021). CO₂ does not contribute to the combustion. It only lowers the heating value of biogas and increases the compression and transportation costs (Ghatak and Mahanta, 2016). Siloxanes can form sticky silicone oxide (SiO₂) deposits in process equipment and combustion engines that could result in their malfunctioning and damage (Angelidaki et al., 2018; Santiago et al., 2020). It is, therefore, imperative to reduce the amount of impurities, especially CO₂ and H₂S, to a minimal level or remove them completely to concentrate the biomethane content and improve its quality (Tippayawong et al., 2010; Hosseinipour and Mehrpooya, 2019). This implies that for biogas to be used as a fuel or viable alternative energy, it has to be purified, sweetened or upgraded. Therefore, the upgrading or sweetening process involves firstly, the

removal of the toxic and harmful impurities or pollutants (such as H₂S, NH₃, N₂, etc.) referred to as “biogas cleaning” (Angelidaki et al., 2018) and secondly, the removal of CO₂ (Kougias et al., 2017; Angelidaki et al., 2018; Moya et al., 2022) or its conversion to CH₄ by reacting with hydrogen (H₂) (Kougias et al., 2017). That is, the upgrading is performed by separating CH₄ from other impurities through the treatment and purification of biogas (Domingues et al., 2021). The major purpose of upgrading biogas is to concentrate the biomethane content to an optimum or maximum level of approximately 95%–99%, to maximise its heating effectiveness or calorific value similar to natural gas (Moya et al., 2022). When biogas is upgraded to pure or clean bio-methane with a methane content of approximately 98%, the biomethane has the same properties as natural gas (Wellinger et al., 2013; Hosseinipour and Mehrpooya, 2019) and can thus be compressed (Bio-CNG) or liquefied (Bio-LNG) (Domingues et al., 2021). Nonetheless, the ultimate or final use of biogas determines its final composition and the type of upgrading technique required. For example, for its use in boilers to generate heat, it only requires the removal of water vapour and H₂S removal (<1,000 ppm) prior to combustion (Bailo 'n, 2012). Biogas utilisation in ICE (internal combustion engines) for the generation of combined heat and power (CHP), requires the removal of H₂S, siloxanes NH₃, halogenated compounds and water levels below 200–1,000 ppm, 5–28, 32–50, and 65–100 mgm⁻³, respectively, which dependent on the manufacturer's specification; whereas for its use in turbines for CHP requires very low amounts of siloxane (<0.1 ppm) and water while it is able to withstand high H₂S (10,000–70,000 ppm) and halogenated carbon (200–1,500 ppm) concentrations (Muñoz et al., 2015; Bailo 'n, 2012; Soreanu et al., 2011). However, biogas used as a vehicle fuel and for its injection into natural gas grids requires the most stringent quality, which usually demands that the biogas should possess CH₄ concentrations (80%–96%), CO₂ (2%–3%), NH₃ (3–20 mgm⁻³), H₂S (5 mgm⁻³), O₂ (0.2%–0.5%) and siloxanes (5–10 mgm⁻³) (Muñoz et al., 2015). Several authors have studied the technologies of upgrading biogas (Sun et al., 2015; Kadam and Panwar, 2017; Khan et al., 2017; Baena-Moreno et al., 2019a). From the studies, the most known technologies for biogas upgrading are absorption technology, adsorption technology, membrane separation technology, cryogenic separation technology and biological technology. Figure 7 shows the schematic representation of the different biogas purification and upgrade technologies. The following sections discuss the biogas purification and upgrade technologies enumerated in Figure 7.

1.8 Absorption scrubbing technology

This method can be subdivided into physical absorption scrubbing and chemical absorption scrubbing.

1.8.1 Chemical absorption scrubbing

This is a method, as shown in Figure 8, that uses aqueous organic or inorganic compounds as absorption scrubbers or agents to capture and remove impurities or pollutants such as CO₂ and H₂S from gases (Ryckebosch et al., 2011; Struk et al., 2020). That is, it is a method that involves the chemical binding of CO₂ and H₂S

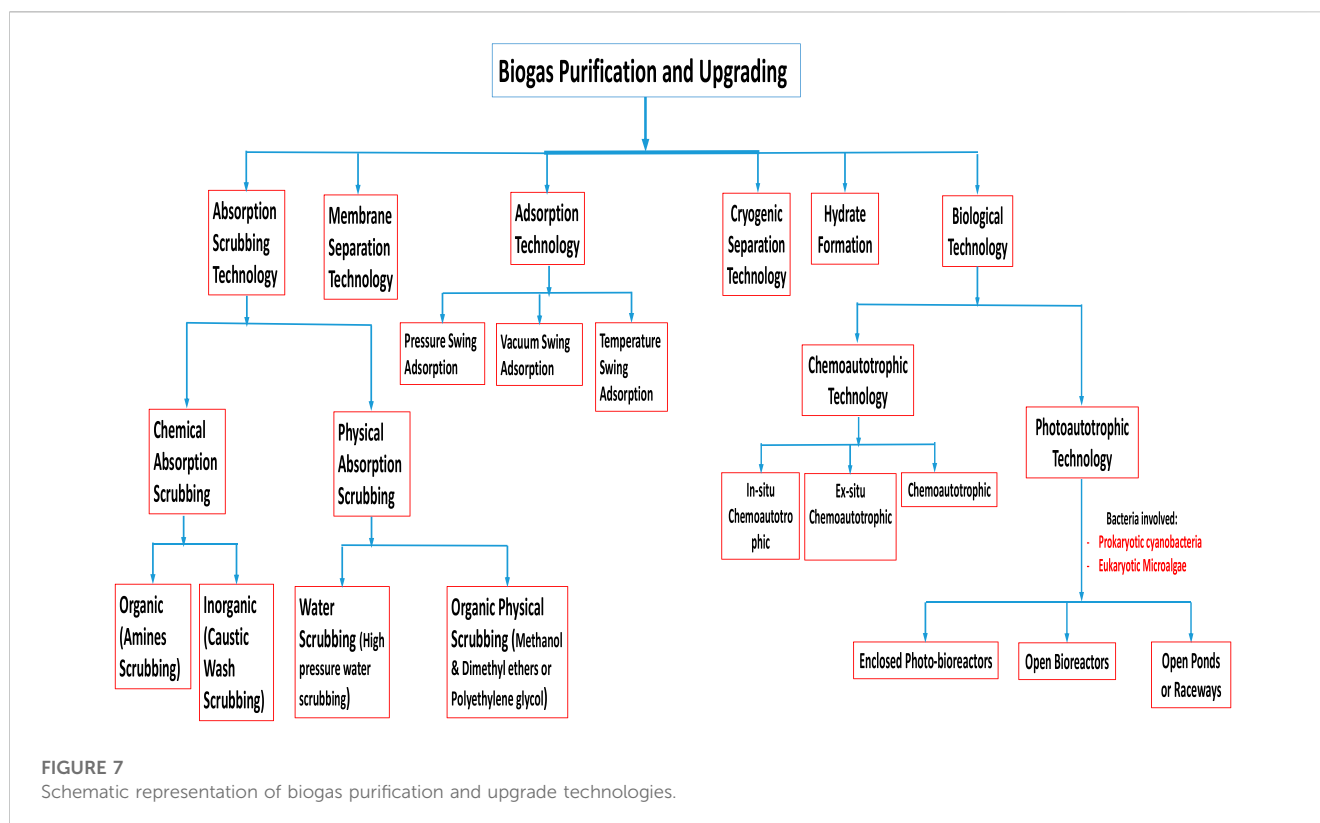


FIGURE 7
Schematic representation of biogas purification and upgrade technologies.

to an organic (Wellinger et al., 2013) or inorganic scrubbing agent (Struk et al., 2020). The method that involves the utilisation of organic absorption scrubbers is referred to as organic-chemical absorption scrubbing, while the one that entails the use of inorganic-chemical absorption scrubbers is called inorganic-chemical absorption scrubbing.

1.8.1.1 Organic-chemical absorption scrubbing

In this method, the most widely used organic scrubbers are the aqueous solutions of amines and alkyl-amines such as monoethanolamine (MEA), aminoethoxyethanol (DGA), diethanolamine (DEA), diisopropanolamine (DIPA), and methyl diethanolamine (MDEA) (Struk et al., 2020). The use of amines for scrubbing can also be referred to as the amine scrubbing method (Kadam and Panwar, 2017; Hosseinipour and Mehrpooya, 2019). These scrubbing systems or absorption columns normally consist of absorber and stripper units. In the absorber unit, the biogas is first compressed or pressurized at a range of 1–2 bars and then injected into the unit from the bottom while the amine solution is made to flow into the unit from the top countercurrent to the biogas (Vega et al., 2017a; Struk et al., 2020). There occurs between the pollutants or biogas impurities (CO_2 and H_2S) and the amine solution reversible exothermic reaction. The resultant amine solution, which now contains CO_2 and H_2S , flows to the stripping column for regeneration, which operates at a pressure range of 1.5–3 bars and a temperature range of 120–160°C (Struk et al., 2020). It is often recommended that H_2S be removed first before using this method for CO_2 removal (Zhou et al., 2017). This is because the presence of H_2S can cause amine poisoning and increase the energy requirement

(Struk et al., 2020). The advantages of this amine scrubbing method include high amine selectivity by CO_2 , no loss of CH_4 and high volume reduction in comparison with other methods (Struk et al., 2020). On the other hand, the drawbacks are the high cost of the amine solvents, increased operational costs due to amine loss caused by evaporation as a result of its high volatility and high energy (i.e., high-temperature heat) requirements for regeneration of solvent (Kadam and Panwar, 2017; Khan et al., 2017; Struk et al., 2020). Organic-chemical absorption scrubbing systems using amine solutions can produce biogas with a final CH_4 content of 99% purity (Abdeen et al., 2016; Struk et al., 2020).

