



N-Benzoimidazole/Oxadiazole Hybrid Universal Electron Acceptors for Highly Efficient Exciplex-Type Thermally Activated Delayed Fluorescence OLEDs

Wenbo Yuan^{1†}, Hannan Yang^{2†}, Mucan Zhang¹, Die Hu¹, Ning Sun^{2*} and Youtian Tao^{1*}

¹ Key Lab for Flexible Electronics and Institute of Advanced Materials (IAM), Nanjing Tech University, Nanjing, China,

² Department of Physics, Yunnan University, Kunming, China

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*Correspondence:

Ning Sun
ning.sun@ynu.edu.cn
Youtian Tao
iamyttao@njtech.edu.cn

[†]These authors have contributed
equally to this work

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Recently, donor/acceptor type exciplex have attracted considerable interests due to the low driving voltages and small singlet-triplet bandgaps for efficient reverse intersystem crossing to achieve 100% excitons for high efficiency thermally activated delayed fluorescence (TADF) OLEDs. Herein, two N-linked benzoimidazole/oxadiazole hybrid electron acceptors were designed and synthesized through simple catalyst-free C-N coupling reaction. 24iPBIOXD and iTPBIOXD exhibited deep-blue emission with peak at 421 and 459 nm in solution, 397 and 419 nm at film state, respectively. The HOMO/LUMO energy levels were $-6.14/-2.80$ eV for 24iPBIOXD and $-6.17/-2.95$ eV for iTPBIOXD. Both compounds could form exciplex with conventional electron donors such as TAPC, TCTA, and mCP. It is found that the electroluminescent performance for exciplex-type OLEDs as well as the delayed lifetime was dependent with the driving force of both HOMO and LUMO energy offsets on exciplex formation. The delayed lifetime from 579 to 2,045 ns was achieved at driving forces close to or larger than 1 eV. Two TAPC based devices possessing large HOMO/LUMO offsets of 1.09–1.34 eV exhibited the best EL performance, with maximum external quantum efficiency (EQE) of 9.3% for 24iPBIOXD and 7.0% for iTPBIOXD acceptor. The TCTA containing exciplex demonstrated moderate energy offsets (0.88–1.03 eV) and EL efficiency ($\sim 4\%$), while mCP systems showed the poorest EL performance (EQE $< 1\%$) and shortest delayed lifetime of < 100 ns due to inadequate driving force of 0.47–0.75 eV for efficient exciplex formation.

Keywords: electron acceptor, electron donor, exciplex, oxadiazole, thermally activated delayed fluorescence, OLEDs

INTRODUCTION

Organic light-emitting diodes (OLEDs) have been developed rapidly in recent years since the pioneer work on low-voltage fluorescence electroluminescence by Tang in 1987 (Tang and VanSlyke, 1987; Ma et al., 1998; Gong et al., 2010; Park et al., 2013; Zhang et al., 2016). According to spin statistics, the ratio for singlet and triplet excitons recombined from electrogenerated holes and electrons is 1:3 (Baldo et al., 1999; Segal et al., 2003). Thus, the first generation of traditional fluorescent OLEDs which solely harvest singlet excitons only shows 25% of maximum internal

quantum efficiency (IQE) (Wen et al., 2005). On the other hand, the second generation of phosphorescent OLEDs (PHOLEDs) based on heavy metal complexes and third generation of thermally activated delayed fluorescence (TADF) OLEDs could both reach 100% IQE in theory by utilizing all singlet and triplet excitons through intersystem crossing (ISC) and reverse inter-system crossing (RISC), respectively (Baldo et al., 1998; Adachi et al., 2001; Su et al., 2008; Lo et al., 2009; Goushi et al., 2012; Uoyama et al., 2012; Zhang and Forrest, 2012; Li et al., 2016; Cao et al., 2017; Huang et al., 2018; Wu Q. et al., 2018). However, to avoid consuming noble metals and achieving reliable true-blue light, TADF OLEDs based on low-cost pure organic emitters have attracted increasing interests as an alternative mechanism to PHOLEDs. TADF emission is realized by an up-conversion process from lower energy triplet states to slightly higher energy singlet states by endothermic reverse inter-system crossing process (Li et al., 2016; Cao et al., 2017; Huang et al., 2018; Wu Q. et al., 2018). Therefore, a small singlet-triplet energy bandgap (ΔE_{ST}) is required for TADF emitters.

It is reported that the small ΔE_{ST} could be attained in (i) intramolecular charge transfer featured single molecule with twisted donor-acceptor structured for effective spatial isolation between the highest occupied molecular orbital (HOMO) and lowest unoccupied molecular orbital (LUMO) on the relevant hole and electron transporting moieties, and (ii) bimolecular-exciplex which contains an electron-donor material mixed with an electron-acceptor material through intermolecular charge transfer characteristics (Cai and Su, 2018; Liu et al., 2018; Sarma and Wong, 2018). High external quantum efficiency (EQE) of 20% for red, 29% for orange, 38% for green, and 37% for light-blue TADF OLEDs have been achieved in single-molecule TADF emitters (Lin et al., 2016; Chen et al., 2018; Wu T.-L. et al., 2018; Zeng et al., 2018). However, the development of bimolecular TADF lags far behind. Most exciplex-type TADF OLEDs showed maximum EQE close to 10% (Jankus et al., 2014; Liu et al., 2015a,b; Oh et al., 2015; Zhang L. et al., 2015; Hung et al., 2016, 2017; Jeon et al., 2016), with only one example approaching to 18% (Liu et al., 2016).