1.8.1.2 Inorganic-chemical absorption scrubbing (caustic wash scrubbing)

The inorganic scrubbers that are mostly used for CO_2 removal are an aqueous solution of alkaline salts (e.g., calcium, potassium and sodium hydroxides) (Kougias et al., 2017). This method of using hydroxides or caustic solutions for CO_2 removal can also be referred to as the caustic wash scrubbing method (Hosseinipour and Mehrpooya, 2019). This method utilizes a countercurrent biogas stream contacting with the aqueous caustic solution in a tray or packed column (Hosseinipour and Mehrpooya, 2019). The inorganic-chemical absorption scrubbing systems using alkaline solutions can result in biogas that has a 94%–97% of final CH_4 content (Abdeen et al., 2016; Struk et al., 2020). The main advantages of sodium and potassium hydroxides, when compared with other solvents, are their greater availability, the possibility of forming a non-polluting by-product and the theoretical capture capacity (Kougias et al., 2010; Yoo et al., 2013). KOH is more

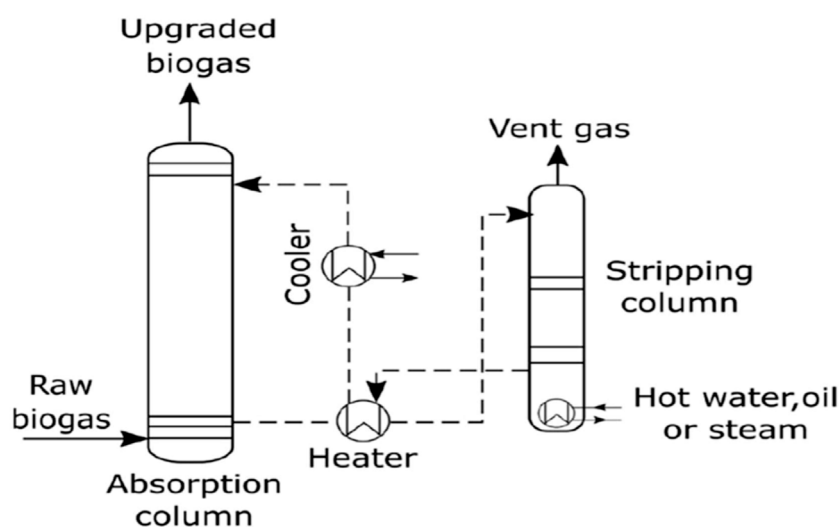


FIGURE 8

Chemical absorption scrubbing system (Adapted from Muñoz et al., 2015; Struk et al., 2020).

expensive than NaOH (Lombardi and Francini, 2020). The disadvantage of the caustic wash method is that it involves high consumption of caustic (hydroxides) solutions due to the formation of bicarbonates (e.g., Na_2CO_3) that cannot be regenerated, leading to the solution being discarded (Hosseinipour and Mehrpooya, 2019). To resolve the issues of solvent loss and energy requirements for solvent regeneration, researchers have reported the use of new solvents like ionic liquids (IL) such as amino acid or carboxylate-based IL (Hospital-Benito et al., 2020) and a super acid mixture of sulphuric acid and glacial acetic acid (Charry Prada et al., 2020). Novel super acid solvent systems absorbed up to 83% and ionic solvents removed 90% of CO_2 from biogas (Charry Prada et al., 2020; Hospital-Benito et al., 2020). These ionic liquids have lower energy requirements for their regeneration and possess high CO_2 uptake capacity (Hospital-Benito et al., 2020). These ionic liquids removed 90% of CO_2 from biogas (Hospital-Benito et al., 2020). The benefits of the super acid solvent system are that it offers reduced energy requirements for their regeneration, complete degradation of volatile organic compounds (VOCs) and siloxanes, as well as simultaneous removal of CO_2 and H_2S (Charry Prada et al., 2020). The super acid (sulphuric acid and glacial acetic acid) solvent system captured up to 83% of CO_2 (Charry Prada et al., 2020).

1.8.2 Physical absorption scrubbing

This method involves the use of water (Wellinger et al., 2013) and/or organic solvents (Muñoz et al., 2015) as absorption scrubbers to remove CO_2 as impurities from gases. Thus using only water as the scrubber or absorbent method is referred to as the water scrubbing method (Angelidaki et al., 2018), while the use of organic solvent only is called organic physical scrubbing (Struk et al., 2020).

1.8.2.1 Water scrubbing

Water scrubbing is a widely applied technique for the purification (cleaning) of biogas and upgrading (Angelidaki et al., 2018). The

method relies on the higher CO_2 and H_2S solubility in water compared to CH_4 . Although this method can get through with an H_2S concentration range of 300–2,500 ppm, however, the H_2S solubility in water is higher than that of CO_2 and thus can be removed alongside CO_2 . But because dissolved H_2S can result in corrosion and offensive odour, hence H_2S is recommended to be removed prior to the use of water scrubbing for CO_2 removal (Ryckebosch et al., 2011; Sun et al., 2015). The water scrubbing method, as presented in Figure 9, is operated at a high-pressure range of 6–10 bar and a temperature of up to 40 °C (Zhou et al., 2017; Angelidaki et al., 2018). In high-pressure water scrubbing, gas enters the absorption packed or tray column from the bottom, and water is then sprayed or introduced into the column from the top and flows down the column counter-current to the biogas (Bauer et al., 2013b; Angelidaki et al., 2018). The H_2S and CO_2 in the raw biogas then dissolve into the water, and the upgraded biogas (i.e., biomethane) leaves and is obtained from the top of the absorption column, while the water that contains dissolved CO_2 , H_2S and small amounts of CH_4 leaves the absorption column from the bottom and flows into a flash tank from where an additional amount of CH_4 can be retrieved back using a lower pressure range of 2.5–3.5 bar (Angelidaki et al., 2018). The water scrubbing method results in high CH_4 losses and possess the inability to remove O_2 and N_2 . The loss of CH_4 is due mainly to its dissolution in water. Based on theoretical calculations, the loss usually lies between 3% and 5% (Sun et al., 2015). However, Persson (Persson, 2003) has reported CH_4 loss of about 8%–10% and Sun et al. (Sun et al., 2015) reported CH_4 loss of less than 2%, while 18% CH_4 loss has been the highest reported to date (223). With this method, a biomethane purity of more than 96% can be obtained after drying (Sun et al., 2015; Hosseinipour and Mehrpooya, 2019), which depends on the non-condensable gases (N_2 and O_2) volume that cannot be removed from CH_4 (10). The main merits of this method are its design simplicity and overall cost-effectiveness (Struk et al., 2020). The limitations or disadvantages of the method are high-water consumption requirements (Starr et al., 2012) and lower energy efficiency (Bauer et al., 2013a). Thus, water regeneration is a vital step for this method. Therefore, most

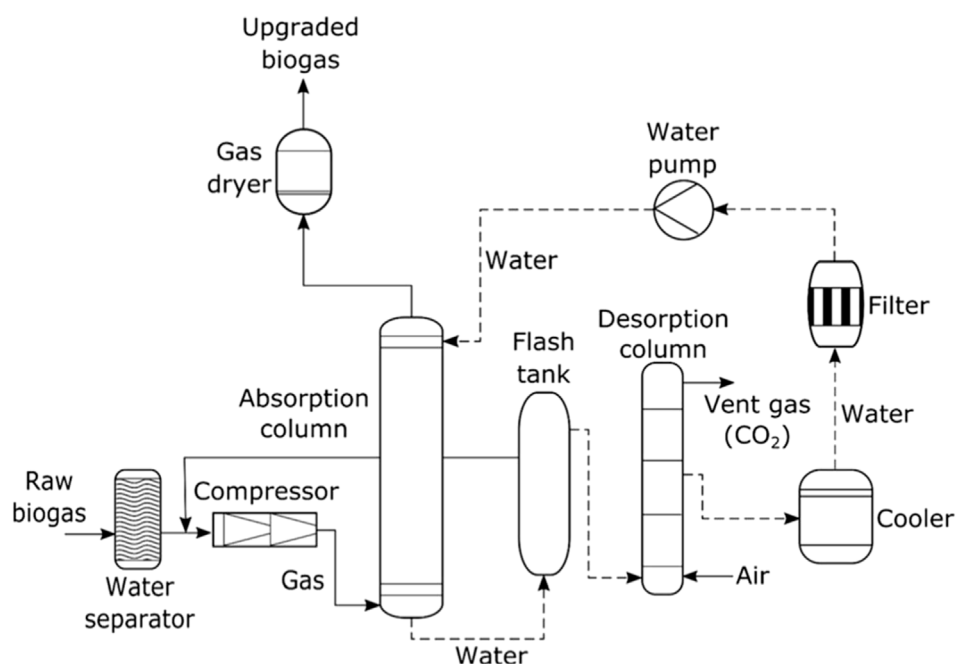


FIGURE 9
Water scrubbing system (Adapted from Baena-Moreno et al., 2019b; Struk et al., 2020).