In electron donor/acceptor formed exciplex systems, compared with commercially available various electron-donor materials, such as 4,4',4''-tris[3-methylphenyl(phenyl)amino]-triphenylamine (*m*-MTDATA), *N,N'*-bis(1-naphthyl)-*N,N'*-diphenyl-[1,1'-biphenyl]-4,4'-diamine (NPB), 4,4'-(cyclohexane-1,1-diyl)bis(*N*-phenyl-*N-p*-tolylaniline) (TAPC), 4,4',4''-tris(*N*-carbazolyl) triphenylamine (TCTA), 4,4'-bis(*N*-carbazolyl)-1,1'-biphenyl (CBP), and *N,N'*-dicarbazolyl-3,5-benzene (mCP) etc., the types of efficient and low-cost electron-acceptor materials are scarce (Goushi and Adachi, 2012; Goushi et al., 2012; Sun et al., 2014; Lee et al., 2015; Liu et al., 2015b). Thus, the exploration of electron accepting materials is essential for constructing exciplex systems with outstanding optoelectronic performance. Therefore, in this work, we designed and synthesized two new electron-acceptors of 2-(2,4-bis(2-phenyl-1*H*-benzo[*d*]imidazol-1-yl)phenyl)-5-phenyl-1,3,4-oxadiazole (24*i*PBIOXD), and 2-phenyl-5-(2,4,6-tris(2-phenyl-1*H*-benzo[*d*]imidazol-1-yl)phenyl)-1,3,4-oxadiazole (*i*TPBIOXD) through a simple

one-step catalyst-free aromatic nucleophilic substitution reaction. The electron-withdrawing oxadiazole (OXD) unit has been extensively applied in donor-acceptor type bipolar transport host materials, single molecule intramolecular charge transfer type TADF emitters as well as electron transport materials (Tao et al., 2011; Mondal et al., 2013; Olivier et al., 2017; Cooper et al., 2018; Yao et al., 2018; Zhang et al., 2018). By combining OXD building block with our previously reported isomeric *N*-linked benzoimidazole (Hu et al., 2017), both 24*i*PBIOXD, and *i*TPBIOXD exhibited deep HOMO level of \sim -6.15 eV, facilitating the exciplex formation with general electron donor materials of TAPC, TCTA, and mCP due to the compatible HOMO and LUMO energy levels between donor and acceptor materials. The gradient energy offsets ranging from 0.47 to 1.34 eV correlated well with the delayed lifetime and EL efficiencies in exciplex type TADF OLEDs. The TAPC:24*i*PBIOXD exciplex with the largest HOMO/LUMO offsets exhibited the best EL performance, with maximum EQE of 9.3% for green TADF OLEDs.

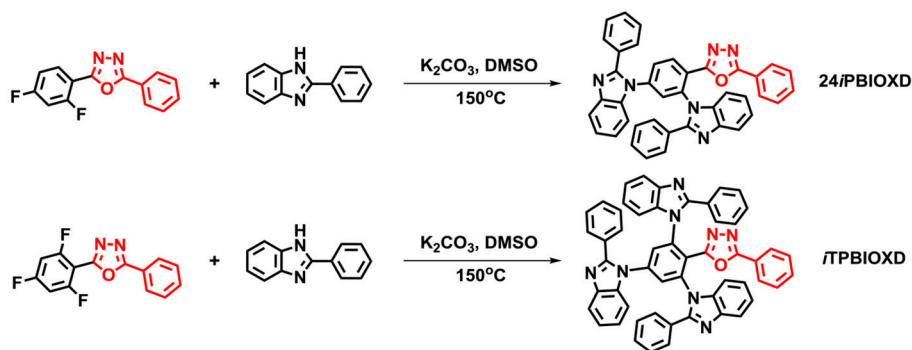
RESULTS AND DISCUSSION

Synthesis and Characterization

Scheme 1 shows the synthetic routes and molecular structures of 24*i*PBIOXD and *i*TPBIOXD. The two compounds could be facilely synthesized by a simple one-step catalyst free C-N coupling reaction. This nucleophilic substitution reaction was carried out in DMSO solvent with K_2CO_3 base at high yields over 80% by using di/tri-fluorine substituted oxadiazole derivatives as electrophiles and 2-phenyl-1*H*-benzo[*d*]imidazole as nucleophiles. The considerably high yields and environmentally eco-friendly conditions demonstrated the superiority than common metal-catalyzed Ullman reactions (Son et al., 2008; Liu et al., 2011; Volz et al., 2013). In addition, the directly connection of the isomeric *N*-linked benzoimidazole to the central phenyl ring avoided the complicated multistep ring-closing synthetic process for the normal C-linked benzoimidazole in traditional electron transport material of 2,2,2-(1,3,5-phenylene)-tris(1-phenyl-1*H*-benzoimidazole) (TPBI) or its derivatives. The chemical structures of the new compounds were fully characterized by 1H NMR, ^{13}C NMR, mass spectrometry (MALDI-TOF) and element analysis (**Figure S1**). The good thermal stability of the two compounds was confirmed by thermogravimetric analysis (TGA) and differential scanning calorimetry (DSC) (**Figure 1**). The decomposition temperatures (T_d , corresponding to a 5% weight loss) from TGA curves for 24*i*PBIOXD and *i*TPBIOXD were determined at 443 and 461°C, respectively. Additionally, the melting point (T_m) of *i*TPBIOXD was observed at 327°C, which was much higher than 286°C of 24*i*PBIOXD. The glass transition temperature (T_g) of both materials can be detected from the second heating cycles from DSC, with values of 126°C for 24*i*PBIOXD and 165°C for *i*TPBIOXD, indicating their reasonable thermal stability.

Photophysical Properties

The room temperature UV-Vis absorption and photoluminescence (PL) spectra of 24*i*PBIOXD and *i*TPBIOXD



SCHEME 1 | Synthesis of compounds 24/*i*PBIOXD and *i*TPBIOXD.

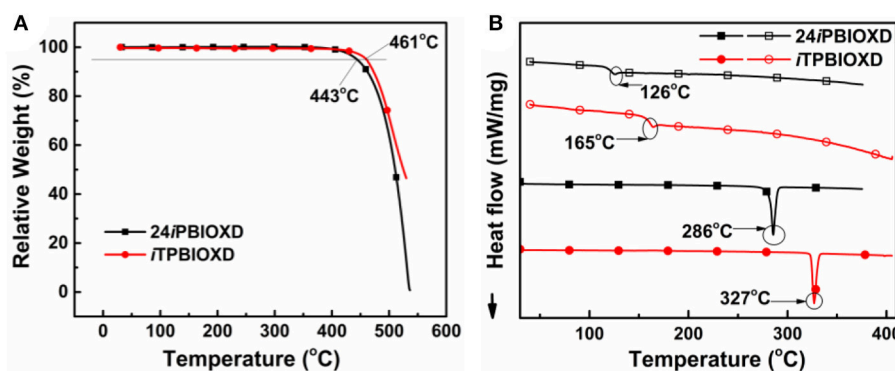


FIGURE 1 | (A) Thermogravimetric analysis (TGA) and (B) differential scanning calorimetry (DSC) (solid symbols represent for first heating scan and open symbols for second heating scan) curves for 24/*i*PBIOXD and *i*TPBIOXD.

in CH_2Cl_2 solution are shown in **Figure 2A**. Both compounds exhibited an intense absorption with peaks at 289 and 283 nm in solution, 297 and 288 nm in film, respectively, which can be ascribed to the π - π^* transition of molecules. The optical bandgap (E_g) was calculated to be 3.34 eV for 24/*i*PBIOXD and 3.22 eV for *i*TPBIOXD, according to the film-state absorption edge. On the other hand, 24/*i*PBIOXD and *i*TPBIOXD showed unimodal photoluminescence peaking at 421 and 459 nm in solution, whereas significantly blue-shift to 397 and 419 nm in film state (**Table 1**). By analyzing the highest-energy vibronic sub-band of low-temperature fluorescence and phosphorescence spectrum (**Figure 2B**), the singlet (E_S) and triplet (E_T) energy levels could be determined to be 3.31/2.55 and 3.18/2.53 eV for 24/*i*PBIOXD and *i*TPBIOXD, respectively. In addition, the E_S/E_T energy levels of three hole-transport electron donor materials were also calculated to be 3.54/2.95 eV for mCP, 3.79/2.82 eV for TAPC, and 3.66/2.84 eV for TCTA (**Figure 2C**). The PL spectra for the neat film of electron donors such as mCP, TAPC, and TCTA, the two new electron-acceptors of 24/*i*PBIOXD and *i*TPBIOXD as well as their corresponding mixtures in a 1:1 weight ratio were investigated. As shown in **Figure 3** and **Figure S2**, all blended films showed bathochromic shifted PL spectra compared with the emission of neat 24/*i*PBIOXD/*i*TPBIOXD and the corresponding donor-material,

indicating the successful formation of exciplex (Zhang T. et al., 2015). In addition, it is found that exciplex based on 24/*i*PBIOXD acceptors all exhibited about 20–30 nm blue-shifted emission than *i*TPBIOXD based exciplex systems. The exciplex emission color could be tuned from deep-blue of mCP:24/*i*PBIOXD with peak at 419 nm to light-blue of TCTA:24/*i*PBIOXD (501 nm) and further to green of TAPC:24/*i*PBIOXD (518 nm). Besides, transient photoluminescence (PL) measurements were carried out for all six exciplexes (**Figure 4**). The exciplexes comprising TAPC or TCTA donor all possessed significantly longer delayed decay lifetime, with values of 579 ns for TCTA:24/*i*PBIOXD, 1,907 ns for TCTA:*i*TPBIOXD, 1,520 and 2,045 ns for TAPC:24/*i*PBIOXD, TAPC:*i*TPBIOXD exciplex, respectively. However, the mCP:24/*i*PBIOXD and mCP:*i*TPBIOXD exciplex systems displayed greatly shorter delayed decay lifetime of only 42 and 72 ns (**Table 1**). Besides, the temperature dependent PL transients for the representative TAPC:24/*i*PBIOXD and TCTA:*i*TPBIOXD exciplexes (**Figure S3**) both demonstrated a more significant decay from 100 to 300 K at the longer lifetime range, suggesting the potential existence of endothermic reverse inter-system crossing. It is expected the obvious variations on delayed decay time for different exciplexes may demonstrate some relationships with the device efficiency in exciplex-TADF OLEDs.