of the modern absorption column units consist of a two-stage air stripping unit for CO₂ and water regeneration (Bauer et al., 2013a; Muñoz et al., 2015; Bauer et al., 2013b). The dissolved CO₂ present in the water can be obtained using the air stripping method, so it is feasible to achieve high CO₂ purity of 80%–90% (Sun et al., 2015). In addition, efforts have been made in recent times by Noorain et al. (Noorain et al., 2019) and Benizri et al. (Benizri et al., 2019) to improve the removal efficiency of absorption columns by operating at atmospheric pressure and high pressure, respectively. Noorain et al. (Noorain et al., 2019) used sponge materials to pack the column to increase the hydraulic retention time, and they achieved complete removal of H₂S from artificial biogas with CH₄ having a 90% purity. Benizri et al. (Benizri et al., 2019) constructed a CH₄ leak-proof high-pressure water scrubbing system that included a static mixer for CO₂ desorption enhancement, and they reported a high recovery efficiency of 94% CH₄ and a low energy consumption of 0.26 kWh/Nm³.

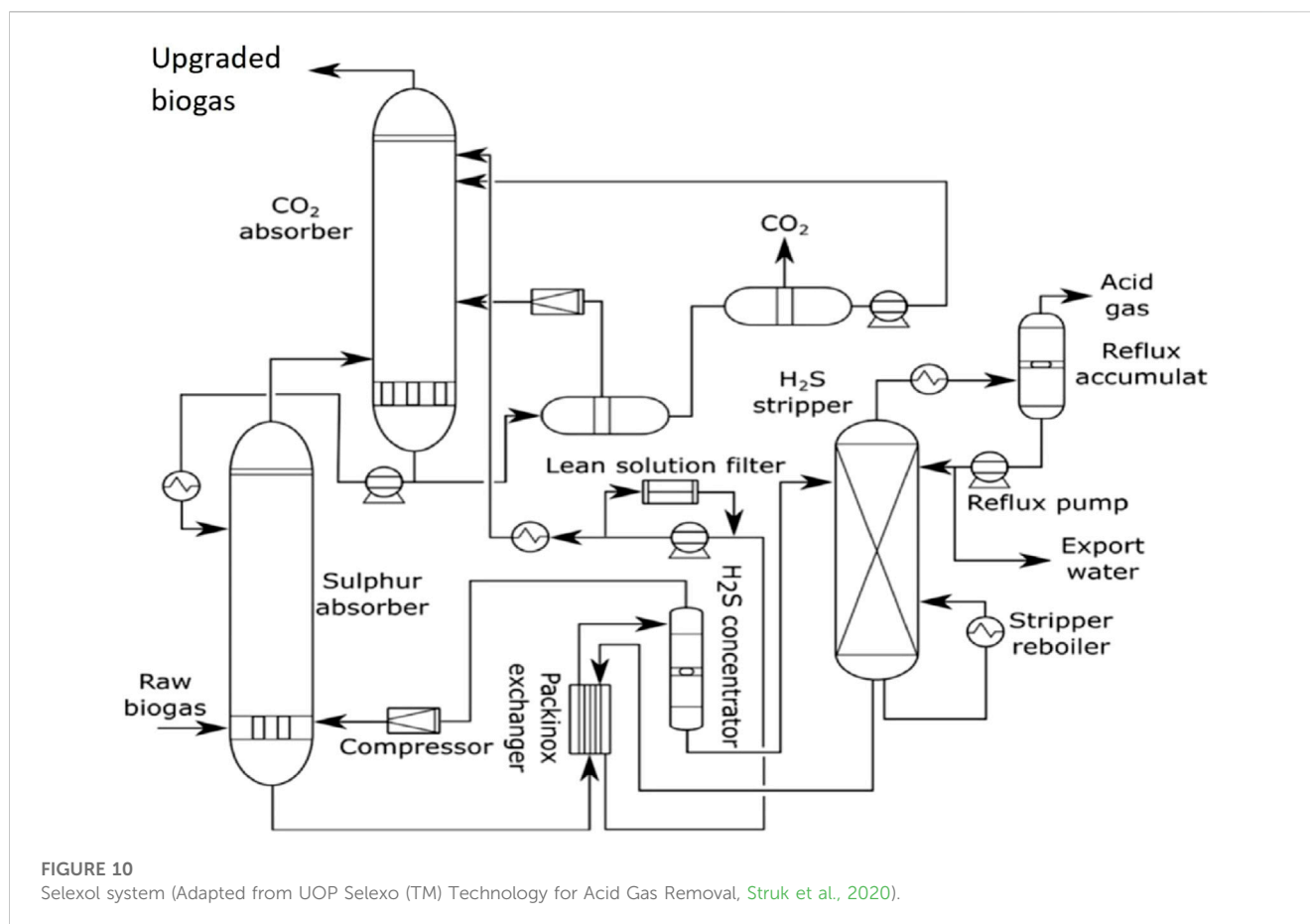
1.8.2.2 Organic physical scrubbing

Organic-physical absorption scrubbing technique, as shown in Figure 10, has the same principle as water scrubbing (Struk et al., 2020). This method entails the use of organic solvents that have higher affinity for H₂S and CO₂ (Angelidaki et al., 2018; Struk et al., 2020). Organic solvents such as methanol and dimethyl ethers of polyethylene glycol (DMPEG) are employed in this method (Awe et al., 2017; Struk et al., 2020). CO₂, H₂S and H₂O can simultaneously be absorbed since they all have a higher solubility than methane in organic solvent (DMPEG) (Awe et al., 2017; Struk et al., 2020). There are commercial organic scrubbers that are sold under the trade names Genosorb and Selexol, which exhibit higher CO₂ and NH₃ solubility compared to water (Struk et al., 2020). The first step in this method involves the compression of the biogas at 7–8 bar and thereafter cooling

at 20 °C. The compressed and cooled biogas is then introduced into the absorption column from the bottom. A cooled organic solvent is injected into the column from the top to flow countercurrent to the biogas (Bauer et al., 2013a). This method can achieve an upgraded biogas that has up to 96%–98.5% CH₄ with less than 2% loss of CH₄ (Bauer et al., 2013a; Bauer et al., 2013b; Sun et al., 2015). The organic solvent regeneration is conducted in the desorption column at a reduced pressure of 1 bar, heating it to 80 °C (Sun et al., 2015; Angelidaki et al., 2018). Presently due to increased pressure towards the use of environmentally friendly organic solutions, the use of green solvents is gradually replacing conventional organic solvents. In recent times, there has been the application of deep eutectic solvents (DES) like Reline and Glyceline as organic scrubbers (Hagbakhsh and Raeissi, 2019). This DES consists of two or more components that are majorly hydrogen bond donors and hydrogen bond acceptors having properties of very low vapour pressure, lower melting point and biodegradable. The potential of Reline and Glyceline to capture and remove CO₂ from natural gas and their performance compared with commercial Selexol was investigated by Hagbakhsh and Raeissi (Hagbakhsh and Raeissi, 2019). The workers reported that Reline, Glyceline and Selexol achieved 79.6%, 90.1% and 89.1% CH₄ purity, respectively, and that both Reline Glyceline attained an average of 98% CO₂ capture as compared to 94.9% achieved by Selexol.

1.9 Membrane separation technology

Membrane separation technology is a method of separation at the molecular level through a selective membrane barrier (Sun et al., 2015) as illustrated in Figure 11.



Membrane separation technology possesses a number of advantages, such as easy process, energy efficiency and low cost (Sun et al., 2015). The permeability of the membrane is based on the electric charges differential of the different molecules, concentration gradient, temperature gradient and pressure gradient (Hashiba et al., 2017; Khan et al., 2017). For membrane separation, there are gas-liquid (wet) and gas-gas (dry) techniques (Struk et al., 2020). The wet and the dry membrane separation techniques differ in their micro-porous membranes' hydrophobic properties (Struk et al., 2020). The liquid system that can be used for the gas-liquid separation system is alkanol amines or aqueous alkali solutions. In gas-liquid membrane systems operated at atmospheric pressure, the biogas diffuses through the membrane and is absorbed by the liquid media passing or flowing counter-current to the biogas (Ryckebosch et al., 2011; Bauer et al., 2013a). In gas-gas membrane systems, the biogas is pressurised at 20–40 bars (or 6–20 bars) such that the CO_2 , H_2S and some amount of CH_4 (about 10%–15%) will diffuse or pass through the membrane to the permeate side with the lower pressure while a significant amount of CH_4 will remain or be retained on the membrane inlet side with the higher pressure (Bauer et al., 2013b; Sun et al., 2015; Struk et al., 2020). The gas-gas membrane separation units are produced under different configurations: single-pass or one-stage and multiple-stage units having internal permeate and retentates recirculation (Ryckebosch et al., 2011). The gas-gas multi-stage membrane units can achieve about 96% or more of CH_4 purity, while one-stage membrane units can result to 92% purity of CH_4 (Allegue and Hinge, 2014). Compared to

the gas-liquid membrane system, the efficiency is lower (Allegue and Hinge, 2014). The most suitable commercial membranes for biogas separation and enrichment are those produced from polymeric materials such as polyimide and cellulose acetate-based membranes (Basu et al., 2010; Scholz et al., 2013).

1.10 Adsorption technology

This technology utilises a porous solid adsorbent to remove the impurities from the biogas (Angelidaki et al., 2018; Moya et al., 2022). The properties of adsorbent types influences the efficiency of this method. The most commonly and widely utilised commercial adsorbents for biogas treatment are carbon-based adsorbents (activated carbon), silica gels, and metal-organic frameworks (Abdelhafez et al., 2014; Augelletti et al., 2017; Lin et al., 2019). The common sources of activated carbon are lignocellulosic biomass, coals and industrial by-products (Bamdad et al., 2018). The metal-organic frameworks (MOF) are a hybrid of organic-inorganic crystalline porous materials consisting of an assemblage of positively charged metal salts ions that are surrounded by organic ligands (Chaemchuen et al., 2013). The MOFs design and synthesis is an emerging area with several possibilities for industrial applications (Chaemchuen et al., 2013; Hanak et al., 2015). The other adsorbent materials include synthetic or natural zeolites, commercial metal oxide-based adsorbents such as SulfaTreat, Sulphur-Rite or SOXSIA (230), copper, alkaline and

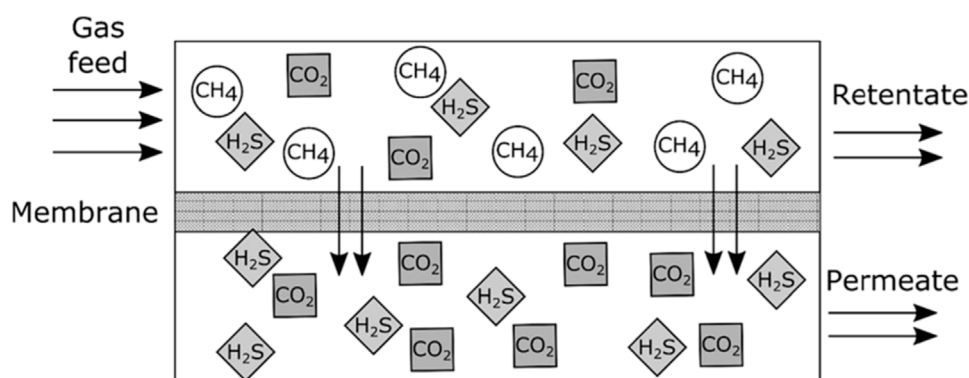


FIGURE 11

Principle of membrane separation (Adapted from Allegue and Hinge 2014; Struk et al., 2020).

iron-based metal oxides and unusual adsorbents such as CeO_2 , La_2O_3 , and TiO_2 (Hakim et al., 2016). The attractive use of iron metal oxides and alkaline metal oxides for CO_2 capture and removal is due to their abundance, favourable adsorption and desorption kinetic, good adsorption capacity, and low costs (Hakim et al., 2016; Bamdad et al., 2018). There are different forms of adsorption technology which include, pressure swing adsorption (PSA) (Wiheeb et al., 2016; Zhou et al., 2017), vacuum swing adsorption (VSA) (Zhou et al., 2017) and temperature swing adsorption (TSA) (Zhou et al., 2017).

1.10.1 Pressure swing adsorption

Pressure swing adsorption (PSA) is based on the selective adsorption of biogas impurities (CO_2) over CH_4 onto solid surfaces under high pressure according to their different molecular sizes and adsorbent affinity (Sun et al., 2015; Ferella et al., 2017; Canevesi et al., 2018). It is called PSA if the adsorbent regeneration occurs when the pressure decreases up to the level of the atmospheric pressure. The PSA technology can be employed to remove CH_4 from N_2 , O_2 and CO_2 since the CH_4 is larger than the rest of the biogas impurities (Sun et al., 2015). The PSA technology, as shown in Figure 12, is carried out in adsorbent-packed vertical columns, and it consists of four steps sequence, which includes adsorption, depressurisation, desorption and pressurisation (Augelletti et al., 2017). As biogas flows or passes through the high-pressurized column, O_2 , CO_2 , H_2S , and N_2 are captured and adsorbed on the adsorbent. Siloxanes and H_2S are irreversibly adsorbing onto the adsorbent. For instance, H_2S forms insoluble metal sulphides with metal-oxide adsorbent, and it is therefore advised to get them removed alongside water vapour prior to the subjection of biogas to the PSA technology (Bauer et al., 2013a). For continuous operation, multiple adsorption columns can be employed (Bauer et al., 2013b). The CH_4 content after upgrading with PSA can get up to a range of 96%–98%; however, with about 2%–4% CH_4 is lost (Sun et al., 2015). Generally, to achieve higher concentrated CH_4 or purity, more CH_4 is usually lost (Sun et al., 2015).

1.10.2 Vacuum swing adsorption

Vacuum swing adsorption (VSA) has a similar operating system to PSA, except that the desorption or regeneration step is conducted at a pressure lower than the atmospheric pressure (Zhou et al., 2017). That is, it is called VSA, if the adsorbent regeneration occurs by applying a vacuum after when the inlet valve has been closed. In recent times, there is more focus on improving the adsorbents and optimizing the PSA/VSA technology. Some of these improvements include silica gel modification by grafting amine functional groups into the pore walls of the silica (Wang and Yang, 2014) temperature vacuum swing adsorption system, which employed the use of nanogel particles containing amines supported by honeycomb shaped-carbon fibre (Gao et al., 2020) and adsorbents tailor-made into 3D printed-patterned structures (Middelkoop et al., 2019). Gao et al. (Gao et al., 2020) reported an increase in CO_2 recovery with the use of nanogel particles containing amines supported by honeycomb shaped-carbon fibre. The different adsorbents for CO_2 and H_2S removal are mostly utilised in the form of beads. When the performance of 139 zeolite beads was compared with the 3D printed-patterned structures, it was reported that higher adsorption and desorption rates were exhibited by the 3D printed-patterned structures. In contrast, the 139 zeolite beads displayed higher adsorption capacity than the 3D printed-patterned structures made from the same zeolite (Middelkoop et al., 2019).

1.10.3 Temperature swing adsorption

Temperature swing adsorption (TSA) is also similar in operation to PSA, except that adsorbent regeneration takes place at varying temperature between 30 and 120 °C while pressure remains constant [Sahota et al., 2018]. Through lower temperatures, the adsorption process becomes more effective, with the regeneration of the adsorbent taking place with increasing temperature (Santos and Joppert, 2018). In TSA, it is hot gases or steam that is utilised in transferring heat to the column (Zhou et al., 2017). TSA is usually restricted to the removal of small amounts of impurities or pollutants. That is, TSA should be applied for the removal of CO_2 concentration that is 1% or less, while PSA is preferred for CO_2 concentration that is greater than 10% (Zhou et al., 2017).

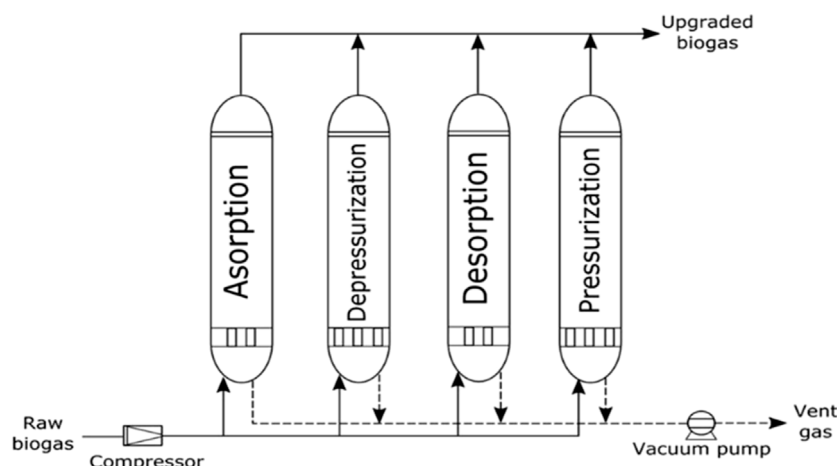


FIGURE 12

Pressure swing adsorption system (Adapted from Kadam and Panwar, 2017; Struk et al., 2020).