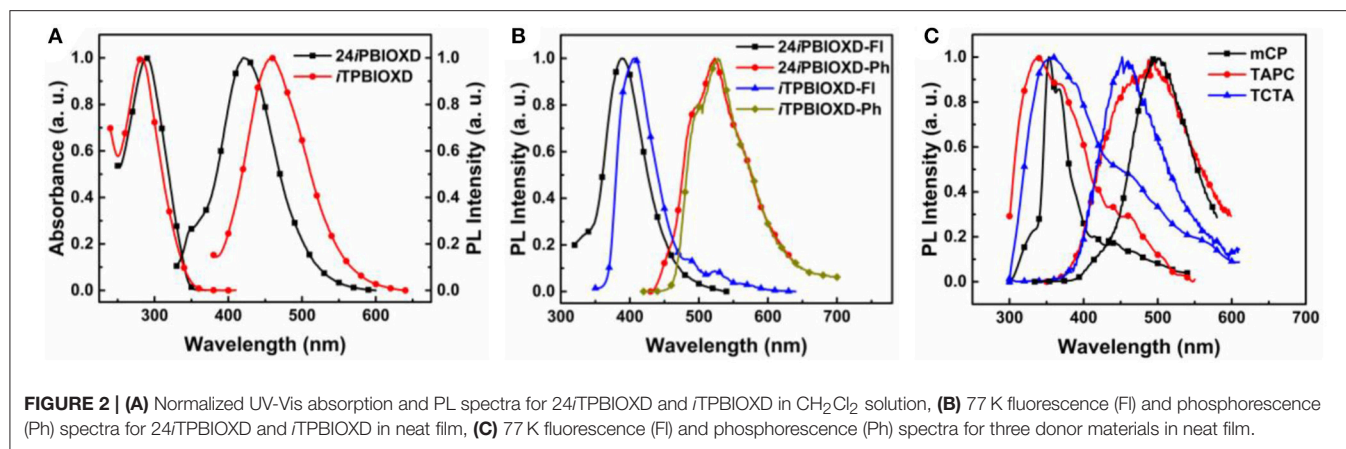


TABLE 1 | Physical properties of compounds 24iPBIOXD and iTPBIOXD.

Compounds	$\lambda_{\text{abs}}/\lambda_{\text{em}}^{\text{a}}$ [nm]	$\lambda_{\text{abs}}/\lambda_{\text{em}}^{\text{b}}$ [nm]	$E_{\text{S}}/E_{\text{T}}^{\text{c}}$ [eV]	E_{g}^{d} [eV]	HOMO/LUMO ^e [eV]	$T_{\text{g}}/T_{\text{m}}/T_{\text{d}}^{\text{f}}$ [°C]	λ_{CT} (nm)/ E_{CT} (eV)/ τ_{d} (ns)/ ΔE (eV) ^g		
							mCP	TCTA	TAPC
24iPBIOXD	289/421	297/397	3.31/2.55	3.34	-6.14 (-5.75)/ -2.80 (-2.09)	126/286/443	419/2.96/42	501/2.48/579	518/2.39/1520
iTPBIOXD	283/459	288/419	3.18/2.53	3.22	-6.17 (-5.95)/ -2.95(-2.05)	165/327/461	443/2.80/72	531/2.36/1907	544/2.28/2045

^a Measured in CH₂Cl₂ solution at room temperature.

^b Measured in film.

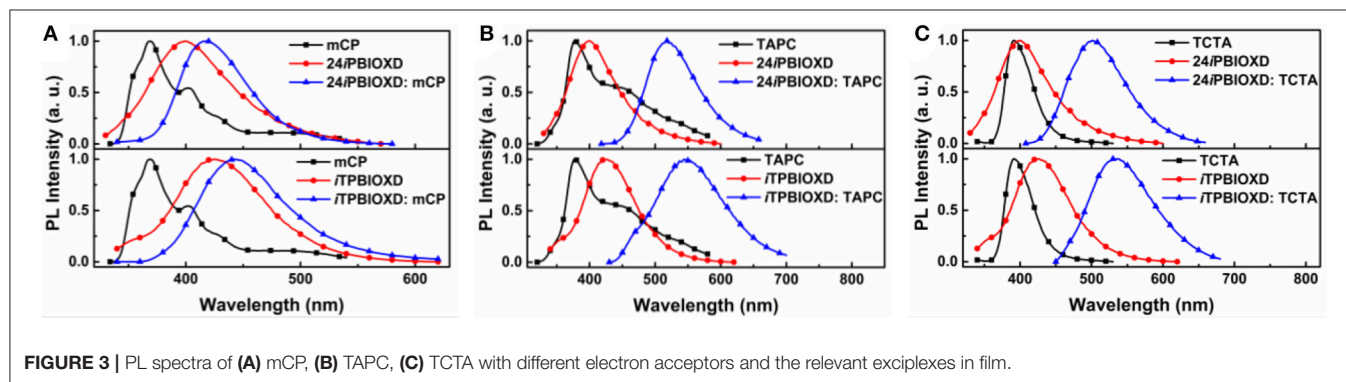
^c Singlet energy and triplet energy was calculated from low temperature (77 K) fluorescence spectra and phosphorescence spectrum.

^d Optical bandgap (E_{g}) calculated from the absorption edge of film state UV-Vis spectra.

^e LUMO measured from the onset of reduction curves from CV and HOMO calculated from the difference between LUMO and E_{g} , values in parentheses from DFT calculations.

^f Glass transition temperature/melting point/decomposition temperature.

^g Emission maxima, charge transfer state energy, delayed decay lifetime and HOMO/LUMO energy offsets for various exciplexes.



Theoretical Calculations and Electrochemical Properties

In order to gain insights into the frontier molecular orbital and excited states level distribution of 24iPBIOXD and iTPBIOXD, density functional theory (DFT) calculation was conducted at the B3LYP level (Francl et al., 1982; Becke, 1988; Lee et al., 1988). From the optimized geometry shown in **Figure 5**, the dihedral angles between the central phenyl and oxadiazole ring were 22.0 and 50.3° for 24iPBIOXD and iTPBIOXD, respectively, the values between the benzoimidazoles and the

central phenyl rings ranged from 50.4 to 77.4°, indicating a twisted structure for both compounds. Furthermore, in the ground state, the highest occupied molecular orbital (HOMO) were almost completely located on one of the ortho-positioned phenylbenzoimidazole units, indicating the electron-donating characteristics of *N*-linked phenylbenzoimidazole, which was quite different from the *C*-isomerized phenylbenzoimidazole containing TPBI (Hu et al., 2017). And the lowest unoccupied molecular orbital (LUMO) were mainly localized on 2,5-diphenyl-1,3,4-oxadiazole, along with mildly distribution over

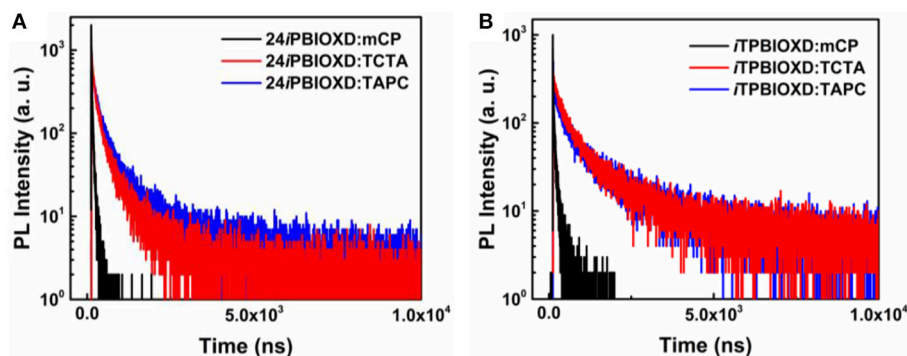


FIGURE 4 | Transient decay curves of (A) 24*i*PBIOXD and (B) *i*TPBIOXD-based exciplexes at film state.

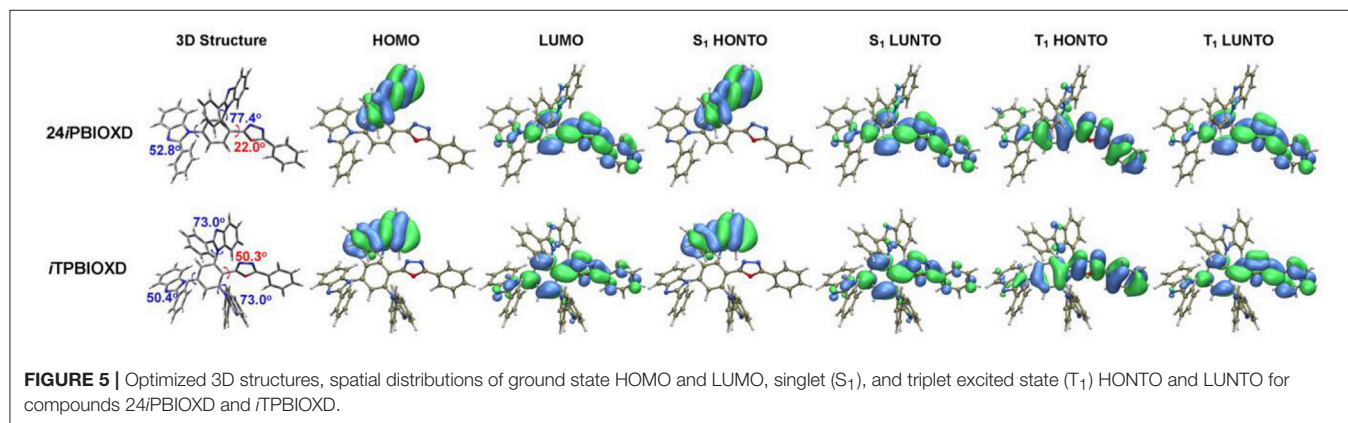


FIGURE 5 | Optimized 3D structures, spatial distributions of ground state HOMO and LUMO, singlet (S_1), and triplet excited state (T_1) HONTO and LUNTO for compounds 24*i*PBIOXD and *i*TPBIOXD.

the penta-heterocyclic imidazoles, suggesting the weak electron-withdrawing property to gently participate electron-transport for the imidazoles. Similar distribution can be observed in the highest occupied natural transition orbital (HONTO) and the lowest unoccupied natural transition orbital (LUNTO) at singlet excited state. It should be noted that the HONTO distribution at triplet excited state was completely different from S_0 and S_1 for both compounds, which was mainly delocalized through the 2,5-diphenyl-1,3,4-oxadiazole skeleton, similar with the LUNTO distribution.

The electrochemical features were measured by cyclic voltammetry (CV) (Figure 6). Both compounds exhibited reversible reduction whereas undetectable oxidation behavior. The LUMO energy levels calculated from the onset of reduction curves for 24*i*PBIOXD and *i*TPBIOXD were measured to be -2.80 and -2.95 eV, while the HOMO energy levels calculated from the different between the LUMO and optical bandgaps (E_g) were evaluated to be -6.14 and -6.17 eV, respectively. The values were in good agreement with the theoretical calculation. Besides, the energy levels for electron donor materials were also measured, with HOMO estimated from the onset of electro-oxidation curves and LUMO calculated from HOMO and optical bandgaps. The HOMO/LUMO energy level values for mCP, TAPC, and TCTA were $-5.67/-2.20$, $-5.05/-1.61$, and $-5.19/-1.92$ eV, respectively. The deep HOMO and LUMO for

the two new electron acceptors of 24*i*PBIOXD and *i*TPBIOXD, provided sufficient driving forces on HOMO/LUMO energy offsets for the exciplex formation (Figure 6A). As shown in Figure 7A, the HOMO energy level offsets between the electron donor of TAPC, TCTA, or mCP and the electron acceptors of 24*i*PBIOXD/*i*TPBIOXD were calculated to be 1.09/1.12, 0.95/0.98, or 0.47/0.5 eV, and the corresponding LUMO offsets were 1.19/1.34, 0.88/1.03, or 0.6/0.75 eV, respectively. It is noted in both acceptor systems, the TAPC donor based exciplex presented the highest driving force, followed by TCTA, while the mCP donor demonstrated the lowest HOMO/LUMO offsets.