1.11 Cryogenic separation technology

The cryogenic separation technology (Figure 13) is a distillation based technology that involves the separation of the different biogas mixtures hinged on their different boiling points performed through a gradual sequential temperature reduction (Muñoz et al., 2015) in four stages. The technology demands high pressure of about 50–200 bar for initial compression and cryogenic temperature (i.e., low temperature that is close to -125°C) (Hosseinipour and Mehrpooya, 2019). Components of the biogas possess different temperatures of liquefaction (condensation) and solidification, which gives room for CH_4 , H_2S , H_2O , and CO_2 selective separation (Bauer et al., 2013a; Tan et al., 2017b). The cryogenic separation process begins first with biogas compression to 80 bar and temperature reduction to -25°C . At this first stage, the H_2S , siloxanes, halogens and H_2O are removed. After the first stage, in the second stage, the purified biogas is compressed to a pressure of 1000 kPa (80 bar) and then passed through a heat exchanger where the biogas temperature is further cooled or reduced to -25°C . The cooled biogas then enters into the third stage, where the temperature is further cooled or reduced to -55°C , during which most of the CO_2 is liquefied or condensed. The temperature of this cooled purified biogas is finally cooled or reduced to -85°C in the fourth stage, during which the remaining CO_2 solidifies and is completely separated or removed (Ryckebosch et al., 2011; Riva et al., 2014; Baena-Moreno et al., 2019a). Also, after CH_4 condensation, N_2 and O_2 can be removed. To avoid the problem of freezing (i.e., formation of ice) and other operational problems, such as pipe or heat exchanger clogging in the cryogenic process of separation, H_2S and H_2O are to be removed prior to cryogenic separation application (Bauer et al., 2013a; Sun et al., 2015; Baena-Moreno et al., 2019a). The cryogenic separation technology can result in over 97% CH_4 purity and less than 1 or 2% of CH_4 loss (Andriani et al., 2014; Baena-Moreno et al., 2019a) as well as in 98% CO_2 purity (Sun et al., 2015). However, the use of this technology is limited by large or high energy requirement, high capital investment and operation costs (Angelidaki et al., 2018; Ryckebosch et al., 2011; Muñoz et al., 2015). In order to provide solutions to the limitations presented by

cryogenic separation technology, some workers have proposed the use of cryogenic hybrid technologies such as cryogenic hydrates systems (Sreenivasulu et al., 2015; Baena-Moreno et al., 2019b), cryogenic adsorption systems (Fong et al., 2016; Moreira et al., 2017; Song et al., 2019), cryogenic absorption systems (Hanak et al., 2015; Song et al., 2018) and cryogenic membrane systems (Song et al., 2017; Song et al., 2019). Presently, the cryogenic membrane process has been the most intense combination studied.

1.12 Hydrate formation technology

The contaminants of biogas can be removed or separated based on the difference in their formation of gas hydrate (Sun et al., 2015). This method has been utilised to successfully remove CO_2 from impure natural gas (Van Denderen et al., 2009). It has been reported that this technology consumes a large amount of energy due to the high pressure needed for hydrate formation (Awe et al., 2017). This technology requires improvement by reducing its energy consumption and methane loss for future use.

1.13 Biological technology

The biological technologies that can be applied for biogas upgrading can generally be categorised into chemoautotrophic and photoautotrophic technologies (Angelidaki et al., 2018).

1.13.1 Chemoautotrophic technology

The chemoautotrophic technology, which is an end-of-the-pipe biotechnology, can be applied for CO_2 conversion to CH_4 and for biogas desulphurization (i.e., H_2S removal) (Muñoz et al., 2015; Struk et al., 2020). For CO_2 conversion to CH_4 , the technology is based on the application of hydrogenotrophic methanogenic bacteria in converting CO_2 to CH_4 through the utilisation of H_2 as an electron donor (Muñoz et al., 2015). This hydrogen-assisted biogas upgrading technology can be categorised into three, namely,

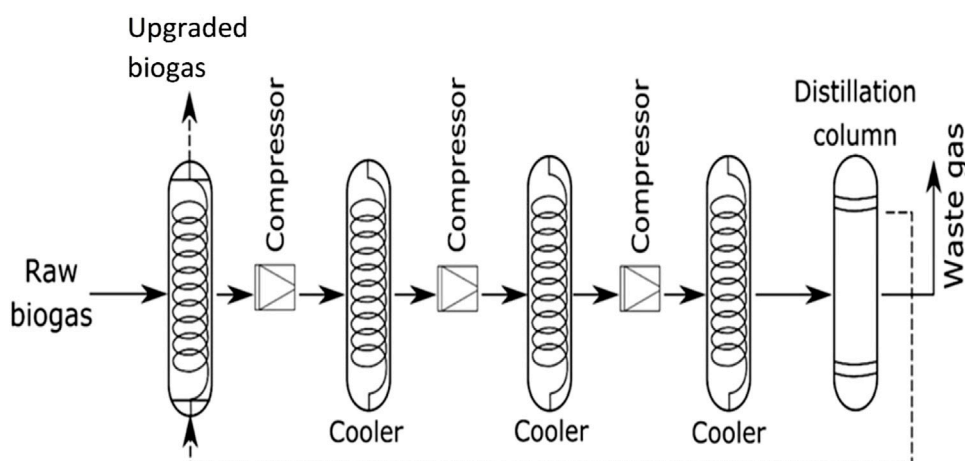


FIGURE 13

Cryogenic separation system. (Adapted from Baena-Moreno et al., 2019b; Struk et al., 2020).

in-situ, *ex-situ* and hybrid chemoautotrophic technologies (Kougias et al., 2017).

1.13.1.1 *In-situ* chemoautotrophic technology for CO₂ conversion

In the *in-situ* chemoautotrophic technology, H₂ is introduced or injected into the biogas digester or bioreactor so that the endogenous CO₂ produced in the digester can react with it and be converted into CH₄ through the activity of the autochthonous methanogens (Kougias et al., 2017). These methanogens which belong to the Archaeal genera include *Methanosarcina* sp., *Methanobacterium* sp., *Methanoculleus*, *Methanothermobacter* sp., *Methanospirillum* sp., *Methanosaeta* sp., and *Methanococcus* sp. (Muñoz et al., 2015). The main challenge of this technology is related to pH elevation to values above 8.5, which often leads to methanogenesis inhibition (Luo et al., 2012) as well as the low gas-liquid H₂ mass transfer rate (Muñoz et al., 2015). The pH elevation is attributed to bicarbonate removal, which is the main biogas process buffer (Angelidaki et al., 2018). In order to alleviate this technical challenge, co-digestion with acidic waste was proposed to arrest the increase of pH (Luo and Angelidaki, 2013) as well as the application of pH control (Luo et al., 2014). The use of this technology can achieve approximately 80%–100% final CH₄ content in the biogas (Wang et al., 2013; Luo et al., 2014; Bassani et al., 2016; Agneessens et al., 2017; Mulat et al., 2017) depending on the material and module type utilised for H₂ injection, gas recirculation flow, reactor design and pH control (Bassani et al., 2016; Angelidaki et al., 2018).

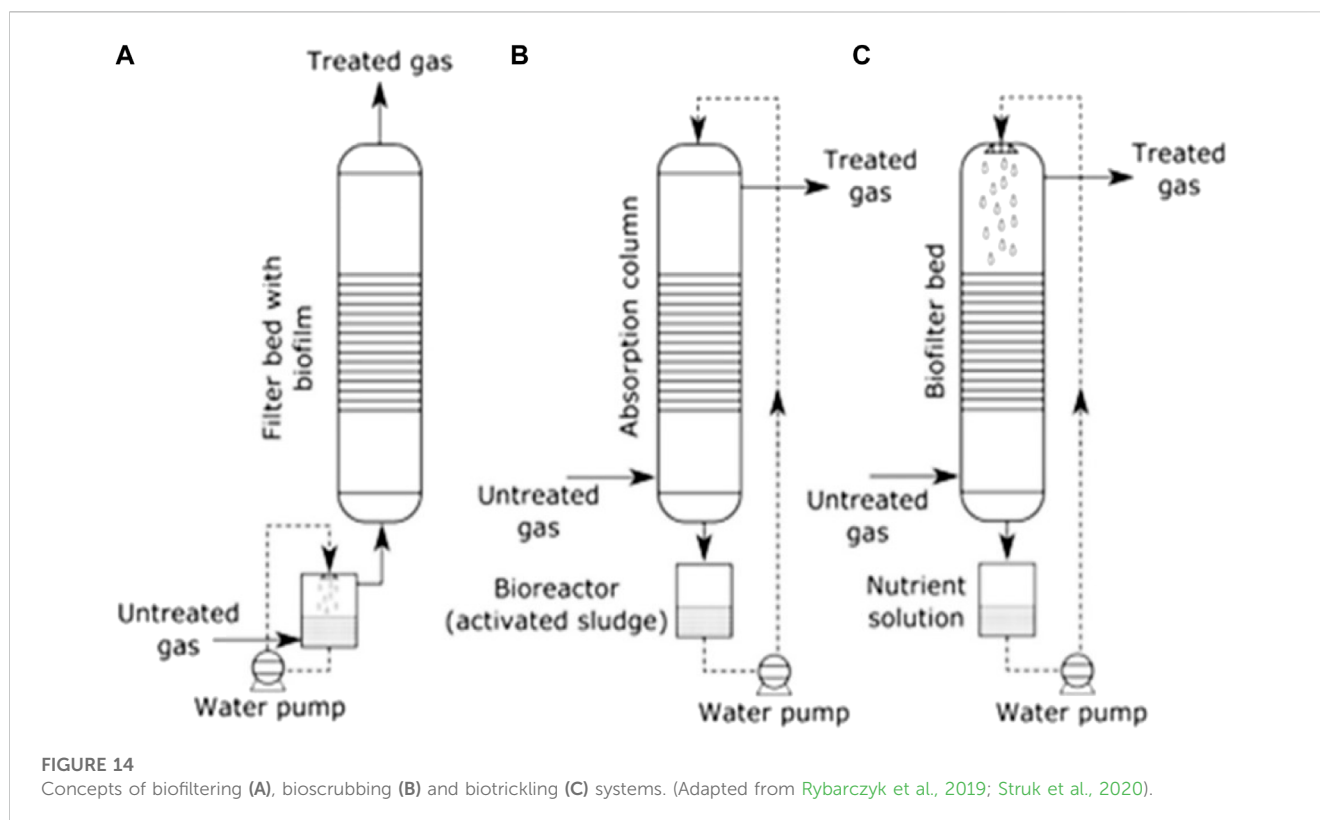
1.13.1.2 *Ex-situ* chemoautotrophic technology for CO₂ conversion