Electroluminescence Properties

To investigate the charge transport properties of the two new *N*-linked isomeric benzoimidazole containing electron acceptors, single carrier electron-only device was prepared to find out the electron inject and transport properties of 24*i*PBIOXD and *i*TPBIOXD. The device structure was ITO/24*i*PBIOXD, *i*TPBIOXD, or TPBI (50 nm)/LiF (1 nm)/Al (150 nm), where the commercial electron transport material of 2,2,2-(1,3,5-phenylene)-tris(1-phenyl-1*H*-benzoimidazole) (TPBI) with *C*-linkage in benzoimidazole was selected for comparison. As shown in Figure 8, at the same operating voltage, TPBI based device exhibited the highest current density among all the three devices. Since the LUMO energy of TPBI (2.7–2.9 eV)

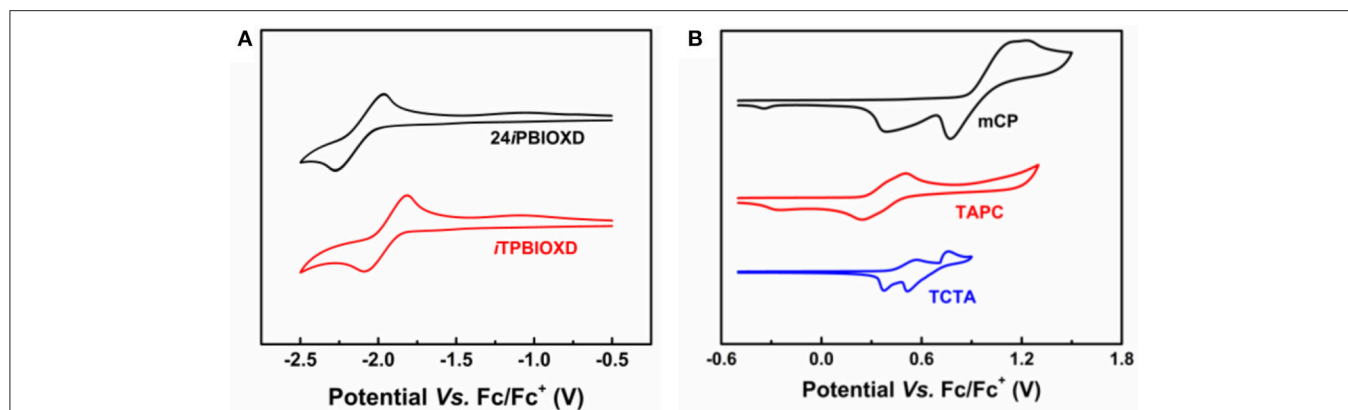


FIGURE 6 | Cyclic voltammograms of (A) 24iPBIOXD and iTPBIOXD in THF solution for reduction scan; (B) conventional electron donors (mCP, TAPC, and TCTA) in CH₂Cl₂ solution for oxidation scan.

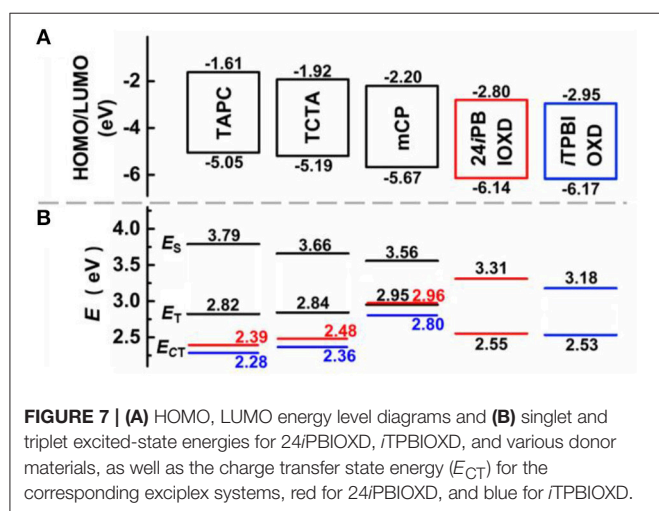


FIGURE 7 | (A) HOMO, LUMO energy level diagrams and (B) singlet and triplet excited-state energies for 24iPBIOXD, iTPBIOXD, and various donor materials, as well as the charge transfer state energy (E_{CT}) for the corresponding exciplex systems, red for 24iPBIOXD, and blue for iTPBIOXD.

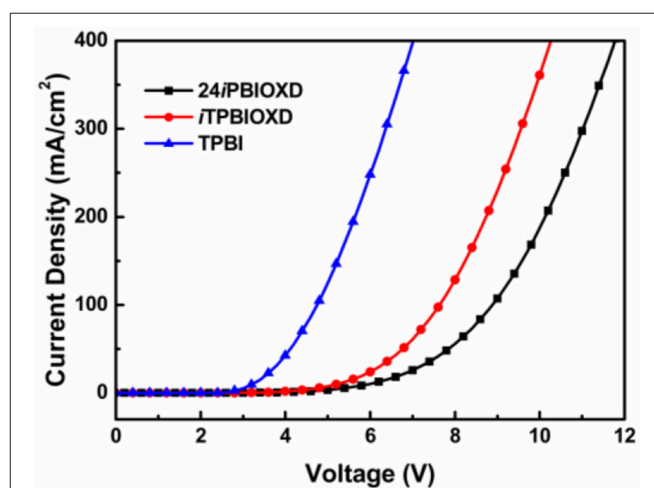


FIGURE 8 | J-V characteristic of nominal single-electron-only devices based on compounds 24iPBIOXD, iTPBIOXD, and TPBI [device structures: ITO/EML (50 nm)/LiF (1 nm)/Al (150 nm)].

(Bian et al., 2018; Jou et al., 2018) was almost the same as 24iPBIOXD and iTPBIOXD, which manifested their similar injection barrier for efficient electron injection. Therefore, the significantly higher current for TPBI indicated better electron transporting property than 24iPBIOXD and iTPBIOXD. On the other hand, the current density in iTPBIOXD device was slightly higher than 24iPBIOXD, as depicted in **Figure 8**, the LUMO level of iTPBIOXD was 0.15 eV lower than 24iPBIOXD, therefore a mildly efficient electron-injection could be attained in iTPBIOXD device due to its lower injection barriers. Thus, the electron-transport performance for both electron acceptors may be comparable.

To conduct a comprehensive comparison on the EL performance for the exciplex-TADF OLEDs among diverse electron-donor and acceptor systems, a series of vacuum deposited devices A-F were fabricated. Due to the highest HOMO level of TAPC for efficient hole-injection, the device configuration for TAPC-based OLEDs was ITO/MoO₃ (1 nm)/TAPC:24iPBIOXD or iTPBIOXD (1:1, 70 nm)/LiF (1 nm)/Al (100 nm). To reduce the hole-injection

barrier, TCTA-based device was constructed by ITO/MoO₃ (1 nm)/TAPC (40 nm)/TCTA:24iPBIOXD or iTPBIOXD (1:1, 30 nm)/TmPyPB (40 nm)/LiF (1 nm)/Al (100 nm), while a further 10 nm TCTA thin film was inserted between the TAPC layer and emissive layer (EML) in mCP-based devices. Among them, MoO₃ and LiF were used as hole- and electron-injection materials, respectively; TAPC and 1,3,5-tri[(3-pyridyl)-phen-3-yl] benzene (TmPyPB) were functionalized as hole- and electron-transport materials, respectively, an extra TCTA layer was aimed to promote the hole-injection and block the electrons.

The current density-voltage-luminance (J - V - L), electroluminescence (EL) spectra, together with the current and power efficiency, external quantum efficiency vs. luminance curves are shown in **Figure 9**. The device fabrication details are stated in Supporting Information. According to the key EL data listed in **Table 2**, the turn-on voltage for TAPC, TCTA, and mCP containing devices was gradually increased from 2.8, 3.0

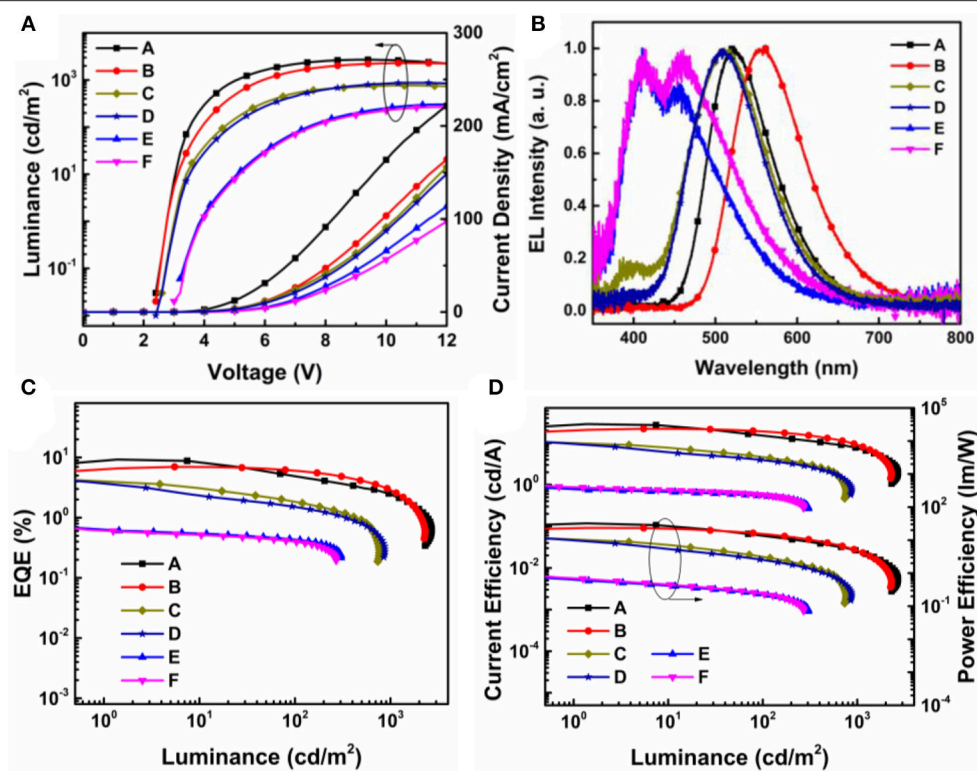


FIGURE 9 | (A) L - V - J characteristics; **(B)** normalized electroluminescent (EL) spectra; **(C)** external quantum efficiency (EQE) vs. luminance curves; **(D)** current efficiency and power efficiency vs. luminance curves of device A–F.

TABLE 2 | Electroluminescence characteristics for the devices.