The *ex-situ* chemoautotrophic technology entails the biogas upgrade being performed in a separate unit or chamber that contains pure or enriched hydrogenotrophic microbial culture in which external CO₂ is injected in order for it to be coupled with the H₂ present in the biogas and through the action of the culture is subsequently converted to CH₄ (Kougias et al., 2017). This biogas

upgrading technology, strongly depending on the diffusion device utilised for CO₂ injection, gas recirculation flow, reactor configuration and pH control, can result in final CH₄ content that ranges from 79% to 98% as well as a range of 71%–100% CO₂ removal (Diaz et al., 2015; Bassani et al., 2016; Kougias et al., 2017; Savvas et al., 2017). The major limitation or challenge of this technology is the low gas-liquid mass transfer (Angelidaki et al., 2018).

1.13.1.3 Chemoautotrophic technology for biogas desulphurization

For biogas desulphurization, the chemoautotrophic technology is based on the oxidation of H₂S to sulphate through the metabolic activity of lithotrophic or chemotrophic sulphur-oxidizing bacteria (SOB) using CO₂ as a carbon source and O₂ in the biogas or NO₃ as electron acceptor (Madigan et al., 2014; Struk et al., 2020). Examples of these SOB belongs to the genera *Thiospira*, *Thiobacillus*, *Sulfolobus*, *Acidithiobacillus*, *Thiothrix*, *Thiovulum*, *Paracoccus*, *Halothiobacillus* and *Sulfurimonas*, which could be strictly aerobes or facultative anaerobes (Muñoz et al., 2015). This can be performed using biofilters, biotrickling filters and bioscrubbers (Struk et al., 2020), as presented in Figure 14. A biofilter consists of a bioreactor that contains moist packed bed material with microbial biofilm immobilized on it (Mudliar et al., 2010). Due to their indigenous microbial consortia and high performance, natural organic materials (such as peats, coconut fibre, composts, woodchips, and bark) can be utilised as packed bed materials in biofilters (Struk et al., 2020). In carrying out the chemoautotrophic technology using a biofilter, the biogas is injected into the biofilter from the bottom and flows through the moist, packed bed material with microbial biofilm. The packed bed material provides the nutrient, and occasionally, a nutrient solution is added from the top of the biofilter. In addition, 5%–10% of oxygen is required, which may not be present in the biogas, and this level is usually controlled by direct injection of air into the biogas stream (Syed et al., 2006). Despite the benefits of the low cost of operation with no

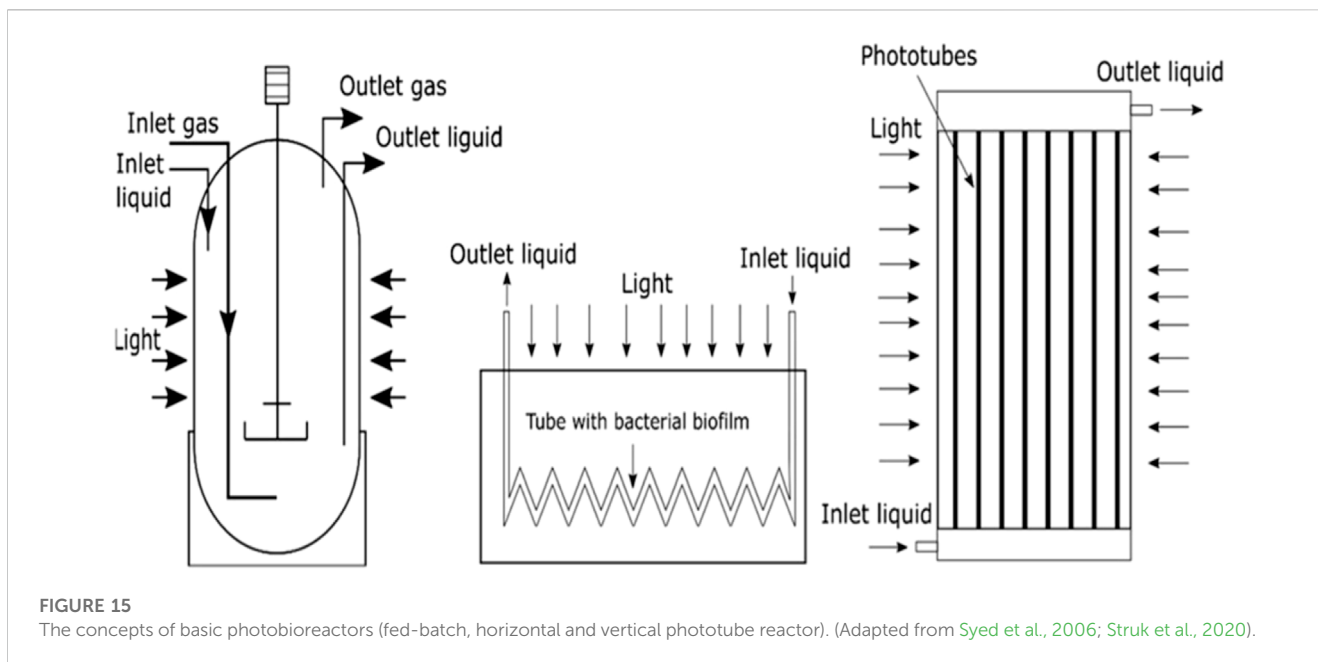


chemical required, the use of this system still has some limitations, such as the occurrence of packed bed acidification due to sulphuric acid formation, inefficient or no sufficient mixing, and pH and moisture control problems (Mudliar et al., 2010; Brinkmann et al., 2016; Struk et al., 2020). To circumvent these limitations, the use of acid-tolerant mixed culture, the use of packed bed materials with alkaline properties, periodic filter media washing and alkaline solution addition have been proposed (Mudliar et al., 2010; Allegue and Hinge, 2012; Struk et al., 2020). The principle of biotrickling filters is similar to that of biofilters. However, the difference is in the utilisation of inert packing bed materials and the need for the continuous supply of nutrient solution into the system. The biotrickling system has plenty of advantages over the biofiltering system and some of which include easy control and regulation of operational factors (e.g., temperature and pH), less space requirement, high process stability, lower flow resistance and continuous nutrient provision (Schiavon et al., 2015; Barbusinski et al., 2017; Rybarczyk et al., 2019; Struk et al., 2020). The biotrickling system is a widely applied technology that is commercially available as Biogas Cleaner (BioGas clean), Biopuric process (Biothane Corporation), Bidox (Colsen B.V.) and BioSulfurex (DMT Environmental Technology) (Allegue and Hinge, 2014). The bed materials used as matrix or carrier for microbial biofilm immobilization that are commonly utilised in biotrickling filters are porous ceramics, plastic and metallic pall rings, polyurethane foam, and plastic supports activated granular carbon (Mudliar et al., 2010; Fernandez et al., 2014; Montebello et al., 2014; Rodriguez et al., 2014; Struk et al., 2020). The biotrickling system is able to achieve approximately 80%–100% H₂S removal (Fernandez et al., 2014; Montebello et al., 2014; Rodriguez et al.,

2014). The bioscrubber system consists majorly of two reactor units (absorption tower and activated sludge reactor) but in some cases, three (e.g., sedimentation tank) depending on the type of contaminant to be removed. For H₂S removal, the biogas is injected into the absorption tower where the H₂S is absorbed in the countercurrent flowing liquid phase (e.g., alkaline solution) and then flows into the activated sludge reactor, wherein the presence of O₂, degradation takes place by the microbial consortia (Struk et al., 2020). The resultant effluent moves into the sedimentation tank to collect the elemental sulphur with O₂ used as the oxidant and then recirculates back to the absorption tower. The bioscrubber system does not require O₂ and N₂ injection into the biogas but requires the addition of nutrient, continuous excess biomass and by-products purging, and pH and O₂ regulation and control to maintain the optimal growth and metabolic activities of the microorganisms (Allegue and Hinge, 2014; Brinkmann et al., 2016). The commercially well-known bioscrubbers employed for the removal of H₂S from biogas are Sulfothane™ (France) and THIOPAQ process (Netherlands) (Janssen et al., 2009). The major significant demerit of the bioscrubber system is its relatively high initial capital investment (Allegue and Hinge, 2012).

1.13.2 Photoautotrophic technology

The photoautotrophic technology is based on the photosynthetic capability of phototrophic organisms (such as prokaryotic cyanobacteria or eukaryotic microalgae) to sequester the CO₂ present in the biogas as to obtain a CH₄-rich biogas (Angelidaki et al., 2018). This technology involves the use of enclosed photo-bioreactors (Prandini et al., 2016) or open photo-bioreactors (278) and open ponds or raceways (Bahr et al., 2014; Serejo et al., 2015). In



this technology (Figure 15), the biogas is directly injected into the photobioreactors containing the photoautotrophic microorganisms, and these organisms efficiently sequester CO₂ using nutrients, solar irradiation and water to produce oxygen, biomass and heat. The CH₄ recovery of this technology can get up to about 97% depending on the selected type of algal species (Bahr et al., 2014; Serejo et al., 2015) and a maximum range of 2%–6% CO₂ in the final biogas (Meier et al., 2015). Moreover, the biomass produced in the course of biogas upgrading can be utilised either for value-added products' extraction (Guedes et al., 2011) or as biogas production feedstock oriented to a circular economy (Mussnug et al., 2010). Some of the microalgae or cyanobacteria possessing high photosynthetic efficiency that have been successfully employed for biogas upgrading using this technology include *Chlorella* species, *Arthrospira platensis* and *Spirulina platensis* (Muñoz et al., 2015; Converti et al., 2009; Kao et al., 2012; Bahr et al., 2013). For example, utilizing *Chlorella* sp., *A. platensis*, and *S. platensis* respectively achieved up to 85%, 100%, and 86% photosynthetic CO₂ sequestration or uptake, while it resulted in 90%, 82% and 74% final CH₄ content in the biogas, respectively (Muñoz et al., 2015; Converti et al., 2009; Kao et al., 2012; Bahr et al., 2013). This technology can also be applied for the removal of H₂S (biogas desulphurization), while the amount of CO₂ consumed can be greater than 54% (Angelidaki et al., 2018). To be able to achieve H₂S removal, a mixed consortium of microalgae and sulphur-oxidizing bacteria (SOB) (i.e., algal-bacterial symbiosis) can be employed such that the SOB will oxidize the H₂S to sulphate utilizing the O₂ present in the biogas as an electron acceptor (Bahr et al., 2013; Bahr et al., 2014).

1.14 Energy efficiency of biogas purification and upgrading technologies

The energy efficiency (η) of biogas purification and upgrading technologies is defined as given in Eq. 9:

$$\eta = \frac{\text{Energy}_{\text{upgraded biogas}}}{\text{Energy}_{\text{raw biogas}} + \text{Energy}_{\text{upgrading}}} \quad (9)$$

Energy efficiency can be employed as a criteria for comparing the efficiency or performance of the different biogas upgrading technologies.

1.15 Comparison of biogas upgrading technologies, challenges and mitigations

The performance of physical, chemical and biological biogas upgrading technologies is compared (Table 4) in terms of different parameters or criteria: gas pre-cleaning requirements, H₂S removal requirement, N₂ and O₂ removal, working pressure, working temperature, CH₄ concentration in upgraded biogas, CH₄ loss, CH₄ recovery, specific energy consumption (kWh/Nm³ upgraded biogas), heat requirement, quality of upgraded gas, investment cost €/Nm³ biogas, maintenance cost, and technical availability. Table 4 data revealed that the conventional biogas upgrading technologies (physical absorption, chemical absorption, adsorption and membrane separation) and the emerging biogas upgrading technologies (hydrate formation, cryogenic and biological) displayed different responses to the performance criteria. The information in Table 4 shows that all the different biogas upgrading technologies relatively result in high CH₄ purity levels. Amongst these technologies, organic chemical scrubbing technology suggests resulting in the highest CH₄ purity level (99%) of raw biogas with the least CH₄ loss (0.04%–0.1%) and the lowest level of energy consumption (0.05–0.25 kWh/Nm³ biomethane). However, its heat requirement, investment and maintenance costs seem to be relatively very high in comparison with other physical and chemical technologies. Therefore, it is not only appropriate to focus on the quality or purity of the biomethane alone but equally very important to consider the cost (i.e., investment/operational cost, maintenance cost, and energy cost) as core

TABLE 4 Comparisons of biogas upgrading technologies [Source: Patterson et al., 2011; Castellani et al., 2014; Masebinu et al., 2014; Muñoz et al., 2015; Sun et al., 2015; Khan et al., 2017; Singhal et al., 2017; Vrbová and Ciahotný, 2017; Kapoor et al., 2019; Khan et al., 2021].

Parameter	Physical absorption scrubbing		Chemical absorption scrubbing		Adsorption	Membrane separation	Cryogenic separation	Hydrate formation	Biological technology	
	Water scrubbing	Organic physical scrubbing	Organic chemical absorption	Inorganic chemical absorption (caustic wash scrubbing)	Pressure/ Vacuum/ Temperature swing adsorption				Chemoautotrophic	Photoautotrophic
Gas pre-cleaning requirement	Required	Not required	Required	Required	Required	Required	Required & Recommended	Required	Required	Required
H ₂ S removal requirement	Not required	No required	Required	Required	Required	Required	Required & Recommended	Required	Not required	Not required
N ₂ and O ₂ removal	No	No	No	No	Possible	Partial	No	No	No	No
Working pressure (bar)	6–10	4–7	1	1	4–8	20–40	50–200	40–70	Not required	Not required
Temperature (°C)	20–40	10–20	35–50	35–50	5–30	25–60	–25–85	3–5	30–55	15–35
Heat requirement	Not required	Medium, 70°C–80 °C	Very high (120°C–160 °C)	Very high (120°C–160 °C)	Not required	Not required	Not required	Not required	Not required	Not required
Energy Consumption (kWh/Nm ³ biomethane)	0.2–0.5	0.10–0.33	0.05–0.25	Not available	0.16–0.43	0.18–0.35	0.2–0.79	Not available	Not available	Not available
%CH ₄ purity in biogas	95–98	>96, 96–98.5	99	94–97	>96–98	92–97 96	>97, 99	84	95–96	95–96
CH ₄ loss (%)	1–3	1.5–4	0.04–0.1		1–3.5	0.5–20	0.5–3	High	0	0
Efficiency	88.9–92.8	90–95.5	88.5–97.7	Not available	84.8–93.6	Not available	82.4–98	84.9–96.7	Not available	Not available
Technical availability	96	96	91	Not available	94	98	Not available	Not available	Not available	Not available
Investment cost (€/Nm ³ biogas)	0.13–1.5	0.25	0.28	No cost data found	0.26	0.22	0.44	Not available	Not available	Not available
Maintenance cost (€/year)	15,000	39,000	59,000	No cost data found	56,000	25,000	No cost data found	Not available	Not available	Not available

TABLE 5 Challenges of the biogas purification and upgrading technologies and their recommended mitigations.