Device	Emitting layer	V_{on}^a (V)	η_c^b (cd/A)	η_p^c (lm/W)	η_{ext}^d (%)	CIE (x,y)
A	TAPC: 24iPBIOXD	2.8	28.8	32.3	9.3	(0.31, 0.55)
B	TAPC: iTPBIOXD	2.8	22.1	23.4	7.0	(0.43, 0.54)
C	TCTA: 24iPBIOXD	3.0	10.1	10.6	4.0	(0.26, 0.43)
D	TCTA: iTPBIOXD	3.0	10.6	10.2	3.9	(0.25, 0.44)
E	mCP: 24iPBIOXD	3.8	0.8	0.66	0.65	(0.19, 0.20)
F	mCP: iTPBIOXD	3.8	0.9	0.74	0.63	(0.20, 0.23)

^aTurn-on voltage at 1 cd/m^2 .

^bMaximum current efficiency.

^cMaximum power efficiency.

^dMaximum external quantum efficiency.

to 3.8 eV. The as high as 0.48 eV hole-injection barrier between TCTA and mCP lead to the highest operating voltage in devices E and F. The EL performance trend was in consistent with the values of HOMO/LUMO energy offsets, and TAPC-analog bearing the highest driving forces for convenient exciplex formation demonstrated the best highest EL efficiency. The best performance was attained from device A with TAPC:24iPBIOXD exciplex, corresponding to a maximum current efficiency (CE), power efficiency (PE), and external quantum efficiency (EQE) of 28.8 cd/A, 32.3 lm/W, and 9.3%. And the TAPC:iTPBIOXD based device B demonstrated slightly poorer EL performance with maximum current efficiency, power efficiency and external

quantum efficiency of 22.1 cd/A, 23.41 m/W, and 7.0%, respectively. Device C and D based on TCTA electron donor showed comparable EL efficiency, with maximum EQE of 4.0 and 3.9% for 24iPBIOXD and iTPBIOXD electron acceptors, respectively. The EL performance for mCP device was rather poor, with maximum EQE of <1% in both device E and F. As depicted in **Figure 9B**, devices A and B with TAPC donor depicted smooth exciplex-TADF emission, with EL peak at 519 and 556 nm, respectively, which is in agreement with the relevant PL spectra. Commission Internationale de L'Éclairage (CIE) values for device A and B was measured at (0.31, 0.55) and (0.43, 0.54), corresponding to green and yellow emission,

respectively. The TCTA based device C and D both exhibited blueish-green emission, with a gentle shoulder peak at around 400 nm for 2*i*PBIOXD. The two mCP-based devices displayed blue emission with CIE *x*, *y* each at ~0.20. However, the EL spectra of device E and F revealed bimodal emission showing comparable intensity for the two peaks. It is hypothesized that the inadequate HOMO and LUMO energy offsets (< ~1 eV) for TCTA:2*i*PBIOXD, mCP:2*i*PBIOXD, and mCP:*i*TPBIOXD, resulted in the unexpected shorter wavelength EL emission peak, which was ascribed to pure mCP emission (Chiu and Lee, 2012; Shahaliazad et al., 2017). In addition, since the E_{CT} of TAPC and TCTA based exciplexes were lower than the triplet energy of both donor and acceptor materials, which was beneficial to restrict triplet excitons in the exciplex states for the efficient RISC. However, E_{CT} of 2.8–2.96 eV (Figure 7B) for mCP based exciplex was significantly higher than the triplet energy (~2.55 eV) of the two electron acceptors, which provided a potential way for energy leakage from exciplex states to the T_1 excited state of 2*i*PBIOXD and *i*TPBIOXD. Thus, devices based on mCP donors demonstrated the lowest EL efficiency and inadequate TADF emission.

CONCLUSION

In summary, we have designed and synthesized two universal *N*-linked benzimidazole/oxadiazole hybrid electron acceptors through a simple nucleophilic substitution reaction. Diverse deep-blue to yellow emissive exciplex could be formed between various conventional donor materials and the two acceptors due to their deep HOMO levels of ~6.15 eV. The HOMO and LUMO energy level offsets which were also named as the driving forces for exciplex formation were gradiently increased from 0.47 to 1.12 and 0.6 to 1.34 eV in mCP, TCTA, to TAPC based exciplexes. We have found that both HOMO and LUMO offsets ≥ 1 eV was required to form efficient and stable intermolecular charge transfer exciplex. When the driving forces were as low as

0.47–0.75 eV, which is far <1 eV, the two mCP based exciplex demonstrated considerably short delayed component lifetime, with values of only 42 and 72 ns for 2*i*PBIOXD and *i*TPBIOXD acceptors, respectively. Additionally, the exciplex-type device EQE was lower than 1%. When the driving forces were slightly lower or approaching 1 eV, the two TCTA exciplexes displayed moderate EL efficiency of about 4%. And the best EL performance was achieved in TAPC containing exciplex-type TADF OLEDs, with relatively low turn-on voltage of 2.8 V, maximum efficiency of 28.8 cd/A CE, 32.3 lm/W PE, and 9.3% for 2*i*PBIOXD acceptor and 22.1 cd/A CE, 23.4 lm/W and 7.0% for *i*TPBIOXD acceptor. Our results provide guidance on the exploration of efficient exciplex type TADF OLEDs.

AUTHOR CONTRIBUTIONS

WY, MZ, and DH designed and synthesized the materials. WY and MZ did most of the experimental work and data analyses. OLED device fabrication and electroluminescent performance studies were carried out by HY and NS. YT had the idea, led the project. WY and YT prepared the manuscript. All authors contributed to the manuscript preparation.

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SUPPLEMENTARY MATERIAL

The Supplementary Material for this article can be found online at: <https://www.frontiersin.org/articles/10.3389/fchem.2019.00187/full#supplementary-material>

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Conflict of Interest Statement: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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