Biogas upgrading technology	Challenge	Recommended mitigation
Chemical absorption scrubbing		
Organic chemical absorption	The major challenge of this technology is that the investment and operational costs are very high due to the high cost of amine solvent as well as amine solvent loss caused by evaporation as a result of its high volatility and high energy (i.e., high-temperature heat) requirements for solvent regeneration (Kadam and Panwar, 2017; Khan et al., 2017; Struk et al., 2020)	To address this challenge, it is recommended that new amine solution that possesses high uptake capacity with lower volatility and less energy requirement for solvent regeneration as well as an efficient heat exchanger, should be developed
Inorganic chemical absorption (Caustic wash scrubbing)	There is high consumption of caustic (hydroxides) solutions due to the formation of bicarbonates that cannot be regenerated, leading to solvent loss and high energy requirement for solvent regeneration (Hosseinipour and Mehrpooya, 2019)	The issue of solvent loss and high energy requirement for solvent regeneration can be resolved by making use of new solvents that has high uptake capacity with less energy requirement for regeneration. For example, ionic liquids (IL) such as amino acid or carboxylate-based IL as well as super acid mixture of sulphuric acid and glacial acetic acid, are solvents that are being researched to address this issue (Charry Prada et al., 2020; Hospital-Benito et al., 2020)
Physical absorption scrubbing		
(a) Water scrubbing	The major challenge is that it requires high water consumption, and the energy efficiency is low due to the high loss of CH ₄ (Bauer et al., 2013a). In addition, the H ₂ S that dissolves in the water causes corrosion and thus leads to equipment damage	The high water consumption requirement and regeneration can be addressed by integrating the water scrubbing technology with a wastewater treatment plant. In addition, the H ₂ S should be removed from the raw biogas before subjection to water scrubbing as to prevent corrosion that leads to equipment damage
(b) Organic physical scrubbing	The major challenge is that the technology involves heating or high energy requirement for solvent regeneration while it is also expensive due to high investment and operational costs (Wagner et al., 2018)	To circumvent these challenges, it is recommended that alternative energy sources should be utilized, unnecessary material flow to the upgrading unit should be reduced, the biogas upgrading plant should be optimized, and the system should be automatized and remotely operated
Adsorption: Pressure swing adsorption (PSA)	The challenges are that the technology is expensive due to high investment and operating cost that emanates from additional biogas pretreatment with complicated process control and high CH ₄ loss (Hjorth et al., 2011; Zhang et al., 2014)	To address these challenges, <i>in-situ</i> biogas-cleaning or purification using inexpensive and efficient kinetic-based adsorbents (e.g., activated carbon/carbon molecular sieve 3 K) or equilibrium-based adsorbents (e.g., Metal-Organic Frameworks (MOFs)) should be incorporated at the beginning into the anaerobic bioreactor such that H ₂ S, CO ₂ , and some other impurities would have been removed at the level of the biogas production. Adsorbent materials such as Zeolitic Imidazolite Frameworks (ZIF) and Metal-Organic Frameworks (MOFs) have been developed by some researchers (Chaemchuen et al., 2013) and their adsorption efficiency can further be tested. Furthermore, to reduce CH ₄ losses in the desorbed gas, the PSA system should be re-designed in such a way that desorbed gases should be made to recirculate back to the PSA process
Membrane Separation	The major challenges of this technology include medium to high CH ₄ loss, membrane replacement in 1–5 years due to membrane's degradation with time, and clogging and fouling of membranes	To circumvent these challenges, raw biogas pre-treatment is recommended before carrying out the membrane separation to prevent membrane's deterioration and degradation as to prolong the membrane's life, as well as to avoid clogging of the membrane. In addition, to reduce CH ₄ losses in the off-line gas, the membrane separation system should be re-designed in such a way that off-line gases should be made to recirculate back to the membrane process
Cryogenic Separation	The technology is faced with the challenges of being multi-faceted and technically complex to operate due to the utilization of a number of equipments such as compressors, turbines, heat exchangers and distillation column (Bauer et al., 2013), large or high energy requirement for compression and refrigeration of biogas (Zhang et al., 2014), and high capital investment and operation costs (Paudel et al., 2017)	To mitigate these challenges, it is recommended that new cryogenic hybrid configuration technologies with reduced technical complexities and lower energy consumption should be developed and applied. Some workers have proposed the use of cryogenic hydrates systems (Sreenivasulu et al., 2015), cryogenic adsorption systems (Fong et al., 2016; Moreira et al., 2017; Song et al., 2019), cryogenic absorption systems (Hanak et al., 2015; Song et al., 2018), and cryogenic membrane systems (Song et al., 2017; Song et al., 2019). Furthermore, <i>in-situ</i> biogas-

(Continued on following page)

TABLE 5 (Continued) Challenges of the biogas purification and upgrading technologies and their recommended mitigations.

Biogas upgrading technology	Challenge	Recommended mitigation
		cleaning or purification using inexpensive and efficient kinetic-based/equilibrium-based adsorbents (e.g., activated carbon/carbon molecular sieve 3 K or metal-organic frameworks) should be incorporated at the beginning into the anaerobic biodigester such that H ₂ S, CO ₂ , and some other impurities would have been removed at the level of the biogas production. Also, ways to recover the energy utilized for raw biogas condensation should be developed, and this can be achieved by carrying out the liquefaction of CH ₄ at 15 bar and -125 °C and this minimises the energy consumption
Hydrate Formation	The challenges include the consumption of large amounts of energy and high-pressure requirement	These challenges can be circumvented using hybrid technology such as hydrate-membrane technology
Biological Technology		
<i>In-situ</i> Chemoautotrophic	The major challenges faced involve the occurrence of low gas-liquid H ₂ mass transfer rate (Muñoz et al., 2015) as well as the occurrence of pH elevation to values above 8.5, which often leads to methanogenesis inhibition (Luo et al., 2012)	To mitigate these challenges, it is recommended that the materials in the biogas digester should be co-digested with acidic waste materials as to arrest pH increase (Luo and Angelidaki, 2013) as well as to apply the use of pH control agents (Luo et al., 2014). In addition, the diffusion device utilized for H ₂ injection into the biogas digester should be modified so as to allow for increased gas-liquid H ₂ mass transfer in the biogas digester
<i>Ex-situ</i> chemoautotrophic	The major challenge is the occurrence of low gas (CO ₂)-liquid mass transfer	To resolve this challenge, it is recommended that the diffusion equipment to be used for the injection of external CO ₂ into the biogas reactor should be modified in order to allow for increased gas-liquid mass transfer
Chemoautotrophic-biofilter system	The challenges faced with the use of biofilters in carrying out chemoautotrophic technology include the existence of low gas-liquid mass transfer, packed bed acidification due to sulphuric acid formation, inefficient or no sufficient mixing pH and moisture control problems (Mudliar et al., 2010; Brinkmann et al., 2016; Struk et al., 2020)	To mitigate these challenges, it is recommended that the packing bed materials to be used in the biofilter should possess alkaline properties (i.e., alkaline packing bed material) with high mass transfer coefficient so as to increase the gas-liquid mass transfer and to solve the problem of acidification. Alternatively, acid-tolerant mixed cultures can be employed. There should also be periodic filter media washing and the addition of alkaline solution to address the challenge of pH and moisture control (Mudliar et al., 2010; Allegue and Hinge, 2012; Struk et al., 2020)
Photoautotrophic	The fundamental challenge is the occurrence of poor gas-liquid mass transfer of CO ₂ and H ₂ (Angelidaki et al., 2018)	To overcome this challenge, it is recommended that an indirect biogas upgrading system should be adopted. This entails capturing the CO ₂ in a carbonate solution (e.g., potassium carbonate) and this potassium carbonate solution is fed into the microalgal culture. The carbonate solution is utilised by the microalgae as a source of carbon for growth and later regenerates the carbonate for a next biogas upgrading cycle (Nguyen et al., 2020)

criteria in the selection of appropriate biogas upgrading technology. To this end, among the physical and chemical biogas upgrading technologies, membrane based upgrading technology seems to meet up with these said criteria. Nevertheless, many of these conventional technologies (physical and chemical) require a high energy input and/or costly chemicals/materials for upgrading the biogas to CH₄, increasing operational and maintenance costs. In this regard, the biological biogas upgrading technologies can play a pertinent role with their lower energy requirements and no requirement for costly chemicals or materials. Therefore, at this present stage, with the development and laboratory validation of new cost-efficient biological technologies showing very good performance, it becomes difficult to predict or select which of all these (conventional and emerging) technologies will dominate the

biogas upgrading process. Furthermore, the challenges faced by these technologies and the suggested or recommended mitigation(s) are presented in Table 5.

2 Conclusion and recommendation

This paper reviewed the process intensification strategies required for enhancing AD performance as well as the biogas cleaning and upgrading technologies. It can be concluded from this review that the performance of the AD process for optimum delivery and stability has been seriously affected by the feedstock-induced operational problems (i.e., feedstock physical and chemical compositions) as well as other operational factors classified as

physical, chemical and biological, respectively. Pretreatment, co-digestion, bioaugmentation (i.e., biological additives), biostimulation (addition of macro and micronutrients), additions of carbon-based conductive materials ion-exchange capacity materials as well as improved reactor configuration and operating parameters can be applied for accelerating or enhancing the AD Process technology performance to achieve greater biogas and biomethane yield or production. However, the large or full-scale application of bioaugmentation is still scarce due to its economic risks and uncertainties. Pretreatment and co-digestion promote feedstock accessibility to the microbial community involved in the AD process. However, choosing the type of pretreatment method depends on the feedstock type; hence, there should be a compromise between the cost and energy involved. There is no comprehensive information with regard to the financial feasibility of the enzyme pretreatment. The review showed that the investment cost in the production of commercial enzymes is the greatest obstacle hindering its industrial or full-scale application as a pretreatment method. Thus, future research should be geared towards investigating new approaches to reducing the cost of enzyme pretreatment cost so as to make it economically feasible for full-scale application. Furthermore, depending on the specific biogas application, different levels of cleaning or purification and upgrading technologies (absorption, adsorption, cryogenic, membrane, hydrate and biological separation) are required. These technologies differ in their overall efficiency, cost and energy requirements. Some of these technologies, such as cryogenic separation, hydrate separation and biotechnological methods, are novel emerging technologies that are still under development. However, cost minimization and gas quality requirements are the fundamental criteria for selecting an appropriate cleaning and upgrading technology. To further reduce biogas upgrading costs, the utilisation of separated CO₂ for other purposes like algae production and enhanced oil recovery can offer additional benefits reinforcing the concept of circular bioeconomy. Since separated CO₂ can be used for other beneficial purposes, it is,

therefore, necessary that more research attention should be focused on the separation of CO₂ and its quality as there is not much available information on it.

Author contributions

OAw, AA, and OAg contributed to the conception and design of the study. SA, OO, OL, and IF organised the database. OAw, AA, and OAg wrote the first draft of the manuscript. OO, MK, and IF reviewed and edited the manuscript. All authors contributed to the article and approved the submitted version.

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Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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